

Article

Transformation of Corn Stalk Residue to Humus-Like Substances during Solid-State Fermentation

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Abstract: Lignocellulase production from straw fermentation has been widely investigated but the research has neglected to quantify fermentation-derived residue transformation to the humus-like substance (HULIS). To investigate the conversion efficacy of corn stalk residue to HULIS, the amount of HULIS associated with chemical composition and structural changes of humic acid-like substances (HAL) was investigated in a 30 L solid-state fermentation tank during a short period of eight days. The results show that the highest decomposition rate of corn stalk and the highest activity of cellulase, xylanase, and β -glucosidase appeared at the fourth day. At the end of fermenting process, the amount of humic acid-like substances (HAL) and the percentage of HAL in humus acid (PQ value) increased 17.5% and 8.9%, respectively, indicating *Trichoderma reesei* facilitates the transformation of corn stalk residue to HAL. Fatty acids decreased while aromatic carbon and carboxyl content significantly increased during the ongoing fermentation, which had a positive impact on the HAL thermal stability. The FTIR spectral and thermal analysis revealed an improvement in HAL degrees of condensation, oxidation, and aromatization. The present study suggests that the residue of corn stalks fermented with *T. reesei* might be a good fertilizer to improve soil characteristics.

Keywords: corn stalk residue; humic-like substance (HULIS); solid-state fermentation; *Trichoderma reesei* (*T. reesei*); humification

1. Introduction

Proper management of crop residues is becoming more critical due to its high correlation with increased crop production. In some developing countries, crop residues are usually burned in the open field, resulting in emissions of particles and hazardous gases inducing serious air pollution [1]. For preventing air pollution caused by the direct combustion of crop residues, many local governments and farmers are seeking alternative treatments. To solve the straw surplus problem, crop residues have been treated for the production of bioenergy and environmentally friendly materials [2,3]. However, due to the high reluctant content of cellulose (25%–35%), hemicellulose (20%–40%), and lignin (10%–25%) in crop residues, an effective pretreatment method is usually required in the conversion process. Monomicrobial fermentation or physicochemical microbial methods have been widely used to pretreat crop residues due to their low cost and high environmental benefits [4–6]. A number of studies attempted to use filamentous fungi (*Aspergillus*, *Penicillium*, or *Trichoderma*) to produce

high concentrations of lignocellulases from crop residues [7–10]. Cellulase is a multicomponent enzyme system that plays a synergistic role in degrading cellulose and contains endocellulase, exoglucanase, and β -glucosidase. Endocellulase cleaves internal glycosidic bonds; exoglucanase cuts the cellulose chain from either the reducing or nonreducing end; and β -glucosidase completes the final step of hydrolyzing cellulose to produce glucose [11,12]. *T. reesei* is a mesophilic, filamentous fungus, which is a major industrial source of cellulase due to its easier culture and excellent capacity for degrading various substrates into lignocellulase and cellulase, such as horticultural waste [13], stalks [14], and apple pomace [15]. The optimal fermenting conditions of *T. reesei* in the biodegradation of various substrates could be different, for example, a pH of 5.0–6.6 and water content of 70%–94% for wheat bran [16] and 4% sodium hydroxide pretreatment for wheat stalks [17]. The highest cellulase activity was observed in 25–30 °C and a water content of 55%–70% when fermented rice bran with *T. reesei* [18].

As stated above, many studies have focused on cellulase or lignocellulase production rates from various substrates, functional mechanism of enzymes, and the optimal fermenting conditions. However, the incorporation of fermentation-derived residues in the humus-like substance (HULIS) is not completely understood. HULIS is mainly composed of humic acid-like substances (HAL), fulvic acid-like substances (FAL), and humin-like substances (HML), with similar physicochemical properties to natural soil humus, which are well known for their contribution to soil fertility [19,20]. The amount, the chemical composition, and the structural features of fermentation-derived HULIS are still ambiguous, despite its potential in facilitating the agricultural ecological cycle.

