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New Antibacterial Phenone Derivatives Asperphenone A–C from Mangrove-Derived Fungus *Aspergillus* sp. YHZ-1

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Abstract: Marine fungi are a promising source of novel bioactive natural products with diverse structure. In our search for new bioactive natural products from marine fungi, three new phenone derivatives, asperphenone A–C (**1–3**), have been isolated from the ethyl acetate extract of the fermentation broth of the mangrove-derived fungus, *Aspergillus* sp. YHZ-1. The chemical structures of these natural products were elucidated on the basis of mass spectrometry, one- and two-dimensional NMR spectroscopic analysis and asperphenone A and B were confirmed by single-crystal X-ray crystallography. Compounds **1** and **2** exhibited weak antibacterial activity against four Gram-positive bacteria, *Staphylococcus aureus* CMCC(B) 26003, *Streptococcus pyogenes* ATCC19615, *Bacillus subtilis* CICC 10283 and *Micrococcus luteus*, with the MIC values higher than 32.0 μ M.

Keywords: mangrove endophyte; *Aspergillus* sp.; marine natural product; asperphenone; antibacterial activity

1. Introduction

Marine filamentous fungi are a rich source of antimicrobial compounds, anti-inflammatory, anticancer and antiviral agents [1–5]. Among all the marine-sourced fungi, the species in the genus *Aspergillus* (Trichocomaceae) are an important source for novel pharmacological metabolites such as polyketides, alkaloids and terpenoids [6–9]. The mangrove plants inhabit the intertidal zones in the tropics and subtropics, providing a very unique habitat for animals and microbes. Thus, the mangrove-associated fungi, with the majority coming from endophytic species, have attracted much attention to discover structurally diverse and bioactive secondary metabolites. In the past several years, our research group has explored many novel secondary metabolites from mangrove-derived endophytic fungi, such as sesquiterpenoids diaporols A-I, diaporine with regulation activity of macrophage differentiation and other polyketides [10–12]. As part of our ongoing search for novel biologically active natural products from microbe from special environment [13,14], we chemically investigated an endophytic fungus, *Aspergillus* sp. YHZ-1, from mangrove plant from Hainan Island, China. Fractionation of the ethyl acetate extract of its liquid fermentation broth led to the discovery of three new phenone derivatives, which we have named asperphenone A–C (**1–3**).

Compounds **1** and **2** were tested and displayed weak antibacterial activity against four Gram-positive bacteria. Herein, we report the isolation and structure elucidation of these new compounds and their antibacterial activity.

2. Results

A large-scale fermentation broth (50 L) of *Aspergillus* sp. YHZ-1 was collected and extracted with ethyl acetate three times to produce a crude extract. The subsequent fractionation by repeated column chromatography over silica gel, octadecylsilyl silica gel (ODS), Sephadex LH-20 and semi-preparative reversed-phase high performance liquid chromatography (HPLC) yielded three new phenone derivatives, asperphenone A–C (**1–3**) (Figure 1).

