





# Electrophoretic Deposition of WS<sub>2</sub> Flakes on Nanoholes Arrays—Role of Used Suspension Medium

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### **Detailed XPS Analysis**

### Commercial 2H-WS<sub>2</sub> (before exfoliation)

Commercial WS<sub>2</sub> from Sigma-Aldrich was analyzed in order to get the reference for hexagonal 2Hphase BE and FWHM values, both for W 4f and S 2p lines. Moreover, since the surface oxidation due to air exposure generates WO<sub>3</sub>, the values for this species was used as reference too.



Figure S1. W 4f (left) and S 2p (right) core levels of commercial WS2.





 $\label{eq:stable_stab$ 

W 4f			S 2p		
Species	BE (eV)	% at.	Species	BE (eV)	% at.
$2H-WS_2$	32.3	93.8	$2H-WS_2$	162.1	92.1
WO <sub>3</sub>	35.7	6.2	SOx	168.8	7.9

Pristine 1T-WS<sub>2</sub>



Figure S2. W 4f (left) and S 2p (right) core levels of Pristine 1T-WS<sub>2</sub>.

W 4f			S 2p			
Species	BE (eV)	% at.	Species	BE (eV)	% at.	
1T-WS <sub>2</sub>	31.5	74.2	1T-WS <sub>2</sub>	161.4	82.8	
$2H-WS_2$	32.3	15.5	$2H-WS_2$	162.1	17.2	
$W^V S_x O_y$	34.9	5.0	SOx	-	-	
WO <sub>3</sub>	35.7	5.3				







Figure S3. W 4f (left) and S 2p (right) core levels of 1T-WS<sub>2</sub> deposited from B-1.

Table S3. Multipeak analysis of W 4f and S 2p photoemission lines for 1T-WS<sub>2</sub> deposited from B-1.

W 4f			S 2p			
Species	BE (eV)	% at.	Species	BE (eV)	% at.	
$1T$ -WS $_2$	31.5	69.2	1T-WS <sub>2</sub>	161.5	80.8	
$2H-WS_2$	32.3	16.3	2H-WS <sub>2</sub>	162.1	19.2	
$W^V S_x O_y$	34.8	9.2	SOx	-	-	
WO <sub>3</sub>	35.7	5.3				





### 1T-WS<sub>2</sub> deposited from MES 10 mM - pH 3



The presence of  $SO_x$  in S 2p line can be associated with adsorbed MES molecules.

Figure S4. W 4f (left) and S 2p (right) core levels of 1T-WS<sub>2</sub> deposited from MES 10 mM - pH 3.

**Table S4.** Multipeak analysis of W 4f and S 2p photoemission lines for 1T-WS<sub>2</sub> deposited from MES 10 mM - pH 3.

W 4f			S 2p		
Species	BE (eV)	% at.	Species	BE (eV)	% at.
1T-WS <sub>2</sub>	31.6	19.2	$1T$ -WS $_2$	161.5	64.1
$2H-WS_2$	32.4	7.4	$2H-WS_2$	162.1	24.6
$W^{v}S_{x}O_{y}$	34.8	65.3	SOx	168.4	11.3
WO <sub>3</sub>	35.7	8.1			





 $1T\text{-}WS_2$  deposited from MES 10 mM – pH 5



Figure S5. W 4f (left) and S 2p (right) core levels of 1T-WS<sub>2</sub> deposited from MES 10 mM - pH 5.

**Table S5.** Multipeak analysis of W 4f and S 2p photoemission lines for 1T-WS2 deposited from MES10 mM - pH 5.

W 4f			S 2p		
Species	BE (eV)	% at.	Species	BE (eV)	% at.
1T-WS <sub>2</sub>	31.6	54.7	1T-WS <sub>2</sub>	161.6	77.3
2H-WS <sub>2</sub>	32.4	10.6	2H-WS₂	162.1	15.0
$W^{V}S_{x}O_{y}$	34.7	22.7	SO <sub>x</sub>	167.2	7.7
WO3	35.7	12.0			





# $1T\text{-}WS_2$ deposited from MES 10 mM - pH 7



Figure S6. W 4f (left) and S 2p (right) core levels of 1T-WS<sub>2</sub> deposited from MES 10 mM - pH 7.