This study aimed to investigate conversion efficacy of fermentation residues to HULIS, considering the impacts of *T. reesei* on corn stalk degradation. The cellulose and lignin content were compared along with the cellulase, xylanase, and β -glucosidase activity. Additionally, the dynamic changes of HULIS in terms of elemental composition and structural characteristics were determined by elemental analysis, thermogravimetric analysis, and infrared spectroscopy.

2. Materials and Methods

2.1. Strains and Medium

T. reesei (MCG77), strain purchased from American Type Culture Collection (ATCC), were inoculated on a medium containing 30 mL potato dextrose agar (PDA) and placed in an incubator at 28 °C for 72 hours to obtain mature mycelia; then, 2 mL of sterile water was added and oscillated at 130 rpm for 2 min. The solution was removed, placed in a sterile test tube, and diluted immediately after the oscillation. It was counted afterward by a hemocytometer plate to make a spore suspension of 10^7 unit per mL. A total of 20 mL \times 3 of the diluted solution was transferred to a liquid medium, considering a 1:10 ratio, and then cultured for 6 days at 30 °C at 100 rpm to obtain the stock (total 600 mL) containing benton mycelium.

2.2. Organic Materials

The corn stalks were collected at Jilin Agricultural University experimental farm, naturally air-dried, then crushed to 0.5 cm in length, and stored in a sealed container. Basic properties of the corn stalk were 419.34 g kg⁻¹ total organic carbon (TOC), 6.91 g kg⁻¹ total nitrogen, 7.70 g kg⁻¹ total phosphorus, 4.50 g kg⁻¹ total potassium, and a C/N ratio of 61.

2.3. Experimental Design

The experiment was performed in a solid-state fermentation tank (Baoxing BIOTECH-30SS, Shanghai) of 30 L, able to automatically stir materials and control humidity (60%) and temperature (30 °C). Fermentation speed was set at 6.0 rpm. Inorganic mineral salt was added for *T. reesei*'s nutrition and was made up by urea 4.2 g L⁻¹, ammonium sulfate 19.6 g L⁻¹, calcium chloride 0.028 g L⁻¹, potassium

dihydrogen phosphate 28 g L⁻¹, magnesium sulfate 4.2 g L⁻¹, ferrous sulfate 0.07 g L⁻¹, manganese sulfate 0.021 g L⁻¹, zinc sulfate 0.019 g L⁻¹, and yeast extract 7 g L⁻¹, with an adjusted pH of 5.

In total, 1.5 kg of crushed corn stalks were added to the fermentation tank and sterilized at 121 °C for 1 hour. After the temperature dropping to the set temperature (30 °C), 600 mL *T. reesei* stock and 3.75 L sterile inorganic mineral salt were added. A round of fermentation was 8 days and 3 rounds were repeated. A treatment of corn stalk without adding *T. reesei* was set as control, denoted as T0. Three phases of fermented materials with *T. reesei* were sampled for 2, 4, and 8 days, denoted as T1, T2, and T3, respectively. Samples were collected after every 24 h period. All samples were stored at −20 °C.

2.4. Determination of Cellulose, Lignin, and Enzyme Activity

The cellulose and lignin contents were analyzed and calculated based on previous reported methods [21].

A 6 cm Whatman filter paper (width 1 cm) was folded and put into a sterile test tube. One milliliter citrate buffer (pH 4.8) was added in the tube with 0.5 mL diluted spore suspension in five series: 0.04, 0.05, 0.067, 0.1, and 0.2 g mL⁻¹. After reacting at 50 °C for 60 min, 3 mL 3,5-dinitrosalicylic acid (DNS) was added, the mixture was heated in a boiling water bath for 5 min, then cooled immediately. Subsequently, 200 µL of above solutions were diluted with 2.5 mL deionized (DI) water. Absorbance values were measured on visible spectrophotometer (SP-722E) with optical density at 540 nm (OD 540). The ratio of 0.37, corresponding to the enzyme concentration that released 2 mg glucose, was defined as the cellulose activity [22].