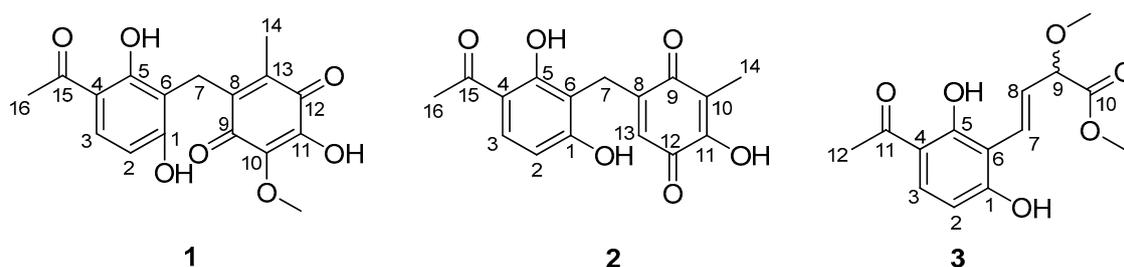


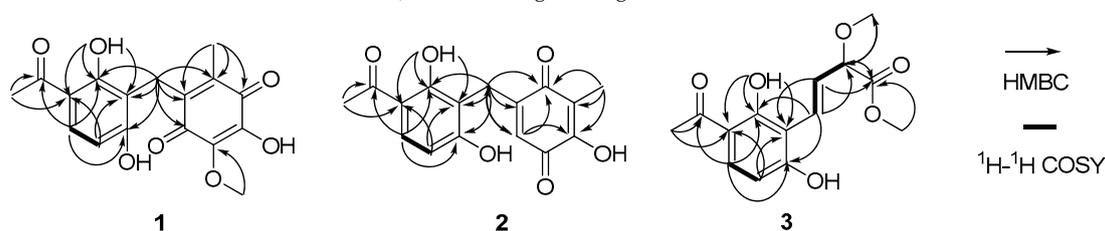
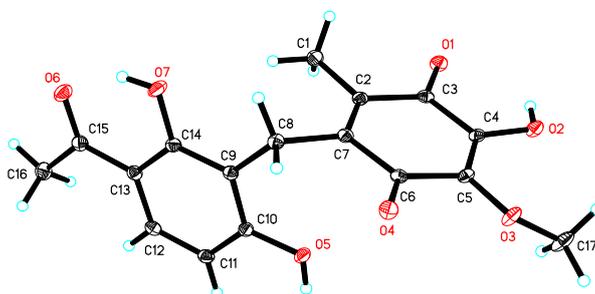
Figure 1. The structures of asperphenone A–C (**1–3**) isolated from *Aspergillus* sp. YHZ-1.

Asperphenone A (**1**) was isolated as a reddish-brown needle crystal. The molecular formula of **1** was determined to be $C_{17}H_{16}O_7$ on the basis of the high-resolution (HR) ESI-MS (m/z 355.2716 $[M + Na]^+$, calcd. 355.2706) along with the 1H and ^{13}C NMR data (Table 1), indicating ten degrees of unsaturation. The 1H , ^{13}C and HSQC NMR spectra (in supplementary materials) in $DMSO-d_6$ revealed signals of two methyls (δ_C/δ_H 11.6/1.73, C-14/ H_3 -14; δ_C/δ_H 26.6/2.52, C-16/ H_3 -16), one oxygenated methyl (δ_C/δ_H 60.3/3.78, 10-OMe), one methylene (δ_C/δ_H 19.4/3.70, C-7/ H_2 -7), one ketone carbon (δ_C 203.8, C-15), two α,β -unsaturated carbonyl carbons (δ_C 182.7, C-9; δ_C 185.1, C-12), three phenolic hydroxyl protons (δ_H 10.25, 1-OH; δ_H 13.21, 5-OH; δ_H 10.85, 11-OH) and ten aromatic/olefinic carbons, whose chemical shift values indicated the presence of one 1,2,3,4-tetrasubstituted benzene ring (δ_C 162.7, C-1; δ_C/δ_H 107.8/6.42 (d, $J = 8.8$ Hz), C-2/ H -2; δ_C/δ_H 131.8/7.67 (d, $J = 8.8$ Hz), C-3/ H -3; δ_C 112.8, C-4; δ_C 162.4, C-5; δ_C 112.0, C-6) and four quaternary carbons (δ_C 136.0, C-8; δ_C 138.8, C-10; δ_C 143.3, C-11; δ_C 144.5, C-13). The tetrasubstituted benzene ring was determined as a 1,3-dihydroxy-4-acetophenone moiety, which was deduced by analysis of the 1H - 1H COSY correlation from H-2 to H-3 and the HMBC correlations from H-2 to C-4 and C-6, from H-3 to C-1, C-5 and C-15, from H_3 -16 to C-4 and C-15 and from 5-OH to C-4, C-5 and C-6 (Figure 2). HMBC correlations from H_2 -7 to C-1, C-5 and C-6 permitted the linkage of C-7 to C-6 of the acetophenone moiety. The presence of a *p*-benzoquinone moiety could be deduced from the chemical shifts of C8–C13 in **1**, which was partially confirmed by the observation of HMBC correlations from H_3 -14 to C-8, C-12 and C-13 and from H_2 -7 to C-8, C-9 and C-13. Meanwhile, the connectivity of the *p*-benzoquinone and acetophenone moieties via the methylene C-7 was established by the relevant HMBC correlations. However, the positions of the hydroxyl group and methoxyl group on the *p*-benzoquinone moiety cannot be determined as their 2D correlations cannot be observed. Finally, a high-quality single crystal of **1** was obtained and the full structure of **1** was determined by the X-ray crystallographic analysis in a Cu $K\alpha$ radiation in low temperature (Figure 3), which also confirmed the proposed structure of **1**.

