materials

MDPI

Supplementary material:

A feasible One-Step Synthesis of Hierarchical Zeolite Beta with Uniform Nanocrystals via CTAB

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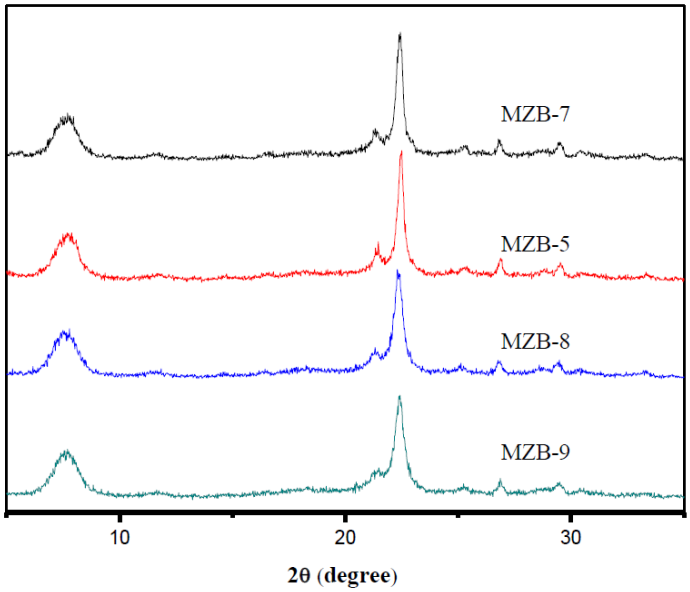


Figure S1. XRD patterns of as-synthesized zeolite Beta samples with different H_2O amount.

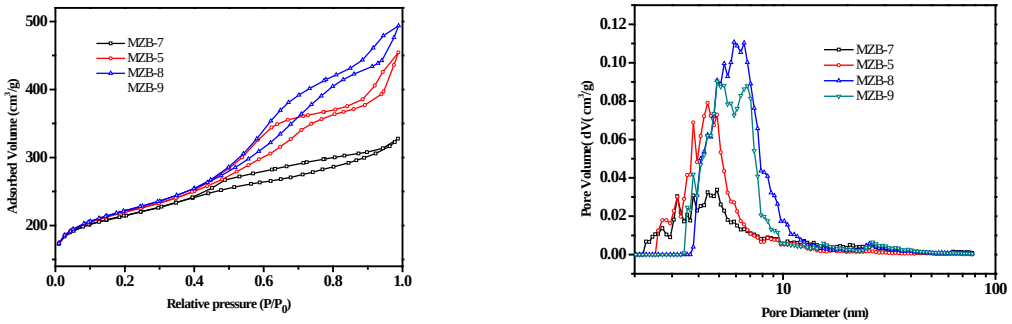
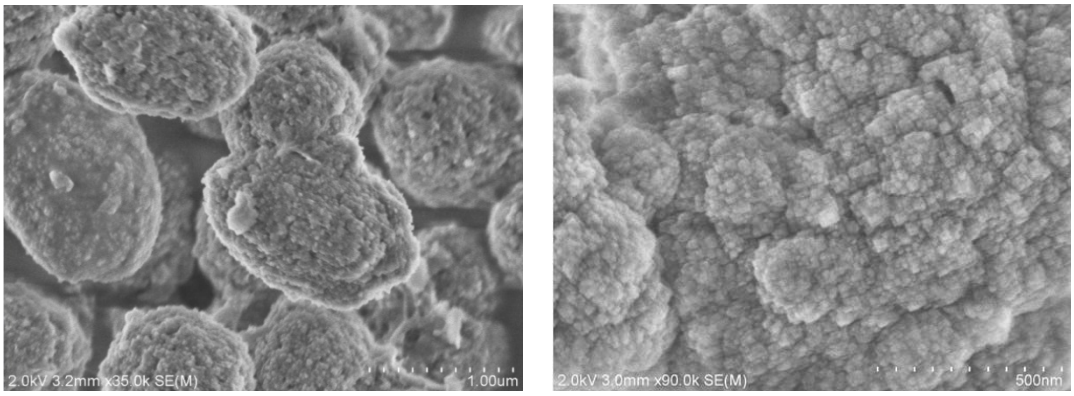


Figure S2. N_2 adsorption-desorption isotherms of synthesized zeolite Beta samples with different H2O amount (left) and pore distribution (right).



