

## Article

# Comparison of Research Data of Diesel–Biodiesel–Isopropanol and Diesel–Rapeseed Oil–Isopropanol Fuel Blends Mixed at Different Proportions on a CI Engine

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**Abstract:** Depletion in the levels of fossil fuels and increasing environmental concerns associated with the rise in consumption of conventional fuels are among the top global concerns. Finding an alternative sustainable fuel that matches the performance characteristics of diesel/petrol fuels as well as decreases the exhaust emissions has been a challenging task. After deliberate research, it is found that every alternative fuel is associated with different problems when they are used independently, thereby limiting its benefits. Scientists suggest that using different fuel blends might lead to sustainability. This article is the analysis of data obtained from the experimentation based on two different alternative fuels, Rapeseed Methyl Ester (RME)-based biodiesel and Rapeseed Oil (RO), blended with diesel (D) and Isopropanol (P) into three different proportions each. Tests were carried out in a compression ignition (CI) engine, and comparisons are based on the resulted performance and exhaust emission characteristics. The two different alternative fuels are blended into the following proportions to make six fuel mixtures, D50RME30P20, D50RME40P10, D50RME45P5, D50RO30P20, D50RO40P10 and D50RO45P5. The tests are carried out at different loads (*BMEP*) and are compared to that of pure diesel. Using the experimentation results, we also obtained the combustion characteristics of all fuel mixtures for further evaluation

**Keywords:** diesel engine; rapeseed methyl ester; biodiesel; rapeseed oil; isopropanol; performance characteristics; exhaust emission characteristics



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## 1. Introduction

The automobile world is concerned with two major problems: decreasing fossil fuel use and the rise in harmful emissions. Research result shows that automotive accounts for almost 14% of the total greenhouse gases and 25% of global carbon dioxide (CO<sub>2</sub>) emissions [1–6]. Internal combustion engines are widely used for various purposes all over the world from small-scale to large-scale applications. Statistics shows that 99.8% of the world transport is IC engine-powered, out of which 95% uses fuels such as petrol and diesel [7]. While we see the increasing demand of diesel vehicles, the amount of pollution caused is also increasing. The pollutants that are released after the combustion of fossil fuels are divided into regulated and unregulated pollutants. The limited of the amount of pollutants released that are regulated by legislation are called regulated pollutants; those for which limits are not specified are categorized into unregulated [8]. Regulated pollutants include nitrogen oxides (NO<sub>x</sub>), carbon monoxide (CO), hydrocarbons (HC) and particulate matter (PM) [9]. Due to their high-performance parameters, fossil fuels are being used in internal combustion engines [10].

Increasing demand of conventional diesel fuel in the transport sector is one of the primary reasons for the scarcity of fossil fuels [4–6]. The major reserves of fossil fuels are confined to a particular region. Countries that are low in these reserves are facing economic

and energy crisis [9]. In the wake of these problems, researchers are focusing on finding a naturally available sustainable fuel made locally from vegetable oils, biodiesel and even alcohols to balance the performance characteristics and decrease the emissions [11–13].

Biodiesels produced by the transesterification process using primary alcohols are now being greatly studied [14], investigated and implemented in the replacement of fossil fuels [15,16]. Biodiesels showed higher CO and lower brake thermal efficiency (BTE), HC and NO<sub>x</sub> emissions. Lower peak in-cylinder pressure, maximum heat release rate ( $HRR_{max}$ ) and longer combustion duration are observed in biofuels. Combustion noise and vibrations from biofuels are also found to be lower [17]. Blending biodiesel with diesel fuel lowered the emissions but increased the fuel consumption with minimum effect on efficiency [18–21]. Another research result of Agarwal [9] stated that using biodiesel in conventional diesel fueled engines reduces the emission of unburnt HC, CO and PM. However, the application of biodiesel as a replacement on a full scale has been limited due to its poor low-temperature flow property and high viscosity [22].

To balance the viscosity and solidification complications associated with biodiesel, addition of alcohols to the mixture has been found to solve the problem [23]. Blending oxygen-rich alcohols with diesel fuels in low percentages does not require any modifications in the engine fuel system [24,25]. Results of the previous studies show that, by blending diesel fuel with alcohol, emission characteristics such as smoke, carbon monoxide and nitrogen oxide are found to be improved but at the cost of decreased performance and increased fuel consumption [26–29]. Low cetane numbers and poor solubility restrict the direct use of alcohols with diesel fuel.

The benefits are limited, and different problems emerge when a single alternative fuel is used [30]. Scientists suggest that sustainability can be achieved by using fuel blends to balance the performance and emission characteristics [7,17,31–34]. Recent research reveals that using alcohols and biodiesels is one of the primary ways to decrease the emissions that disturb the environment [22,35–38]. The use of higher alcohols and biodiesels is widely recognized and researched as a replacement for conventional fuels, as they are easily produced and available [9]. Research conducted by Erdiwansyah [39] reveals that using alcohol and biodiesel as an alternative fuel decreases the emission of NO<sub>x</sub>, CO and HC.

The objective of this research is to check different engine performance and emission characteristics, followed by examining the combustion parameters of two different biofuels, Rapeseed Methyl Ester-based biodiesel and Rapeseed oil, blended with Isopropanol and diesel fuel in three different by volume proportions. The proportions at which the testing carried out were 50% diesel fuel, 30% RME and 20% Isopropanol; 50% diesel fuel, 30% RO and 20% Isopropanol; 50% diesel fuel, 40% RME and 10% Isopropanol; 50% diesel fuel, 40% RO and 10% Isopropanol; 50% diesel fuel, 45% RME and 5% Isopropanol; and 50% diesel fuel, 45% RO and 5% Isopropanol.

The selection of fuel mixture components was determined in order to evaluate the sustainability of the most commonly used biodiesel resource in Lithuania, Rapeseed Oil and its esterification product (RME) used as alternative fuels in internal combustion engines. As the use of pure RO or RME as fuel requires engine modifications [40] (such as changes in valve diameters, etc.), isopropanol alcohol was used to bring its properties closer to that of diesel fuel and to avoid additional engine modifications. This alcohol has advantages over methanol and ethanol (e.g., does not corrode the elements of the fuel supply system), so its calorific value approaches that of diesel fuel, which is directly reflected in the fuel consumption figures.