Table 1. ^1H and ^{13}C NMR data for compounds 1–3.

No.	1 ^a		2 ^a		3 ^b	
	δ_{H} , Mult. (J in Hz)	δ_{C} , Type	δ_{H} , Mult. (J in Hz)	δ_{C} , Type	δ_{H} , Mult. (J in Hz)	δ_{C} , Type
1		162.7, C		162.3, C		163.3, C
1-OH	10.25 ^c , brs		10.75, brs			
2	6.42, d (8.8)	107.8, CH	6.52, d (8.8)	107.4, CH	6.55, d (8.6)	108.3, CH
3	7.67, d (8.8)	131.8, CH	7.77, d (8.8)	132.1, CH	7.71, d (8.6)	132.8, CH
4		112.8, C		112.5, C		113.8, C
5		162.4, C		162.7, C		164.4, C
5-OH	13.21, s		13.09, s		13.73, s	
6		112.0, C		109.5, C		111.4, C
7	3.70, s	19.4, CH ₂	3.61, d (2.0)	22.0, CH ₂	7.07, dd (16.2, 0.8)	124.1, CH
8		136.0 ^d , C		148.5, C	6.78, dd (16.2, 7.0)	128.6, CH
9		182.7, C		187.2, C	4.42, dd (7.0, 0.8)	83.3, CH
9-OMe					3.37, s	57.0, CH ₃
10		138.8, C		116.9, C		171.7, C
10-OMe	3.78, s	60.3, CH ₃			3.70, s	52.0, CH ₃
11		143.3, C		153.8, C		204.3, C
11-OH	10.85 ^c , br s		10.75, brs			
12		185.1, C		183.0, C	2.56, s	26.3, CH ₃
13		144.5 ^d , C	5.76, t (2.0)	127.1, CH		
14	1.73, s	11.6, CH ₃	1.82, s	8.15, CH ₃		
15		203.8, C		203.4, C		
16	2.52, s	26.6, CH ₃	2.55, s	26.2, CH ₃		

^a Acquired at 400 MHz for ^1H NMR and 100 MHz for ^{13}C NMR in DMSO-*d*₆; ^b Acquired at 600 MHz for ^1H NMR and 150 MHz for ^{13}C NMR in acetone-*d*₆; ^{c,d} Interchangeable signals.

**Figure 2.** The key 2D NMR correlations of asperphenone A–C (1–3).**Figure 3.** X-ray crystal structure of 1.

Asperphenone B (2) was obtained as a yellow needle crystal, with the molecular formula C₁₆H₁₄O₆ (ten degrees of unsaturation) as derived from the HRESIMS data (*m/z* 325. 1326 [M + Na]⁺, calcd. 325.0683). The ^{13}C NMR spectrum showed sixteen carbon signals for two methyls (C-14 and C-16), one methylene (C-7), one ketone carbon (C-15), two α,β -unsaturated carbonyl carbons (C-9 and C-12)

and ten aromatic/olefinic carbons. Comparison of the MS, ^1H and ^{13}C NMR data of **2** with those of **1** revealed the presence of the same 1,3-dihydroxy-4-acetophenone moiety as **1** and a different substituted *p*-benzoquinone moiety. Further HMBC correlations (Figure 2) from H₃-14 (δ_{H} 1.82) to C-9 (δ_{C} 187.2), C-10 (δ_{C} 116.9) and C-11 (δ_{C} 153.8), from H-13 (δ_{H} 5.76) to C-7 (δ_{C} 22.0), C-8 (δ_{C} 148.5), C-9 and C-11 and from H₂-7 (δ_{H} 3.61) to C-8, C-9 and C-13 (δ_{C} 127.1) along with the unsaturation requirement of **2** confirmed the substituted pattern of the *p*-benzoquinone moiety. Finally, the connection between the two moieties mentioned above was through C-7 by interpretation of the relevant HMBC correlations from H₂-7. Detailed NMR analysis allowed the assignment of **2** as shown in Figure 1, which was also confirmed by a low-temperature single-crystal X-ray diffraction experiment (Figure 4).

