

Short Note

1-Phenyl-3,3-di(1*H*-pyrazol-1-yl)propan-1-one

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Abstract: The title compound, 1-phenyl-3,3-di(1*H*-pyrazol-1-yl)propan-1-one, was synthesized in a 52% yield for the first time by a one-step reaction between 1*H*-pyrazole and 1-phenylprop-2-yn-1-one (their ratio being 2:1) in solid Al₂O₃ at room temperature. The product was characterized by ¹H-NMR, ¹³C-NMR, IR spectroscopy, X-ray diffraction and elemental analysis.

Keywords: pyrazole; 1-phenylprop-2-yn-1-one; di(1*H*-pyrazol-1-yl)propan-1-one; Al₂O₃

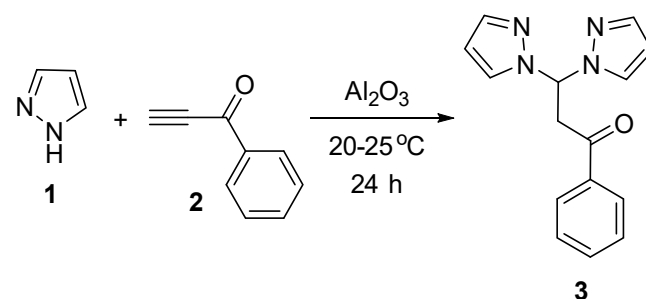
1. Introduction

Dipyrazole with a gem-disposition of pyrazole rings exhibit a wide spectrum of biological activity [1,2]. Furthermore, dipyrazolymethanes are promising ligands [3,4]. Complex compounds with such dipyrazolyl ligands are active against the trypanosome pathogen [5], while scorpionate copper complexes of dipyrazolymethanecarboxylic acids possess anticancer activity [6]. Complexes based on dipyrazolyl chelate ligands are active catalysts in polymerization processes [7].

Among many methods for the synthesis of compounds containing gem-dipyrazole moieties [8], a special place is occupied by those using acetylene-derived molecules. These are the addition of pyrazoles to electron-deficient alkynes in the presence of Lewis acids [9,10] and a silver(I)-catalyzed reaction between pyrazole and propargyl acetates [11]. However, these reactions take place under rather harsh conditions (reflux during 24 h), and they require expensive catalyst salts (Sc(OTf)₃, Au(OTf)₃, AgOTf, AgPF₄ and AgNO₃).

2. Results and Discussion

We have shown that 1*H*-pyrazole (1) readily reacted with available 1-phenylprop-2-yn-1-one (2) (the ratio being 2:3, 2:1) in solid Al₂O₃ at room temperature to give a ketone with two pyrazole rings, 1-phenyl-3,3-di(1*H*-pyrazol-1-yl)propan-1-one (3), in a 52% yield (Scheme 1).



Scheme 1. Synthesis of 1-phenyl-3,3-di(1*H*-pyrazol-1-yl)propan-1-one (3).

The structure and composition of the synthesized dipyrazole were confirmed by ¹H, ¹³C NMR, IR spectroscopy, X-Ray diffraction (Figure 1) and elemental analysis (see Supplementary Materials). Elemental analysis establishes the chemical formula of compound 3. The ¹H NMR spectrum of 3 shows signals of the pyrazole rings at 7.65, 7.53 and 6.27 ppm,



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the signal of the proton of sp^2 -carbon at 7.17 ppm ($CHCH_2$) and the signal of the proton of sp^3 -carbon at 4.40 ppm ($CHCH_2$). The characteristic signal of the carbonyl group is observed in the ^{13}C NMR spectrum at 194.7 ppm. The IR spectrum reveals characteristic bands of the $C=O$ bond (1663 cm^{-1}) and $C=N$, $C=C$ bonds ($1570\text{--}1615\text{ cm}^{-1}$).

