



# Project Report On the Sustainability and Progress of Energy Neutral Mineral Processing

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**Abstract:** A number of primary ores such as phosphate rock, gold-, copper- and rare earth ores contain considerable amounts of accompanying uranium and other critical materials. Energy neutral mineral processing is the extraction of unconventional uranium during primary ore processing to use it, after enrichment and fuel production, to generate greenhouse gas lean energy in a nuclear reactor. Energy neutrality is reached if the energy produced from the extracted uranium is equal to or larger than the energy required for primary ore processing, uranium extraction, -conversion, -enrichment and -fuel production. This work discusses the sustainability of energy neutral mineral processing and provides an overview of the current progress of a multinational research project on that topic conducted under the umbrella of the International Atomic Energy Agency.

**Keywords:** energy neutral mineral processing; unconventional uranium extraction; comprehensive extraction; high temperature reactors

#### 1. Introduction to Energy Neutral Mineral Processing

Energy neutral mineral processing is the extraction of unconventional uranium (and/or thorium) during primary ore processing for use as raw material to produce nuclear reactor fuel [1]. Energy neutrality is reached if the extracted unconventional uranium is used to generate energy equivalent to or larger than the amount of energy required for mineral processing of the primary ore and uranium extraction, -conversion, -enrichment and -fuel production. Figure 1 illustrates the very basic idea of energy neutral mineral processing.

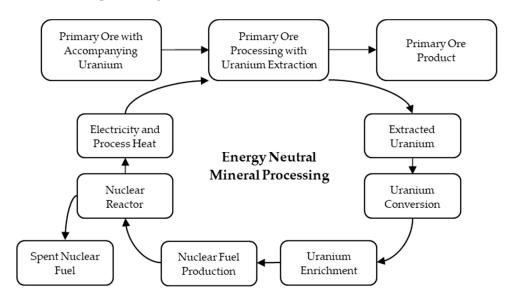


Figure 1. Brief overview of energy neutral mineral processing.

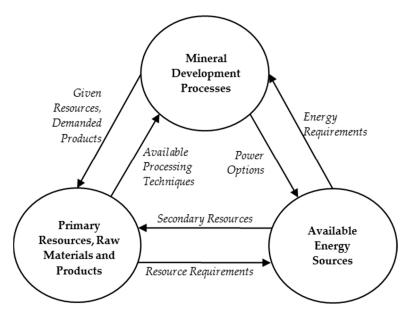
Extracted uranium, usually shipped as uranium ore concentrate (UOC), can be sent for uranium conversion/enrichment and later nuclear fuel production at a nuclear fuel manufacturer in the same way traditional uranium mines handle their product. Poly-metallic mines such as the Olympic Dam mine in Australia (copper, uranium, silver and gold mine) are already in operation today [2]. The efficiency of energy neutral mineral processing may further be enhanced if the energy source is deployed in the vicinity of the processing plant and even more so if used to directly supply process heat for energy intensive mineral ore development and/or supporting energy intensive operations such as water desalination [3–6]. Spent nuclear fuel from the energy source can be safely stored [7] or reprocessed.

## 2. Motivation Behind Energy Neutral Mineral Processing

Today between 8% and 10% of the world total energy consumption is dedicated to the extraction of materials that the society demands [8]. This number does not take into account metallurgical processes, transport and other mining-related activities. Ore grades worldwide are depleting rapidly while the demand for mineral commodities is constantly rising [9–13]. Processing lower grade mineral resources constitutes a general trend that is associated with often exponentially increasing energy requirements [14–17] and larger amounts of mine tailings. Environmental concerns such as greenhouse gas emissions, water and land usage, waste treatment, etc. are becoming more important in the mineral processing industry today [18–20] leading to planned or already imposed legislations regarding the

use of cleaner energy sources, as well as increased responsibilities to beneficiate/process mine tailings or waste materials. The majority of mineral processing operations today are powered by burning fossil fuels. In addition to renewable energy sources nuclear power may be a viable option to provide the large amounts of electricity and process heat required for present and future mineral processing operations with a drastically reduced environmental footprint.

Available resources and raw materials, as well as demanded products made from these, will determine the mineral development processes that have to be used for ore development. The required mineral processes will determine the energy demand that will have to be covered by an available energy source. The energy source will have certain resource requirements that may partly or, in the case of energy neutral mineral processing, entirely be covered by unconventional resources from byproduct extraction. The described interdependency is illustrated in Figure 2. While thermal nuclear power plants presently use uranium fuel (and in the future possibly thorium fuel), present day renewable energy sources require relevant amounts of rare earth elements (REE) for their production [21–27]. These REEs may also be (at least partly) provided as a byproduct from primary ore processing.

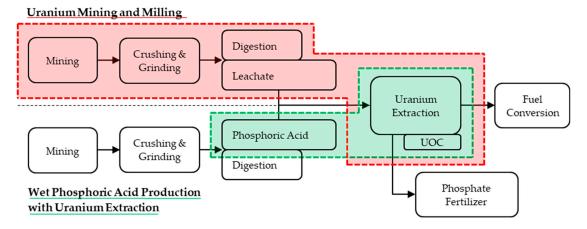


**Figure 2.** Interdependence of available primary resources, raw materials and demanded products (**bottom left**), required mineral development processes (**top**) and available energy sources (**bottom right**).

Besides currently deployed mineral processes that will require increasing amounts of energy to process lower grade ores, certain lower grade resources may have to be developed using alternative processing techniques. At the moment, for instance, most (>90%) phosphate rock is processed using the wet phosphoric acid (WPA) process. The WPA process cannot develop lower grade phosphate rock. Pre-concentration of phosphate rock that ranges from simple scrubbing and screening to more advanced techniques such as flotation and calcination [28–31] is required to separate impurities prior to subsequent WPA processing. In addition to more advanced pre-concentration techniques, energy intensive thermal phosphate rock processing that can process lower grade ores may have to be used. Phosphate rock is mined for its phosphorus component that is vital to global food security. The availability of higher grade phosphate rock resources that can be processed using the WPA process is part of an active scientific discussion [32–36].

Nuclear power does not directly emit greenhouse gas emissions. Greenhouse gas emissions are, however, emitted indirectly as a result of uranium mining/milling and -enrichment as well as plant construction, -operation and -decommissioning. Besides greenhouse gas emissions uranium mining and milling operations generate radiotoxic mine tailings that can pose harm to the environment [37–39].

Uranium mining/milling already accounts for a considerable share of the environmental impact of nuclear power that may further increase with decreasing uranium ore grades [40,41]. In the long term this environmental impact of nuclear energy may be reduced if fast breeder reactors are deployed [42]. In the short term this environmental impact of nuclear power can partly be reduced if uranium is extracted as a byproduct from another primary ore. Again, phosphate rock is used as an example to illustrate the environmental impact of uranium byproduct extraction vs. traditional uranium mining and milling. Uranium byproduct extraction from merchant grade phosphoric acid, an intermediate product during WPA processing, has been practiced on an industrial scale in the past [43–46] and is at the edge of being profitable again today [47–49]. Phosphate rocks contain considerable amounts of accompanying uranium in quantity and concentration [50–54]. Figure 3 briefly compares uranium mining and milling (top) to uranium extraction from wet phosphoric acid (bottom). Extracted UOC is further transported to a facility for fuel conversion and subsequent uranium enrichment. Enriched uranium is used for nuclear reactor fuel production. It is anticipated that uranium ore is only mined for its uranium content so that the full environmental impact (red box Figure 3) needs to be taken into account. Since phosphate rock is mined, crushed/ground and digested for its phosphorus content anyway only the environmental impact for additional byproduct uranium extraction (green box Figure 3) has to be considered in a direct comparison.



**Figure 3.** Traditional uranium mining (**top**) in comparison with uranium extraction from wet phosphoric acid (WPA) (**bottom**).

In the case of phosphate rock, uranium extraction may be particularly desirable as radiotoxic uranium which is not extracted primarily (80–90%) transfers to the final fertilizer product [55] which is brought out onto agricultural soils [56,57]. Guidelines of the International Atomic Energy Agency (IAEA) for unconventional uranium extraction are in place and if applied correctly ensure that unconventional byproduct uranium extraction does not result in an additional risk regarding nuclear proliferation [58].

