

Figure S1. The chemical construction of the investigated molecules.

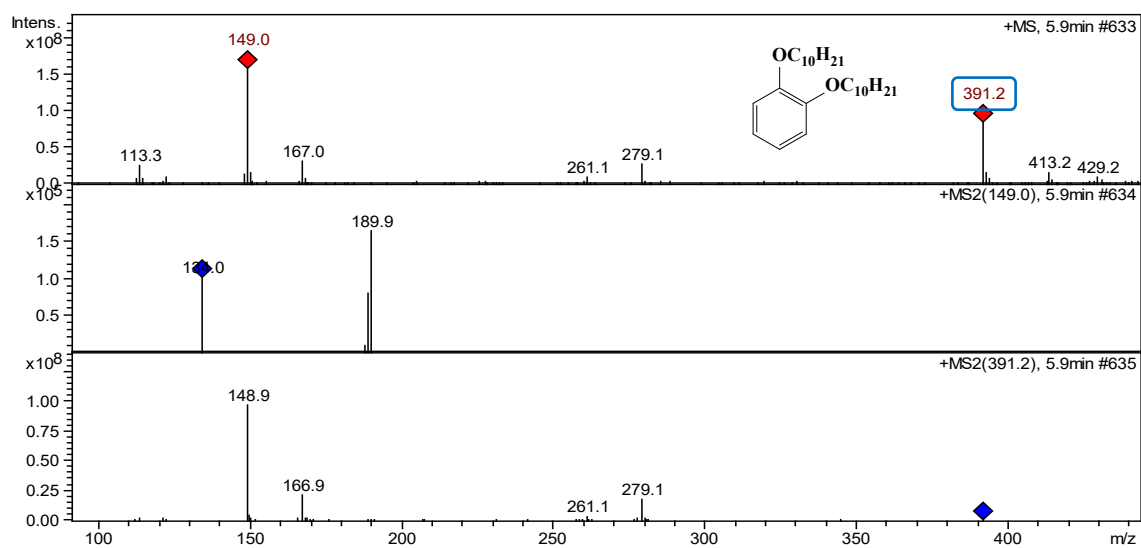


Figure S2. Mass spectrum of B1.

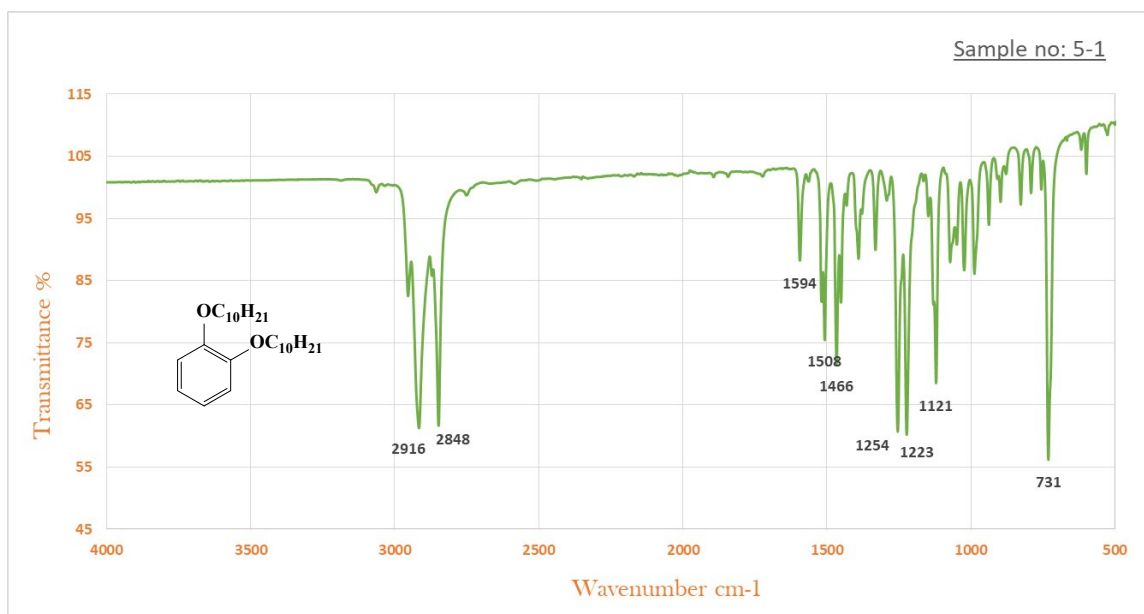


Figure S3. IR spectrum of B1.

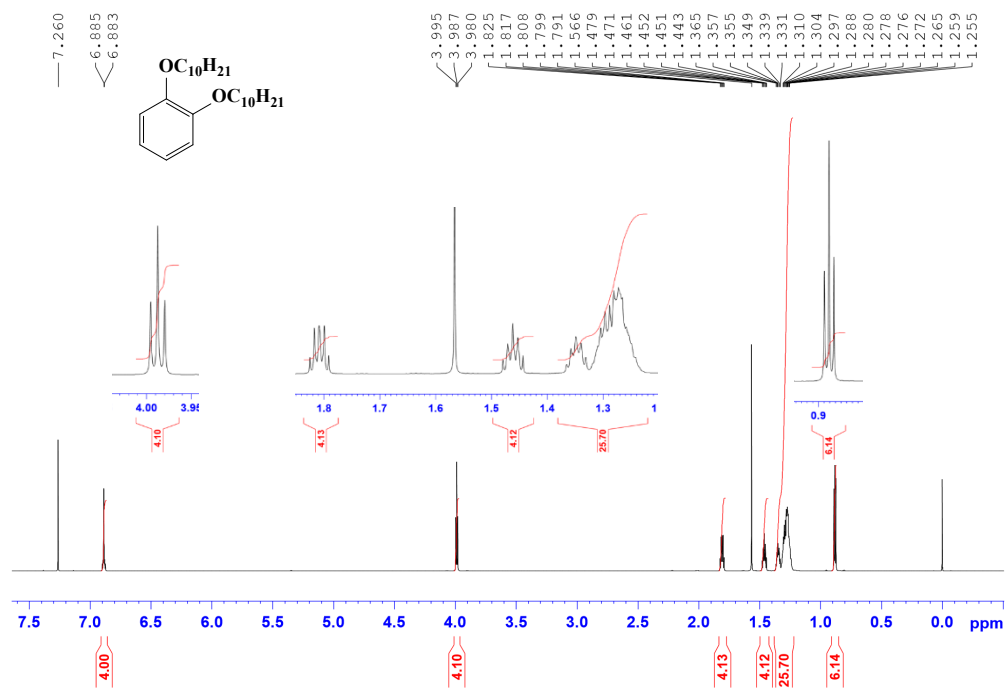


Figure S4. ¹H NMR (CDCl₃) spectrum of B1.

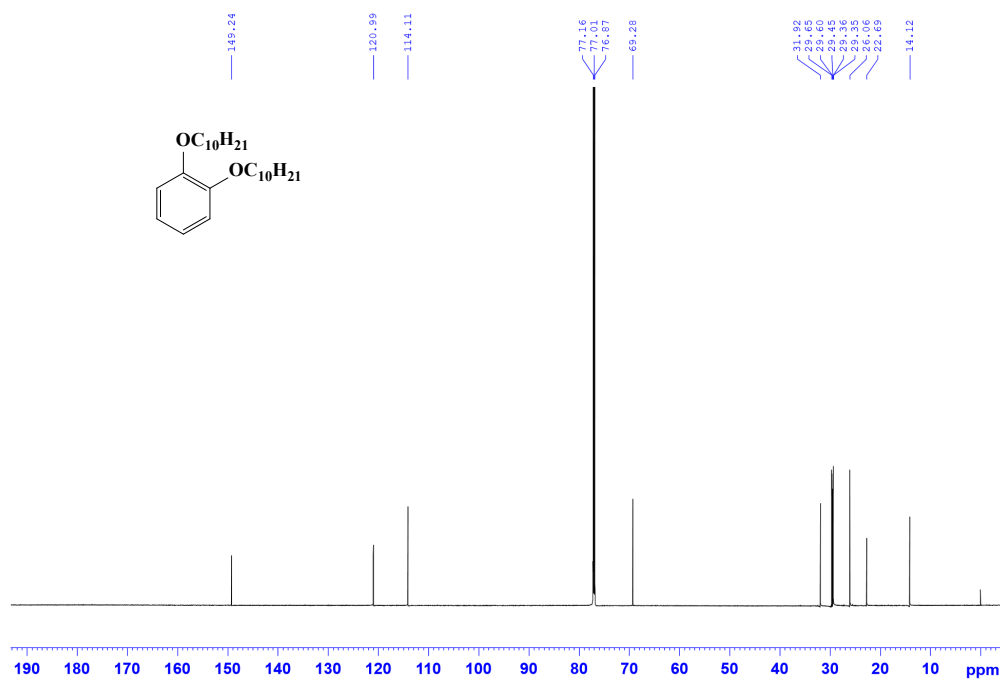


Figure S5. ¹³C NMR (CDCl₃) spectrum of B1.

