

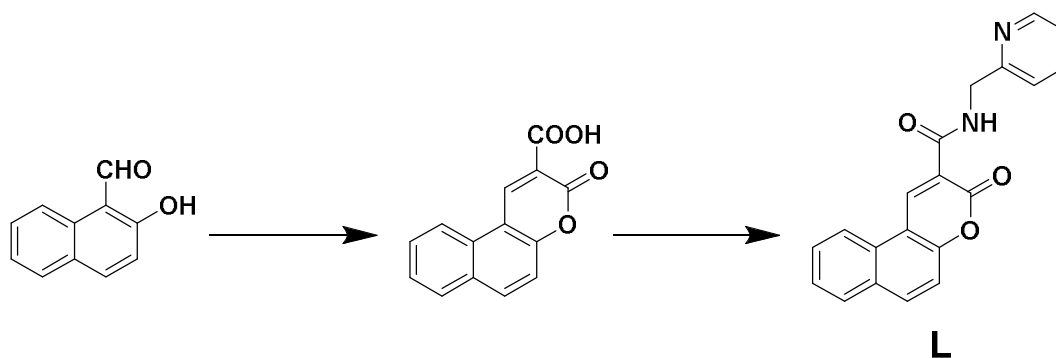
Supplementary Materials

A Copper (II) Ensemble Based Fluorescence Chemosensor and the Application in the “Naked–eye” Detection of Biothiols in Human Urine

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Scheme S1. Synthetic procedure of the fluorescent ligand, **L**

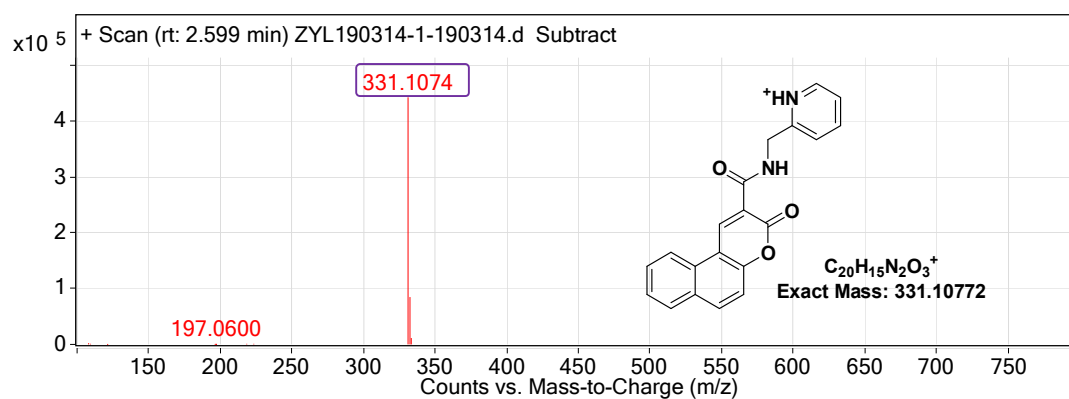


Figure S1. HRMS of fluorescent ligand, **L**.