One gram of xylan was dissolved into 80 mL 50 mM citrate buffer, stabilized in a 60 °C water bath for 1 hour, and then transferred to another water bath at 100 °C until boiling. Absorbance was measured on OD 540. The amount of enzyme to produce 1 µmol xylose per minute was defined as one unit of xylanase activity [23].

A total of 200 µL fermented sample solution was diluted 1:200 with citrate buffer (pH 4.8). Then, 1 mL of the above solution and 1 mL 15 mmol/L cellobiose were pipetted into a sterile centrifuge tube, placed in a 50 °C water bath for 30 min, and then transferred to 100 °C water bath for 5 minutes. The tube was removed and cooled. The mixed solution was used to measure glucose content with a glucose kit (Huili Biotech, China). Another 1 mL of diluted sample solution and 1 mL of citrate buffer were added into a sterile centrifuge tube as an enzyme blank. The ratio of 0.0926, corresponding to the enzyme concentration in 1 mg/ml of the glucose content in the sample, was defined as the activity of the glycosidase [24]. The ratio of 0.0926, corresponding to the enzyme concentration in 1 mg/ml of glucose content in the sample, was defined as the activity of the glycosidase [24].

2.5. Extraction and Determination of Humus

We weighed 0.25 g of dried (55 °C for 6 h) and fine ground corn stalk (passed through a sieve with 0.25 mm in diameter). Extraction protocol was based on the modified method of soil humus composition [25]. HULIS was extracted from corn stalk and mixed 0.1 mol·L⁻¹ sodium hydroxide and 0.1 mol·L⁻¹ sodium pyrophosphate. HAL and FAL were separated with 0.5 mol·L⁻¹ sulfuric acid. Residues left were considered as HML. Carbon content of each component was determined according to the potassium dichromate volumetric method. The percentage of HAL in humus acid (PQ value) was calculated to evaluate the dynamic changes between FAL and HAL and measure the corn stalk humification degree [26], which is displayed as:

$$PQ = \frac{HAL (g \cdot kg^{-1})}{HAL (g \cdot kg^{-1}) + FAL (g \cdot kg^{-1})} \times 100\% \quad (1)$$

where HAL and FAL denote humic acid-like and fulvic acid-like substances, respectively, extracted from fermented corn stalk residue.

2.6. Purification and Determination of HAL

Corn stalk samples were dissolved in $0.1 \text{ mol} \cdot \text{L}^{-1}$ hydrochloric acid with a ratio (solid to liquid) 1:10 (V:V). Then, $0.1 \text{ mol} \cdot \text{L}^{-1}$ sodium hydroxide was added and the pH of the mixed solution was acidified by $6 \text{ mol} \cdot \text{L}^{-1}$ hydrochloric acid to 1.5 to get rough HAL. The purified HAL samples were obtained by high-speed centrifugation, electrodialysis, rotary evaporation, and lyophilization, as recommended by the International Humic Substances Society (IHSS) [27]. HAL elemental composition was measured by the Vario EL III elemental analyzer (Elementar, German) in C/H/N mode. Freeze-dried HAL was analyzed for functionalization on an AVATAR 360 Fourier transform infrared spectrometer (FTIR, Nicolet, America); and the differential thermal analysis (DTA) and thermogravimetric analysis (TG) were performed by STA 2500 thermogravimetric analyzer (Netzsch, German).

2.7. Data Processing

Data were collected and analyzed by Graphpad Prism 6.0. One-way analysis of variance (ANOVA) and Pearson correlation with two tailed analysis were used to determine the difference and correlation of corn stalk quantitative and structural change after fermentation. All statistical tests were evaluated at the 95% confidence level.

3. Results

3.1. Quantitative Change of Corn Stalk

The quantitative change of corn stalk is ascribed to the change of material weight; TOC; cumulative degradation rate of lignin and cellulose; and enzyme activities of cellulase, xylanase, and β -glucosidase. There was approximately 9.83% stalk loss during the first three days, and about 25.09% loss in the following five days. TOC decreased 39.66% during the fermentation and the degradation rate reached the highest at the fourth day (Figure 1a). The cumulative degradation rate of lignin and cellulose revealed a sharpened increase tendency in the last four days (Figure 1b).