## 3. Mineral Processes Currently Considered

A number of ores show relevant amounts of accompanying uranium and could therefore be considered for energy neutral mineral processing. In the IAEA coordinated research project discussed here [59,60] more than a dozen different countries are participating and studying a variety of ores, determining the content of uranium and thorium, and testing different extraction processes. In Table 1 some detail is provided such as the list of IAEA member states presently participating in the study, the ores or other material forms being studied and some of the techniques and equipment used to study the ores and to evaluate its composition and uranium content. The list does not include all possible ores with potential for unconventional uranium extraction and is by no means complete but aims to

indicate where the initial work in this project was undertaken. In the sub-sections below some of these ores and processes are further discussed.

Countries Participating	Ores and Other Forms Considered in the Study	Techniques and Equipment Used
Argentina, China, Egypt, Germany, India, Indonesia, Kuwait, Malaysia, Mexico, Morocco, Philippines, Poland, Tanzania, Tunisia, Venezuela	Carbonates, Columbite-Tantalite, Copper Tailings, Ilmenite, Monazite, Oil Sludge, Phosphate Rock, Phosphogypsum, Phosphoric Acid, Red Mud (Cerro Impacto Laterite), Tin Slag, Xenotime	Alpha Spectroscopy, Atomic Absorption, Electronic Microscope, Field Emission Scanning Electron Microscope, Gamma Spectroscopy, Gas Chromatography, Inductive Coupled Plasma—Optical Emission Spectroscopy, Inductive Coupled Plasma—Mass Spectroscopy, Ion Chromatography, Petrological Microscope, Pyrometallurgy Reduction, Nuclear Activation Analysis, X-ray Fluorescence

Table 1. Details of the research project participation and areas of study.

## 3.1. Phosphate Rock

Phosphate rock is mined for its phosphorous content used for fertilizer production. Phosphate rock contains considerable amounts of uranium and REEs [52,61]. Phosphate rock can be developed using the wet-acid or the thermal route. At the moment, most (>90%) phosphate rock is processed using the wet-acid route (WPA process) while back in the 1950s, phosphate rock was developed to equal quantities using the wet-acid and thermal route in the U.S., the largest phosphate rock producer at that time. Uranium extraction from WPA is a well-known process and was practiced on an industrial scale in the 1980s–1990s until it became unprofitable due to decreasing uranium prices. With the chance of increasing uranium prices, uranium extraction from WPA is once again a very active field of research and new molecules and improved extraction techniques are being developed that may rival traditional open-pit and underground uranium mining as well as in situ leach operations in the near future. During WPA production nuclear process heat from a high temperature reactor (HTR) may best be used for energy intensive calcination prior to the digestion process [62] or providing energy for thermal phosphate rock processing [63–65]. At present, approximately 10% of all phosphate rock processed using the wet-acid route is calcined to drive off undesired impurities, notably carbonates and organic matter. Figure 4 illustrates the very basic flow sheet of wet-acid phosphate rock processing with calcination and uranium extraction currently investigated as part of the research project.

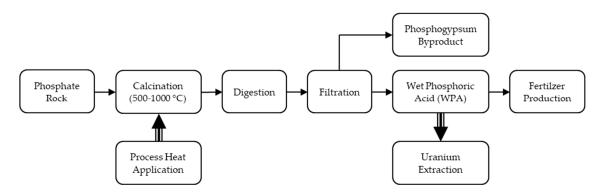


Figure 4. Uranium extraction during wet-phosphoric acid (WPA) production with calcination using process heat.

If calcination is used as a means of beneficiating phosphate rock prior to the digestion process, a cleaner ("light-green") WPA is produced. Based on lab-scale experiments it is believed that

WPA is considerably cheaper than uranium extraction from

uranium extraction from "light-green" WPA is considerably cheaper than uranium extraction from "merchant-grade" WPA produced using flotation as a means of beneficiation. However, this does not consider the costs for energy intensive phosphate rock calcination.

In addition to using HTRs to provide process heat for phosphate rock calcination concentrated solar power (CSP) is considered for this purpose in the research project. CSP has already been considered for calcination of limestone and solar calcination experiments were successfully conducted by Flamant et al. [66,67], Licht et al. [68], Meier et al. [69–72], as well as Salman and Kraishi [73]. Conveniently, the largest phosphate rock reserves are found in Northern Africa and the Middle East [74,75] where solar radiation is good to excellent. It was found that in the future phosphate rock calcination with both CSP and HTR may become economically competitive [76]. In addition to phosphate rock this research project is actively looking into processing byproduct phosphogypsum [77]. The WPA process generates some 5 tons of phosphogypsum per ton of phosphoric acid produced [55,78]. Phosphogypsum contains naturally occurring <sup>238</sup>U, <sup>226</sup>Ra, <sup>232</sup>Th and <sup>40</sup>K [79] and is presently stored indefinitely at most locations because of the resulting weak radioactivity.

#### 3.2. Rare Earth Element Ore

REE ores are usually leached using sulfuric-, hydrochloric- or nitric acid. Most minerals containing REEs are either phosphates (monazite and xenotime) or fluoro-carbonates (bastnaesite). To increase the solubility, low-temperature (<300 °C) or high-temperature (>300 °C) acid roasting is used prior to water leaching. The general acid roasting process for bastnaesite and monazite is provided in Figure 5 [80–82]. Uranium, thorium and REEs can be extracted from the leachate after several additional treatments [83,84].

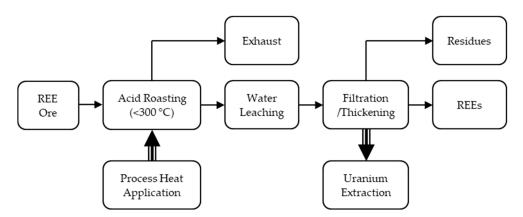


Figure 5. General high-temperature roasting and leaching process for bastnaesite and monazite.

HTRs and CSP are foreseen as greenhouse gas lean energy sources to provide process heat and/or electricity for acid roasting. If relevant uranium concentrations are found in the ore, uranium may be extracted as part of the process.

Other, rather unconventional, REE ores are laterites that result from the leaching and weathering of certain rock formations leading to the concentration of elements and resulting in interesting deposits that may contain thorium, REE and other valuable metals. Such is the case of the Cerro Impacto mineral deposit in Venezuela [85–87] that is investigated as part of this research project. It was found that the material is amenable to sulfuric acid leaching followed by solvent extraction for the recovery of niobium thorium and REE [88]. The process does however lead to highly contaminating acid effluents incompatible with the deep forest environment of its location. The use of a high temperature pyrometallurgical reduction process [89] has shown promise for a less contaminant procedure on a lab scale. With appropriate reductant and high temperature conditions, achieved with HTR process heat

and electricity, reduction to a metallic phase containing the iron, niobium, vanadium and nickel may be feasible.

#### 3.3. Copper Ore

Copper is one of the four most used metals in society and demand is expected to further increase [16]. Two main routes of primary copper production: pyrometallurgical and hydrometallurgical can be differentiated. The pyrometallurgical route accounts for some 80% of the world's primary copper production. Copper sulphide mineral deposits that can show relevant quantities of byproduct uranium are processed using the pyrometallurgical route. Byproduct uranium was economically successfully recovered at the Parabola mine in South Africa [90] and is economically successfully mined at the Olympic Dam mine in Australia [2,91]. Figure 6 provides a brief overview of pyrometallurgical copper production with uranium recovery. Furthermore, uranium can be recovered from copper mine tailings as for instance discussed by Chmielewski et al. [92].

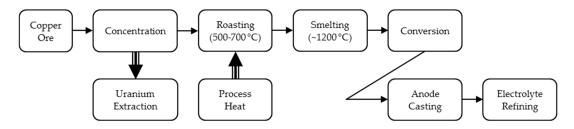


Figure 6. Pyrometallurgical copper ore processing with uranium recovery.

