

Supplementary Materials

Deep Red Photoluminescence from Cr³⁺ in Fluorine-Doped Lithium Aluminate Host Material

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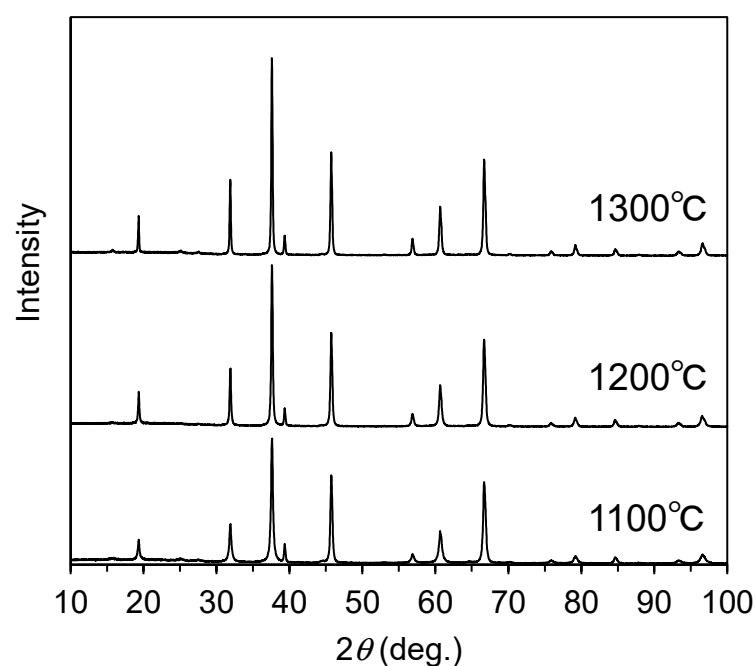


Figure S1 XRD patterns of aluminum lithium fluoride oxide (ALFO) samples prepared at 1100°C, 1200°C, and 1300°C.

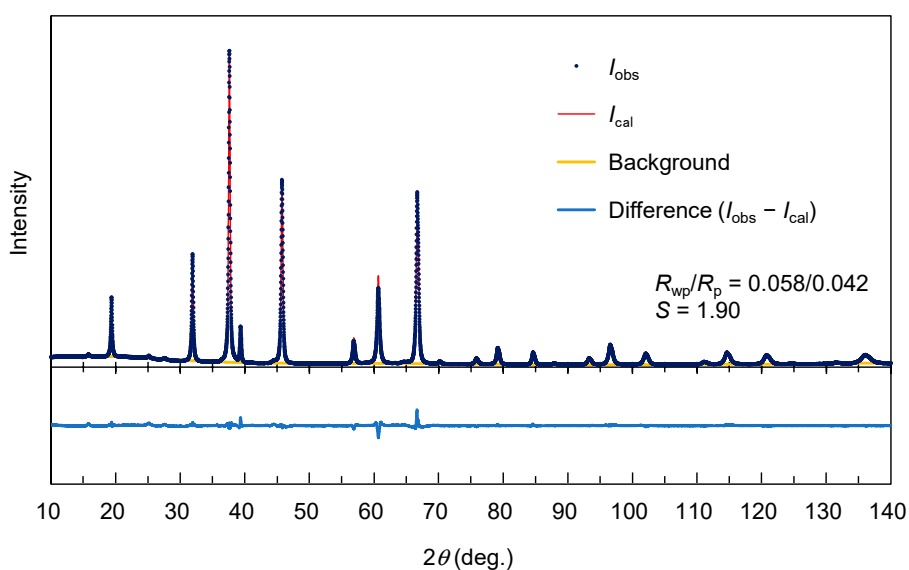


Figure S2 The result of the Rietveld analysis for non-doped ALFO sample ($Al_{4.73}Li_{1.27}F_{0.17}O_{7.65}$) prepared at 1200 °C.