(**a**) (**b**)

Figure S3. SEM images of synthesized zeolite Beta samples MZB-5 (a) and MZB-9 (b) with different H_2O amount.

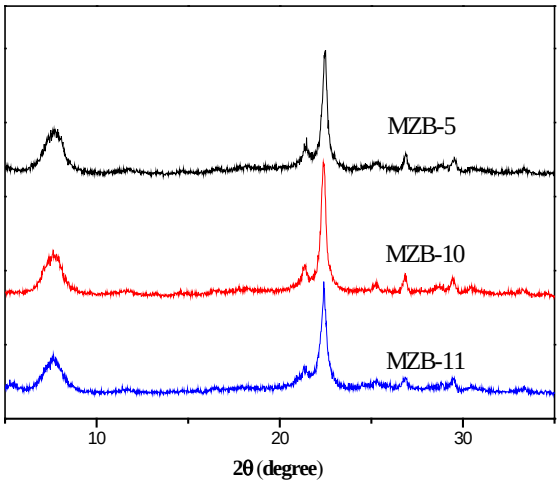
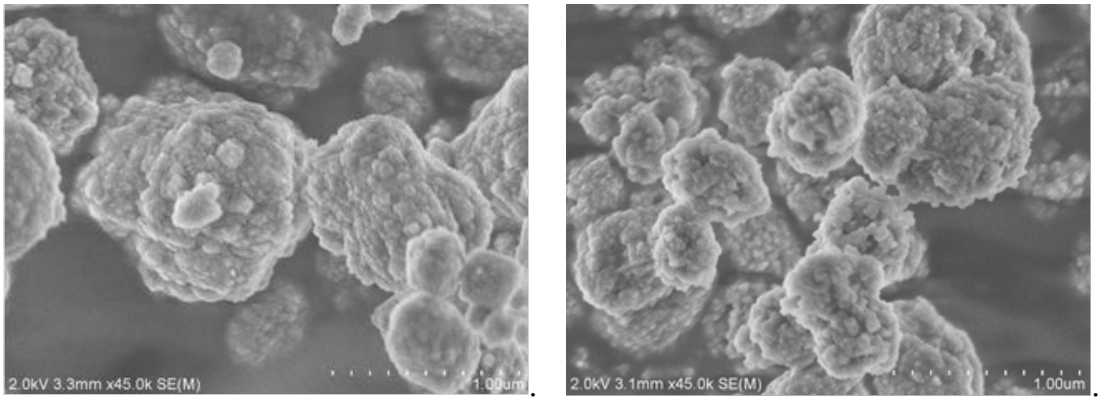


Figure S4. XRD patterns of synthesized zeolite Beta samples with different pre-crystallization time.



(**a**) (**b**)

Figure S5. SEM images of synthesized zeolite Beta samples MZB-10 (a) and MZB-11(b) with different pre-crystallization time.