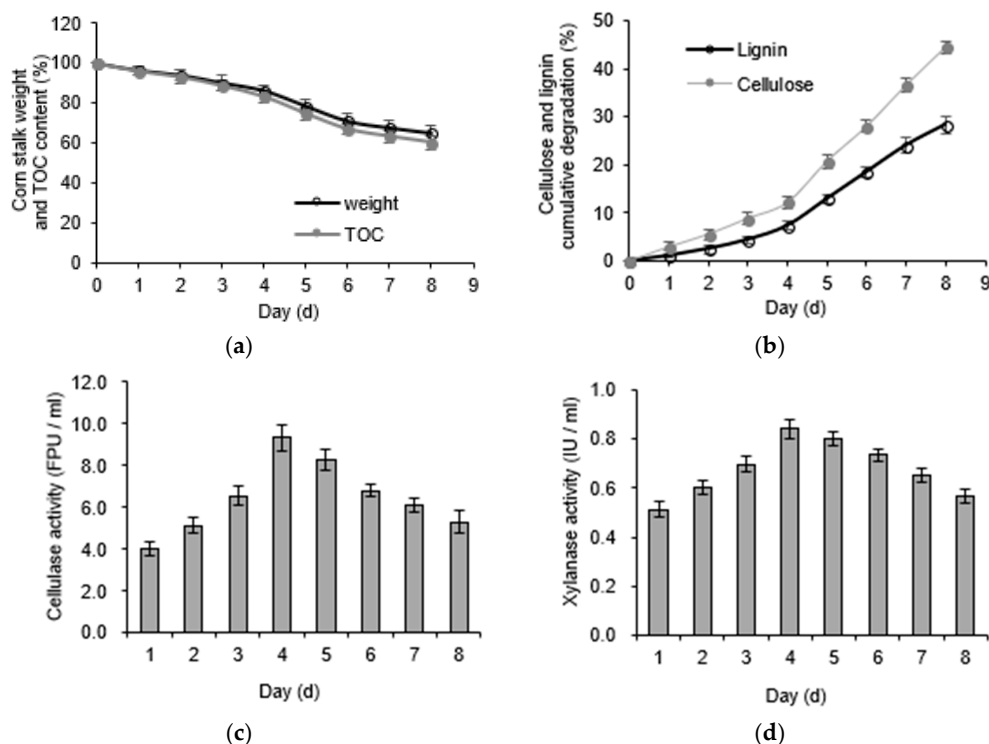


Figure 1. Cont.

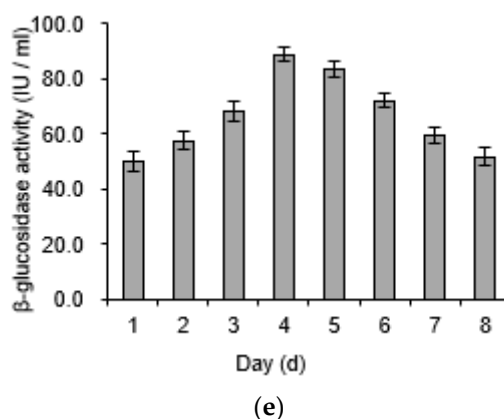


Figure 1. Quantitative change of corn stalk and enzyme activities at different fermentation times. (a) weight and total organic carbon (TOC); (b) cellulose and lignin; (c) cellulase; (d) xylanase; (e) β -glucosidase. Data represent the mean \pm SEM of triplicate.

After eight days fermentation, the cumulative degradation rate of cellulose and lignin arrived at 43.84% and 30.85%, respectively. β -glucosidase showed a higher enzyme activity of 88.91 IU/m on the fourth day (Figure 1e), while xylanase had the lowest enzyme activity with 0.84 IU/mL, as shown in Figure 1d. A high level of β -glucosidase was considered as a strong inhibitor of endocellulase and exoglucanase to regulate the conversion of cellulose to glucose [12].