Copper production, especially pyrometallurgical copper production, is an energy intensive process [93,94]. HTRs do not reach the high temperatures required to deliver process heat for the smelting step. They may, however, be used to provide process heat for the roasting of the copper concentrate. In addition, HTRs could provide electricity for the production of pure copper and distilled water (through desalination of seawater [95]) of which plenty is used during the whole ore development process [96,97]. In addition to HTRs, CSP may again be a viable alternative in areas with high solar irradiation. This may be the case for Chile, the largest copper producing country in the world [98].

## 3.4. Tin Slag

The smelting process in tin production generates slag which contains radioactive elements such as uranium and thorium. Therefore tin slag often has to be categorized as technologically enhanced naturally occurring radioactive material (TENORM). Valuable elements such as REEs, niobium-tantalum, zirconium and titanium are contained in tin slag. The radioactive elements should be removed prior to extraction of those valuable elements. Some studies of tin slag characterization show the existence of an amorphous silica structure and resulting poor performance in direct leaching [99,100]. Pre-treatment of tin slag is required to increase the effectiveness of acid leaching. The silica structure can be broken by reaction with sodium hydroxide at high temperature (alkali fusion). As a result, silica is converted into a water-soluble form that can be dissolved in a subsequent water leaching step. In the next stage, direct leaching using sulfuric acid can dissolve uranium and thorium, while REEs stay in the solid phase [101]. Figure 7 provides a brief overview of the tin slag processing with uranium and thorium extraction that is presently investigated as part of the research project.

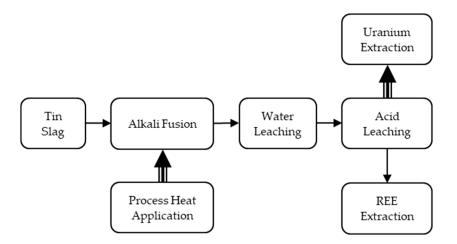


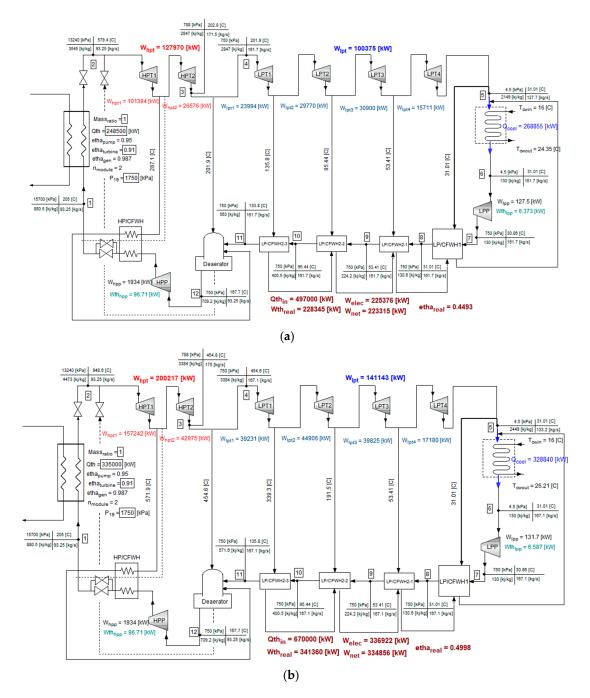
Figure 7. Tin slag processing with uranium and thorium recovery.

### 4. Modeling of the Coupled System

As part of the research project modeling of the coupled system consisting of HTRs and mineral processing plants is performed at IAEA. For now, hypothetical high temperature gas-cooled reactor (HTGR) steam cycle models are considered. These models can be extended to include intermediate heat exchangers (for higher temperature applications). Furthermore, simplified models can be used if only low temperature (waste heat recovery) is required. Details on the energy and heat requirements for mineral processing (temperatures, transport medium, availability, etc.) are derived for the different ores and processes. The HTGR models are designed based on the HTR-PM (high-temperature gas-cooled reactor pebble-bed module) demonstation plant that is presently constructed in China [102]. Since the thermal input required for mineral processing differs for each ore, the properties and systems are amended constantly. Currently, the model focuses on the Rankine cycle. Figure 8 shows two cases: the first case (top) is for 580 °C steam right after passing through the steam generator and the second case (bottom) is for 950 °C. One bleed stream from the high pressure turbine, three bleed streams from the low pressure turbine and one bleed stream from the deaerator between the high pressure turbine and low pressure turbine are considered. For each bleed stream a closed feed water heater is implemented. Besides, one more closed feed water heater is included for waste steam. The temperature of the cooling water of the condenser is designed to be 16 °C. The energy provided by the pumps is designed to be added at static pressure depending on efficiency while the rest is added at enthalpy value. The whole calculation is performed assuming that there are no heat and pressure losses. Due to this assumption, the thermal efficiency is slightly overestimated.

The models determine the total electrical output of the reactor as a result of non-electrical, high temperature heat applications for mineral processing. The models are used to evaluate the technical feasibility of coupling HTGRs to different mineral processing applications.





**Figure 8.** Hypothetical high temperature gas-cooled reactor (HTGR) models for the research project. (a) 580 °C steam cycle model; (b) 950 °C high-temperature model.

## 5. Identified Challenges

Small modular reactors, nuclear reactors with power levels less than or equal to 300 MWe, receive increased attention from numerous countries around the world. These small reactors are believed to fill a gap in the energy market as they may be constructed in a short time, can work on less developed energy grids and do not require the considerable upfront capital costs associated with currently operated large NPPs (nuclear power plants) that make purchasing them economically challenging for most countries [103–116]. Small modular reactors caused a similar euphoria in the 1960s and again in the 1980s that did not materialize due to the smaller reactors overall less favorable economic performance when compared to large (>1000 MWe) nuclear power plants (NPPs) [117,118] or other

energy generating technology. Present-day small modular reactor developers need to prove that their product can be built, operated and decommissioned economically in today's energy market to be accepted.

If small modular reactors are going to be built on a large scale, safe operation of these reactors is paramount. Most small modular reactors have technical features that make them different from currently deployed large (>1000 MWe) commercial light water reactors. Licensing small modular light water reactors with new features, such as passive cooling, is challenging and in most countries associated with increased costs for the licensing procedure and uncertainties regarding the outcome of the licensing process [119,120]. NuScale's present attempts to license their small modular reactor design in the U.S. may be a good example of this [121,122]. Licensing small modular reactors that are not only different in power level but show fundamental differences to currently deployed light water reactors such as a different coolant or moderator creates additional uncertainties for the licensor and thus increase the risk of the applicant to provide additional information, that results in increased costs for the overall licensing procedure [123–128]. Ramana et al. [129] provide an overview of the present status of licensing small modular reactors globally. Discussed are countries that already have a nuclear infrastructure in place and operate or operated commercial light water NPPs. Countries that cannot profit from this experience need to gain it so that the licensing process is further prolonged. Challenges associated with building an infrastructure to license small modular reactors are for instance discussed for Jordan [130].

In the case of energy neutral mineral processing, though it may be realized, HTGRs could be the first choice over other emerging HTR technology [131–133], such as the compact high temperature reactor (CHTR) [134,135], fluoride salt-cooled, high-temperature reactor (FHR) [136–139], gas-cooled fast reactor (GFR) [140–142], lead-cooled fast reactor (LFR) [143–145], molten salt reactor (MSR) [146] and others that may deliver process heat at temperatures higher than or equal to 600 °C. Five countries: Great Britain, the U.S., Germany, Japan and China have experience with operating and thus licensing HTGRs. An overview of past HTGR plants and projects is provided in Figure 9. The Republic of South Africa is included since considerable knowledge was gained as part of the PBMR (Pebble Bed Modular Reactor) project. However, the project did not result in actually building and operating an HTGR.



Figure 9. Past and present high temperature gas-cooled reactor (HTGR) plants/projects by country.

