

Article

Tribological Properties of WS₂ Thin Films Containing Graphite-like Carbon and Ni Interlayers

Roman I. Romanov ¹, Dmitry V. Fominski ¹, Maxim V. Demin ², Mariya D. Gritskevich ¹, Natalia V. Doroshina ³, Valentin S. Volkov ³  and Vyacheslav Yu. Fominski ^{1,*} 

¹ National Research Nuclear University MEPhI (Moscow Engineering Physics Institute), Kashirskoe sh., 31, Moscow 115409, Russia

² Immanuel Kant Baltic Federal University, A. Nevskogo St 14, Kaliningrad 236016, Russia

³ Center for Photonics and 2D Materials, Moscow Institute of Physics and Technology (MIPT), Dolgoprudny 141701, Russia

* Correspondence: vyfominskij@mephi.ru

Abstract: The development and production of thin-film coatings having very low friction is an urgent problem of materials science. One of the most promising solutions is the fabrication of special nanocomposites containing transition-metal dichalcogenides and various carbon-based nanophases. This study aims to explore the influence of graphite-like carbon (g-C) and Ni interface layers on the tribological properties of thin WS₂ films. Nanocrystalline WS₂ films were created by reactive pulsed laser deposition (PLD) in H₂S at 500 °C. Between the two WS₂ nanolayers, g-C and Ni nanofilms were fabricated by PLD at 700 and 22 °C, respectively. Tribotesting was carried out in a nitrogen-enriched atmosphere by the reciprocal sliding of a steel counterbody under a relatively low load of 1 N. For single-layer WS₂ films, the friction coefficient was ~0.04. The application of g-C films did not noticeably improve the tribological properties of WS₂-based films. However, the application of thin films of g-C and Ni reduced the friction coefficient to 0.013, thus, approaching superlubricity. The island morphology of the Ni nanofilm ensured WS₂ retention and altered the contact area between the counterbody and the film surface. The catalytic properties of nickel facilitated the introduction of S and H atoms into g-C. The sliding of WS₂ nanoplates against an amorphous g-C(S, H) nanolayer caused a lower coefficient of friction than the relative sliding of WS₂ nanoplates. The detected behavior of the prepared thin films suggests a new strategy of designing antifriction coatings for practical applications and highlights the ample opportunities of laser techniques in the formation of promising thin-film coatings.

Keywords: ultralow friction; wear; nanolayered coatings; tribochemistry; WS₂ thin films; graphitic-like thin films; pulsed laser deposition



Citation: Romanov, R.I.; Fominski, D.V.; Demin, M.V.; Gritskevich, M.D.; Doroshina, N.V.; Volkov, V.S.; Fominski, V.Y. Tribological Properties of WS₂ Thin Films Containing Graphite-like Carbon and Ni Interlayers. *Materials* **2023**, *16*, 282. <https://doi.org/10.3390/ma16010282>

Academic Editors: Hui Sun and Jing Xu

Received: 20 November 2022

Revised: 23 December 2022

Accepted: 27 December 2022

Published: 28 December 2022



Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (<https://creativecommons.org/licenses/by/4.0/>).

1. Introduction

Scientific interest in thin-film coatings of pure transition-metal dichalcogenides (TMD) has been growing actively since the end of the last century. This holds especially for Mo/W/S₂/Se₂, which have good antifriction properties when liquid lubricants are not applicable [1–4]. This interest has been drummed up by the rapid development of aerospace technology, vacuum technology, micro- and nano-electromechanical systems (MEMS, NEMS), and the search for alternatives to liquid lubricants used in weapons and metal machining—the latter inspired by environmental concerns [5–8]. Comprehensive studies of the morphology, structure, and chemical composition have revealed the main advantages and disadvantages of TMD coatings formed by ion-plasma and pulsed laser deposition (PLD) [9–12]. New avenues for modifying the composition and structure of such coatings are developing to overcome deficiencies such as low wear resistance, especially at high contact loads. TMD coatings have been doped with metal atoms (Ni, Ti, and others), as well as with carbon and nitrogen atoms [13–17]. Another approach to improving the tribological

properties of TMD-based coatings is to create multilayer coatings in which TMD layers are alternated with layers of other TMD materials or carbon [18–21].

The latter strand of research has gained considerable momentum since the superlubricity effect is achieved in coatings formed from TMD and carbon nanolayers. When this effect is manifested, the friction coefficient does not exceed 0.01. The mechanisms behind this phenomenon, however, may differ. For this effect to occur, special interactions between the contacting surfaces are required at the atomic level. Incommensurate contact in the tribopair at an atomic level is considered a necessary condition for such an effect. The incommensurability can be achieved at the contact of surfaces with both ordered (crystalline) atomic packing and disordered local packing [22–24]. These conditions require an optimal structural state on the surface of the coating (in frictional interfaces) and the counterbody at the stage of friction pair preparation. A more promising and practical approach to achieving the superlubricity effect, which creates prerequisites for material self-organization in the contact area between the coating surface and the counterbody or between the coating layers. Triboinduced/tribochemical processes can initiate the formation of new states/phases in the contact area and, thus, reduce friction significantly. Tribochemical interaction between TMD and C-based components in the tribofilm can cause structural/chemical changes in both components. These alternations manifest themselves in the formation of new nanoparticles of different morphology, which change both the contact area in the friction pair and the friction mechanism [25–28].

The success of “constructing” new nanolayers depends on scientific advances in the understanding of triboadaptation mechanisms and the development of techniques for the formation of nanolayers based on TMD and carbon components. A combination of plasma-enhanced chemical vapor deposition (PECVD) and physical vapor deposition (PVD, ion-sputtering/magnetron deposition) techniques are commonly used to obtain such nanomaterials [18,29–31]. Sometimes, MoS₂ flakes are deposited by drop-casting, ensuring the deposition of ultra-thin (2D/quasi-2D) MoS₂ layers with the most perfect basal plane packing [32]. The PVD method helps to obtain TMD films whose nanocrystalline structure is usually amorphous or highly disordered. In this case, the conditions for ultralow friction are achieved under sufficiently high counterbody loads. The structural/chemical modification of the TMD layer is possible only under these conditions. Additionally, this also applies to the low shear force and incommensurability contact between adjacent TMD layers [29,31,32].

We investigated the formation of a nanolayer structure containing ultra-thin WS₂ films, graphite-like carbon, and nickel, using pulsed laser deposition (PLD), including reactive PLD (RPLD). Studies in the field have mostly focused on MoS₂-based films, whereas the application of WS₂-based ultra-thin films to achieve superior lubrication with hierarchical/multilayer structures have not been treated in much detail. The well-known WS₂ and MoS₂ solid lubricants are extensively used materials today since they both have the most pronounced friction and wear reduced effect. The WS₂ and MoS₂ materials have both close-packed hexagonal laminar structure, and their crystal lattices contains thin layer units, which are formed by three plane layers of S, W, and S and S, Mo, and S, respectively. During friction, slippage develops easily along the closed packed plane. It is considered that MoS₂-based coatings are soft; their antifriction properties are very good, but they have poorer corrosion/wear resistance, while WS₂-based coatings are hard, and their corrosion/wear resistance and thermal stability are better. Formed during friction, WO₃ is slightly more protective and provides a lower friction coefficient than MoO₃ [19,33–37]. However, the generalizing characteristics are not universal, and deviations have been found in some experimental studies. Watanabe et al. [38,39] revealed that the friction-reduced properties of WS₂ coating might be better than those of MoS₂ coating.

Laser-based techniques open broad possibilities for the formation of ultra-thin TMD films with an adjustable number of basal planes and sufficiently perfect atomic packing [40–42]. The utility of PLD as a technique of obtaining graphite-like carbon films and metal films is well known [43,44]. However, interfacial solid-phase reactions can occur during the formation

of multilayer films at high temperatures (up to 700 °C) ensuring the required characteristics of individual nanolayers. Further work is required to explore this phenomenon. A nickel film was used to initiate graphene-like phase formation in a carbon film [45,46]. We chose graphite-like carbon interlayers, which were obtained by vacuum PLD at an elevated temperature in the substrate. The deposition of carbon laser plasma at this condition provided sp^2 -bond formation in the g-C film, as well as the formation of W–C or Ni–C chemical bonds at the corresponding interfaces during the deposition of WS_2 /g-C/ WS_2 or WS_2 /g-C/Ni/ WS_2 thin-film coatings. It was also assumed that the local layered structure of g-C films can facilitate the penetration of heteroatoms from neighbor layers resulting in graphene-like structure formation during triboinduced processes.

Using RPLD/PLD, nanometer-thick films were created, consisting of one WS_2 layer, three WS_2 /g-C/ WS_2 layers, and four WS_2 /g-C/Ni/ WS_2 layers. Tribotests were performed in an N_2 -enriched atmosphere, with a relatively low counterbody load. The superlubricity effect can be expected as a result of WS_2 nanocrystals with basal atomic planes sliding along the surface of a g-C film modified by introducing heteroatoms. The retention of WS_2 solid lubricating material on the surface of a carbon nanolayer can be achieved by the optimal disturbance of the smooth surface of the carbon layer.

2. Materials and Methods

Layered thin-film coatings were created on polished silicon substrates in a single vacuum cycle. An Nd:YAG laser (LQ529, Solar LS, Minsk, Belarus) with a laser wavelength of 1064 nm, pulse duration of 15 ns, pulse energy of 40 mJ, and pulse repetition frequency of 20 Hz was used. The vacuum chamber was evacuated by a turbomolecular pump to a residual pressure of 10^{-3} Pa. WS_2 nanometer-thick films were grown by RPLD. Using this technique, a WO_3 target was irradiated in a reaction gas atmosphere (H_2S) at 40 Pa. At the same time, the Si substrate was heated to a temperature of 500 °C. The deposition time was 30 s. During the deposition of nickel and carbon layers, targets of nickel foil and glass carbon plates were laser-irradiated in vacuum. The deposition time was 2 min and 1 min, respectively. The Ni layer was deposited at room temperature of the substrate; the g-C layer, at 700 °C. The thickness of all layers was 20–30 nm. In special cases (e.g., in MRS studies), the film thickness was increased to exclude the influence of the Si substrate signal on the measurements. In the following text, abbreviations will be used to refer to the coatings: WS_2 (C) for the three-layer WS_2 /g-C/ WS_2 ; WS_2 (C, Ni) for the four-layer WS_2 /g-C/Ni/ WS_2 (Figure 1). A scheme of fabrication of multilayer thin-film coatings using the laser technique is shown in Figure S1.

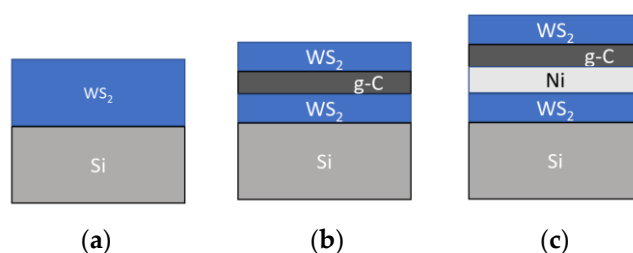


Figure 1. Schematic illustrations of thin-film coatings with different structures: (a) WS_2 , (b) WS_2 (C), and (c) WS_2 (C, Ni).