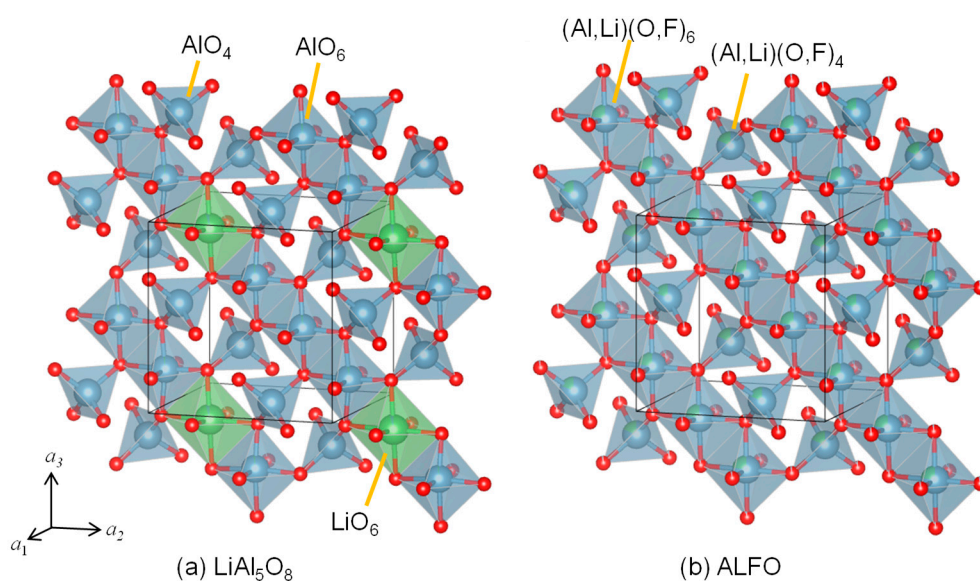


Figure S3 Perspective view of the crystal structures of ordered $LiAl_5O_8$ (a) and aluminum lithium fluoride oxide (ALFO) (b) in the range $x = 0.2$ to 0.6 along $\langle 100 \rangle$, illustrating the difference in the arrangement of the tetrahedra and the octahedra in these host materials.

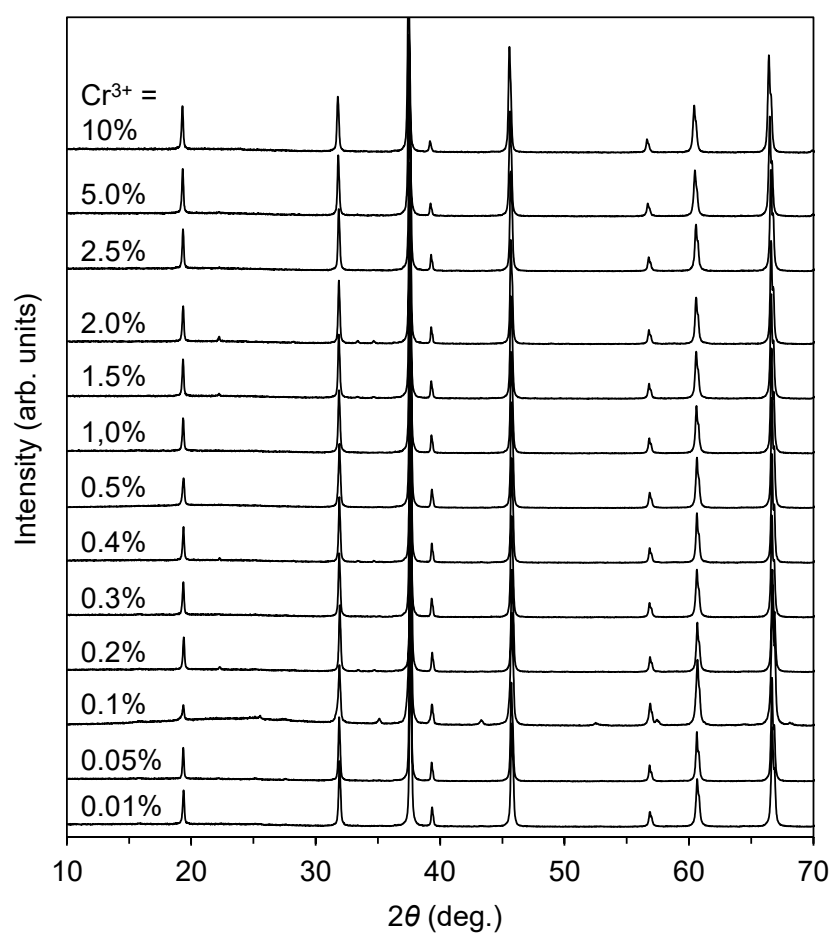


Figure S4 XRD patterns of aluminum lithium fluoride oxide (ALFO) samples prepared at 1200°C with different Cr³⁺ concentrations.

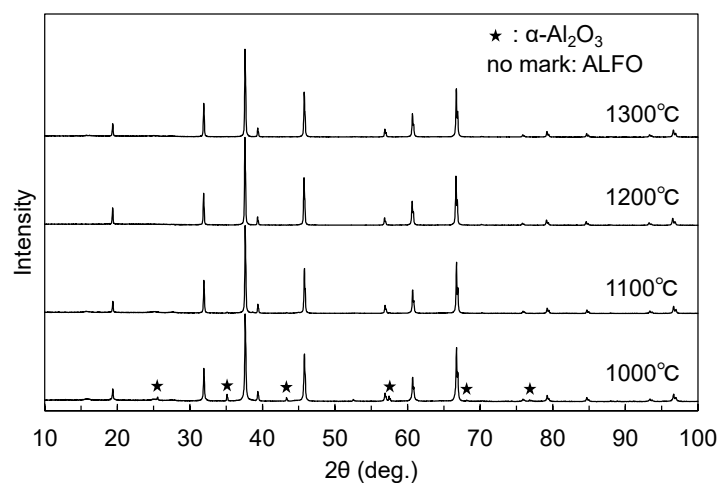


Figure S5 XRD patterns of aluminum lithium fluoride oxide (ALFO) samples prepared at different temperatures.

