



Article

A Novel Solar System of Electricity and Heat

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Abstract: Thermoelectric devices may have an essential role in the development of fuel-saving, environmentallyfriendly, and cost-effective energy sources for power generation based on the direct conversion of heat into electrical energy. A wide usage of thermoelectric energy systems already exhibits high reliability and long operation time in the space industry and gas pipe systems. The development and application of solar thermoelectric generators (TEGs) are limited mainly by relatively low thermoelectric conversion efficiency. Forthe first time, we propose to use the direct energy conversion of solar energy by TEGs based on the high-performance multilayer thermoelectric modules with electric efficiency of ~15%. Solar energy was absorbed and converted to thermal energy, which is accumulated by a phase-change material (aluminum alloys at solidification temperature ~900 K). The heat flow from the accumulator through the thermoelectric convertor (generator) allows electrical power to be obtained and the exhaust energy to be used for household purposes (heating and hot water supply) or for the operation of a plant for thermal desalination of water.

Keywords: solar energy; TEG; thermoelectric module; figure of merit; heat accumulation



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1. Introduction

Energy production and environmental conservation are among the most important global challenges of modern humanity [1–4]. Complex problems are encountered on the way to strategically balancing these issues. Solar energy is an important sustainable, widespread, although sparse, energy source in many regions of the globe [5–8]. There are many ways to utilize solar energy: photochemistry, solar desalination, electricity generation, solar water heaters, and room temperature control. Direct conversion of solar energy into electricity is known using photovoltaic (PV) cells and thermoelectric converters (TEG) [9,10]. Currently, the most popular, widely used, and commercially available photovoltaic solar cells are based on Si, CdTe, and GaAs. Having a sufficiently large band gap, they convert only a part of the solar spectrum, while the excess energy of photons exceeding the band gap is lost due to the thermalization of the generated electron–hole pairs, which leads to the heating of solar panels [11,12]. The power generated by a solar cell is the product of the photovoltage on the photocurrent. The photovoltage is proportional to the band gap, while the photocurrent is reversely dependent on the band gap. This is another viewpoint of the solar energy lost.

Spectral splitting is an effective method for managing the spectrum of light that reaches PV cells, and it has great potential for preventing unnecessary heating of the cells. For instance, Si solar cells can effectively utilize light within the range of 300 to 1100 nm, while GaAs solar cells can utilize light within the range of 300 to 870 nm. With spectral splitting, the solar spectrum is separated into different parts, and only the part that is within the effective range of the PV cell is directed to the cell for generating electricity. The rest of

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the spectrum, which cannot be electrically utilized by the cell, is sent to a separate thermal absorber for thermal energy generation. The thermal absorber is thermally decoupled from the PV cells, which means that it can generate high-temperature thermal energy without overheating of the PV cells. This is especially true when using PV systems with concentrators, as the concentration of solar energy may result in a too high operating temperature for the PV cells. Thus, spectral splitting helps to increase the lifespan of the cells and improve the efficiency of PV generation of electricity and, simultaneously, to store high-temperature thermal energy.

The development of solar hybrid systems has become an important area of research in recent years due to the limitations of traditional photovoltaic (PV) systems [13–15]. The photovoltaic generator—thermal station (PVT) [16] combines a photovoltaic generator with a thermal station to improve the overall efficiency. The thermal station captures the waste heat generated by the PV cells and uses it to produce additional electricity through a thermodynamic cycle. This cycle involves using the waste heat to generate steam, which is then used to power a turbine to generate additional electricity. The PVT system has several advantages over traditional PV systems. First, it can generate both electrical and thermal energy. Second, it can operate more efficiently in high-temperature environments, which is particularly useful when using PV systems with concentrators. Third, it can generate electricity even during periods of low solar radiation or at night by utilizing the stored thermal energy.

The photovoltaic station—thermoelectric generator (PV-TEG) [17–22], on the other hand, integrates a thermoelectric generator with a PV station. The thermoelectric generator utilizes the temperature difference between the hot PV cells and the cooler ambient temperature to generate electricity. This approach eliminates the need for a thermal station and is, therefore, simpler than the photovoltaic generator—thermal station. Both of these hybrid systems have the potential to significantly improve the efficiency of PV systems with concentrators. They are also able to generate electricity, even during periods of low solar radiation, by utilizing the stored thermal energy.