The UK operated the first HTGR research reactor called DRAGON from 1963 to 1976 [147]. The UK heavily relied on gas-cooled commercial reactors (other than HTGRs) in the past so that licensing HTGR plants may be less of a stretch for the UK licensing body than it could be in other countries that primarily relied on light water reactor technology. The U.S. and Germany are the only countries that operated research- (PB-1 and AVR) as well as prototype (FSR and THTR) HTGRs with China soon to join this group once the prototype HTR-PM has been constructed and is operational. Besides the HTR PM prototype reactor currently under construction in Shandong Province [102], China plans to build additional commercial HTGRs in the near future [148]. From a regulatory point of view, licensing a prototype reactor is somewhat different to licensing a research reactor. Research reactors are operated at research institutions or areas controlled by the military and are therefore (in most cases) easier to license than larger prototype reactors built primarily for power production. Besides, research reactors are usually much smaller and show greatly increased safety margins since economics are less of an issue and they are not built for power production [149,150].

Licensing HTGRs for process heat application is yet another challenge as the interconnection with the process heat receiving chemical plant needs to be taken into account. This may be particularly challenging in the case of hydrogen production [151–154]. The first experience on this was gained in Germany with the THTR that did not provide process heat but was erected in an industrial complex to learn about potential challenges for future HTGR process heat licensing procedures.

Countries that did not operate HTGRs in the past but operate large nuclear plants (Mexico, Brazil, South Africa, Russia, and India) may profit from this knowledge. Ideas for turnkey modular reactors, where maintenance, fuel supply and disposal is taken care of by the vendor/operator may also be an option in the future given the modular nature of past and present HTGRs. This may allow the deployment of HTGRs in newcomer countries that could profit from energy neutral mineral processing in the very far future.

## 6. Conclusions

Decreasing ore grades that result in much larger energy requirements for processing these ores as well as increasing public awareness regarding the way this energy is produced creates global interest in energy neutral mineral processing as a more sustainable way to power mineral development. At present, more than a dozen member states of the International Atomic Energy Agency (IAEA) investigate forms of energy neutral mineral processing. This work aims to provide a brief overview of the latest developments of these investigations. At present, the most prominently investigated ores are phosphate rock, rare earth element (REE) ores, copper ore and tin slag. High temperature gas-cooled reactors (HTGRs) such as the HTR-PM currently constructed in China and concentrated solar power (CSP) plants such as the Gemasolar plant operating in Spain are foreseen as greenhouse gas lean energy sources. Besides technical challenges, regulatory burdens have been identified as the largest challenge for energy neutral mineral processing.

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## References

- Haneklaus, N.; Reitsma, F.; Tulsidas, H.; Schnug, E.; Lottermoser, B.G.; Allelein, H.J. Energy Neutral Mineral Development Processes—An Overview. In Proceedings of the AIMS 2016, Aachen, Germany, 18–19 May 2016.
- 2. Robertson, A.; Grant, D.; Liebezeit, V.; Ehrig, K.; Badenhorst, C.; Durandt, G. Olympic Dam Mine, BHP Biliton. Australasian Mining Metallurgical Operating Practices. 2013. Available online: https://www.ausimm.com.au/publications/publication.aspx?ID=15461 (accessed on 17 January 2018).
- 3. Yan, X.; Noguchi, H.; Sato, H.; Tachibana, Y.; Kunitomi, K.; Hino, R. A hybrid HTGR system producing electricity, hydrogen and such other products as water demanded in the Middle East. *Nucl. Eng. Des.* **2014**, 271, 20–29. [CrossRef]
- 4. Dardour, S.; Nisan, S.; Charbit, F. Utilisation of waste heat from GT-MHR and PBMR reactors for nuclear desalination. *Desalination* **2007**, 205, 254–268. [CrossRef]
- Nisan, S.; Benzarti, N. A comprehensive economic evaluation of integrated desalination systems using fossil fuelled and nuclear energies and including their environmental costs. *Desalination* 2008, 229, 125–146. [CrossRef]
- Nisan, S.; Dardour, S. Economic evaluation of nuclear desalination systems. *Desalination* 2007, 205, 231–242. [CrossRef]
- 7. Peterson, P.F. Spent Nuclear Fuel is not the Problem. Proc. IEEE 2017, 105, 411–414. [CrossRef]

- IEA. World Energy Outlook 2016 (Executive Summary). World Energy Outlook. 2016, pp. 1–8. Available online: http://www.iea.org/publications/freepublications/publication/WEB\_WorldEnergyOutlook2015 ExecutiveSummaryEnglishFinal.pdf (accessed on 12 January 2018).
- 9. Norgate, T.; Jahanshahi, S. Low grade ores—Smelt, leach or concentrate? *Miner. Eng.* 2010, 23, 65–73. [CrossRef]
- 10. Ragnarsdóttir, K.V. Rare metals getting rarer. Nat. Geosci. 2008, 1, 720–721. [CrossRef]
- 11. West, J. Decreasing Metal Ore Grades: Are They Really Being Driven by the Depletion of High-Grade Deposits? *J. Ind. Ecol.* **2011**, *15*, 165–168. [CrossRef]
- 12. Dittmar, M. The end of cheap uranium. Sci. Total Environ. 2013, 461–462, 792–798. [CrossRef] [PubMed]
- 13. Prior, T.; Giurco, D.; Mudd, G.; Mason, L.; Behrisch, J. Resource depletion, peak minerals and the implications for sustainable resource management. *Glob. Environ. Chang.* **2012**, *22*, 577–587. [CrossRef]
- 14. Norgate, T.; Haque, N. Energy and greenhouse gas impacts of mining and mineral processing operations. *J. Clean. Prod.* **2010**, *18*, 266–274. [CrossRef]
- 15. Northey, S.; Mohr, S.; Mudd, G.M.; Weng, Z.; Giurco, D. Modelling future copper ore grade decline based on a detailed assessment of copper resources and mining. *Resour. Conserv. Recycl.* 2014, *83*, 190–201. [CrossRef]
- 16. Elshkaki, A.; Graedel, T.E.; Ciacci, L.; Reck, B. Copper demand, supply, and associated energy use to 2050. *Glob. Environ. Chang.* **2016**, *39*, 305–315. [CrossRef]
- 17. Calvo, G.; Mudd, G.; Valero, A.; Valero, A. Decreasing Ore Grades in Global Metallic Mining: A Theoretical Issue or a Global Reality? *Resources* **2016**, *5*, 36. [CrossRef]
- 18. Mudd, G.M. The Environmental sustainability of mining in Australia: Key mega-trends and looming constraints. *Resour. Policy* **2010**, *35*, 98–115. [CrossRef]
- 19. Hilson, G. Corporate Social Responsibility in the extractive industries: Experiences from developing countries. *Resour. Policy* **2012**, *37*, 131–137. [CrossRef]
- 20. Owen, J.R.; Kemp, D. Social licence and mining: A critical perspective. *Resour. Policy* **2013**, *38*, 29–35. [CrossRef]
- 21. Vesborg, P.C.K.; Jaramillo, T.F. Addressing the terawatt challenge: Scalability in the supply of chemical elements for renewable energy. *RSC Adv.* **2012**, *2*, 7933–7947. [CrossRef]
- 22. Bradshaw, A.M.; Reuter, B.; Hamacher, T. The potential scarcity of rare elements for the Energiewende. *Green* **2013**, *3*, 93–111. [CrossRef]
- 23. Habib, K.; Wenzel, H. Exploring rare earths supply constraints for the emerging clean energy technologies and the role of recycling. *J. Clean. Prod.* **2014**, *84*, 348–359. [CrossRef]
- 24. Arent, D.; Pless, J.; Mai, T.; Wiser, R.; Hand, M.; Baldwin, S.; Heath, G.; Macknick, J.; Bazilian, M.; Schlosser, A.; et al. Implications of high renewable electricity penetration in the U.S. for water use, greenhouse gas emissions, land-use, and materials supply. *Appl. Energy* **2014**, *123*, 368–377. [CrossRef]
- 25. Moss, R.L.; Tzimas, E.; Kara, H.; Willis, P.; Kooroshy, J. The potential risks from metals bottlenecks to the deployment of Strategic Energy Technologies. *Energy Policy* **2013**, *55*, 556–564. [CrossRef]
- 26. Baldi, L.; Peri, M.; Vandone, D. Clean energy industries and rare earth materials: Economic and financial issues. *Energy Policy* **2014**, *66*, 53–61. [CrossRef]
- 27. Stegen, K.S. Heavy rare earths, permanent magnets, and renewable energies: An imminent crisis. *Energy Policy* **2015**, *79*, 1–8. [CrossRef]
- 28. Kawatra, S.K.; Carlson, J.T. Beneficiation of Phosphate Ore; SME: Englewood, CO, USA, 2013; p. 168.
- 29. Abouzeid, A. Upgrading of Phosphate Ores-A Review. J. ORE Dress. 2007, 9, 15-20.
- 30. Zhang, P. Comprehensive recovery and sustainable development of phosphate resources. *Procedia Eng.* **2014**, *83*, 37–51. [CrossRef]
- 31. El-Shall, H.; Zhang, P.; Abdel-Khalek, N.; El-Mofty, S. Beneficiation technology of phosphates: Challenges and solutions. *Miner. Metall. Process.* **2004**, *21*, 17–26.
- Ulrich, A.E.; Frossard, E. On the history of a reoccurring concept: Phosphorus scarcity. *Sci. Total Environ.* 2014, 490, 694–707. [CrossRef] [PubMed]
- 33. Cordell, D.; White, S. Peak phosphorus: Clarifying the key issues of a vigorous debate about long-term phosphorus security. *Sustainability* **2011**, *3*, 2027–2049. [CrossRef]
- Smil, V. Phousphorus in the environment: Natural Flows and Human Interferences. *Annu. Rev. Energy Environ.* 2000, 25, 53–88. [CrossRef]