The morphology of sample surfaces and sample composition were studied by scanning electron microscopy (SEM) and energy-dispersive X-ray spectroscopy (EDS) using a Tescan LYRA3 device (Brno, Czech Republic). Structural analysis was carried out by X-ray diffraction (XRD) and micro-Raman spectroscopy (MRS). XRD measurements were carried out on an ARL X'tra diffractometer equipped with a parabolic mirror ($Cu-K\alpha$ radiation was used). Measurements were performed in the Θ – 2Θ geometry. Micro-Raman spectroscopy (MRS) provided insights into the detailed film structure. A LabRAM Evolution (Horiba Scientific, Kyoto, Japan) instrument with a 532 nm laser source with a 1 cm^{-1} spectral

resolution was used to perform spectral measurements; the laser spot diameter was 0.45 μm . Laser intensity was kept under 0.2 mW to prevent laser-induced artefacts. Measurements were performed in a temperature-controlled room at standard conditions. The 520 cm^{-1} phonon mode from the silicon wafer was used for calibration.

The chemical state and composition of the films were analyzed by X-ray photoelectron spectroscopy (XPS), using a Theta Probe spectrometer under high-vacuum conditions (base pressure $< 2 \times 10^{-9}$ mbar) with a monochromatic Al-K α X-ray source (1486.6 eV). Photoelectron spectra were acquired using the fixed analyzer transmission mode at 50 eV pass energy. For elemental composition XPS analysis, Scofield's factors were employed in the calculations.

The tribological testing of thin-film coatings was carried out with the help of an Anton Paar TRB3 tribometer in the reciprocating motion mode, using a steel ball (100Cr6) with a diameter of 6 mm as a counterbody. The load on the ball was 1 N, and the Hertzian contact stress was ~ 660 MPa. The average speed of the ball over a coated substrate was 1 cm/s. The length of the wear track was 5 mm. The tests were carried out at a reduced atmospheric humidity (relative humidity RH $\sim 8\%$), which was achieved by pumping N_2 gas through the testing chamber. The sample temperature was 22 $^\circ\text{C}$. The wear tracks and debris were studied by MRS, SEM, and optical microscopy.

3. Results

3.1. The Surface Morphology and Structure of the Films

RPLD/PLD made it possible to form sufficiently smooth WS_2 and carbon film on an area necessary for studying their tribological properties. Figure 2a shows the SEM image of a $\text{WS}_2(\text{C})$ film before the tribotest. Rare round submicron particles were visible on the smooth surface of the film. These could be WO_x and carbon particles formed during the laser ablation of WO_3 and glassy carbon target, respectively. Once within the track, such particles could cause abrasive wear and increase the coefficient of friction. The results of three friction coefficient measurements were used to analyze the tribological properties of the coatings.

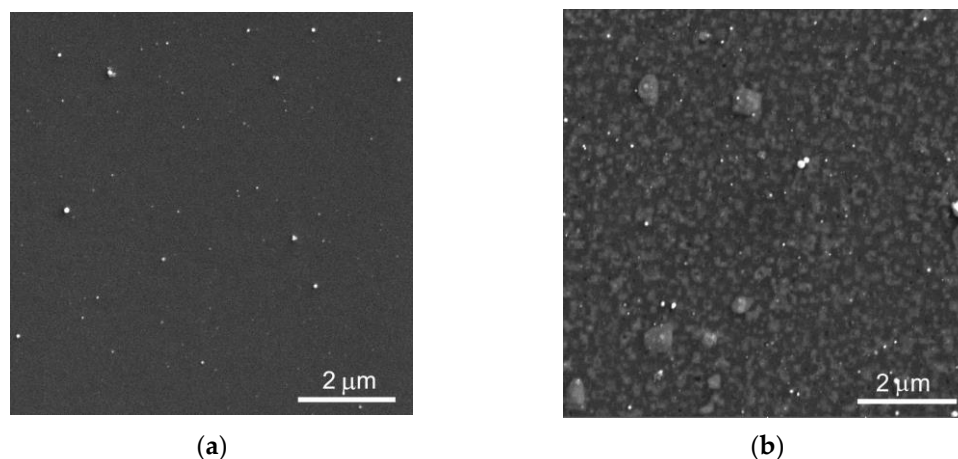


Figure 2. SEM surface images of (a) $\text{WS}_2(\text{C})$ and (b) $\text{WS}_2(\text{C, Ni})$ thin-film coatings on a Si substrate. The inhomogeneity of the Ni-containing film caused by the deposition of Ni droplets and the formation of Ni clusters on the surface of the WS_2 underlayer (see Figure S2).

Figure 2b shows that Ni film deposition is associated with marked changes in the surface morphology of the $\text{WS}_2(\text{C, Ni})$ coating. These alterations were due to an increase in the surface concentration of micron-sized Ni particles increased during nickel deposition and the formation of a network of submicron-sized Ni clusters. Micron-sized particles were formed during the ablation of the Ni target and transported by the plasma plume to the coating surface. The surface concentration of microparticles was low, and of submicron

particles, rather high. This difference was due to the particle formation mechanism. Coalescence and the growth of an island structure in the Ni layer were likely to occur during the deposition of the nickel atom flow onto the WS₂ layer. These processes might have taken place when the bilayer Ni/WS₂ film was heated before the carbon film deposition.

The data shown in Figures 3, 4 and 5a shed light on the characteristics of the single-layer WS₂ film formed by RPLD on a Si substrate. The XRD pattern in Figure 3 shows not only the peaks of the silicon substrate but also an intense peak of the WS₂ at 14°. This peak corresponds to basal planes (002) for the 2H-WS₂ phase. This means that these basal planes mostly lay parallel to the substrate surface.

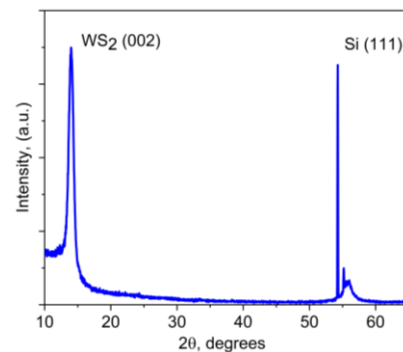


Figure 3. XRD pattern for the WS₂ thin film obtained by RPLD on a Si substrate at 500 °C.

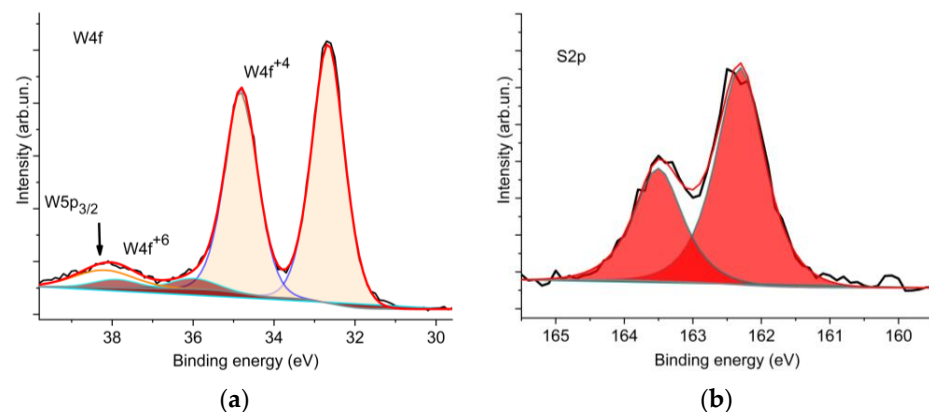


Figure 4. XPS spectra of (a) W4f and (b) S2p for the WS₂ thin film obtained by RPLD at 500 °C.

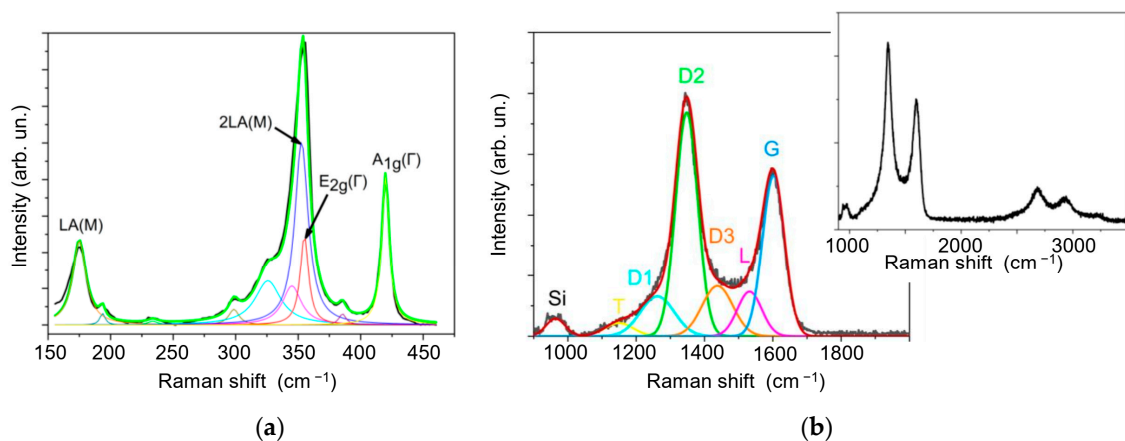


Figure 5. The MRS spectra of (a) WS₂ and (b) g-C thin films obtained at 500 °C and 750 °C, respectively. The inset in (b) shows the MRS spectrum for a g-C film with second-order peaks typical of graphite-like carbon.

XPS spectra were analyzed to determine the chemical states of the WS₂ films (Figure 4). The W4f spectrum was decomposed into two doublets. The principal doublet with W4f_{5/2} at 32.7 eV and a spin-orbit splitting of 2.1 eV corresponds to the state of W⁴⁺ in WS₂. The weaker doublet with W4f_{7/2} at 36.0 eV and the same spin-orbit splitting value corresponds to the state of W⁶⁺ in WO₃. In the compound, WO₃ accounted for 5% at most. The S2p spectrum shows a doublet with S2p_{3/2} at 162.3 eV and a spin-orbit splitting of 1.3 eV, which corresponds to sulfur in WS₂. As the figure shows, the selected RPLD mode ensures the effective sulfurization of tungsten and the removal of oxygen upon interaction between the deposited thin WO_x layer and hydrogen sulfide. The presence of a low-intensity peak of tungsten oxide in the XPS spectrum in Figure 4 is probably due to the introduction into the film of microparticles formed during the laser ablation of a WO₃ target. The microparticles were not sulfurized completely.

