

Stamping nanoparticles onto the electrode for rapid electrochemical analysis in microfluidics

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Supplement Information

Additional experimental details, figures, and tables are provided to support the content in the main text.

Supplemental Figures

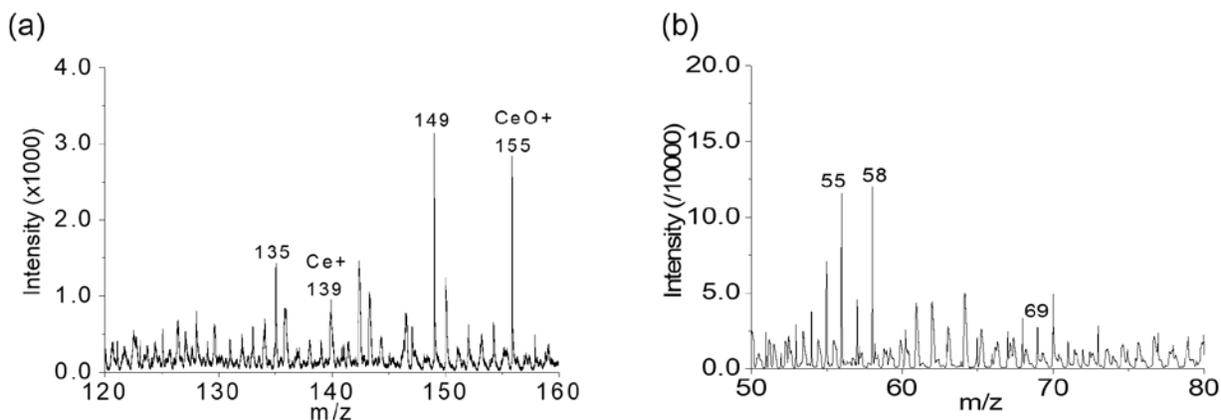


Figure. S1. ToF-SIMS Spectral plots from the stamped SiN window: (a) surface spectrum of the SiN membrane with CeO₂ sequence stamping and (b) surface spectrum of the SiN membrane with graphite applied by mixed stamping.

Figure S1 depicts ToF-SIMS spectral showing mass fragment peaks of targeted nanoparticles, Ce⁺ m/z⁺ 139.9178 and CeO⁺ m/z⁺ 155.8964 show cerium fragment peaks from the CeO₂ nanoparticle (Figure S1a). Fewer significant numbers were used in the spectra due to space constraints. Figure S1b shows graphite fragment peaks C₄H₁₀⁺ m/z⁺ 58.0604 and C₅H₈⁺ m/z⁺ 69.0462, and both are previously reported in graphite electrode studies [1, 2].

