

Article

Influence of Permeability and Injection Orientation Variations on Dispersion Coefficient during Enhanced Gas Recovery by CO₂ Injection

Muhammad Kabir Abba *, Athari Al-Otaibi, Abubakar Jibrin Abbas, Ghasem Ghavami Nasr and Martin Burby

Petroleum and Gas Research Group, The University of Salford, Manchester M4 5WT, UK; A.al-otaibi@edu.salford.ac.uk (A.A.-O.); a.j.abbas@salford.ac.uk (A.J.A.); g.g.nasr@salford.ac.uk (G.G.N.);

m.burby@salford.ac.uk (M.B.)

* Correspondence: m.k.abba1@salford.ac.uk

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Abstract: This investigation was carried out to highlight the influence of the variation of permeability of the porous media with respect to the injection orientations during enhanced gas recovery (EGR) by CO_2 injection using different core samples of different petrophysical properties. The laboratory investigation was performed using core flooding technique at 1300 psig and 50 °C. The injection rates were expressed in terms of the interstitial velocities to give an indication of its magnitude and variation based on the petrophysical properties of each core sample tested. *Bandera Grey, Grey Berea*, and *Buff Berea* sandstone core samples were used with measured permeabilities of 16.08, 217.04, and 560.63 md, respectively. The dispersion coefficient was observed to increase with a decrease in permeability, with Bandera Grey having the highest dispersion coefficient and invariably higher mixing between the injected CO_2 and the nascent CH_4 . Furthermore, this dispersion was more pronounced in the horizontal injection orientation compared to the vertical orientation by about 50%. This study highlights the importance of the permeability variation in the design of the injection strategy of EGR and provides a revision of the CO_2 plume propagation at reservoir conditions during injection.

Keywords: enhanced gas recovery; longitudinal dispersion coefficient; injection orientation; supercritical CO₂; CO₂ permeability

1. Introduction

The effects of greenhouse gas (GHG) emissions in the form of global warming with the resulting subsequent climate change cannot be overemphasized. The authors of [1] reported that the most significant of the GHG emissions are carbon dioxide (CO₂), methane (CH₄), dinitrogen oxide (N₂O), and other gases. Amongst all these gases, they iterated that CO₂ makes up about 76% of the total global emissions of GHGs. Incidentally, combustion of fossil fuels is the central source of the global CO₂ emissions and the oil and gas industry alone accounts for 65% of global CO₂ emissions [2]. Furthermore, CO₂ emissions from the oil and gas industry is ever increasing as a result of the high energy demand and at a rate of 1.7% per annum between the 1990s and early 2000s, and at an even higher rate of 3.1% per annum between 2000 and 2010 [3]. Therefore, the environmental consequences associated with CO₂ emissions have forced researchers in the oil and gas industry to come up with technologies to curb the proliferation of anthropogenic CO₂ due to the oil and gas activities. An avenue with a growing potential to address this issue is by the injection of the CO₂ emissions into geological



formations like oil and gas reservoirs [4] using enhanced oil/gas recovery processes. Enhanced gas recovery (EGR) by CO_2 injection and sequestration is a simultaneous process whereby CO_2 is injected into a natural gas (CH₄) reservoir to displace CH₄ and store CO_2 in the reservoir.

EGR is still in its pilot/test phase and has not been widely accepted due to the nature of the gas-gas displacement whereby CO_2 disperses into CH_4 during the process and the recovered CH_4 will be heavily contaminated with the injected CO_2 . This will affect the market value of the recovered CH_4 given that one of the reasons for the choice of CH_4 reservoirs as potential storage sites is that the recovered natural gas will offset part of the cost of the sequestration process [5]. Therefore, efforts are being made to stall the incessant mixing during EGR, and this is only achievable when the physics of the mixing are better understood. Pivotal to the adoption of the technique is understanding the mechanisms and factors which influence the interaction between the gases in situ, which leads to mixing at reservoir conditions. This will eventually provide an avenue to characterise gas systems for better injection scenarios and explore the potential and the viability of EGR as an adopted method of CO_2 sequestration.

