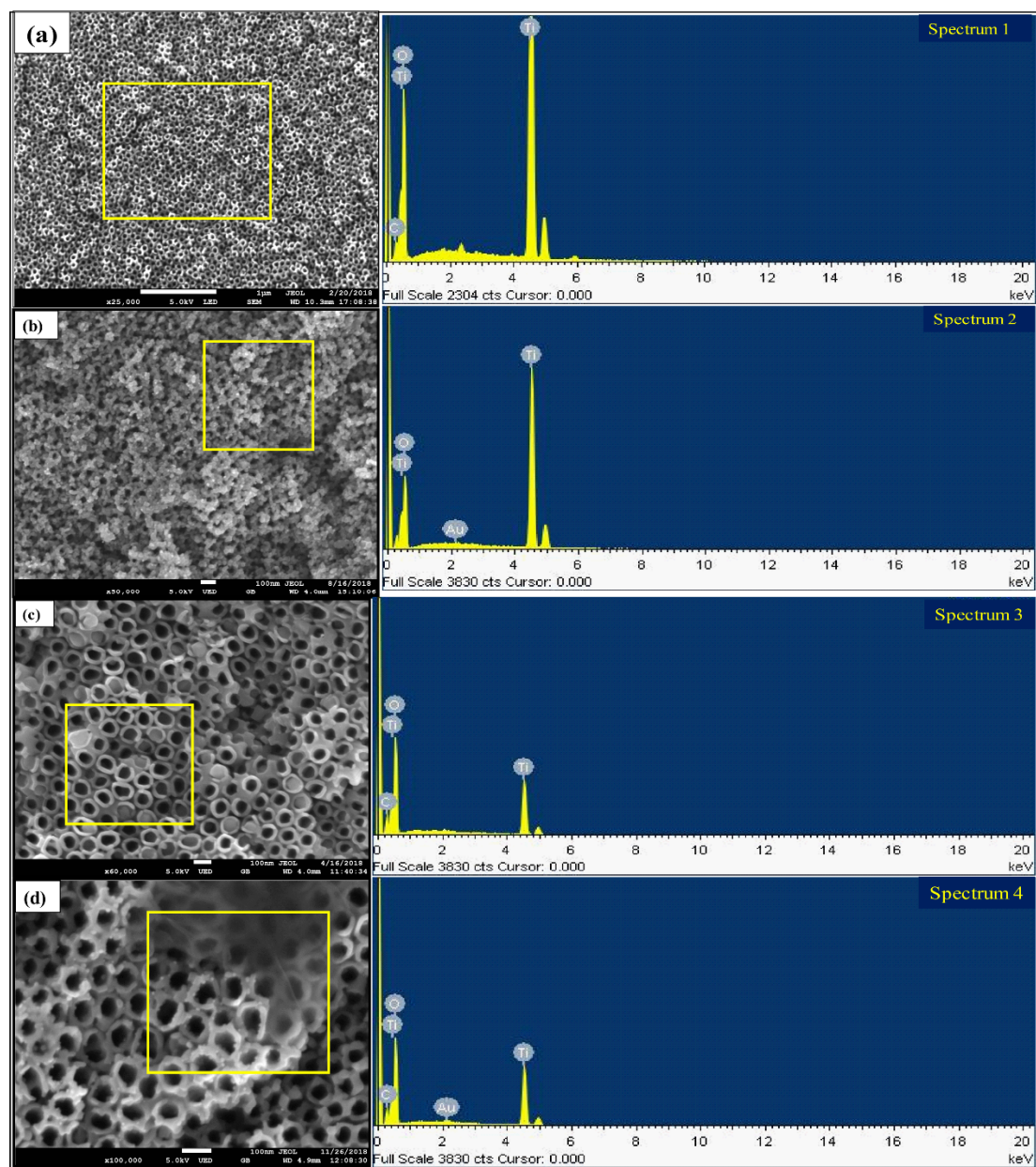


# Supporting Information

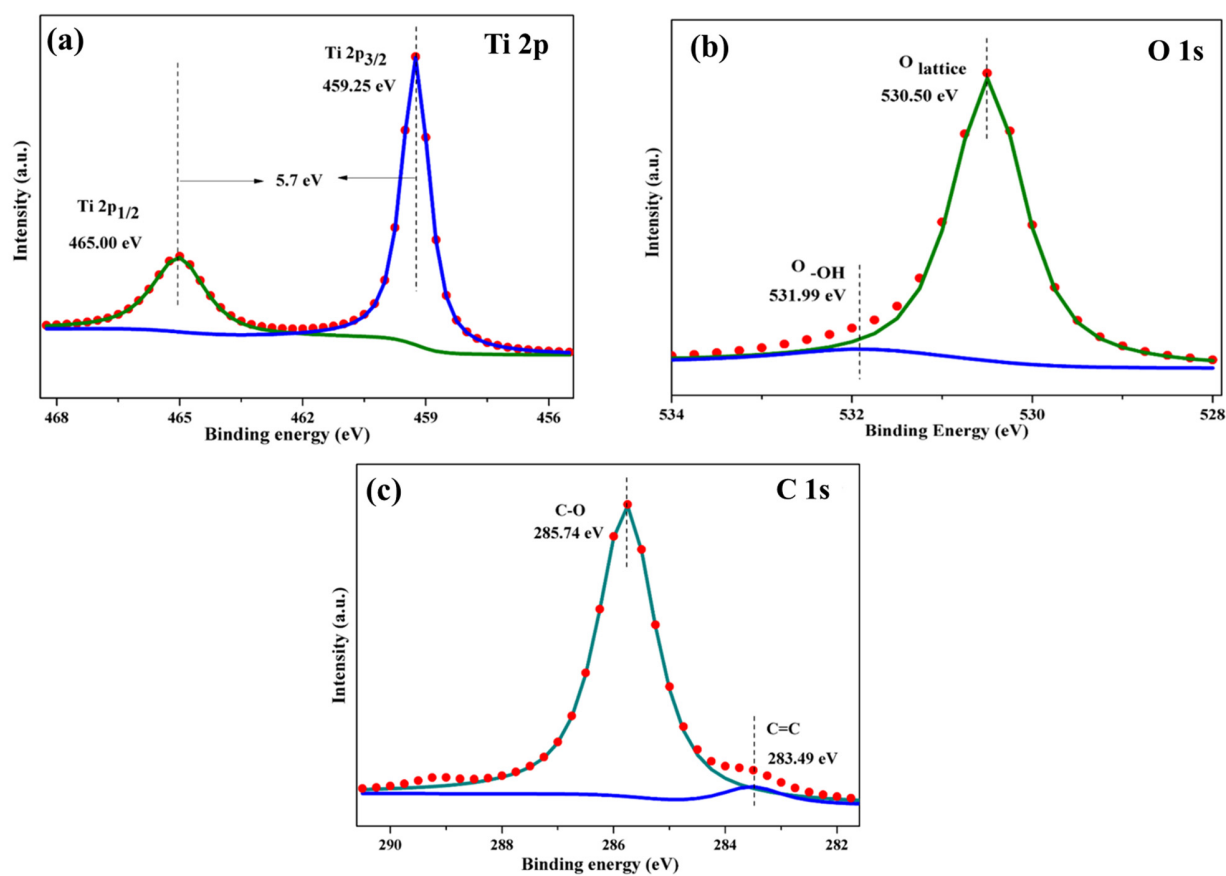
## Enhanced Photocatalytic CO<sub>2</sub> Reduction to CH<sub>4</sub> Using Novel Ternary Photocatalyst RGO/Au-TNTAs