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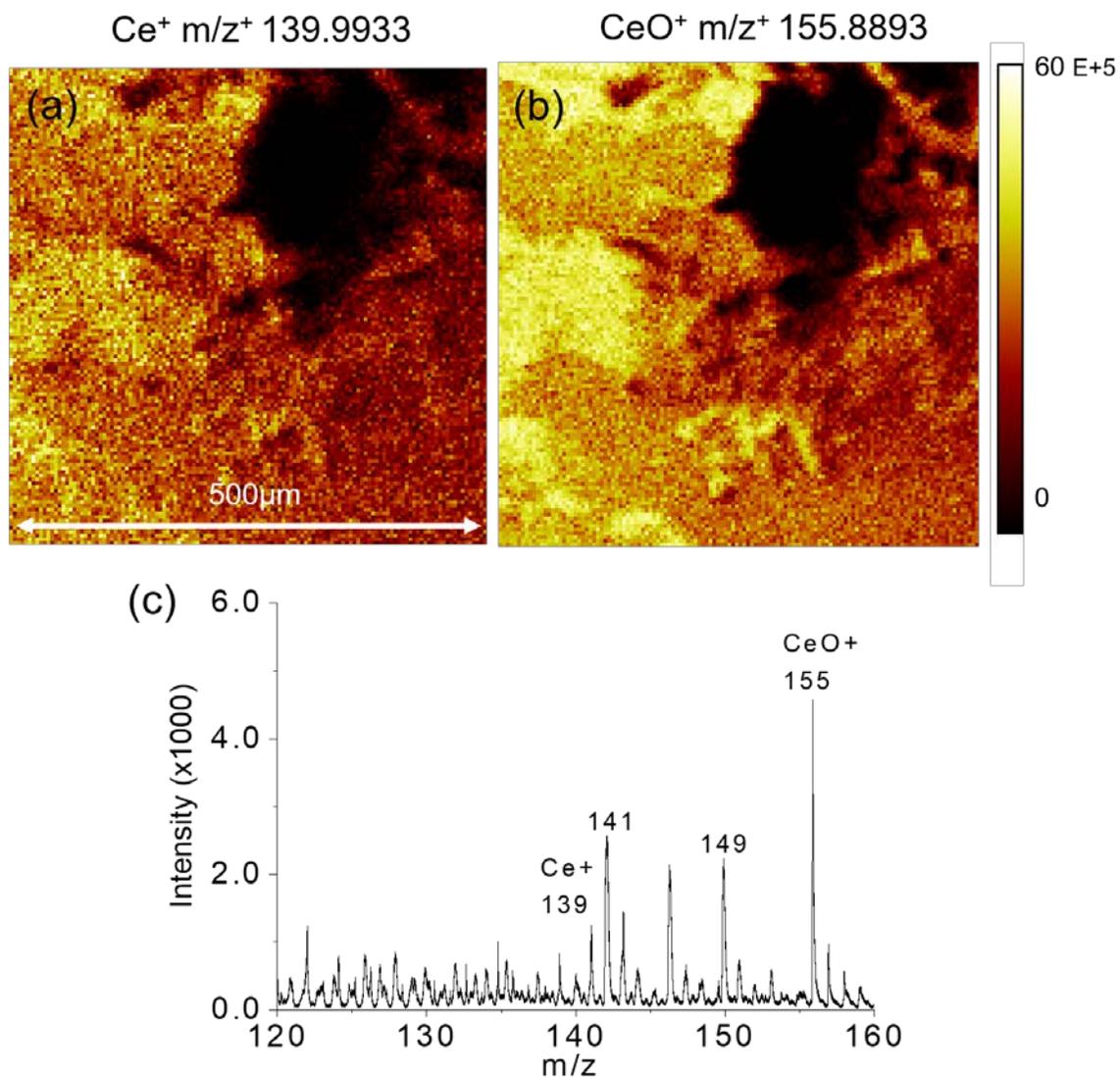


Figure. S2. ToF-SIMS 2D images and a spectrum plot of the CeO_2 mix stamped on SiN membrane, (a) SIMS 2D image of the Ce^+ , (b) CeO^+ , and (c) spectral plot showing Ce^+ ion related peaks.

Fewer significant numbers were used in the spectra due to space constraints.

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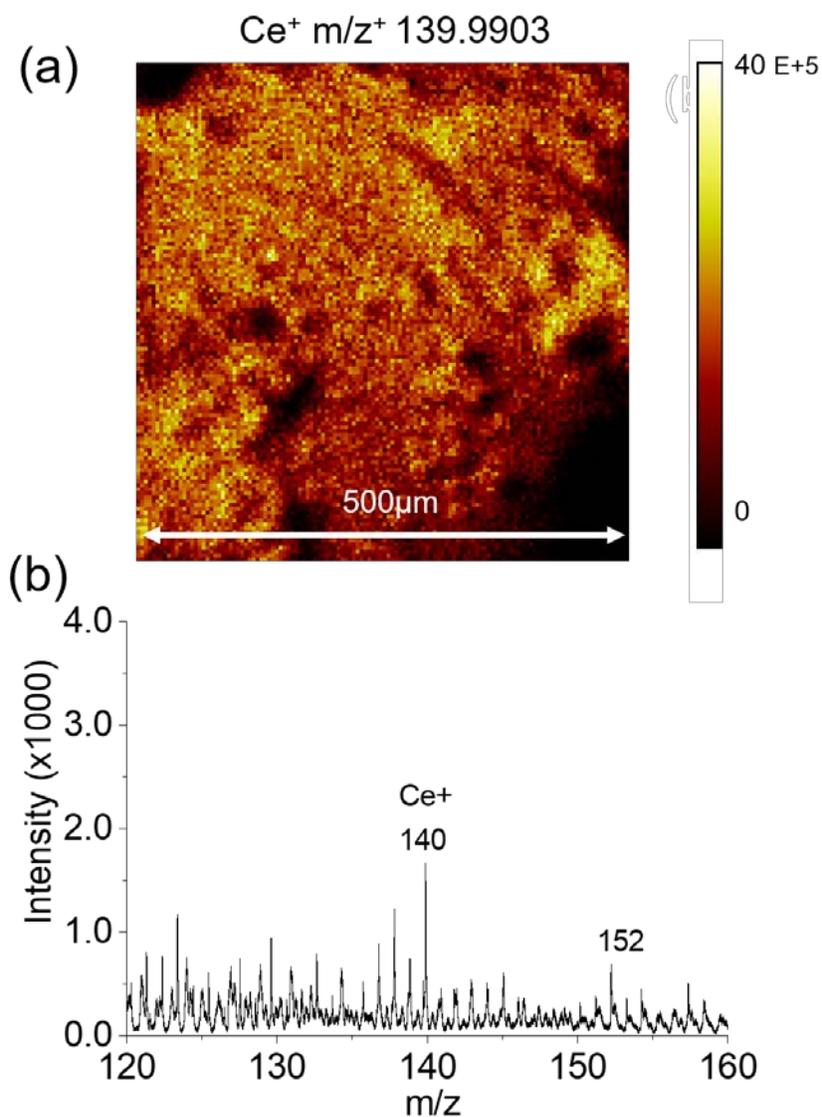