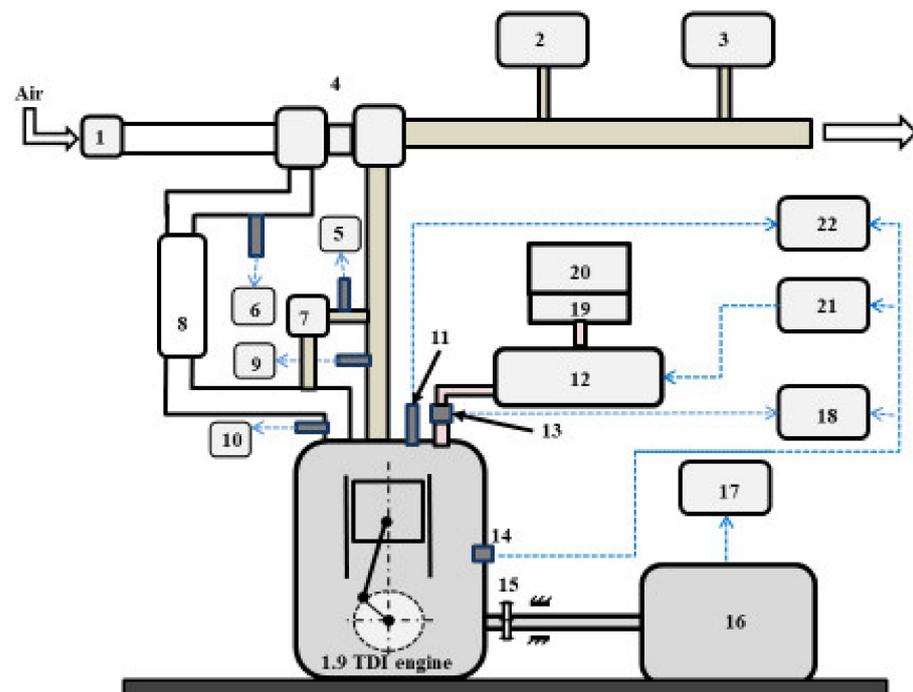
## 2. Materials and Methods

### 2.1. Engine Testing Equipment

The primary testing equipment is a turbocharged direct injection diesel engine (Table 1) with an electronically controlled distribution fuel pump. It works on a single-injection strategy, and the engine ECU controls the injection timing. The tested engine details can be understood by the schematic diagram presented in Figure 1.

**Table 1.** The main parameters of the tested diesel engine.

Parameter	Value
Displacement (cm <sup>3</sup> )	1896
Number of cylinders	4
Piston bore (mm)	79.5
Piston stroke (mm)	95.5
Compression ratio	19.5
Torque (Nm) at 2000 rpm	182
Power (kW) at 4000 rpm	66
Fuel injection	Direct (Single)
Nozzle opening pressure (bar)	190



**Figure 1.** Test bench: 1—Air Mass Flow Meter; 2—Exhaust Gas Analyzer; 3—Smoke Analyzer; 4—Turbocharger; 5—Temperature Sensor; 6—Turbocharger Pressure Meter; 7—EGR Valve; 8—Air Cooler; 9—Exhaust Gas Temperature Meter; 10—Intake Gas Temperature Meter; 11—Cylinder Pressure Sensor; 12—Injection Pump; 13—Fuel Injection Timing Sensor; 14—Crankshaft Position Sensor; 15—Connecting Shaft; 16—Engine Load Plate; 17—Engine Torque and Rotational Speed Recording Equipment; 18—Fuel Injection Moment Recording Equipment; 19—Fuel Consumption Calculation Equipment; 20—Fuel Tank; 21—Engine Electronic Control Unit (ECU); 22—Cylinder Pressure Recording Equipment.

The main parameters of the laboratory equipment used during the tests and the measurement errors are presented in Table 2. Fuel injection timing information is obtained by analyzing the fuel injection timing sensor signal. Data obtained from the piezoelectric cylinder pressure sensor (which is replaced for a glow plug) and the crankshaft position sensor (encoder) were processed using LabView Real software to obtain the in-cylinder pressure diagram. This information was used to analyze the combustion process and determine the rate of heat release, temperature rise and pressure rise in the cylinder. AVL BOOST's subprogram Burn was used for numerical analysis of the combustion process. Exhaust gas concentration (CO<sub>2</sub>, CO, HC, NO<sub>x</sub>) was determined using the AVL DiCom 4000 gas analyzer. The specific emissions (g/kWh) of different pollutants were determined by calculating the exhaust gas flow rate. Exhaust smoke was determined using an opacity measuring chamber.

**Table 2.** Uncertainties of the parameters.

Parameters	Instrument	Measurement Error
Brake torque $M_B$ (Nm)	load bench	$\pm 1.2$ Nm
Hourly fuel consumption $B_f$ (kg/h)	electronic weighing scales	0.5%
Cylinder pressure	piezoelectric sensor	$15.8 \pm 0.09$ pC/bar
Value recording for cylinder pressure	LabView Real software	interval of 0.176 CAD
Turbocharger pressure	Pressure gauge	$\pm 0.0002$ MPa
Intake air and exhaust gas temperature	K-type thermocouples	$\pm 1.5$ °C
Exhaust gas concentration	AVL DiCom 4000 gas analyzer	
CO		0.01% vol
HC		1 ppm
O <sub>2</sub>		0.01%
NO <sub>x</sub>		1 ppm
Opacity		0.1%

## 2.2. Fuels and Test Conditions

The prepared fuels for testing purposes were 100% pure diesel fuel, 100% pure Rapeseed Methyl Ester (RME), 100% pure Rapeseed Oil and 100% pure Isopropanol. Properties of all the four pure fuels presented in Table 3. As presented earlier, tests were carried out by blending Rapeseed Methyl Ester (meets the requirements of the standard EN 14,214) or Rapeseed Oil with diesel fuel (meets the requirements of the standard EN 590) and Isopropanol in three different volume proportions. First mixture consists of 50% diesel fuel, 30% Rapeseed Methyl Ester/Rapeseed Oil and 20% Isopropanol (D50RME/RO30P20), second mixture contains 50% diesel fuel, 40% Rapeseed Methyl Ester/Rapeseed Oil and 10% Isopropanol (D50RME/RO40P10) and third mixture contains 50% diesel fuel, 45% Rapeseed Methyl Ester/Rapeseed Oil and 5% Isopropanol (D50RME/RO45P5).