3.2. HULIS Composition and Transformation

Compared to T0, the carbon content of HAL with the *T. reesei* treatment increased by 17.5% at the end of the fermenting process (Table 1), the C content of FAL and HML decreased by 5.54% and 7.38%, and the PQ value increased by 8.9%, suggesting that either the formation rate of HAL was greater than FAL, or FAL degradation rate was higher than HAL. Thus, the addition of *T. reesei* to the fermentation process enhanced the degree of humification of corn stalk.

Table 1. Effects of *T. reesei* on composition and percentage of humic acid-like substances (HAL) in humus acid (PQ) value of corn stalk humus.

Treatment	Day (d)	HAL ² (g·kg ⁻¹)	FAL ² (g·kg ⁻¹)	HML ² (g·kg ⁻¹)	PQ ³ (%)
T0 ¹	0	50.63 \pm 0.44 d	36.08 \pm 0.07 a	269.25 \pm 0.10 a	58.39 \pm 0.25 d
T1 ¹	1–2	55.70 \pm 0.07 c	35.98 \pm 0.18 a	268.88 \pm 0.15 b	60.75 \pm 0.11 bc
T2 ¹	4–5	57.10 \pm 0.48 b	35.16 \pm 0.08 b	256.72 \pm 0.15 c	61.89 \pm 0.18 ab
T3 ¹	7–8	59.47 \pm 0.13 a	34.09 \pm 0.07 c	249.39 \pm 0.28 d	63.56 \pm 0.10 a

Note: One-way analysis of variance (ANOVA) was used to determine the difference in the corn stalk composition. Multiple comparisons between every two treatments were performed using the least significant difference (LSD). Different lower-case letters in the same column mean significant difference at $p < 0.05$ level; ¹ T0 was a control treatment of corn stalk without added *T. reesei*; T1, T2, and T3 were three phases of fermented materials with *T. reesei* sampled in 2, 4, and 8 days, respectively; ² HAL, FAL, and HML represent three substances that, like humic acid, fulvic acid, and humin, derived from fermented straw residue, respectively; ³ PQ = [HAL/(HAL + FAL)] \times 100%.

3.3. Elemental Composition of HAL

The elemental composition of HAL is used to evaluate the structural change of HAL molecules and provide information of humus-like formation. The molecular structure of HAL is characterized by the molecular ratio of (O + S)/C and H/C. A higher ratio of (O + S)/C indicates more oxygen-containing groups in HAL; conversely a lower H/C ratio suggests a higher degree of aromatization and condensation [28].

The elemental composition of HAL is shown in Table 2. Compared to T0, the ratio of HAL (O + S)/C increased by 32.71%. H/C ratio was lower than T0, decreasing by 5.53% by the end of fermentation.

Table 2. Effect of *T. reesei* on the elemental composition of HAL in corn stalk.

Treatment	C (g kg ⁻¹)	N (g kg ⁻¹)	H (g kg ⁻¹)	O + S (g kg ⁻¹)	C/N	(O + S)/C	H/C
T0 ¹	387.4	16.82	68.35	527.4	26.87	1.021	2.117
T1 ¹	376.2	16.42	64.72	542.7	26.73	1.082	2.064
T2 ¹	355.4	15.77	60.74	568.1	26.29	1.199	2.059
T3 ¹	331.2	15.03	55.37	598.4	25.71	1.355	2.006

Note: One-way analysis of variance (ANOVA) was used to determine the difference in the corn stalk composition. Multiple comparisons between every two treatments were performed using the least significant difference (LSD). Different lower-case letters in the same column mean significant difference at $p < 0.05$ level; ¹ T0 was a control treatment of corn stalk without adding *T. reesei*; T1, T2, and T3 were three phases of fermented materials with *T. reesei* sampled in 2, 4, and 8 days, respectively.