Table S6. Multipeak analysis of W 4f and S 2p photoemission lines for 1T-WS2 deposited from ME	5
10 mM - pH 7.	

W 4f			S 2p			
Species	BE (eV)	% at.	Species	BE (eV)	% at.	
1T-WS <sub>2</sub>	31.6	57.7	1T-WS <sub>2</sub>	161.5	71.6	
$2H-WS_2$	32.4	10.8	2H-WS <sub>2</sub>	162.1	13.5	
$W^{v}S_{x}O_{y}$	34.8	23.2	SOx	168.0	14.9	
WO <sub>3</sub>	35.7	8.3				





#### 1T-WS<sub>2</sub> deposited from MES 10 mM – pH 8

In this sample the SO<sub>x</sub> band is particularly intense: due to basic pH, sulfonic acid is mainly in the deprotonated form, that may favor the adsorption on WS<sub>2</sub> sheets.



Figure S7. W 4f (left) and S 2p (right) core levels of 1T-WS<sub>2</sub> deposited from MES 10 mM – pH 8.

**Table S7.** Multipeak analysis of W 4f and S 2p photoemission lines for 1T-WS<sub>2</sub> deposited from MES 10 mM – pH 8.

W 4f			S 2p			
Species	BE (eV)	% at.	Species	BE (eV)	% at.	
1T-WS <sub>2</sub>	31.5	45.0	$1T$ -WS $_2$	161.5	59.3	
$2H-WS_2$	32.3	8.3	$2H-WS_2$	162.1	10.9	
$W^{V}S_{x}O_{y}$	34.9	30.6	SOx	168.2	29.8	
WO <sub>3</sub>	35.7	16.1				





# 1T-WS<sub>2</sub> deposited from PBS 10 mM – pH 7.4



Figure S8. W 4f (left) and S 2p (right) core levels of 1T-WS<sub>2</sub> deposited from PBS 10 mM – pH 7.4.

**Table S8.** Multipeak analysis of W 4f and S 2p photoemission lines for 1T-WS<sub>2</sub> deposited from PBS 10 mM – pH 7.4.

W 4f			S 2p			
Species	BE (eV)	% at.	Species	BE (eV)	% at.	
1T-WS <sub>2</sub>	31.6	66.9	$1T$ -WS $_2$	161.6	84.3	
$2H-WS_2$	32.4	9.2	$2H-WS_2$	162.1	15.7	
$W^V S_x O_y$	34.9	12.5	SOx	-	-	
WO <sub>3</sub>	35.7	11.4				





# 1T-WS<sub>2</sub> deposited from PBS 1 mM – pH 7.4



Figure S9. W 4f (left) and S 2p (right) core levels of 1T-WS<sub>2</sub> deposited from PBS 1 mM – pH 7.4.

**Table S9.** Multipeak analysis of W 4f and S 2p photoemission lines for 1T-WS<sub>2</sub> deposited from PBS 1 mM – pH 7.4.

W 4f			S 2p			
Species	BE (eV)	% at.	Species	BE (eV)	% at.	
1T-WS <sub>2</sub>	31.6	44.1	$1T$ -WS $_2$	161.5	75.7	
2 <i>H</i> -WS <sub>2</sub>	32.4	9.4	$2H-WS_2$	162.1	16.2	
$W^{v}S_{x}O_{y}$	34.8	35.4	SOx	168.0	8.1	
WO <sub>3</sub>	35.7	11.1				





# 1T-WS<sub>2</sub> deposited from PBS 0.1 mM – pH 7.4



Figure S10. W 4f (left) and S 2p (right) core levels of 1T-WS<sub>2</sub> deposited from PBS 0.1 mM – pH 7.4.

**Table S10.** Multipeak analysis of W 4f and S 2p photoemission lines for 1T-WS<sub>2</sub> deposited from PBS 0.1 mM – pH 7.4.

W 4f			S 2p		
Species	BE (eV)	% at.	Species	BE (eV)	% at.
$1T-WS_2$	31.6	53.6	$1T$ -WS $_2$	161.5	85.1
$2H-WS_2$	32.4	9.4	$2H-WS_2$	162.1	14.9
$W^{V}S_{x}O_{y}$	34.9	24.5	SOx	-	-
WO <sub>3</sub>	35.7	12.4			