



Short Note **2-(2,7-Bis(pyridin-3-ylethynyl)fluoren-9-ylidene)malononitrile**

Enrico Podda ^{1,2,*}, Massimiliano Arca ¹, Anna Pintus ¹, Francesca Meloni ¹, Vito Lippolis ¹, Giulio Ferino ², James B. Orton ³, Simon J. Coles ³ and Maria Carla Aragoni ^{1,*}

- ¹ Department of Chemical and Geological Sciences, University of Cagliari, Cittadella Universitaria, S.S. 554 Bivio Sestu, 09042 Monserrato, Italy; marca@unica.it (M.A.)
- ² Centre for Research University Services (CeSAR), University of Cagliari, Cittadella Universitaria, S.S. 554 Bivio Sestu, 09042 Monserrato, Italy
- ³ UK National Crystallography Service, School of Chemistry, Faculty of Engineering and Physical Sciences, University of Southampton, Southampton SO17 1BJ, UK
- * Correspondence: enrico.podda@unica.it (E.P.); aragoni@unica.it (M.C.A.)

Abstract: The 2-(2,7-bis(pyridin-3-ylethynyl)fluoren-9-ylidene)malononitrile (1) was synthesized by reaction of 2,7-bis(pyridin-3-ylethynyl)fluoren-9-one with malononitrile in DMSO solution. The structural characterization of 1 by SC-XRD analysis was accompanied by elemental analysis, FT-IR, NMR, and MS measurements.

Keywords: fluoren-9-ylidene malononitrile; polypyridyl donors; SC-XRD

1. Introduction

Fluorene and its derivatives are essential tools for the fabrication of functional materials, including organic photovoltaics [1] and electronic devices [2]. Purely organic [3] and hybrid metal–organic frameworks [4] with tunable absorption and emission properties have been prepared.

Fluoren-9-ylidene malononitrile derivatives are an interesting family of fluorene compounds as they are easily accessible from fluoren-9-one precursors. The electron-rich nature of fluorene is modified by the presence of the two strong electron-withdrawing cyano groups, which cause stabilization of the lowest unoccupied molecular orbital (LUMO) [2,5].

In order to design mixed Py/CN polytopic building blocks and to explore their potential use in supramolecular chemistry, additional moieties (such as pyridyl groups) can be appended to the fluoren-9-ylidene malononitrile core. In fact, pyridyl derivatives have proven to be extremely versatile in the field of supramolecular chemistry since they can be reacted with a large variety of complementary building blocks such as halogens [6,7], boron derivatives [8], and metal ions/complexes [9]. Herein, we report the synthesis and structural characterization of the novel pyridyl-functionalized fluoren-9-ylidene malononitrile derivative 2-(2,7-bis(pyridin-3-ylethynyl)fluoren-9-ylidene)malononitrile (1) (Scheme 1).



Scheme 1. Synthesis of 2-(2,7-bis(pyridin-3-ylethynyl)fluoren-9-ylidene)malononitrile (1).

2. Results

Compound 2,7-bis(pyridin-3-ylethynyl)fluoren-9-one was prepared by Sonogashira coupling reaction [10] and then reacted with malononitrile in DMSO (Scheme 1) to give compound 1.



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The multitopic ligand **1** was fully characterized by means of elemental analysis, ESI-MS, FT-IR, ¹H, and ¹³C NMR spectroscopy. The middle infrared spectrum shows the characteristic peaks for the C \equiv N and C \equiv C stretching modes at 2225 and 2198 cm⁻¹, respectively (Figure S1). The ¹H NMR spectrum in CDCl₃ (Figure S2) shows the H-pyridyl signals at 8.80, 8.59, 7.87, and 7.33 ppm, and those of the relevant protons of the fluoren-9-ylidene fragment at 8.55, 7.71, and 7.60 ppm. The ¹³C{¹H} spectrum in the same solvent (Figure S3) shows a signal at 159.9 ppm assigned to the C–9 of the fluoren-9-ylidene moiety and three signals resonating at 91.5, 88.6, and 77.9 ppm ascribed to the alkyne and nitrile functionalities, respectively. The MS spectrum in MeCN is consistent with the chemical structure of compound **1** (Figure S4). Recrystallization of the crude product from chloroform yielded well-defined purple needle-shaped crystals, which were structurally characterized by SC-XRD analysis. Compound **1** crystallizes in the monoclinic space group *P*2₁/*c* with one molecule in the asymmetric unit featuring a periplanar conformation where the nitrogen atoms of both pyridyl rings face the same side of the malononitrile substituent (Figure 1) [7].