The efficiency of conversion of thermal energy to electricity by TEGs improves with increasing temperature difference. Thus, TEGs are the natural choice for utilization of the high-temperature thermal energy stored in the above-mentioned spectral splitting systems. They have also the capability to transform into electricity part of the waste heat generated on the surfaces of PV cells during their operation. TEGs may be operated alone to utilize the entire solar spectrum once it is absorbed and tunneled into a heat flux across the converter. Forthe first time, Prof. Z. Dashevsky with colleagues developed and tested a solar water-lifting unit with external-heat-supply engine, which directly converts the solar energy to thermal energy [23,24]. A schematic view of this system is presented in Figure 1. The unit consists of a parabolic concentrator (1) with a diameter of 2 m. The heat absorber is at the focus of concentrator. The heat absorber is directly connected with the heat exchanger of a Stirling power convertor (3). This convertor activates a water pump. The Stirling convertor has efficiency of ~20% and a power of 200 W.

The efficiency of a thermoelectric converter (generator) η is the ratio of the produced electrical energy to the thermal energy and is the product of two terms: the Carnot efficiency and the thermoelectric efficiency, which depends of an average dimensionless figure of merit (ZT)_{av}, which is [25]:

$$(ZT)_{av} = \frac{1}{T_h - T_c} \int_{T_c}^{T_h} ZT \cdot dT,$$

where T_h is the hot-side temperature and T_c is the cold-side temperature.

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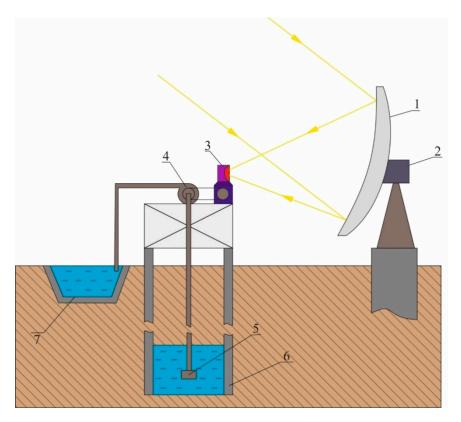


Figure 1. A solar water-lifting system with Stirling power convertor [21]. 1—parabolic concentrator. 2—tracking system. 3—Stirling power converter. 4—water pump. 5—valve. 6—water well. 7—water tank.

The dimensionless figure of merit is expressed as:

$$ZT = \frac{S^2 \sigma}{k} T,$$

where *S* is the Seebeck coefficient, σ is the electrical conductivity, *k* is the thermal conductivities, and *T* is the absolute temperature.

The figure of merit *Z* mainly depends on three fundamental parameters of the semi-conductor material [26]:

- Effective mass for the main carriers (m_n^* or m_p^* for electrons or holes for n- or p-type semiconductor, respectively);
- Drift mobility of the main carriers μ;
- Phonon part of the thermal conductivity k_L .

$$Z \sim m_n^{*3/2} (\mu/k_L)$$

The values of ZT for different thermoelectric materials in the 300–1200 K temperature range were published in [27–43] and are reproduced in Figure 2. It shows that, in order to obtain the highest ZT over a wide temperature range, TEGs should be multilayered, with appropriate material selection for different temperature ranges.

In the present contribution, we offer a hybrid system consisting of a phase-change thermal storage of solar energy, high-efficiency conversion of the heat flux to electricity by TEG, and secondary use of the waste heat for domestic heating or for water desalination, with an estimated total efficiency of up to 70%.

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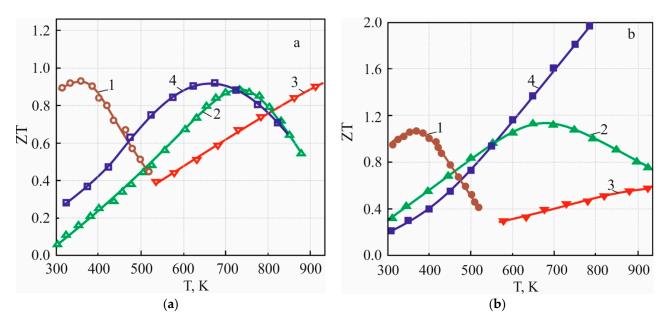


Figure 2. Temperature dependence of dimensionless figure of merit ZT as a function for different thermoelectric materials over 300–900 K temperature range [27–43]. (a) n-type: 1—Bi $_2$ Te $_3$ _xSe $_x$. 2—PbTe. 3—Si $_1$ _xGe $_x$. 4—Pb $_0$.75Sn $_0$.25Te. (b) p-type: 1—Bi $_2$ _xSb $_x$ Te $_3$. 2—GeTe. 3—Si $_1$ _xGe $_x$. 4—PbTe.

