



# Short Note **Cocrystal of 4-Nitrophenol and 2,1,3-Benzoselenadizole**

Honghong Lan \*, Shaobin Miao and Weizhou Wang 💿

College of Chemistry and Chemical Engineering, Luoyang Normal University, Luoyang 471934, China; miaoshaobin@126.com (S.M.); wzw@lynu.edu.cn (W.W.) \* Correspondence: lantianlh@126.com; Tel.: +86-379-68618321

**Abstract:** The 1:1 cocrystal of 4-nitrophenol (NP) and 2,1,3-benzoselenadiazole (BSA) was successfully synthesized. The X-ray single-crystal diffraction analysis revealed that the structure contained a [Se–N]<sub>2</sub> cyclic supramolecular synthon. The synthons were connected into a one-dimensional ribbon by O–H…N hydrogen bonds and N–Se…O chalcogen bonds. Furthermore, adjacent ribbons were stabilized by the  $\pi$ … $\pi$  stacking interactions between two 2,1,3-benzoselenadiazole molecules, leading to the formation of a two-dimensional network.

Keywords: cocrystal; X-ray crystallography; non-covalent interaction

## 1. Introduction

In recent years, organic cocrystals of variable stoichiometry based on intermolecular interactions became an active area in crystal engineering [1–3]. Non-covalent interactions between the cocrystal-building units play very important roles for the formation of molecular aggregates [4–6]. Among these interactions, the hydrogen bond is the most important and appears in most crystal structures [7]. Simultaneously, a  $\pi \cdots \pi$  stacking interaction, halogen bond, and chalcogen bond are also extensively studied in supramolecular assembling [8]. The synthon approach was successfully used in the construction of supramolecular cocrystals. The aromatic heterocyclic compound 2,1,3-benzoselenadiazole (BSA) has received considerable attention for its ability to form a dimeric homosynthon based on a pair of Se…N chalcogen bonds, showing a short Se…N distance (2.80~3.26 Å) in a series of cocrystal compounds [9]. Meanwhile, The supramolecular heterosynthon including 2,1,3-benzoselenadiazole was also found in crystal engineering as a useful synthetic tool [10,11]. In this context, 2,1,3-benzoselenadiazole is often used as building block in a variety of supramolecular architectures; here, 2,1,3-benzoselenadiazole was used in the reaction with 4-nitrophenol (NP), and a new supramolecular cocrystal compound (1) was assembled and structurally characterized.

### 2. Results

The synthesized supramolecular compound was crystallized in a monoclinic  $P2_1/c$  space group, having an equimolar ratio (1:1) of 2,1,3-benzoselenadiazole and 4-nitrophenol in the asymmetric unit (Figure 1). In 1, two BSA molecules are held together by a pair of Se…N chalcogen bonds [d(Se…N) 2.88 Å], generating the [Se–N]<sub>2</sub> cyclic supramolecular synthon. Each synthon is bonded to four NP molecules via two intermolecular O–H…N hydrogen bonds [d(H…N) 1.96 Å] and two N–Se…O chalcogen bonds [d(Se…O) 3.37 Å]. With these interactions, adjacent BSA and NP molecules are linked together into an infinite one-dimensional chain, in which BSA synthons and NP molecules appear alternately. Furthermore, the chains are stabilized by the head-to-head  $\pi \dots \pi$  stacking interactions between the BSA molecules in two chains, forming a two-dimensional supramolecular network, as shown in Figure 2.



Citation: Lan, H.; Miao, S.; Wang, W. Cocrystal of 4-Nitrophenol and 2,1,3-Benzoselenadizole. *Molbank* 2023, 2023, M1685. https:// doi.org/10.3390/M1685

Academic Editor: René T. Boeré

Received: 27 March 2023 Revised: 9 June 2023 Accepted: 29 June 2023 Published: 3 July 2023



**Copyright:** © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/).



**Figure 1.** The molecular structure of **1** showing atom labeling and displacement ellipsoids at the 50% probability.



Figure 2. The 2D structure of 1.

## 3. Materials and Methods

#### 3.1. General Information

All chemicals and solvents were purchased from commercial sources and used without purification. The X-ray diffraction data were collected at 291 K on an Oxford Diffraction Gemini (Rigaku Corporation, Oxford, United Kingdom) with graphite-monochromated Mok $\alpha$  radiation ( $\lambda$  = 0.71073 Å). The Olex2 program was used as an interface, together with the SHELXT and SHELXL programs to solve the structure [12,13].

#### 3.2. Synthesis and Characterization of **1**

A mixture of 2,1,3-benzoselenadiazole (0.183 g, 1 mmol) and 4-nitrophenol (0.139 g, 1 mmol) in 30 mL methanol was refluxed for 2 h. The resulting solution was allowed to slowly evaporate under ambient conditions. After a week, the product was collected as colorless crystals. Yield: 0.181 g, 56%.

Crystal data for C11H10N4S (1): M = 322.18, monoclinic, P2<sub>1</sub>/c, a = 14.1004(10), b = 3.9390(2), c = 23.4020(15) Å, V = 1250.73(14) Å<sup>3</sup>, Z = 4,  $D_x$  = 1.711 g cm<sup>-3</sup>, F(000) = 640, and  $\mu$  = 3.009 mm<sup>-1</sup>. CCDC deposition number: 2250661.

The Supplementary Materials containing check CIF report, <sup>1</sup>H, <sup>13</sup>C NMR and XRD spectra of the title compound.