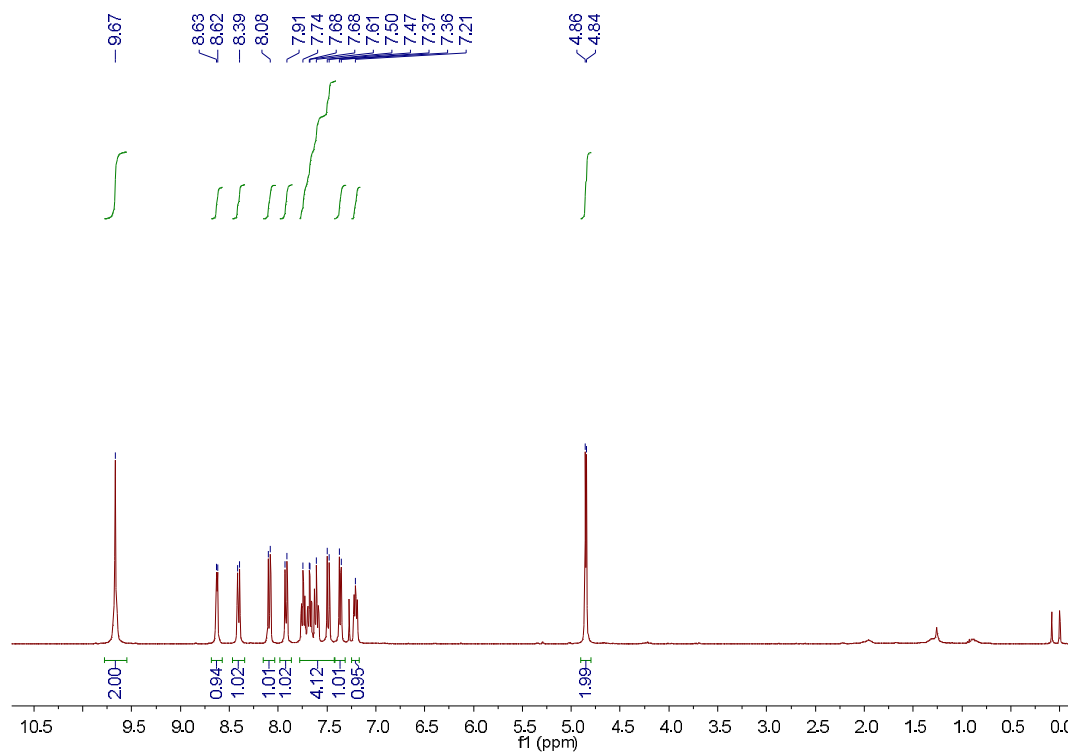


Figure S2. 1H NMR of fluorescent ligand, **L** ($CDCl_3$, 600 MHz)

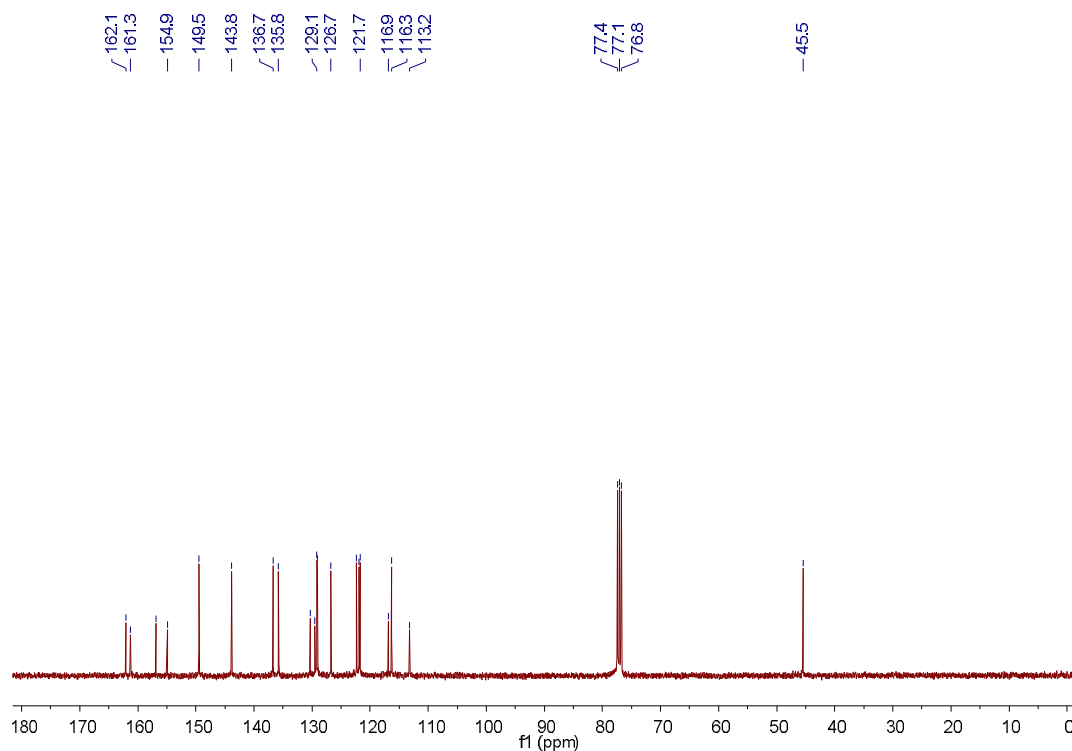


Figure S3. ^{13}C NMR of fluorescent ligand, **L** (CDCl_3 , 150 Hz).