Figure 1. Ellipsoid plot of compound **1** with the numbering scheme adopted. Displacement ellipsoids are drawn at 50 % probability level.

Crystal data for compound 1: $C_{30}H_{14}N_4$, (*Mr* = 430.45 g mol⁻¹) monoclinic, *P*2₁/*c* (No. 14), a = 7.0164(2) Å, b = 28.2624(6) Å, c = 11.2725(3) Å, β = 107.350(3) °, α = γ = 90 °, *V* = 2133.63(10) Å³, *T* = 100(2) K, *Z* = 4, *Z'* = 1, μ (Cu *K* α) = 0.637 mm⁻¹, 17,137 reflections measured, 3,904 unique (*R*_{int} = 0.0573) which were used in all calculations. The final *wR*₂ was 0.1540 (all data), and *R*₁ was 0.0538 (I ≥ 2 σ (I)).

Compound **1** adopts a quasi-planar geometry with pyridyl rings twisted by about 7° and 8° with respect to the fluorenylidene moiety, resulting in an overall slightly wavy shape. The central bis(fluoren-9-ylidene)malononitrile core is involved in two intramolecular CH…N hydrogen bonds between fluorenyl protons and nitrile functionalities identified as interactions a and b in Figure 2b and Table 1. A Full Interaction Map [11] (FIM) was calculated for compound 1 (based on IsoStar data from the CSD) to gain a better insight into the interaction preferences for the title compound (Figure 2a).

The FIM indicates four main hydrogen bond acceptor sites (highlighted in blue) close to the heteroatoms and a diffuse hydrogen bond donor region (highlighted in red) with hotspots close to the aryl protons of both the pyridyl and fluorenyl moieties. In agreement with the knowledge-based predicted scenario, the crystal packing is characterized by multiple intermolecular CH…N bonds between adjacent molecules of compound **1**, generating wavy sheets parallel to the [0 1 0] direction (Figure 3; see interactions c–d in Figure 2b and Table 1).



Figure 2. (a) Full interaction map calculated for compound **1** showing regions of interaction's likelihood for acceptors (red), donors (blue), and hydrophobic groups (orange). (b) Partial view of the crystal packing for compound **1** along the [1 0 0] direction showing intra- and intermolecular interactions labelled according to Table 1.

Table 1. Intra- and intermolecular interactions of compound 1.

Interaction		d _{D-Н} (Å)	$d_{\mathbf{H}\cdots\mathbf{A}}$ (Å)	d _{D…A} (Å)	$\alpha_{D-H\cdots A}$ (°)
а	C3-H3N3	0.95	2.62	3.440(2)	145.1
b	C12-H12N4	0.95	2.52	3.348(2)	145.9
с	C9 ⁱ -H9 ⁱ …N1	0.95	2.71	3.617(2)	159.2
d	C10 ⁱⁱ –H10 ⁱⁱ …N2	0.95	2.48	3.414(2)	167.2

 $\overline{i} = 1 + x, + y, 1 + z; ii = 1 - x, \frac{1}{2} + y, \frac{1}{2} - z.$



Figure 3. Packing diagrams of compound **1** viewed along [1 0 0] and [1 0 1] directions in (**a**,**b**), respectively, showing corrugated sheets. Molecules from adjacent sheets are depicted in different colors. H-bonds are represented as dashed lines.

Adjacent sheets interact through π - π stacking interactions (Figures 3 and S5), in good agreement with the FIM-calculated hydrophobic regions (highlighted in orange in Figure 2a).