2. Experimental Procedure

Polycrystalline ingots: n- and p-type Bi₂Te₃-based compounds and A^{IV}B^{VI} semiconductors (n-type PbTe doped with In and I; p-type GeTe doped with Bi) were melted at 1000–1300 K in quartz ampoules under a vacuum of 10^{-5} Torr in a rocking furnace [29,40–42]. Some Bi₂Te_{3-x}Se_x ingots were doped with 0.1 mol% SbI₃. Iodine is a donor in these compounds [24]. The ampoules were quenched into water from a temperature of ~1000 K for Bi₂Te₃-based compounds and T~1300 K for A^{IV}B^{VI} semiconductors. The powder of these ingots was prepared in the planetary ball mill. Then, the powder was compacted into tablets with a diameter of 20 mm and a thickness of 5 mm under the pressure of 1 GPa. The tablets were densified by spark plasma sintering (SPS) technique [29,40–42,44]. A schematic view of the SPS technique is presented in Figure 3. The temperature of sintering was T~900 K during 20 min under a compressive stress of 60 MPa in an argon atmosphere. The heating/cooling rate was 30–50 K/min.

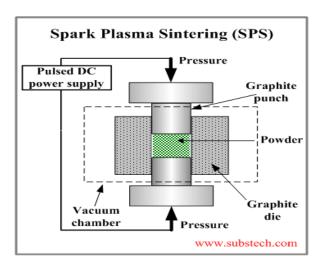


Figure 3. Scheme of the spark plasma sintering (SPS) technique [44].

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The Seebeck coefficient S and the electrical conductivity σ were measured by commercial apparatus SBA 458 Nemesis developed by Netzsch. Measurements were in argon atmosphere over a 300–900 K temperature range [41].

Thermal diffusivity α was measured using standard techniques (Netzsch LFA 457 equipment) also in argon atmosphere over a 300–900 K temperature range [41]. The thermal conductivity was calculated as follows:

$$\kappa = \alpha \rho C p$$

where the density ρ was obtained by Archimedes principle and the specific heat capacity C_p was determined by comparison to a reference Pyroceram 9606 sample [40].

The output characteristics of thermoelectric unicouples over a wide temperature range (300–900 K) were measured in an original set-up, which is presented in [29]. The distinction of this set-up from standard equipment is the measurement of two thermoelectric unicouples ("twins") simultaneously.

3. Results and Discussion

3.1. The Thermoelectric Materials for Application at 300-600 K Temperature Range

Presently, the most effective thermoelectric materials operating at the 300–600 K temperature range are solid solutions of Bi₂Te₃-Bi₂Se₃ (Bi₂Te_{3-x}Se_x) for n-type thermoelectrics. The best p-type thermoelectrics are solid solutions of Bi₂Te₃-Sb₂Te₃ (Bi_{2-x}Sb_xTe₃), as shown in Figure 1 [25–29,43]. The temperature dependences of the thermoelectric characteristics (S, σ , κ , and ZT) over the 300–600 K temperature range for these p-type materials prepared by SPS were presented in [29].

3.2. The Thermoelectric Materials for Application at 600–900 K Temperature Range 3.2.1. n-Type

The best n-type thermoelectric materials for the 600–900 K temperature range are PbTe-based compounds (Figure 1a [32–34]). As shown in [40,41], indium doping in lead telluride creates an In local level in the conduction band. This leads to localization of Fermi level close to the edge of the conduction band, which imparts an optimal value of ZT [45–48]. Increasing ZT has been achieved by co-doping of PbTe with indium and iodine (I is a donor impurity in PbTe [32]) [41]. The temperature dependence of the thermoelectric characteristics over the 600–900 K temperature range for n-type Pb_{1-x}In_xTe_{1-y}I_y thermoelectrics were presented in [41]. A value of the dimensionless figure of merit ZT for Pb_{0.999}In_{0.001}Te_{0.999}Io_{.001} is ~1.3 at operating temperature T = 750 K, which is one of the maximal values for n-type PbTe.

