

Article

Phytochemical Analysis Using cLC-DAD, Nutritional Importance and Assessment of Antioxidant, Antidiabetic and Anticholinesterase Activities of *Ruta tuberculata* Forssk Organic Extracts

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Abstract: Among the Mediterranean aromatic plants, *Ruta tuberculata* Forssk. (Rutaceae) has been widely used as a traditional natural remedy against various disorders resulting from its divers' pharmacological virtues. The aim of this study is to characterize for the first time the phenolic profile of its ethyl acetate (EtOAcE) and acetonic (AcE) extracts and to screen their in vitro antioxidant, antidiabetic, and neuroprotective activities. Phenolic content was determined using spectrophotometric and cLC-DAD analysis. Pharmacologically, in vitro antioxidant power was evaluated using six different antioxidant methods. Moreover, the antidiabetic and neuroprotective capacities were assessed in vitro by determining the α -amylase, α -glucosidase, and acetylcholinesterase inhibitory activities. Phytochemically, the highest flavonoid content was found in EtOAcE where the major identified compounds were myrecetin, rutin, sylimarin, naringenin, and quercetin. In presence of other phenolic acids, gallic acid was exclusively detected in AcE. Furthermore, both *R. tuberculata* extracts showed significantly remarkable antioxidant activities, especially the EtOAcE. Interestingly, AcE strongly inhibited the acetylcholinesterase and α -glucosidase, with the respective IC₅₀ values of 20.48 ± 0.2 and 104.5 ± 1.8 μ g/mL. In this study, we also reported the nutritional quality associated with the identified phytochemicals. *R. tuberculata* organic extracts may offer exciting reserves to achieve new anti-diabetic and anti-Alzheimer drugs which have also antioxidant potential.

Keywords: *Ruta tuberculata*; antioxidant; anti- α -amylase; anti- α -glucosidase; anticholinesterase; phenolic analysis

1. Introduction

Diabetes mellitus (DM) is an epidemic disease, that raises the most public health challenges in the entire world. Considering its current alarming incidence rate, the world health organization predicted that in the year 2030, DM will expand to be the seventh major cause of death worldwide [1]. In this metabolic disorder, the blood glucose level increases markedly as a result of insufficient insulin synthesis or secretion or the inefficiency in binding to its receptor (IR). Indeed, children, adults, and pregnant women were the frequent targets of types (1,2), and gestational diabetes [2,3]. Indeed, it has been documented that hyperglycemia but also fasting hypoglycemia may enhance the Alzheimer's incidence in T2DM cases [4]. Besides, these distinct illnesses may share the common causes of their progression such as insulin signaling pathway, glucose metabolism relationship, chronic inflammation, and diseases associated with oxidative stress [4–6]. Indeed, it was reported that the disruption of insulin signaling in muscle or glucose transport via GLUTs in neurons may be involved in the progression of many neurodegenerative diseases such as AD [4,5]. Indeed, the high circulating levels of glucose and lipids may increase the accumulation of amyloid beta ($A\beta$) oligomers. These oligomers can markedly inhibit the auto-phosphorylation of insulin receptors (IR) in dendrites of hippocampal neurons, which in return decrease the ability of insulin to stimulate specific signaling pathways in the AD brain, such as the PI3K/AKT/m-TOR pathway, promoting synaptic loss [5].

In addition, chronic inflammatory phenomena and oxidative stress are presumed as the two key risk factors related to diabetes and AD [5–7]. However, some undesirable effects may be seen in T2DM and AD patients resulting from the daily consumption of synthetic anti-diabetic and/or anti-cholinesterase inhibitor drugs as a basic treatment. Moreover, the excessive use of some non-steroidal anti-inflammatory and antioxidant drugs, to reduce the complications of those pathologies, often may increase the severity of the medication's side effects [3,6,8,9].

Recently, many pharmacological investigations demonstrated that natural compounds can display a key role in the prevention of chronic diseases [9]. Indeed, the free radicals' overproduction or their accumulation may cause or involve complicating factors in several disorders, in particular, Alzheimer's and diabetes mellitus injuries as well as cancer [9–11]. Nowadays, the recent poly-pharmacology trend is to elaborate on new drugs that can target common biological pathways involving at once in AD and T2DM. Indeed, the researchers are showing an increased interest in medicinal plants and their benefic effects as the desired resource of natural drugs which are able to enhance or prevent various chronic illnesses and degenerative diseases [3,12,13]. In fact, herbs have been reported to exhibit potent pharmacological acts including antioxidant, anti-inflammatory, antidiabetic, and anticancer activities [3,6,14]. Moreover, their neuroprotective and anti-cholinesterase capacities were newly documented in the absence of available therapies for the management of progressive neuron impairment and neuronal loss [4,15].

Ruta tuberculata Forssk. is a perennial aromatic herb that belongs to the *Ruta* (*haplopyllum*) genus from the Rutaceae family which has been growing in all Mediterranean regions, especially the tropical and temperate areas [7,13,16]. Based on green foliage with yellow flowers, more than 1800 plants were classified as *Ruta* species [9,13,17]. For its safety and benefic effects, Algerian *R. tuberculata* is traditionally used as a natural therapy in the treatment of skin pathologies, and gastric and intestinal disorders, and as calming and analgesic agents to prevent some neurological problems in children [9,13,16]. Furthermore, this plant was likely prescribed to also treat divers' metabolic disorders such as diabetes, cardiovascular and hypertension illness and to alleviate rheumatism, tonsillitis, menses disturbance, and infertility in women [18]. In fact, many investigations affirmed that some

Ruta species seemed to have various therapeutic virtues and antioxidant potential with an attractive capacity to preserve diverse injuries resulting from oxidative stress, through the inhibition of free radical production mechanisms and enhancement of antioxidant signaling pathways [16,19,20]. Moreover, their extensive anti-inflammatory and neuroprotective effects have also been studied in different neurodegenerative disorders models, particularly, cerebral ischemia, and Parkinson's and Alzheimer's diseases [6,13,15,20,21]. In fact, it has been reported that *Ruta* plants revealed mostly a high level of essential oils [16,20], coumarine derivatives and alkaloids [12], lignanes [22], and polyphenols [6,19,23]. Indeed, the beneficial properties, which *Ruta* species displayed, were demonstrated to be related to their content in diverse classes of polyphenols [9,13,23,24]. Alternatively, the effective lack of studies that provide information in-depth on the phenolic characterization of Algerian *R. tuberculata* extracts was noted in literature.

Phenolic compounds could also act as a protective metabolic barrier within the intestinal mucosa where the nutritional quality of them is probably related to their remarkable physicochemical and pharmacological properties, leading to the prevention of several diseases and helping to digest easily complex macromolecules [11].

The aims of this study are to characterize for the first time the phenolic profile of the ethyl acetate and acetonic extracts from Algerian *R. tuberculata* aerial parts and to assess their in vitro antioxidant, anti-diabetic, and neuroprotective activities.

2. Materials and Methods

2.1. Chemicals

Methanol and other organic solvents (HPLC-grade) used in the present work were purchased from Sigma-Aldrich. Acetylthiocholine iodide (ACI) reagent was purchased from BioChemica. Pure phenolic standards, Acetylcholinesterase (AChE), Yeast α -glucosidase enzyme, Pancreatic α -amylase enzyme (1U), Acarbose, starch, Neocuproine, Galanthamine, 1,1-Diphenyl-2-picryl-hydrazyl (DPPH), 2,2'-azino-bis(3-ethylbenzo-thiazoline-6-sulfonic acid) (ABTS), Butylated hydroxytoluene BHT (C₁₅H₂₄O), Butylate hydroxyanisole BHA (C₁₁H₁₆O₂), Vanillin (C₈H₈O₃), Ascorbic Acid (C₆H₈O₆) and the other chemicals were obtained from Sigma-Aldrich and Fluka.

2.2. Plant Material and Extraction

Ruta tuberculata Forssk. aerial parts (Voucher number: Fl.AA 86 86 1775/TRO-28101397; <https://www.ipni.org/n/775152-1> accessed on 18 July 2022) were obtained in March 2017 from the southern area Rase El-miaade of Ouled djellal-Algeria (latitude 34.1846; longitude 4.44955).

Acetonic (AcE) and ethyl acetate (EtOAcE) extracts of *R. tuberculata* aerial parts were prepared according to the previous methods [25,26]. Firstly, the vegetal material was subjected to a cold maceration in petroleum ether for 4 h in ratio 1:10 (w:v) and then was filtrated. Then, the obtained residues from the petroleum ether extraction were macerated in the EtOAc or acetone in ratio 1:10 (w:v) for 48 days. Acetonic and EtOAc crude extracts were filtered under reduced pressure then dried in the oven at 35 °C and stored at −4 °C. Obtained extracts were weighted to determine the extraction yields using the following equation Equation (1):

$$\% \text{ Yield} = (\text{Weight of dry extract} / \text{Weight of taken plant for extraction}) \times 100. \quad (1)$$

2.3. Phytochemical Investigation

2.3.1. Spectrophotometric Determination of Total Phenolic Content

The total polyphenols content (TPC) was determined using the standardized Folin-Ciocalteu's method [27]. The mixture containing 1 mL of Folin-Ciocalteu's (10% in distilled H₂O) and 200 μ L of each extract at different concentrations or gallic acid (20–200 μ g/mL) was prepared and stored for 5 min before the addition of carbonate sodium (Na₂CO₃)

solution (7.5%, 800 μ L). Absorbances were readied at 765 nm after an incubation period of 2 h at room temperature.

Total flavonoid content (TFC) was estimated by following yellowish complex (AlCl_3 -flavonoid) formation as previously described [28]. A mixture of 1 mL of aluminum chloride (AlCl_3) solution and 1 mL of each extract or quercetin as standard (4–20 $\mu\text{g}/\text{mL}$) was prepared and measured at 430 nm. After an incubation period of 10 min in dark, TPC value of the studied extracts was determined according to the linear regression equation of quercetin curve ($y = 0.036x + 0.127$; $R^2 = 0.997$) and the results were expressed as mg quercetin equivalent (mg QE/g d.E).

The quantification of the total condensed tannins content (CTC) present in *R. tuberculata* crudes extracts was carried out spectrophotometrically [29]. Briefly, 1.5 mL of vanillin solution (4%) was added to the reaction mixture of 25 μL of each extract or catechin as standard (20–200 $\mu\text{g}/\text{mL}$) and 750 μL of hydrochloric acid HCl (30%). After 15 min of the incubation period, the absorbances of the mixtures were determined at 500 nm. CTC amount was calculated from the calibration equation of catechin curve ($y = x 0.0022 + 0.043$; $R^2 = 0.999$). Results were expressed in mg catechin equivalent per g of dried extract (mg CE/g d.E).