In our previous works [6–9], we studied some of the factors that affect in situ mixing between the injected CO_2 and the displaced CH_4 which have not been previously considered. These factors include the connate water salinity, the injection orientation and, also, the flow behaviour of CO_2 in its supercritical state during EGR displacement. This research, however, focuses on the effect of injection orientation of the CO_2 on the dispersion of the gas in consolidated sandstone cores with an emphasis on permeability variation. The injection orientation determines the path of the CO_2 plume propagation through the pore spaces of the reservoir rock, which is also very important when it comes to the dip angle between the injector and producer wells. Consistent with our previous results, horizontal injections showed that the segregation of CO_2 to the bottom of the core sample—due to its higher density of supercritical CO_2 compared to that of CH_4 (showed in Figure 1)—led to higher residence time of the CO_2 traversing the length of the core sample, thereby increasing CO_2 dispersion. Invariably, dispersion in the vertical injection orientation of the CO_2 was significantly less than that in the horizontal orientation.

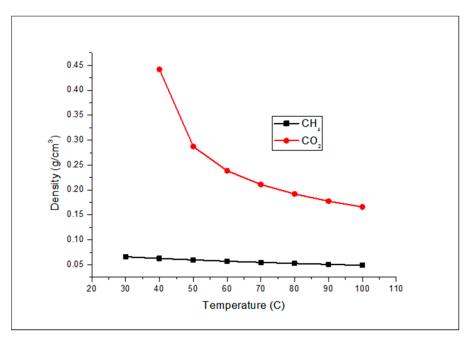


Figure 1. CH₄ and CO₂ density as a function of temperature at 1300 psi.

The authors of [10] carried out a horizontal dispersion of CO_2 in CH_4 in sand packs as the porous medium and also concluded that gravity plays a significant role in the dispersion of CO_2 during displacement applications. Their study included long and short sand packs with similar permeabilities,

while here, different consolidated core samples with different petrophysical properties and equal length were employed to ascertain the effects of their variation on dispersion. The reasoning behind the use of equal length core samples was to minimise systemic disparities in our measurements. Given that permeability is a function of length, the pore structure variation which is responsible for the permeability difference between the core sample was explored, and the interplay between the core sample and the gases was evaluated. Therefore, the variation of the injection orientation and permeability was analysed to showcase their influence on dispersion coefficient.

2. Materials and Methods

The core samples were obtained from Kocurek Industries USA whose petrophysical properties are shown in Table 1. Gas permeability was obtained using a core flooding and the porosity was evaluated using Helium Porosimetry. Research grade CO_2 and CH_4 were obtained from BOC UK both with purity of >99.999%.

Table 1. Dimensions and petrophysical properties of core samples.	petrophysical properties of core samples.
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Core Samples	Length (mm)	Diameter (mm)	* Porosity (%)	* Permeability (md)
Grey Berea	76.27	25.22	19–20	200-315
Bandera Grey	76.00	25.47	21	30
Buff Berea	76.18	24.95	26	350-600

* Properties provided by suppliers.

Apparatus and Procedure

(1) Porosity and permeability measurements

The Helium Porosimeter PORG 200[™] (Corelab, OK, USA and Gas Permeater PERG 200 [™] (Corelab, OK, USA) were used to measure the porosity and permeability of the core samples respectively. Details of the equipment description, principles of operation, and procedure can be found in our previous works [8]. The results are shown in Table 2.

Core Samples	Porosimetry Porosity (%)	Measured Permeability (md)
Bandera Grey	17	16.08
Grey Berea	20	217.04
Buff Berea	26.27	560.63

Table 2. Measured petrophysical properties.

(2) Core flooding

Core flooding equipment, also from CoreLab Oklahoma, UFS 200 (details are presented in [7]), was used in this work. It simulated the displacement of CH_4 by supercritical CO_2 at reservoir conditions. The core holder was originally horizontally orientated; however, a stand was constructed to change this orientation to vertical in order to vary the injection orientation. The same procedure was adopted as the one employed in [9], the only difference being in the using different core samples of different permeabilities and porosities. The schematics of the setup are shown in Figure 2. After carrying out the same experiments at the same conditions in the horizontal orientation for each core sample, the core holder was installed vertically, and the same experiments were repeated using the same conditions as in the previous horizontal experiments.