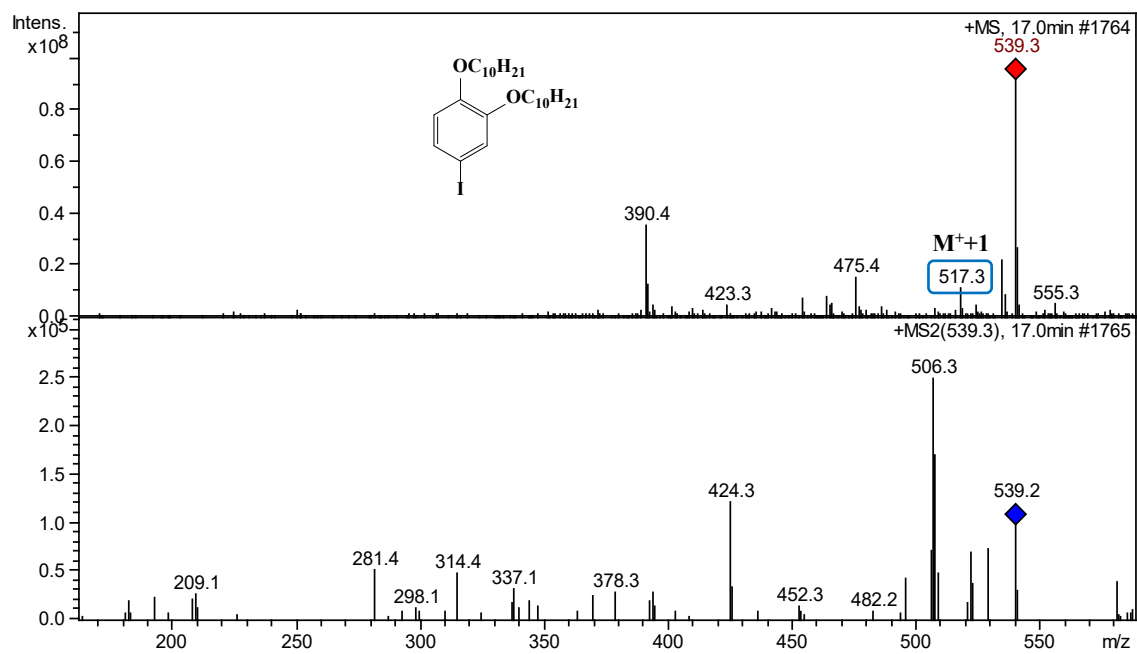


Figure S6. Mass spectrum of B2.

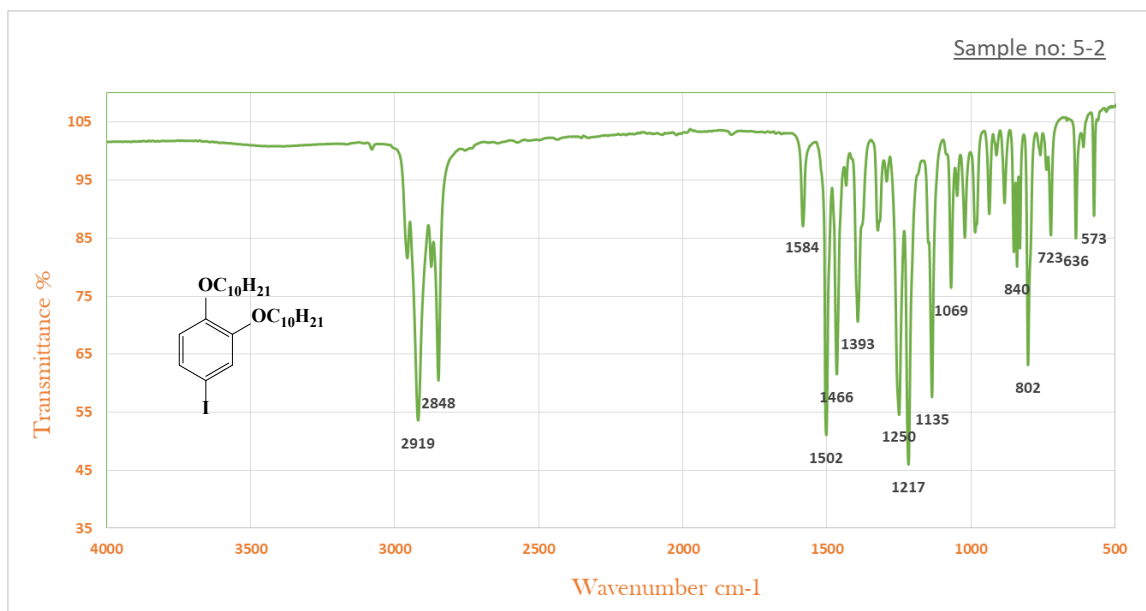


Figure S7. IR spectrum of B2.

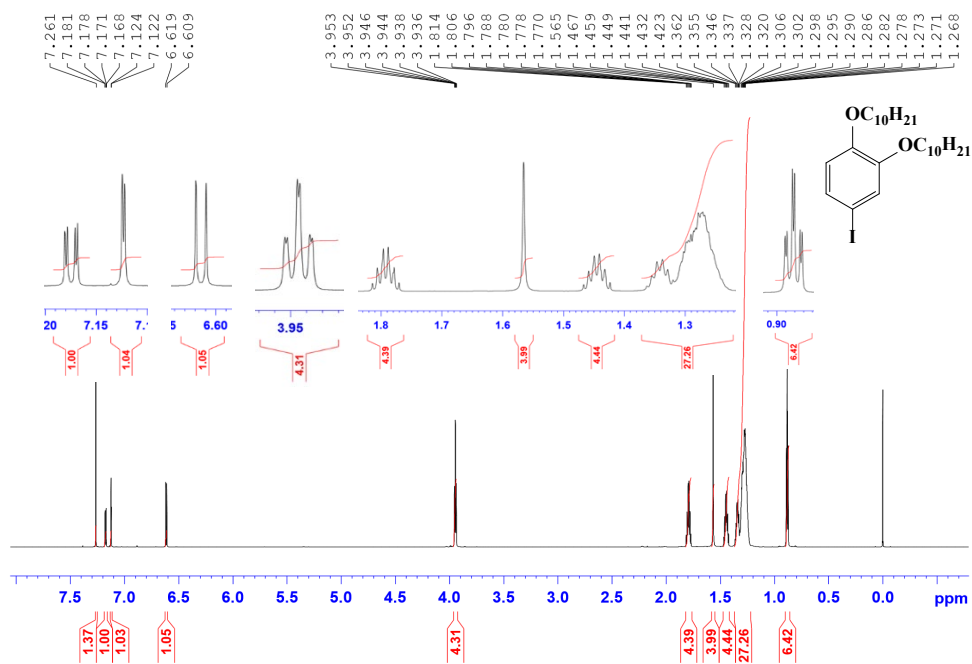


Figure S8. ¹H NMR (CDCl₃) spectrum of **B2**.

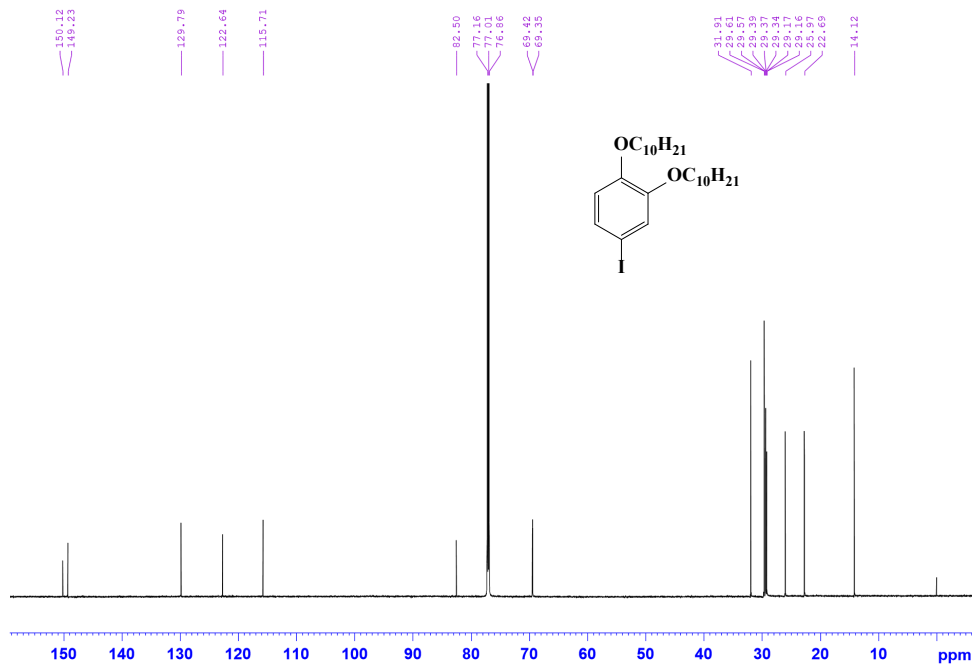


Figure S9. ¹³C NMR (CDCl₃) spectrum of **B2**.