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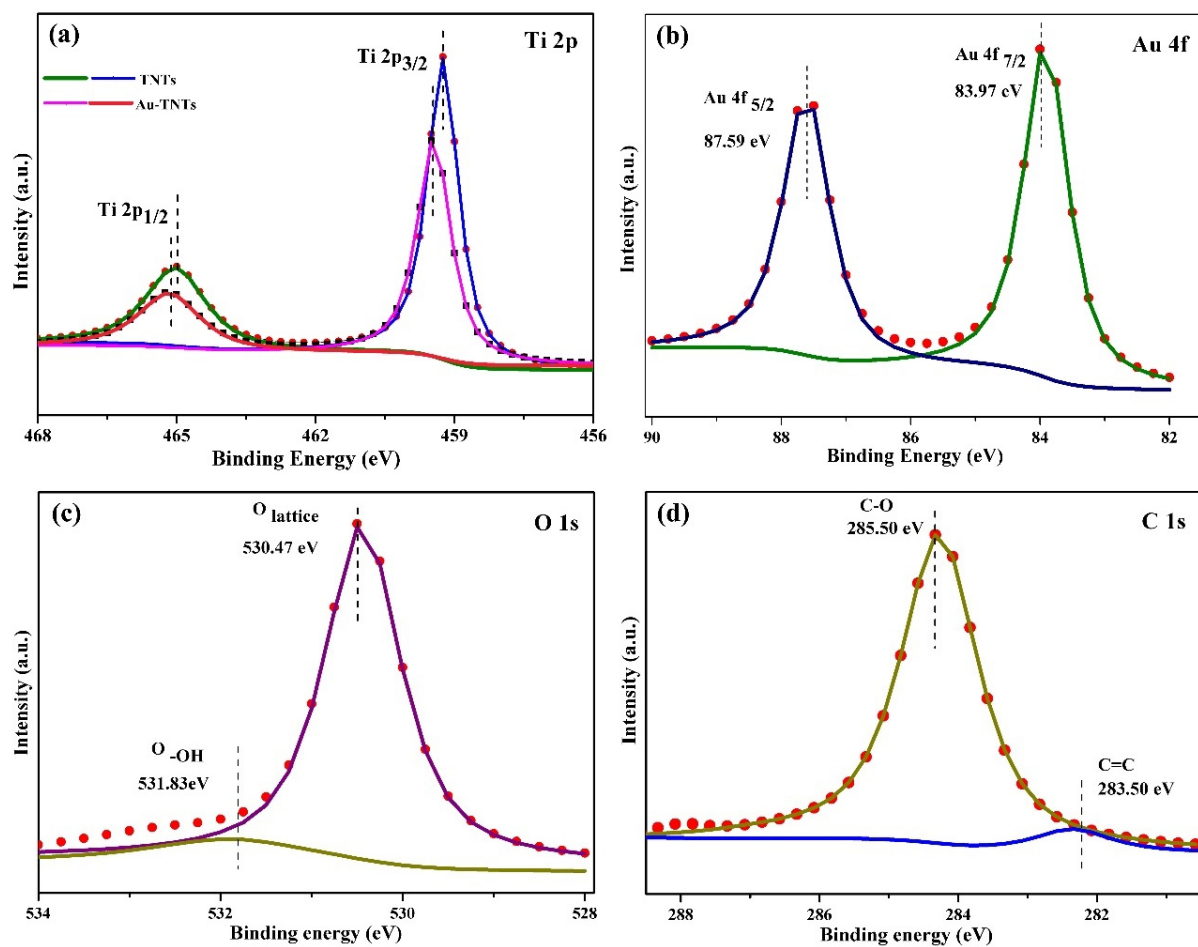
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