

Supplementary Materials

Evaluation of Rhodamine B Photocatalytic Degradation over BaTiO₃-MnO₂ Ceramic Materials

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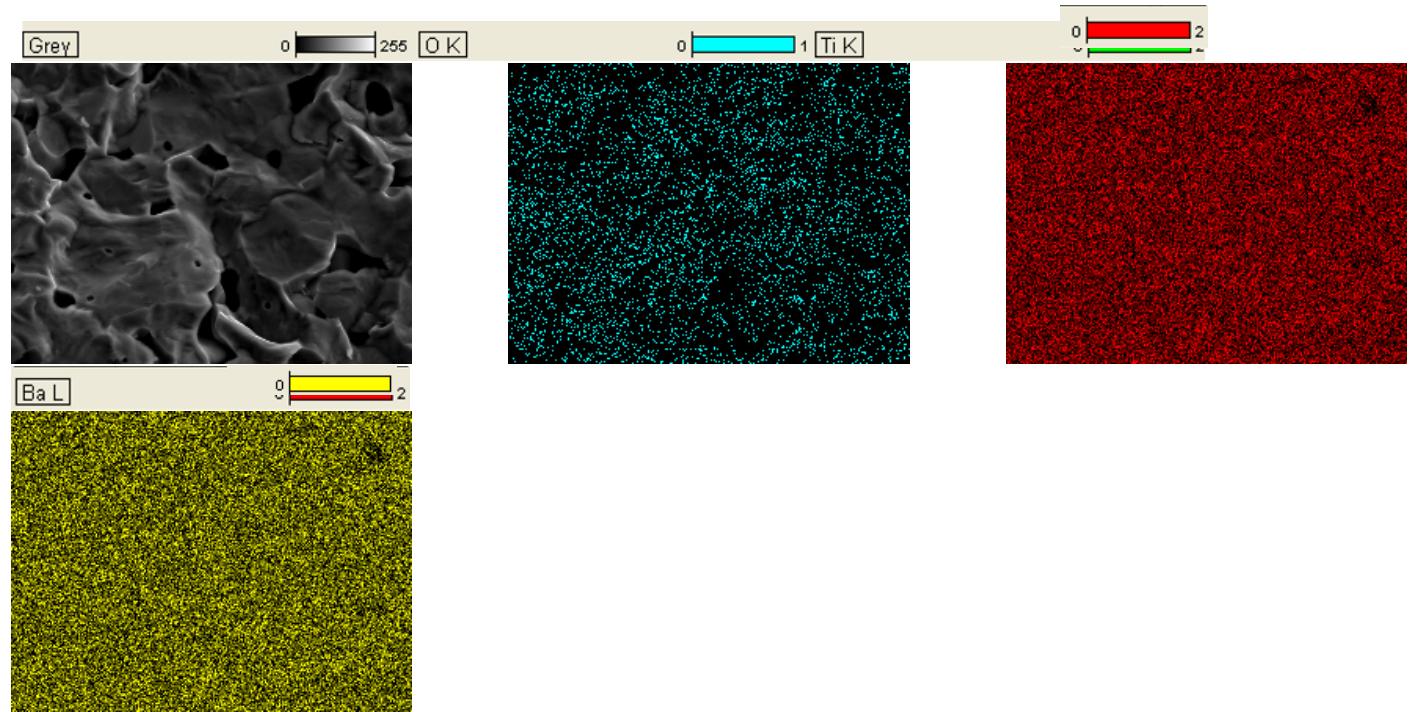
Table S1. Photodecolorization of RhB and calculated reaction rates of different photocatalysts.

Photocatalyst	C/C ₀	k ^a (min ⁻¹)	R ²
BTO	0.88	3.0×10 ⁻⁴	0.78
BTO_1	0.47 (0.76) ^b	1.4×10 ⁻³ (8.0×10 ⁻⁴) ^b	0.99 (0.99) ^b
BTO_2	0.40 (0.64) ^b	1.9×10 ⁻³ (1.2×10 ⁻³) ^b	0.97 (0.99) ^b
BTO_3	0.30 (0.70) ^b	3.3×10 ⁻³ (1.0×10 ⁻³) ^b	0.91 (0.98) ^b

