

Supplementary

Enhancement of Titania Photoanode Performance by Sandwiching Copper between Two Titania Layers

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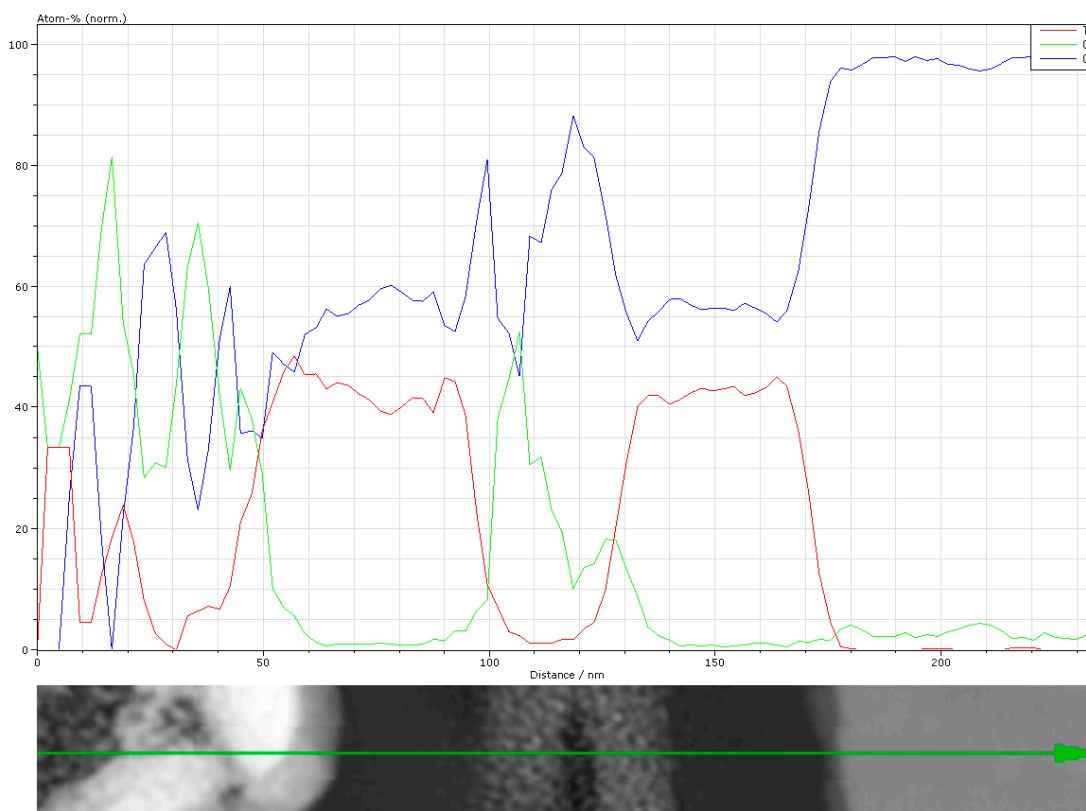


Figure S1. the atomic dispersion of TCT-4 from the line scanning model.