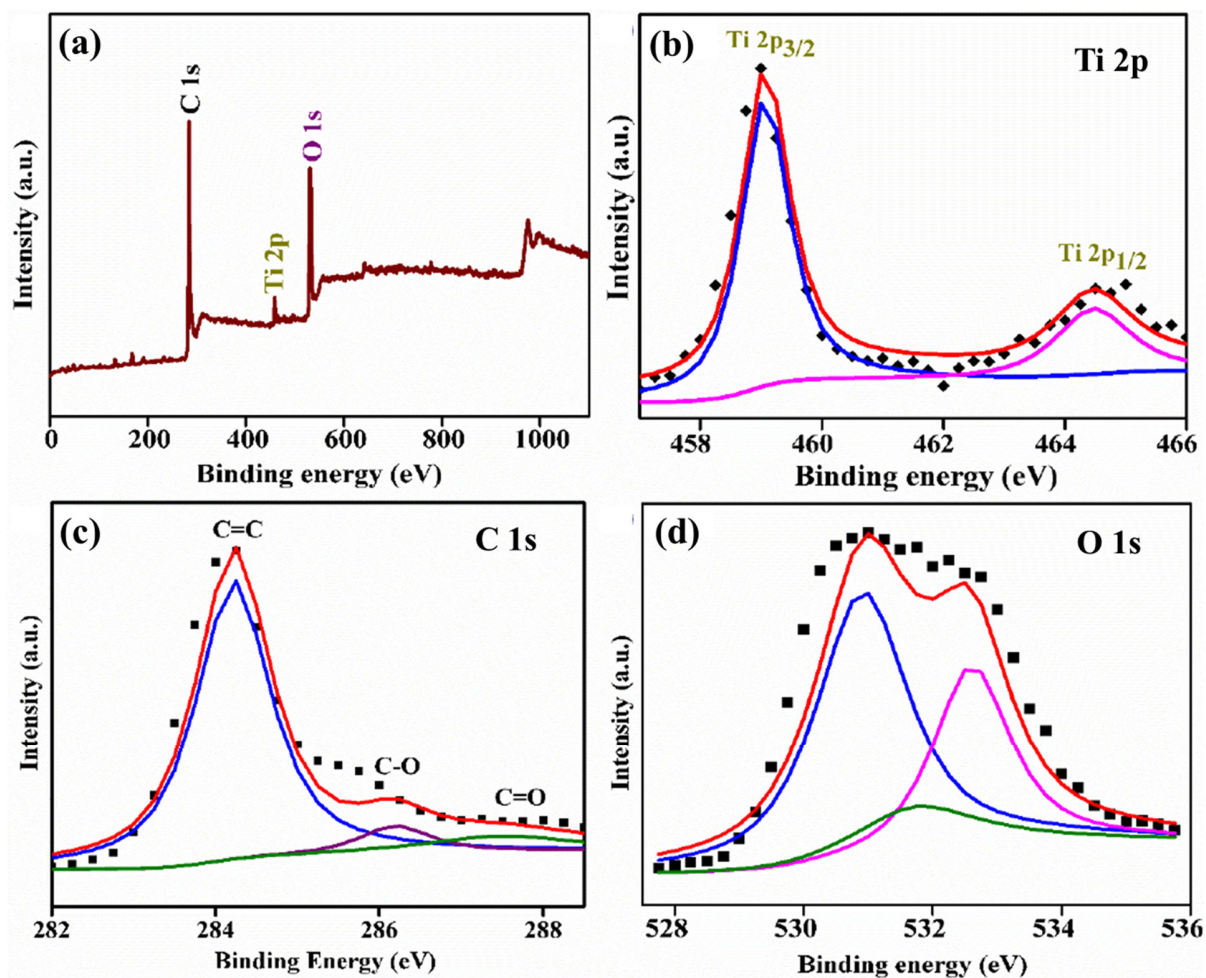
**Figure S1.** EDX of (a) TNTAs, (b) Au-TNTAs, (c) RGO-TNTAs and (d) RGO/Au-TNTAs.