Figure 5a shows the MRS spectra for a relatively thick WS₂ film obtained by RPLD on a silicon substrate. The wavelength, 532 nm, for WS₂ corresponds to the resonance of the exciton peak, C. Under these conditions, the obtained spectra contained first-order peaks (denoted as E_{2g}¹ (Γ) and A_{1g} (Γ)) corresponding to oscillatory modes inside the S-W-S layer in the parallel and perpendicular directions; overtones and combination peaks were also present. The most intense peak, with the center at about 352 cm^{−1}, corresponds to the 2 LA(M) mode. There is also a peak at 175 cm^{−1}, corresponding to the LA(M) mode. This first-order mode correlates with the acoustic phonon at the M point of the Brillouin zone. The relative intensity of this peak is used to assess the severity of defects in materials [47,48]. For microcrystalline WS₂ films, the common type of defect is crystalline domain boundaries. Therefore, the relative intensity of the LA(M) peak correlates with the size of the domains, quite in accordance with earlier reports. In general, the spectrum of the film is typical of structures containing sub-μm grains [49,50]. The decomposition of the spectra was performed using the Lorentz function. The peaks were identified based on data from the literature.

Figure 5b shows the MRS spectrum for the g-C/Ni sample consisting of a g-C film deposited by PLD on a Si substrate with a Ni layer. This spectrum is characteristic of microcrystalline graphite [51]. The first order in the spectrum (1000–1800 cm^{−1}) contains two well-defined peaks centered at about 1350 cm^{−1} and 1590 cm^{−1}. In the literature, these peaks are referred to as the D and G lines. The G line corresponds to the doubly degenerate E_{2g} phonon mode at the center of the Brillouin zone. The position of this line is sensitive to changes in the length and direction of interatomic bonds in the grid of sp²-hybridized states. The position and FWHM of MRS peaks are usually taken as a measure of the disordering of the carbon network, present as distorted hexagonal rings and chains. The more defects in the structure, the smaller the Raman shift of the G line and the greater its width [52]. The D line appears in the presence of bond breaks. It corresponds to the A_{1g} phonon mode at the K point of the Brillouin zone. These lines can be used to estimate the size of crystal domains and the defectiveness of carbon films. The lateral size of the domains was estimated using the formula [53]: $L_a = C(\lambda)I(G)/I(D)$, where $C(\lambda)$ is the parameter determined by the wavelength of exciting laser radiation and $I(D)$ and $I(G)$ are the intensities of the D and G lines, respectively. The calculated L_a value was ~15 nm.

The spectrum was decomposed into Gaussian functions. The line centered at approximately 1150 cm^{−1} corresponds to trans-poly-acetylene chains [54]. The lines centered at about 1250 cm^{−1} (D1), 1350 cm^{−1} (D2), and 1440 cm^{−1} (D3) correspond to correlate with the breathing modes of benzene rings containing five, six, and seven atoms [55]. The line at 1530 cm^{−1} (L) is usually associated with vibrations in disordered carbon structures containing a certain number of sp³-hybridized states. It may be interpreted as a G line in amorphous diamond-like films [56,57]. The peak centered at 1590 cm^{−1} represents the G line for graphite.

As the WS₂ film was deposited on the surface of the g-C film, the latter was exposed to laser plasma from a WO₃ target, interacting chemically with the reaction gas H₂S at a rather high temperature. This could change the chemical state of the interface between WS₂

and g-C. Figure 6 shows the MRS spectra of the g-C/WS₂ and WS₂/g-C/WS₂ samples (i.e., WS₂(C)). The relative intensities of the D1, D3, and L lines increase in the decompositions of these spectra compared to the spectrum of the g-C/Ni sample (Figure 5b). The intensity of the peak at 2930 cm⁻¹, which corresponds to the D + G mode, grew in the second order of the carbon spectrum. The ratio of the total intensities of the D1, D2, and D3 peaks to the G peak intensity (I(D)/I(G)) changed slightly (from 1.95 to 2.01 for all the samples). The FWHM of the G peak for the g-C/WS₂ remained almost unchanged at 40 cm⁻¹. Thus, it can be assumed that the size of crystallites does not change significantly, and the observed alterations in the spectra are due to the formation of lattice defects representing five- and seven-atom rings and amorphized regions.

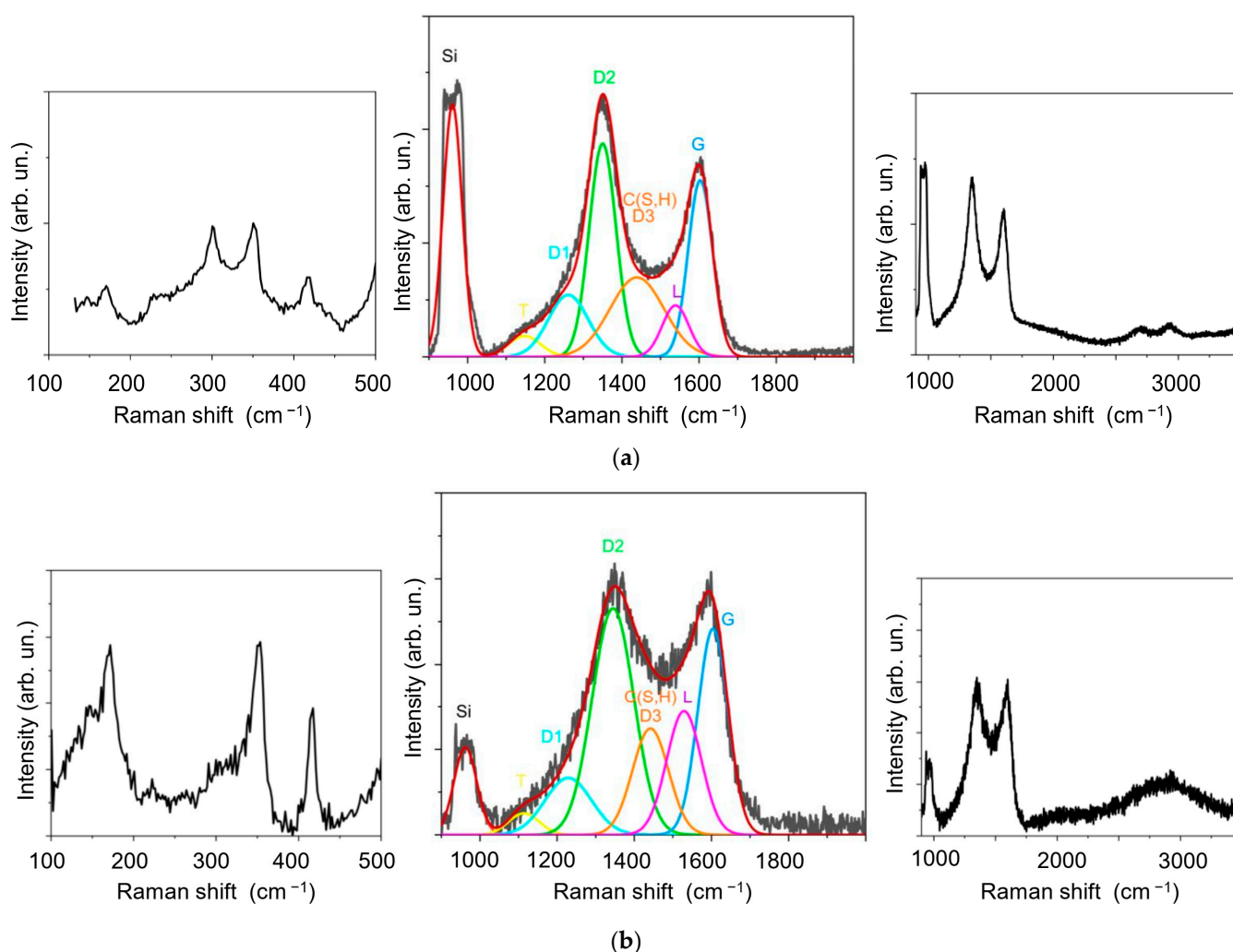


Figure 6. MRS spectra of g-C films for (a) g-C/WS₂/Si and (b) WS₂/g-C/WS₂/Si samples. Insets show the MRS spectra for the WS₂ (left) and g-C film (right), including second-order peaks.

These differences can be explained by the different influences of the underlayer. In the first case, it is a catalytic metal; in the second, WS₂. Another possible cause is variations in the thickness of the g-C layer. It should be noted that the MRS peak at 1440 cm⁻¹ may be of a different nature, namely, due to the formation of C(S, H) species following the introduction of sulfur and/or hydrogen into amorphous carbon. This was established by the authors of this study in earlier MRS studies of carbon films obtained by RPLD in hydrogen sulfide [58].

In the spectra decomposition for the g-C film in the WS₂(C, Ni) sample, the intensity of lines D1, D3, and L grew even further (see Figure S3a). For instance, the ratio I(D3)/I(D) reached 0.32 compared to 0.15 for the g-C/Ni sample. The FWHM(G) value was as high as

60 cm^{-1} . Another distinctive feature of the sample spectrum was the shift of the D3/C(H, S) peak from 1440 cm^{-1} to 1470 cm^{-1} . The studies performed pointed out that the selected modes of forming a multilayer thin-film structure did not cause significant changes at the interfaces of the deposited nanolayers (e.g., Figure S3b). Yet, it can be assumed that the formation of the $\text{WS}_2(\text{C}, \text{Ni})$ thin film coating was accompanied by modifications in the g-C layer under the influence of WS_2 film deposition. This phenomenon manifested itself in changes in defectiveness and the possible introduction of S and H atoms into the surface of the g-C layer.

3.2. The Tribological Performance of the Obtained Films

Figure 7 shows the results of changes in the coefficient of friction when testing thin-film WS_2 , $\text{WS}_2(\text{C})$, and $\text{WS}_2(\text{C}, \text{Ni})$ coatings on a silicon substrate. The single-layer WS_2 nanocoating withstood up to 200 cycles, and the coefficient of friction was approximately 0.04. The carbon interlayer had no significant effect on either the coefficient of friction or the durability of the $\text{WS}_2(\text{C})$ coating. However, the sliding was accompanied by marked changes in the coefficient of friction. For the $\text{WS}_2(\text{C}, \text{Ni})$ coating, the sliding of the counterbody occurred with minimum fluctuations in the coefficient, which decreased to 0.02 after running-in over several cycles and to 0.013 after 20 cycles. A rapid increase in the coefficient of friction was observed after ~ 130 cycles.