2.3.2. Characterization of Phenolic Compounds Using cLC-DAD Analysis

The chromatographic screening of the studied extracts was carried out as described by Erenler et al. [30]. The mobile phase consisted of eluent A which contained 0.1% formic acid in water and eluent B, containing methanol (HPLC-grade). A volume of 20 μL from the mobile phase was injected with a flow rate of 0.7 mL/min at 40 °C. The extracts and standards to be analyzed have been solubilized at a dose of 1 mg/mL in a methanol HPLC-grade and then filtrated using a cellulose filter membrane (part no. 3150-0576, 47 mm, pore size 0.45 μm , 100/pk, Germany).

2.4. In Vitro Antioxidant Assay

2.4.1. DPPH Free Radical Test

The scavenging activity of its crude EtOAc and acetonic extracts of the plant was assessed using the spectrophotometric method of Blois [31], based on the color turn of the purple solution of the stable free radical 1,1-Diphenyl-2-picryl-hydrazyl (DPPH) to yellow-colored solution in presence of an antioxidant substance. The synthetic antioxidants BHA and BHT (0.78–50 $\mu\text{g}/\text{mL}$) or pure phenolic compounds gallic acid and quercetine (10–200 $\mu\text{g}/\text{mL}$) were used as references. In microplate 96 wells, Methanol DPPH solution 0.004% (160 μL) was added to a volume of 40 μL of each sample (12.5–400 $\mu\text{g}/\text{mL}$) or antioxidant standards. The reaction mixing solution was agitated and preserved in the dark for 30 min at ambient temperature. Absorbance has been read at 517 nm. Decreased absorbance of the mixing solution signified the higher free radical-scavenging capacity of the tested sample where the concentration which provided 50% extinction of the DPPH radical (EC_{50}) was calculated using the Equation (2) resulting from the plotting of the extinction percentage inhibitions:

$$\text{I\%} = [(A_{\text{control}} - A_{\text{sample}}) / A_{\text{control}}] \times 100 \quad (2)$$

A_{control} is the absorbance of blank and the A_{sample} is that of extract or standard sample.

2.4.2. ABTS Scavenging Assay

The hydrogen donating capacity of AcE and EtOAcE was evaluated even against the ABTS (2,2-azino-bis (3-ethylbenzothiazloine-6-sulfonic acid)) cation radicals as described in the literature [32], with slight modifications. $\text{ABTS}^{\cdot+}$ solution occurred by mixing ABTS (2 mM in H_2O) with 2.45 mM potassium persulfate ($\text{K}_2\text{S}_2\text{O}_8$). The resulting solution was stored in the dark for 16 h then it was adjusted to obtain an absorbance value of 0.700 ± 0.02 at 734 nm. A mixture consisting of 160 μL of the ABTS cation solution and 40 μL of each extract (12.5–400 $\mu\text{g}/\text{mL}$) or the synthetic antioxidants BHA and BHT (0.78–50 $\mu\text{g}/\text{mL}$)

was prepared and incubated for 30 min at room temperature. Absorbance was measured at 734 nm, noting that Equation (2) was used to calculate the inhibition rate.

2.4.3. Total Antioxidant Capacity by Phosphomolybdenum Assay (TAC)

The total antioxidant activity of the studied extracts was investigated by measuring their ability to reduce the molybdate ion Mo (VI) to form a green phosphate/Mo (V) complex at acid pH according to the previous method [33]. A volume of 100 μ L of each sample was added to 1 mL of reagent solution containing 4 mM ammonium molybdate and 28 mM sodium phosphate in 0.6 M sulfuric acid. The reaction mixture was incubated for 90 min at 95 °C. After cooling the mixture at room temperature, the absorbance was read at 695 nm. Ascorbic acid (10–300 μ g/mL) was used as an antioxidant and its curve was used to estimate the total antioxidant capacity.

2.4.4. Ferric Reducing Antioxidant Power (FRAP) Test

This test was performed according to the Oyaizu [34] method, with minor modifications. Synthetic antioxidants BHA, BHT and ascorbic acid (0.78–50 μ g/mL) with pure phenolic compounds gallic acid and quercetin (2–50 μ g/mL) were employed as standards. A volume of 1% potassium ferricyanide ($K_3Fe(CN)_6$) was added to a mixture consisting of sodium phosphate buffer (0.2 M, pH 6.6) and 10 μ L of each extract (12.5–800 μ g/mL) or antioxidant standards. The reaction was incubated for 20 min at 50 °C. To acidify the previous mixture, a volume of 50 μ L trichloroacetic acid 10% (TCA) was added. After the addition of 40 μ L of distilled water and 10 μ L of 0.1% $FeCl_3$ to this mixture, the reading of absorbance was done at 700 nm.

2.4.5. Cupric Reducing Antioxidant Capacity (CUPRAC)

The potential of the corresponding extracts to reduce the Copper (II) to copper (I) ions was evaluated using CUPRAC bioassay [35] by mixing a volume of 50 μ L of copper (II) chloride solution (10 mM) with 50 μ L of neocuprine in ethanol (7.5 mM) and 60 μ L of ammonium acetate buffer solution (1 M, pH 7.0). In presence of neocuprine, a stable complex with maximal absorbance at 450 nm resulted from this reduction. To the initial mixture, a volume of 40 μ L of each sample (12.5–800 μ g/mL) or standard (0.78–50 μ g/mL) was added to obtain a final volume of 200 μ L. Absorbance was recorded at 450 nm.

2.4.6. β -Carotene/Linoleic Acid Bleaching Assay

The lipid-peroxidation inhibitory activity of the plant extracts was investigated via the previous method previously described [36], whereas, BHT and BHA were used as antioxidant standards. To prepare the β -carotene emulsion, β -carotene (0.5 mg) was dissolved in 1 mL of chloroform in a round-bottomed flask and mixed with 200 mg of Tween 40 and 25 μ L of linoleic acid. Under reduced pressure, chloroform was removed in a rotary evaporator then 100 mL of H_2O_2 was added to this emulsion before vigorous shaking. After adjustment of absorbance to 0.8–0.9 at 470 nm, 160 μ L of this emulsion was added to 40 μ L of studied extracts (12.5–800 μ g/mL) or standards (0.78–50 μ g/mL). The reaction mixture was incubated for up to 120 min at 50 °C. β -carotene bleaching was monitored spectrophotometrically at 470 nm. The antioxidant capacity of β -carotene bleaching was calculated according to previous research work [36].

2.5. Anti-Diabetic Test

2.5.1. Anti- α -Amylase Assay

This assay was achieved using the iodine/potassium iodine (IKI) method [17], with minor modifications. A volume of 50 μ L of α -amylase solution (1U) prepared in phosphate buffer [PBS (pH 6.9 supplemented with 6 mM sodium chloride)] was mixed with 25 μ L of each extract dissolved in MeOH or acarbose, as a positive control, at different concentrations (6.25–400 μ g/mL) in 96-well microplate and incubated for 10 min at 37 °C. 50 μ L of starch solution (0.1%) was then added to initiate the reaction. The reaction mixture was incubated

again for 10 min at 37 °C. The reaction was stopped with the addition of HCl (25 µL, 1M) and 100 µL of IKI solution. Similarly, a blank was prepared by adding the sample solution to all reagents without the enzyme α -amylase solution, and acarbose was used as a positive control. Absorbances were measured at 630 nm, where the absorbance of the blank was subtracted from that recorded by the samples. The α -amylase inhibitory effect was calculated according to Equation (3):

$$\% \text{ inhibition} = 1 - [(Ac - Ae) - (As - Ab)] / (Ac - Ae) \quad (3)$$

Ac: Absorbance of the mixture of starch + IKI + HCl + MeOH + phosphate buffer solutions.

Ae: Absorbance of the mixture of the enzyme + starch + IKI + HCl + MeOH solutions.

As: Absorbance of the mixture of the enzyme + sample + starch + IKI + HCl solutions.

Ab: Absorbance of the mixture of the sample + IKI + phosphate buffer solutions.

2.5.2. Anti- α -Glucosidase Assay

The α -glucosidase inhibitory capacity of EtOAcE and AcE of *R. tuberculata* was determined according to the developed method [37], where acarbose (78.125–5000 µg/mL) and quercetin (0.48–125 µg/mL) were used as a positive control. A volume of 20 µL of each extract (15.625–1000 µg/mL) or acarbose solution at different concentrations was mixed with 100 µL of α -glucosidase solution (0.1 U) prepared in phosphate buffer (100 mM, pH 6.9) and 80 µL PNPG (5 mM) in 96-well microplate and incubated for 30 min at 37 °C. Similarly, a blank was prepared by the addition of sample solution to all reaction reagents without enzyme solution. The reaction was stopped by the addition of the sodium carbonate solution (0.2 M). The absorbance was then read at 405 nm in initial time (0) and after 30 min.

2.6. Anti-Alzheimer Effect (Anti-Cholinesterase Activity)

The neuroprotective effect of the target extracts was performed by determining Acetylcholinesterase (AChE) inhibition capacity as previously described in the literature [38], using Galantamine as a positive control. Briefly, a volume of 20 µL of AChE enzyme solution (5.32×10^{-3} U) was added to a mixture containing 150 µL of sodium phosphate buffer (0.1 M, pH 8.02) and 10 µL of each extract or standard at different concentrations (3.25–200 µg/mL). After an incubation period of 15 min at 37 °C, a volume of 10 µL of the revealing solution of DTNB (5.50-Dithio-bis (2-nitrobenzoic) acid (0.5 mM)) and 10 µL of the substrate of the reaction (acetylthiocholine iodide ACI (0.73 mM)) were added. A blank was prepared by adding the sample solution to all reaction reagents without enzyme solution. The hydrolysis reaction of the acetylthiocholine iodide by acetylcholinesterase was monitored at 412 nm.

2.7. Statistical Analysis

The actual experiments were carried out in triplicates and the results were expressed as mean \pm SD. Graph Pad Prism version 8.0.2 (Graph Pad Software Inc., San Diego, CA, USA) was used for the analysis of variance (One-way ANOVA followed by Tukey's multiple comparison tests). The statistical significance level was identified at $p < 0.05$.