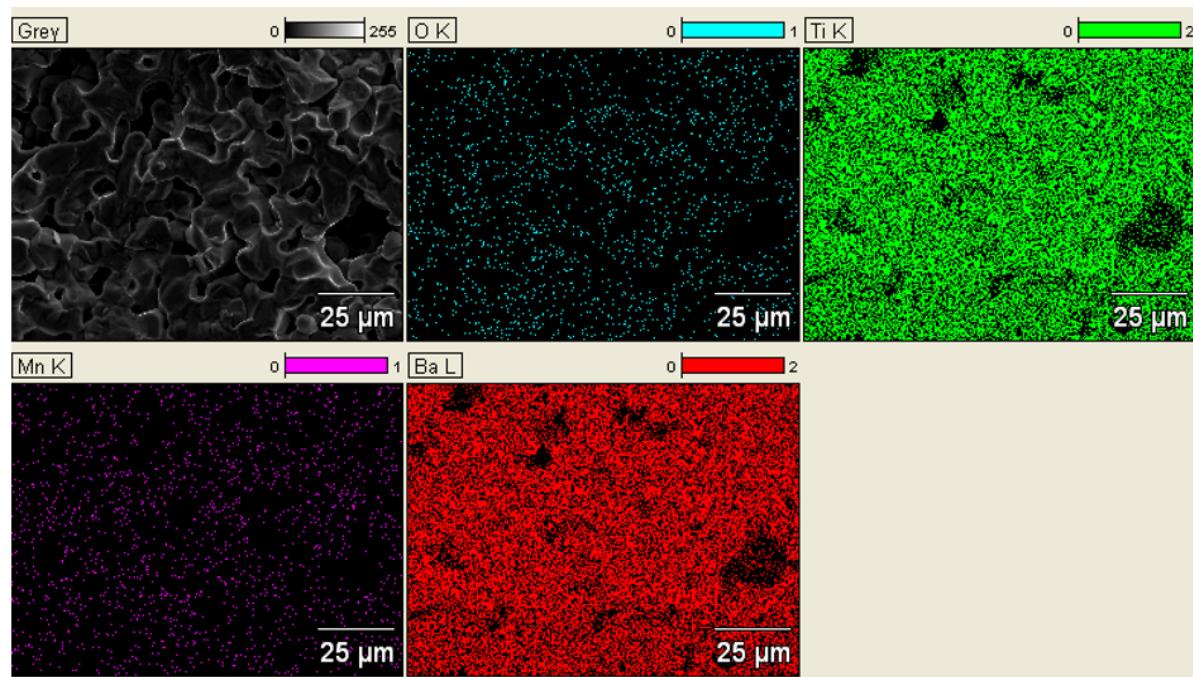
^a k – first order rate constant calculated from $\ln(C/C_0) = -kt$, where C₀ and C (mg L⁻¹) are the concentrations of RhB at time t (min); ^b values in parenthesis for 2nd photoreaction cycle

Figure S1. Elemental area-mappings for BTO, BTO_1 and BTO_3, representative examples