Figure S3. ToF-SIMS 2D images and a spectrum plot of the CeO_2 droplet stamped on SiN membrane, SIMS 2D image of Ce^+ (a) spectral plot showing Ce^+ ion related peaks.

Note that the $\text{Ce}^+ m/z^+ 139.9178$ peak has different ion counts as Ce^+ in the [Figure S3c](#) spectral plot because the sample surface is not as smooth. Fewer significant numbers were used in the spectra due to space constraints.

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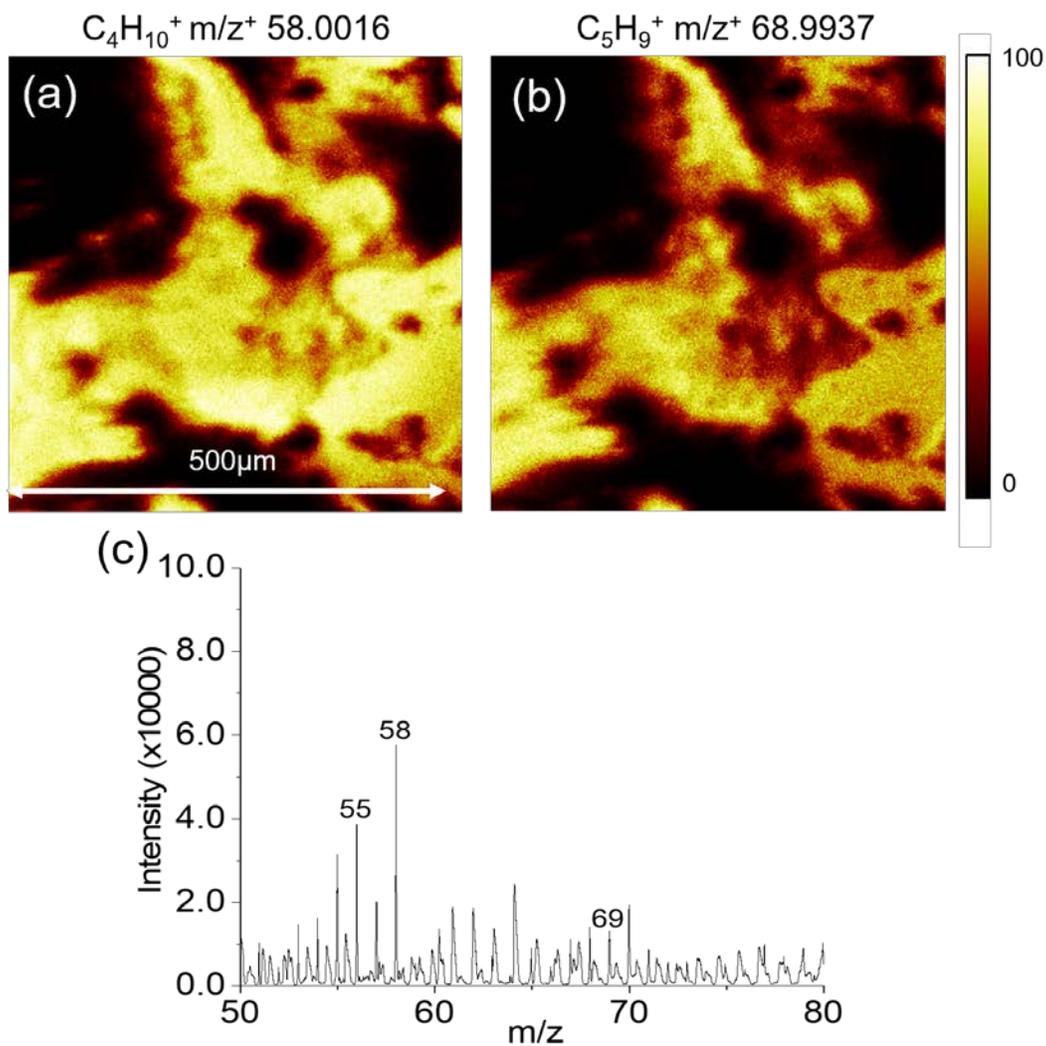