Injection was made from the bottom of the core holder in the vertical orientation. The injection orientation adopted in this study was not based on the dip angle of injection during the flooding and recovery processes. The experiments were not field scale depictions of the injection strategy, rather, we were looking at core to pore scale investigation of the flow physics of CO₂ plume as it displaces nascent CH₄. Central to the experiments was the interaction and mixing between CO₂ in its

supercritical state and the CH_4 in the reservoir given that the CO_2 plume occupies the lower echelon of the CH_4 zone due to its higher density at supercritical condition as we have iterated in our previous works. Vertical and horizontal flows of the CO_2 plume are the two extreme flow conditions in terms of propagation direction.

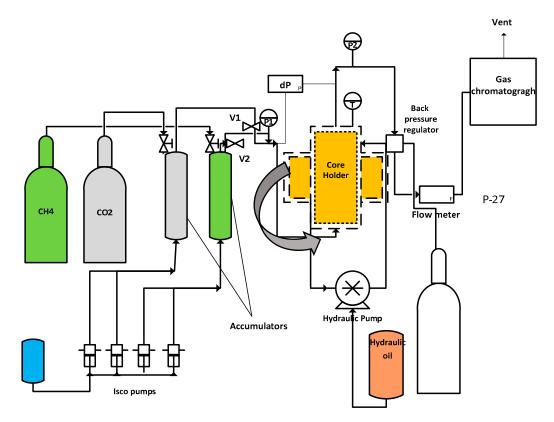


Figure 2. Core flooding set up schematics.

(3) Data analysis

A single parameter diffusion equation was used by [11] in the description of the longitudinal dispersion coefficient for gas flow in porous media which is shown in Equation (1):

$$K_l \frac{\partial^2 C}{\partial x^2} - u \frac{\partial C}{\partial x} = \frac{\partial C}{\partial t},\tag{1}$$

where C-CO₂ concentration at time t, location x, K_l —the longitudinal dispersion coefficient, and u is the interstitial velocity. The dimensionless for of Equation (1) is written as

$$\frac{1}{P_e} \frac{\partial^2 C}{\partial x_D^2} - \frac{\partial C}{\partial x_D} = \frac{\partial C}{\partial t_D},$$
(2)

where each parameter is defined in Table 3.

Parameter	Expression
P_e	$\frac{uL}{K_l}$
t_D	$\frac{t\dot{u}}{L}$
x_D	$\frac{x}{L}$
и	$\frac{Q}{\pi r^2 \phi}$

Table 3. Dimensionless parameters.

This one-dimensional advection–dispersion (ADE) equation was used to measure/evaluate, analytically, the longitudinal dispersion of CO₂ into CH₄ using the solution of the equation and also assuming that the dispersion coefficient and interstitial velocity of the displacing species are not affected by the concentration. With the following boundary conditions, the initial condition—C = 0 at $t_D = 0$, boundary conditions—C = 1 at $x_D = 0$, $C \rightarrow 0$ as $x_D \rightarrow \infty$, the dimensionless solution to (Equation (1)) thus becomes [12–14]

$$C = \frac{1}{2} \left\{ erfc\left(\frac{x_D - t_D}{2\sqrt{t_D/P_e}}\right) + e^{P_e x_D} erfc\left(\frac{x_D + t_D}{2\sqrt{t_D/P_e}}\right) \right\}.$$
(3)

Using the obtained concentration profiles from core flooding experiments, Equation (3) was used to fit the results from the experiments and the dispersion coefficient was obtained analytically using the least squares regression method with the dispersion coefficient as the fitting parameter.

Perkins and Johnston (1963) described the medium *Péclet* number denoted by P_{ex} , which defines the displacement mechanism that is dominant in gas dispersion in porous media as

$$P_{ex} = \frac{u_m d}{D},\tag{4}$$

where *D*—diffusion coefficient (m²/s), P_{em} —medium *Péclet* number, u_m —mean interstitial velocity (m/s), and *d* is the characteristic length scale of mixing in the porous medium. Largely, at the values of $P_{em} < 0.1$, diffusion mixing dominates and, equally, at $P_{ex} > 10$, advective mixing dominates during the gas dispersion process. In this range of P_{ex} , [15] related diffusion to dispersion coefficients as presented in (Equation (6)):

$$\frac{K_l}{D} = \frac{1}{\tau} + \alpha \frac{u_m^n}{D},\tag{5}$$

where α represents the dispersivity of the porous medium in *m*, τ is the tortuosity of the porous medium, and *n* is an exponent.