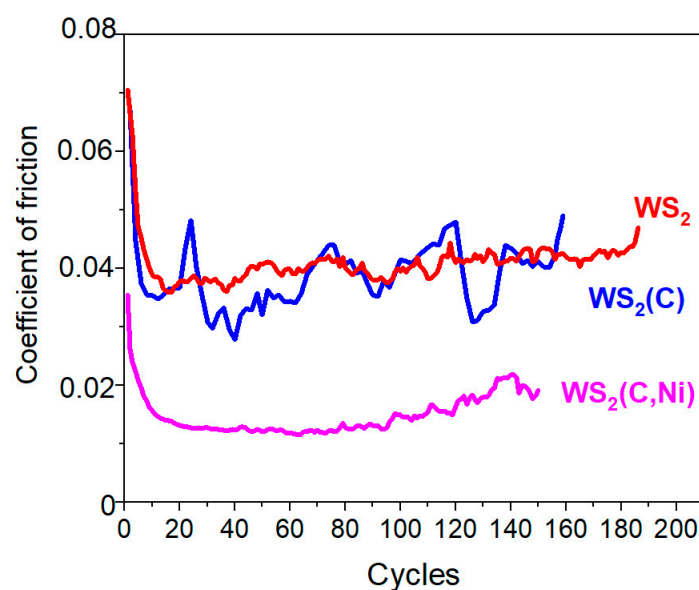


Figure 7. The characteristic evolution of the friction coefficient as a function of the number of cycles for different WS_2 -based thin-film coatings. For thin films, the tests were stopped as signs of a substantial increase in the coefficient of friction appeared.

The coating composition had a significant effect on the wear pattern of the coating and the counterbody (Figure 8). When the WS_2 coating was worn, wear debris effectively adhered as nanoplatelets to the coating surface at the counterbody reversal points. The wear of the counterbody was minimal. The WS_2 coating wear plates barely adhered to the counterbody. The depth of the wear track was $\sim 12\text{ nm}$, and its width did not exceed 80 nm . The wear of the $\text{WS}_2(\text{C})$ coating was a result of the effective adhesion of the wear debris on the surface of the counterbody. The depth of the wear track exceeded 17 nm , and the maximum width was 90 nm . During wear, the $\text{WS}_2(\text{C}, \text{Ni})$ coating had the least amount of wear products, which adhered mainly to the counterbody. At the same time, the wear of the counterbody was most noticeable, which caused the wear track to widen to 90 nm . The depth of the track was approximately 14 nm .

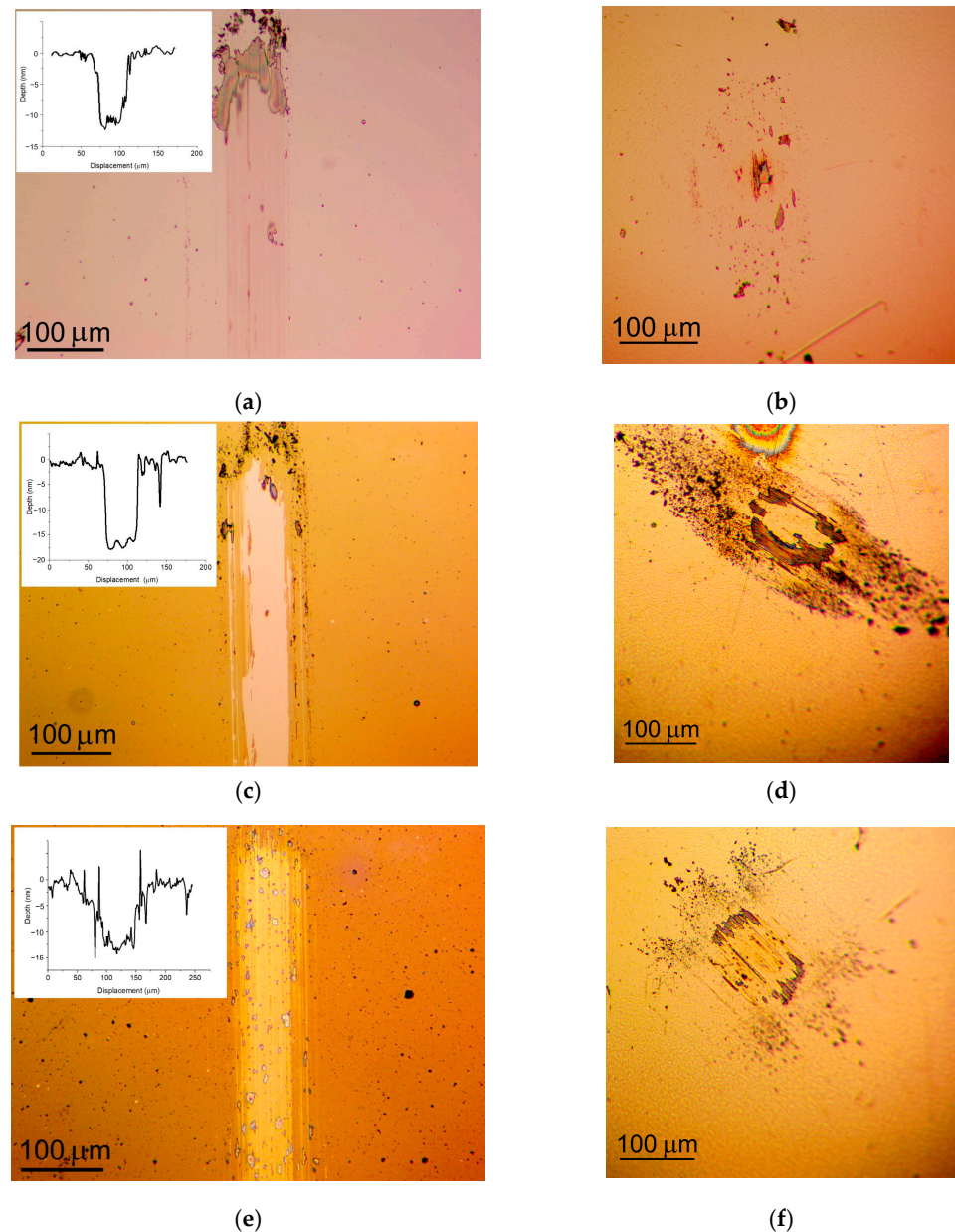


Figure 8. Optical images of the wear tracks (**left**) and the wear scars (**right**) on the coatings and steel balls, respectively: (a,b) WS_2 , (c,d) $\text{WS}_2(\text{C})$, and (e,f) $\text{WS}_2(\text{C}, \text{Ni})$ thin-film coatings on a Si substrate. Figure 7 indicates test durations.

3.3. Analysis of the Friction and Tribomodification of the Films

MRS spectra were measured inside the tracks at different points to analyze triboinduced changes in the coating structure. Wear debris located both inside the tracks and at the ends of the tracks (at the counterbody reversal points) was also examined. Figure 9 shows the results obtained for the thin-film WS_2 coating. Although there was no coating in the center of the WS_2 track, the coefficient of friction remained low. This could be due to a very thin tribofilm preserved in the track. The same film could remain on the surface of the counterbody. The MRS spectra for the wear debris localized both in the track and on its boundaries coincided with the spectra of the original film, i.e., had the structure characteristic of the crystalline WS_2 phase. Thus, the MRS study showed that triboexposure does not cause appreciable changes in the structure of WS_2 nanocoatings. When the counterbody was exposed, the WS_2 nanolayers were removed layer-by-layer, accumulating mainly at the counterbody reversal points.

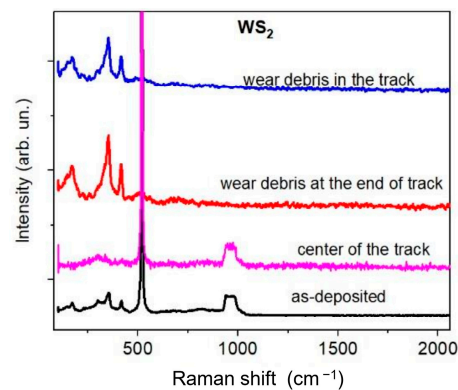


Figure 9. MRS spectra of the WS_2 thin-film coating measured before and after tribotesting.

An MRS study of the $WS_2(C)$ thin-film coatings showed that the wear particles consisted predominantly of carbon: the peaks in the low-frequency region corresponding to WS_2 had very low intensities (Figure 10). These peaks could be due to scattering on the WS_2 film, which deposited on the silicon and remained under the wear debris. The intensity of the carbon spectrum, however, increased compared to the spectra of the original films. In the decomposition of the wear debris spectrum for the $WS_2(C)$ nanocoating, the relative intensities of lines D1, D3, and L lines diminished compared to the spectrum of the original sample. The $I(D3)/I(D)$ ratio plummeted to 0.2, and the $I(D)/I(G)$ ratio dipped from 1.95 to 1.85. The $FWHM(G)$ value decreased to 40 cm^{-1} . This led us to assume that tribotesting caused the dispersion of WS_2 nanoplates in the wear debris of the carbon layer, which underwent only weak structural changes upon contact with the counterbody.

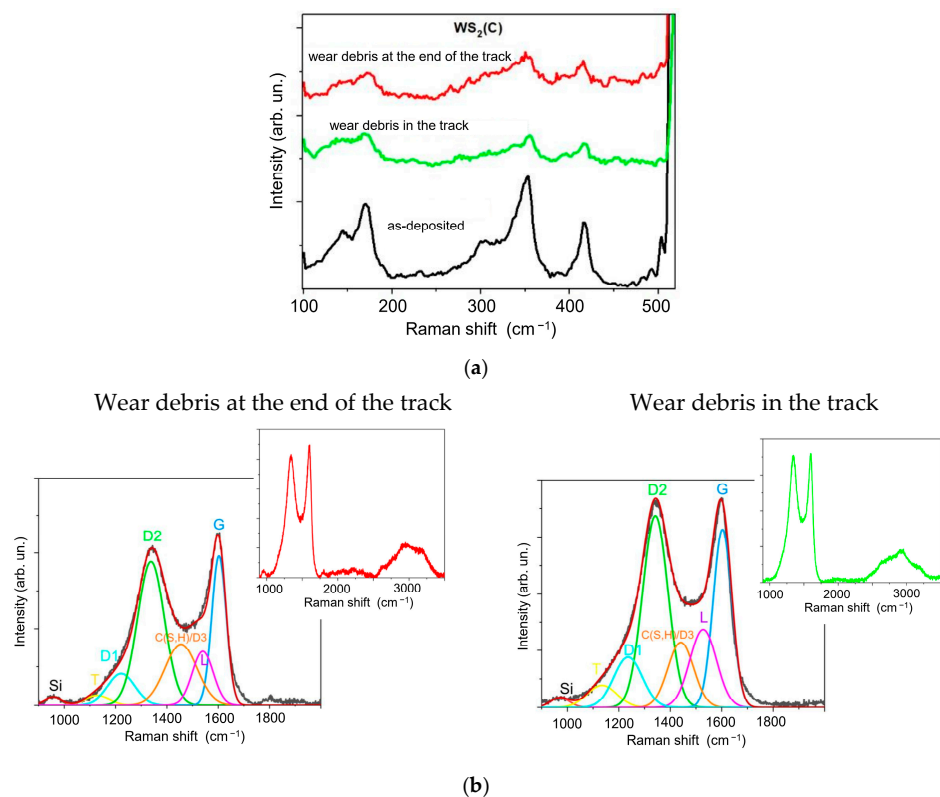


Figure 10. MRS spectra of the $WS_2(C)$ thin-film coating measured after tribotesting in (a) low- and (b) high-frequency intervals. The spectrum of as-deposited $WS_2(C)$ film is presented for comparison to illustrate the removal of WS_2 film. The inserts in (b) show MRS spectra, which include second-order peaks. Spectra were measured for wear debris located at the end of the track and inside the track.