Figure. S4. ToF-SIMS 2D images and a spectrum plot of the graphite sequence stamped on SiN membrane, SIMS 2D image of the $C_4H_{10}^+$ (a) and $C_5H_9^+$ (b), and (c) spectral plot showing graphite fragment related peaks.

Fewer significant numbers were used in the spectra due to space constraints.

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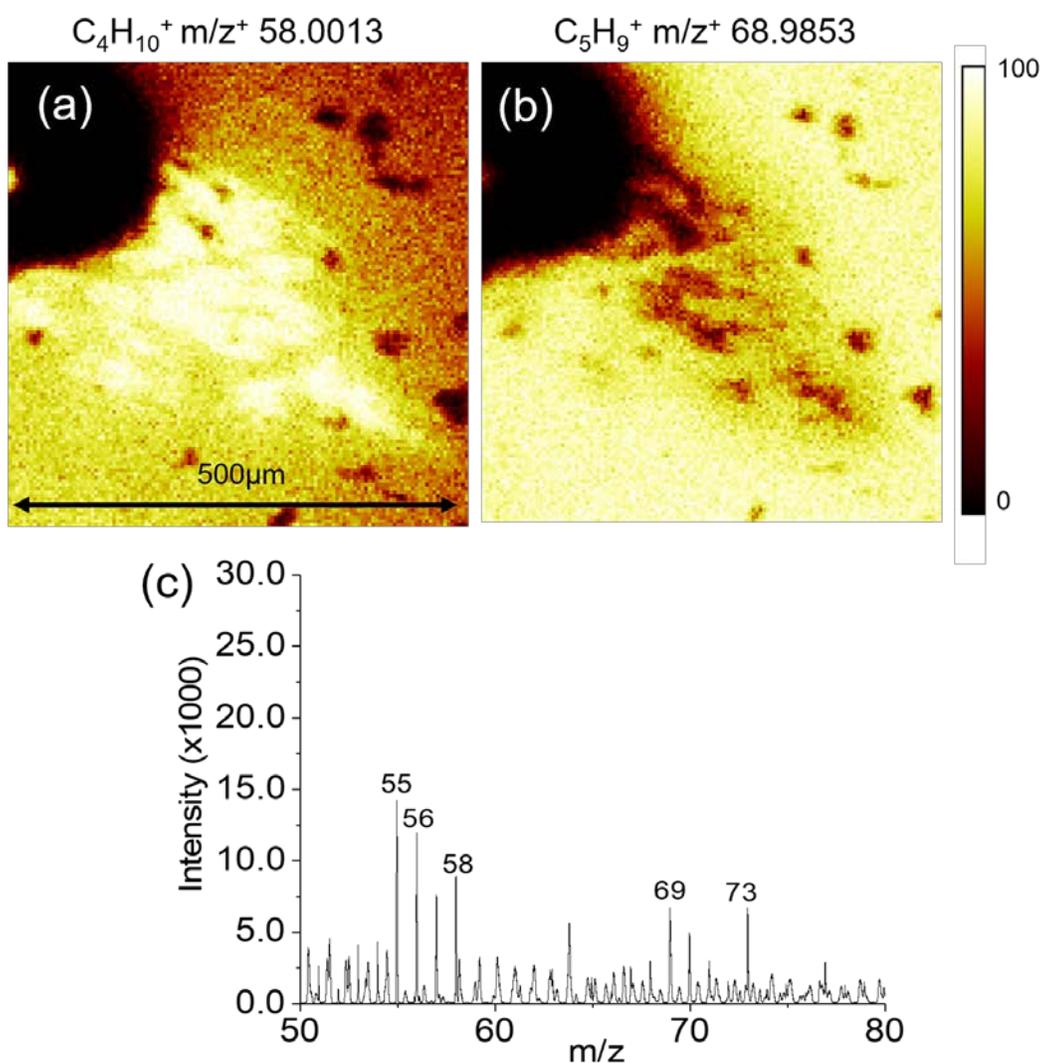


Figure. S5. ToF-SIMS 2D images and a spectrum plot of the graphite droplet stamped on SiN membrane, (a) SIMS 2D image of the $C_4H_{10}^+$ and (b) $C_5H_9^+$, and (c) spectral plot showing graphite fragment ion related peaks.