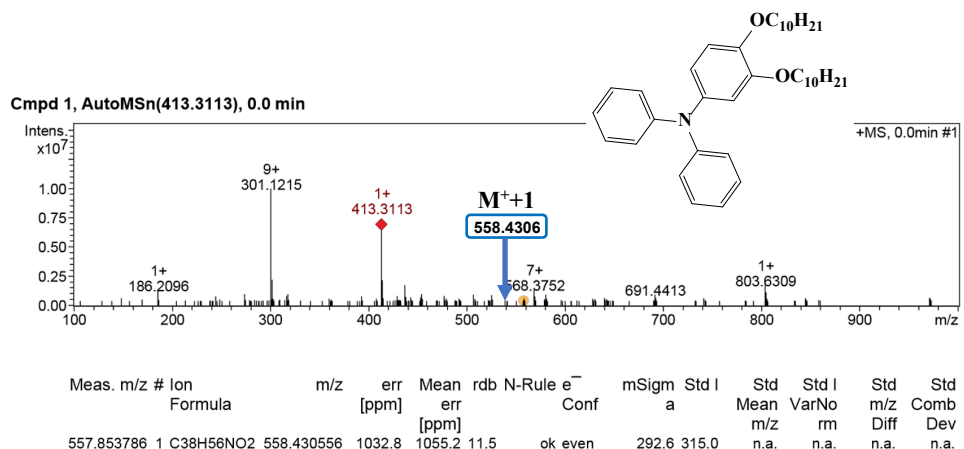


Figure S10. High resolution mass spectrum of **compound 1**.

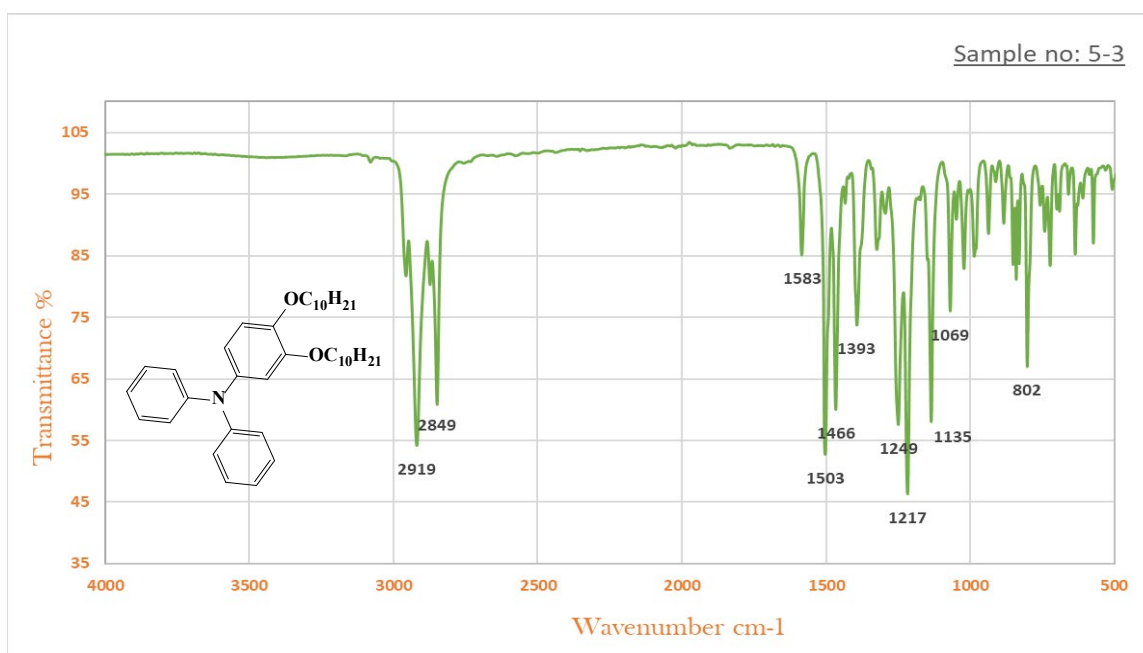


Figure S11. IR spectrum of **compound 1**.

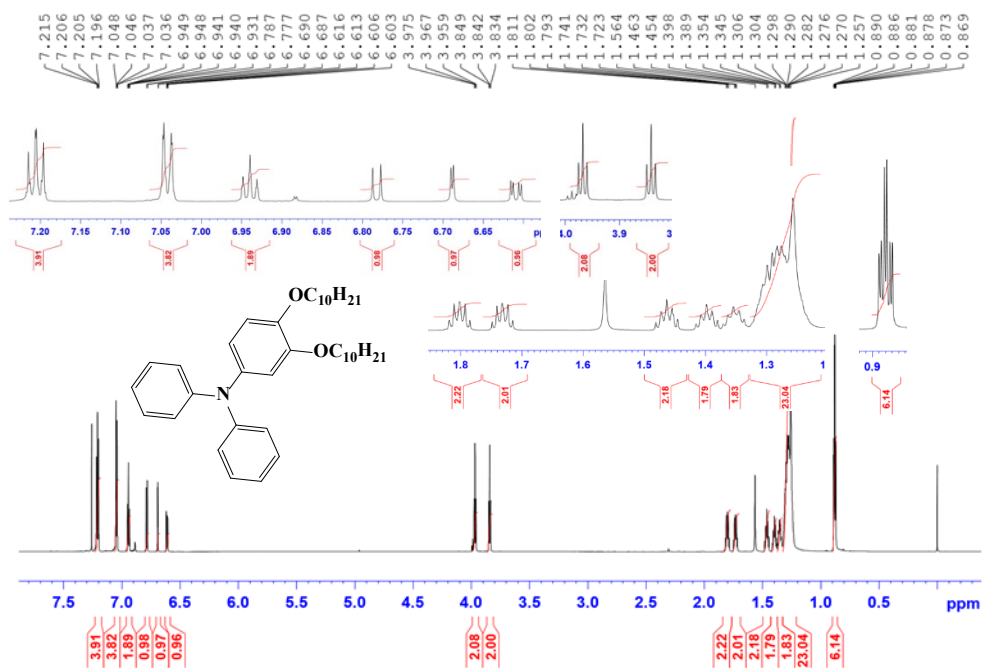


Figure S12. ¹H NMR (CDCl₃) spectrum of **compound 1**.

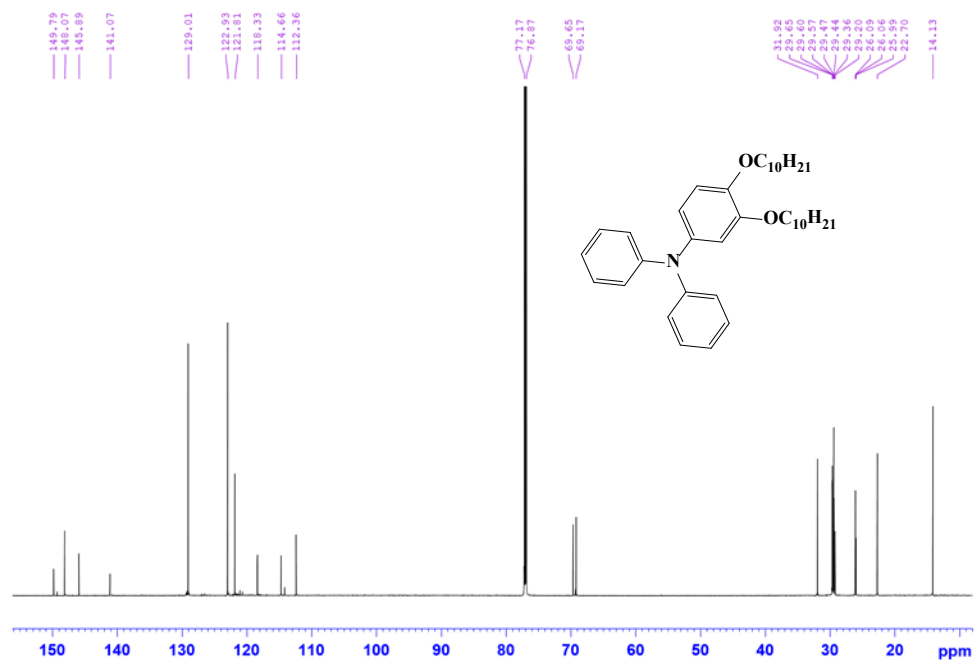


Figure S13. ¹³C NMR (CDCl₃) spectrum of **compound 1**.

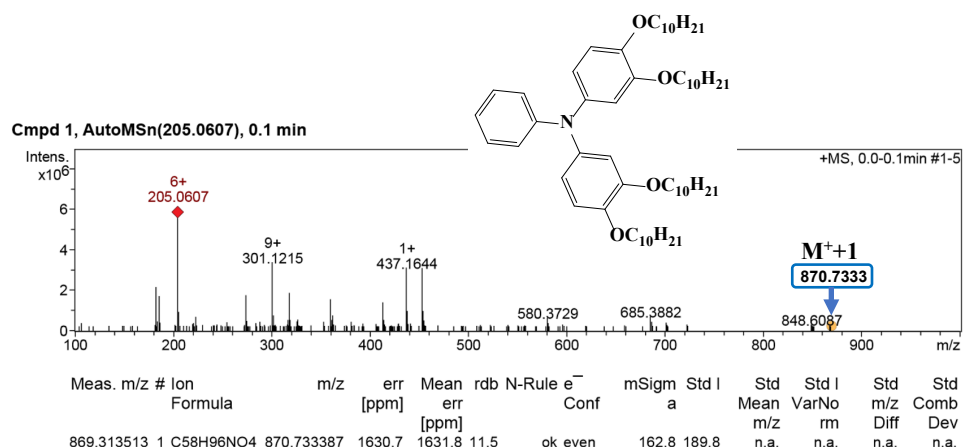


Figure S14. High resolution mass spectrum of compound 2.