- 35. Cordell, D.; Drangert, J.O.; White, S. The story of phosphorus: Global food security and food for thought. *Glob. Environ. Chang.* **2009**, *19*, 292–305. [CrossRef]
- 36. Van Vuuren, D.P.; Bouwman, A.F.; Beusen, A.H.W. Phosphorus demand for the 1970–2100 period: A scenario analysis of resource depletion. *Glob. Environ. Chang.* **2010**, *20*, 428–439. [CrossRef]
- Carvalho, F.P.; Madruga, M.J.; Reis, M.C.; Alves, J.G.; Oliveira, J.M.; Gouveia, J.; Silva, L. Radioactivity in the environment around past radium and uranium mining sites of Portugal. *J. Environ. Radioact.* 2007, *96*, 39–46. [CrossRef] [PubMed]
- Tripathi, R.M.; Sahoo, S.K.; Jha, V.N.; Khan, A.H.; Puranik, V.D. Assessment of environmental radioactivity at uranium mining, processing and tailings management facility at Jaduguda, India. *Appl. Radiat. Isot.* 2008, 66, 1666–1670. [CrossRef] [PubMed]
- 39. Abdelouas, A. Uranium mill tailings: Geochemistry, mineralogy, and environmental impact. *Elements* **2006**, 2, 335–341. [CrossRef]
- 40. Norgate, T.; Haque, N.; Koltun, P. The impact of uranium ore grade on the greenhouse gas footprint of nuclear power. *J. Clean. Prod.* **2014**, *84*, 360–367. [CrossRef]
- 41. Lenzen, M. Life cycle energy and greenhouse gas emissions of nuclear energy: A review. *Energy Convers. Manag.* **2008**, *49*, 2178–2199. [CrossRef]
- 42. Poinssot, C.; Bourg, S.; Ouvrier, N.; Combernoux, N.; Rostaing, C.; Vargas-Gonzalez, M.; Bruno, J. Assessment of the environmental footprint of nuclear energy systems. Comparison between closed and open fuel cycles. *Energy* **2014**, *69*, 199–211. [CrossRef]
- 43. Beltrami, D.; Cote, G.; Mokhtari, H.; Courtaud, B.; Moyer, B.A.; Chagnes, A. Recovery of Uranium from Wet Process Phosphoric Acid by Solvent Extraction. *Chem. Rev.* **2014**, *114*, 12002–12023. [CrossRef] [PubMed]
- 44. Singh, D.K.; Mondal, S.; Chakravartty, J.K. Recovery of Uranium from Phosphoric Acid: A Review. *Solvent Extr. Ion Exch.* **2016**, *34*, 201–225. [CrossRef]
- 45. Bunus, F.T. Uranium and Rare Earth Recovery from Phosphate Fertilizer Industry by Solvent Extraction. *Miner. Process. Extr. Metall. Rev.* **2000**, *21*, 381–478. [CrossRef]
- 46. Astley, V.; Stana, R. There and Back Again 2.5Again Who did What in Solvent Extraction? A Demonstrated & Proven Technology for Uranium Recovery from Phosphoric Acid. *Procedia Eng.* **2014**, *83*, 270–278. [CrossRef]
- 47. Kim, H.; Eggert, R.G.; Carlsen, B.W.; Dixon, B.W. Potential uranium supply from phosphoric acid: A U.S. analysis comparing solvent extraction and Ion exchange recovery. *Resour. Policy* **2016**, *49*, 222–231. [CrossRef]
- 48. Haneklaus, N.; Sun, Y.; Bol, R.; Lottermoser, B.; Schnug, E. To Extract, or not to Extract Uranium from Phosphate Rock, that is the Question. *Environ. Sci. Technol.* **2016**, 753–754. [CrossRef] [PubMed]
- 49. Hore-Lacy, I. 9-Production of byproduct uranium and uranium from unconventional resources. *Uranium Nucl. Power* **2016**, 239–251. [CrossRef]
- 50. Gabriel, S.; Baschwitz, A.; Mathonnière, G.; Eleouet, T.; Fizaine, F. A critical assessment of global uranium resources, including uranium in phosphate rocks, and the possible impact of uranium shortages on nuclear power fleets. *Ann. Nucl. Energy* **2013**, *58*, 213–220. [CrossRef]
- 51. Schnug, E.; Haneklaus, N. Uranium, the hidden treasure in phosphates. *Procedia Eng.* **2014**, *83*, 265–269. [CrossRef]
- 52. Chen, M.; Graedel, T.E. The potential for mining trace elements from phosphate rock. J. Clean. Prod. 2015, 91, 337–346. [CrossRef]
- 53. Ulrich, A.E.; Schnug, E.; Prasser, H.M.; Frossard, E. Uranium endowments in phosphate rock. *Sci. Total Environ.* 2014, 478, 226–234. [CrossRef] [PubMed]
- 54. Schnug, E.; Haneklaus, N. Energetic and economic significance of uranium in mineral phosphorous fertilizers. In *The New Uranium Mining Boom*; Springer: Berlin/Heidelberg, Germany, 2011.
- Rutherford, P.M.; Dudas, M.J.; Samek, R.A. Environmental impacts of phosphogypsum. *Sci. Total Environ*. 1994, 149, 1–38. [CrossRef]
- 56. Sabiha-Javied; Mehmood, T.; Chaudhry, M.M.; Tufail, M.; Irfan, N. Heavy metal pollution from phosphate rock used for the production of fertilizer in Pakistan. *Micorchem. J.* **2009**, *91*, 94–99.
- Schnug, E.; Lottermoser, B.G. Fertilizer-Derived Uranium and its Threat to Human Health. *Environ. Sci. Technol.* 2013, 47, 2433–2434. [CrossRef] [PubMed]
- 58. Haneklaus, N.; Bayok, A.; Fedchenko, V. Phosphate Rocks and Nuclear Proliferation. *Sci. Glob. Secur.* 2017, 25, 143–158. [CrossRef]