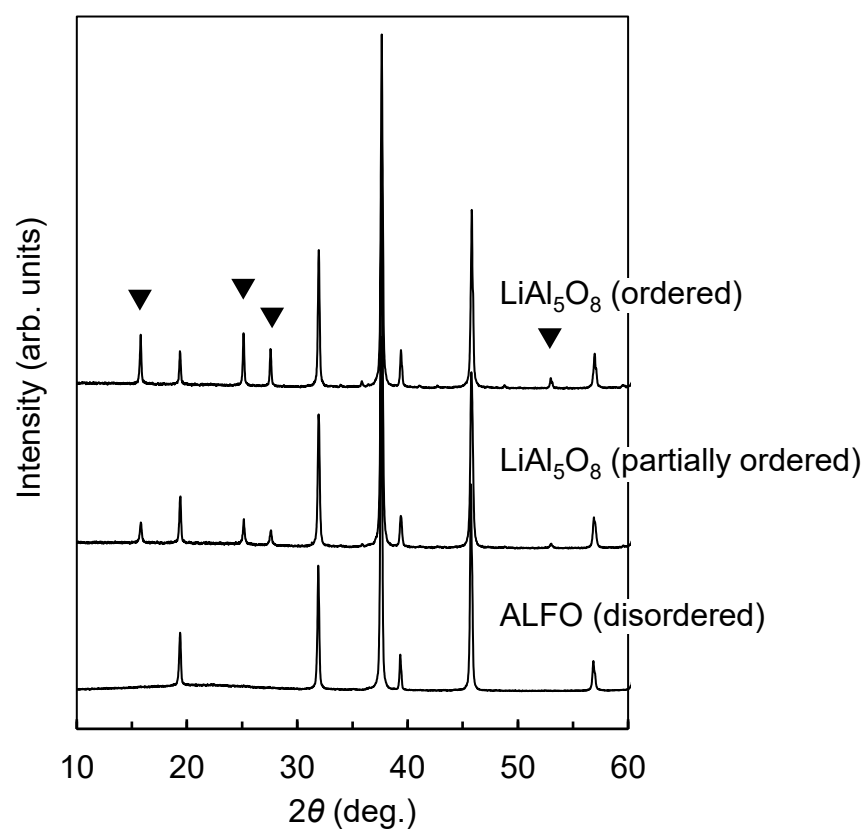


Figure S6 Comparison of XRD patterns of disordered spinel (ALFO), ordered spinel (LiAl₅O₈), and partially ordered LiAl₅O₈ prepared by quenching from 1400°C.

Table S1 The structural parameters refined in the Rietveld analysis for non-doped ALFO sample ($\text{Al}_{4.73}\text{Li}_{1.27}\text{F}_{0.17}\text{O}_{7.65}$) prepared at 1200°C.

Space group		$Fd\bar{3}m$				
Lattice constant		7.9222(8) Å				
$R_{wp} / R_p / S$		0.058 / 0.042 / 1.90				
Atom	Wyckoff position	Occupancy, g	x	y	z	B (Å ²)
Al1	8a	0.817(1)	0	0	0	0.67(1)
Li1	8a	0.183	0	0	0	$= B(\text{Al1})$
		$= 1 - g(\text{Al1})$				
Al2	16d	0.774	5/8	5/8	5/8	0.79(1)
Li2	16d	0.226	5/8	5/8	5/8	$= B(\text{Al2})$
		$= 1 - g(\text{Al2})$				
O	32e	0.9563	0.38139(4)	$= x$	$= x$	0.72(1)
F	32e	0.0213	$= x(\text{O})$	$= x(\text{O})$	$= x(\text{O})$	$= B(\text{O})$

Table S2 Bond angles in CrO_6 octahedra in $\alpha\text{-Al}_2\text{O}_3$ and ordered LiAl_5O_8 after relaxation by MD.

$\alpha\text{-Al}_2\text{O}_3\text{:Cr}^{3+}$		$\text{LiAl}_5\text{O}_8\text{:Cr}^{3+}$	
Bond angles (°)		Bond angle (°)	
76.3	O1-Cr-O2, O2-Cr-O3, O2-Cr-O3	81.3	O2-Cr-O5, O4-Cr-O6
84.8-9	O1-Cr-O4, O2-Cr-O5, O3-Cr-O6	84.4	O1-Cr-O5, O3-Cr-O6
89.5-7	O1-Cr-O5, O2-Cr-O6, O3-Cr-O4	86.5-6	O2-Cr-O6, O4-Cr-O5
105.2-3	O4-Cr-O5, O4-Cr-O6, O5-Cr-O6	90.9-91.4	O1-Cr-O4, O2-Cr-O3, O5-Cr-O6
		100.2-7	O1-Cr-O2, O1-Cr-O3, O3-Cr-O4

Table S3 Assignment of the ESR peaks of the polycrystalline sample of Cr^{3+} in $\alpha\text{-Al}_2\text{O}_3$.

$\alpha\text{-Al}_2\text{O}_3\text{:Cr}^{3+}$		
g value	Transition	Angle (θ)
3.79	$-3/2 \rightarrow -1/2$	90°
2.26	$-3/2 \rightarrow -1/2$	40°
1.72	$-1/2 \rightarrow +1/2$	40°
1.46	$+1/2 \rightarrow +3/2$	90°