3. Results and Discussion

It is commonly acknowledged that herbal polyphenols and flavonoids possess various therapeutic properties and exhibit antioxidant, anti-inflammatory, antidiabetic, and neuroprotective activities. Against this background, a moderate number of phytochemical and pharmacological data could be available in the literature on *Ruta tuberculata* organic fractions compared to the other *Ruta* species. While no reports concerning the phytochemical composition and the pharmacological properties of the acetonic extract were published for all species of genus *Ruta*, as the best of the author's acknowledgment. Indeed, most studies of *Ruta* plants focused mostly on their alkaloid derivatives and essential oil compounds [12,20,22,39], whereas their phenolic contents and pharmacotherapies of them were

rarely investigated [17,19,21,40]. That heightened our desire to characterize for the first time the individual phenolic profiles of the EtOAc and acetonc extracts of *R. tuberculata* aerial parts and to investigate their pharmacological properties.

3.1. Extraction Yield and Spectrophotometric Determination of Total Bioactive Contents

In the recent study, the maceration method of *R. tuberculata* aerial parts, using different organic solvents, allowed us to obtain greenish semisolid crude EtOAc and acetonc extracts that represented extraction yield values in order of 2.3 ± 0.3 and $1.5 \pm 0.2\%$, respectively. As shown in Table 1, the EtOAc extraction yield was higher than that recorded by the acetonc extraction. In accordance with our results, Zengin et al. [17] proved that ethyl acetate extraction yield (3.5%) of *Hapophyllum myrtifolium* (Rutaceae) was higher than that registered by the other nonpolar extracts such as petroleum ether solvent. Indeed, diverse factors including the genetic patrimony, the collected plant status, climatic conditions, method, and solvent system used for extraction can significantly affect the extraction yield and have an impact often on the chemical diversity between *Ruta* species [18,30].

Table 1. Colorimetric determination of total polyphenol, total flavonoid, and condensed tannin contents in *R. tuberculata* extracts.

Extract	Extraction Yield	TPC	FTC	CTC
	%	mg GAE/g (d.E)	mg QE/g (d.E)	mg CE/g (d.E)
AcE	1.5 ± 0.2^b	40.9 ± 0.2^a	14.3 ± 0.7	39.4 ± 0.1^a
EtOAcE	2.3 ± 0.3	51.0 ± 0.1	16.4 ± 0.9	43.2 ± 0.6

AcE: Results were expressed as Mean \pm SD ($n = 3$). Mean followed by different letters are significantly different (Level of Significance: $^a p < 0.05$, $^b p < 0.01$). EtOAcE: ethyl acetate extract; AcE: acetonc extract; d.E: dry extract.

The phenolic content of EtOAc and acetonc crude extracts from *R. tuberculata* aerial parts were estimated and illustrated in Table 1. Based on the obtained results, both target extracts showed to be rich in various classes of bioactive phenols but with varying levels. In evidence, EtOAc extract revealed the highest significant value ($p < 0.05$) of total polyphenols (51.0 ± 0.1 mg GAE/g d.E) and condensed tannin (43.2 ± 0.6 mg CE/g d.E) than that recorded by acetonc extract. However, no significant difference has been noted in FTC values between the studied extracts which means that they showed a similar amount of total flavonoids.

Our results accord with that reported by Hamdi et al. [40], who proved with overwhelming evidence that the highest TPC and FTC of Tunisian *R. tuberculata* aerial parts were found in their EtOAc fraction with the respective values of 262 mg GAE/g d.E and 99.1 mg QE/g d.E. In fact, they are most three times considerable than our findings. In contrast, the monitored CTC by Hamdi et al. [40] in the same extract is almost four times lower than that obtained. According to this result, Zengin et al. [17] indicated that ethyl acetate extract of *H. myrtifolium* showed nearly a similar TPC value of 52.5 ± 1.5 mg GAE/g d.E to that obtained in *R. tuberculata* EtOAc extract, but it was the richest one in flavonoids (25.6 ± 0.5 mg QE/g d.E). However, the TPC value obtained in our study was also lower than that investigated by Kacem et al. [26] in the same extract of Tunisian *R. chalepensis* aerial parts (142.5 mg GAE/g d.E), as well as the phenolic content was further much smaller in hexane extract (15 mg GAE/g d.E) than that one obtained in the studied AcE. Moreover, the finding results showed that *R. tuberculata* EtOAc extract is richer even in FTC and CTC compared to that found by Kacem et al. [26] in *R. chalepensis* EtOAc extract. Overall, it has also been reported [6] that the EtOAc fraction from Algerian *R. chalepensis* was the second richest extract in TPC (175.23 ± 5.64 GAE/g d.E) and FTC (78.95 ± 1.02 mg QE/g d.E) amounts comparing to the other fractions.

On the whole, several reports offered by Eissa et al. [19], Kacem et al. [26], and Gali and Bedjou [6] indicated the same distribution of the total phenolic and flavonoids contents between the solvent systems but with higher amounts when compared with the obtained

results. Indeed, these components revealed differences in their contents depending to the solvent's polarities, whether TPC and FTC contents increase depending on the polarity of the solvent, which reached the maximum in the ethyl acetate and decreased gradually with the nonpolar solvents as suggesting by previous reports [9,11,17]. Based on our findings, we could assume further that ethyl acetate is the most adequate solvent able to extract a wide range of the middle and nonpolar classes of polyphenols.

3.2. Phenolic Characterization Using cLC-DAD Analysis

Depending on the molecular segregation of the phenolic compounds at four wavelengths and the retention time, the phenolic profiles of the corresponding extracts of Algerian *R. tuberculata* were assessed in Figures 1 and 2, while the results of the individual phenolic quantification were summarized in Table 2. Our findings revealed that various classes of polyphenols were detected in both extracts of *R. tuberculata* aerial parts. However, EtOAc extract showed high amounts of polyphenols, mainly flavonoids. Furthermore, greater amounts of cinnamic derivatives were found in this extract such as trans-cinnamic, *p*-coumaric, caffeic, and trans-ferulic acids, respectively (Table 2). More than benzoic derivatives as hydroxyl-benzoic acid, vanillin acid was also identified in both studied extracts, as well as gallic acid being exclusively detected in the acetonic extract. Moreover, the most abundant phenolic acids in EtOAc extract were trans-cinnamic acid (87.01 mg/100 g d.E) followed by cinnamic acid (74.95 mg/100 g d.E), *p*-hydroxybenzoic acid (53.85 mg/100 g d.E), vanillic acid (37.03 mg/100 g d.E) and *p*-coumaric acid (28.17 mg/100 g d.E) with the nonappearance of gallic acid. Likewise, it was noted that the cinnamic acid (97.41 mg/100 g d.E) is the main phenolic acid in the acetonic extract followed by gallic acid (64.025 mg/100 g d.E), trans-cinnamic acid (86.47 mg/100 g d.E), *p*-coumaric acid (35.97 mg/100 g d.E), vanillin acid (18.20 mg/100 g d.E) and *p*-hydroxybenzoic acid (14.06 mg/100 g d.E).

Table 2. Characterization of polyphenols in *R. tuberculata* organic extracts using cLC-DAD analysis.

N°	Compound	RT, UV (min, nm)	cLC-DAD	
			EtOAcE	AcE
			Extracted Amount (mg/100 g d.E)	
1	Gallic acid	10.37 (280)	Nd	64.02
2	Catechin	14.12 (280)	147.02	22.24
3	<i>p</i> -hydroxybenzoic acid	14.80 (254)	53.85	14.06
4	Vanillic acid	16.68 (280)	37.03	18.20
5	Caffeic acid	15.62 (325)	11.15	1.34
6	<i>p</i> -coumaric acid	17.32 (280)	28.17	35.97
7	Trans-Ferulic acid	17.94 (325)	16.08	2.9
8	Rutin	19.14 (360)	645.6	172.77
9	Myricetin	19.51 (360)	4253.86	2111.03
10	Sylimarin	19.96 (280)	345.6	149.29
11	Naringenin	18.02 (280)	29.06	9.98
12	Quercetin	20.81 (360)	12.12	13.89
13	Kaempferol	Nt	Nt	Nt
14	Hesperidin	Nt	Nt	Nt
15	Cinnamic acid	20.89 (280)	87.01	97.41
16	Trans-cinnamic acid	21.36 (280)	74.95	86.47

EtOAcE: ethyl acetate extract; AcE: acetonic extract; RT: retention time; Nt: not tested.

