

SUPPORTING INFORMATION

Communication

Exploring the Halogen-Bonded Cocrystallization Potential of a Metal-Organic Unit Derived from Copper(II) Chloride and 4-Aminoacetophenone

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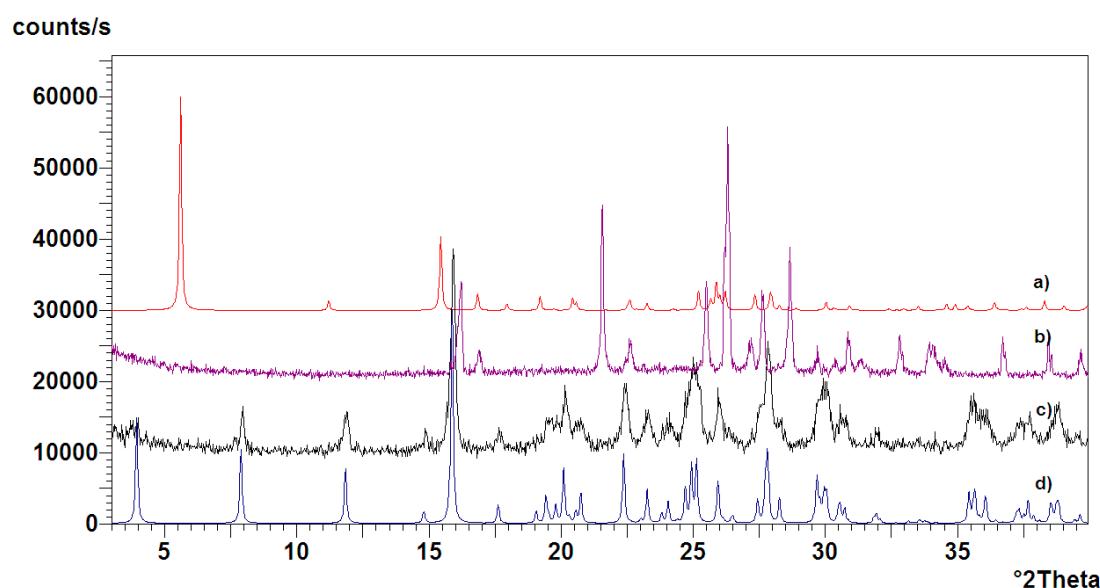


Figure S1. PXRD patterns of **a)** $\text{CuCl}_2(\text{aap})_2$, **b)** **14tfib**, **c)** product obtained by grinding $\text{CuCl}_2(\text{aap})_2$ and **14tfib** in a 1:1 stoichiometric ratio, **d)** calculated pattern from $[\text{CuCl}_2(\text{aap})_2](\text{14tfib})$ single crystal data.

