

Supplementary Material

Mn-Fe layered double hydroxide intercalated with ethylene-diaminetetraacetate anion: Synthesis and application for removal of As(III) from aqueous solution around pH 2-11

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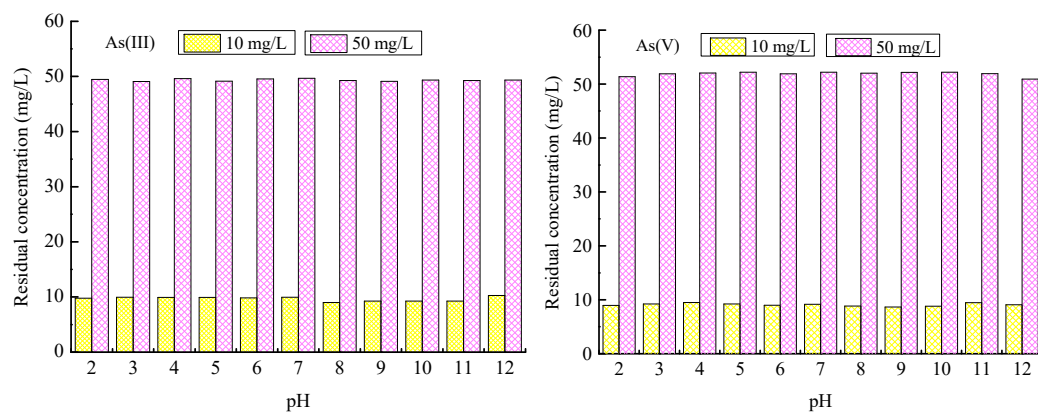


Figure S1. Effect of pH on the residual concentration of 10 and 50 mg/L As(III/V) stock solution (temperature $25\pm1^\circ\text{C}$).

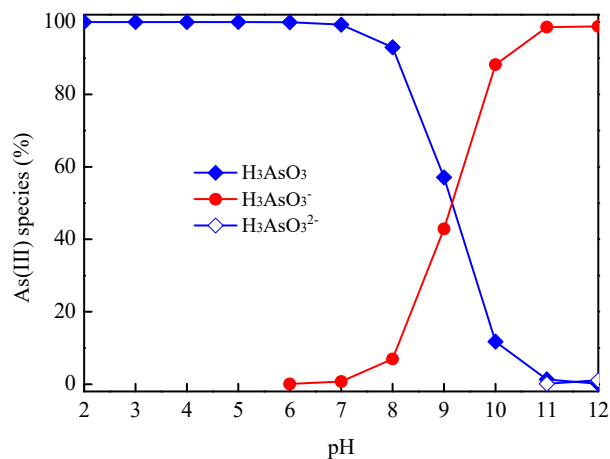


Figure S2. As(III) species with the change of the solution pH (As(III)=50 mg/L; chloride ion strength of 0.01 mol/L).

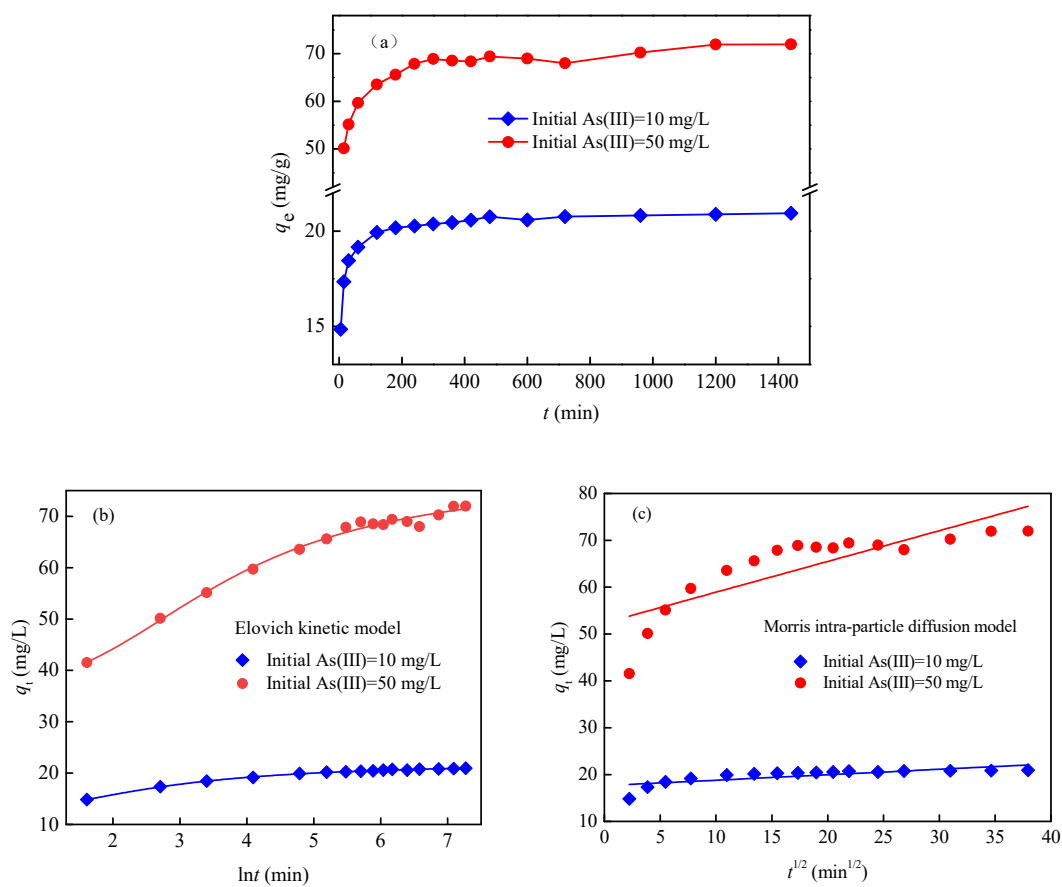


Figure S3. Effect of contact time on As(III) adsorption onto EDTA@MF-LDHs (a); Fitting with Elovich kinetic model (b); Fitting with the Morris intra-particle diffusion model (c). (Initial As(III) concentration 10 and 50 mg/L; sorbent dose 0.5 g/L; pH 7.0±0.2; temperature 25±1°C)

Table S1. EDS analysis of EDTA@MF-LDHs before and after As(III) adsorption (%).

EDTA@MF-LDHs	elements	region 1	region 2	region 3	average
before	C	29.58	31.40	31.04	30.67
	O	21.66	16.38	12.37	16.80
	Cl	2.93	2.69	2.63	2.75
	Mn	29.54	32.57	35.92	32.68
	Fe	16.30	16.96	18.04	17.10
after	C	7.26	9.66	16.94	11.29
	O	33.27	37.31	36.95	35.84
	Cl	–	–	–	–
	Mn	31.86	24.97	23.21	26.68
	Fe	19.10	18.56	14.47	17.38

Note:-, not detected

Table S2. Kinetics parameter for As(III) adsorption onto EDTA@MF-LDHs

Elovich constant			
Initial As(III) concentration (mg/L)	a_e (mg/g·min)	b_e (g/mg)	R^2
10	5.10×10^{-6}	1.053	0.998
50	0.007	0.195	0.990
Morris intra-particle diffusion constant (fitting data with single straight line)			
Initial As(III) concentration (mg/L)	k_d (mg/(g·min ^{1/2}))	R^2	
10	0.116	0.563	
50	0.656	0.665	