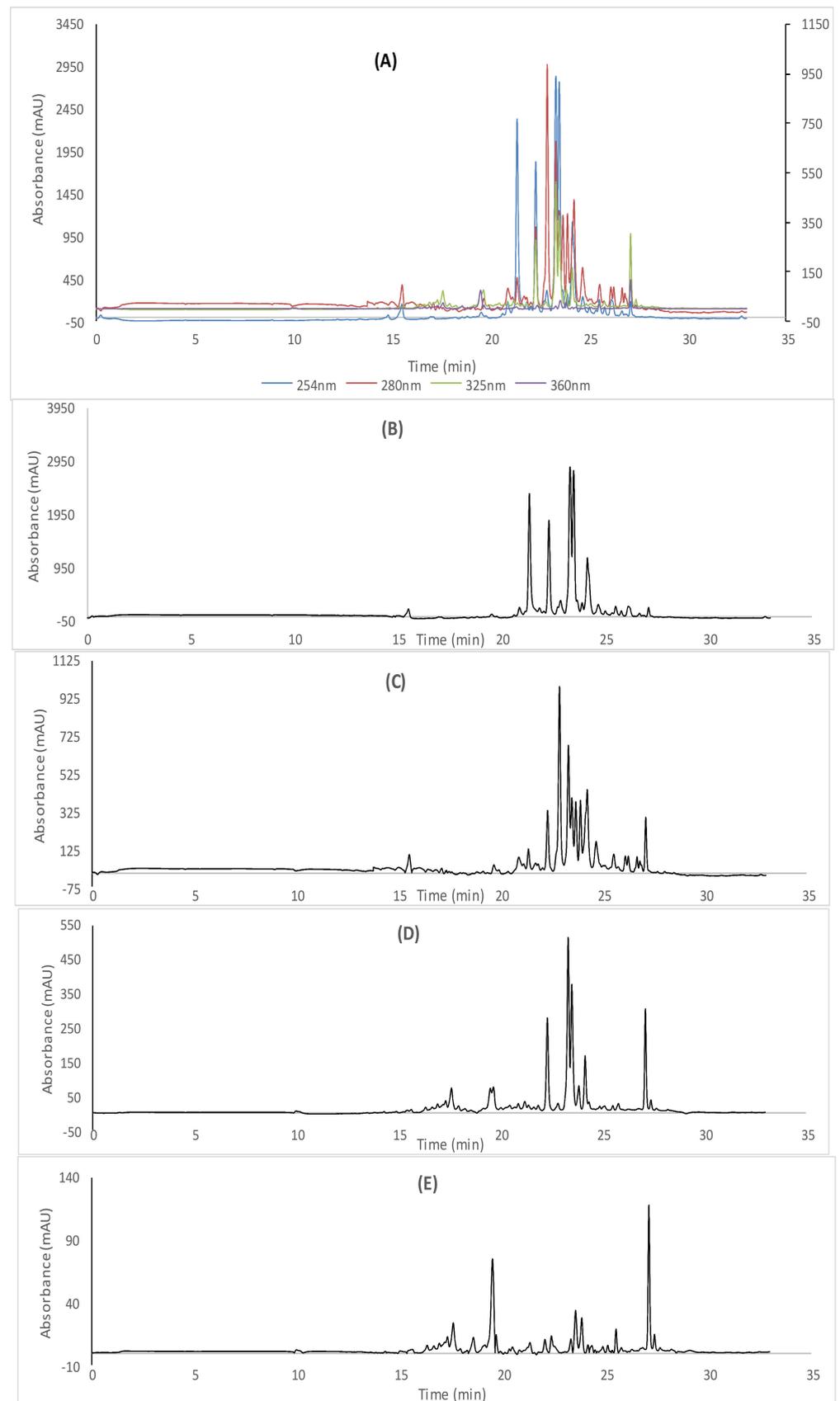


Figure 1. cLC–DAD chromatogram of ethyl acetate extracts; (A) combined chromatograms, (B) 254 nm, (C) 280 nm (D) 325 nm (E) 360 nm.

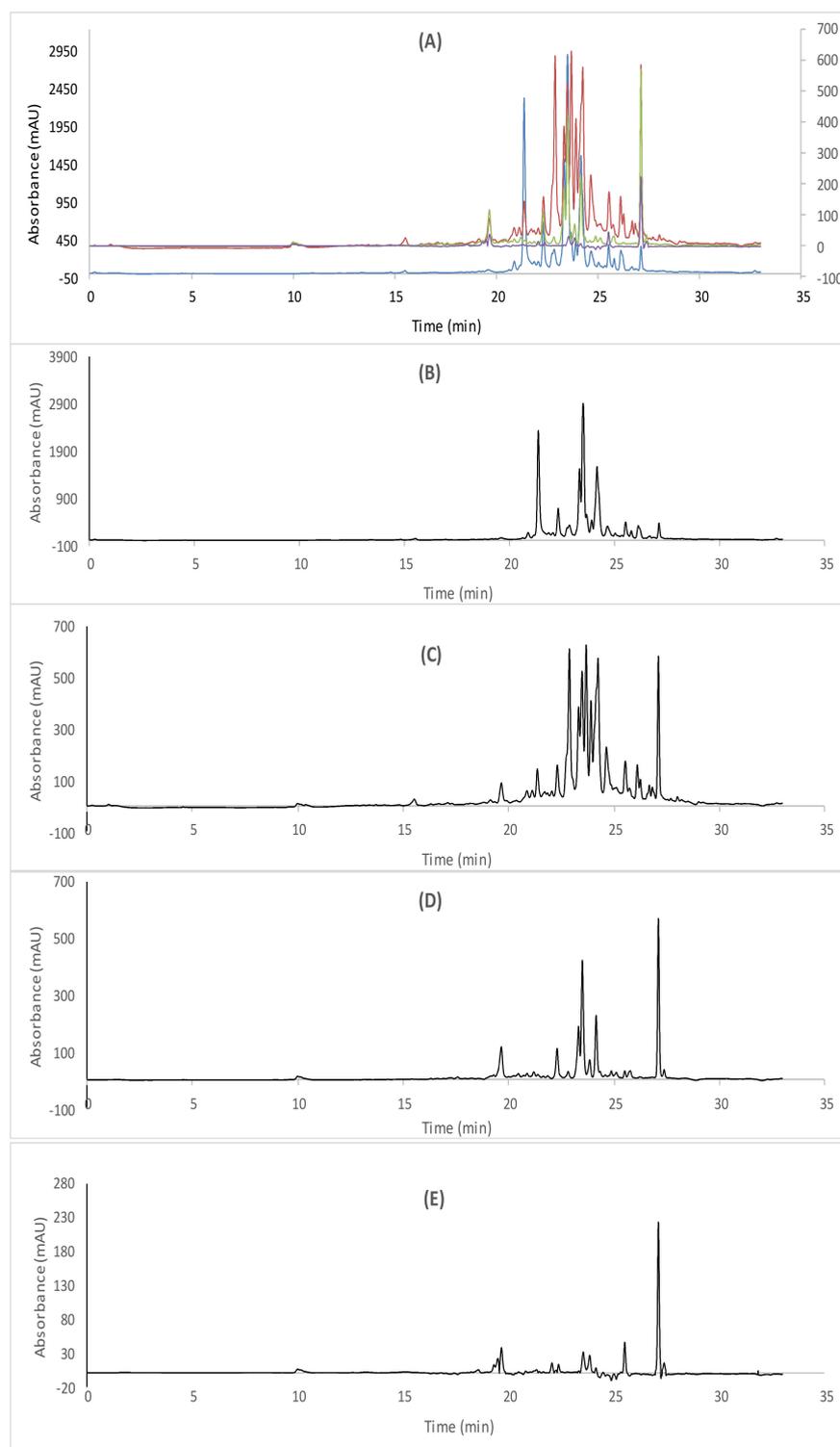


Figure 2. cLC–DAD chromatogram of acetonic extracts; (A) combined chromatograms, (B) 254 nm, (C) 280 nm (D) 325 nm (E) 360 nm.

Interestingly, both studied extracts showed to contain diverse flavonoid classes in particular EtOAc extract which presented the highest levels of myricetin and quercetin as flavonols, naringenin as flavonones, catechin as flavanol, rutin as flavone, and silymarin as flavanonol derivate. Relying on cCL-DAD analysis results (Table 2), the most abundant flavonoid detected in EtOAc extract was myricetin (4253.86 mg/100 g d.E) followed by rutin (645.59 mg/100 g d.E), silymarin (345.61 mg/100 g d.E) and catechin (147.02 mg/100 g d.E).

In the same order, myricetin was considerably the major compound in acetic acid with the highest amount of 2111.03 mg/100 g d.E, followed by rutin (172.77 mg/100 g d.E), silymarin (149.29 mg/100 g d.E), and catechin (22.24 mg/100 g d.E). However, naringenin was found in both extracts, in which the higher values were depicted in the EtOAcE extract. Otherwise, it is important to note that a high level of quercetin was observed in AcE compared to EtOAcE.

Hamdi et al. [40] identified both vanillin and caffeic acids in the EtOAc fraction of Tunisian *R. tuberculata* (or *H. tuberculatum*) aerial parts using a UPLC-ESI-MS. According to our findings, the phenolic screening data [19] using LC-MS depicted closely the same diversity of small phenolic compounds which were ferulic, caffeic, and *p*-OH cinnamic acids as cinnamic acids derivatives and vanillic acid as benzoic acids derivative.

Using different analytical techniques, several phytochemical studies demonstrated that alcoholic extracts of many *Ruta* species, such as Algerian *R. tuberculata* [9,13] and Tunisian *R. graveolens* L. [18], and *R. chalepensis* [19,41,42], showed their richness in diverse flavonoid categories that affirmed effectively our findings. Align with these previous reports, our suggestion on the wide phenolic diversity contained in *R. tuberculata* aerial parts could be certainly confirmed but supposedly with some variation in biocompound levels compared to the other *Ruta* species. Indeed, the chemical structure of phenolic compounds and their amounts in *Ruta* plants can vary under diverse conditions as documented in previous data [18,30,40]. When refereeing to literature, the present study is the first report that characterized the flavonoid compounds mainly myricetin, catechin, silymarin naringenin, and quercetin in both EtOAc and acetic extracts from the aerial parts of Algerian *R. tuberculata*.

3.3. Antioxidant Activity

Various universal bioassays have been practiced as in vitro models to assess the likely antioxidant potential of samples, but without specifying details about their mode of action. Among those methods, six approaches were carried out in the actual work to evaluate the eventual antioxidant activities of *R. tuberculata* EtOAc and acetic extracts. The obtained results were expressed as the respective values of ascorbic acid equivalents, IC₅₀ and A_{0.5} (mean ± SD) as shown in Table 3. Noting that standards values have been reported in our previous study [9].

Table 3. Antioxidant activities of AcE and EtOAcE from *R. tuberculata* aerial parts.

Extracts	Radical Scavenging		Lipid Peroxidation (β-Carotene-Linoleic Acid)	Reducing Power		Total Antioxidant Capacity (TAC)
	DPPH IC ₅₀ (μg/mL)	ABT'S IC ₅₀ (μg/mL)	IC ₅₀ (μg/mL)	FRAP A _{0.5} (μg/mL)	CUPRAC A _{0.5} (μg/mL)	AAE (mg/g E)
AcE	82.6 ± 1.1 ^{a,b,c,d}	123.58 ± 1.4 ^d	60.93 ± 0.5 ^{a,d}	120.93 ± 0.9 ^d	184.13 ± 0.7 ^d	295.62 ± 2.2 ^d
EtOAcE	186.86 ± 1.0 ^d	62.65 ± 1.5 ^d	62.19 ± 1.8 ^d	111.13 ± 0.3 ^d	213.88 ± 0.9 ^d	216.60 ± 1.0

Values are expressed as means ± S.D (*n* = 3) and are presented as ascorbic acid equivalent, IC₅₀, and/or A_{0.5} values. Mean values followed by different letters are significantly different (based on one-way ANOVA followed by Tukey's multiple comparison tests, Level of Significance: ^a *p* < 0.05, ^b *p* < 0.01, ^c *p* < 0.001, ^d *p* < 0.0001). EtOAcE: ethyl acetate extract; AcE: acetic extract.