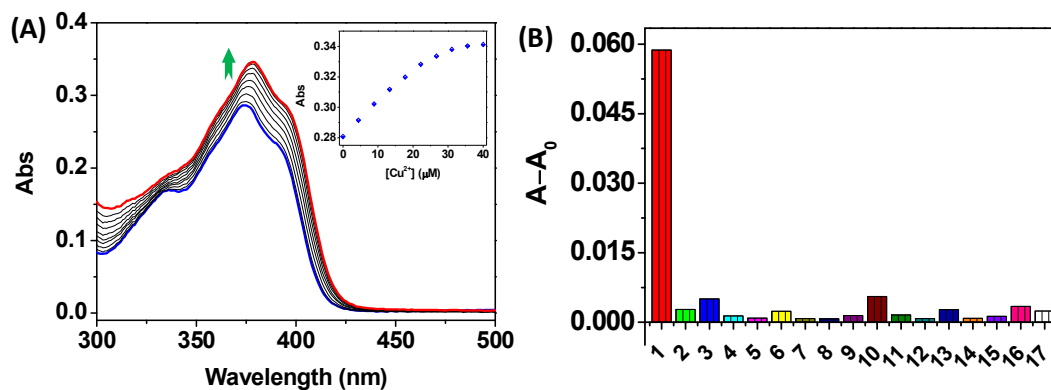


Figure S4. (A) UV-vis absorption spectra of **L** (10 μM) in the presence of different amounts of Cu^{2+} (0–60 μM) in the DMF/HEPES mixed solution (7:3, v/v, pH=7.4). Inset: normalized absorption intensities of **L** at 378 nm as a function of Cu^{2+} (0–60 μM). (B) UV-vis absorption spectra of **L** (10 μM) in the DMF/HEPES mixed solution (7:3, v/v, pH=7.4) upon addition of 60 μM various cations: 1. Cu^{2+} , 2. Pb^{2+} , 3. Ba^{2+} , 4. Ag^{2+} , 5. Al^{3+} , 6. Cd^{2+} , 7. Ca^{2+} , 8. Mg^{2+} , 9. Co^{2+} , 10. Fe^{3+} , 11. Cr^{2+} , 12. Ni^{2+} , 13. Hg^{2+} , 14. Li^{+} , 15. Na^{+} , 16. K^{+} , 17. Zn^{2+} . A_0 and A represent the absorption intensities of **L** at 378 nm in the absence and in the presence of Cu^{2+} .

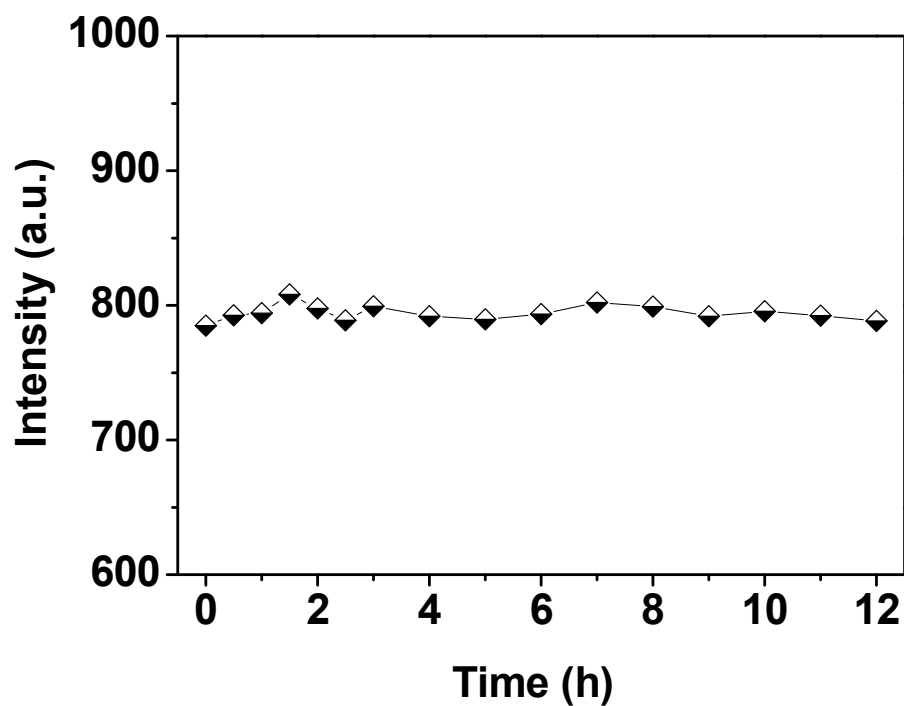


Figure S5. Fluorescence of **L** (10 μM) at different times in the DMF/HEPES mixed solution (7:3, v/v, pH=7.4). The intensities were recorded at 443 nm, and excitation was performed at 378 nm.