SEM and EDS studies confirm that the sliding of the counterbody when testing the WS₂(C)/Si sample caused wear to spread across the carbon layer, which was almost completely removed towards the end of the test (Figure 11). Therefore, a wear mechanism is possible where the mechanical mixing of WS₂ and g-C nanoparticles takes place. The dispersion of WS₂ nanoparticles in the tribofilm was responsible for the relatively low coefficient of friction against the WS₂(C) coating, which is comparable to the coefficient of a single-layer WS₂ coating.

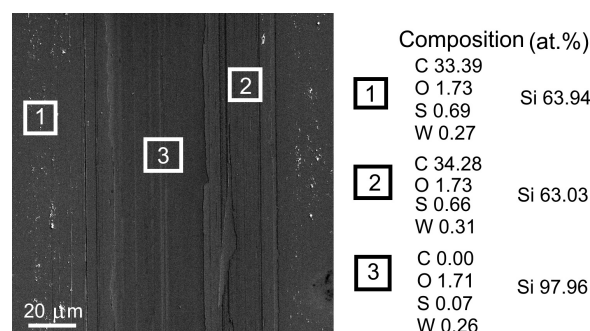


Figure 11. SEM image of the wear track formed after testing on the surface of the WS₂(C)/Si sample. The composition was measured by EDS. Figure S4 shows the corresponding EDS spectra; Figure S5, the lateral distributions of elements across the wear track.

Figure 12 shows the MRS spectra measured for the WS₂(C, Ni)/Si sample after tribotesting. Intense peaks corresponding to tungsten disulfide can be observed in the low-frequency region of the Raman spectra of the WS₂(C, Ni) sample's wear debris. The shape and relative intensity of these peaks are almost the same as in the spectra of the original films. For wear debris, the difference is that the position of the A_{1g} peak is shifted towards a decrease in the wave number by 1.7 cm^{−1}. For ultrathin WS₂ films, the shift between the E_{2g} and A_{1g} peaks is sensitive to the number of monolayers [59,60]. In this case, this shift changed from 65.3 cm^{−1} in the original film to 63.6 cm^{−1} in the wear debris. This difference corresponds to a decrease in the number of monolayers from about 10 to 5 [60]. Consequently, WS₂ crystallites are ground into thinner flakes during the tribomodification.

For the WS₂(C, Ni) sample, the carbon spectra of the wear particles in the track were slightly different from those of the original coating. Neither the I(D3)/I(D) nor the I(D)/I(G) ratio changed, remaining at 0.25 and 1.95, respectively. The FWHM(G) value (54 cm^{−1}) did not alter either. The most significant changes were the shift of line D3 by 1440 cm^{−1} and a decrease in the I(D1)/I(D) ratio from 1.8 for the original sample to 1.1. Even more noticeable were the variations in the spectrum of the wear debris accumulated at the end of the wear track. The main change was the increase in the relative intensity of the peaks at ~1440 and ~2900 cm^{−1}. According to the literature, the introduction of sulfur into exfoliated graphene and carbon nanotubes caused amorphization [61,62]. The annealing of the resulting nonequilibrium structures leads to the appearance of narrower bands with centers at approximately 1250 and 1440 cm^{−1}. According to [63], the annealing of graphene in sulfur vapor results in the appearance of lines with centers at 1436 and 1530 cm^{−1} in the Raman spectra. To confirm this, Raman spectra were measured using a higher intensity of exciting laser radiation (2 mW). The partial annealing of the WS₂(C, Ni) films took place under these conditions. As a result, peaks with centers at approximately 1440 and 2890 cm^{−1} became clearly visible in the spectra (Figure S6). In addition, the increase in the relative intensity of line L may also be due to the formation of C-H bonds [64]. Hydrogen could be introduced into coatings during RPLD in hydrogen sulfide. Comparing the results of the MRS studies of triboinduced changes in WS₂(C) and WS₂(Ni, C) thin-film coatings suggests that the introduction of sulfur/hydrogen in the g-C layer is largely due to the catalytic effect of nickel.

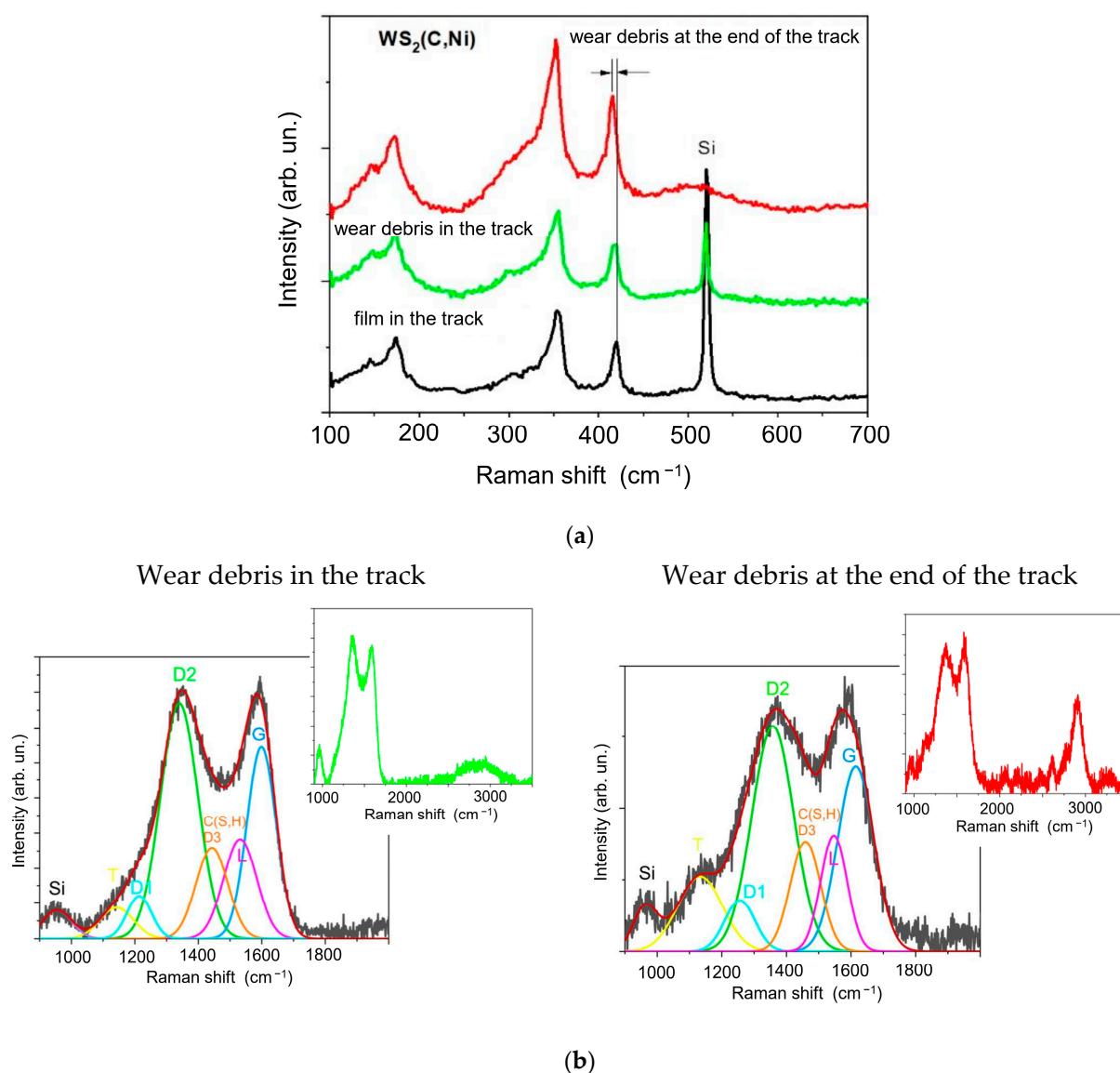


Figure 12. MRS spectra of the WS₂(C, Ni)/Si sample after tribotesting in different regions of the wear track in (a) low- and (b) high-frequency intervals. MRS Si peak is shown for controlling the thickness of thin-film coating. The inserts in (b) show the MRS spectra, which include second-order peaks.

SEM and EDS studies of WS₂(C, Ni) coatings on a Si substrate showed that tribointeraction disturbed the uniform distribution of WS₂ in the track (Figures 13 and 14). Figure S6 shows the EDS spectra for the selected area of this sample. Micro-regions were formed in the track in which WS₂ accumulated. These regions appeared greyish in the SEM images. The distribution of elements in one of them is shown in Figure S7. An EDS study demonstrated that WS₂-enriched areas were subject to partial oxidation. However, no appreciable peaks from WO_x were detected in the frequency range of 700–900 cm⁻¹ in the sample's MRS spectra. The presence of oxygen could be due to surface contamination. In the other sections of the track, the WS₂ content decreased significantly. The carbon and nickel content barely changed, indicating that the tribomodification affected mainly the WS₂ layer and the WS₂/g-C interface. The nanostructured surface morphology of the Ni layer was preserved. The wear debris located in the track (125 μm coordinate) consisted mainly of C, O, S, and probably W. The study of tungsten by EDS in WS₂-containing thin films deposited on a silicon substrate has its own deficiencies, which are due to the W Ma1 peak overlapping the intense Si Ka1 peak.

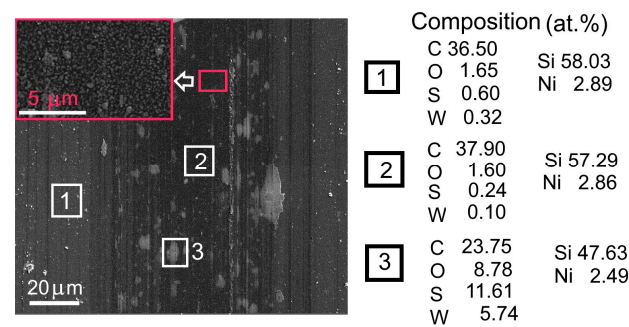


Figure 13. SEM image of the wear track formed after tribotesting on the WS₂(C, Ni)/Si sample in the central region of the track. The EDS-measured compositions of the indicated areas near and inside the track are also shown.

