

Biosorption and Bioleaching of Heavy Metals from Electronic Waste Varied with Microbial Genera

Table S1. Advantages and Disadvantages of conventional and biological methods for efficient metal recovery from e-waste.

Conventional methods	Advantages	Disadvantages	References
Chemical precipitation	<ul style="list-style-type: none"> • Integrated physico-chemical process, • low capital cost, • simple in operation, • adaptive to high pollutant loads, • efficient for removal of metals and fluorides 	<ul style="list-style-type: none"> • Inefficient for metals present in less amount, • pH dependent • requires chemicals such as lime, oxidants and hydrogen sulfide, • oxidation required if metals are complexed, • maximum sludge generation during the entire process • requires extra cost for sludge management 	[5,6]
Coagulation/Flocculation	<ul style="list-style-type: none"> • Integrated physico-chemical technique, • simple process, • low capital cost, • efficient for removal of soluble, insoluble metals and colloidal particles 	<ul style="list-style-type: none"> • pH dependent • requires non-reusable chemicals, • maximum sludge generation • requires extra cost for sludge management • not efficient for As removal 	[7,8]
Ion exchange	<ul style="list-style-type: none"> • Technologically simple process, • rapid, • efficient process for removal of valuable metals up to ppb levels, • can be integrated with other processes (precipitation and filtration) 	<ul style="list-style-type: none"> • High operational cost (maintenance and initial cost of the resin), • not efficient for dispersal dyes and drugs, • large volume requires large columns, • rapid saturation and clogging of the reactor, • easily contaminated with organic matter, oils • requires physico-chemical pretreatment 	[9]
Filtration	<ul style="list-style-type: none"> • Simple process, • smaller space requirement, • high separation selectivity with fast kinetics, • efficient for removal of wide range of pollutants 	<ul style="list-style-type: none"> • High operational cost due to reactor clogging, • require several types of adsorbents, • regeneration is quite costly and results in loss of material, 	[6]

		<ul style="list-style-type: none"> not efficient for certain metals and dye stuffs 	
Electro-dialysis	<ul style="list-style-type: none"> Simple process, efficient even at high concentration, no chemicals required, efficient for removal of dyes, salts, minerals and dissolved inorganic matter with high separation selectivity 	<ul style="list-style-type: none"> High maintenance cost high energy requirement, specific process, rapid clogging of membrane, limited flow rates low throughput 	[10]
Photo-catalysis	<ul style="list-style-type: none"> Simple, highly efficient process for removal of organic pollutants and metals simultaneously, production of energy 	<ul style="list-style-type: none"> Initial investment costs, formation of dioxins and other pollutants, high operational and maintenance cost, limited applications 	[11]
Biological methods	<ul style="list-style-type: none"> Low cost, high efficiency, potential metal recovery, regeneration of biosorbent, economically attractive, non-requirement of additional nutrient, utilization of large number of microbial species either in pure or mixed cultures, environment-friendly, minimization of sludge 	<ul style="list-style-type: none"> Possibility of bio-transformation or biodegradation, slow process, complexity of microbiological mechanisms, necessary to maintain optimized conditions, possible sludge bulking and foaming 	[12–14]

Table S2. Research studies emphasizing the potential of biological remediation of the E-waste.

Physico chemical practices	Biological practices	Remarks
<ul style="list-style-type: none"> Hydrometallurgy: Gold mining involves use of leach reagent i.e. cyanide <ul style="list-style-type: none"> Exhibits high gold recovery Robust and low cost environmental damage mismanagement produces noxious side products which tempers recycling process <ul style="list-style-type: none"> Cu, Pb and Sn recovery from scrap printed circuit boards (PCBs) by combination of leaching, electrochemical ion exchange and electrodeposition Use of aq. HNO₃ stripping solution for Cu and Pb extraction Sn precipitation as 2SnO₃ at high acid concentrations (>4 M dm⁻³) 	<ul style="list-style-type: none"> No leaching agents are required Microorganisms themselves act as leaching agents Environment-prudent No huge capital investment is required 	<p>[25]</p> <p>[26]</p>

- Poor current efficiency due to cathodic Pb deposition
 - Requirement of huge capital investment
 - Physical processes require low capital
 - suffer from huge metal loss (10-35%)
 - brominated flame retardant dust problem
 - No Brominated flame retardant dust problem was observed
- [27]
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