Additionally, [16–18] reported a correlation between the molecular diffusion coefficient, pressure, and temperature to obtain accurate diffusivity at conditions relevant to EGR as follows:

$$D = \frac{\left(-4.3844 \times 10^{-13} p + 8.55440 \times 10^{-11}\right) T^{1.75}}{p},\tag{6}$$

where D (m²/s) is the molecular diffusion coefficient of CO₂ in CH₄ pressure p (MPa) and at temperature T (K).

3. Results and Discussion

The core flooding experiments were carried out at 1300 psig, and 50 °C in both vertical and horizontal orientations for each core sample. The interstitial velocities were varied according to individual core samples given their different porosities. A range of interstitial velocities was based on the injection rates in our previous work [8], which were evaluated using Equation (7):

$$u = \frac{Q}{\phi A},\tag{7}$$

where *Q*—injection rate in m³/s, *A*—cross-sectional area of core sample, and φ —porosity of core sample. The results are shown in Table 4.

SN	Core Sample	Porosity (%)	Interstitial Velocity (µm/s)
1	Grey Berea	20.10	33.5-83.8
2	Bandera Grey	17.20	39.3–98.3
3	Buff Berea	26.27	26.2–65.6

Table 4. Interstitial velocities employed in each core sample.

Accordingly, the results of each individual core sample will be presented, discussed, and analysed based on the injection orientation and the interstitial velocity. *Grey Berea* core sample will be presented first, followed by *Bandera Grey* and, finally, *Buff Berea*.

The mole fractions of the injected CO_2 were used to develop the concentration profile which assesses rate of mixing between the injected CO_2 and the CH_4 , in situ, using Equation (3) to fit the equation to the experimental data and varying K_L , which is the longitudinal dispersion coefficient (keeping the interstitial velocity constant as assumed in the 1D ADE), until the analytical solution fits the experimental results. The L_{exp} was also adjusted in the regression to provide a good fit as carried out by [16] and adopted by [17]. Least square regression analysis was the method used in the curve fitting process.

Curve fitting was carried out using OriginPro 8 software and the curve-fitted concentration profiles for each u for *Grey Berea* core sample are shown in Figures 3 and 4 for horizontal and vertical orientation respectively. The dispersion coefficients were evaluated, and it was shown that the K_L increases with increase in the interstitial velocity. This was done for all core samples and injection orientation experiments in this investigation. There was early breakthrough at higher values of u run and a late breakthrough at lower values as expected in all the core samples. The fitting of the 1D ADE to the experimental results was meagre as a result of systemics like entry and exit effects as described in the works of [17]. This was noted for all the runs in the entire experiments. However, these do not affect the evaluation of the parameter i.e., dispersion coefficient. For all subsequent experiments, these systemic effects were noticed and are presented as such.

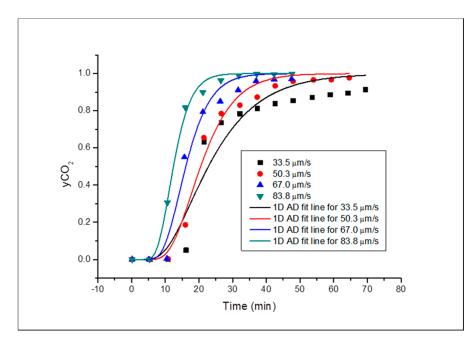


Figure 3. Concentration profile for Grey Berea in horizontal orientation.

Using Equation (6), the diffusion coefficients, D, were evaluated at the experimental conditions. This is essential when describing the dispersivity, α , and the P_{ex} of the core sample, and also when comparing the results to those in literature, which will further reaffirm the accuracy of the experiments.

The dispersivity can be analytically evaluated by fitting Equation (5) to the plot of u/D against k/D which is a straight line as shown in Figure 4.