3.3.1. Radicals Scavenging Activity

A high percentage of radical scavenging was recorded at a low concentration of the tested samples which reflects their strong antioxidant activity. For this reason, we can suggest that remarkable antiradical capacities were shown by *R. tuberculata* EtOAcE and AcE against both DPPH and ABTS stable radicals. Statistically, AcE was seen to be (*p* < 0.001) the stronger DPPH free radical scavenger agent, with an IC₅₀ value of 82.6 ± 1.06 μg/mL which is 2 times lower than that exhibited by EtOAc extract (186.86 ± 1.01 μg/mL) when its phenolic content is relatively the lowest one (Table 3). It was interested to underline

that its DPPH scavenging activity seems to be similar to that exerted by quercetin, but it remains significantly lower than that recorded by the other antioxidants drugs such as the synthetic standards (BHT, BHA) and the natural pure compounds (ascorbic acid and gallic acid). However, the anti-radical activity of EtOAcE against the DPPH radicals was statistically ($p < 0.001$) weaker than that shown by all antioxidant standards. Among these references, BHA expressed significantly ($p < 0.001$) the potent antiradical capacity with the weakest IC_{50} value ($15.74 \pm 0.5 \mu\text{g/mL}$) tracked by the ascorbic acid ($26.38 \pm 0.5 \mu\text{g/mL}$). Whereas, no statistical difference was noted between the BHT ($49.77 \pm 0.1 \mu\text{g/mL}$), gallic acid ($53.03 \pm 0.0 \mu\text{g/mL}$), and quercetin ($60.77 \pm 0.0 \mu\text{g/mL}$) which scavenged similarly the DPPH radicals. Compared to that, EtOAcE seemed to have the most significant ($p < 0.001$) quenching activity against the ABTS cation radical than that recorded by the AcE with IC_{50} values of 62.65 ± 1.5 and $123.58 \pm 1.4 \mu\text{g/mL}$, respectively. It is important to underline that BHT ($1.55 \pm 0.3 \mu\text{g/mL}$) showed to be the best antioxidant drug against ABTS cation radical with no significant difference compared to BHA ($7.54 \pm 0.7 \mu\text{g/mL}$).

Our results are in complete agreement with those reported by Hamdi et al. [40], who demonstrated that the EtOAc fraction of *H. tuberculatum*, from Egypt, has the highest scavenging capacity against both DPPH and ABTS than the other studied fractions, including chloroform and ether petroleum fractions, with IC_{50} values ranged from 20 to $29 \mu\text{g/mL}$, respectively. While these reported IC_{50} values of the same extract against both ABTS and DPPH were mostly lower than those presented by the analyzed extracts in our study. However, a similar DPPH scavenging activity was found in the literature [17] by EtOAc extract of *R. chalepensis* L., another species of *Ruta* genus [26]. Otherwise, EtOAc extract from *H. myrtifolium* seemed to be most effective against ABTS than DPPH free radical. When compared to previous report data on Algerian *R. chalepensis* [6], the EtOAc fraction from its aerial parts seems to be more effective in DPPH and ABTS reduction process than *R. tuberculata* EtOAc extract, with IC_{50} values reached to 54.98 ± 0.5 and $31.75 \pm 0.95 \mu\text{g/mL}$, respectively. Moreover, the IC_{50} values produced by the nonpolar fractions of *R. chalepensis*, in particular n-hexane and chloroform fractions showed to be closely higher than that one obtained by the acetonic extract against both DPPH and ABTS radical that makes it the most effective one against free radicals [7,9,19,41,42].

Regarding the current results, we supposed that the remarkable scavenging activity of *R. tuberculata* extracts might be due to their richness in several phenolic biocompounds, particularly in the presence of the cinnamic, trans-cinnamic, and gallic acids. This last one, which was detected only in acetonic extract, has been reported as a potent antiradical agent [43,44]. Among the diver antioxidant mechanisms, the hydrogen or hydroxyl groups contained in the phenolic structures succeed easily in the transfer of the acidic protons to the free radicals, which leads to stabilizing them effectively. Corollary, the formation of a stable phenoxyl radical resulting from the delocalization of the unpaired electron, as documented par several data [2,30,45]. Furthermore, the significant antioxidant effect of the studied extracts against the free radical may be correlated to the flavonoid compounds previously detected in them. Several investigations presumed that the structural features of flavonoids, particularly the binding of hydroxyl groups on C3 and C4 in the B ring and the double bound between C2 and C3 in the C ring, provided further evidence about their interest in antiradical ability [10,30,46,47].

3.3.2. Reducing Power

The ability of *R. tuberculata* extracts to reduce iron and copper ions was assessed using based ferric reducing antioxidant power (FRAP) and copper reducing antioxidant power (CUPRC) assays, which lead to measuring the reducers contained in samples. The data was presented in Table 3 as $A_{0.5}$ values estimated from absorbance curves at corresponding wavelengths, 700 and 450 nm, respectively as previously described in Section 2. The reduction capacity exhibited by *R. tuberculata* extracts evolved in concentration depending on both assays.

Compared to BHT ($A_{0.5} > 50 \mu\text{g/mL}$), EtOAc extract reduced effectively the iron ions with the lowest $A_{0.5}$ value of $111.13 \pm 0.3 \mu\text{g/mL}$. Thus, the acetonic extract showed to be ($p < 0.001$) the fewer one with an $A_{0.5}$ value of $120.93 \pm 0.9 \mu\text{g/mL}$. Statistically, the iron reducing capacity of *R. tuberculata* extracts remains to be weaker than that registered by the antioxidant standards, which used as a positive control. In contrast to the FRAP bioassay, the high copper reducing capacity was recorded by acetonic extract ($A_{0.5} = 184.13 \pm 0.7 \mu\text{g/mL}$) despite its few phenolic contents compared to that presented by EtOAc extract. With no statistical difference ($p > 0.05$) between them, BHA and BHT as, synthetic antioxidants, exhibited interestingly the best Cu^{+2} reduction effect ($p < 0.001$) with the weakest $A_{0.5}$ values ranging from 3.64 ± 0.2 and $9.62 \pm 0.9 \mu\text{g/mL}$, respectively, compared to the corresponding extracts.

As far as is known to the authors, no data about the possible reducing power of *R. tuberculata* organic extracts were documented which means that this study is the first investigation of their eventual reducing power using FRAP and CUPRAC approaches. Our results were appreciable but not significant to that reported data [6], where it is indicated that the EtOAc fraction of Algerian *R. chalepensis* presents interesting copper and iron reducing power capacities with weakest values of $A_{0.5}$ (40.7 ± 0.9 and $45.28 \pm 0.4 \mu\text{g/mL}$, respectively) than n-hexane and chloroform fractions, which showed the low reduction ability in both tests with $A_{0.5}$ values $> 200 \mu\text{g/mL}$. However, both studied extracts showed a potent reducing capacity with the lowest FRAP values when compared to those cultivated and spontaneous Tunisian *R. chalepensis* obtained from its flowers and leaves. Furthermore, the reducing power capacities of EtOH extracts from Tunisian *R. mentana* and *R. chalepensis* were investigated by Khadhri et al. [7], who affirmed that the highest iron reducing ability was found with *R. mentana* extracts, mainly in EtOH extract from its stem and leaves.

The electron-donating power acted by *R. tuberculata* extracts is presumably ascribed to the redox proprieties of their polyphenols content which are considered natural redactors able to donate easily the electron and hydrogen atoms. Similarly to our suggestion, it has been documented that the small phenols and their related compounds in plants are the main antioxidants that are responsible for their reducing power based on their most effective ability to deliver more free atoms [11,13,45,47]. Indeed, these natural biocompounds may act through many transfer mechanisms to scavenge free radicals, chelate metal ions, and modulate the endogenous antioxidant systems [45].

3.3.3. Total Antioxidant Activity (TAC)

The phospho-molybdenum assay is frequently performed to determine the total antioxidant activity of samples. Mo(VI) is reduced to Mo(V) by the antioxidants contained in extracts which form a stable green-colored phosphate/Mo(V), this complex has a maximal absorbance at 695 nm [33]. Results are expressed as ascorbic acid equivalent and registered in Table 3. Relevantly, EtOAc extract of *R. tuberculata* aerial parts showed to have a lower TAC value ($216.6 \pm 1.0 \text{ AAE mg/g E}$) than that established by acetonic extract ($295.62 \pm 2.2 \text{ AAE mg/g E}$), which reflect significantly ($p < 0.001$) the best TAC rate of EtOAc extract.

Our findings disagree with ones reported on *R. chalepensis* by Gali and Bedjou [6], who demonstrated that the EtOAc fraction presents the highest TAC compared to the nonpolar fractions, as well as n-hexane and chloroform fractions which gave the lower TAC values of 108.1 ± 3.8 and $49.30 \pm 1.4 \text{ AAE mg/g d.E}$, respectively. However, Zengin et al. [17] showed that the EtOAc fraction of *H. myrtifolium* had almost 1.7 times better efficiency than its MeOH extract. In addition, Kacem et al. [26] found that the TAC rates were decreased in the order of EtOH extract $>$ MeOH (50%) extract $>$ MeOH (100%) extract $>$ EtOAc extract ($92.38 \pm 4.0 \text{ AAE mg/g E}$) $>$ water extract $>$ n-hexane extract ($3.05 \pm 0.1 \text{ AAE mg/g E}$), which affirmed resolutely the positive correlation between the polyphenols amount in sample and its total antioxidant capacity. Furthermore, the variation in antioxidant capacity could be correlated with the high amount of the same small phenolic acids in EtOAc extract than in acetonic extracts, such as *p*-hydroxybenzoic, vanillic acid, caffeic acid, and

trans-ferulic acid that may act as great reducer agents. Besides, a high amount of gallic acid was found in the acetonic extract, which is an excellent antioxidant agent [48], that can elucidate often the great bioactivity of this letter. On the other hand, it is interesting to highlight the correlation between TAC rate and the richness of the studied plant extracts in flavonoids mainly rutin, myricetin, sylimarin, and naringenin, which are known as potent antioxidants compounds [24,44,48].

3.3.4. Lipid Peroxidation Inhibition Capacity

This bioassay, which is considered a hydrogen transfer-based test, is widely used to assess the ability of samples to neutralize lipid free radicals by preventing the decomposition of hydro-peroxides into free radicals and the continued hydrogen abstraction [6,47]. Indeed, β -carotene is a natural vitamin A precursor with a red-orange color. This pigment is quickly decolorized when is oxidized in presence of radicals or oxidizing ingredients. Once the reaction medium is enriched up with oxygen during the β -carotene bioassay, the linoleic acid well is autoxidized to lipid radical, which oxidases and relatively decolorizes this pigment.

The obtained results showed that both acetonic and EtOAc extracts of *R. tuberculata* exhibited similarly a remarkable inhibitory effect on β -carotene blanching phenomenon in a dose-dependent manner with maximal inhibition rate values of 95.31 ± 2.0 and $96.97 \pm 0.7\%$ for the highest tested concentration, at the end of the test. As shown in Table 3, no significant difference was statistically noted ($p > 0.05$) between the inhibition activities of these extracts on linoleic acid oxidation. Apparently, AcE seemed to be slightly more effective than EtOAc extract, nearly with equal IC_{50} values of 60.93 ± 0.5 and $62.19 \pm 1.8 \mu\text{g/mL}$, respectively. Nevertheless, their inhibitory activity remains weaker than that exerted by the synthetic antioxidants BHT and BHA which present close IC_{50} values of 1.24 ± 0.00 and $1.26 \pm 0.00 \mu\text{g/mL}$, respectively.