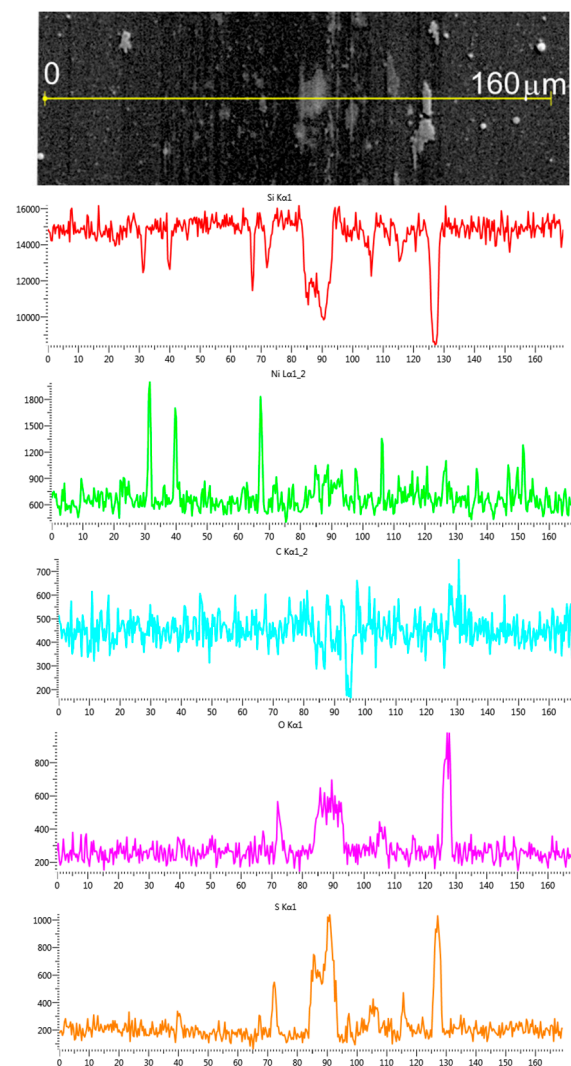


Figure 14. Distribution of different elements across the wear track for the WS₂(C, Ni)/Si sample.

4. Discussion

The tribotests show that the single-layer WS₂ nanocoatings are characterized by fairly good adhesion to the Si substrate. The coating consisted of 2H-WS₂ nanocrystals having basal orientation. The perfect/non-defect basal plane of 2H-WS₂ is characterized by weak chemical activity. However, during PLD, the surface of the Si substrate and the interface with the deposited WS₂ film are bombarded by ions from the laser plasma. The ion

energy may exceed 100 eV. Ion implantation could result in defect and new chemical bond formation on the surface of the underlayer and boost the adhesion of the deposited films. Our result correlates with the results of several studies, which found an improvement in the adhesion of WS₂-based films to Si substrates and/or to pre-deposited W or Si-based interlayers (e.g., [35,65,66]). During friction, the nanocrystals could slide against each other in the near-surface layer. The wear occurred as a result of the displacement of WS₂ surface nanoplates towards the ends of the tracks. The structure in the track and the wear debris almost did not differ from the original structure of the thin-film WS₂ coating. This coating exhibited the highest durability.

The WS₂(C) thin-film coating consisted of three rather smooth WS₂/g-C/WS₂ layers. A thin layer of graphite-like carbon doped with sulfur and/or hydrogen formed on the WS₂/g-C interface. Probably, the sliding of the counterbody against the outer WS₂ layer caused the WS₂ nanoplates to slide against each other and the surface of the g-C film. In the case of weak layer adhesion at the g-C/WS₂ interface, a mechanical mixture of WS₂ and g-C nanoparticles could form on the surface of the underlying WS₂ film. As the counterbody slid, the mixture of WS₂ and g-C nanoparticles caused the coefficient of friction to fluctuate, which could cause the underlying WS₂ film to crack and detach from the silicon substrate. The deep track profile with sharp edges pointed to this fracture mechanism. A thin g-C(S, H) layer was absent at the g-C/WS₂ interface, which probably had an important influence on the friction and wear mechanism of the WS₂(C) coating.

Figure 15 shows the morphology and structure of the prepared WS₂(C, Ni) thin-film coating schematically. The deposition of the nanostructured Ni film caused the formation of irregularities on the coating surface. These irregularities, on the one hand, ensured the retention of the WS₂ film on the surface of the WS₂/g-C/Ni/WS₂/Si sample. On the other hand, the irregularities contributed to the local pressure in the contact area between the counterbody and the coating. The sliding of the counterbody over the WS₂ islands created the conditions for the permanent triboinduced transfer of the solid lubricant WS₂ phase over the entire surface of the track. One might assume that the sliding of WS₂ nanoplates against the surface of the interface g-C(S, H) layer creates the conditions for very low friction. There was no g-C(S, H) nanomaterial on the track due to wear accumulated at the ends of the track when the counterbody stopped. Remarkably, the doping of the g-C film surface with sulfur could continue as long as WS₂ islands were present. During the triboinduced contact of the WS₂ film with the g-C layer, sulfur atoms could penetrate the carbon, which probably contributed to the formation of new carbon forms, such as graphene-like carbon [25,55]. The efficiency of this process was probably controlled by the catalytic properties of nickel. The sliding of WS₂ nanoplates having basal orientation against the amorphous g-C(S, H) layer with graphene-like nanophase inclusions could contribute to a very low friction coefficient. The conditions for very low friction disappear when the carbon layer is removed from the tops of the irregularities caused by the island-like nature of the Ni film structure and the more intensive wear of the counterbody begins.

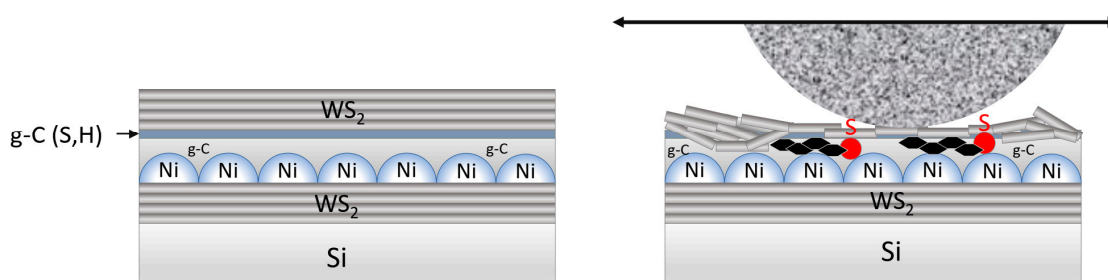


Figure 15. Schematic representation of the WS₂/g-C/Ni/WS₂/Si sample structure before (left) and during (right) contact with the counterbody during tribotests. Explanations are given in the text.

It should be noted that the ultralow coefficient of friction for the WS₂(C, Ni) thin-film coatings was due to the new realized coating architecture. Cao et al. [37] revealed that

for nanocomposite WS₂/a-C coatings when tested in low humidity conditions (5% RH), the friction coefficient was 0.021 at a counterbody load of 5 N. The work [37] contains a comparative analysis of the antifriction properties of WS₂/a-C coatings with literature data on coatings of various compositions. For WS₂/a-C coatings, ultrashort WS₂ nanoplatelets were randomly distributed in an amorphous carbon matrix. The WS₂ nanocrystallites form via selective atomic rearrangement from the amorphous bulk and join into longer crystallites because of defect climbing driven by frictional contact. It was shown that the single-layer WS₂ coating we created provided a friction coefficient of ~0.04. Additional studies have shown that in the case of a higher load (5 N) on the counterbody, the coefficient of sliding friction over such a coating could decrease to ~0.02.

5. Conclusions

Multilayer thin-film coatings containing WS₂, g-C, and Ni nanolayers were created using pulsed laser deposition. The structural and chemical state of these layers and their thickness (in the nanometer range) were controlled by PLD/RPLD modes and the deposition time. When sliding in a nitrogen atmosphere (RH~8%), a single-layer nanocoating consisting of 2H-WS₂ nanocrystals having a basal orientation was subjected to wear by the mechanism of the layer-by-layer removal of the coating material. The coefficient of friction was ~0.04; it was determined by the resistance to the relative slip of the basal planes in the 2H-WS₂ structure.

The alternation of WS₂ and g-C nanolayers in the WS₂(C) thin-film coating did not reduce the friction coefficient. Tribointeraction caused the formation of a mechanical mixture of WS₂ and g-C nanoparticles, providing a coefficient of friction of ~0.04; the mixture was relatively quickly removed from the wear track. The incorporation of the Ni nanolayer into the WS₂(C, Ni) structure caused the coefficient of friction to decrease to values approaching superlubricity. The lowest value of the friction coefficient in nitrogen with a counterbody load of 1 N was 0.013. This was due to the movement of WS₂ nanoplates against the amorphous g-C layer doped with sulfur and/or hydrogen. S and/or H atoms could be introduced into the carbon layer from hydrogen sulfide (during the RPLD of a WS₂ film on the g-C nanolayer) and as a result of the tribochemical reaction between WS₂ and g-C. The decrease in the friction coefficient could be explained by the transformation of the graphite-like state of carbon into the graphene-like one under the catalytic effect of nickel. Nickel deposition was accompanied by the formation of submicron-sized (in plane) Ni particles, which changed the contact conditions between the counterbody and the thin-film WS₂(C, Ni) coating and contributed to the ultralow coefficient of friction at a relatively low load (1 N) on the counterbody.

Supplementary Materials: The following supporting information can be downloaded at: <https://www.mdpi.com/article/10.3390/ma16010282/s1>, Figure S1: A scheme of fabrication of multilayer thin-film coatings using the laser technique: (a) RPLD of thin WS₂ film on a Si substrate; (b) PLD of thin Ni film on the Si substrate covered with WS₂; and (c) PLD of thin g-C film on the Si substrate covered with bilayer Ni/WS₂ film. To obtain multilayer WS₂/g-C/Ni/WS₂ coating, RPLD of thin WS₂ film was applied again. Figure S2: SEM image and EDS surface distribution of different elements in the WS₂(C, Ni) thin film coating deposited on Si substrate. The distribution of W Ma1 is not shown because it was identical to the distribution of silicon. Quantification of tungsten in the presence of silicon is difficult due to the overlap of the W Ma1 peak with the Si Ka1 of the substrate. The figure illustrates the presence of Ni particles of micron, submicron, and nanometer sizes. Carbon rather effectively covers Ni particles of submicron and nanometer sizes; however, Ni particles with a size close to a micron are not uniformly covered with carbon. This is probably responsible for its modification during RPLD of the WS₂ in H₂S film. An increase in the intensity of Ka1 peak around Ni particle (located at ~3–4 μm) indicates the possibility of sulfurization of the Ni microparticle. Figure S3: (a) MRS spectrum for g-C layer; (b) cross-section TEM image of the interface layer in WS₂(C, Ni) thin-film coating. The structure of the WS₂/g-C interface layer was studied by high-resolution transmission electron microscopy (HRTEM), using a Carl Zeiss Libra 200FE microscope. The HRTEM image confirms the formation of layered 2H-WS₂ film on the disordered surface of

g-C layer. Figure S4: SEM image and EDS spectra for different regions in the wear track for WS₂(C) thin-film coating. Figure S5: SEM image and EDS-measured surface distribution of different elements in the WS₂(C) thin-film coating across the wear track. Figure S6: MRS spectrum of WS₂(C, Ni) thin-film coating, which was measured with increased laser intensity for the wear debris located at the end of the wear track. Figure S7: SEM image and EDS spectra for different regions in the wear track for WS₂(C, Ni) thin-film coating. Figure S8: SEM image and EDS-measured surface distribution of different elements in the WS₂(C, Ni) thin-film coating around the S-enriched area in the wear track.