3.4. FTIR Spectra

Through semi-quantitative analysis, it is known that the relative intensity of HAL at the peaks of 2920 cm⁻¹ and 2850 cm⁻¹ decreased, which indicate a great alleviation of vibration of polymethylene carbon and terminal carbon in aliphatic hydrocarbon; however, the relative intensities at the peaks of 1620 cm⁻¹ and 1720 cm⁻¹ decreased first and then raised up with time, indicating that fatty acids decreased while aromatic carbon and carboxyl content increased during the fermenting process. Several organic functional groups, such as -COOH in 1230 cm⁻¹ and Si-O in 1034 cm⁻¹, were also found, but overlapped for impurities (Table 3 and Figure 2).

Table 3. HAL relative intensities of the main absorption peaks in FTIR spectrum.

Treatment	Relative Intensity (%)				Ratio	
	2920 cm ⁻¹	2850 cm ⁻¹	1720 cm ⁻¹	1620 cm ⁻¹	2920/1720	2920/1620
T0 ¹	22.24 a	10.42 a	14.35 a	21.04 a	1.550 a	1.057 a
T1 ¹	22.09 ab	9.630 b	13.98 b	12.62 b	1.580 a	1.750 b
T2 ¹	21.58 b	7.210 c	16.91 c	15.62 c	1.276 b	1.382 c
T3 ¹	13.73 c	7.120 c	18.80 d	22.96 d	0.730 c	0.598 d

Note: One-way analysis of variance (ANOVA) was used to determine HAL differences in the relative intensities of the main absorption peaks of the FTIR spectrum. Multiple comparisons between every two treatments were performed using the least significant difference (LSD). Different lower-case letters in the same column mean significant difference at $p < 0.05$ level; ¹ T0 was a control treatment of corn stalk without adding *T. reesei*; T1, T2, and T3 were three phases of fermented materials with *T. reesei* sampled in 2, 4, and 8 days, respectively.

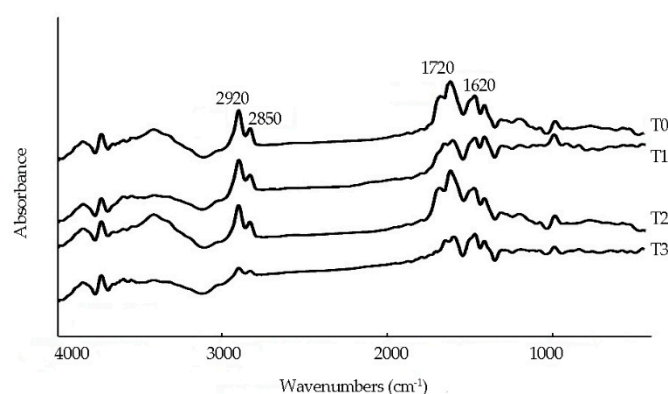


Figure 2. FTIR spectra of HAL in corn stalk. HAL represents substances that, like humic acid, are derived from fermented straw residue.

3.5. HAL Thermal Stability

To better understand HAL structural changes of corn stalk fermentation, HAL thermal stability and the corresponding mass change were accessed by DTA and TA. The thermal decomposition of samples

was exothermic and divided into medium temperature phase (270–362 °C) and high temperature phase (474–509 °C) (Figure 3). Aliphatic pyrolysis occurred during the medium temperature phase; while in the high temperature phase, carboxyl and aromatic compounds were degraded under thermal action. There are two similar exothermic peaks between T0 and T1. The exothermic peak in T2 appeared at 270 °C, which probably accounted for exothermic decarboxylation in the HAL peripheral functional group [29]. Further, *T. reesei* was prone to be active in this phase, suggesting that HAL composition was getting complex in the middle phase of degradation. The higher ratio of the heat released or the weight lost in the high temperature phase compared to that in the medium temperature phase means higher HAL thermal stability. In this experiment, both of the ratios were higher than that in T0 and increased with time, which suggested that the fermentation could increase the proportions of carboxyl and aromatic compounds and enhance the HAL thermal stability (Table 4).