**Table 3.** Properties of 100% pure Diesel fuel, Rapeseed Methyl Ester, Rapeseed Oil and Isopropanol.

PROPERTIES	D100	100RME	100RO	100P
Density (kg/m <sup>3</sup> )	825	878	915	785
Viscosity (mPa·s) at 40 °C	2.352	3.575	2.745	1.138
Mass Fraction (%): Carbon	86.3	77.5	78.0	60.0
Hydrogen	13.7	12.0	10.0	13.4
Oxygen	0.0	10.5	12.0	26.6
C/H	6.30	6.46	7.80	4.48
LHV, MJ/kg	43.97	37.39	37.67	31.19
Cetane number	51.0	51.0	36.0	12.0

As the mixtures were prepared by blending the various components in the prescribed volume ratios (V/V), the density was figured by calculating the mass of each component per liter of fuel mixture and determining the mass fraction (m/m) of the individual components. The fuel blends' characteristics of density, mass fraction of carbon, hydrogen and oxygen, C/H ratio, lower heating value were calculated by estimating the mass fraction (m/m) and properties of the individual components of the mixture; the results are presented in Table 4.

The prepared fuel mixtures were tested over a narrow but frequently used operating range of this engine with a speed ( $n$ ) of 2000 rpm and brake torque  $M_B = 30; 60; 90$  Nm. They were ~20%, ~40% and ~60% from of full load engine, and brake mean effective pressure  $BMEP = 0.2; 0.4; 0.6$  MPa. The performance characteristics such as break specific fuel consumption ( $BSFC$ ), brake thermal efficiency ( $BTE$ ), exhaust gas temperature ( $T_{EX}$ ), turbocharger pressure ( $p_{TC}$ ) and exhaust emission characteristics such as carbon dioxide ( $CO_2$ ), smoke (Opacity), hydrocarbons (HC) and nitrogen oxides ( $NO_x$ ) are plotted. The accuracy at which our equipment presented the readings of carbon monoxide (CO) was too low to evaluate the difference. The tests were performed when the engine oil reached the operating temperature (~95 °C). The tests were repeated five times, and the average of the obtained results was given (at 0.2 MPa, all fuel mixtures including diesel fuel showed ~4 g/kWh; at 0.4 MPa, the value was noted to be ~1.1 g/kWh, and at 0.6 MPa it was

~0.8 g/kWh). The results were obtained by experimentally testing methods and after were analyzed by AVL BOOST program after numerical analysis of the combustion process. The heat transfer to the walls of the combustion chamber (the cylinder head, the piston, the cylinder liner) was calculated used the Woschni model published in 1978. The Woschni heat transfer model was adapted for the high-pressure cycle and the maximum error of heat transfer coefficient not bigger than 3.2% [41]. The prepared fuel mixtures were analyzed, and the results were compared with those of the diesel fuel results.

**Table 4.** Comparison of fuel properties of different fuel mixtures used.

Properties	D100	D50RME45P5	D50RO45P5	D50RME40P10	D50RO40P10	D50RME30P20	D50RO30P20
Density (kg/m <sup>3</sup> )	825.0	851.2	868.1	842.2	857.0	832.9	844.0
Mass Fraction (%): Carbon	86.30	80.96	81.13	80.18	80.35	78.56	78.71
Hydrogen	13.70	12.89	11.91	12.96	12.09	13.11	12.44
Oxygen	0.00	6.15	6.96	6.86	7.56	8.33	8.85
C/H	6.30	6.28	6.81	6.19	6.64	5.99	6.33
LHV, MJ/kg	43.97	40.29	40.37	40.03	40.11	39.48	39.54
Cetane number	51.00	49.05	42.3	47.1	41.1	43.2	34.8

### 3. Results and Discussion

In order to be able to perform a more detailed analysis of engine energy and ecological indicators, a numerical analysis of the combustion process was additionally performed, determining various combustion process indicators.

#### 3.1. Combustion Parameters

The analysis of combustion parameters was carried out at an engine load (*BMEP*) of 0.4 MPa and speed (*n*) at 2000 rpm. As in Table 5, start of injection (*SOI*), which was taken from the experiments (signal from fuel injection timing sensor), and start of combustion (*SOC*), which was taken from the combustion analyses performed in the AVL Boost (subprogram Burn), were used to calculate the ignition delay (*ID*) in CAD. From the data presented in Table 5, it is evident that, under same proportions, ignition delay was longer in mixtures with Rapeseed Oil than those with Rapeseed Methyl Ester. Isopropanol was found to prolong the ignition delay even more. The main reason for the prolongation of the *ID* (interval between *SOI* and *SOC*) is the decreasing cetane number of the fuel mixture (Table 4), and this affects the subsequent combustion phases, engine energy and ecology parameters.

**Table 5.** *SOI*, *SOC* and *ID* of all fuel mixtures.

Fuels	<i>SOI</i> , CAD BTDC	<i>SOC</i> , CAD ATDC	<i>ID</i> , CAD
D100	2.1	1	3.1
D50RME45P5	2.0	1.2	3.2
D50RO45P5	1.4	2.1	3.5
D50RME40P10	1.6	2	3.6
D50RO40P10	1.3	2.4	3.7
D50RME30P20	1.2	2.6	3.8
D50RO30P20	1.0	3	4.0

To get a better glimpse of the characteristics, the graphs of individual parameters such as rate of heat release (*ROHR*) and in-cylinder  $T_{rise}$  and  $p_{rise}$  were plotted between −10 and 25 CAD. The pressure values obtained from the experimentation were loaded and analyzed in AVL Boost software to get the combustion parameters.

From Figure 2, it is observed that maximum rate of heat release of diesel at the premixed combustion phase (33.9 J/deg) is found to be lower than all the remaining fuel mixtures. As given in Table 4, addition of alcohol reduces the cetane number, thereby

increasing the ignition delay, which causes the rise in the heat release. D50RME30P20 is found to have the maximum heat release at 48.8 J/deg, which is ~44% more when compared to diesel fuel.