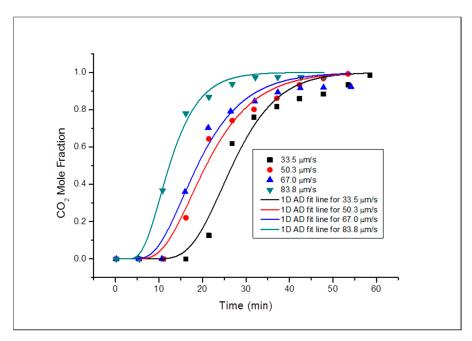


Figure 4. Concentration profile for all Grey Berea runs vertical orientation.

The fitted concentration profiles of the subsequent core sample (*Buff Bera*) is shown in Figures 5 and 6, for horizontal and vertical orientation respectively, which also showed meagre fitting as a result of systemic errors as seen in *Grey Berea*.

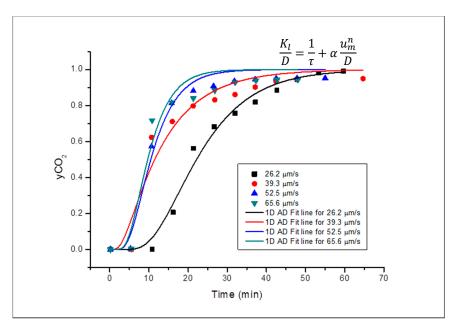


Figure 5. Concentration profile of all the runs for Buff Berea in horizontal orientation.

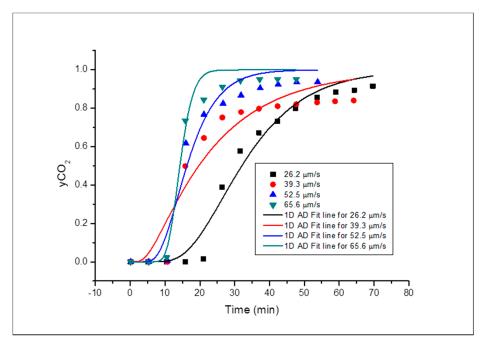


Figure 6. Concentration profile for all Buff Berea runs in vertical orientation.

Similarly, Bandera Grey core sample fitted horizontal and vertical orientation concentration profiles are shown in Figures 7 and 8. The profiles are steeper than the previous ones obtained for the core samples (Buff Berea and Grey Berea) because of the instant mixing during the displacement process due to higher interstitial velocities.

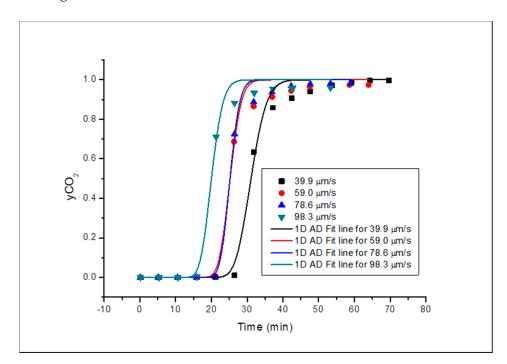


Figure 7. Concentration profile for Bandera Grey in horizontal orientation.

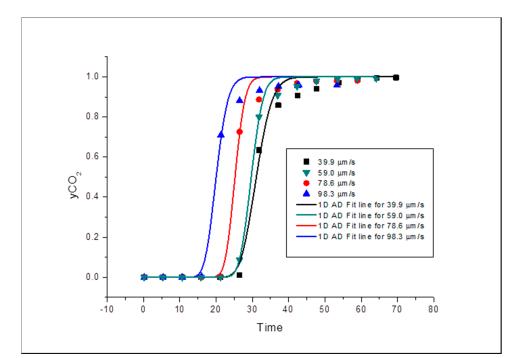


Figure 8. Concentration profile for Bandera Grey runs in vertical orientation.

The works of [16,18,19] presented values of the apparent dispersivity in consolidated porous media that were generally smaller than 0.01 ft (0.003 m). Hughes et al. (2012) obtained dispersivity in a range of 0.0001 to 0.0011 m using a core sample (Donny brook) with similar petrophysical properties as the ones used in this work. This provided a practical input variable for EGR simulations. As dispersivity is a very important porous media property, its accurate determination can provide a befitting technique to establish the optimum injection rate of the CO_2 for a better simulation of the fluid flow in the matrix of the reservoir during EGR. All the dispersivity obtained (from each experiment carried out in this study are well within those obtained from literature. This was evaluated from Equation (5) as earlier stated—a straight-line equation. This further demonstrates the reliability of the experiments.