BTO



BTO_1



BTO_3

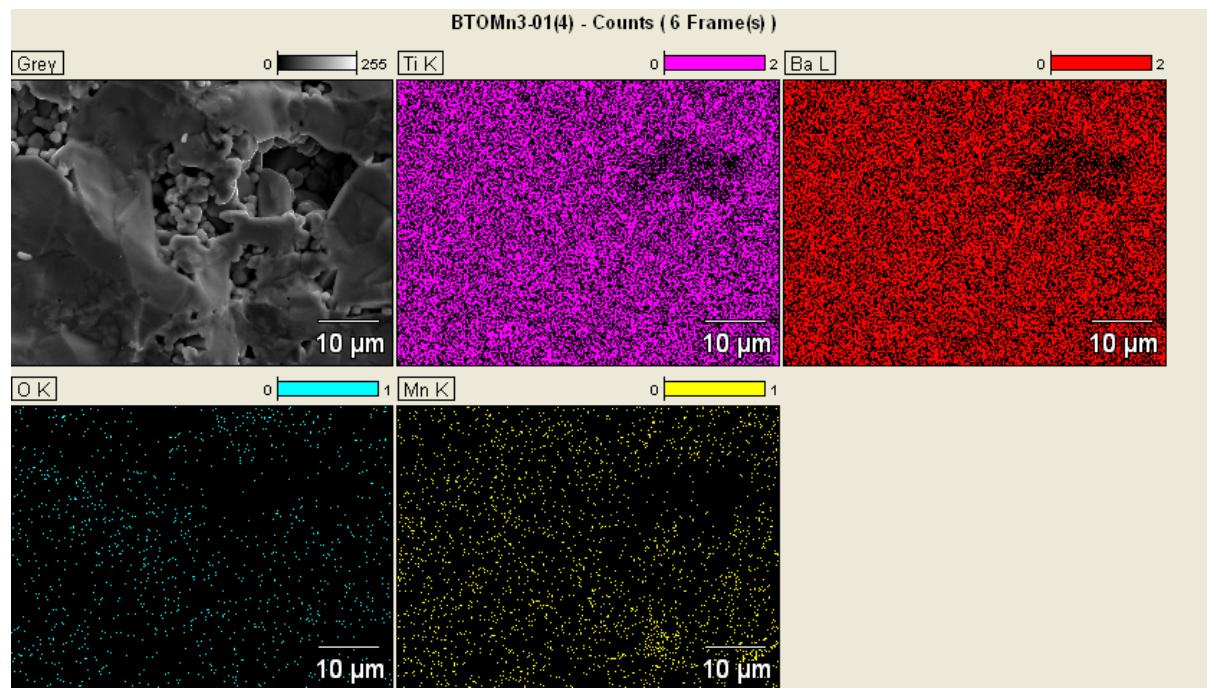


Figure S2. UV-Vis absorption spectra of BaTiO_3 ceramics: BTO (a), BTO_1 (b), BTO_2 (c) and BTO_3 (d) in $220\text{-}500 \text{ cm}^{-1}$ range in Nujol mul

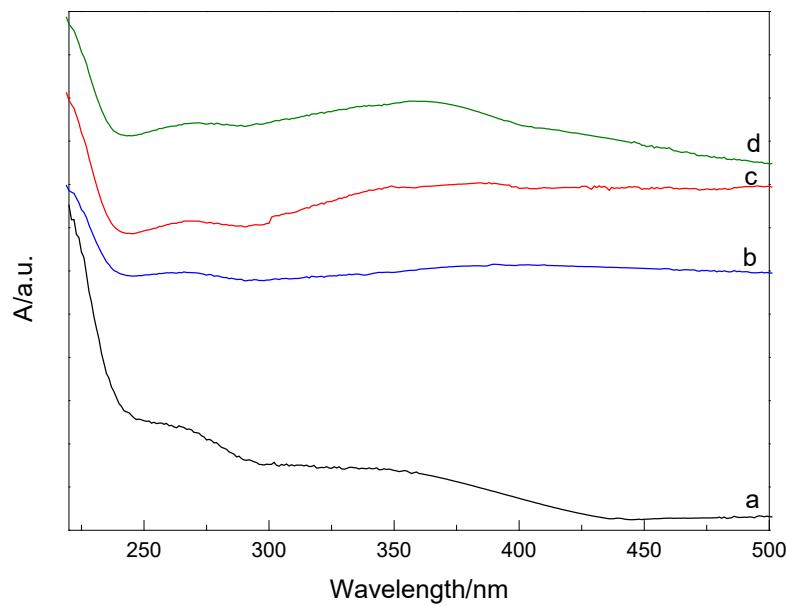


Figure S3. UV-Vis absorption spectra of RhB dye solution with different irradiation time using BTO_1 as photocatalyst