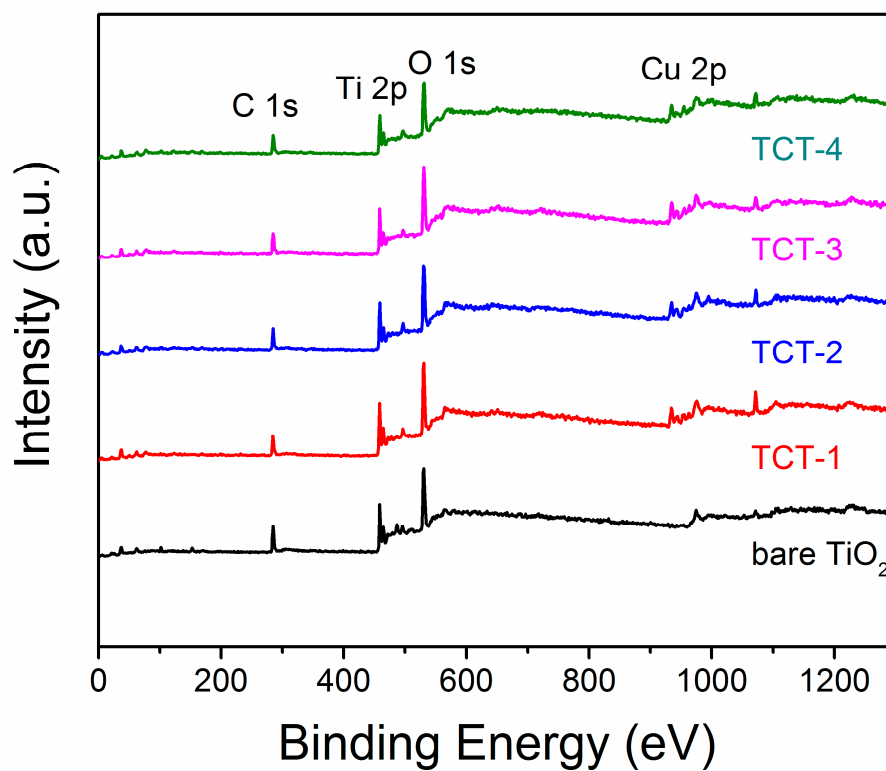


Figure S2. XPS survey spectra of bare TiO₂ and all TCT samples respectively.