Our results agree with the findings data [40], which demonstrated that the best β -carotene blanching inhibitory effect was exerted by the EtOAc fraction of Egyptian *R. tuberculata* ($22 \mu\text{g/mL}$) which is similar to that elaborated by the ascorbic acid ($21 \mu\text{g/mL}$). Nevertheless, their chloroform and petroleum ether fractions showed lower inhibitory activity. According to the phyto-pharmacological report of Gali and Bedjou [6], the EtOAc fraction of *R. chalepensis* revealed an appreciable inhibition of lipid peroxidation followed closely by n-hexane fraction with the respective IC_{50} values of 38.52 ± 0.7 and $46.76 \pm 2.2 \mu\text{g/mL}$. Moreover, Kacem et al. [26] reported that the n-hexane and EtOAc extracts of *R. chalepensis* showed an interesting β -carotene blanching inhibition rate than alcoholic extracts. In contrast, Loizzo et al. [41] affirmed that MeOH extract from *R. chalepensis* leaf has a promising capacity to inhibit lipid peroxidation with an IC_{50} value of $16.9 \mu\text{g/mL}$. In addition, we suggested that the antiradical proprieties of *R. tuberculata* extracts are intensively correlated with their phenolic biocompounds may arrest the reaction chain of free radicals production and prevent lipid peroxidation. Hence, our finding agreed with previous data that found a high correlation between the antioxidant activities of plants and their phenolic contents [9,11,21,26,40]. On the other hand, many approaches were approved by the scientific community to assess the capacity of samples to inhibit free radicals involved in lipid membranes oxidation, among those methods, the ORAC assay which was performed by Eissa et al. [19], who documented that EtOH (70%) fraction of Egyptian *R. tuberculata* expressed a remarkable peroxy radical-trap activity. This important information confirms that *Ruta* species, in particular *tuberculata*, could contain hydrogen donors able to prevent lipid peroxidation such as polyphenols.

Nevertheless, no correlation is observed in our study between the inhibitory effect of the studied extracts on the β -carotene blanching phenomena and their phenolic constituents, that accord with several data. Therefore, it depends mainly on the chemical features of the solvents used to extract effectively the bio-compounds that are able to neutralize free radicals during this assay [6,26]. Indeed, the linoleic acid emulsion constitutes the β -carotene reaction medium that leads the lipophilic biomolecules, as well as

triterpenoids, to easily attack the lipid radicals [6,11]. It was important to note that the ethyl acetate can dissolve the flavonoids aglycones and diterpenes, which are previously well documented [11,44], while triterpanoids are often extracted with non-polar solvents, such as acetone. These biocompounds showed an important inhibition of lipid peroxidation that may explain the effectiveness of the studied extracts against the β -carotene blanching phenomena.

3.4. Anti-Diabetic Activity

One of the most important treatment strategies for type 2 Diabetes (T2D) is the management of postprandial hyperglycemia by delaying of the release glucose from the digestive tract through the inhibition of carbohydrates hydrolase enzymes, mainly pancreatic α -amylase (EC 3.2.1.1.) and intestinal α -glucosidase (EC 3.2.1.21.). This mechanism may retard the gastrointestinal absorption of dietary carbohydrates and decreases consequently the circulating glucose blood levels [1,17,49,50]. For these reasons, the antidiabetic capacity of our plant was investigated in the present work by assessment of their inhibitory effect on α -amylase and α -glucosidase enzymatic activity. The results are expressed as mean \pm SD and provided in Table 4.

Table 4. Anti-diabetic and anti-Alzheimer's activities of *R. tuberculata* extracts.

Extract/ Standard	α -Glucosidase		α -Amylase		Acetyl Cholinesterase	
	Max inhibition (%)	IC ₅₀ (μ g/mL)	Max Inhibition (%)	IC ₅₀ (μ g/mL)	Max Inhibition (%)	IC ₅₀ (μ g/mL)
AcE	80.2 \pm 1.1	104.5 \pm 1.8 ^d	77.9 \pm 0.7	233.04 \pm 0.9 ^d	86.7 \pm 0.5	20.5 \pm 0.2 ^d
EtOAcE	74.6 \pm 0.5	165.4 \pm 1.1 ^d	79.5 \pm 0.07	146.7 \pm 1.6 ^d	77.6 \pm 1.2	45.6 \pm 0.8 ^d
Acarbose	91.05 \pm 0.7	275.4 \pm 1.6 ^d	53.05 \pm 1.6	3650.9 \pm 1.7 ^d	Nt	Nt
Quercetin	95.98 \pm 0.6	4.3 \pm 0.2 ^d	Nt	Nt	Nt	Nt
Galanthamine	Nt	Nt	Nt	Nt	94.7 \pm 0.3	6.3 \pm 1.1 ^d

Inhibition rates and IC₅₀ values are expressed as means \pm S.D ($n = 3$). Mean values followed by different letters are significantly different (based on one-way ANOVA followed by Tukey's multiple comparison tests, Level of Significance: ^d $p < 0.0001$). Nt: not tested.

The obtained results evidenced that both studied extracts exhibited a strong inhibitory effect against both enzymes in a dose-depending manner with maximum inhibition rates $> 75.0\%$ at the high tested dose of 400 μ g/mL against α -glucosidase and 800 μ g/mL against α -amylase, respectively (Figures 3 and 4). As shown in Table 4, the inhibition effect of EtOAc extract (146.7 \pm 1.6 μ g/mL) α -amylase has been significantly ($p < 0.001$) almost 1.6 times higher than that exerted by acetonic extract (IC₅₀ value reached 233.04 \pm 0.9 μ g/mL) and even 25 times than that one of acarbose which was used as a positive control. Notably, the inhibitory activity of acarbose against α -amylase is almost 15 times weaker (3650.9 \pm 1.7 μ g/mL) than that registered by the acetonic extract. However, the natural pure quercetin inhibited effectively α -amylase and established the lowest IC₅₀ value (4.3 \pm 0.2 μ g/mL). Indeed, this activity is higher than that one recorded against the α -amylase. Likewise, EtOAc extract inhibits remarkably the α -glucosidase (165.4 \pm 1.1 μ g/mL) which exerted significantly a high inhibition effect compared to acarbose (275.4 \pm 1.6 μ g/mL). These results can reflect the potent antidiabetic effect of both *R. tuberculata* studied extracts.

Unfortunately, insufficient data about the possible inhibitory activities of *Ruta* species against α -amylase and α -glucosidase were available in the literature, while no reports on *R. tuberculata* inhibitory activities were published. Interestingly, Zengin et al. [17] indicated that EtOAc extract from *H. myrtifolium* leaves showed a higher inhibition effect against both tested enzymes followed by petroleum ether and MeOH extracts, respectively. Similarly, a previous study [41] focused on *R. chalepensis* aerial parts affirmed that its MeOH extract is able to inhibit more effectively α -amylase than α -glucosidase with respective IC₅₀ values of

69.0 and 85.5 $\mu\text{g/mL}$. Indeed, these findings data allows as to suggest that the richest extract in phenolic biocompounds, especially in flavonoids, could exhibit the potent inhibitory effect on both α -amylase and α -glucosidase enzymes, while, other bioactive compounds could be involved to enhance or increase the enzyme inhibition activity of extracts as small phenolic acids, tannins and terpenoids [17,50]. Furthermore, one of the interesting antidiabetic mechanisms of those bioactive compounds is their capacity to regulate carbohydrate digestion through the inhibition of these two enzymes, to enhance glucose uptake and its metabolism in the liver [3,49,51,52]. Moreover, previous studies in vivo reported that flavonoid compounds improve significantly the uptake of blood glucose by accelerating insulin sensitivity and GLUT-4 glucose transporters function [3,14,49,51,52].