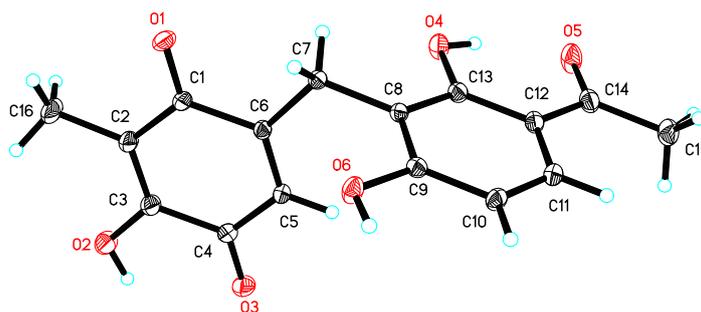


Figure 4. X-ray crystal structure of **2**.

Asperphenone C (**3**) was isolated as a colorless needle crystal that analyzed for the molecular formula $\text{C}_{14}\text{H}_{16}\text{O}_6$ (seven degrees of unsaturation) by HRESIMS (m/z 303. 1215 $[\text{M} + \text{Na}]^+$, calcd. 303.0839) in combination with ^{13}C NMR data (Table 1). The ^{13}C and DEPT135 NMR spectra of **3** displayed signals for fourteen carbons, including one methyl carbon (δ_{C} 26.3, C-12), two oxygenated methyl carbons (δ_{C} 57.0, 9-OMe; δ_{C} 52.0, 10-OMe), one oxygenated methine (δ_{C} 83.3, C-9), one ketone carbon (δ_{C} 204.3, C-11), one carboxylic carbon (δ_{C} 171.7, C-10) and eight aromatic/olefinic carbons. In this molecule, the same moiety, 1,3-dihydroxy-4-acetophenone as found in **1**, could be deduced through analysis of the ^1H - ^1H COSY correlation from H-2 (δ_{H} 6.55, d, $J = 8.6$ Hz) to H-3 (δ_{H} 7.71, d, $J = 8.6$ Hz) and HMBC correlations from H-2 to C-4 (δ_{C} 113.8) and C-6 (δ_{C} 111.4), from H-3 to C-1 (δ_{C} 163.3), C-5 (δ_{C} 164.4) and C-11, from 5-OH (δ_{H} 13.73) to C-4, C-5 and C-6 and from H₃-12 (δ_{H} 2.56) to C-4 and C-11. ^1H - ^1H COSY correlations from H-7 (δ_{H} 7.07, dd, $J = 16.2, 0.8$ Hz) to H-8 (δ_{H} 6.78, dd, $J = 16.2, 7.0$ Hz) and from H-8 to H-9 (δ_{H} 4.42, dd, $J = 7.0, 0.8$ Hz) and HMBC cross-peaks from H-8 to the carboxylic carbon C-10 established the fragment C-7-C-8-C-9-C-10, which was attached to the acetophenone moiety through C-6-C-7 linkage on the basis of the HMBC correlations from H-8 to C-6 and from H-7 to C-1 and C-5. $^3J_{\text{C-H}}$ diagnostic correlations from the oxygenated methyl protons at δ_{H} 3.37 to C-9 and from the oxygenated methyl protons at δ_{H} 3.70 to C-10 demonstrated that the methoxyl groups should be located at C-9 and C-10, respectively. Thus, the structure of compound **3** was elucidated as shown in Figure 1. However, the absolute configuration of C-9 has not been determined.

In the primary screen for antibacterial compounds, compounds **1** and **2** were tested against four Gram-positive bacteria, *Staphylococcus aureus* CMCC(B) 26003, *Streptococcus pyogenes* ATCC19615, *Bacillus subtilis* CICC 10283 and *Micrococcus luteus*. As summarized in Table 2, Both of these two compounds showed weak activity against the tested bacteria, *Staphylococcus aureus* CMCC(B) 26003, *Streptococcus pyogenes* ATCC19615, *Bacillus subtilis* CICC 10283 and *Micrococcus luteus*, with the MIC values higher than 32.0 μM .

Table 2. Antibacterial activities of compounds **1** and **2** (MIC, μM).