The tortuosity (inverse of the intercept on the y-axis of the straight line) was similar in both cases (Figures 9–11), for *Grey Berea*, indicating that regardless of the injection orientation of the core sample, the tortuosity (a property of the core sample) remained unchanged. This shows that core orientation does not alter the pathways of the matrix of the porous media when fluids traverse through the porous medium, further attributing the fluid behaviour to mainly a function of the fluid properties and not the porous media. This was also true for *Buff Berea* depicted in Figures 12 and 13, and a comparison between the vertical and horizontal dispersivity is shown in Figure 14.

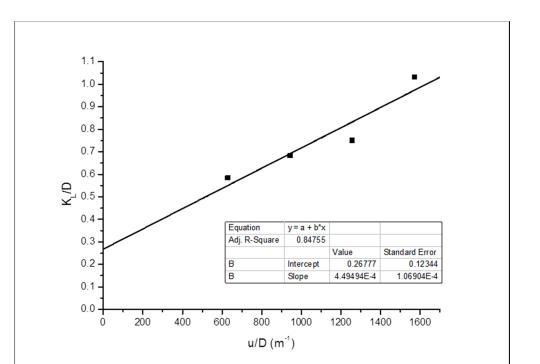


Figure 9. Dispersion to diffusion coefficient ratio against interstitial velocity for Grey Berea (vertical).

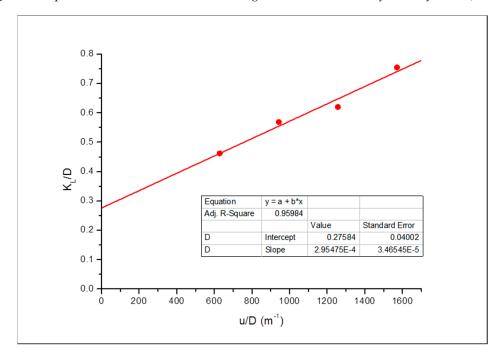


Figure 10. Dispersivity evaluation using the dispersion coefficient and interstitial velocity for Grey Berea (horizontal).

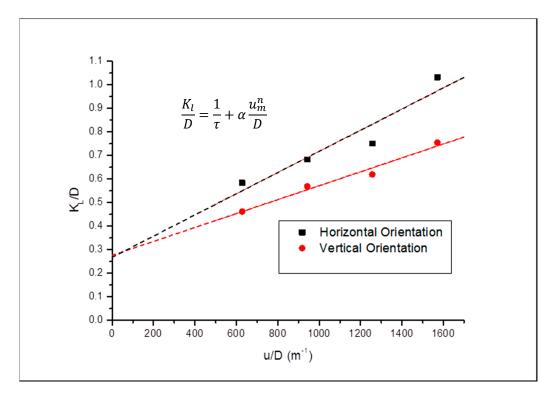


Figure 11. Comparison of the dispersivities in both orientations for Grey Berea.

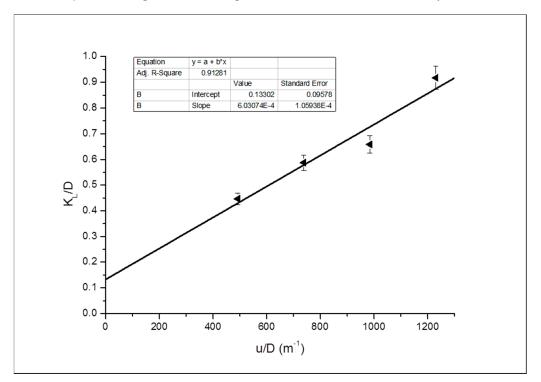


Figure 12. Dispersivity evaluation for Buff Berea in the horizontal orientation.