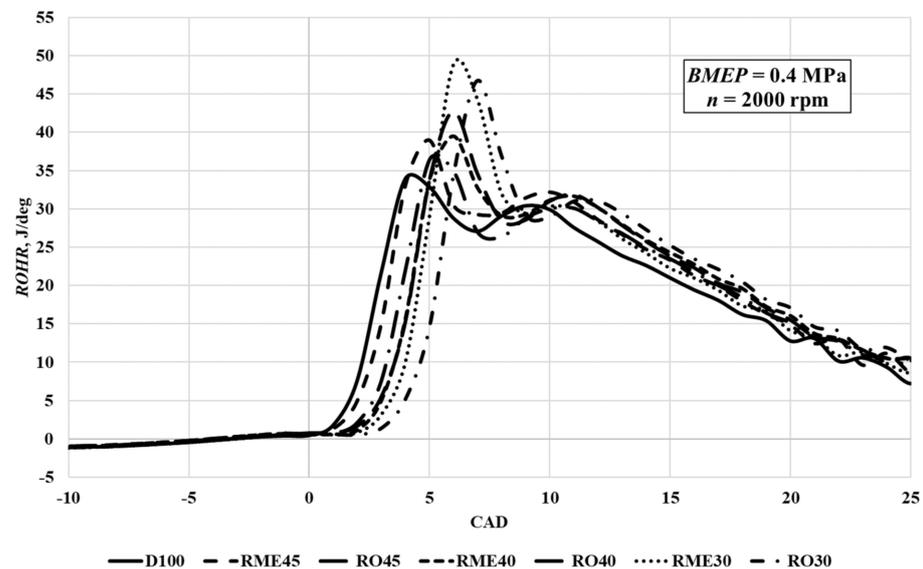


Figure 2. Comparison of ROHR of different fuel mixtures.

The  $T_{rise}$  of specific fuel mixtures was found to increase with the alcohol percentage. When the alcohol percentage in mixtures of Rapeseed Oil raised from ~5 to 10%, the peak temperature rise was found to increase by 13.9%, and with further increase to 20%, the value raised to ~11%. A similar pattern was observed in mixtures of Rapeseed Methyl Ester from Figure 3. With an increase in alcohol percentage from ~5 to 20%, the peak  $T_{rise}$  was found to increase by ~31%. Diesel fuel is found to be the lowest of all with a peak temperature rise of 60.2 K/deg. A higher  $T_{rise}$  increases the heat load of the engine and influences higher NO<sub>x</sub> emissions.

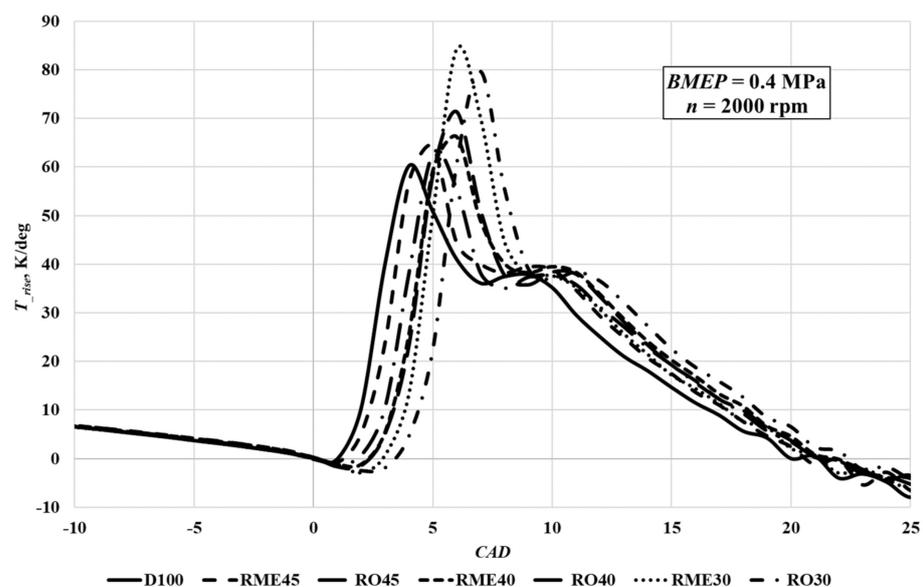
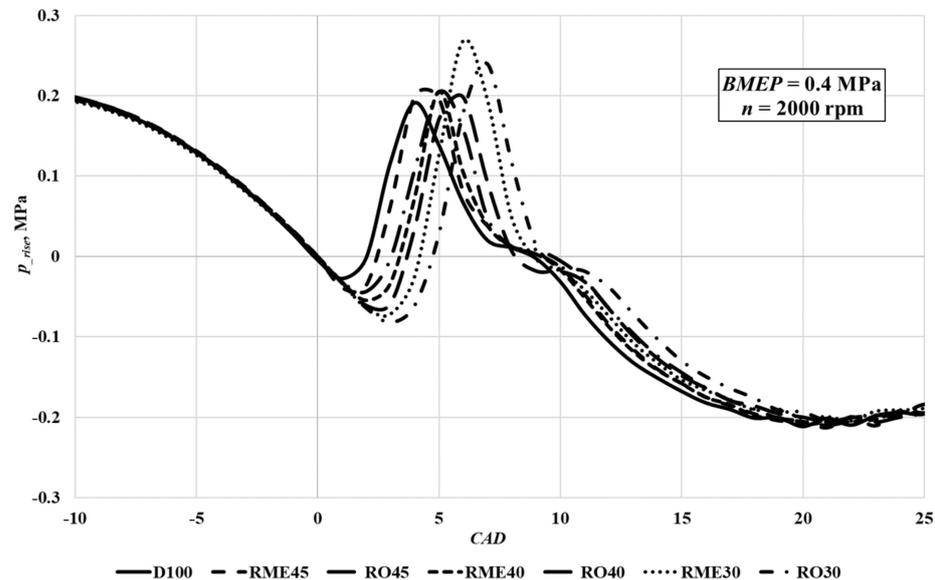


Figure 3. Comparison of  $T_{rise}$  of different fuel mixtures.