Author Contributions: Conceptualization, V.Y.F.; methodology, M.V.D. and R.I.R.; PLD, RPLD, and tribology testing of the films, D.V.F.; investigation, XPS, and SEM studies, R.I.R. and M.D.G.; investigation and MRS, N.V.D. and V.S.V.; writing—original draft preparation, V.Y.F. and R.I.R.; writing—review and editing, R.I.R. All authors have read and agreed to the published version of the manuscript.

Funding: This research was carried out at the National Research Nuclear University MEPhI and funded by the Russian Science Foundation, grant number 22-29-00197.

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Not applicable.

Conflicts of Interest: The authors declare no conflict of interest.

References

- Spalvins, T. A review of recent advances in solid film lubrication. *J. Vac. Sci. Technol.* **1987**, *5*, 212–219. [\[CrossRef\]](#)
- Müller, C.; Menoud, C.; Maillat, M.; Hintermann, H.E. Thick compact MoS₂ coatings. *Surf. Coat. Technol.* **1988**, *36*, 351–359. [\[CrossRef\]](#)
- Fominskii, V.; Markeev, A.; Nevolin, V. Pulsed ion beams for modification of metal surface properties. *Vacuum* **1991**, *42*, 73–74. [\[CrossRef\]](#)
- Pope, L.E.; Panitz, J.K.G. The effect of Hertzian stress and test atmosphere on the friction coefficient of MoS₂ coatings. *Surf. Coat. Technol.* **1988**, *36*, 341–350. [\[CrossRef\]](#)
- Zhuang, W.; Fan, X.; Li, W.; Li, H.; Zhang, L.; Peng, J.; Cai, Z.; Mo, J.; Zhang, G.; Zhu, M. Comparing space adaptability of diamond-like carbon and molybdenum disulfide films toward synergistic lubrication. *Carbon* **2018**, *134*, 163–173. [\[CrossRef\]](#)
- Shi, J.; Ma, G.; Han, C.; Li, G.; Liu, Y.; Liu, Q. Tribological properties and bearing application of Mo-based films in space environment. *Vacuum* **2021**, *188*, 110217. [\[CrossRef\]](#)
- Trivedi, H.K.; Massey, M.M.; Bhattacharya, R.S.; Strahl, G.A.; Collumb, D. Next generation lubrication system for weapons. *Tribol. Lett.* **2001**, *10*, 229–235. [\[CrossRef\]](#)
- Behrens, B.A.; Maier, H.J.; Hübner, S.; Bonk, C.; Almohallami, A.; Lummer, C.; Schein, P.; Scheland, H.; Mücke-Camuz, M. Wear Behavior of MoS₂ Lubricant Layers during Sheet Metal Forming. *Procedia Eng.* **2017**, *183*, 357–362. [\[CrossRef\]](#)
- Zhang, X.; Vitchev, R.G.; Lauwerens, W.; Stals, L.; He, J.; Celis, J.-P. Effect of crystallographic orientation on fretting wear behavior of MoS_x coatings in dry and humid air. *Thin Solid Films* **2001**, *396*, 69–77. [\[CrossRef\]](#)
- Fominski, V.; Demin, M.; Nevolin, V.; Fominski, D.; Romanov, R.; Gritskovich, M.; Smirnov, N. Reactive Pulsed Laser Deposition of Clustered-Type MoS_x ($x \sim 2, 3$, and 4) Films and Their Solid Lubricant Properties at Low Temperature. *Nanomaterials* **2020**, *10*, 653. [\[CrossRef\]](#)
- Ren, S.; Shang, K.; Cui, M.; Wang, L.; Jibin, P.; Yi, P. Structural design of MoS₂-based coatings toward high humidity and wide temperature. *J. Mater. Sci.* **2019**, *54*, 11889–11902. [\[CrossRef\]](#)
- Hu, J.J.; Zabinski, J.S.; Bultman, J.E.; Sanders, J.H.; Voevodin, A.A. Structure characterization of pulsed laser deposited MoS_x-WSe_y composite films of tribological interests. *Tribol. Lett.* **2006**, *24*, 126–135. [\[CrossRef\]](#)
- Fominski, V.Y.; Grigoriev, S.N.; Gnedovets, A.G.; Romanov, R.I. Pulsed laser deposition of composite Mo-Se-Ni-C coatings using standard and shadow mask configuration. *Surf. Coat. Technol.* **2012**, *206*, 5046–5054. [\[CrossRef\]](#)
- Cavaleiro, A.; Polcar, T. Review on self-lubricant transition metal dichalcogenide nanocomposite coatings alloyed with carbon. *Surf. Coat. Technol.* **2011**, *206*, 686–695. [\[CrossRef\]](#)
- Hudeca, T.; Mikula, M.; Satrapinsky, L.; Roch, T.; Truchlý, M.; Švec, P., Jr.; Humeniuc, T.; Polcard, T. Structure, mechanical and tribological properties of Mo-S-N solid lubricant coatings. *Appl. Surf. Sci.* **2019**, *486*, 1–14. [\[CrossRef\]](#)
- Yaqub, T.B.; Vuchkov, T.; Sanguino, P.; Polcar, T.; Cavaleiro, A. Comparative study of DC and RF sputtered MoSe₂ coatings containing carbon—An approach to optimize stoichiometry, microstructure, crystallinity, and hardness. *Coatings* **2020**, *10*, 133. [\[CrossRef\]](#)
- Maharana, H.S.; Mondal, K. Manifestation of Hall–Petch breakdown in nanocrystalline electrodeposited Ni-MoS₂ coating and its structure dependent wear resistance behavior. *Surf. Coat. Technol.* **2021**, *410*, 126950. [\[CrossRef\]](#)

18. Gao, X.; Fu, Y.; Jiang, D.; Wang, D.; Xu, S.; Liu, W.; Weng, L.; Yang, J.; Sun, J.; Hu, M. Constructing WS₂/MoS₂ nano-scale multilayer film and understanding its positive response to space environment. *Surf. Coat. Technol.* **2018**, *353*, 8–17. [\[CrossRef\]](#)
19. Noshiro, J.; Watanabe, S.; Sakurai, T.; Miyake, S. Friction properties of co-sputtered sulfide/DLC solid lubricating films. *Surf. Coat. Technol.* **2006**, *200*, 5849–5854. [\[CrossRef\]](#)
20. Wu, Y.; Li, H.; Ji, L.; Ye, Y.; Chen, J.; Zhou, H. A long-lifetime MoS₂/a-C:H nanoscale multilayer film with extremely low internal stress. *Surf. Coat. Technol.* **2013**, *236*, 438–443. [\[CrossRef\]](#)
21. Xu, J.; He, T.F.; Chai, L.Q.; Qiao, L.; Wang, P.; Liu, W.M. Growth and characteristics of self-assembled MoS₂/Mo-S-C nanoperiod multilayers for enhanced tribological performance. *Sci. Rep.* **2016**, *6*, 25378. [\[CrossRef\]](#)
22. Martin, J.M.; Donnet, C.; Mogne, T.L.; Epicier, T. Superlubricity of molybdenum disulphide. *Phys. Rev. B* **1993**, *48*, 10583. [\[CrossRef\]](#) [\[PubMed\]](#)
23. Martin, D.; Verhoeven, G.S.; Pradeep, N.; Frenken, J.W.M.; Heimberg, J.A.; Zandbergen, H.W. Superlubricity of Graphite. *Phys. Rev. Lett.* **2004**, *92*, 126101. [\[CrossRef\]](#)
24. Qu, C.; Shi, S.; Ma, M.; Zheng, Q. Rotational Instability in Superlubric Joints. *Phys. Rev. Lett.* **2019**, *122*, 246101. [\[CrossRef\]](#) [\[PubMed\]](#)
25. Berman, D.; Narayanan, B.; Cherukara, M.J.; Sankaranarayanan, S.K.R.S.; Erdemir, A.; Zinovev, A.; Sumant, A.V. Operando tribochemical formation of onion-like-carbon leads to macroscale superlubricity. *Nat. Commun.* **2018**, *9*, 1164. [\[CrossRef\]](#) [\[PubMed\]](#)
26. Zhang, R.; Yang, X.; Pu, J.; He, Z.; Xiong, L. Extraordinary macroscale lubricity of sonication-assisted fabrication of MoS₂ nano-ball and investigation of in situ formation mechanism of graphene induced by tribochemical reactions. *Appl. Surf. Sci.* **2020**, *510*, 145456. [\[CrossRef\]](#)
27. Yu, G.; Gong, Z.; Jiang, B.; Wang, D.; Bai, C.; Zhang, J. Superlubricity for hydrogenated diamond like carbon induced by thin MoS₂ and DLC layer in moist air. *Diam. Relat. Mater.* **2020**, *102*, 107668. [\[CrossRef\]](#)
28. Chen, X.; Yin, X.; Qi, W.; Zhang, C.; Choi, J.; Wu, S.; Wang, R.; Luo, J. Atomic-scale insights into the interfacial instability of superlubricity in hydrogenated amorphous carbon films. *Sci. Adv.* **2020**, *6*, eaay1272. [\[CrossRef\]](#)
29. Gong, Z.; Ji, X.; Ma, W.; Zhang, B.; Zhang, J. Hierarchical structure graphitic-like/MoS₂ film as superlubricity material. *Appl. Surf. Sci.* **2017**, *413*, 381–386. [\[CrossRef\]](#)
30. Jiang, A.; Cao, X.; Wang, Z.; Ma, J.; Xiao, J.; Ma, S. Friction performance and corrosion resistance of MoS₂/DLC composite films deposited by magnetron sputtering. *Results Phys.* **2021**, *25*, 104278. [\[CrossRef\]](#)
31. Gao, K.; Lai, Z.; Jia, Q.; Zhang, B.; Wei, X.; Zhang, J. Bilayer a-C:H/MoS₂ film to realize superlubricity in open atmosphere. *Diam. Relat. Mater.* **2020**, *108*, 107973. [\[CrossRef\]](#)
32. Yu, G.; Zhang, Z.; Tian, P.; Gong, Z.; Zhang, J. Macro-scale superlow friction enabled when MoS₂ flakes lubricate hydrogenated diamond-like carbon film. *Ceram. Int.* **2021**, *47*, 10980–10989. [\[CrossRef\]](#)
33. Nossa, A.; Cavaleiro, A.; Carvalho, N.J.M.; Kooi, B.J.; De Hosson, J.T.M. On the microstructure of tungsten disulfide films alloyed with carbon and nitrogen. *Thin Solid Films* **2005**, *484*, 389–395. [\[CrossRef\]](#)
34. Zhu, L.-N.; Wang, C.-B.; Wang, H.-D.; Xu, B.-S.; Zhuang, D.-M.; Liu, J.-J.; Li, G.-L. Microstructure and tribological properties of WS₂/MoS₂ multilayer films. *Appl. Surf. Sci.* **2012**, *258*, 1944–1948. [\[CrossRef\]](#)
35. Xu, S.; Gao, X.; Hu, M.; Sun, J.; Jiang, D.; Zhou, F.; Liu, W.; Weng, L. Nanostructured WS₂-Ni composite films for improved oxidation, resistance and tribological performance. *Appl. Surf. Sci.* **2014**, *288*, 15–25. [\[CrossRef\]](#)
36. Gao, X.; Fu, Y.; Jiang, D.; Wang, D.; Yang, J.; Weng, L.; Hu, M.; Sun, J. Structural, Mechanical, and Tribological Properties of WS₂-Al Nanocomposite Film for Space Application. *Tribol. Lett.* **2018**, *66*, 137. [\[CrossRef\]](#)
37. Cao, H.; Momand, J.; Syari'ati, A.; Wen, F.; Rudolf, P.; Xiao, P.; De Hosson, J.T.M.; Pei, Y. Temperature-Adaptive Ultralubricity of a WS₂/a-C Nanocomposite Coating: Performance from Room Temperature up to 500 °C. *ACS Appl. Mater. Interfaces* **2021**, *13*, 28843–28854. [\[CrossRef\]](#)
38. Watanabe, S.; Noshiro, J.; Miyake, S. Tribological characteristics of WS₂/MoS₂ solid lubricating multilayer films. *Surf. Coat. Technol.* **2004**, *183*, 347–351. [\[CrossRef\]](#)
39. Watanabe, S.; Noshiro, J.; Miyake, S. Friction properties of WS₂/MoS₂ multilayer films under vacuum environment. *Surf. Coat. Technol.* **2004**, *188–189*, 644–648. [\[CrossRef\]](#)
40. Xi, J.; Huang, X.; Hu, M.; Xiang, W. Dependence of laser parameters on structural properties of pulsed laser-deposited MoS₂ thin films applicable for field effect transistors. *J. Mater. Sci. Mater. Electron.* **2020**, *31*, 21118–21127. [\[CrossRef\]](#)
41. Seo, S.; Kim, S.; Choi, H.; Lee, J.; Yoon, H.; Piao, G.; Park, J.-C.; Jung, Y.; Song, J.; Jeong, S.Y.; et al. Direct in situ growth of centimeter-scale multi-heterojunction MoS₂/WS₂/WSe₂ thin-film catalyst for photo-electrochemical hydrogen evolution. *Adv. Sci.* **2019**, *6*, 1900301. [\[CrossRef\]](#) [\[PubMed\]](#)
42. Fominski, V.Y.; Romanov, R.I.; Nevolin, V.N.; Fominski, D.V.; Komleva, O.V.; Popov, V.V. Formation of ultrathin MoS₂ films using laser-based methods. *J. Phys. Conf. Ser.* **2019**, *1238*, 012007. [\[CrossRef\]](#)
43. Ossia, P.M.; Bottania, C.E.; Miotello, A. Pulsed-laser deposition of carbon: From DLC to cluster-assembled films. *Thin Solid Films* **2005**, *482*, 2–8. [\[CrossRef\]](#)
44. Fominski, V.; Nevolin, V.; Romanov, R.; Smirnov, A.; Scharff, W. Atomic mixing and chemical bond formation in MoS_x/Fe thin-film system deposited from a laser plume in a high-intensity electrostatic field. *Thin Solid Films* **2002**, *422*, 39–47. [\[CrossRef\]](#)