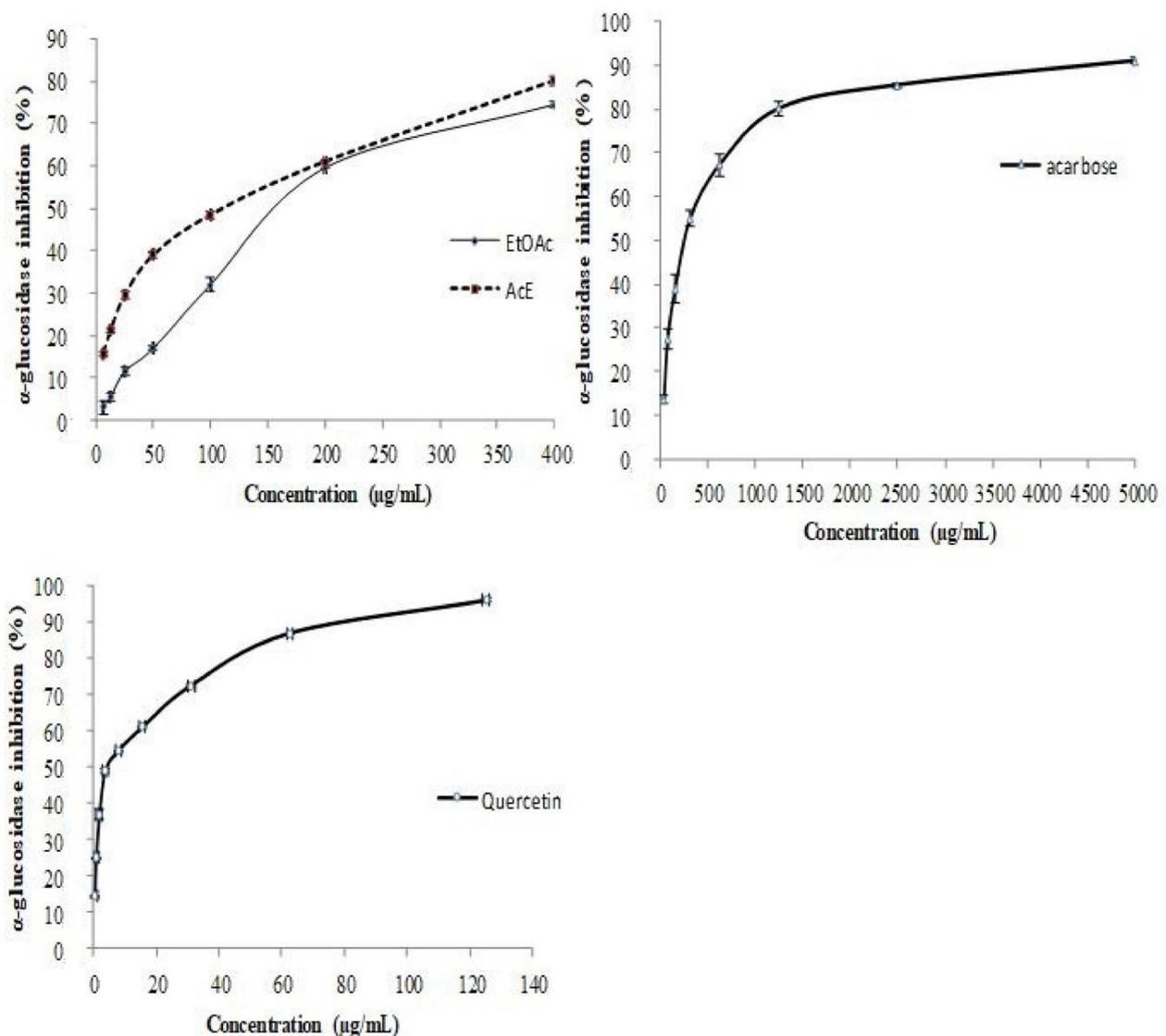


Figure 3. Inhibitory effect of *R. tuberculata* organic extracts, quercetin and acarbose against α -glucosidase. Values are expressed as mean \pm SD ($n = 3$).

Compelling evidence noticed even that naringerin, which was detected in *R. tuberculata* extracts, inhibits significantly in vitro the catalytic activity of intestinal α -glucosidase that decreases the intestinal carbohydrate absorption, and reduce therefore the postprandial blood glucose level [49]. On the other hand, the based-plant flavonoid catechin is known also as the occurring antidiabetic agent which is able to inhibit probably the carbohydrate hydrolyzing enzymes via its virtuous capacity to bind with those enzymatic proteins or with

enzyme-substrate complex [10]. Based on the report of Oboh et al. [53], rutin and quercetin, which were also found in both *R. tuberculata* extracts, have a potent hypoglycemia impact and could inhibit strongly both α -amylase and α -glucosidase. In an enzyme kinetic study performed by Limanto et al. [48], rutin showed to be an excellent competitive inhibitor of α -glucosidase than acarbose by binding to the active site of the free enzyme through non-covalent interactions, while quercetin inhibits this enzyme in a mixed type of competitive and non-competitive inhibition phenomena. Indeed, the chemical structure of flavonoids and the number of free phenolic hydroxyl in the B ring exhibited significantly a positive effect on their enzymatic inhibitory capacity. In fact, the flavonoid aglycones were reported to be the strongest inhibitor of α -glucosidase than glycosylated flavonoids. On the other hand, the large molecular size of those previously cited phytoconstituents may hinder their access to the active site of this enzyme [53–58]. This important information may explain the high efficacy of pure quercetin used as a reference to the studied extracts on α -glucosidase. Thus, *R. tuberculata* acetonetic extract, as the richest extract of quercetin, inhibited more effectively this enzyme.

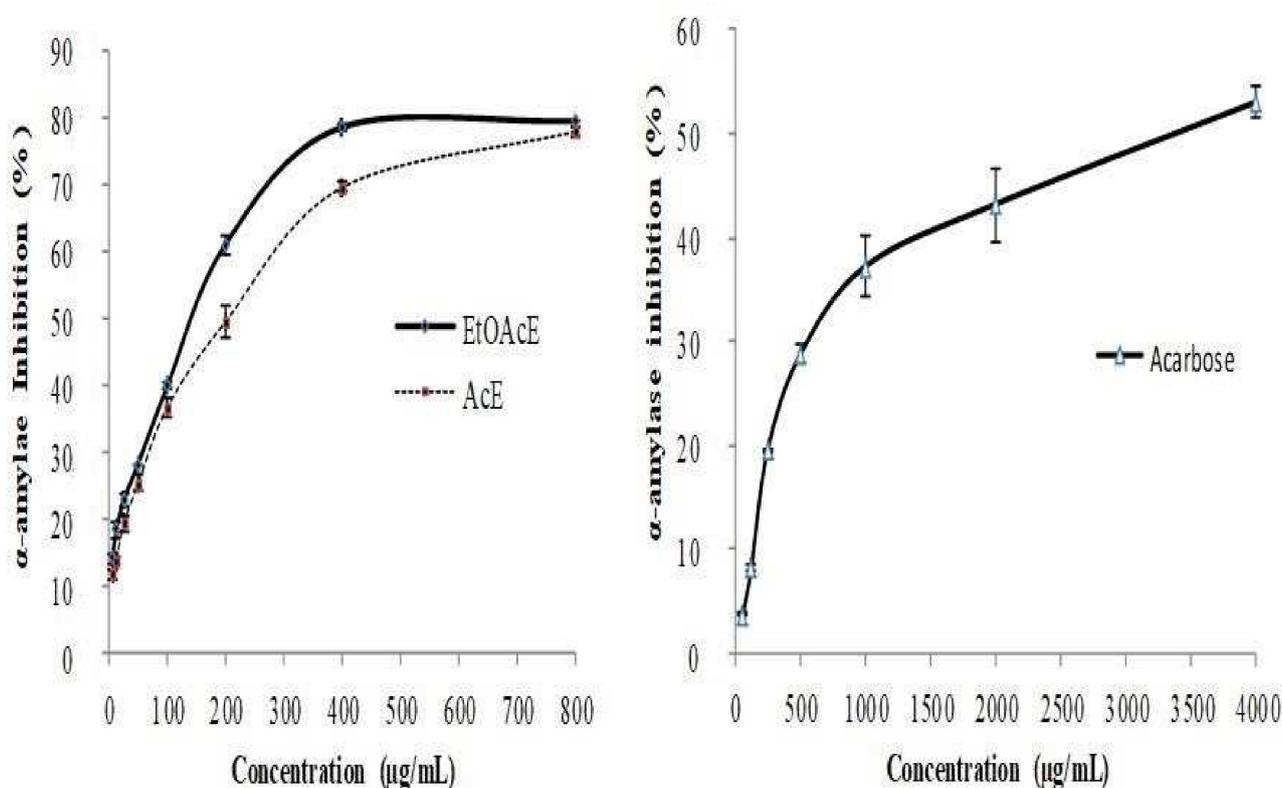


Figure 4. Inhibitory effect of *R. tuberculata* organic extracts and acarbose against α -amylase. Values are expressed as mean \pm SD ($n = 3$).

Furthermore, Limanto and its collaborators [48] suggested that both rutin and quercetin exerted their hypoglycemic effect compounds via multiple mechanisms including the improvement of insulin sensitivity, the increase of β -cell proliferation, and the stimulation of insulin release. Moreover, it was demonstrated that the free radical scavenging ability of these bioflavonoids and their lipid peroxidation inhibition power assist to alleviate streptozolocin-induced oxidative damage in diabetic rats and protecting their pancreatic β cells that improve insulin secretion and decrease lipid profile, hemoglobin A1c (HbA1c) and blood glucose levels [52,54–56]. As the same, the anti-diabetic effect of these natural flavonoids in a patient with T1DM may be mediated through the reduction of % change of fasting blood glucose level via diverse mechanisms including the decrease of the intestinal carbohydrate absorption and gluconeogenesis, prevention of the Langerhans islets cells

against degeneration, stimulation of insulin release and increase tissue glucose uptake [57]. In addition, the treatment of the diabetic rats with quercetin and hesperitin, the derive aglycone of the flavanone hesperidin, improved significantly the hepatic transaminases, α -amylase, and lipase enzymes function and modulated their serum glucose, glucagon and insulin levels [58]. Indeed, these health benefits were evidenced and expressed by their ability to increase the antioxidant defenses system and decrease lipid profile and oxidative stress indicators [54,57,58]. Furthermore, these bioflavonones could enhance GLUT-2 and 4 transporters expression and attenuate the 6-P dehydrogenase, 6-PG dehydrogenase, hexokinase, and glucokinase function in pancreatic tissue. In addition, they may down regulate the insulin receptor (IR), PI3kinase, AMPKinase and IL-1 β expression levels in diabetic rats which alters the IP/PI3K and AMPK signaling pathways and remedies insulin resistance [2,3,58].

Several pharmacological studies have postulated the biological proprieties of small phenolic acids in the preventing or treating of certain metabolic disorders linked to oxidative stress phenomena like cardiovascular disease, cancer, and diabetes. Indeed, these natural bioactive compounds act as antioxidants and could modulate effectively the oxidative stress in the body, thus may prevent often those diseases [49,50]. Moreover, Oboh et al. [59] and Abdel-Moneim et al. [60] reported that gallic and *p*-coumaic acids mostly detected in the studied acetonic extract exhibited a significant inhibitory effect on both target enzymes. This information may explain the great inhibition effect of acetonic extract against α -glucosidase. Furthermore, it was demonstrated that gallic acid is a strong competitive inhibitor of α -glucosidase when compared to acarbose, thus, this inhibitory capacity was resulting from the binding to the active site of this enzyme in free status [48].

3.5. Cholinesterase Inhibitory Activity (Anti-Alzheimer Activities)

Alzheimer's disease (AD) is one of the common neurodegenerative illnesses which affects more than 4% of people over the age of 65% and the excessive hydrolysis of acetylcholine, a key neurotransmitter in nerve terminations, may be considered the principal cause of this disease [6,7]. Actually, the pharmacological research on natural inhibitors of AChE from plants gained the attention to elaborate novel active ingredients that could enhance cognitive function and alleviate other symptoms related to this disease [9]. In this context, the AChE inhibition capacity of the studied plant was screened in vitro as previously described in Section 2, compared to Galanthamine, the natural anti-AChE drug which is notably used in therapeutics. The results are reported in Table 4.