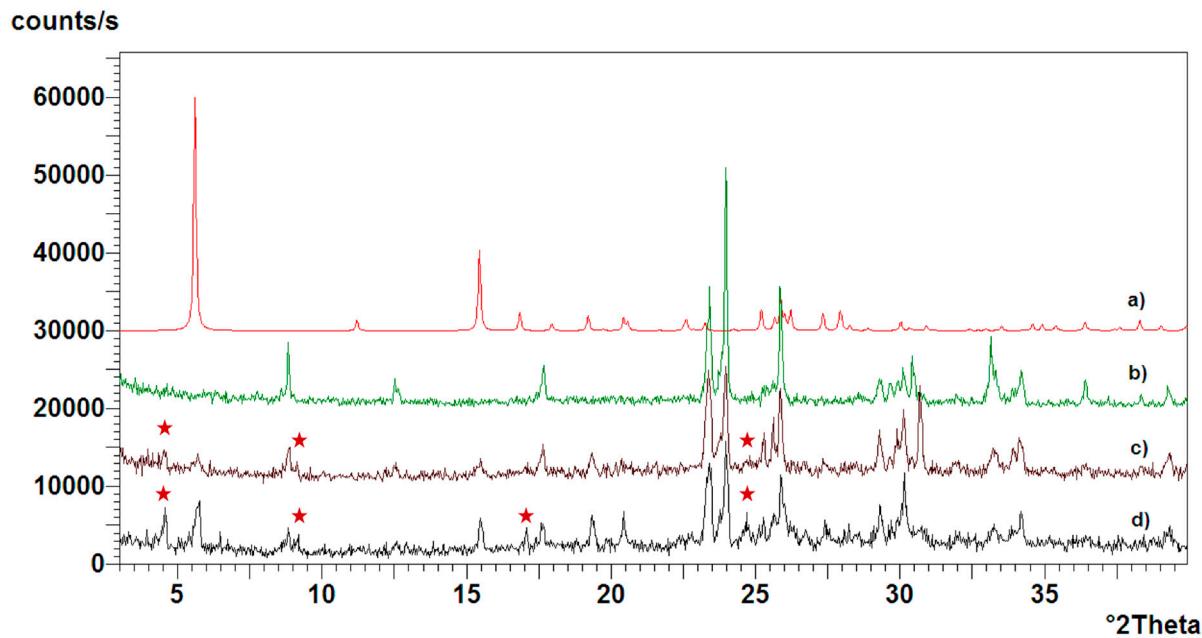


Figure S2. PXRD patterns of a) $\text{CuCl}_2(\text{aap})_2$, b) **12tfib**, c) product obtained by grinding $\text{CuCl}_2(\text{aap})_2$ and **12tfib** in a 1:2 stoichiometric ratio, d) product obtained by grinding $\text{CuCl}_2(\text{aap})_2$ and **12tfib** in a 1:1 stoichiometric ratio. Stars denote small peaks not belonging to either reactant.

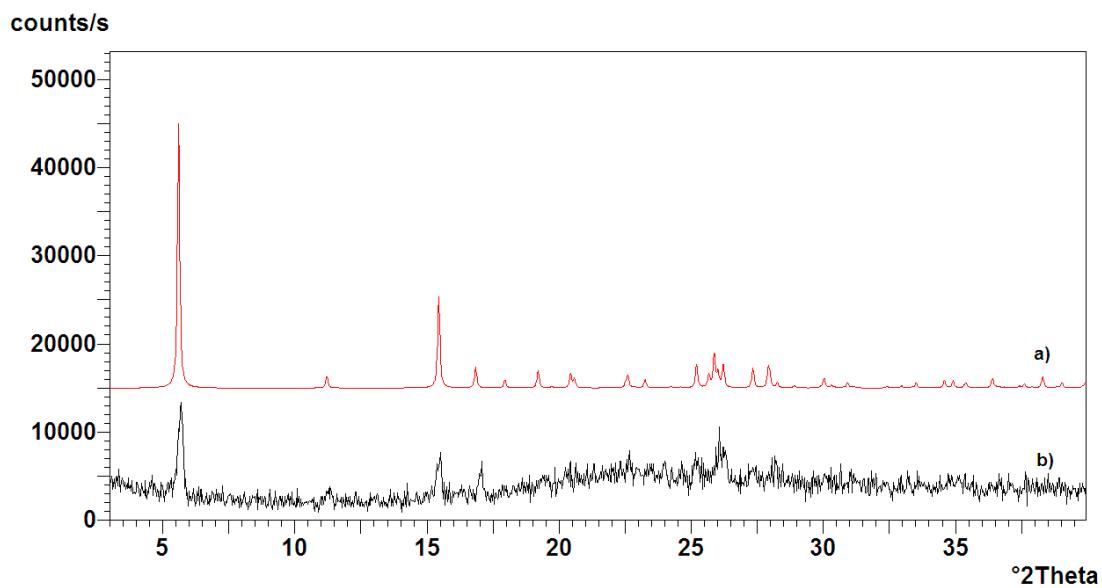


Figure S3. PXRD patterns of a) $\text{CuCl}_2(\text{aap})_2$, b) product obtained by grinding $\text{CuCl}_2(\text{aap})_2$ and **13tfib** in a 1:1 stoichiometric ratio. The other reagent (**13tfib**) is a liquid, so its PXRD pattern is omitted.