3.2.2. *p*-Type

The best p-type thermoelectric materials for the 600–900 K temperature range are $A^{IV}B^{VI}$ semiconductor compounds (PbTe and GeTe) (Figure 1b [35–37]). Recently, significant progress has been made in increasing ZT for GeTe doped by Bi ($Ge_{1-x}Bi_xTe$ solid solutions) [42]. For these compositions, a significant decrease is observed in the hole concentration imparting a high ZT value. Moreover, GeTe thermoelectric has satisfactory mechanical properties. On the other hand, high-efficiency p-type PbTe (Figure 1) has poor mechanical properties [49–51]). This characteristic is necessary for fabrication of thermoelectric modules [27,52,53].

The temperature dependences of the thermoelectric characteristics over a 600–900 K temperature range for p-type $Ge_{1-x-y}Bi_xPb_yTe$, prepared by SPS, were presented in [42]. The parameter ZT for $Ge_{0.96}Bi_{0.04}Te$ has a value of ~2.0 at the temperature T = 700 K and it remains practically constant up to 900 K.

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4. Application

4.1. Thermoelectric Module

To obtain maximum TE efficiency, we developed a thermoelectric module based on a multilayered TE unicouple [25]. The design and composition of each layer is presented in Figure 4. The efficiency of such multilayer unicouples was measured by the set-up presented in [23] and found to have a high value of \sim 14–15% for T_h = 900 K and T_c = 320 K. A thermoelectric module (TEM) is assembled in the following route:

- 1. Replication of many TE unicouples by SPS process.
- 2. Assembly into an aluminum cassette.
- 3. Forming the module by hot pressing at temperature $T \approx 700$ K at pressure $p \approx 100$ T under argon atmosphere.
- 4. Setting the module in a 1 mm thick stainless-steel case [25]. Electrical insulation between the case and the TE module is provided by 0.1 mm thick mica.
- 5. Sealing the protective case and filling it with argon at a pressure $p \approx 760$ Torr.

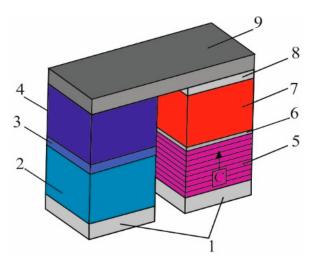


Figure 4. Schema of TE multilayer unicouple, designed to operate at a wide temperature difference between the hot and the cold sides of the unicouple $\Delta T \sim 600$ K. 1—Co contact. 2—n-type Bi_{2.7}Se_{0.3}Te₃ layer. 3—n-type PbTe_{1-x}I_x layer. 4—n-type Pb_{1-x}In_xTe_{1-y}I_ylayer. 5—p-type Bi_{0.5}Sb_{1.5}Te₃ layer (orientation along c axis). 6—CoGe₂ (anti-diffusion) layer. 7—p-type Ge_{1-x}Bi_xTe layer. 8—thin SnTe layer. 9—Fe contact.

The characteristics of the thermoelectric module are presented in Table 1.

Table 1. The characteristics of therr	moelectric module.
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Dimensions, mm-mm-mm	Number of Unicouples	Hot Temperature, T_h , K	Cold Temperature, T_c , K	Supply Voltage <i>U</i> , V	Supply Current I, A	Electric Power P_e , W
150-30-10	100	900	300–320	8	15	120

4.2. Solar Hybrid System Using TEG (SHTES)

Based on TEG, a novel solar hybrid system of electric energy and heat is presented in Figure 5. Solar radiation (1) is collected by the reflector (2), which is oriented by a tracking system (3) (heliostat) onto absorber (5). There, the radiation is converted to thermal energy and accumulated in a phase-change material (PCM). The melting temperature of PCM is close to the optimal operating temperature of the TEG (8). Up to 15% of the thermal energy is converted into electricity due to the difference ΔT between the temperatures of the thermal storage and the radiators (9) on the cold side of TEM.