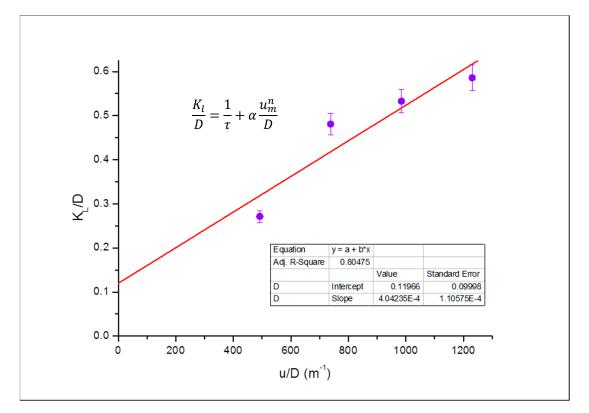


Figure 13. Dispersivity evaluation for Buff Berea in the vertical orientation.

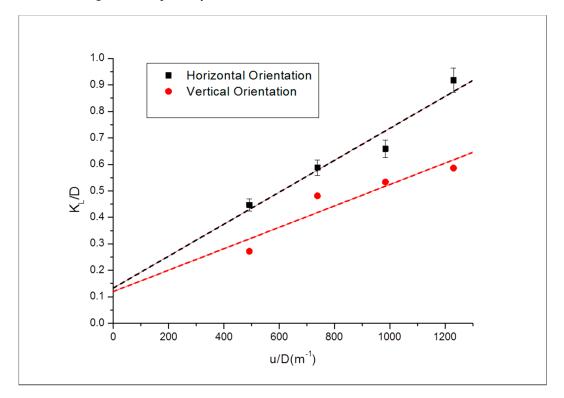


Figure 14. Comparison of the dispersivities in both orientations for Buff Berea.

Bandera Grey core sample exhibited similar trend as the previous core samples discussed in terms of dispersivity as shown in Figures 15–17. Dispersivity, however, is highest in *Bandera Grey* compared to the other core samples. This can be attributed to the grain arrangement of the core sample and the structure of the pore matrix which is tightly packed with its characteristic low permeability.

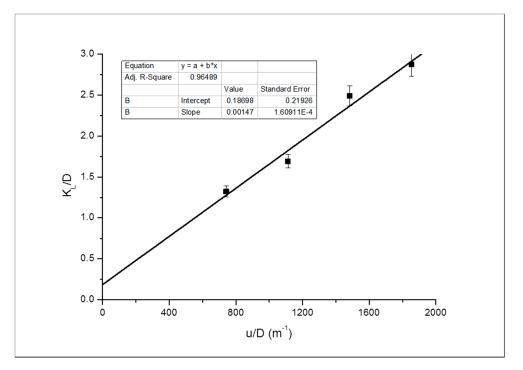


Figure 15. Dispersivity evaluation for Bandera Grey in the horizontal orientation.

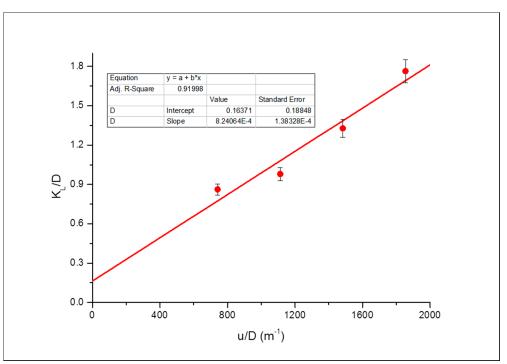


Figure 16. Dispersivity evaluation for Bandera Grey in the vertical orientation.

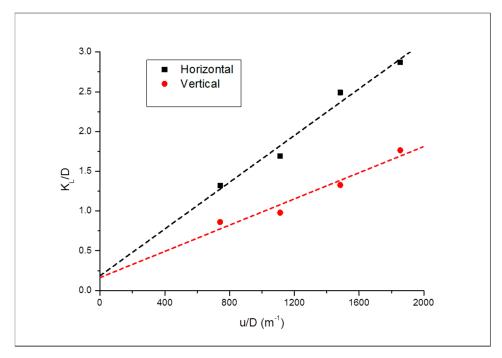


Figure 17. Comparison of the dispersivities in orientations for Bandera Grey.