Fewer significant numbers were used in the spectra due to space constraints.

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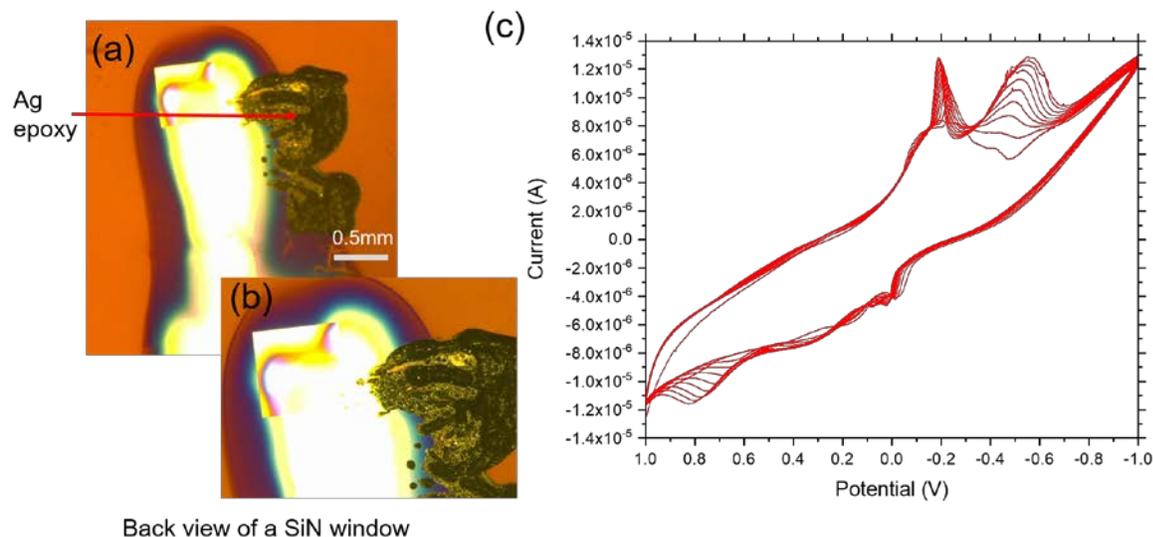


Figure. S6. Optical image of silver epoxy stamped SiN membrane and Cyclic Voltammetry (CV) scan of the silver epoxy control device of 50 x (a) and 200 x (b) magnification, and (c) CV scan plot.

Figures S6a-b show optical microscope images of Ag epoxy stamped SiN and gold electrode boundary of 50x and 200 x magnifications, respectively; and the square structure is a suspended SiN membrane window (0.5 mm x 0.5 mm). Figure S6-c shows the CV scan profile of Ag epoxy applied control device. It has two major peaks at -0.2 V, and -0.5 V during the 1 V to -1 V sweep. Then in the -1 V to 1 V) sweep, there are peaks at -0.1 and 0.8V. This is a different CV profile compared to the blank control device (Figure S7-c). Figure S6 shows similarities between -0.1 V and 0.8 V during the -1 V to 1 V sweep. In Figure S7c there is a major peak at 0.8 V in the 1 V to -1 V sweep; and there are two peaks at -0.1 V and 0.8V during -1 V to 1 V sweep. This represents a CV profile of the Ag epoxy control device inherent of the blank control device. The new peaks are due to electrochemical reaction products characteristic of Ag epoxy.