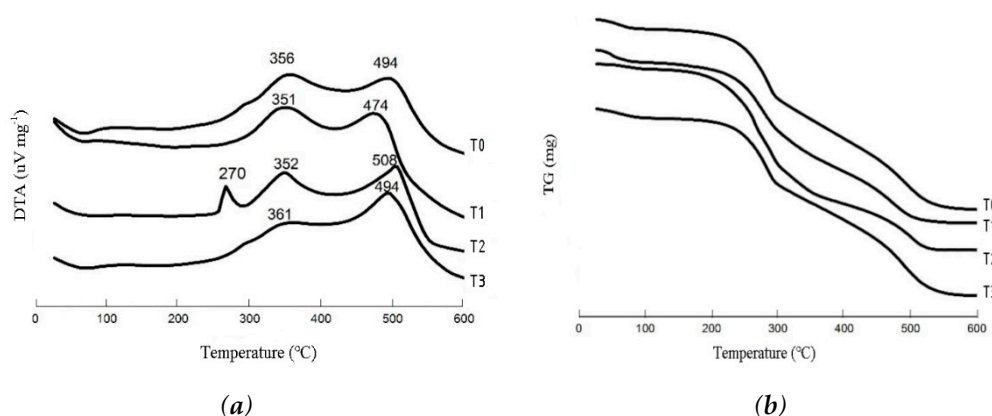


Figure 3. Thermal stability and change of corresponding mass of HAL. (a) Differential thermal analysis (DTA); (b) thermogravimetric analysis (TG). HAL represents substances that, like humic acid, were derived from fermented corn residue.

Table 4. Effects of *T. reesei* on exothermic heat and mass loss of HAL.

Treat-ment	Exothermic Heat (kJ g ⁻¹)		Exothermic Heat Ratio of Moderate to High Temperature	Mass Loss (mg g ⁻¹)		Mass Loss Ratio of Moderate to High Temperature
	Moderate Temperature	High Temperature		Moderate Temperature	High Temperature	
T0 ¹	4.167	3.356	0.805	572.7	297.3	0.519
T1 ¹	3.747	3.790	1.011	469.1	213.3	0.455
T2 ¹	3.120	4.069	1.304	394.3	233.3	0.592
T3 ¹	2.603	5.200	1.998	306.4	397.7	0.848

¹ T0 was a control treatment of corn stalk without adding *T. reesei*; T1, T2, and T3 were three phases of fermented materials with *T. reesei* sampled in 2, 4, and 8 days, respectively.

4. Discussion

Weight and TOC content of corn stalk decreased during the eight days of fermentation. Both the cumulative degradation rates of lignin and cellulose increased more intensely in the last four days. Further, Pearson correlation analysis implied that the weight loss of corn stalk had a significantly positive correlation with TOC ($p < 0.01$), but TOC had a highly negative correlation with the degradation rate of lignin and cellulose ($p < 0.01$). Therefore, along with the weight loss of corn stalk, the easily degradable carbon was firstly exhausted, and then the hardly degradable material, including lignin and cellulose, was significantly decomposed. It was reported that cellulase activity reached the highest value in 77 hours, when the bagasse was degraded by *T. reesei* in solid-state fermentation [30]. Lee reported that cellulase activity reached the highest value at day five, when bagasse and palm meal were mixed in the ratio of 1:1 [31]. The present study suggests that the first several days are important for the enzyme activities of cellulase, xylanase, and β -glucosidase. The three enzymatic activities increased in the first four days, then decreased during the next four days. One possible reason might

be *T. reesei* adapting to a new environment slowly in the early phase and alleviating the nutrients' limitations afterwards.

During the fermentation of corn stalk, *T. reesei* mainly produced: (i) cellulase, which cleaved glycosidic bonds and degraded cellulose to cellobiose; (ii) xylanase, which is a critical hydrolytic enzyme to hydrolyze xylan into oligosaccharides; and (iii) a small amount of xylose and arabinose [11]. At the final step of cellulose hydrolysis, β -glucosidase converted cellobiose to free glucose molecules. Thus, β -glucosidase showed comparatively higher enzyme activity (88.91 IU/mL) than the others on the fourth day, while xylanase had the lowest enzyme activity with 0.84 IU/mL, as shown in Figure 1d,e.