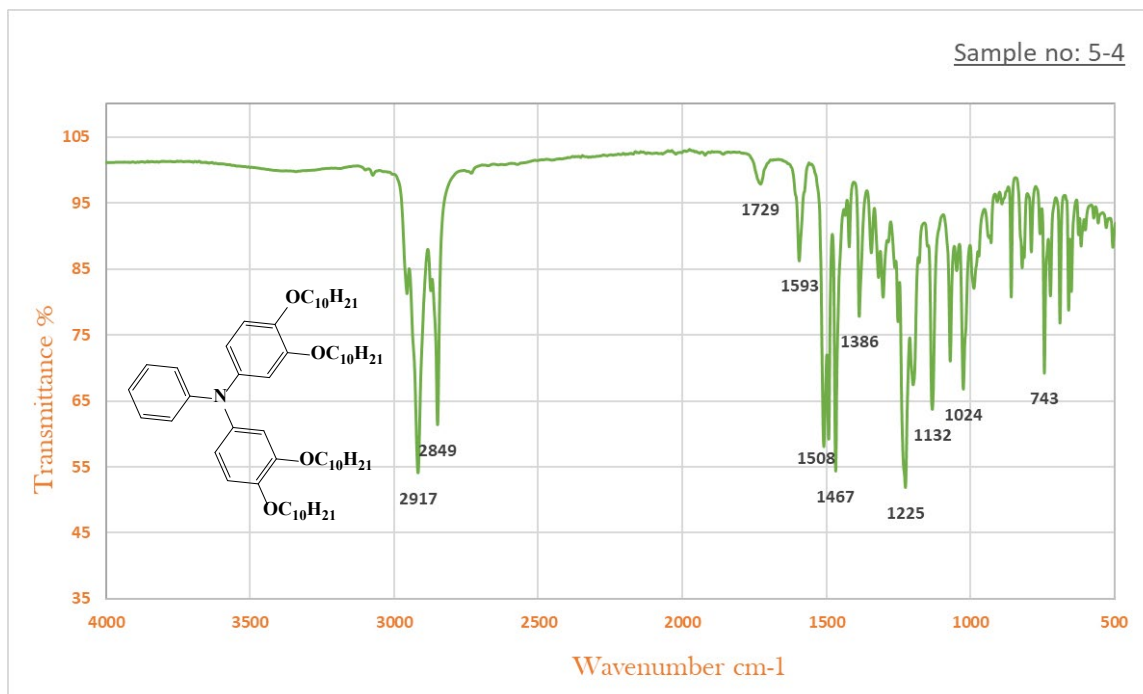


Figure S15. IR spectrum of compound 2.

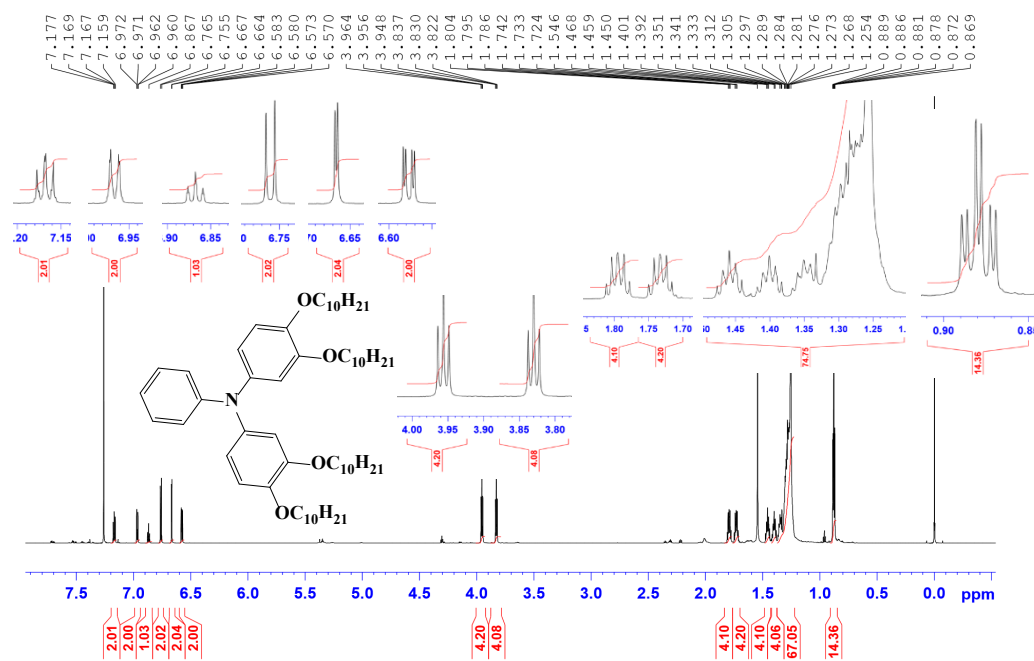


Figure S16. ¹H NMR (CDCl₃) spectrum of compound 2.

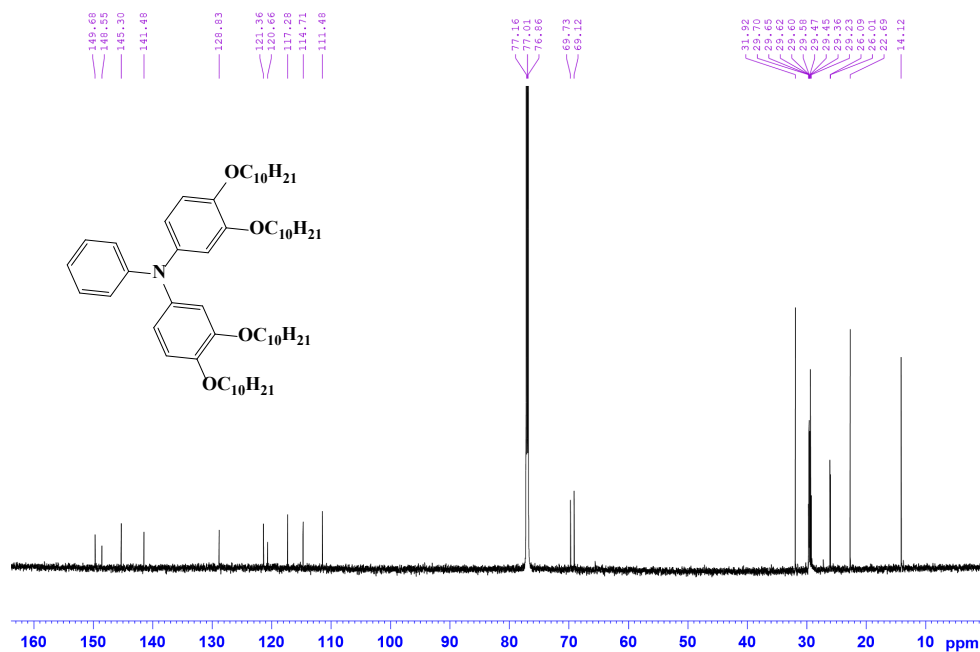


Figure S17. ¹³C NMR (CDCl₃) spectrum of compound 2.

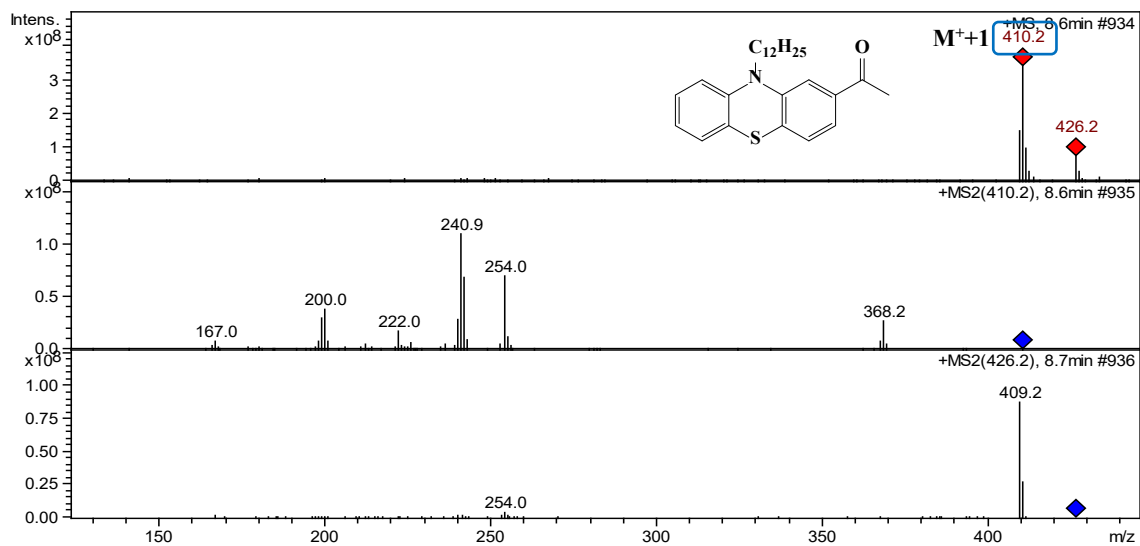


Figure S18. High resolution mass spectrum of A1.