- 59. Haneklaus, N.; Reitsma, F.; Tulsidas, H. High Temperature Reactors for a new IAEA Coordinated Research Project on energy neutral mineral development processes. *Nucl. Eng. Des.* **2016**, *306*, 198–202. [CrossRef]
- 60. Haneklaus, N.; Reitsma, F.; Tulsidas, H.; Tyobeka, B.; Schnug, E.; Allelein, H.J.; Birky, B.; Peterson, P.F.; Dyck, G.; Koshy, T. Using high temperature reactors for energy neutral mineral development processes a proposed IAEA Coordinated Research Project. In Proceedings of the International Symposium on Uranium Raw Material for the Nuclear Fuel Cycle, Vienna, Austria, 23–27 June 2014.
- 61. Emsbo, P.; McLaughlin, P.I.; Breit, G.N.; du Bray, E.A.; Koenig, A.E. Rare earth elements in sedimentary phosphate deposits: Solution to the global REE crisis? *Gondwana Res.* **2015**, *27*, 776–785. [CrossRef]
- 62. Haneklaus, N.; Zheng, Y.; Allelein, H.J. Stop Smoking—Tube-In-Tube Helical System for Flameless Calcination of Minerals. *Processes* 2017, *5*, 67. [CrossRef]
- 63. Haneklaus, N.; Schnug, E.; Tulsidas, H.; Tyobeka, B. Annals of Nuclear Energy Using high temperature gas-cooled reactors for greenhouse gas reduction and energy neutral production of phosphate fertilizers. *Ann. Nucl. Energy* **2015**, *75*, 275–282. [CrossRef]
- 64. Haneklaus, N.; Reyes, R.; Lim, W.G.; Tabora, E.U.; Palattao, B.L.; Petrache, C.; Vargas, E.P.; Kunitomi, K.; Ohashi, H.; Sakaba, N.; et al. Energy Neutral Phosphate Fertilizer Production Using High Temperature Reactors: A Philippine Case Study. *Philipp. J. Sci.* **2015**, *144*, 69–79.
- 65. Haneklaus, N.; Schnug, E. Energy Neutral Phosphate Fertilizer Production Using High Temperature Reactors. In *Phosphorus Agriculture: 100% Zero*; Springer: Dordrecht, The Netherlands, 2016; pp. 309–316.
- 66. Flamant, G.; Hernandez, D.; Bonet, C.; Traverse, J.P. Experimental aspects of the thermochemical conversion of solar energy; Decarbonation of CaCO3. *Sol. Energy* **1980**, *24*, 385–395. [CrossRef]
- 67. Flamant, G.; Gauthier, D.; Boudhari, C.; Flitris, Y. A 50 kW fluidized bed high temperature solar receiver: Heat transfer analysis. *J. Sol. Energy Eng.* **1988**, *110*, 313–320. [CrossRef]
- Licht, S.; Wu, H.; Hettige, C.; Wang, B.; Asercion, J.; Lau, J.; Stuart, J. STEP cement: Solar Thermal Electrochemical Production of CaO without CO<sub>2</sub> emission. *Chem. Commun.* 2012, 48, 6019. [CrossRef] [PubMed]
- 69. Meier, A.; Bonaldi, E.; Cella, G.M.; Lipinski, W. Multitube Rotary Kiln for the Industrial Solar Production of Lime. *J. Sol. Energy Eng.* **2005**, 127, 386–395. [CrossRef]
- 70. Meier, A.; Gremaud, N.; Steinfeld, A. Economic evaluation of the industrial solar production of lime. *Energy Convers. Manag.* **2005**, *46*, 905–926. [CrossRef]
- Meier, A.; Bonaldi, E.; Cella, G.M.; Lipinski, W.; Wuillemin, D.; Palumbo, R. Design and experimental investigation of a horizontal rotary reactor for the solar thermal production of lime. *Energy* 2004, 29, 811–821. [CrossRef]
- 72. Meier, A.; Bonaldi, E.; Cella, G.M.; Lipinski, W.; Wuillemin, D. Solar chemical reactor technology for industrial production of lime. *Sol. Energy* **2006**, *80*, 1355–1362. [CrossRef]
- 73. Salman, O.A.; Kraishi, N. Thermal decomposition of limestone and gypsum by solar energy. *Sol. Energy* **1988**, *41*, 305–308. [CrossRef]
- 74. Cooper, J.; Lombardi, R.; Boardman, D.; Carliell-Marquet, C. The future distribution and production of global phosphate rock reserves. *Resour. Conserv. Recycl.* **2011**, *57*, 78–86. [CrossRef]
- 75. Edixhoven, J.D.; Gupta, J.; Savenije, H.H.G. Recent revisions of phosphate rock reserves and resources: A critique. *Earth Syst. Dyn.* **2014**, *5*, 491–507. [CrossRef]
- Haneklaus, N.; Schröders, S.; Zheng, Y.; Allelein, H.-J. Economic evaluation of flameless phosphate rock calcination with concentrated solar power and high temperature reactors. *Energy* 2017, 140, 1148–1157. [CrossRef]
- 77. Haneklaus, N.; Tulsidas, H.; Reitsma, F.; Schnug, E. Using high temperature reactors for energy neutral phosphate fertilizer and phosphogypsum processing. In *Uranium-Past and Future Challenges*; Merkel, B., Arab, A., Eds.; Springer: Cham, Switzerland, 2014; pp. 785–792.
- 78. Tayibi, H.; Choura, M.; López, F.A.; Alguacil, F.J.; López-Delgado, A. Environmental impact and management of phosphogypsum. *J. Environ. Manag.* **2009**, *90*, 2377–2386. [CrossRef] [PubMed]
- 79. Sahu, S.K.; Ajmal, P.Y.; Bhangare, R.C.; Tiwari, M.; Pandit, G.G. Natural radioactivity assessment of a phosphate fertilizer plant area. *J. Radiat. Res. Appl. Sci.* **2014**, *7*, 123–128. [CrossRef]
- 80. Zhang, J.; Edwards, C. Mineral decomposition and leaching processes for treating rare earth ore concentrates. *Can. Metall. Q.* **2013**, *52*, 243–248. [CrossRef]