**Figure S2.** Core level XPS spectra of TNTAs (a) Ti 2p, (b) O 1s and (c) C 1s.



**Figure S3.** XPS spectra of Au-TNTAs (a) Ti 2p of TNTs and Au-TNTs, (b) Au 4f, (c) O 1s and (d) C 1s.



**Figure S4.** XPS spectra of RGO-TNTAs (a) Fully scanned spectra (b) Ti 2p, (c) C 1s and (d) O 1s.

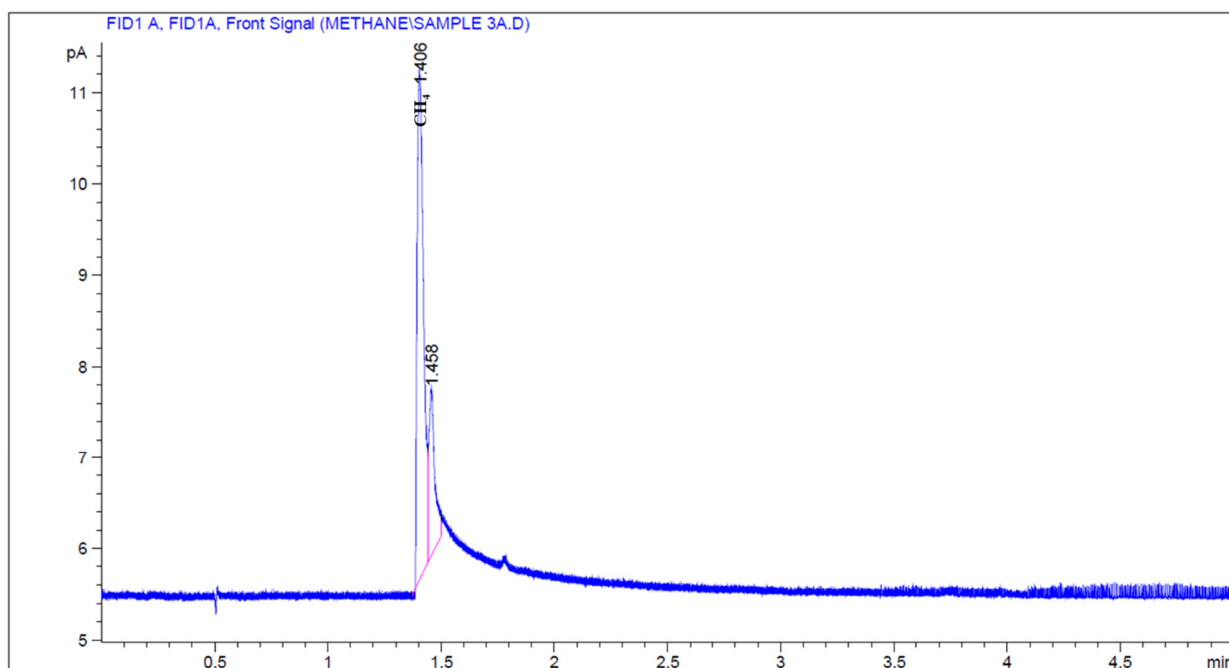
### Quantification of Bandgap of Synthesized Photocatalysts

The bandgap energies were estimated by using the Tauc equation (Equation (S1)).

$$(\alpha h\nu)^2 = A (h\nu - E_g) \quad (S1)$$

where  $\alpha$  is the absorption coefficient,  $h\nu$  is the photon energy,  $A$  is the absorption constant, and  $E_g$  is the energy band gap. By plotting the values of  $(\alpha h\nu)^2$  versus the photon energy values ( $h\nu$ ) the band gap energies of different synthesized photocatalysts were estimated.





**Figure S5.** Chromatograms of hydrocarbon fuels generation using RGO/Au-TNTAs as photocatalyst over a period of 4 h of light irradiation. (Detector: FID/Methanizer)

**Table S1:** Rate of CH<sub>4</sub> production at different irradiation time.

Catalysts	CH <sub>4</sub> production at every 1h irradiation time (ppm cm <sup>-2</sup> h <sup>-1</sup> )				Total CH <sub>4</sub> production (ppm cm <sup>-2</sup> )
	1h	2h	3h	4h	
TNTAs	0.72 ± 0.37	1.04 ± 0.56	1.24 ± 1.03	1.02 ± 0.67	4.02 ± 1.02
Au-TNTAs	3.42 ± 0.33	4.05 ± 0.55	6.02 ± 0.56	8.15 ± 1.02	21.64 ± 2.34
RGO-TNTAs	1.82 ± 0.32	2.21 ± 0.61	4.41 ± 0.83	4.02 ± 0.72	12.46 ± 3.40
RGO/Au-TNTAs	7.33 ± 0.46	7.92 ± 0.64	9.26 ± 1.06	10.62 ± 0.88	35.13 ± 2.33

Note: Number of of repeated experiments, n = 05