With increase in the alcohol percentage the pressure rise is found to increase for all fuel mixtures. As shown in Figure 4, maximum  $p_{rise}$  is found to follow similar pattern as

that of  $ROHR$  and  $T_{rise}$ . Diesel fuel was found to have the lowest peak of pressure rise with  $0.191 \text{ MPa/deg}$ , which is  $\sim 29\%$  less than that of D50RME30P20. A higher-pressure rise increased the mechanical loads on the crankshaft, and the engine emits more noise.



**Figure 4.** Comparison of  $p_{rise}$  of different fuel mixtures.

### 3.2. Performance Characteristics

$BSFC$  of all the fuel mixtures tended to exhibit similar patterns, as shown in Figure 5, at a specific load; with the addition of alcohol,  $BSFC$  tended to increase. Similarly, for a specific fuel mixture, the rise in the load led to the decrease in  $BSFC$ . At  $0.2 \text{ MPa}$ ,  $BSFC$  of D50RO30P20 ( $377.78 \text{ g/kWh}$ ) was found to be  $\sim 13\%$  more than that of diesel fuel. This might be because of the high oxygen content present in Rapeseed Oil and Isopropanol, as shown in Tables 3 and 4. Although the presence of oxygen in the fuel improves combustion, it reduces calorific value. The decreasing trend in the CN can be a factor for the rise in  $BSFC$ . The trend of difference in  $BSFC$  between the highest of all fuel mixtures (D50RO30P20) when compared to diesel fuel decreased with increase in load ( $\sim 13\%$  at  $0.2 \text{ MPa}$ ,  $\sim 12\%$  at  $0.4 \text{ MPa}$  and  $\sim 11\%$  at  $0.6 \text{ MPa}$ ). Diesel's fuel consumption was found to be more than the remaining fuel mixtures because of its higher calorific value (Table 4) [31].

It is observed from Figure 6 that there is an increase in  $BTE$  of all fuel mixtures with the load, because most of the indicative work is converted into energy efficient. At all loads, RME mixtures tended to show greater efficiency than RO mixtures. The percentage of difference at  $0.2 \text{ MPa}$  was found to be  $\sim 3\%$ , and the difference was found to be decreasing with the load. At  $0.2 \text{ MPa}$ ,  $BTE$  of D50RME45P5 was  $\sim 5\%$  greater than D100, followed by  $\sim 3\%$  at  $0.4 \text{ MPa}$  and  $\sim 2\%$  at  $0.6 \text{ MPa}$ .  $BTE$  was enhanced by a better combustion process due to the oxygen contained in the biodiesel.

Exhaust gas temperature ( $T_{EX}$ ), presented in Figure 7, might provide an understanding of combustion process in the engine. It plays an important role in analyzing the emission characteristics of the exhaust gas [32].  $T_{EX}$  was found to be increasing with the load as the amount of fuel burned during the cycle increased. At all loads, exhaust gas temperature of pure diesel fuel was found to be higher than the remaining fuel mixtures. When analyzed, it was found that, at  $0.2 \text{ MPa}$ , temperature at D100 was  $\sim 3\%$  more than that of average of all the remaining fuel mixtures; at  $0.4 \text{ MPa}$  and  $0.6 \text{ MPa}$ , the value was found to be  $\sim 2\%$  and  $\sim 1\%$ , respectively.  $T_{EX}$  was reduced by the use of RME or RO and the addition of Isopropanol, because the oxygen in biofuels results in faster combustion.

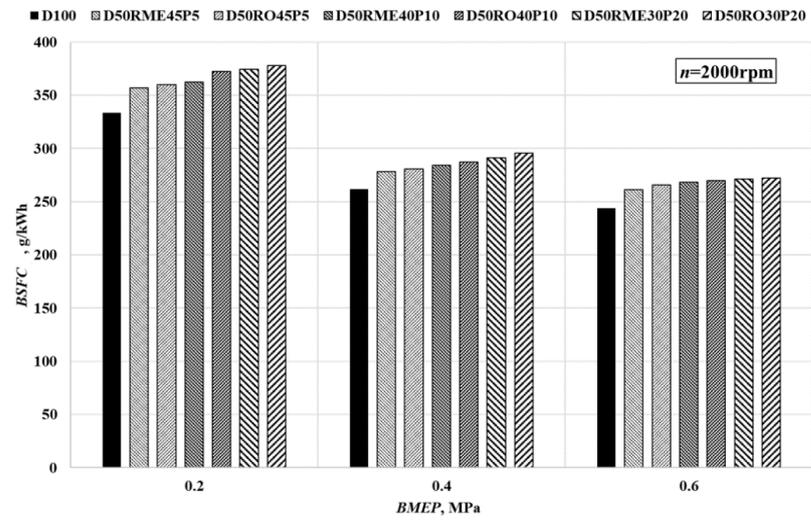


Figure 5. Comparison of BSFC of different fuel mixtures.

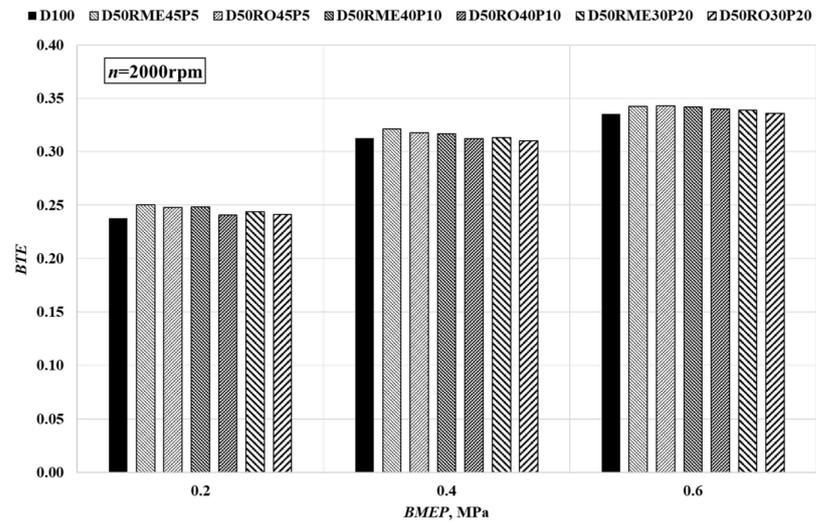


Figure 6. Comparison of BTE of different fuel mixtures.