According to Waxman's theory [32], lignin is the basic composition of humus. Lignin and its degradation products, such as phenolic compounds and aliphatic compounds, form humus-like materials. They are the important precursor materials of humus, and can further form humus through complex reaction mechanism. HULIS is mainly composed of HAL, FAL, and a comparatively large portion of HML, quantified by the content of organic carbon. The results present that the carbon content of FAL and HML decreased. However, both the carbon content of HAL and PQ values increased, indicating the fermentation process promoted the transformation of corn stalk residue to HAL. HAL concentration after eight days of corn stalk fermentation was higher than that of chicken manure compost ($\sim 52.0 \text{ g} \cdot \text{kg}^{-1}$) [33], indicating that the addition of *T. reesei* to the fermentation process enhanced the degree of humification of corn stalk and the corn stalk fermentation product could be a good fertilizer. Part of the undecomposed corn stalks were transferred to HULIS under the action of microorganisms, in which part of the FAL and HML were either degraded or converted to HAL (Table 1, Table 3, and Figure 2). In general, H/C ratio of HAL corn stalk was close to 2.0, making it higher than that of most soil, which is close to 1.0–1.2 [28]. Thus, it may indicate that HAL of corn stalk is not a similar substance with the precursor of humic acid. The fermentation process speeds the oxidation and aromatization of HAL, leading to a more complex molecular structure.

Compared to T0, aliphatic degree was greatly reduced in corn stalk under the fermentation of *T. reesei*, while the oxidation degree, condensation degree, and thermal stability significantly increased. The latter fact suggests that the structure of HAL tended to be complex with lignin decomposition. Yu et al. demonstrated that soil HA was prone to formation by converting reductive functional groups to oxidative functional groups during the process of straw decay [34]. Wang et al. indicated that HA was mainly derived from lignin in the early phase of formation. The final formation of HA needed a long period of soil humification [35]. Meanwhile, in a relatively short time, HAL has begun to form with a low aliphatic, high aromatic degree, and more complex macromolecular compounds through condensates and structural recombination processes.

5. Conclusions

The transformation efficacy from corn stalk to HULIS was studied based on the transform amount of HULIS associated with HAL changes in chemical composition and structure under the action of *T. reesei*. HAL and PQ value increased 17.5% and 8.9%, whereas substances of FAL and HML decreased 5.54% and 7.38%, respectively, indicating that *T. reesei* promoted the transformation of corn stalk residues to HULIS during a short period of eight days, and part of the FAL and HML were degraded or converted to HAL. The highest decomposition rate of corn stalk and the highest activity of cellulase, xylanase, and β -glucosidase appeared at the fourth day. Additionally, the HAL degrees of condensation, oxidation, and aromatization, as well as HAL thermal stability, were improved, which indicated that *T. reesei* produced cellulase and promoted the transformation of corn stalk residues into stable HAL. Therefore, the corn stalk fermentation product could be a good fertilizer to improve soil characteristics, and the results also provide an available use of *T. reesei* in agricultural soil amelioration.

Author Contributions: Y.Y. set up and performed the experiments, and collected and analyzed data; L.W. contributed to method supervision and writing the manuscript preparation; K.S., Y.Z., and L.L. contributed to the experiment set up and by commenting on and editing the paper; X.S., X.L., and X.R., contributed to evaluating the

data and editing the paper; S.D. contributed to designing the experiments and evaluating the data and editing the paper.

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Conflicts of Interest: The authors declare no conflict of interest.

Abbreviations used in this paper:

CSRE	corn stalk residue extract
DI	deionised
DNS	3,5-dinitrosalicylic acid
DTA	differential thermal analysis
FAL	fulvic acid-like substance
FTIR	Fourier transform infrared spectrometer
HAL	humic acid-like substance
HML	humic-like substance
HULIS	humus-like substances
OD 540	optical density at 540 nm
PQ	the percentage of HAL/[HAL + FAL]
TOC	total organic carbon
TG	thermogravimetric analysis

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