3. Materials and Methods

3.1. General

Solvents and reagents were purchased from TCI (Dublin, Ireland), FluoroChem (Dublin, Ireland), and Aldrich (Schnelldorf, Germany). Compound 2,7-bis(pyridin-3ylethynyl)fluoren-9-one was prepared as previously described [10]. FT-IR spectroscopy measurements were recorded at room temperature on a Thermo-Nicolet 5700 (Burladingen, Germany) spectrometer using KBr pellets with a KBr beam splitter and KBr windows (4000–400 cm⁻¹, resolution 4 cm⁻¹). ¹H and ¹³C NMR spectra were carried out in CDCl₃ at room temperature using a Bruker Avance III HD 600 (Mannheim, Germany) spectrometer. Chemical shifts are reported in ppm (δ) and were calibrated to the solvent residue. Coupling constants J are expressed in Hertz (Hz). Positive ESI-MS spectra were recorded on a highresolution LTQ Orbitrap Elite™ mass spectrometer (Thermo Fisher Scientific, Waltham, MA, USA). Solutions were infused into the ESI source at a flow rate of 5.00 μ L/min. Spectra were recorded in the range of m/z 300–600 with a resolution of 240,000 (FWHM). Instrument conditions were as follows. Spray voltage 3,500 V, capillary temperature 275 °C, sheath gas 12 (arbitrary units), auxiliary gas 3 (arbitrary units), sweep gas 0 (arbitrary units), and probe heater temperature 50 °C. Elemental analysis was carried out with a CHNS/O PE 2400 series II elemental analyzer (Thermo Fisher, Dreieich, Germany; T = 925 $^{\circ}$ C). The melting point was determined on a FALC mod. C apparatus. X-ray diffraction data for compound 1 were collected at 100(2) K on a Rigaku FRE+ diffractometer equipped with

VHF Varimax confocal mirrors, an AFC12 goniometer, and an HyPix 6000 detector. The structure was solved with the ShelXT [12] solution program using dual methods, and the model was refined with ShelXL 2018/3 [13] using full matrix least squares minimization on F^2 . Olex2 1.5 [14] was used as the graphical interface.

3.2. Synthesis of 2-(2,7-Bis(pyridin-3-ylethynyl)fluoren-9-ylidene)malononitrile (1)

Compound 1 was synthesized by a slight modification of a previously reported synthetic procedure [15]. 2,7-Bis(pyridin-3-ylethynyl)fluoren-9-one (38.2 mg; 0.100 mmol) and malononitrile (7.2 mg; 0.11 mmol) were reacted in DMSO (0.5 mL) at 110 °C for 5 h. The mixture was cooled to room temperature, and the dark solid was filtered on a Gooch funnel. The solid was then washed thoroughly with acetonitrile and dried under vacuum. Purple needle-shaped crystals suitable for SC-XRD analysis were grown by slow evaporation of a chloroform solution of the product (28.1 mg; 0.065 mmol; Y = 65 %) M. p. = 301 °C. ESI(+)-MS (MeCN solution) m/z 431.1322 for $[C_{30}H_{15}N_4]^+$ [M+H]⁺. Elemental analysis calcd (%) for C₃₀H₁₄N₄: C 83.71, H 3.28, N 13.02. Found: C 83.27, H 2.79, N 13.88. FT-IR (KBr, 4000–400 cm⁻¹): 3055 w, 3028 w, 2987 w, 2225 m ν (C \equiv N), 2198 w ν (C \equiv C), 1963 w, 1930 w, 1894 w, 1861 w, 1604 w, 1562 s, 1481 s, 1465 ms, 1427 m, 1408 ms, 1385 w, 1321 w, 1261 w, 1238 w, 1222 w, 1190 mw, 1176 mw, 1118 m, 1095 mw, 1024 ms, 987 w, 949 w, 922 w, 899 ms, 872 ms, 845 ms, 802 s, 756 ms, 700 s, 638 w, 615 w, 597 m, 551 w, 530 w, 513 w, 486 w, 438 w, 405 mw cm⁻¹. ¹H NMR (600 MHz, CDCl₃) δ: 8.80 (s, 2H, Py), 8.59 (s, 2H, Py), 8.55 (s, 2H, fl), 7.87 (d, J = 6.6 Hz, 2H, Py), 7.71 (d, J = 7.0 Hz, 2H, fl), 7.60 (d, J = 7.0 Hz, 2H, fl), 7.33 (s, 2H, Py) ppm. ¹³C{¹H} NMR (151 MHz, CDCl₃) δ: 159.9, 152.3, 149.1, 141.6, 139.0, 138.1, 134.7, 129.9, 124.3, 123.3, 121.2, 120.0, 113.0, 91.5, 88.7, 77.9 ppm.

4. Conclusions

The compound 2-(2,7-bis(pyridin-3-ylethynyl)fluoren-9-ylidene)malononitrile (1) has been synthesized and fully characterized. Further studies are ongoing in our laboratories to evaluate the potential use of compound 1 as a building block for the formation of supramolecular assemblies.

Supplementary Materials: The following supporting information is available online; Figure S1: FT-IR; Figures S2 and S3: ¹ H and ¹³ C{¹ H} NMR; Figure S4: ESI(+) MS; Figure S5: $\pi - \pi$ stacking interactions; Table S1: Crystal data and refinement parameters; Tables S2 and S3: Bond lengths and angles.

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