- 81. U.S. Geological Survey. *Phosphate Rock*; Mineral Commodity Summaries; U.S. Geological Survey: Reston, VA, USA, 2017; pp. 124–125. [CrossRef]
- 82. Kumari, A.; Panda, R.; Jha, M.K.; Lee, J.Y.; Kumar, J.R.; Kumar, V. Thermal treatment for the separation of phosphate and recovery of rare earth metals (REMs) from Korean monazite. *J. Ind. Eng. Chem.* **2015**, *21*, 696–703. [CrossRef]
- 83. Habashi, F. Extractive metallurgy of rare earths. Can. Metall. Q. 2013, 52, 224–233. [CrossRef]
- 84. Amaral, J.C.B.S.; Morais, C.A. Thorium and uranium extraction from rare earth elements in monazite sulfuric acid liquor through solvent extraction. *Miner. Eng.* **2010**, *23*, 498–503. [CrossRef]
- 85. Aarden, H.M.; Arozena, J.M.I.; Moticska, P.; Navarro, J.; Pasquali, J.; Sifontes, R.S. *El Complejo Geológico del area de Impacto, Distrito Cedeño, Estado Bolivar, Venezuela*; Ministerio de Minas e Hidrocarburos, Dirección de Geología: Caracas, Venezuela, 1973.
- Greaves, E.D.; Pasquali, J.; Sifontes, R.S.; Manrique, M. Cerro Impacto, Venezuela's Thorium Deposit. Thorium Energy Conference. London, UK, 2010. Available online: http://www.nuclear.fis.usb.ve/fn/wpcontent/uploads/2018/01/Greaves-IMPACTO-Venezuela%E2%80%99s-Thorium-Deposit.pdf (accessed on 17 January 2018).
- 87. Eduardo, D.; Greaves, J.; Pasquali, R.S.S.; Impacto, M.M. *Venezuela's Thorium Deposit*; World Thorium Resource; Thiruvantaphuran: Kerala, India, 2011.
- Manrique, M.; Greaves, E.D.; Garcia, C. Extraction of Rare Earths from low-grade lateritic ore by pressure leaching. In Proceedings of the First International Conference on Processing Materials for Properties, Honolulu, HI, USA, 7–10 November 1993; pp. 445–448.
- 89. José, A.H. *"Lixiviación de Niobio y Torio" Tesis de Ingeniería Metalúrgica;* Facultad de Ingeniería Universidad Central de Venezuela: Caracas, Venezuela, 1974.
- 90. Fordt, M.A. Uranium in South Africa. J. South. Afr. Inst. Min. Metall. 1993, 93, 37-58. [CrossRef]
- 91. Gupta, C.; Singh, H. Uranium Resource Processing—Secondary Resources; Springer: Berlin, Germany, 2003.
- 92. Chmielewski, A.G.; Wawszczak, D.; Brykała, M. Possibility of uranium and rare metal recovery in the Polish copper mining industry. *Hydrometallurgy* **2016**, *159*, 12–18. [CrossRef]
- 93. Memary, R.; Giurco, D.; Mudd, G.; Mason, L. Life cycle assessment: A time-series analysis of copper. *J. Clean. Prod.* **2012**, *33*, 97–108. [CrossRef]
- 94. Alvarado, S.; Maldonado, P.; Barrios, A.; Jaques, I. Long term energy-related environmental issues of copper production. *Energy* **2002**, *27*, 183–196. [CrossRef]
- 95. Kim, H.S.; No, H.C. Thermal coupling of HTGRs and MED desalination plants, and its performance and cost analysis for nuclear desalination. *Desalination* **2012**, *303*, 17–22. [CrossRef]
- 96. Pena, C.A.; Huijbregts, M.A.J. The blue water footprint of primary copper production in Northern Chile. *J. Ind. Ecol.* **2014**, *18*, 49–58. [CrossRef]
- 97. Moreno, P.A.; Aral, H.; Cuevas, J.; Monardes, A.; Adaro, M.; Norgate, T.; Bruckard, W. The use of seawater as process water at Las Luces copper-molybdenum beneficiation plant in Taltal (Chile). *Miner. Eng.* **2011**, *24*, 852–858. [CrossRef]
- 98. Moreno-Leiva, S.; Díaz-Ferrán, G.; Haas, J.; Telsnig, T.; Díaz-Alvarado, F.A.; Palma-Behnke, R.; Kracht, W.; Román, R.; Chudinzow, D.; Eltrop, L. Towards solar power supply for copper production in Chile: Assessment of global warming potential using a life-cycle approach. J. Clean. Prod. 2017, 164, 242–249. [CrossRef]
- 99. Gaballah, I.; Allain, E.; Meyer-Joly, M.C.; Malau, K. A Possible Method for the Characterization of Amorphous Slags Recovery of Refractory Metal Oxides from Tin Slags. *Metall. Trans.* **1992**, *23*, 249–259. [CrossRef]
- 100. Subramanian, C.; Suri, A.K.; Atomic, B. Recovery of niobium and tantalum from low grade tin slag—A hydrometallurgical approach. In *Environmental & Waste Management in NoN-Ferrous Metallurgical Industries*; NML: Jamshedpur, India, 1998; pp. 100–107.
- 101. Trinopiawan, K.; Mubarok, M.Z.; Mellawati, J.; Ani, B.Y. Pelindian Logam Tanah Jarang dari Terak Timah dengan Asam Klorida Setelah Proses Fusi Alkali. Buletin Pusat Teknologi Bahan Galian Nuklir 2016, 1, 37. [CrossRef]
- 102. Zhang, Z.; Dong, Y.; Li, F.; Zhang, Z.; Wang, H.; Huang, X.; Li, H.; Liu, B.; Wu, X.; Wang, H.; et al. The Shandong Shidao Bay 200 MWe High-Temperature Gas-Cooled Reactor Pebble-Bed Module (HTR-PM) Demonstration Power Plant: An Engineering and Technological Innovation. *Engineering* 2016, 2, 112–118. [CrossRef]

- 103. Reutler, H.; Lohnert, G.H. Advantages of going modular in HTRs. Nucl. Eng. 1984, 78, 129–136. [CrossRef]
- 104. Carelli, M.D.; Garrone, P.; Locatelli, G.; Mancini, M.; Mycoff, C.; Trucco, P.; Ricotti, M.E. Economic features of integral, modular, small-to-medium size reactors. *Prog. Nucl. Energy* **2010**, *52*, 403–414. [CrossRef]
- Boarin, S.; Ricotti, M.E. An Evaluation of SMR Economic Attractiveness. Sci. Technol. Nucl. Install. 2014, 2014, 803698. [CrossRef]
- 106. Carlsson, J.; Shropshire, D.E.; van Heek, A.; Fütterer, M.A. Economic viability of small nuclear reactors in future European cogeneration markets. *Energy Policy* **2012**, *43*, 396–406. [CrossRef]
- Kessides, I.N.; Kuznetsov, V. Small modular reactors for enhancing energy security in developing countries. Sustainability 2012, 4, 1806–1832. [CrossRef]
- 108. Hampe, J.; Madlener, R. *Economics of High-Temperature Nuclear Reactors for Industrial Cogeneration;* Institute for Future Energy Consumer Needs and Behavior (FCN): Aachen, Germany, 2012.
- Vujić, J.; Bergmann, R.M.; Škoda, R.; Miletić, M. Small modular reactors: Simpler, safer, cheaper? *Energy* 2012, 45, 288–295. [CrossRef]
- 110. Alonso, G.; Ramirez, R.; del Valle, E.; Castillo, R. Process heat cogeneration using a high temperature reactor. *Nucl. Eng. Des.* **2014**, *280*, 137–143. [CrossRef]
- 111. Locatelli, G.; Bingham, C.; Mancini, M. Small modular reactors: A comprehensive overview of their economics and strategic aspects. *Prog. Nucl. Energy* **2014**, *73*, 75–85. [CrossRef]
- Hidayatullah, H.; Susyadi, S.; Subki, M.H. Design and technology development for small modular reactors—Safety expectations, prospects and impediments of their deployment. *Prog. Nucl. Energy* 2015, 79, 127–135. [CrossRef]
- Locatelli, G.; Fiordaliso, A.; Boarin, S.; Ricotti, M.E. Cogeneration: An option to facilitate load following in Small Modular Reactors. *Prog. Nucl. Energy* 2017, 97, 153–161. [CrossRef]
- 114. Ingersoll, D.T. Deliberately small reactors and the second nuclear era. *Prog. Nucl. Energy* **2009**, *51*, 589–603. [CrossRef]
- Shropshire, D. Economic viability of small to medium-sized reactors deployed in future European energy markets. *Prog. Nucl. Energy* 2011, 53, 299–307. [CrossRef]
- 116. Angulo, C.; Bogusch, E.; Bredimas, A.; Delannay, N.; Viala, C.; Ruer, J.; Muguerra, P.; Sibaud, E.; Chauvet, V.; Hittner, D.; et al. EUROPAIRS: The European project on coupling of High Temperature Reactors with industrial processes. *Nucl. Eng. Des.* 2012, 251, 30–37. [CrossRef]
- 117. Sovacool, B.K.; Ramana, M.V. Back to the Future. Sci. Technol. Hum. Values 2015, 40, 96–125. [CrossRef]
- 118. Thomas, S. The Pebble Bed Modular Reactor: An obituary. Energy Policy 2011, 39, 2431–2440. [CrossRef]
- 119. Söderholm, K. Challenges of Smr Licensing Practices. CNL Nucl. Rev. 2012, 1, 19–31. [CrossRef]
- 120. Sainati, T.; Locatelli, G.; Brookes, N. Small Modular Reactors: Licensing constraints and the way forward. *Energy* **2015**, *82*, 1092–1095. [CrossRef]
- Ingersoll, D.T. Integration of Nuscale Smr with Desalination Technologies. In Proceedings of the ASME 2014 Small Modular Reactors Symposium, Washington, DC, USA, 15–17 April 2014; pp. 1–3392. [CrossRef]
- 122. Ingersoll, D.T.; Houghton, Z.J.; Bromm, R.; Desportes, C. NuScale small modular reactor for Co-generation of electricity and water. *Desalination* **2014**, 340, 84–93. [CrossRef]
- Brinkmann, G.; Will, M. Concept licensing procedure for an htr-module nuclear power plant. *Nucl. Eng. Des.* 1990, 121, 293–298. [CrossRef]
- 124. Delaney, M.J.; Apostolakis, G.E.; Driscoll, M.J. Risk-informed design guidance for future reactor systems. *Nucl. Eng. Des.* **2005**, 235, 1537–1556. [CrossRef]
- 125. Scarlat, R.O.; Laufer, M.R.; Blandford, E.D.; Zweibaum, N.; Krumwiede, D.L.; Cisneros, A.T.; Andreades, C.; Forsberg, C.W.; Greenspan, E.; Hu, L.W.; et al. Design and licensing strategies for the fluoride-salt-cooled, high-temperature reactor (FHR) technology. *Prog. Nucl. Energy* 2014, 77, 406–420. [CrossRef]
- 126. Nickel, H.; Hofmann, K.; Wachholz, W.; Weisbrodt, I. The helium-cooled high-temperature reactor in the Federal Republic of Germany: Safety features, integrity concept, outlook for design codes and licensing procedures. *Nucl. Eng. Des.* **1991**, *127*, 181–190. [CrossRef]
- Trikouros, N.G. A Perspective on Small Reactor Licensing and Implementation. *Nucl. Technol.* 2012, 178, 233–239. [CrossRef]
- 128. Silady, F.A.; Cunliffe, J.C.; Walker, L.P. The licensing experience of the Modular High-Temperature Gas-Cooled Reactor (MHTGR). *Energy* **1991**, *16*, 417–424. [CrossRef]