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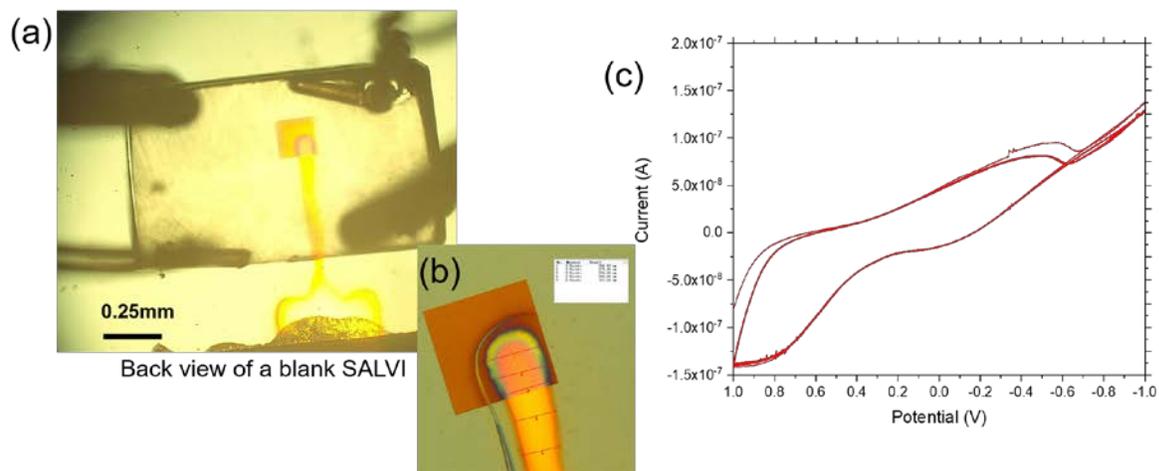


Figure. S7. Optical image of blank SiN membrane with Au electrode and CV scan of the blank control device of 50 x (a) and 200 x (b) magnification, and (c) CV scan plot.

Figure S7c shows CV scan profile of the blank System for Analysis at the Liquid-Vacuum Interface (SALVI) device. Peaks from this plot represent the background characteristic of the device which has similarities to [Figures 4-5](#) of the main text.

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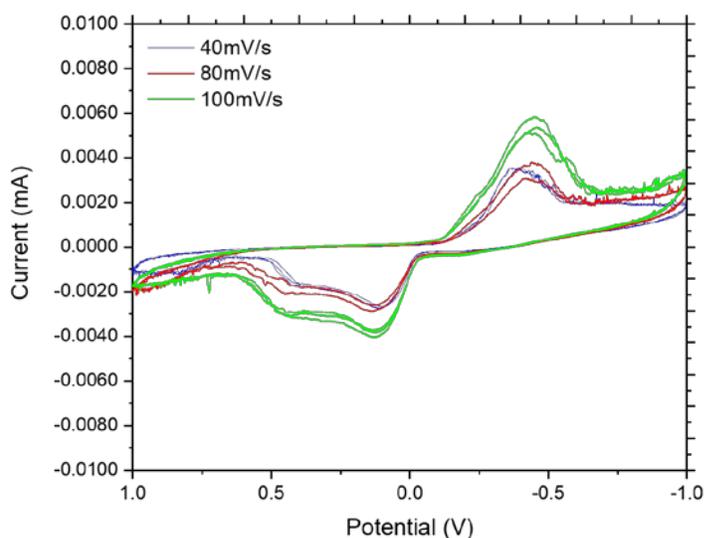


Figure. S8. CV scans of a sequence stamp device with different scan rates, i.e., 40 mV/s, 80 mV/s, and 100 mV/s.

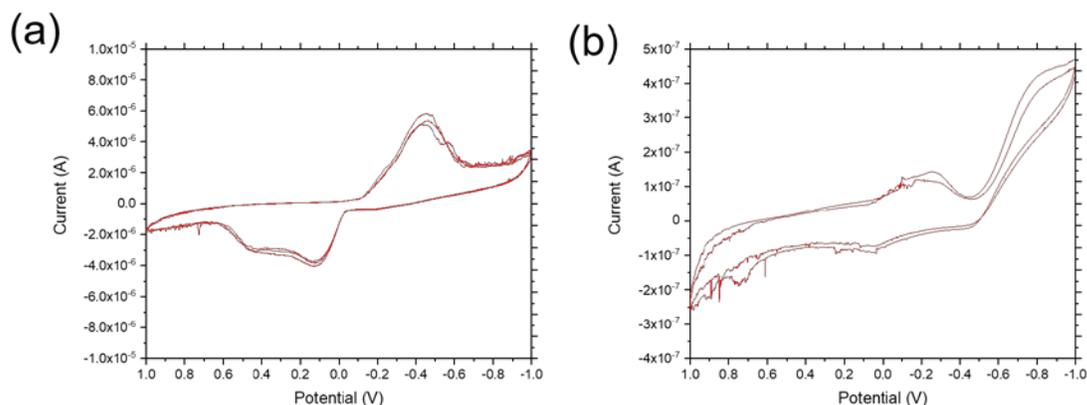


Figure. S9. Comparison of CV scan of the two sequence stamp devices using the scan rate of 100 mV/s: (a) the same device as shown in the main test and (b) a different device.