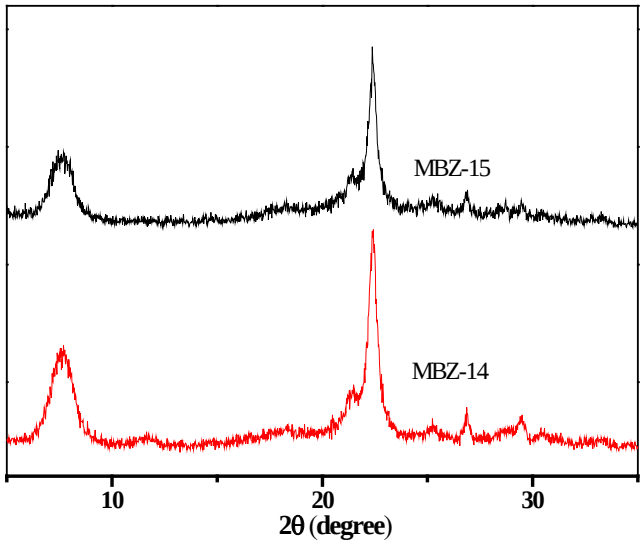


Figure S6. XRD patterns of synthesized zeolite Beta samples with alcohol.

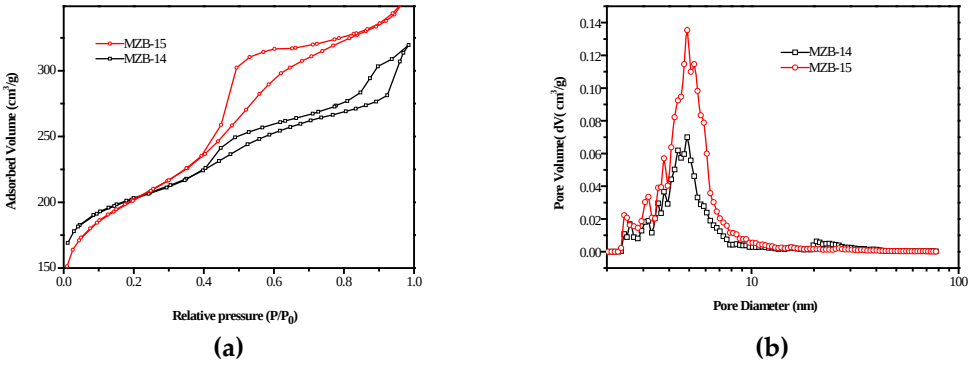


Figure S7. N_2 adsorption-desorption isotherms of synthesized zeolite Beta samples with alcohol (left) and pore distribution (right).

Table S1 Pore structure parameters of hierarchical Beta zeolites with 10L autoclave.

Samples	S_{BET} (m^2/g)	S_{MIC} (m^2/g)	S_{EXT} (m^2/g)	V_{MIC} (cm^3/g)	V_{MESO} (cm^3/g)	V_{pore} (cm^3/g)	HF
MZB-4 (M)	**756**	**496**	**260**	**0.20**	**0.36**	**0.56**	**0.12**
MZB-15(M)	**715**	**394**	**321**	**0.16**	**0.53**	**0.69**	**0.10**

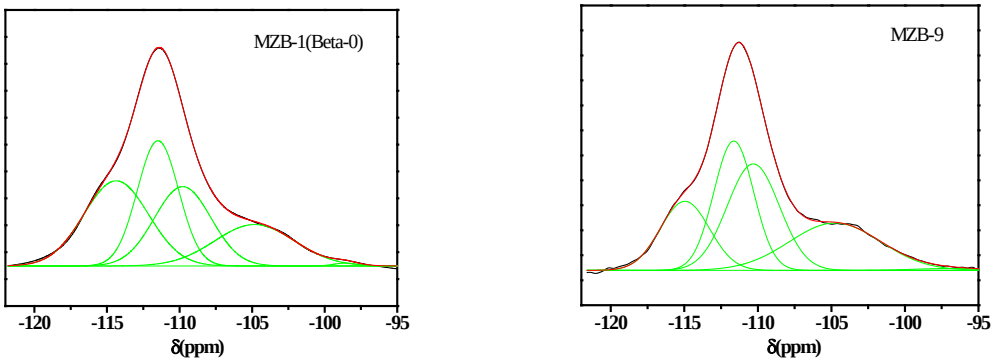


Figure S8. 29Si MAS NMR spectra of microzeolite Beta MZB-1 and hierarchical zeolite Beta MZB-9.