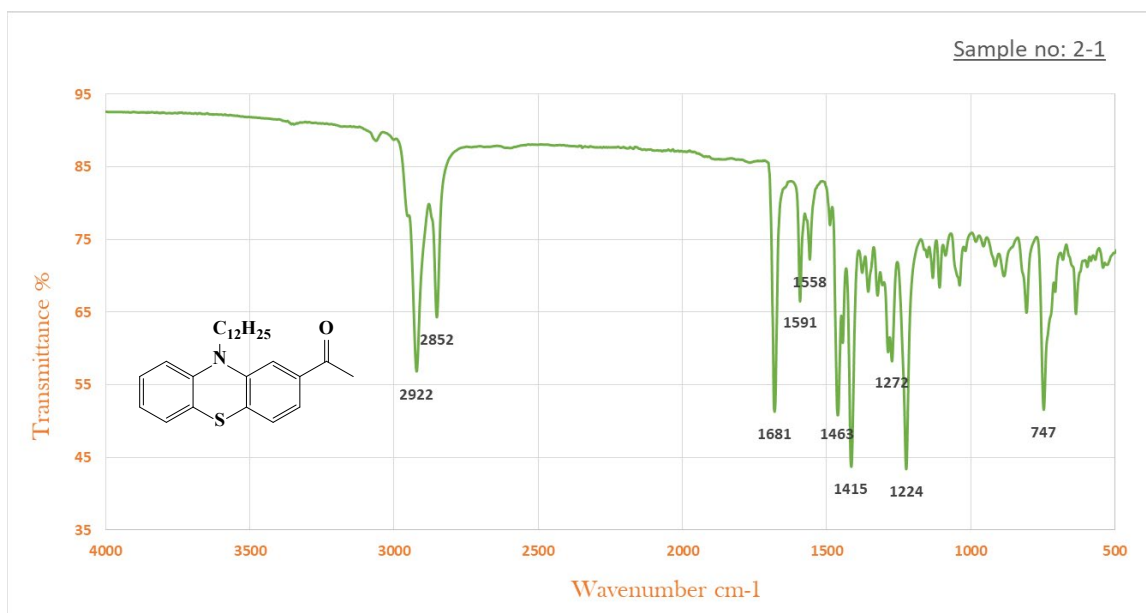


Figure S19. IR spectrum of A1.

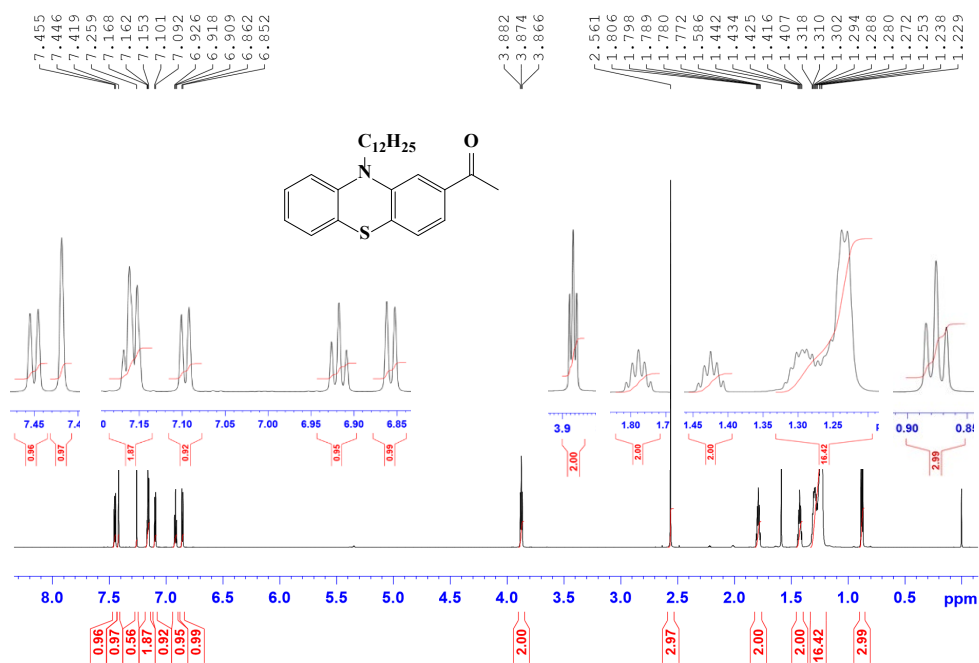


Figure S20. ¹H NMR (CDCl₃) spectrum of A1.

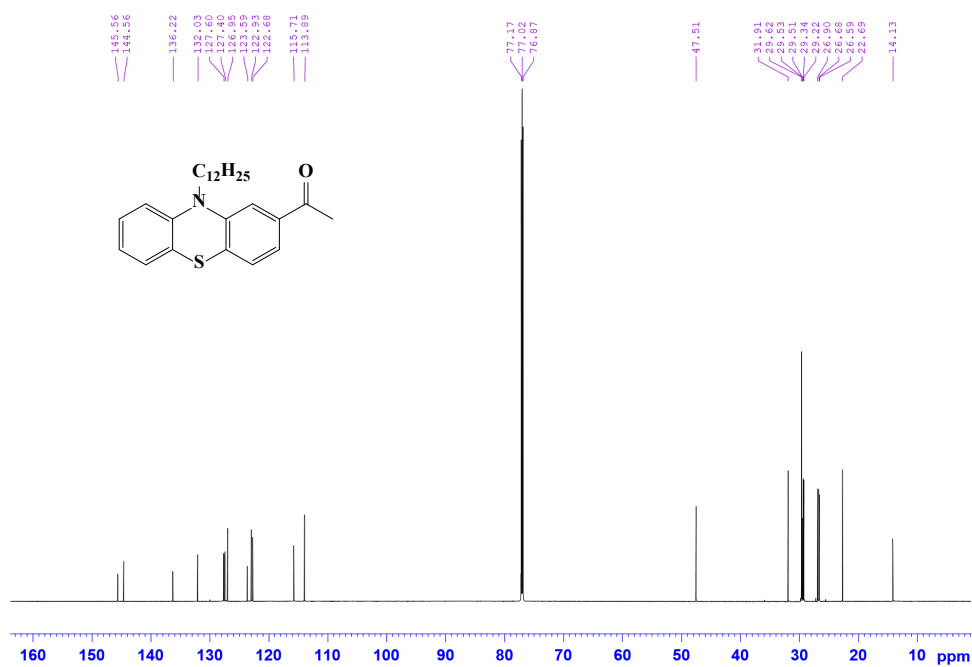


Figure S21. ^{13}C NMR (CDCl₃) spectrum of A1.

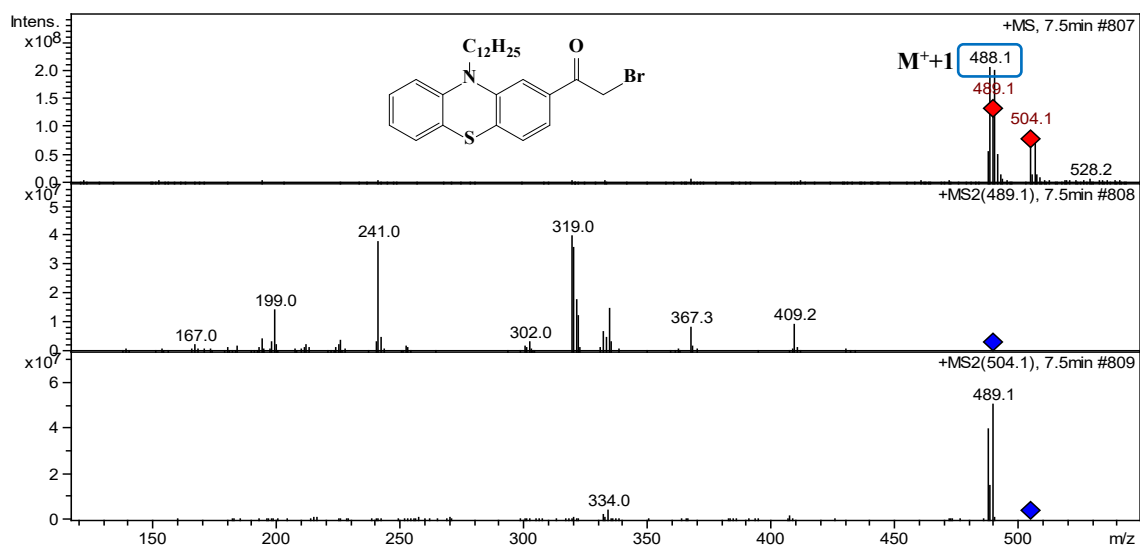


Figure S22. Mass spectrum of A2.