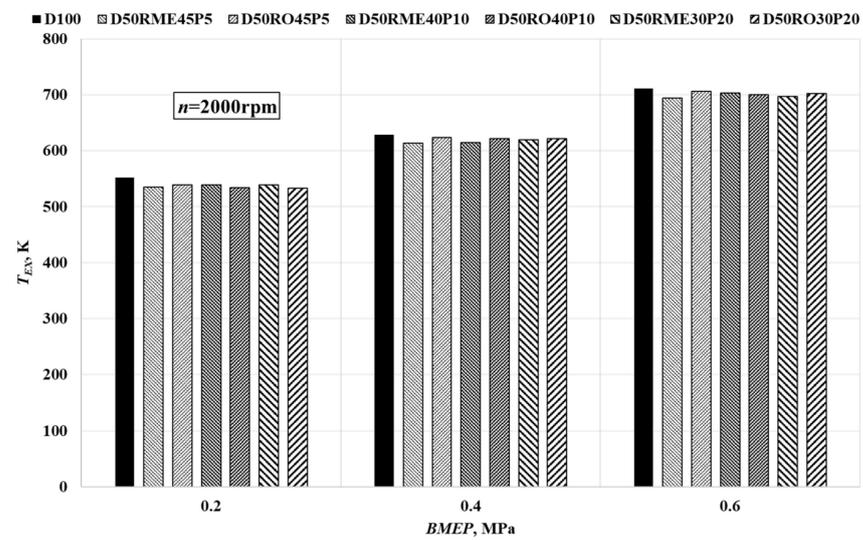


Figure 7. Comparison of T\_EX of different fuel mixtures.

As shown in Figure 8,  $p_{TC}$  is found increase with the load as the exhaust gas flow increases. At all the given loads, diesel fuel was found to have the highest pressure, because, when diesel fuel and biofuel blends are used, the oxygen in the fuel accelerates combustion and reduces exhaust energy. At 0.2 MPa, the  $p_{TC}$  was found to be 1.24 bar, which is ~4% higher than the average of remaining fuel mixtures.

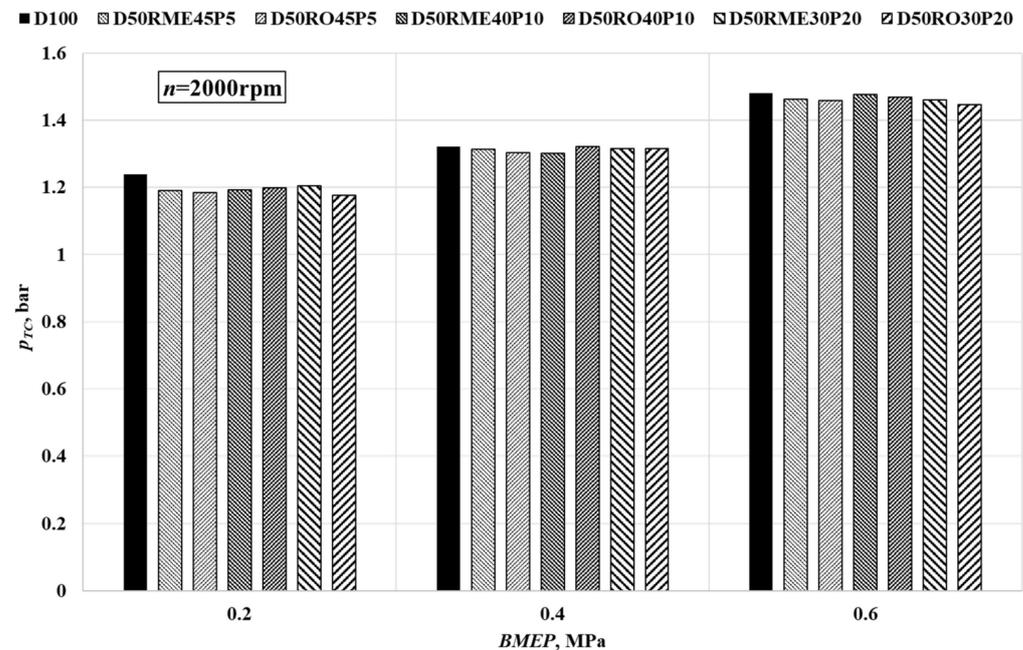


Figure 8. Comparison of  $p_{TC}$  of different fuel mixtures.

### 3.3. Exhaust Emission Characteristics

The tendency of  $\text{CO}_2$ -specific emissions is presented in Figure 9. The carbon dioxide emissions were found to decrease with increase in the load as the percentage of energy efficiency increased. Addition of alcohol to the blends led to the reduction of the C/H ratio in the fuel blends, which, in turn, reduced the emissions [17]. At relatively higher tested loads, 0.6 MPa, all the fuels tended to have nearly similar amounts of emissions. At 0.2 MPa, diesel fuel tended to have the highest amounts of emission at 1047.5 g/kWh, which is ~4% higher than that of D50RO30P20. At 0.4 MPa and 0.6 MPa, the difference was found to be ~2% and ~1.3%, respectively. The difference between these mixtures decreased with increase in load.

As shown in the Figure 10, smoke (opacity) levels were found to increase with load as the air–fuel ratio decreases. At all loads, pure diesel fuel was found to have the highest levels. This might be because of the lack of oxygen content in diesel fuel. It was observed that, at any given load, with the addition of alcohol, the smoke decreased. For fuels with the same alcohol percentage, mixtures with RME tended to be lower than RO because of the low C/H ratio, as given in Table 2. At all loads, mixtures with RME were found to be lower than RO at an average of ~4–5%.