Compounds	<i>S. aureus</i>	<i>B. subtilis</i>	<i>S. pyogenes</i>	<i>M. luteus</i>
1	64.0	64.0	64.0	32.0
2	32.0	64.0	32.0	32.0
Ampicillin	4.0	8.0	2.0	1.0

3. Materials and Methods

3.1. General Experimental Procedures

The HR-ESI-MS data were obtained on an Agilent 6210 time of flight LC-MS instrument (Agilent Technologies Inc., Palo Alto, CA, USA). NMR experiments were conducted on a Bruker DPX-400 NMR spectrometer (400 MHz for ^1H NMR and 100 MHz for ^{13}C NMR) or Bruker DRX-600 spectrometer (600 MHz for ^1H NMR and 150 MHz for ^{13}C NMR) (Bruker Corporation, Karlsruhe, Germany). The chemical shifts were given in δ (ppm) and referenced to the solvent signal (DMSO- d_6 , δ_{H} 2.50, δ_{C} 39.5; acetone- d_6 , δ_{H} 2.05, δ_{C} 29.8). Column chromatography (CC) was accomplished on silica gel (200–300 mesh, Qingdao Marine Chemical Inc., Qingdao, China), ODS (40–70 μm , Merck Company, Darmstadt, Germany) and Sephadex LH-20 (GE Healthcare Bio-Sciences AB, Uppsala, Sweden). Semi-preparative reverse-phase (RP) HPLC was performed on a Hitachi HPLC system with a L-7110 pump, a L-7420 UV/vis detector and an Hypersil RP-C₁₈ column (5 μm , 250 \times 10.0 mm, Thermo Fisher Scientific, Waltham, MA, USA). Thin-layer chromatography (TLC) was conducted on silica gel GF254 (10–20 μm , Qingdao Marine Chemical Inc., Qingdao, China). All chemicals used were of HPLC grade or analytical grade.

3.2. Strain Isolation and Cultivation

The fungus *Aspergillus* sp. YHZ-1 was isolated by one of the authors (Y.-Q.Z.) from an unidentified mangrove plant from Hainan Island, China, in October 2015. The isolate was identified as *Aspergillus* sp. by its morphological characteristics and the voucher specimen (IFB-YHZ-1) was deposited in the Institute of Functional Biomolecules, State Key Laboratory of Pharmaceutical Biotechnology, Nanjing University. The strain was cultivated on MEA agar plate (consisting of 20 g/L malt extract, 20 g/L sucrose, 1 g/L peptone, 20 g/L agar and deionized water) at 30 °C for 5 days. Then small agar plugs with mycelia were inoculated into five 1 L-Erlenmeyer flasks, each containing 400 mL MEA liquid medium, which were cultivated at 28 °C with 108 rpm/min. After 3 days of fermentation, 10 mL of the seed cultures were inoculated into 200 flasks with 250 mL ME liquid medium and fermented on a rotary shaker with 120 rpm/min at 28 °C for 14 days.

3.3. Extraction and Isolation

The entire filtrate of the fermented broth (50 L) was harvested and extracted three times with an equivalent volume of ethyl acetate at room temperature. The organic solvent was evaporated in vacuo to yield 25 g of crude extract. Then the crude extract was fractionated by silica gel (200–300 mesh) CC eluting with a gradient of CH_2Cl_2 -MeOH mixtures (v/v , 100:0, 100:1, 100:2, 100:4, 100:8, 100:16, 100:32, 0:100) to give 8 fractions (Fr.1–8). Fr.3 (CH_2Cl_2 -MeOH, v/v , 100:2) was subsequently subjected to ODS CC with MeOH- H_2O (v/v , 30:70, 40:60, 50:50, 60:40, 70:30, 100:0) to afford 6 sub fractions (Fr.3.1–Fr.3.6). Finally, compounds **1** (12.5 mg), **2** (11.2 mg), **3** (6.3 mg) were purified from Fr.3.4 and Fr.3.5 by semi-preparative reverse-phase HPLC (2 mL/min, detector UV λ_{max} 254 nm, CH_3CN - H_2O 55:45).

3.4. Crystal Data of **1**

The single crystal X-ray diffraction data of compound **1** was collected on a Bruker APEX-II diffractometer at 130 K with Cu $K\alpha$ radiation ($\lambda = 1.54178 \text{ \AA}$). The structure was solved using the program SHELXS-97 and refined by full-matrix least-squares on F^2 . Crystal data of compound **1** have

been deposited with the Cambridge Crystallographic Data Centre (deposition No. CCDC 1584390), which can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif.