Sample no: 2-2

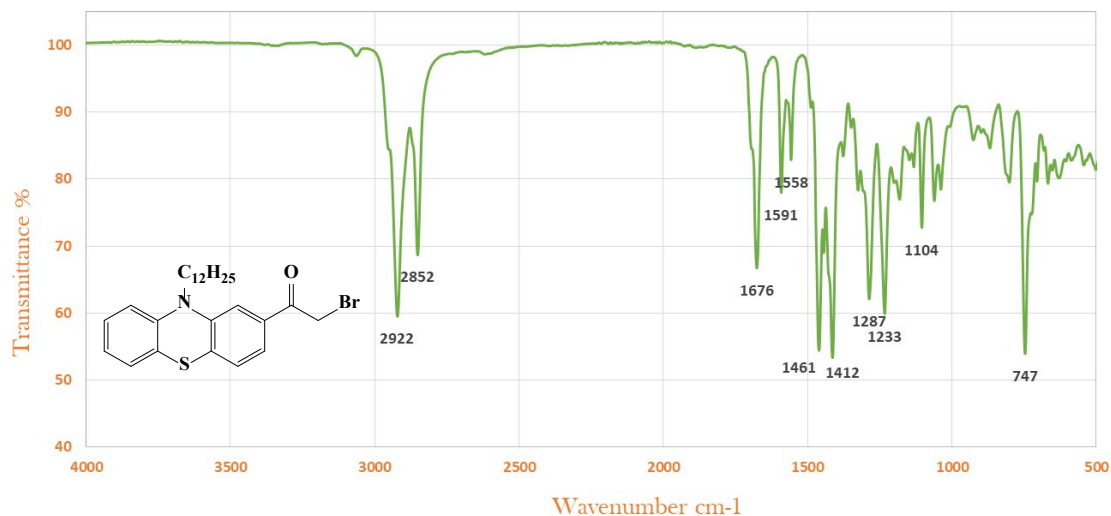


Figure S23. IR spectrum of A2.

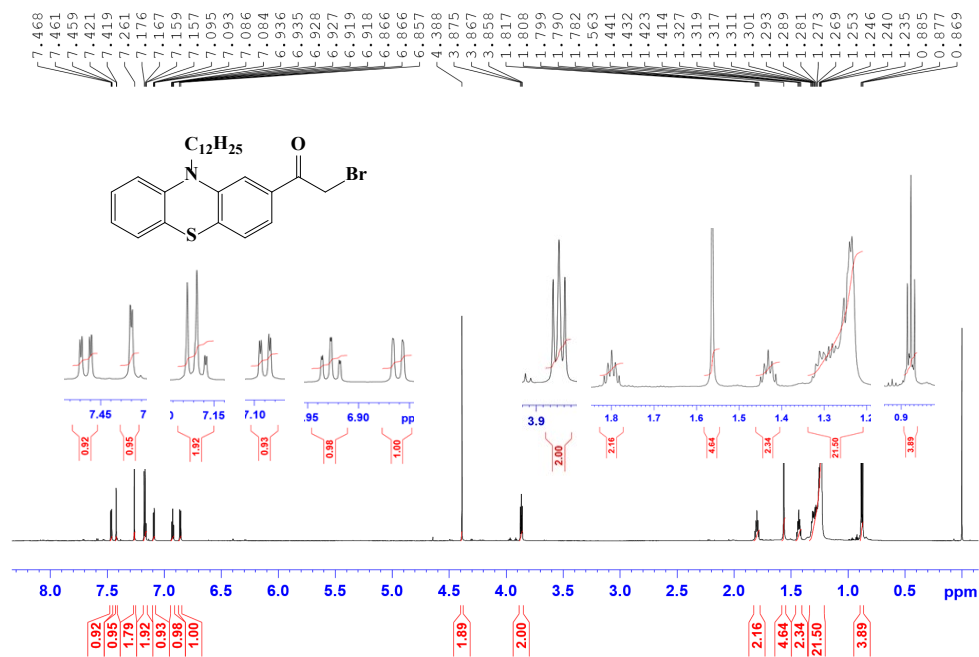


Figure S24. ¹H NMR (CDCl₃) spectrum of A2.

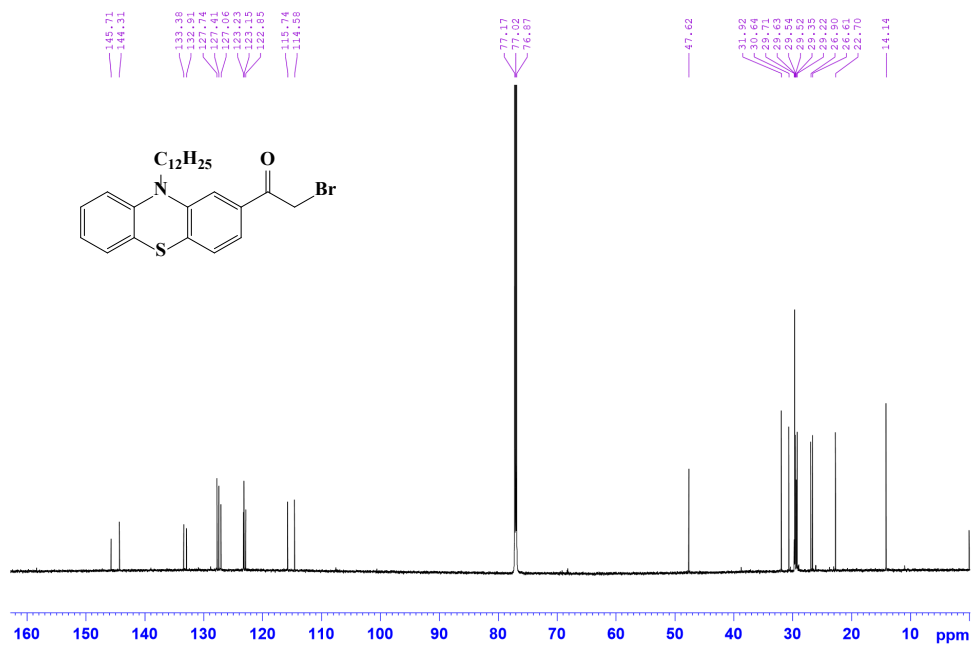


Figure S25. ¹³C NMR (CDCl₃) spectrum of A2.

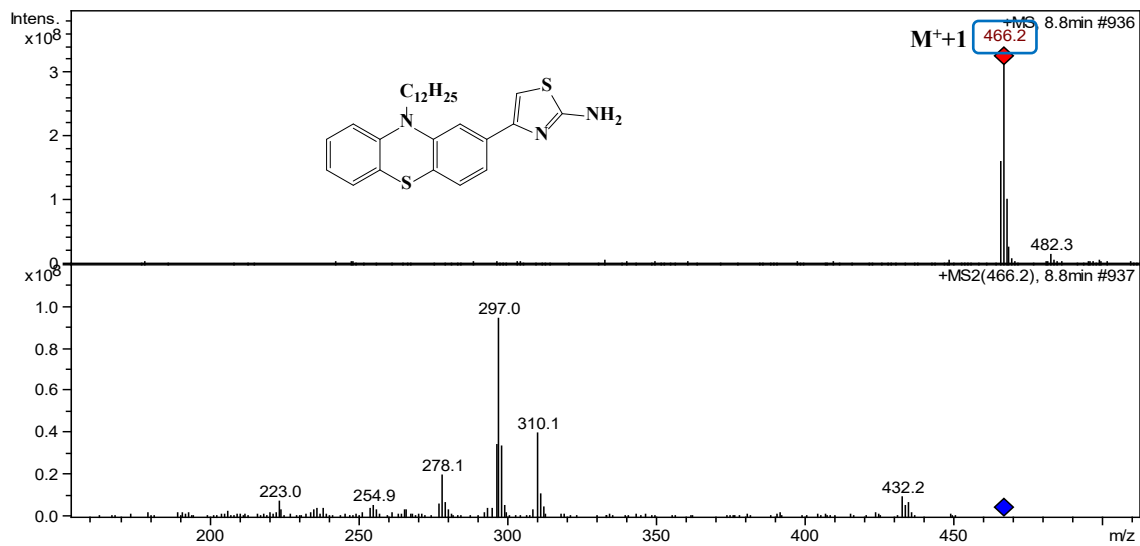


Figure S26. Mass spectrum of A3.

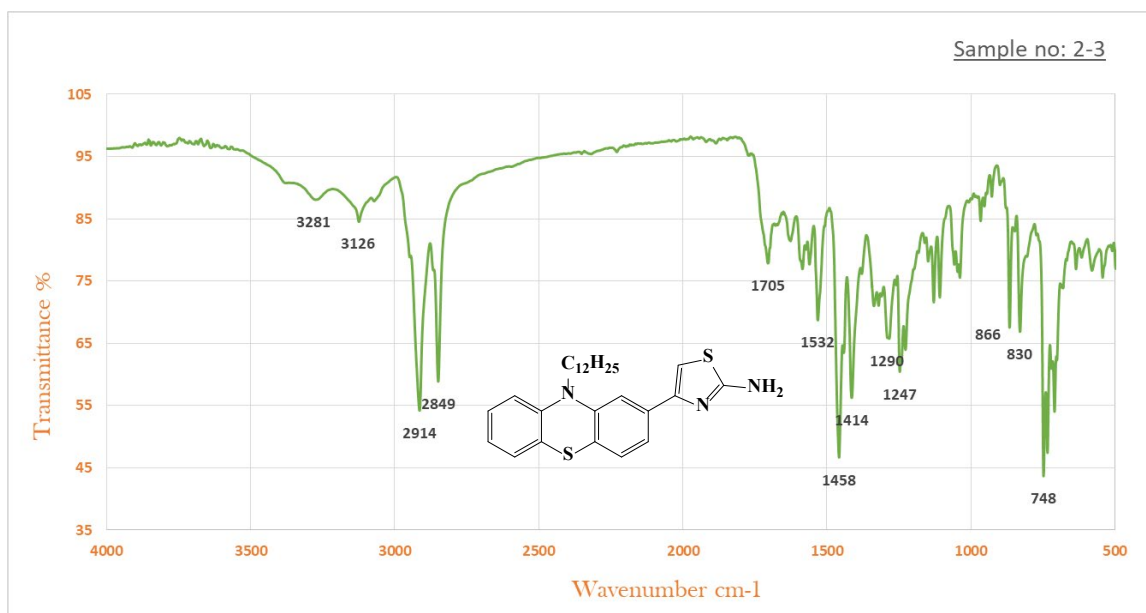


Figure S27. IR spectrum of A3.

