



## Defect, Diffusion and Dopant Properties of NaNiO<sub>2</sub>: Atomistic Simulation Study

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<b>Two-body</b> $[\Phi_{ij}(r_{ij}) = A_{ij} \exp(-r_{ij}/\rho_{ij}) - C_{ij}/r_{ij}^6]$					
Interaction	A / eV	ρ / Å	$C / eV \cdot Å^6$	Y / e	K / eV·Å <sup>-2</sup>
Na+O2-[1]	1497.830598	0.287483	0.00	1.000	99999
Ni <sup>3+</sup> –O <sup>2–</sup> [2]	1018.36	0.3299	0.00	4.97	304.7
O <sup>2-</sup> –O <sup>2–</sup> [3]	22764.30	0.149	43.0	-2.86	42.0
Mg <sup>2+</sup> O <sup>2-</sup> [4]	946.627	0.31813	0.000	0.000	99999
Co <sup>2+</sup> –O <sup>2–</sup> [5]	1670.2416	0.2859	0.000	-1.5030	110.5
Fe <sup>2+</sup> -O <sup>2-</sup> [6]	1207.6	0.3084	0.000	2.000	99999
Ca <sup>2+</sup> –O <sup>2–</sup> [5]	1090.4	0.3372	0.0000	1.26	34.00
Sr <sup>2+</sup> -O <sup>2-</sup> [5]	1400.0	0.3500	0.0000	1.33	21.53
Ba <sup>2+</sup> -O <sup>2-</sup> [5]	931.7	0.3949	0.000	1.46	14.78
Al <sup>3+</sup> -O <sup>2-</sup> [7]	1725.20	0.28971	0.000	3.000	99999
Sc <sup>3+</sup> –O <sup>2–</sup> [5]	1299.4	0.3312	0.000	3.000	99999
Ga <sup>3+</sup> -O <sup>2-</sup> [5]	2901.12	0.2742	0.000	3.000	99999
Fe <sup>3+</sup> -O <sup>2-</sup> [5]	1156.36	0.3299	0.000	4.970	304.7
In <sup>3+</sup> –O <sup>2–</sup> [5]	1495.65	0.3327	4.33	3.000	99999
Gd <sup>3+</sup> -O <sup>2-</sup> [5]	1885.75	0.3399	20.34	3.000	99999
Y <sup>3+</sup> -O <sup>2-</sup> [5]	1345.10	0.3491	0.00	3.000	99999
La <sup>3+</sup> -O <sup>2-</sup> [5]	1545.21	0.3590	0.00	-0.250	99999
Sn4+-O2-[8]	1414.32	0.3479	13.66	4.000	99999
Zr <sup>4+</sup> -O <sup>2-</sup> [5]	985.869	0.3760	0.00	1.350	169.617
Ce4+-O2-[5]	1986.83	0.3511	20.40	7.700	291.75
$Ti^{4+} - O^{2-}[9]$	5111.7	0.2625	0.000	-0.10	314.0
Ge4+-O2-[10]	1497.3996	0.325646	16.00	4.0000	99999
Si <sup>4+</sup> -O <sup>2-</sup> [11]	1315.2478	0.317759	10.141118	0.000	99999

Table S1. Interatomic potential parameters used in the atomistic simulations of NaNiO2.

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