- 129. Ramana, M.V.; Hopkins, L.B.; Glaser, A. Licensing small modular reactors. *Energy* 2013, *61*, 555–564. [CrossRef]
- 130. Ramana, M.V.; Ahmad, A. Wishful thinking and real problems: Small modular reactors, planning constraints, and nuclear power in Jordan. *Energy Policy* **2016**, *93*, 236–245. [CrossRef]
- 131. Abram, T.; Ion, S. Generation-IV nuclear power: A review of the state of the science. *Energy Policy* **2008**, *36*, 4323–4330. [CrossRef]
- 132. Locatelli, G.; Mancini, M.; Todeschini, N. Generation IV nuclear reactors: Current status and future prospects. *Energy Policy* **2013**, *61*, 1503–1520. [CrossRef]
- 133. Magwood, W.D.; Paillere, H. Looking ahead at reactor development. *Prog. Nucl. Energy* 2017, 1–10. [CrossRef]
- 134. Dulera, I.V.; Sinha, R.K. High temperature reactors. J. Nucl. Mater. 2008, 383, 183–188. [CrossRef]
- 135. Sinha, R.K. Advanced nuclear reactor systems—An Indian perspective. *Energy Procedia* 2011, 7, 34–50. [CrossRef]
- 136. Qualls, A.L.; Betzler, B.R.; Brown, N.R.; Carbajo, J.J.; Greenwood, M.S.; Hale, R.; Harrison, T.J.; Powers, J.J.; Robb, K.R.; Terrell, J.; et al. Preconceptual design of a fluoride high temperature salt-cooled engineering demonstration reactor: Motivation and overview. *Ann. Nucl. Energy* 2017, 107, 144–155. [CrossRef]
- 137. Krumwiede, D.; Andreades, C.; Choi, J.K.; Cisneros, A.T.; Huddar, L.; Huff, K.D.; Laufer, M.R.; Munk, M.; Scarlat, R.O.; Seifried, J.E.; et al. Design of the Mark-1 Pebble-Bed, Fluoride-salt-cooled, High-Temperature Reactor Commercial Power Plant. In Proceedings of the ICAPP 2014, Charlotte, NC, USA, 6–9 April 2014.
- 138. Andreades, C.; Cisneros, A.T.; Choi, J.K.; Chong, A.Y.K.; Fratoni, M.; Hong, S.; Huddar, L.R.; Huff, K.D.; Kendrick, J.; Krumwiede, D.L.; et al. Design Summary of the Mark-I Pebble-Bed, Fluoride Salt–Cooled, High-Temperature Reactor Commercial Power Plant. *Nucl. Technol.* 2016, 195, 223–238. [CrossRef]
- Forsberg, C.W.; Hu, L.W.; Peterson, P.F.; Fratoni, M.; Sridharan, K.; Blandford, E. Progress in Development of Fluoride-Salt-Cooled High-Temperature Reactors (FHRs). 2017. Available online: https://www.osti.gov/ scitech/biblio/1183687 (accessed on 12 January 2018).
- Van Rooijen, W.F.G. Gas-Cooled Fast Reactor: A Historical Overview and Future Outlook. *Sci. Technol. Nucl. Install.* 2009, 2009, 965757. [CrossRef]
- 141. Stainsby, R.; Peers, K.; Mitchell, C.; Poette, C.; Mikityuk, K.; Somers, J. Gas cooled fast reactor research and development in the European Union. *Sci. Technol. Nucl. Install.* **2009**, 2009. [CrossRef]
- 142. Stainsby, R.; Peers, K.; Mitchell, C.; Poette, C.; Mikityuk, K.; Somers, J. Gas cooled fast reactor research in Europe. *Nucl. Eng. Des.* **2011**, *241*, 3481–3489. [CrossRef]
- 143. Alemberti, A.; Smirnov, V.; Smith, C.F.; Takahashi, M. Overview of lead-cooled fast reactor activities. *Prog. Nucl. Energy* **2014**, *77*, 300–307. [CrossRef]
- 144. Smith, C.F.; Halsey, W.G.; Brown, N.W.; Sienicki, J.J.; Moisseytsev, A.; Wade, D.C. SSTAR: The US lead-cooled fast reactor (LFR). *J. Nucl. Mater.* 2008, *376*, 255–259. [CrossRef]
- 145. Alemberti, A. The Lead Fast Reactor: An Opportunity for the Future? Engineering 2016, 2, 59–62. [CrossRef]
- 146. Serp, J.; Allibert, M.; Beneš, O.; Delpech, S.; Feynberg, O.; Ghetta, V.; Heuer, D.; Holcomb, D.; Ignatiev, V.; Kloosterman, J.L.; et al. The molten salt reactor (MSR) in generation IV: Overview and perspectives. *Prog. Nucl. Energy* 2014, 77, 308–319. [CrossRef]
- 147. Price, M.S.T. The Dragon Project origins, achievements and legacies. *Nucl. Eng. Des.* **2012**, 251, 60–68. [CrossRef]
- 148. Fang, C.; Min, Q.; Yang, Y.; Sun, Y. Process heat applications of HTR-PM600 in Chinese petrochemical industry: Preliminary study of adaptability and economy. *Ann. Nucl. Energy* **2017**, *110*, 73–78. [CrossRef]
- 149. U.S. NRC. Research & Test Reactors 2017. Available online: https://www.nrc.gov/reactors/non-power.html. (accessed on 12 January 2018).
- 150. World Nuclear Association. Research Reactors 2017. Available online: http://www.world-nuclear.org/ information-library/non-power-nuclear-applications/radioisotopes-research/research-reactors.aspx (accessed on 17 November 2017).
- 151. Elder, R.; Allen, R. Nuclear heat for hydrogen production: Coupling a very high/high temperature reactor to a hydrogen production plant. *Prog. Nucl. Energy* **2009**, *51*, 500–525. [CrossRef]

- 152. Scarlat, R.O.; Cisneros, A.T.; Koutchesfahani, T.; Hong, R.; Peterson, P.F. Preliminary safety analysis of a PBMR supplying process heat to a co-located ethylene production plant. *Nucl. Eng. Des.* 2012, 251, 53–59. [CrossRef]
- 153. Verfondern, K.; Yan, X.; Nishihara, T.; Allelein, H.J. Safety concept of nuclear cogeneration of hydrogen and electricity. *Int. J. Hydrog. Energy* **2017**, *42*, 7551–7559. [CrossRef]
- 154. Sato, H.; Ohashi, H.; Nakagawa, S.; Tachibana, Y.; Kunitomi, K. Safety design consideration for HTGR coupling with hydrogen production plant. *Prog. Nucl. Energy* **2015**, *82*, 46–52. [CrossRef]



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