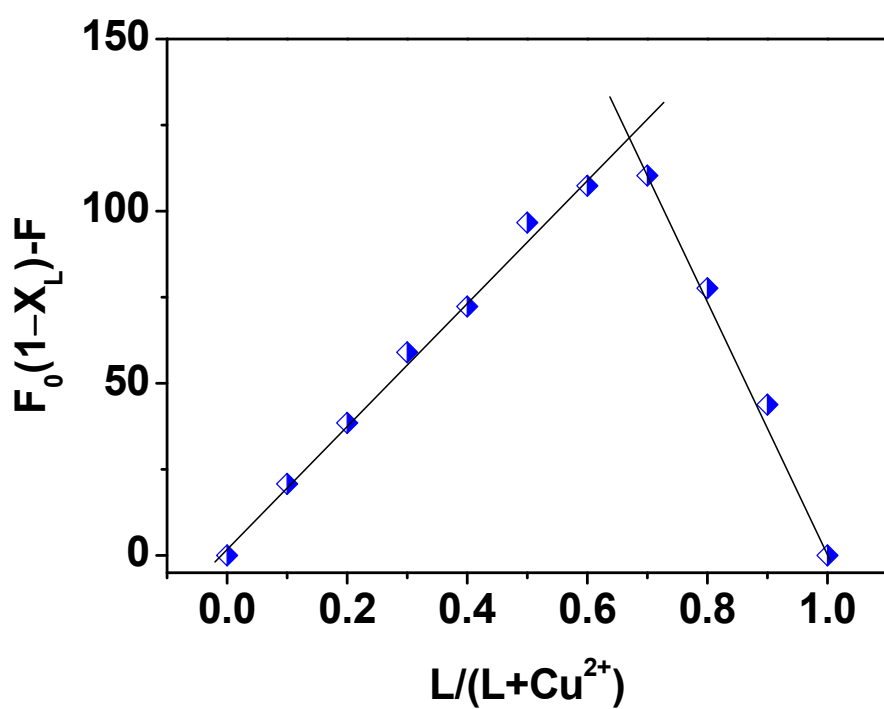


Figure S6. Job's plots according to the method for continuous variations. The total concentration of **L** (10 μM) and Cu^{2+} is 10 μM . X_L represents the proportion of **L** to the total amounts of **L** and Cu^{2+} . The intensities were recorded at 443 nm, and excitation was performed at 378 nm.