#### 4. Conclusions

Thus, a new supramolecular cocrystal of 2,1,3-benzoselenadiazole and 4-nitrophenol was synthesized and characterized by single-crystal X-ray diffraction. The  $[Se-N]_2$  homosynthon was also present in the cocrystal. The synthons were connected by O-H···N hydrogen bonds and N-Se···O chalcogen bonds into a one-dimensional chain, and the

two-dimensional network was formed by  $\pi \cdots \pi$  stacking interactions between chains. The result shows that the synthon approach is helpful in the construction of supramolecular cocrystals.

**Supplementary Materials:** The following supporting information can be downloaded. Crystallographic data for (1) in crystallographic information file (CIF) format. CCDC 2250661 also contains the supplementary crystallographic data for this paper.

**Author Contributions:** W.W. conceived and designed the experiments; H.L. performed the experiments; S.M. and W.W. analyzed the data; H.L. wrote the paper. All authors have read and agreed to the published version of the manuscript.

**Funding:** This research was funded by the Natural Science Foundation of Henan Province of China (Grant No. 232300421147 to W.W.).

**Data Availability Statement:** The authors confirm that the data supporting the findings of this study are available in the article and its Supplementary Materials.

**Acknowledgments:** We thank the X-ray crystallography laboratory at Luoyang Normal University for the X-ray intensity data.

Conflicts of Interest: The author declares no conflict of interest.

#### References

- 1. Lee, R.; Yufit, D.S.; Probert, M.R.; Steed, J.W. High Pressure/low temperature polymorphism in 2,6-dimethylpyridine-formic acid cocrystals. *Cryst. Growth Des.* **2017**, *17*, 1647–1653. [CrossRef]
- 2. Wang, H.; Jin, W.J. Cocrystal assembled by 1,4-diiodotetrafluorobenzene and phenothiazine based on C–I···π/N/S halogen bond and other assisting interactions. *Acta Cryst.* **2017**, *B73*, 210–216. [CrossRef]
- 3. Baykov, S.V.; Ivanov, D.M.; Kasatkina, S.O.; Galmés, B.; Frontera, A.; Resnati, G.; Kukushkin, Y. Stacking interactions: A supramolecular approach to upgrade weak halogen bond donors. *Chem. Eur. J.* **2022**, *28*, e202201869. [CrossRef] [PubMed]
- 4. Amombo Noa, F.M.; Bourne, S.A.; Su, H.; Nassimbeni, L.R. Secondary interactions in halogenated Werner clathrates. *Cryst. Growth Des.* **2017**, *17*, 1876–1883. [CrossRef]
- 5. Huynh, H.T.; Jeannin, O.; Fourmigué, M. Organic selenocyanates as strong and directional chalcogen bond donors for crystal engineering. *Chem. Commun.* 2017, *53*, 8467–8469. [CrossRef] [PubMed]
- 6. Wang, H.; Liu, J.; Wang, W.Z. Intermolecular and very strong intramolecular C–Se…O/N chalcogen bonds in nitrophenyl selenocyanate crystals. *Phys. Chem. Chem. Phys.* **2018**, *20*, 5227–5234. [CrossRef] [PubMed]
- 7. Steiner, T. The hydrogen bond in the solid state. Angew. Chem. Int. Ed. 2002, 41, 48–76. [CrossRef]
- 8. Kumar, S.; Body, C.; Leyssens, T.; Van Hecke, K.; Berger, G.; Van der Lee, A.; Laurencin, D.; Richeter, S.; Chément, S.; Meyer, F. Halogen-bonded thiophene derivatives prepared by solution and/or mechanochemical synthesis. Evidence of N…S chalcogen bonds in homo- and cocrystals. *Cryst. Growth Des.* **2023**, *23*, 2442–2454. [CrossRef]
- 9. Eichstaedt, K.; Wasilewska, A.; Wicher, B.; Gdaniec, M.; Połoński, T. Supramolecular synthesis based on a combination of Se…N secondary bonding interactions with hydrogen and halogen bonds. *Cryst. Growth Des.* **2016**, *16*, 1282–1293. [CrossRef]
- 10. Miao, S.B.; Zhang, Y.F.; Shan, L.J.; Xu, M.Y.; Wang, J.G.; Zhang, Y.; Wang, W.Z. A robust supramolecular heterosynthon assembled by a hydrogen bond and a chalcogen bond. *Crystals* **2021**, *11*, 1309. [CrossRef]
- 11. Lan, H.H.; Miao, S.B.; Zhang, Y.; Wang, W.Z. On the inverse correlation between the hydrogen bond strength and chalogen bond strength in the cyclic supramolecular hetersynthon [-Se-N=]...[HOOC-]. *CrystEngComm* **2023**, 25, 2159–2164. [CrossRef]
- 12. Sheldrick, G.M. Crystal structure refinement with SHELXL. Acta Crystallogr. 2015, C71, 3-8.
- Sheldrick, G.M. SHELXT-integrated space-group and crystal-structure determination. Acta Crystallogr. 2015, A71, 3–8. [CrossRef] [PubMed]

**Disclaimer/Publisher's Note:** The statements, opinions and data contained in all publications are solely those of the individual author(s) and contributor(s) and not of MDPI and/or the editor(s). MDPI and/or the editor(s) disclaim responsibility for any injury to people or property resulting from any ideas, methods, instructions or products referred to in the content.