45. Boubiche, N.; Hamouchi, J.E.; Hulik, J.; Abdesslam, M.; Speisser, C.; Djefal, F.; Normand, F.L. Kinetics of graphitization of thin diamond-like carbon (DLC) films catalyzed by transition metal. *Diam. Relat. Mater.* **2019**, *91*, 190–198. [[CrossRef](#)]
46. Berman, D.; Deshmukh, S.A.; Narayanan, B.; Sankaranarayanan, S.K.R.S.; Yan, Z.; Balandin, A.A.; Zinovev, A.; Rosenmann, D.; Sumant, A.V. Metal-induced rapid transformation of diamond into single and multilayer graphene on wafer scale. *Nat. Commun.* **2016**, *7*, 12099. [[CrossRef](#)]
47. Mignuzzi, S.; Pollard, A.J.; Bonini, N.; Brennan, B.; Gilmore, I.S.; Pimenta, M.A.; Richards, D.; Roy, D. Effect of disorder on Raman scattering of single-layer MoS₂. *Phys. Rev. B* **2015**, *91*, 195411. [[CrossRef](#)]
48. Li, J.; Su, W.; Chen, F.; Fu, L.; Ding, S.; Song, K.; Huang, X.; Zhang, L. Atypical defect-mediated photoluminescence and resonance Raman spectroscopy of monolayer WS₂. *J. Phys. Chem. C* **2019**, *123*, 3900–3907. [[CrossRef](#)]
49. Markeev, A.M.; Kozodaev, M.G.; Slavich, A.S.; Romanov, R.I.; Zarubin, S.S. Influence of reducing agent on properties of thin WS₂ nanosheets prepared by sulfurization of atomic layer-deposited WO₃. *J. Phys. Chem. C* **2020**, *124*, 28169–28177. [[CrossRef](#)]
50. Romanov, R.I.; Kozodaev, M.G.; Chernikova, A.G.; Zabrosaev, I.V.; Chouprik, A.A.; Zarubin, S.S.; Novikov, S.M.; Volkov, V.S.; Markeev, A.M. Thickness-Dependent Structural and Electrical Properties of WS₂ Nanosheets Obtained via the ALD-Grown WO₃ Sulfurization Technique as a Channel Material for Field-Effect Transistors. *ACS Omega* **2021**, *6*, 34429–34437. [[CrossRef](#)]
51. Pimenta, M.A.; Dresselhaus, G.; Dresselhaus, M.S.; Cancado, L.G.; Jorio, A.; Saito, R. Studying disorder in graphite-based systems by Raman spectroscopy. *Phys. Chem. Chem. Phys.* **2007**, *9*, 1276–1291. [[CrossRef](#)] [[PubMed](#)]
52. Ferrari, A.C.; Robertson, J. Raman spectroscopy of amorphous, nanostructured, diamond-like carbon, and nanodiamond. *Philos. Trans. R. Soc. A* **2004**, *362*, 2477–2512. [[CrossRef](#)] [[PubMed](#)]
53. Ferrari, A.C.; Rodil, S.E.; Robertson, J. Interpretation of infrared and Raman spectra of amorphous carbon nitrides. *Phys. Rev. B* **2003**, *67*, 155306. [[CrossRef](#)]
54. Ferrari, A.C.; Robertson, J. Origin of the 1150 cm^{−1} Raman mode in nanocrystalline diamond. *Phys. Rev. B* **2001**, *63*, 121405. [[CrossRef](#)]
55. Sun, L.; Gao, K.; Jia, Q.; Bai, C.; Zhang, B.; Tan, X.; Zhang, J. Grown of superlubricity a-C:H/MoS₂ film on 9Cr18Mo steel for industrial application. *Diam. Relat. Mater.* **2021**, *117*, 108479. [[CrossRef](#)]
56. Ungár, T.; Gubicza, J.; Trichy, G.; Pantea, C.; Zerda, T.W. Size and shape of crystallites and internal stresses in carbon blacks. *Compos. Part A Appl. Sci. Manuf.* **2005**, *36*, 431–436. [[CrossRef](#)]
57. Cuesta, A.; Dhemelinco, P.; Laureyns, J.; Martinez-Alonso, A.; Tascon, J.M.D. Raman microprobe studies on carbon materials. *Carbon* **1994**, *32*, 1523–1532. [[CrossRef](#)]
58. Fominski, V.; Fominski, D.; Romanov, R.; Gritskevich, M.; Demin, M.; Shvets, P.; Maksimova, K.; Goikhman, A. Specific features of reactive pulsed laser deposition of solid lubricating nanocomposite Mo-S-C-H thin-film coatings. *Nanomaterials* **2020**, *10*, 2456. [[CrossRef](#)]
59. Wang, Q.H.; Kalantar-Zadeh, K.; Kis, A.; Coleman, J.N.; Strano, M.S. Electronics and optoelectronics of two-dimensional transition metal dichalcogenides. *Nat. Nanotechnol.* **2012**, *7*, 699–712. [[CrossRef](#)]
60. Qin, Z.; Zeng, D.; Zhang, J.; Wu, C.; Wen, Y.; Shan, B.; Xie, C. Effect of layer number on recovery rate of WS₂ nanosheets for ammonia detection at room temperature. *Appl. Surf. Sci.* **2017**, *414*, 244–250. [[CrossRef](#)]
61. Wu, H.; Li, W.; Li, W.; Dai, Y.; Guo, J.; Wang, S.; Song, J.; Odunmbaku, G.O.; Zhang, H.; Boi, F.S. Evidence for increased metallicity arising from carbon-sulfur bonding and amorphization effects in sulfur-doped pyrolytic graphite. *Diam. Relat. Mater.* **2022**, *121*, 108729. [[CrossRef](#)]
62. Odunmbakua, O.; Songa, J.; Wang, S.; Taallaha, A.; Dai, Y.; Li, W.; Li, W.; He, Y.; Guo, J.; Zhanga, H.; et al. Nucleation of carbon-sulfur phases by manipulation of vertically-aligned mm-long films of iron-filled few-wall/multiwall carbon nanotubes. *Carbon Trends* **2021**, *5*, 100102. [[CrossRef](#)]
63. Bautista-Flores, C.; Arellano-Peraza, J.S.; Sato-Berrú, R.Y.; Camps, E.; Mendoza, D. Sulfur and few-layer graphene interaction under thermal treatments. *Chem. Phys. Lett.* **2016**, *665*, 121–126. [[CrossRef](#)]
64. Jia, Q.; Yang, Z.; Sun, L.; Gao, K.; Zhang, B.; Zhang, X.; Zhang, J. Catalytic superlubricity via in-situ formation of graphene during sliding friction on Au@a-C:H films. *Carbon* **2022**, *186*, 180–192. [[CrossRef](#)]
65. Sundberg, J.; Nyberg, H.; Särhammar, E.; Nyberg, T.; Jacobson, S.; Jansson, U. Influence of composition, structure and testing atmosphere on the tribological performance of W-S-N coatings. *Surf. Coat. Technol.* **2014**, *285*, 86–94. [[CrossRef](#)]
66. Nossa, A.; Cavaleiro, A. The influence of the addition of C and N on the wear behaviour of W-S-C/N coatings. *Surf. Coat. Technol.* **2001**, *142–144*, 984–991. [[CrossRef](#)]

Disclaimer/Publisher's Note: The statements, opinions and data contained in all publications are solely those of the individual author(s) and contributor(s) and not of MDPI and/or the editor(s). MDPI and/or the editor(s) disclaim responsibility for any injury to people or property resulting from any ideas, methods, instructions or products referred to in the content.