In a dose-depending manner, the target extracts and Galantamine exhibited a significant inhibition activity ($p < 0.001$) on AChE with inhibitory rates maximally reached to 79.6 ± 1.2 , 86.7 ± 0.5 , and $94.77 \pm 0.3\%$ at the high tested dose (200 $\mu\text{g}/\text{mL}$) as depicted in Figure 5. Notably, the acetonic extract showed to have a potent anti-AChE capacity with a weaker IC_{50} value ($20.48 \pm 0.2 \mu\text{g}/\text{mL}$) when compared to EtOAc extract, which inhibited moderately the AChE with an IC_{50} value ranging from $45.62 \pm 0.8 \mu\text{g}/\text{mL}$ which is almost seven times lower than Galanthamine ($6.27 \pm 1.1 \mu\text{g}/\text{mL}$). However, the inhibitory activity of acetonic extract rests almost three times lower than that of Galanthamine. These findings allow us to presume that the organic solvents effectively extract the high level of the natural AChE inhibitor drugs. This agrees with that found by Zengin et al. [17], who indicated that petroleum ether extract of *H. myrtifulum* has the best anti-AChE activity followed by EtOAc and MeOH extracts. Moreover, other previous data [6] affirmed that organic extracts of some plants of the genus *Ruta* possess a significant capacity to inhibit AChE, for example, the chloroform fraction of Algerian *R. chalepensis* was the most active against AChE (IC_{50} value of $41.14 \pm 2.8 \mu\text{g}/\text{mL}$) despite of its low phenolic content, compared to the other analyzed fractions, in particular, EtOAc fraction which presents a higher IC_{50} value of $108 \pm 4.6 \mu\text{g}/\text{mL}$. Moreover, n-hexane extract of *R. graveolens*, as an organic extract, showed a remarkable anti-AChE activity [61,62]. However, EtOH extract of *R. chalepensis* leaves demonstrated a great inhibitory effect ($12 \pm 1.1 \mu\text{g}/\text{mL}$) than the same one of *R. montana* which present an IC_{50} value of $76 \pm 1.6 \mu\text{g}/\text{mL}$ [7].

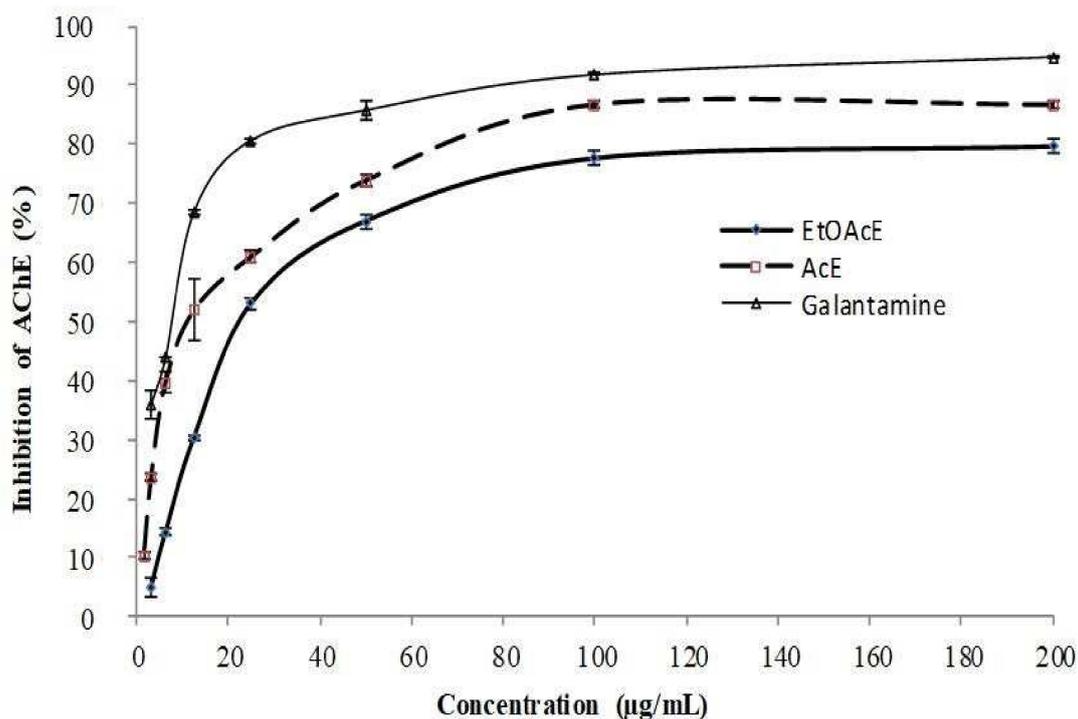


Figure 5. Inhibitory effect of *R. tuberculata* organic extracts and Galantamine against acetylcholinesterase. Values are expressed as mean \pm SD ($n = 3$).

Based on our results, we share even suggestions with Eissa et al. [19] and Saidi et al. [9], who signaled that *R. tuberculata* (or *H. tuberculatum*) may have substantial therapy features that could prevent Alzheimer's and Parkinson's diseases as neurodegenerative disorders or reduce often the behavior changes in AD sufferers via its main polyphenols content. Indeed, many reports underlined that diverse small phenolic acids which is particularly detected in acetonic extract of *R. tuberculata*, whereas, its antioxidant power was also documented [8,43]. That can elucidate the interesting Anti-AChE effect of this extract. Moreover, the significant neuroprotective effect of vanillic acid was demonstrated as evidence of LPS-induced neurotoxicity by regulating the JNK signaling pathway in the mouse brain. Therefore, this phytophenolic compound can preserve also the neuro-inflammatory process and improve the memory impairment phenomena [63]. Noting also that *p*-coumarin acid can alleviate cerebral ischemia-reperfusion injuries via its antioxidant potential that interestingly prevent hippocampal neuronal death by enhancing catalase and dismutase activity and decreasing malondialdehyde (MDA) level [64]. It was also proved that *p*-hydroxybenzoic acid inhibited potently the AChE with an IC_{50} value of 20.07 $\mu\text{g}/\text{mL}$ which can affirm their ability to maintain the cognitive function at a normal level in T2DM patients [65]

A study realized by Hernandez and his collaborators [66] showed that the bioflavonoids rutin and quercetin, which were also found in both target extracts, exhibited an interesting anti-AChE activity with the respective IC_{50} values of 86.0 and 62.0 $\mu\text{g}/\text{mL}$ that reflects their ability to prevent considerably many neurodegenerative illnesses such as Alzheimer and Parkinson diseases. However, Khandare et al. [67] and Zhao et al. [68] proved that quercetin exerted a potential neuroprotective effect on streptozolocin-induced diabetic neuropathy in rats resulting from its capacity to modulate oxidative stress indicators and reduce the release of the anti-inflammatory cytokine as well as $\text{TNF-}\alpha$ and $\text{IL-1}\beta$ via the $\text{TLR4}/\text{My88}/\text{NF-}\kappa\text{B}$ signaling pathway down-regulation. Moreover, it was proved that kaempferol exhibited a noticeable neuroprotective effect on chlorpyrifos-induced oxidative stress and memory deficit in rats which inhibited significantly the $\text{GSK3}\beta$ gene expression through the modulating of the Nrf2 signaling pathway in cerebral tissue [69]. On the other hand, the neuroprotective activity of silymarin was investigated in various models of

Alzheimer's and Parkinson's diseases where this health benefit is considerably assigned to its ability to enhance the IR1-AKT signaling pathway and to decrease effectively the oxidative stress and the inflammatory cytokines genes expression in cerebral tissue that frequently associated with the probable alteration of the cellular apoptosis machinery in there [15,44]. It was important to note that catechin revealed also a great anti-AChE and anti-BuChE activities that lead to restoring effectively the synaptic acetylcholine level, thus, it may be considered a great candidate to up-regulate the Alzheimer's disease progression [70–72].

3.6. Nutritional Importance of the Identified Compounds

Besides the pharmacological potential of *Ruta* species, these plants are also well known for their importance at the nutritional level. Indeed, it has been reported that the nutritional status of rats can considerably be improved following a daily oral administration of *R. graveolens* extracts for a period of three weeks, especially regarding total food intake and protein efficiency ratio [73]. The same research team also reported that this plant can significantly ameliorate both kidney and liver functions and maintain a coherent Albumin/globulin ratio, which could be partially linked to the non-toxic character of *Ruta* species [74].

Myricetin is an important phyto-compound, well recognized for its nutraceuticals value, and already integrated as a key ingredient for various foods and beverages. In this study, the phytochemical investigation revealed that this compound is present in both tested extracts with a considerable value, considered the best. This information is crucial since many health benefits are attributed to myricetin since this compound is effective for the treatment of obesity, diabetes, hypertension, and metabolic syndrome [75]. Indeed, an *in vivo* investigation made by Akindehin et al. [76] revealed that myricetin can reduce body weight by 11% but may also improve glucose tolerance, and fatty acid consumption and significantly decrease mitochondrial proteins acetylation in adipose tissue.

In our study, we identified a key flavonoid in a high amount called rutin, which is well known for its strong antioxidant properties but may also help the human body produce collagen and correctly use vitamin C, which suggests that this flavonoid can improve the immune system by stimulating the activity of white blood cells [77]. Noting that this bio-compound has also been identified in *R. graveolens* and revealed a non-negligible neuroprotective effect, especially to enhance spatial memory and learning performance of animals [78], but may also be involved in the regulation of energy metabolism and methane production in dairy cows [79]. *Chalepensis* leaves could also be considered healthy food ingredients since it contains a high amount of hesperidin and rutin, and their combination can generate beneficial impacts on blood vessels and thus maintain a healthy circulation [80].

Green tea, chocolate, and blueberries are gaining immense popularity among consumers in the world due to their considerable antioxidant, and relaxing effect but also to limit the risk of cardiovascular disease and stroke [81,82]. The fact that these aliments contain a high amount of catechin could explain their pharmacological aspect. Indeed, several studies reported that catechin can increase the resistance of LDL to oxidation and promotion of gut health but also inhibit the production and activation of harmful bacteria [83,84]. It is also interesting to underline [85], which may help provide energy for the contracting muscles. Noting that catechin has also been identified in our study with an important amount. These compounds may also lower blood sugar and cholesterol levels but may also improve memory, prevent neurodegenerative diseases, and decrease inflammatory pain by inhibiting neutrophil recruitment [86–88].

4. Conclusions

R. tuberculata organic extracts exhibited remarkable antioxidant activities through different antioxidant mechanisms. Moreover, they showed potent anti-diabetic and anti-AChE activities that may be due to their natural biocompounds which were identified for the first time in them, mainly naringinin, myricetin, kaempferol, catechin, sylimarin,

and hesperidin. Indeed, these pharmacological acts may justify the traditional use of this plant and its involvement in Mediterranean traditional medicine. Therefore, it may offer an exciting reserve of natural bioactive drugs which able to avoid or treat many disorders especially diabetes and Alzheimer's diseases. However, additional phytochemical studies are required to purify the biocompounds contained in this plant and to investigate their pharmacological activities using in vivo and in vitro bioassays.

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