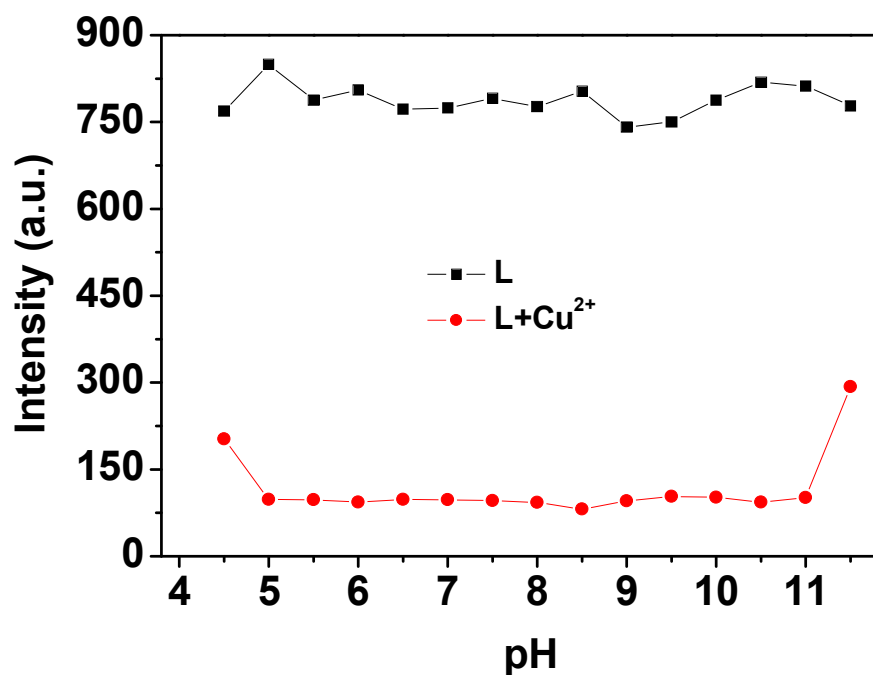


Figure S7. Effect of pH on the fluorescence intensities of **L** (10 μM) in the absence and presence of Cu²⁺ (60 μM) in the DMF/H₂O mixed solution. The intensities were recorded at 443 nm, and excitation was performed at 378 nm.

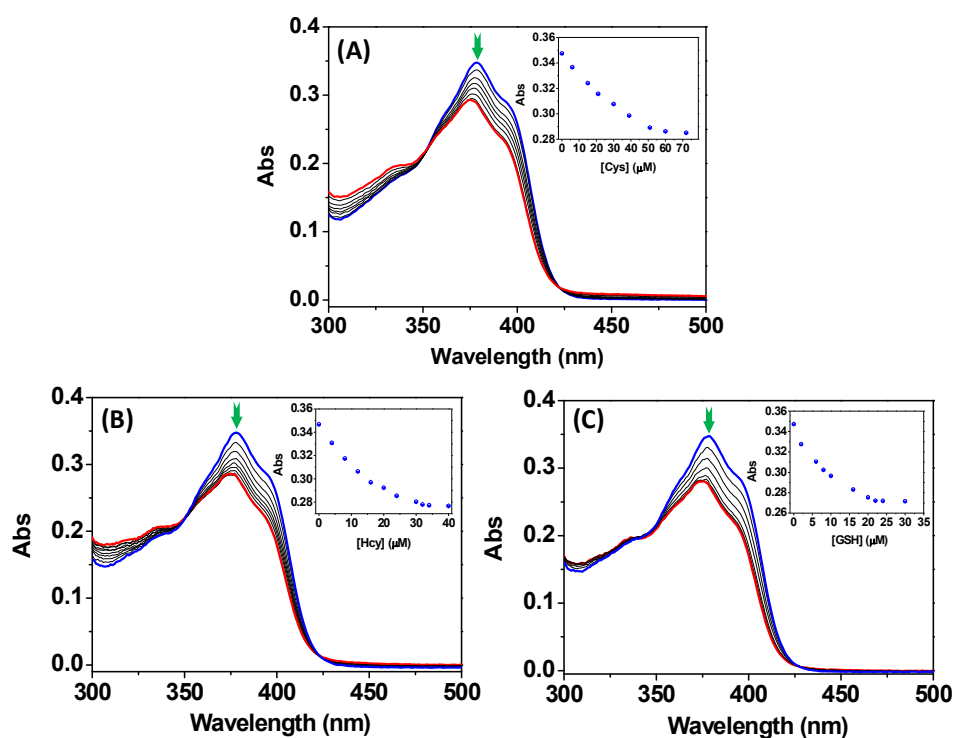


Figure S8. Absorption spectra of **L-Cu²⁺** (10 μM) in the presence of different amounts of Cys (0–70 μM), Hcy (0–40 μM), GSH (0–30 μM) in the DMF/HEPES mixed solution (7:3, v/v, pH=7.4).