Crystal data for **1**: molecular formula $C_{17}H_{16}O_7$, $M_r = 332.30$, monoclinic crystals, $a = 13.2620$ (3) Å, $b = 16.2202$ (4) Å, $c = 7.0447$ (2) Å, $\alpha = 90.00^\circ$, $\beta = 97.7510^\circ$ (10), $\gamma = 90.00^\circ$, $Z = 4$, $\mu = 0.977 \text{ mm}^{-1}$, $F(000) = 696$ and $T = 130 \text{ K}$; Crystal dimensions: $0.15 \times 0.1 \times 0.05 \text{ mm}^3$, Volume = 1501.56 (7) Å³, 10,794 reflections measured, 2783 independent reflections ($R_{int} = 0.0472$), the final R indices [$I > 2\sigma(I)$] $R_1 = 0.0469$, $wR_2 = 0.1286$, R indices (all data) $R_1 = 0.0500$, $wR_2 = 0.1330$. The goodness of fit on F^2 was 1.026.

3.5. Crystal Data of **2**

The single crystal data of compound **2** was collected on a Bruker APEX-II diffractometer at 130 K with Cu K α radiation ($\lambda = 1.54178 \text{ \AA}$). The structure was solved using the program SHELXS-97 and refined by full-matrix least-squares on F^2 . Crystal data of compound **2** have been deposited with the Cambridge Crystallographic Data Centre (deposition No. CCDC 1584387), which can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif.

Crystal data for **2**: molecular formula $C_{16}H_{14}O_6$, $M_r = 302.27$, monoclinic crystals, $a = 11.4611$ (4) Å, $b = 4.9487$ (2) Å, $c = 13.6864$ (5) Å, $\alpha = 90.00^\circ$, $\beta = 111.933^\circ$ (2), $\gamma = 90.00^\circ$, $Z = 2$, $\mu = 0.909 \text{ mm}^{-1}$, $F(000) = 316$ and $T = 130 \text{ K}$; Crystal dimensions: $0.22 \times 0.2 \times 0.15 \text{ mm}^3$, Volume = 720.07 (5) Å³, 4697 reflections measured, 1843 independent reflections ($R_{int} = 0.0350$), the final R indices [$I > 2\sigma(I)$] $R_1 = 0.0385$, $wR_2 = 0.1115$, R indices (all data) $R_1 = 0.0389$, $wR_2 = 0.1119$. The goodness of fit on F^2 was 1.095.

3.6. Antibacterial Activity Assay

The in vitro antibacterial activity of compounds **1** and **2** were evaluated against four bacteria including *Staphylococcus aureus* CMCC(B) 26003, *Streptococcus pyogenes* ATCC19615, *Bacillus subtilis* CICC 10283 and *Micrococcus luteus* in accordance with previously reported methods [14,15]. In the assays, the medium used in the test was Müller-Hinton (MH) broth. All test compounds and the positive control ampicillin were dissolved in dimethyl sulfoxide (DMSO). The minimum inhibitory concentration (MIC) values were determined in the 96-well plates (triplicate) and determined as the lowest sample concentration exhibiting no bacterial growth.

4. Conclusions

Three new phenone derivatives, asperphenone A–C (**1–3**), were isolated from the ethyl acetate extract of the fermentation broth of the mangrove-derived fungus *Aspergillus* sp. YHZ-1. The chemical structures of these compounds were elucidated on the basis of HR-ESI-MS, 1D and 2D NMR spectroscopic analysis, as well as X-ray crystallographic data. Both of the tested compounds **1** and **2** displayed weak antibacterial activity against four Gram-positive bacteria, *Staphylococcus aureus* CMCC(B) 26003, *Streptococcus pyogenes* ATCC19615, *Bacillus subtilis* CICC 10283 and *Micrococcus luteus*, indicating that the mangrove-associated fungi are still a rich source for discovering diverse new bioactive natural products which could be used as lead compounds in drug development.

Supplementary Materials: The 1D and 2D NMR spectra for compounds **1–3** are available online at <http://www.mdpi.com/1660-3397/16/2/45/s1>.

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Author Contributions: Rui-Hua Jiao and Hui-Ming Ge conceived and designed the experiments; Yi-Qin Zhou and Zhi-Kai Guo cultured, isolated and identified the compounds; Xin-Zhao Deng performed the biological tests; Zhi-Kai Guo, Yi-Qin Zhou, Hao Han, Wen Wang and Lang Xiang analyzed the data; Zhi-Kai Guo, Hui-Ming Ge and Rui-Hua Jiao wrote the paper.

Conflicts of Interest: The authors declare no conflict of interest.

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