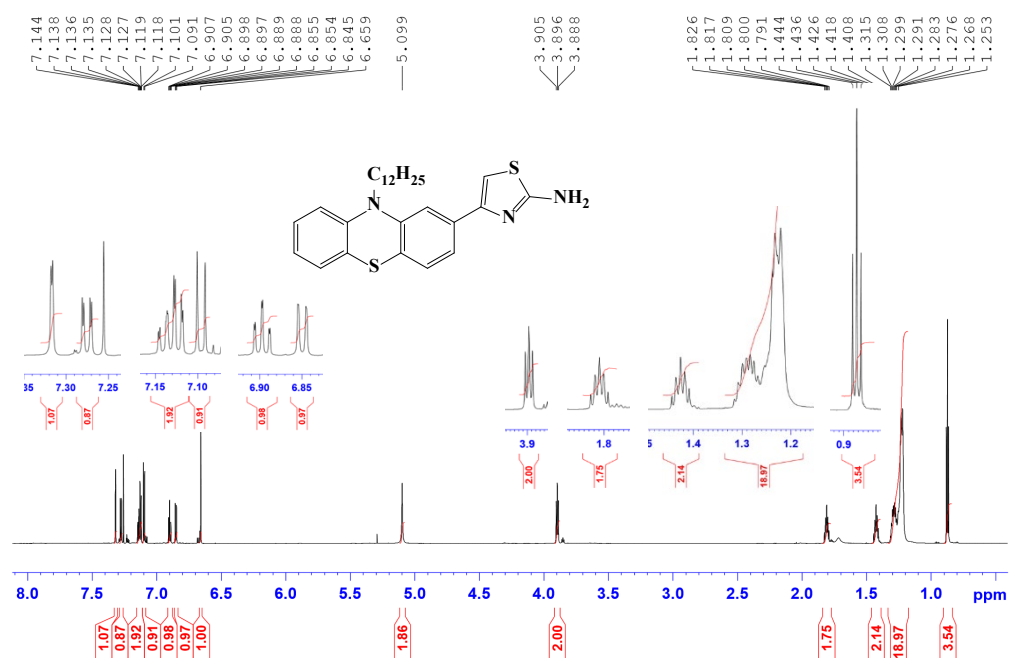


Figure S28. ^1H NMR (CDCl_3) spectrum of A3.

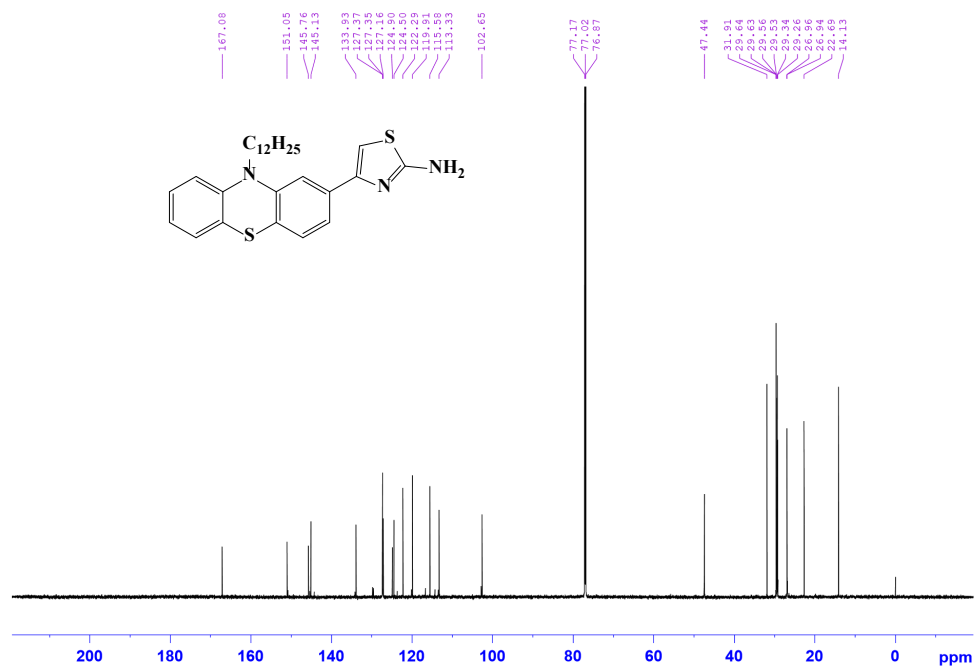


Figure S29. ¹³C NMR (CDCl₃) spectrum of A3.

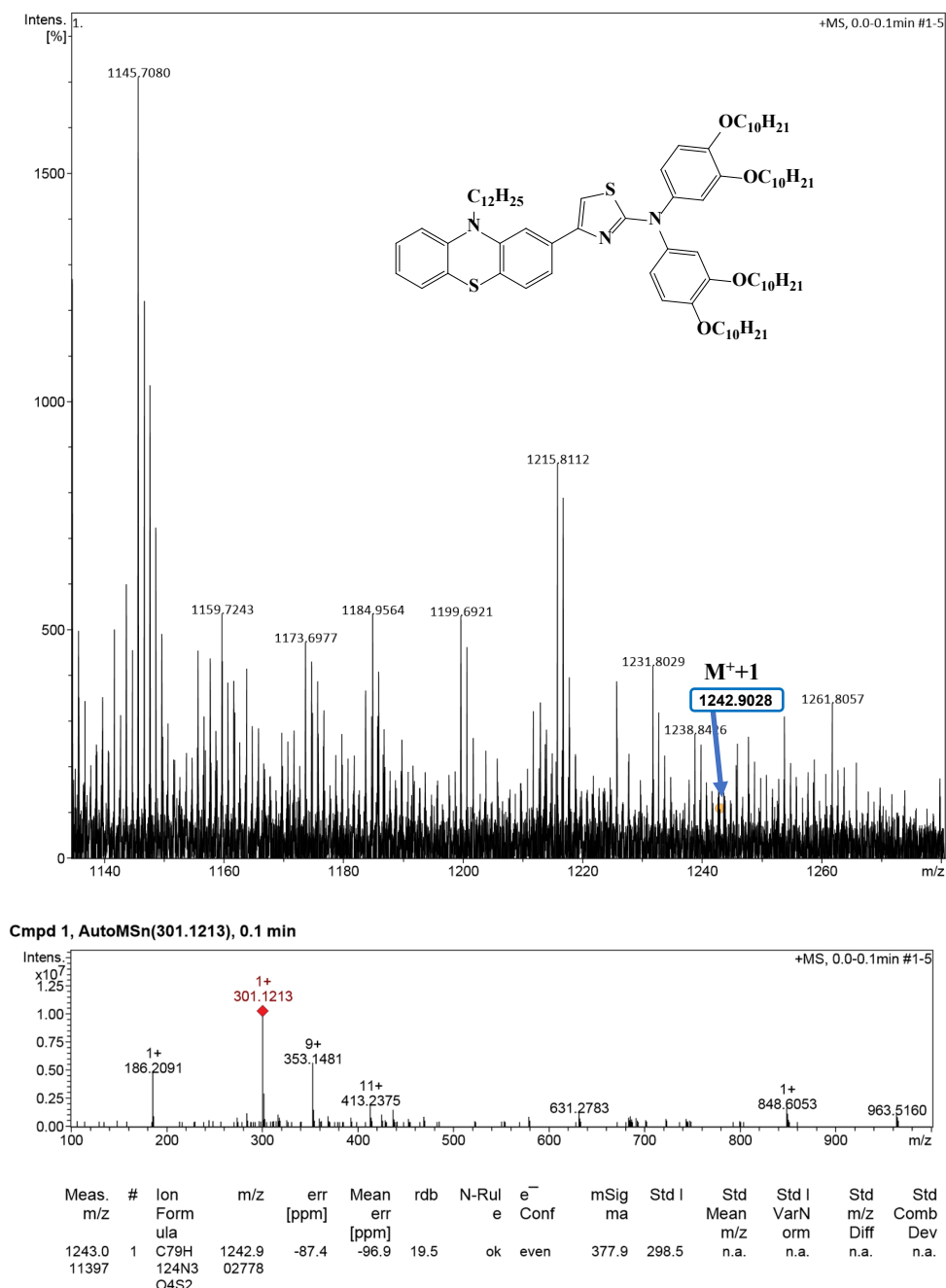


Figure S30. High resolution mass spectrum of **compound 3**.