Figure S8 shows CV plots of the CeO₂ sequence stamped device using different scan rates (i.e., 40, 80, 100 mV/s) from the same device shown in Figure 4-a. Figure S9 gives a CV comparison between two CeO₂ sequence stamped devices with a scan rate of 100 mV/s, namely, a different device and the device from Figure 4-a. Figure S8 shows the same CV characteristics as in Figure 4-a CeO₂ using the sequence stamped device. Results in Figure S9-b also share similar peak characteristics as those in Figure S9-a, which is same plot as Figure 4-a. Similarity peaks are -0.4 V (sweeping from 1 V to -1 V) and 0.1V (sweeping from -1 V to 1 V). Slight noise is seen in Figure S9-b due to some impurity on the electrode surface which can be improved in future work.

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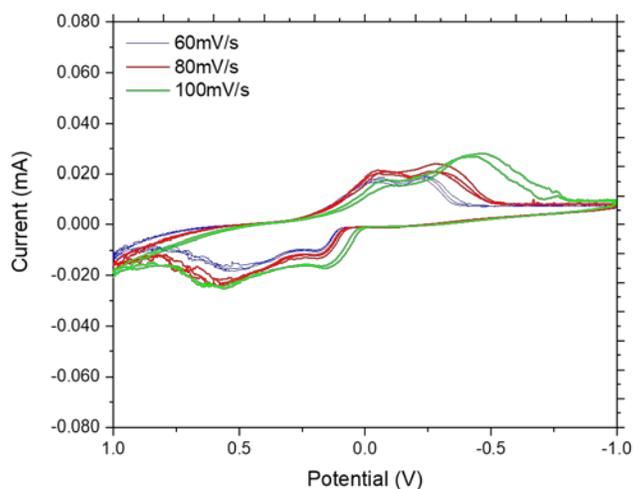


Figure. S10. CV scans of a mix stamp device with different scan rates, i.e., 60 mV/s, 80 mV/s, and 100 mV/s.

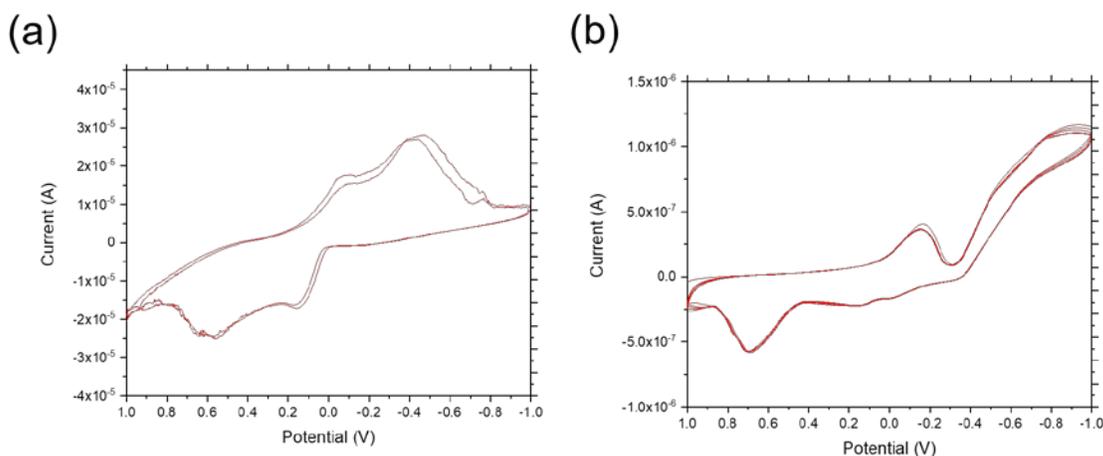


Figure. S11. Comparison of CV scan of the two mix stamp devices using the scan rate of 100 mV/s: (a) the same device as shown in the main test and (b) a different device.