There certainly are similarities between the vertical and horizontal orientations in all the core samples used in this study in terms of dispersion coefficient. This discussion is highlighted in the next section, which summarises the dispersion coefficient variation with the permeability.

Summary of the Dispersion Coefficient Investigation

A summary of the dispersion coefficients for all the core samples is shown in Table 5.

Core Sample	<i>U</i> (μm/s)	$K_L (10^{-8} \text{ m}^2/\text{s})$	
		Horizontal	Vertical
	39.3	7.01	4.56
$P_{audaua}(k - 16.09 \text{ md})$	59.0	8.97	5.19
Bandera ($k = 16.08 \text{ md}$)	78.6	13.20	7.03
	98.3	15.23	9.35
	33.5	3.11	2.46
<i>Grey Berea (k = 217.04 md)</i>	50.3	3.64	3.03
Grey bereu ($\kappa = 217.04 \text{ md}$)	67.0	4.01	3.70
	83.8	5.51	4.02
	26.2	2.38	1.44
<i>Buff Berea (k = 560.63 md)</i>	39.3	3.13	2.56
<i>Dujj Dereu (k – 500.03 IIId)</i>	52.5	3.51	2.84
	65.6	4.89	3.12

Table 5.	Summary	of all	dispersion	coefficients.
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Dispersion coefficient generally decreases with increase in permeability as seen in Table 5. Hence, the core sample with the least permeability (*Bandera Grey*) showed a significantly higher dispersion coefficient. Another realisation from the Table is that in all the runs, those in the horizontal orientation appear to have the higher dispersion coefficient compared to the vertical counterparts. This can be attributed to the effect of gravity on the CO_2 as it traverses the core sample. Also, since the interstitial velocity is a function of porosity, the core sample with the most porosity had the lowest interstitial velocity and, hence, a lower dispersion coefficient at lower injection rates. The dispersion

coefficient increases significantly at higher injection rates in all the runs regardless of the orientation. Furthermore, in the higher permeability core sample, the interaction between the injected CO_2 and the nascent CH_4 was limited as the CO_2 interstitial velocity was lower and, thus, the dispersion coefficient was lower compared to the subsequent core samples. The opposite was realised in the low permeability sample with the highest interstitial velocity was characterised by higher agitation of the gas molecules within the porous medium which eventually led to higher interaction and mixing and of course higher dispersion coefficient—the rate of mixing. Permeability is one of the vast factors that influence dispersion.

The dispersivity also increases with increase in permeability. This being a function of the core sample is evident with this trend, as the absolute permeability of the core sample is also a property of the core sample. Basically, the higher the permeability of the core sample, the higher the rate of mixing when CO_2 is injected to displace CH_4 . However, dispersivity is scale-dependent [20,21] and, albeit being at laboratory scale, this finding is an indication of the effects of the petrophysical properties on the mixing taking place during EGR. Finding the right injection scenario is vital in achieving the best recovery efficiency whilst storing substantial volumes of CO_2 .

4. Conclusions

The effects of permeability on the dispersion coefficient have been shown to be significant during the displacement of CH_4 by supercritical CO_2 . This effect was more pronounced in the horizontal orientation compared to the vertical orientation which was attributed to the gravity effects on the supercritical CO₂ as it traverses the core sample. The magnitude difference in the dispersion coefficients was about 20%–30% higher in the horizontal orientation compared to the vertical orientation. Bandera Grey, with the lowest permeability and porosity, exhibited a wider contrast (40%-50%) between the dispersion coefficients in both orientations as a result of higher interstitial velocity and tortuosity of the core sample compared to the other core samples tested. Therefore, the effect of permeability and orientation on mixing between the displacing and displaced gas during EGR are vital. This study highlights the influence of the permeability variation and plume propagation orientation on the mixing between CO₂ and CH₄ during EGR. It provides reservoir engineers with an insight into characterising gas systems during the design of the EGR technology. The injection rates, injection pressures, and dip angle of injection from the injector to the producer can be evaluated accurately during simulation studies prior to field scale application and implementation. Inclusion of such hysteresis will provide a better representation of the injection process given that the main reason why EGR has not been widely adopted is the its economic drawback brought on by incessant mixing and contamination of the recovered CH₄.

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