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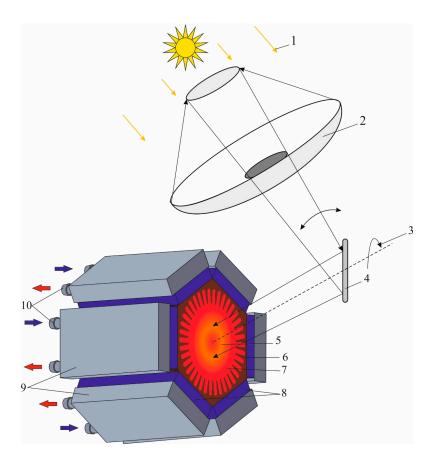


Figure 5. Schematic view of solar hybrid system using thermoelectric generator (SHTES). 1—solar radiation. 2—reflector. 3—tracking system. 4—mirror. 5—solar receiver. 6—heat exchanger. 7—thermal storage. 8—thermoelectric module (TEM). 9—heat sink. 10—water connectors.

4.2.1. Reflector

In Figure 5, the reflector (2) is a parabolic dish that tracks the sun. It is constructed using modular segments that can be combined to create dishes ranging from 3 to 10 m in diameter, depending on the desired power and corresponding reflective surface area. This reflector is designed for ease of use, high precision, and low weight and cost. The low weight allows for precise tracking using small motors and drive mechanisms. The reflective surface is made of mirror-coated facets with a reflectivity of approximately 95%. These facets are identical, curved in a special way, allowing for individual alignment. Additionally, the individual alignment of the facets can be used to control the distribution of energy on the target, resulting in high precision with an accuracy of less than 0.5 milliradians.

4.2.2. Sun Tracking System

The sun tracking system presented (3) in Figure 5 is based on a polar co-ordinate system, thus enabling easy and accurate tracking. The developed design, which combines the main axis of motion with the optical axis, enables directing of the concentrated solar radiation onto a stationary receiver with minimal optical losses and without the need to mobilize a thermal receiver.

4.2.3. Thermal Energy Storage

The solar receiver and absorber (5) are integrated into the thermal storage vessel (7) to reduce thermal and hydrostatic loads and ensure uniform absorption. Solar energy absorbed by the receiver makes the PCM in the accumulator heat and melt. The phase transition keeps the temperature of the hot end of the thermal converter nearly constant and allows generation of electricity during the night. It is proposed to use Al alloys with a melting point around 900–930 K as PCM. Al is environmentally harmless, nontoxic,

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and readily available. It has relatively high storage density (400 Joules per gram) and its melting point can be tuned by alloying with copper. According to data from the National Renewable Energy Laboratory, the average solar insolation (or solar radiation) in Israel is about 5.5 kilowatt-hours per square meter per day. To supply TEG with ~15% efficiency and electric power output of 700 W, we need about 5 kW heat power flow. Thus, to support this power flow during 24 h, we need $24 \times 5 = 120$ kWh per day of solar energy, which can be collected from $120/5.5 \approx 22$ m² of insolated area. For thermal power of Q = 5 kW, a rough estimate of the amount of Al gives about 540 kg. This amount of Al is necessary to store the solar energy collected during the day time (12 h) as energy of melting, taking into account the heat of melting $h_{Al} = 10.79$ kJ/mol.

4.2.4. Solar Thermoelectric Generator (STEG)

When STEG is used as a stand-alone solar power converter, it can be characterized in terms of power, operating temperature, location, and heat utilization. The temperature of the heat exchanger 6 (Figure 5) is maintained at about 900 K. The STEG consists of six thermoelectric modules (8), connected thermally in parallel and electrically in series. Each cold side of the TEM (8) is connected to a radiator (9). Water circulates in the heat sinks at a temperature of 320–340 K. This hot water can be stored in a thermostated tank, such as a solar boiler, and have secondary usage, such as domestic heating, hot water supply, or thermal desalination of water.

It is expected that a prototype of such a hybrid STEG can produce ~700 W of electricity [23]. The remaining ~3000 W of thermal energy can have secondary usage. The advantage of such a system is high stability without operational maintenance during the year and a long service life of more than ~20–25 years.