Figure S10 shows CV plots of the CeO_2 mix stamped device using different scan rates (i.e., 60, 80, 100 mV/s) from the same device shown in Figure 4-a. Figure S11-b gives a CV comparison between two CeO_2 mix stamped devices with the scan rate of 100 mV/s, namely, a different device and the device from Figure 4-a. Figure S10 shows the same CV characteristics as in Figure 4-a CeO_2 using the mix stamped device. Results in Figure S11-b also share similar peak characteristics as those in Figure S11-b, which is same plot as Figure 4-a. Similarity peaks are -0.1 V (sweeping from 1 V to -1 V) and 0.2 V (sweeping from -1 V to 1 V). Slight noise is seen in Figure S11-b due to some impurity on the electrode surface which can be improved in future work. Figure S11-b also shows a slight left shift of 0.55V (sweeping from -1 V to 1 V) to 0.6 V (sweeping from -1 V to 1 V), which can be explained as potential shift due to slight physical difference of nanoparticle applied electrode.

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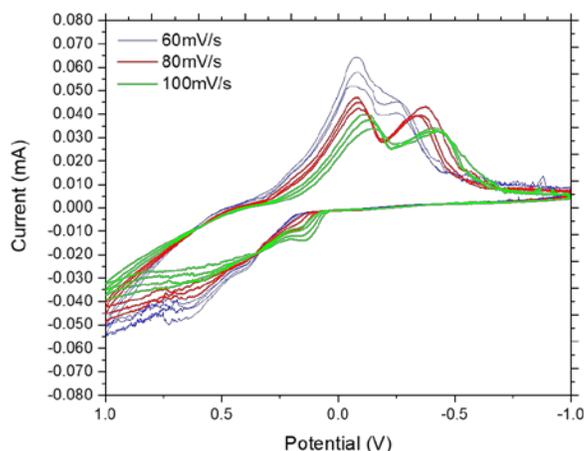


Figure. S12. CV scans of a droplet stamp device with different scan rates, i.e., 60 mV/s, 80 mV/s, and 100 mV/s.

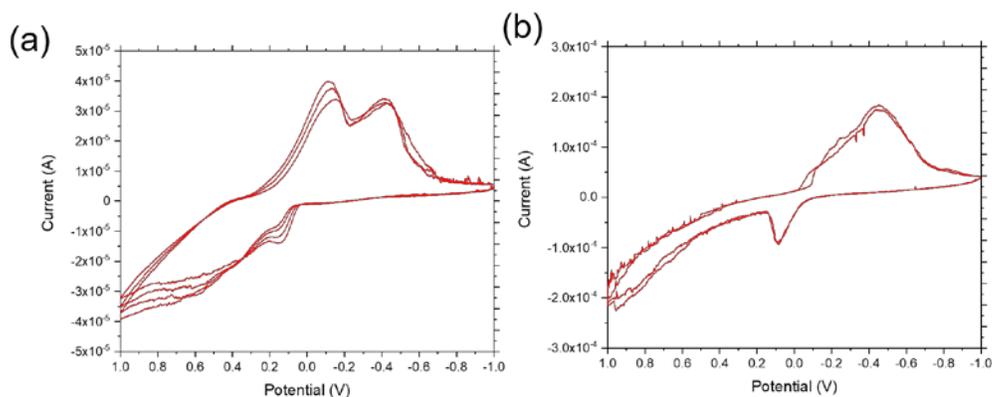


Figure. S13. Comparison of CV scan of the two droplet stamp devices using the scan rate of 100 mV/s: (a) the same device as shown in the main test and (b) a different device.

Figure S12 shows CV plots of the CeO_2 droplet stamped device using different scan rates (i.e., 60, 80, 100 mV/s) from the same device shown in Figure 4-a. Figure S13-b gives a CV comparison between two CeO_2 droplet stamped devices with the scan rate of 100 mV/s, namely, a different device and the device from Figure 4-a. Figure S12 shows the same CV characteristics as in Figure 4-a CeO_2 using the droplet stamp device. Results in Figure S13-b also share similar peak characteristics as those in Figure S4-a. Figure S13-b also shares similar characteristics with Figure S13-a. The latter is same plot from Figure 4-a. Similarity peaks are -0.12 V, -0.42 V (sweeping from 1 V to -1 V), and 0.1V (-1 V to 1 V).

Supplement Information

Reference

1. Smith, T.E., S. McCrory, and M.L. Dunzik-Gougar, *Limited Oxidation of Irradiated Graphite Waste to Remove Surface Carbon-14*. Nuclear Engineering and Technology, 2013. **45**(2): p. 211-218.
2. Deslandes, A., M. Jasieniak, M. Ionescu, J.G. Shapter, C. Fairman, J.J. Gooding, D.B. Hibbert, and J.S. Quinton, *ToF-SIMS characterisation of methane- and hydrogen-plasma-modified graphite using principal component analysis*. Surface and Interface Analysis, 2009. **41**(3): p. 216-224.