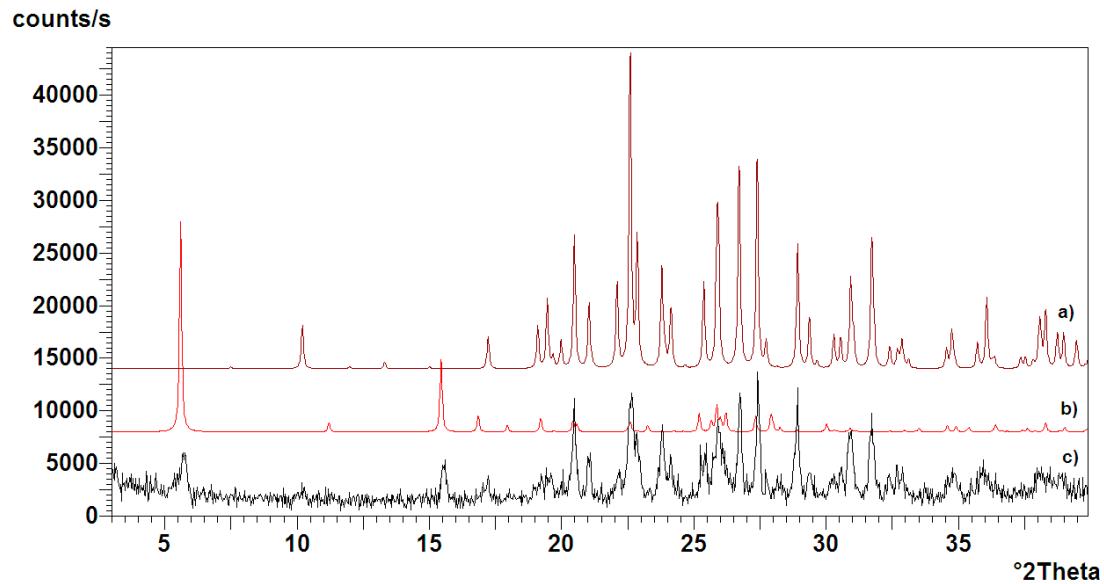


Figure S4. PXRD patterns of a) **135tfib**, b) $\text{CuCl}_2(\text{aap})_2$, c) product obtained by grinding $\text{CuCl}_2(\text{aap})_2$ and **135tfib** in a 1:1 stoichiometric ratio.

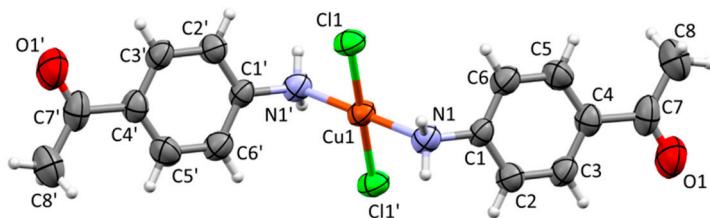


Figure S5. Molecular structure of $\text{CuCl}_2(\text{aap})_2$ showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50 % probability level, and H atoms are shown as small spheres of arbitrary radius. Symmetry codes of symmetry equivalent atoms marked with an ' symbol are listed in Table S4.