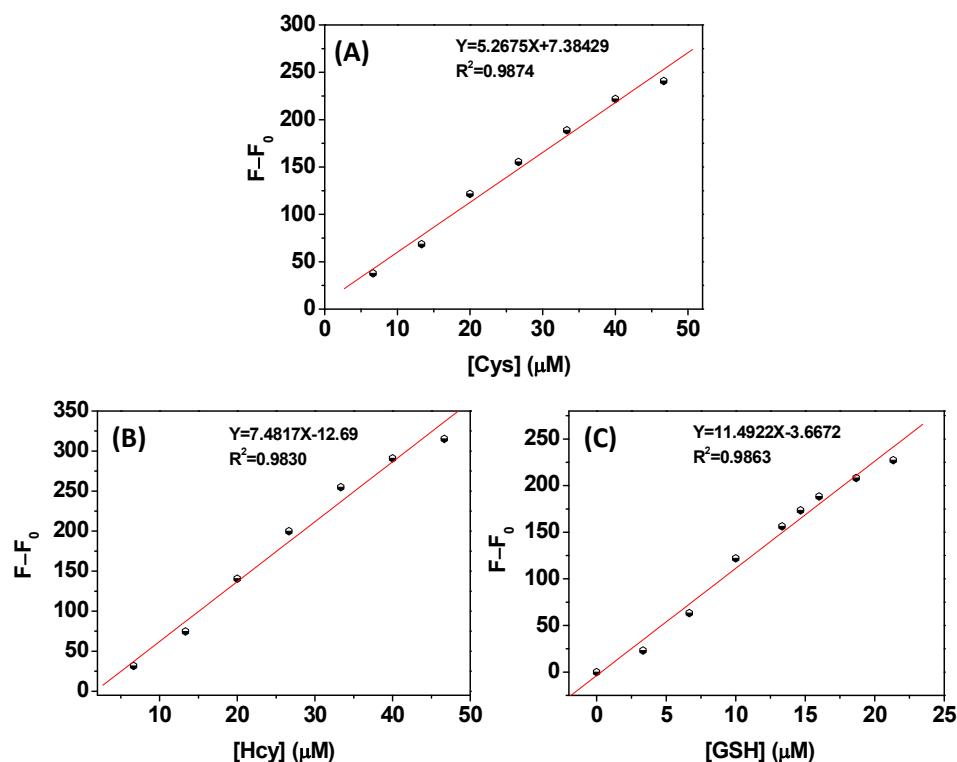


Figure S9. The linear relationship between fluorescence intensity of **L**-Cu²⁺ (1 μ M) at 443 nm versus the concentration of (A) Cys, (B) Hcy and (C) GSH in the DMF/HEPES mixed solution (7:3, v/v, pH=7.4). Excitation was performed at 378 nm. Intensities were recorded at 443 nm.

Table S1. Comparison of this work with reported fluorescent chemosensors for biothiols detection.

Probes	Selectivity	Detection limit	Response time	Detection of Biothiols by "Naked-eye"	Detection of biothiols in Human Urine	Test papers	Ref.
RhAN probe	GSH	0.1 μ M	< 5 s	Yes	No	No	S1
	biothiols	GSH 3.7 μ M	~ 120 min	No	No	No	S2
	biothiols	GSH 10.3 nM	< 5 s	Yes	No	No	S3
	Cys	1.4 nM	~ 30 min	No	No	No	S4
SATZ	Cys/Hcy	2.843 μ M (Cys)	12 min	Yes	No	No	S5
CI	Cys	0.014 μ M	25 min	Yes	No	No	S6
	Hcy	0.081 μ M	55 min				
	GSH	0.097 μ M	50 min				
BT-AC	Cys	36.2 nM	--	No	No	test strip	S7
	Cys	12 nM	40 min	No	No	No	S8
CPR	Hcy	13 nM	100 min				
	GSH	30 nM	38 min				

NIRHA	Cys	77.6 μ M	15 min	Yes	No	No	S9
SHCy-C	Cys	21.2 nM	5 min	No	No	No	S10
	Cys	31 μ M					
HN-NBD	Hcy	66 μ M	5 min	No	No	No	S11
	GSH	58 μ M					
CPA	Cys	49 nM	~3 min	No	No	No	S12
	Hcy	51 nM					
Nap-Cys	Cys	1.8 μ M	5 min	No	No	No	S13
PYR	Cys	22 nM	5min	Yes	No	No	S14
	Hcy	23nM	10min				
Lyso-RC	Cys	27 nM					
	Hcy	33 nM	--	No	No	No	S15
	GSH	16 nM					
	H ₂ S						
probe 1	Cys						
	Hcy	Cys 0.11 μ M	20 min	No	No	No	S16
	GSH						
EuTc-H₂O₂	Cys	100 nM					
	Hcy	200 nM	--	No	Yes	No	S17
	GSH	400 nM					

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