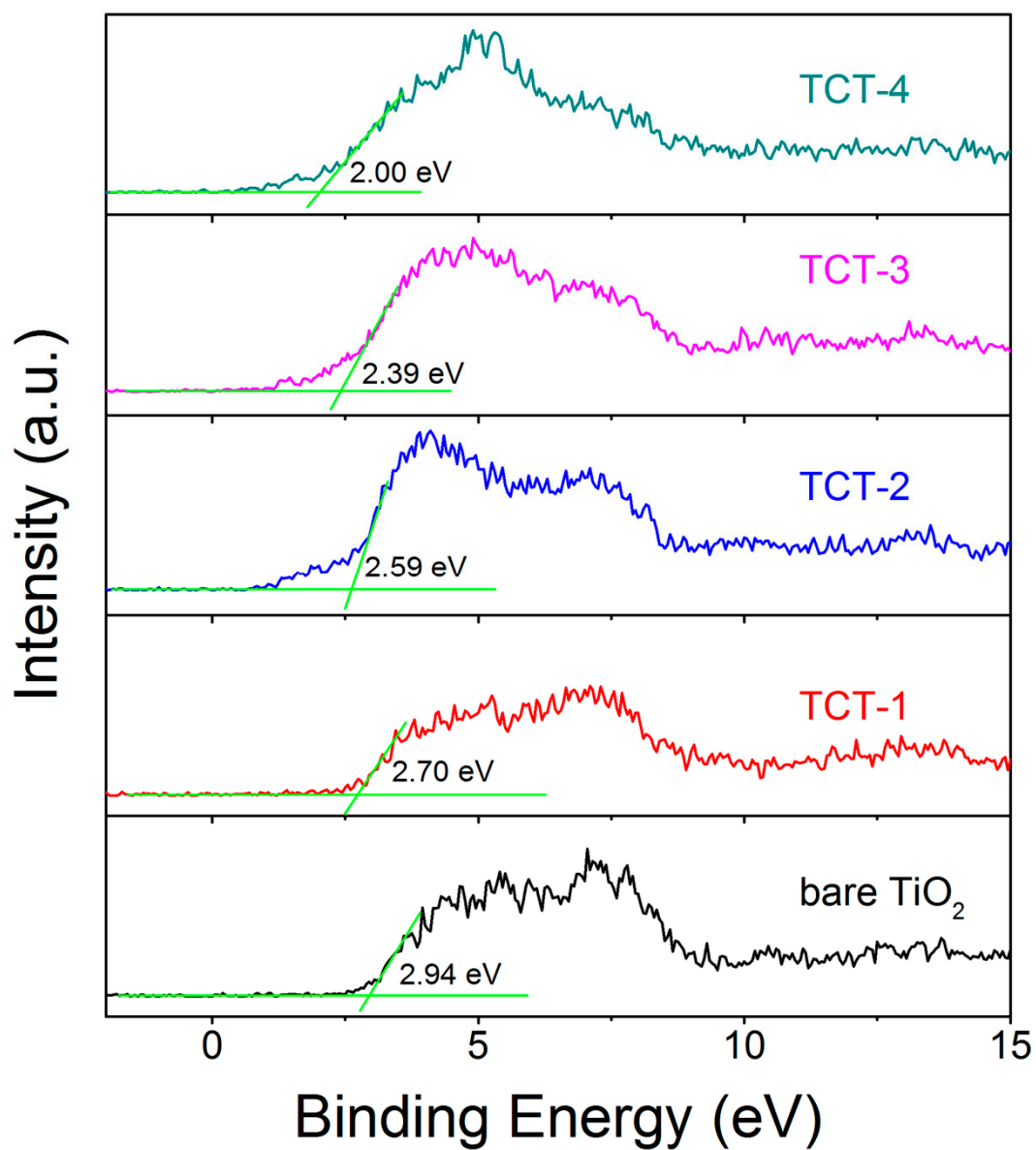


Figure S3. XPS Valence band spectra of bare TiO_2 and all TCT samples respectively.

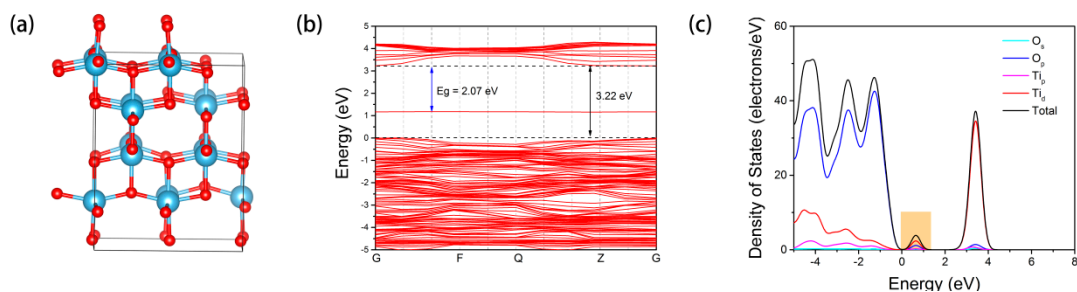


Figure S4. The lattice structure, calculated band structure and density of states (DOS) of $\text{TiO}_2\text{-Vo}$.

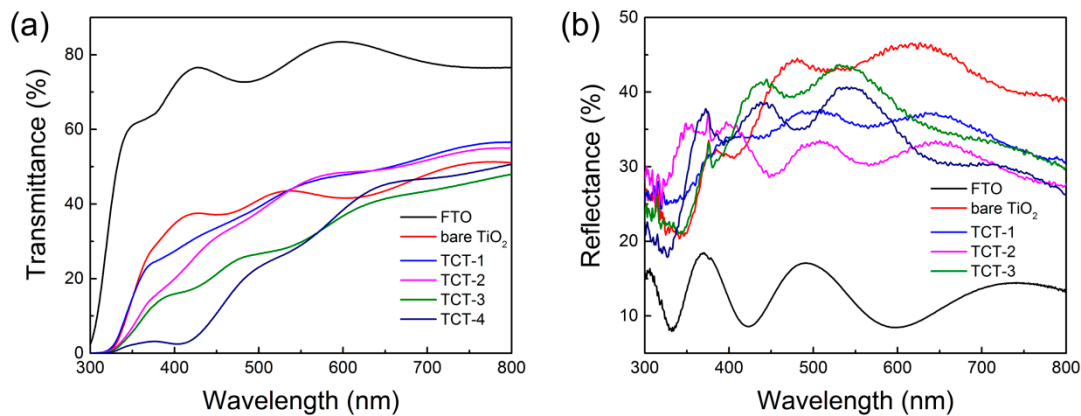


Figure S5. The transmittance (a) and reflectance; (b) of five samples from UV-vis measurement.

The absorbance spectra was calculated by the Equation:

$$A = 1 - T - R \quad (1)$$

Calculations

The IPCE can be calculated by the Equation:

$$\text{IPCE}(\%) = (1240 \times J_{ph}) / (\lambda \times P_{in}) \quad (2)$$

where J_{ph} is the measured photocurrent density of the experimental sample, λ is the wavelength and P_{in} is the intensity of the incident light.

Mott-schottky Equation:

$$\frac{1}{C^2} = \frac{2}{N_D e \epsilon_0 \epsilon} (E - E_{FB} - \frac{kT}{e}) \quad (3)$$

Where C is the capacitance, N_D is the charge carrier density, e is the elemental charge, ϵ is the relative permittivity of material, ϵ_0 is vacuum permittivity, E is the applied potential in electrochemical measurement, E_{FB} is the flat band potential, k is the Boltzman constant and T is measurement temperature.



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