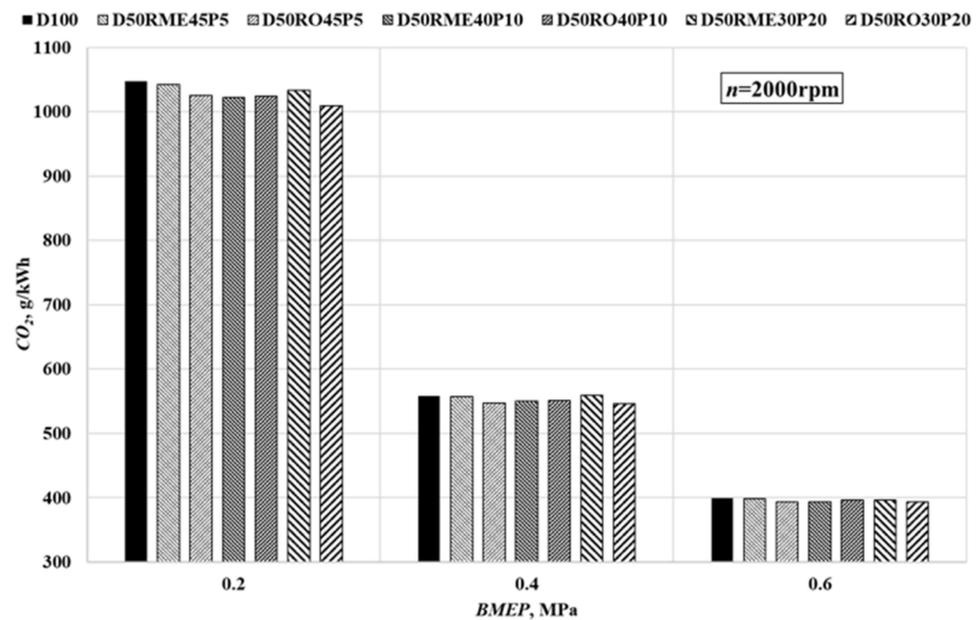


Figure 9. Comparison of CO<sub>2</sub> of different fuel mixtures.

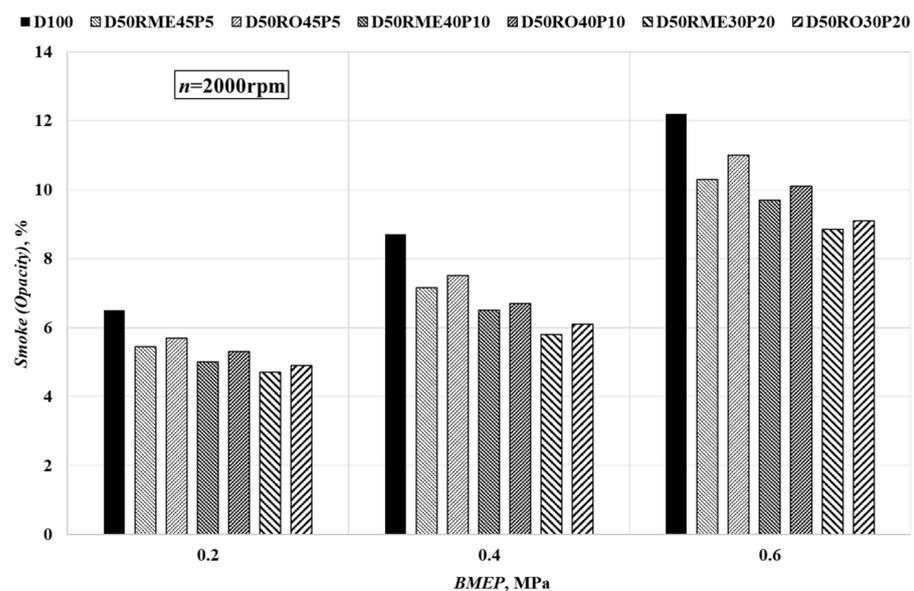


Figure 10. Comparison of Smoke (opacity) of different fuel mixtures.

Hydrocarbon emissions of all fuel mixtures including diesel fuel were found to be decreasing with increase in the load, as shown in Figure 11. This might be because of the rise in combustion temperature. At a specific load, diesel was found to be lowest, followed by RME and then RO. This is because of diesel's high cetane number and lower ignition delay [24]. With the increase in alcohol percentage, mixtures tended to show an increase in HC emissions. At 0.2 MPa, the average of emissions of mixtures of RO were found to be ~44% higher than that of RME. Similarly, the value was found to be ~43% and ~30% at 0.4 MPa and 0.6 MPa, respectively.

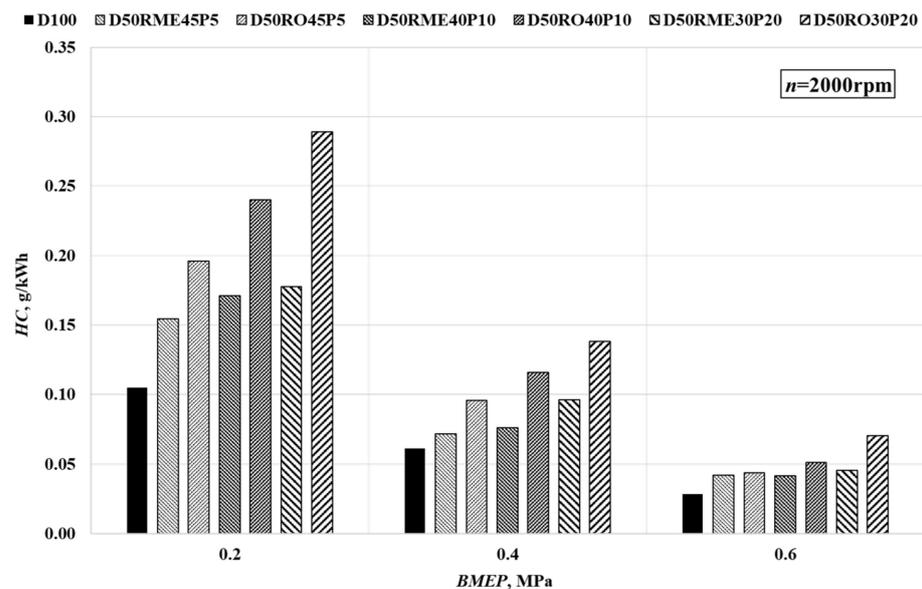


Figure 11. Comparison of HC of different fuel mixtures.