4.3. Solar Thermal Desalination of Water

In arid zones, the heat concentrated on TEG may be utilized to produce renewable electric power, while all the waste heat may be used to produce another necessary resource by thermal desalination of water [54,55]. Thermal desalination is a process of removing salt and other impurities from water by using heat to evaporate the water and then condense the vapor into pure water. There are two main approaches for thermal desalination: direct and indirect.

Direct methods of thermal desalination involve using a single device where evaporation and condensation take place simultaneously. These techniques are typically used to produce water for small communities and households. Examples of direct solar thermal desalination methods include solar stills, water cones, and water pyramids. Solar stills work by using sunlight to heat a shallow container of saltwater. As the water evaporates, it condenses on a sloping surface and drips into a collection container, leaving behind the salt and other impurities. Water cones and water pyramids are similar in design, using a sloping surface to collect and condense water vapor. These methods of thermal desalination are relatively simple and inexpensive, making them a good option for small-scale water desalination projects in arid regions. However, they may not be efficient enough to meet the needs of larger communities or industries.

Indirect methods ordinarily have two steps: a solar collector and a desalination unit. The solar collector is designed to absorb solar radiation and convert it into heat energy. This heat energy is then transferred to a desalination unit, where the actual process of water desalination takes place. The solar collector used in indirect methods can contain a TEG that produces electricity and supplies the exhaust heat in a series connection to the desalination system. These systems may be membrane distillation, multistage flash, and multiple-effect evaporation systems. In the multistage flash method, hot water is delivered from the steam boiler to the first stage, where a portion of water flashes. The vapors condensate and deliver heat to the feed water. The remaining water continues to the second stage, where it flashes at lower pressure and temperature and so on. In the multiple-effect evaporation method, hot steam is delivered from the steam boiler to the first stage, where it condensates and

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delivers heat to the feed water, which evaporates at lower pressure and temperature and continues to the next step.

Some solar collectors, such as solar ponds, can use low-grade heat up to 350 K. Solar ponds are shallow ponds filled with saltwater and have a layer of black sediment at the bottom that absorbs sunlight and converts it into thermal energy. This method is particularly useful in regions where there is abundant sunlight but low wind speeds. Other desalination systems require higher temperatures of up to 500 K to achieve higher thermodynamic efficiency. These systems are typically more complex and expensive, but they are more suitable for large-scale water desalination projects. Thus, indirect methods of thermal desalination are more efficient and can handle larger volumes of water than direct methods, making them ideal for industrial and commercial applications in arid regions.

5. Conclusions

A unique hybrid solar system using a high-performance STEG was developed. The main advantages of this system are the following:

- All the energy of the solar spectrum is converted by STEG to electrical and heat energies.
- The solar energy is absorbed by a PCM thermal storage, which saves the heat during long time and temperature ~900 K due to the reversible transition of aluminum alloy between the solid and liquid states.
- This system does not require buffer electric accumulators.
- A high conversion efficiency up to 15% (electric part) multilayer thermoelectric module was developed. For each leg of each TE unicouple, two types of high-efficiency thermoelectric materials were selected to ensure high efficiency along a wide range of operating temperatures. The low-temperature materials (for the 300–600 K temperature range) are based on Bi₂Te₃ compound. The middle-temperature materials (for the 600–900 K temperature range) are *n*-type PbTe doped by indium, which provided a practically constant value of average figure of merit *ZT* at a wide temperature range. The thermoelectric of *p*-type is GeTe semiconductor compound doped by Bi up to 5%.
- The waste solar energy due to Carnot cycle can be collected in the heat radiators at a temperature of 320–350 K and has secondary usage, such as domestic heating or water thermal desalination.

We expect that a prototype of such a solar hybrid system can produce ~700 W of electrical energy and ~3 kW of heat energy for heating and hot water supply (in the domestic variant).

Author Contributions: S.M.—principal investigator; R.Z.S.—methodology; B.D.—investigation and visualization; F.G.—project management and conceptualization; Z.D.—supervision. All authors have read and agreed to the published version of the manuscript.

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Data Availability Statement: The raw/processed data required to reproduce these findings cannot be shared at this time as the data also form part of an ongoing study.

Conflicts of Interest: The authors declare no conflict of interest.

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