Supporting Information:

Magnetic, Structural and Spectroscopic Properties of Iron(II)-Octacyanoniobate(IV) Crystalline Film Obtained by Ion-Exchange Synthesis

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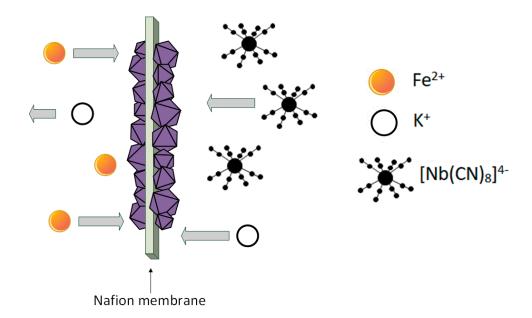
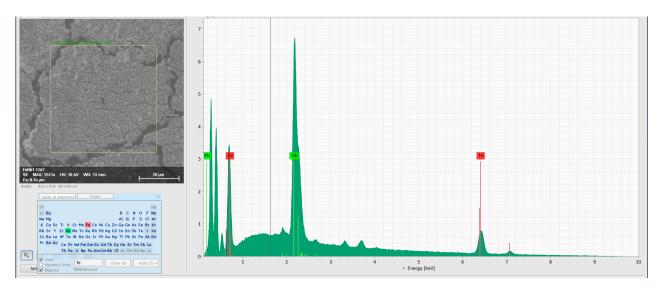


Figure S1. Schematic illustration of the growth process of $[Fe^{II}(H_2O)_2]_2[Nb^{IV}(CN)_8]\cdot 4H_2O$ crystalline film during ion exchange synthesis, where a transparent Nafion membrane is used as a solid substrate.



 $\label{eq:Figure S2} \textbf{Figure S2}. \ \ \text{Energy dispersive x-ray spectroscopy (EDS) analysis data of } \ [\text{Fe}^{II}(\text{H2O})_2]_2[\text{Nb}^{IV}(\text{CN})_8] \cdot 4\text{H2O}$ film deposited on the Nafion membrane during ion- exchange synthesis with the time 60s.

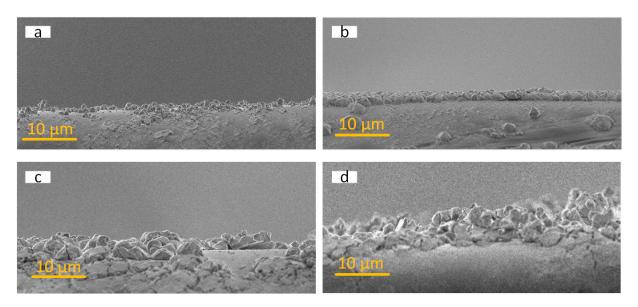


Figure S3. The cross section of the $[Fe^{II}(H_2O)_2]_2[Nb^{IV}(CN)_8]\cdot 4H_2O$ film deposited on the Nafion membrane during ion-exchange synthesis with the time (a) 10s, (b) 30s, (c) 60s and (d) 180s.