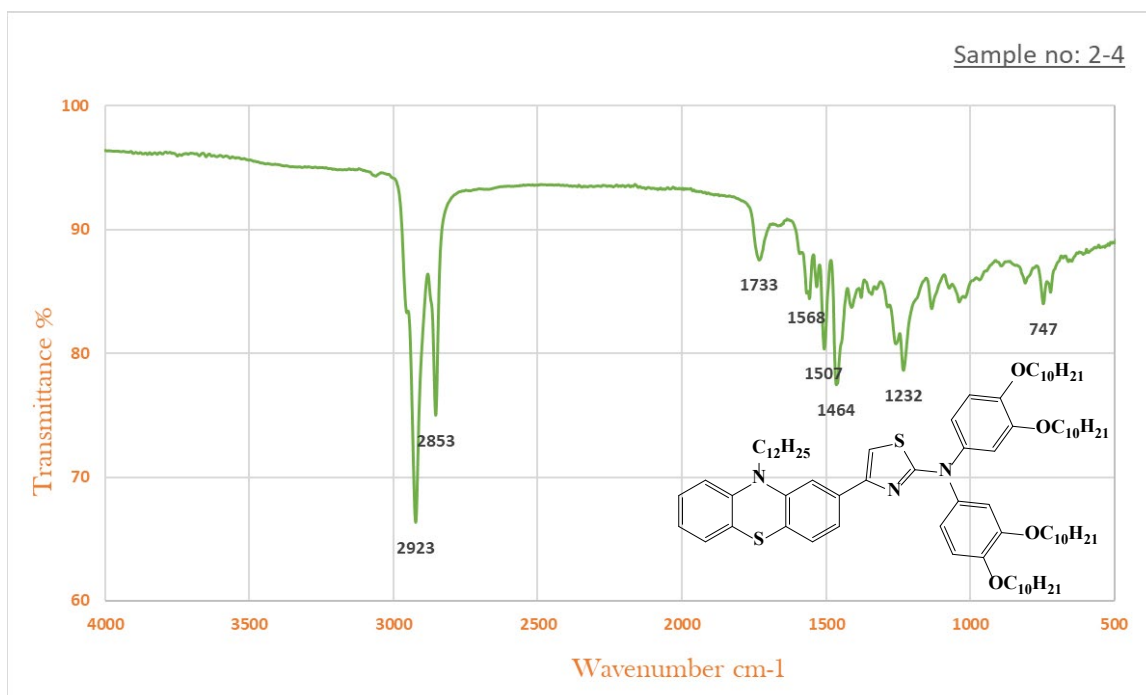


Figure S31. IR spectrum of compound 3.

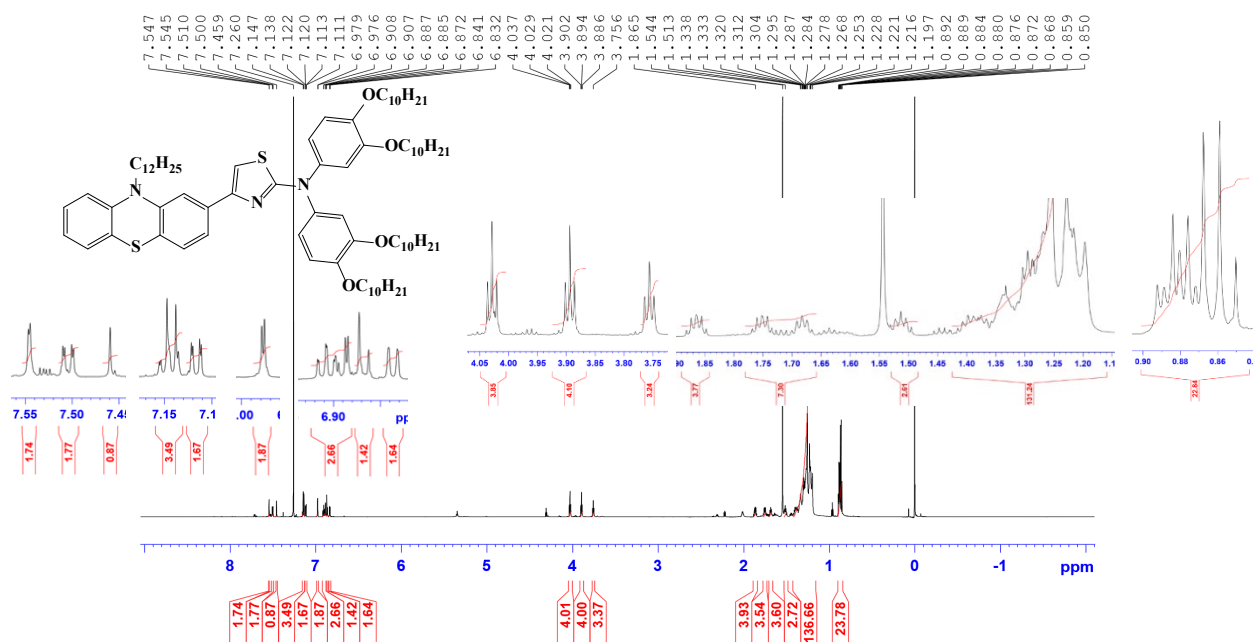


Figure S32. ^1H NMR (CDCl_3) spectrum of compound 3.

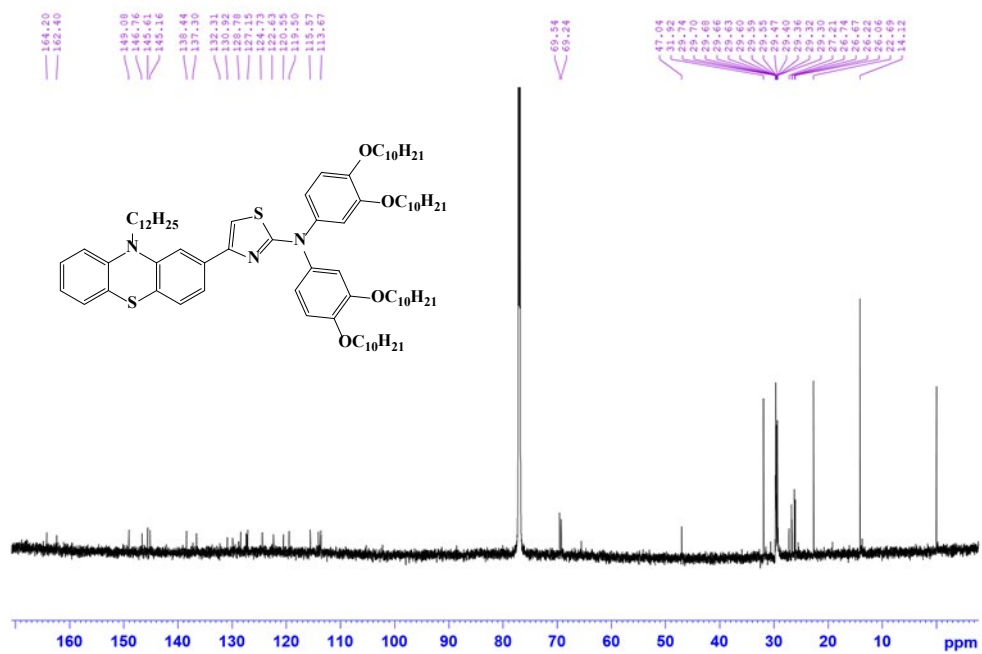


Table S1. UV-Vis Absorption properties of **Compounds 1, 2, and 3.**

Molecules	λ_1	λ_2	λ_3
Compound 1	229	300	-
Compound 2	229	299	-
Compound 3	229	256	292 (Shoulder)

Table S2. Bond lengths and dihedral angles of three compounds

	Compound 1	Compound 2	Compound 3
Bond lengths (Å)			
L ₁ -L ₂	1.4246	1.4231	1.4360
L ₁ -L ₃	1.4215	1.4201	1.3796
L ₁ -L ₄	1.4201	1.4234	1.4329
L ₅ -L ₆	/	/	1.4767
Dihedral angles (°)			
ϕ_1	41.6702	42.1668	16.3468
ϕ_2	39.8371	45.1419	46.0301
ϕ_3	46.2855	44.2488	60.1911
ϕ_4	/	/	6.9894

Table S3. Contribution with Partial density of states analysis.

Molecules	Orbitals	A	B	C	N	D
Compounds 1	HOMO	29%	22%	22%	26%	-
	LUMO	33%	25%	41%	0%	-
Compounds 2	HOMO	28%	27%	21%	25%	-
	LUMO	15%	29%	55%	1%	-
Compounds 3	HOMO	4%	6%	24%	7%	60%
	LUMO	3%	2%	44%	1%	50%

Table S4. The Vertical excitation energy (cm⁻¹), Absorption peak (nm), Oscillator strengths (*f*), Electron transitions and excited-state lifetime τ_2 (ns) of three investigated molecules.

Molecules	State	Energy	λ	<i>f</i>	CI	τ_2
Compound 1	S1	36013.97342	277.67	0.2767	H- L/0.66564	4.17686
Compound 2	S1	35953.11714	278.14	0.3412	H- L/0.64282	3.39875
Compound 3	S1	33034.91791	302.71	0.1619	H- L/0.57662	8.48415