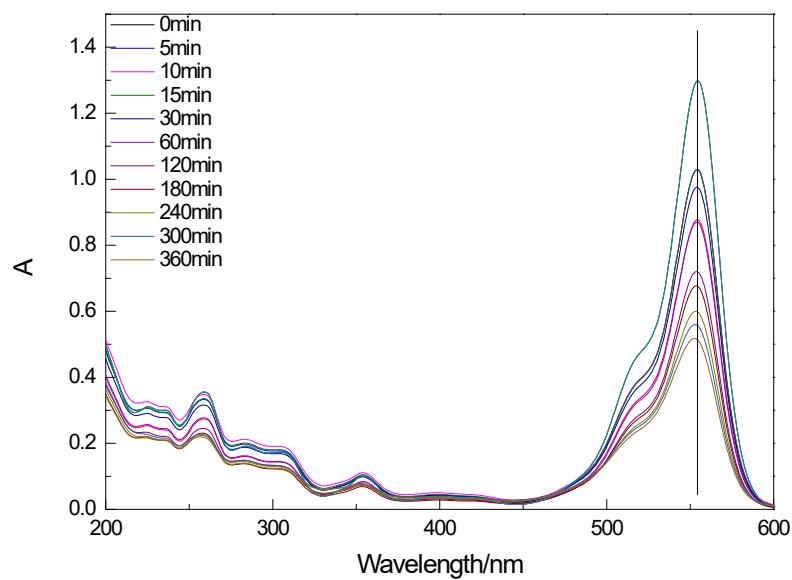


Figure S4. UV-Vis absorption spectra of RhB dye solution with different irradiation time using BTO_2 as photocatalyst

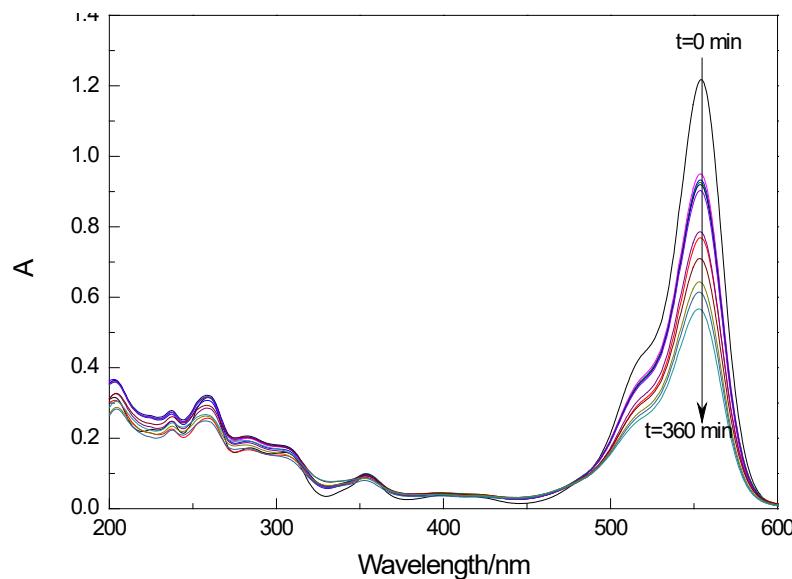


Figure S5. UV-Vis absorption spectra of RhB dye solution with different irradiation time using BTO_3 as photocatalyst

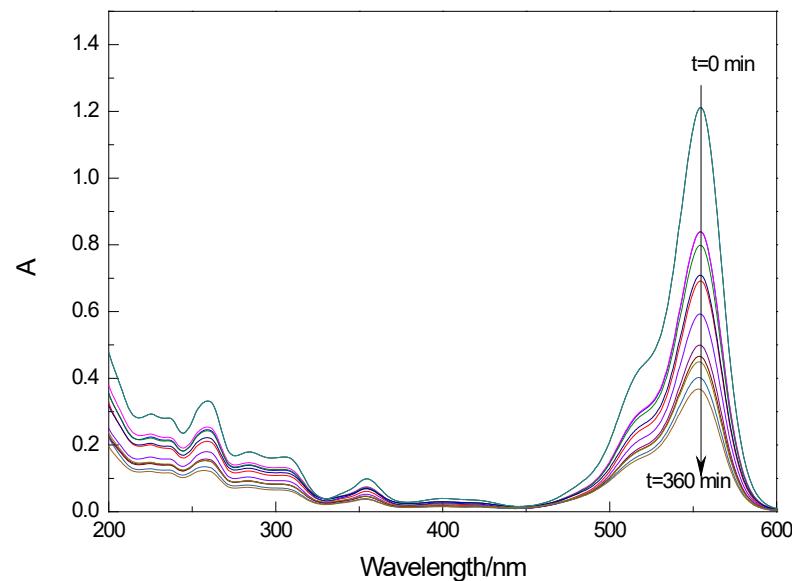


Figure S6. FTIR spectra of BTO_1 (a), BTO_2 (b), BTO_3 (c) and BTO (d) after second photocatalytic cycle.