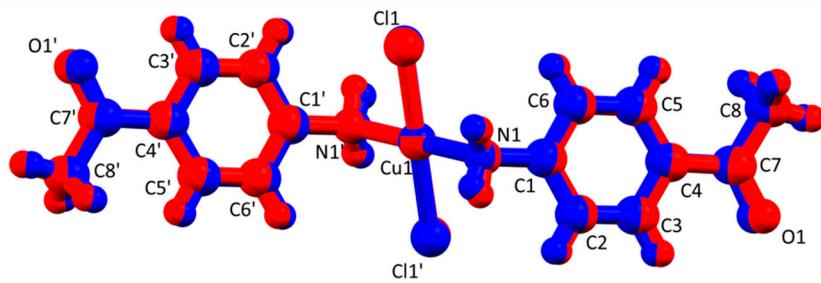
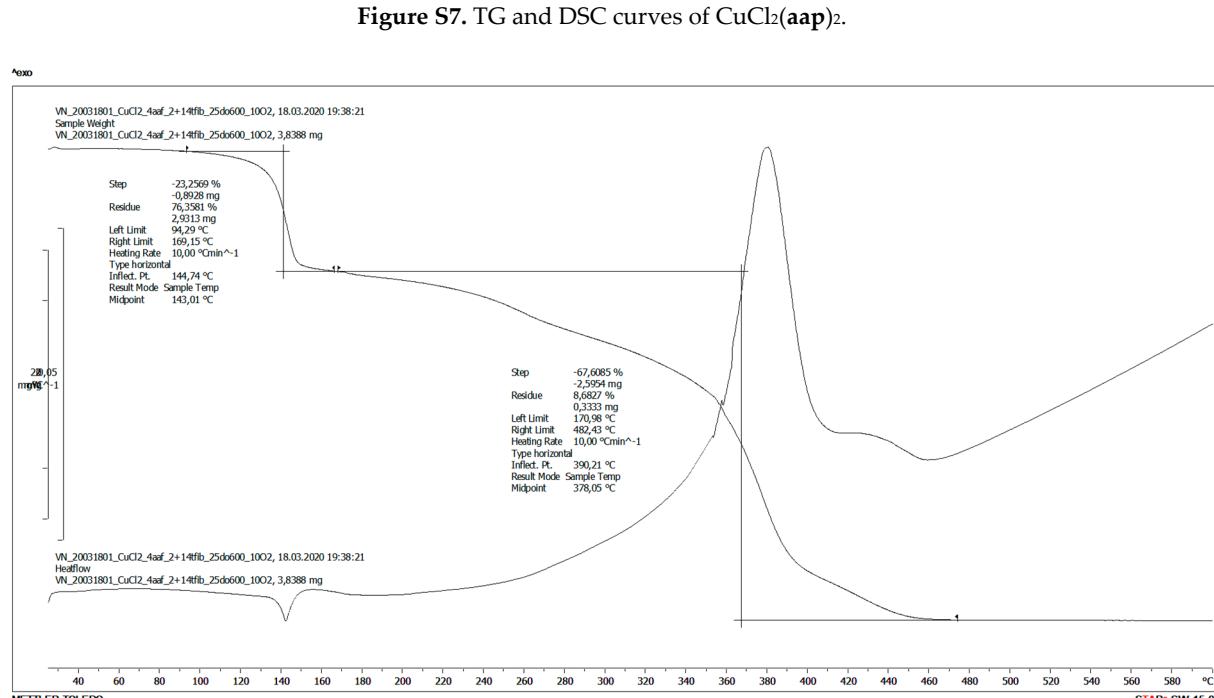
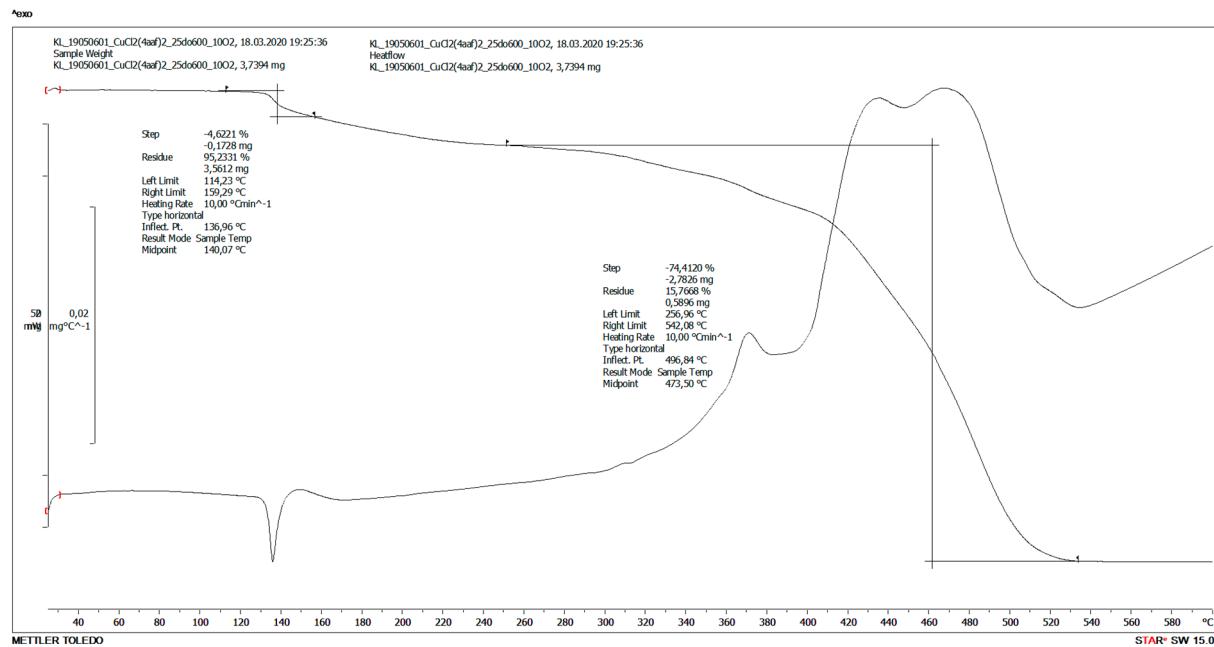


