



Supplementary Materials Ultra-Highly Efficient Removal of Methylene Blue Based on Graphene Oxide/TiO₂/Bentonite Sponge

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Figure S1. FTIR spectra of GO, 10% GO/Bent and 10% GO/TiO₂/Bent sponges.



Figure S2. Photodegradation of MB by GO/TiO₂/Bent sponge with different GO proportions under simulated sunlight irradiation after 30 min dark adsorption.

The effect of the graphene oxide (GO) content has been investigated on the photocatalytic efficiency of sponges under simulated sunlight irradiation for degradation of MB. It can be seen from Figure S1 that 10% GO/TiO₂/Bent exhibits the highest photocatalytic efficiency. 5% GO/TiO₂/Bent presents the lowest efficiency compared with others. This is probably because the amount of GO and Bent is too low to promote effectively the photocatalytic capacity of TiO₂. 15% GO/TiO₂/Bent doesn't present the best probably due to the blockage of light by GO and Bent to TiO₂ surface [1]. Zhang et al. investigated MB degradation on GO/TiO₂ and also found that the higher addition of graphene in GO/TiO₂ leads to a decreased photocatalytic activity [2].



Figure S3. The N₂ adsorption-desorption isotherms of TiO₂, GO, TiO₂/Bent, 10% GO/TiO₂ and 10% GO/TiO₂/Bent.

Table S1. The BET surface area of G	O, TiO2,	TiO ₂ /Bent,	10% GO/TiO	2 and 10% G	O/TiO2/Bent.
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Sample	BET Surface Area (m ² /g)			
TiO ₂	17.80			
GO	40.07			
TiO ₂ /Bent	19.96			
10% GO/TiO2	19.18			
10% GO/TiO ₂ /Bent	33.59			

Our study found that BET values of each catalyst are not related to the photocatalytic dye degradation activity. In Figure S2 and Table S1, we observed the order of BET surface area values from big to small is GO > 10% GO/TiO₂/Bent > TiO₂/Bent > 10% GO/TiO₂ > TiO₂. In Figure 6, the order of photocatalytic dye degradation activity from high to low is 10% GO/TiO₂/Bent > 10% GO/TiO₂ > TiO₂/Bent > TiO₂/Bent > TiO₂/Bent > TiO₂/Bent > 10% GO/TiO₂/Bent > 10% GO/TiO₂ > TiO₂.

References

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