As shown in Figure 12, nitrogen oxides emissions were found to be lower in diesel fuel at all loads. At any specific load, the emissions were found to be gradually increasing with addition of alcohol percentage. Addition of alcohol implies that the oxygen concentration has been increasing, justifying the rise in  $\text{NO}_x$  emissions.  $\text{NO}_x$  emissions were also increased by prolonged ignition delay due to decreasing cetane number. This resulted in an increase in  $\text{ROHR}$  and temperature rise at the premixed combustion phase, which had a direct effect on  $\text{NO}_x$  formation. At 0.2 MPa, the fuel mixture D50RO30P20 was found to be ~16% more than that of diesel fuel. At 0.4 MPa, the value was found to be more than halved at ~8%. At maximum load, the difference between fuel mixture with highest emissions and diesel fuel was found to be ~4%.

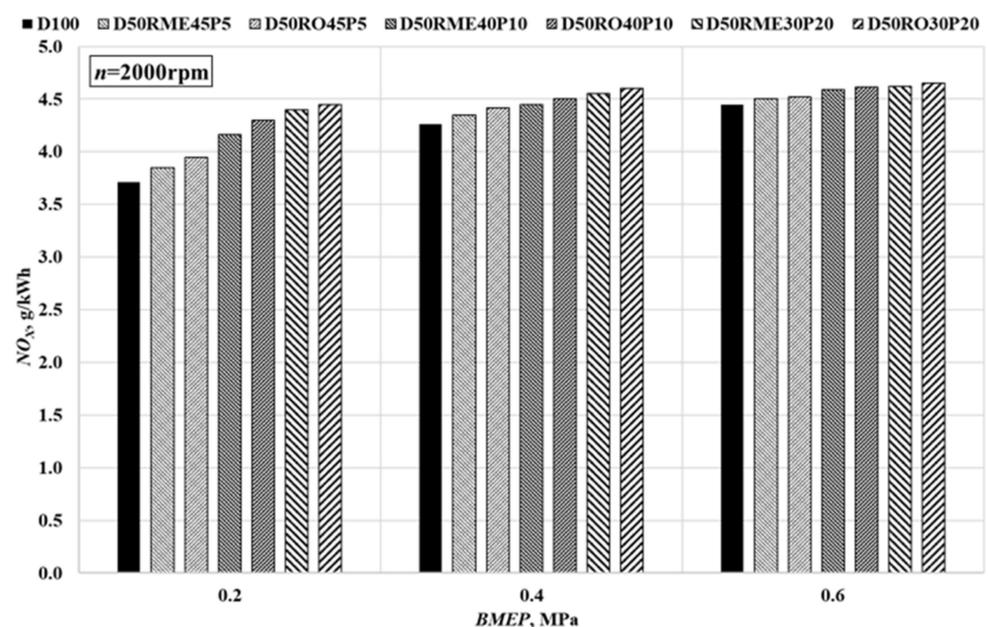


Figure 12. Comparison of  $\text{NO}_x$  of different fuel mixtures.

#### 4. Conclusions

After experimentally finding the values of performance and exhaust emission characteristics of the prepared fuel mixtures (D50RME30P20, D50RME40P10, D50RME45P5, D50RO30P20, D50RO40P10 and D50RO45P5) over a narrow but frequently used operating range, we tested a CI engine (engine speed  $n = 2000$  rpm; three loads—20%, 40% and 60%). Our results were compared with those of pure diesel fuel. The pressure values in cylinder and consumptions of fuel and air thus obtained were used to obtain the theoretical combustion parameters. After carefully analyzing the obtained data, the following conclusions are made:

1. Adding alcohol such as Isopropanol and biofuels such as Rapeseed Methyl Ester or Rapeseed Oil led to the increase of oxygen concentration and decrease of  $LHV$  and cetane number of fuel blends. The C/H ratio of blends with RME additive was lower compared to diesel fuel, but the RO additive C/H ratio increased.
2. Maximum rate of heat release of diesel fuel was lower than all the remaining fuel mixtures, because the addition of alcohol reduced the cetane number, thereby increasing the ignition delay, which caused the heat release to rise at the premixed phase of combustion. Increased ROHR with increasing isopropanol concentration resulted in higher  $p_{rise}$  and  $T_{rise}$ , and this increased the mechanical and thermal load on the engine parts.
3. At 0.2 MPa,  $BSFC$  of D50RO30P20 was found to be ~13.2% more than that of diesel fuel. With increase in oxygen content of the fuel mixtures,  $BSFC$  increased. The trend of difference in  $BSFC$  between the highest of all fuel mixtures (D50RO30P20) when compared to diesel fuel decreased with increase in load (difference when calculated in percentage was found to be ~13.2% at 0.2 MPa, ~12.87% at 0.4 MPa and ~11.6% at 0.6 MPa).
4. At all loads, RME mixtures tend to show greater efficiency than RO mixtures. The percentage of difference at 0.2 MPa was found to be ~1.2–2%, and the difference was found to be decreasing with the load. At 0.2 MPa, BTE of D50RME45P5 was ~5.9% greater than that of D100, followed by ~3.2% at 0.4 MPa and ~2.3% at 0.6 MPa. Both  $T_{EG}$  and  $p_{TC}$  were found to decrease when using diesel fuel mixtures with RME or RO and Isopropanol. This indicates that these mixtures burn faster due to the increased oxygen concentration.
5. At relatively high tested loads, 0.6 MPa, all the fuels tended to have nearly similar amounts of emissions. At 0.2 MPa, diesel fuel tended to have the highest amounts of  $CO_2$  emission, that is, ~3.6% higher than that of D50RO30P20. At 0.4 MPa and 0.6 MPa, the difference was found to be ~2.2% and ~1.3%. The difference between these mixtures decreased with increase in load.
6. Among all the tested loads, diesel fuel was found to record the highest level of smoke opacity due to the lack of oxygen content. With the addition of alcohol, the smoke decreased gradually. At all loads, mixtures with RME were found to be lower than RO at an average difference of ~4–5% because of its low C/H ratio.
7. HC emissions of all fuel mixtures including diesel fuel were found to be decreasing with increase in the load because of the rise in combustion temperature. With the increase in alcohol percentage, mixtures tended to show an increase in HC emissions.
8.  $NO_x$  emissions were increasing with alcohol. The difference in the increase of emissions in mixtures was found to decrease with increase in load. At 0.2 MPa, the fuel mixture D50RO30P20 was found to be ~16.09% more than that of diesel fuel. At 0.4 MPa, the value was ~7.9%, and at maximum load, ~4.3%.

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