Figure S6. Molecular overlay of the metal complex molecule obtained from $\text{CuCl}_2(\text{aap})_2$ data (colored red) with the metal complex molecule obtained from $[\text{CuCl}_2(\text{aap})_2](\text{14tfib})$ data (colored blue). The metal complex structures are in good agreement with a root mean square deviation (RMSD) value of 0.1418 and a maximum distance of 0.2619.

**Table S1.** Mechanochemical Synthesis Parameters.

Reactants	Mole Ratio	$t_{\text{milling}} = 30 \text{ min}, v_{\text{milling}} = 25 \text{ Hz}$		
		$m(\text{CuCl}_2(\text{aap})_2) / \text{mg}$	$m \text{ or } V(\text{donor})$	Liquid
$\text{CuCl}_2(\text{aap})_2 : \text{14tfib}$	1 : 1	99.6	100.0 mg	40.0 μL acetonitrile
$\text{CuCl}_2(\text{aap})_2 : \text{12tfib}$	1 : 2	67.0	133.0 mg	40.0 μL acetonitrile
$\text{CuCl}_2(\text{aap})_2 : \text{13tfib}$	1 : 1	100.0	99.3 mg	40.0 μL acetonitrile
$\text{CuCl}_2(\text{aap})_2 : \text{135tfib}$	1 : 1	100.0	37.5 μL	10.0 μL acetonitrile

Table S2. Crystal data and refinement details for the prepared compounds.

	$\text{CuCl}_2(\text{aap})_2$	$[\text{CuCl}_2(\text{aap})_2](\text{14tfib})$
Molecular formula	$\text{C}_{16}\text{H}_{18}\text{CuCl}_2\text{N}_2\text{O}_2$	$(\text{C}_{16}\text{H}_{18}\text{CuCl}_2\text{N}_2\text{O}_2)(\text{C}_6\text{F}_4\text{I}_2)$
M_r	404.76	806.62
Crystal system	triclinic	triclinic
Space group	$P \bar{1}$	$P \bar{1}$
Crystal data:		
$a / \text{\AA}$	4.5094(12)	4.6532(3)
$b / \text{\AA}$	6.0113(14)	6.0374(3)
$c / \text{\AA}$	15.666(4)	22.5789(12)
$\alpha / {}^\circ$	93.87(2)	82.905(4)
$\beta / {}^\circ$	90.18(2)	89.166(4)
$\gamma / {}^\circ$	92.38(2)	89.192(4)
$V / \text{\AA}^3$	423.31(19)	629.34(6)
Z	1	1
$D_{\text{calc}} / \text{g cm}^{-3}$	1.588	2.928
$\lambda(\text{MoK}_\alpha) / \text{\AA}$	0.71073	0.71073
T / K	295	295
Crystal size / mm ³	0.35 x 0.26 x 0.03	0.60 x 0.40 x 0.07
μ / mm^{-1}	1.615	3.586
$F(000)$	207	385
Refl.	2885 / 1481	8303 / 2400
collected/unique		
Parameters/restraints	113/0	169/0
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}} / \text{e \AA}^{-3}$	1.090; -1.135	0.372; -0.556
$R[F^2 > 4\sigma(F^2)]$	0.0868	0.0276
wR(F^2)	0.2560	0.0664
Goodness-of-fit, S	1.092	1.066

Table S3. Parameters of the supramolecular interactions and corresponding symmetry operators present in the prepared compounds.

Cocrystal	D…A	$d / \text{\AA}$	$\angle (\text{X}-\text{D}…\text{A}) / {}^\circ$	Symmetry Operator
$\text{CuCl}_2(\text{aap})_2$	C8–H8C…O1	3.61(2)	168.0	x, y, z
	N1–H2N…Cl1	3.466(8)	171(9)	x, y, z
$[\text{CuCl}_2(\text{aap})_2](\text{14tfib})$	I1…O1	2.989(2)	172.6(1)	$-x, -y, 1-z$
	N1–H1A…Cl1	3.474(3)	165(4)	x, y, z

Table S4. Atom list and symmetry codes of symmetry equivalent atoms (marked with an ' symbol) in the prepared compounds.

$\text{CuCl}_2(\text{aap})_2$				$[\text{CuCl}_2(\text{aap})_2](\text{14tfib})$			
Atom	Symmetry Operator	Atom	Symmetry Operator	Atom	Symmetry Operator	Atom	Symmetry Operator
C1	x, y, z	H5	x, y, z	C1	x, y, z	F1'	$-x, -y, 1-z$
C2	x, y, z	H6	x, y, z	C2	x, y, z	F2'	$-x, -y, 1-z$
C3	x, y, z	H8A	x, y, z	C3	x, y, z	I1	x, y, z
C4	x, y, z	H8B	x, y, z	C4	x, y, z	I1'	$-x, -y, 1-z$
C5	x, y, z	H8C	x, y, z	C5	x, y, z	N1	x, y, z

C6	x, y, z	H1N'	$-x, 1-y, -z$	C6	x, y, z	N1'	$1-x, -y, -z$
C7	x, y, z	H2N'	$-x, 1-y, -z$	C7	x, y, z	O1	x, y, z
C8	x, y, z	H2'	$-x, 1-y, -z$	C8	x, y, z	O1'	$1-x, -y, -z$
C1'	$-x, 1-y, -z$	H3'	$-x, 1-y, -z$	C9	x, y, z	H1A	x, y, z
C2'	$-x, 1-y, -z$	H5'	$-x, 1-y, -z$	C10	x, y, z	H1B	x, y, z
C3'	$-x, 1-y, -z$	H6'	$-x, 1-y, -z$	C11	x, y, z	H2	x, y, z
C4'	$-x, 1-y, -z$	H8A'	$-x, 1-y, -z$	C1'	$1-x, -y, -z$	H3	x, y, z
C5'	$-x, 1-y, -z$	H8B'	$-x, 1-y, -z$	C2'	$1-x, -y, -z$	H5	x, y, z
C6'	$-x, 1-y, -z$	H8C'	$-x, 1-y, -z$	C3'	$1-x, -y, -z$	H6	x, y, z
C7'	$-x, 1-y, -z$			C4'	$1-x, -y, -z$	H8A	x, y, z
C8'	$-x, 1-y, -z$			C5'	$1-x, -y, -z$	H8B	x, y, z
Cu1	x, y, z			C6'	$1-x, -y, -z$	H8C	x, y, z
Cl1	x, y, z			C7'	$1-x, -y, -z$	H1A'	$1-x, -y, -z$
Cl1'	$-x, 1-y, -z$			C8'	$1-x, -y, -z$	H1B'	$1-x, -y, -z$
N1	x, y, z			C9'	$-x, -y, 1-z$	H2'	$1-x, -y, -z$
N1'	$-x, 1-y, -z$			C10'	$-x, -y, 1-z$	H3'	$1-x, -y, -z$
O1	x, y, z			C11'	$-x, -y, 1-z$	H5'	$1-x, -y, -z$
O1'	$-x, 1-y, -z$			Cu1	x, y, z	H6'	$1-x, -y, -z$
H1N	x, y, z			Cl1	x, y, z	H8A'	$1-x, -y, -z$
H2N	x, y, z			Cl1'	$1-x, -y, -z$	H8B'	$1-x, -y, -z$
H2	x, y, z			F1	x, y, z	H8C'	$1-x, -y, -z$
H3	x, y, z			F2	x, y, z		