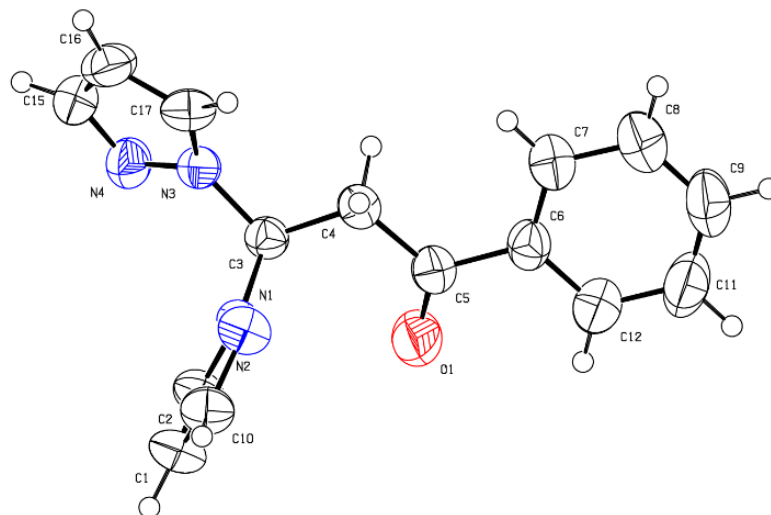


Figure 1. X-ray structure of 1-phenyl-3,3-di(1*H*-pyrazol-1-yl)propan-1-one (**3**). Thermal ellipsoids set at 50% probability.

Thus, we have synthesized a dipyrazole functionalized with a ketone group, 1-phenyl-3,3-di(1*H*-pyrazol-1-yl)propan-1-one, which contains two pharmacologically valuable counterparts (pyrazole and benzoyl ethyl moieties) and can be considered as a prospective tripodal scorpionate-like ligand for coordination chemistry and metal complex catalysis.

3. Materials and Methods

General. NMR spectra were recorded on a Bruker DPX-400 spectrometer (Bruker, Billerica, MA, USA) (400.1 MHz for 1H and 100.6 MHz for ^{13}C) in $CDCl_3$. The internal standards were HMDS (for 1H) and the residual solvent signals (for ^{13}C). Coupling constants (J) were measured from one-dimensional spectra, and multiplicities were abbreviated as follows: s (singlet), d (doublet), dd (doublet of doublets), q (quartet), t (triplet) and m (multiplet). The determination of the unit cell and the data collection for 1-phenyl-3,3-di(1*H*-pyrazol-1-yl)propan-1-one (**3**) were performed on a Bruker D8 VENTURE PHOTON 100 CMOS diffractometer with MoK_{α} radiation ($\lambda = 0.71073$) at 293.0(2) K using the ω - ϕ scan technique. A specimen of $C_{15}H_{14}N_4O$, with approximate dimensions of 0.13 mm \times 0.14 mm \times 0.26 mm, was used for the X-ray crystallographic analysis. The X-ray intensity data were measured. The integration of the data using a monoclinic unit cell with a $P2_1/c$ space group yielded a total of 27,301 reflections to a maximum θ angle of 26.1° (0.81 Å resolution), of which 2675 were independent (completeness = 100.0%, $R_{int} = 6.54\%$, $R_{sig} = 2.68\%$) and 1826 were greater than $2\sigma(F_2)$. The final cell constants of $a = 14.6160(7)$ Å, $b = 9.0561(4)$ Å, $c = 10.4156(4)$ Å, $Z = 4$, volume = 1356.55(10) Å³. Data were corrected for absorption effects using the multi-scan method (SADABS). The calculated minimum and maximum transmission coefficients (based on the crystal size) were 0.9671 and 0.9837. The structure was solved using the Bruker SHELXTL Software Package [12] and refined using the Olex2 [13] package. All H atoms were treated as riding atoms. IR spectra were recorded on a two-beam Bruker Vertex 70 spectrometer (Bruker, Billerica, MA, USA), in a KBr pellet. Elemental analyses (C, H, N) were performed on an EA FLASH 1112 Series (CHN Analyzer) instrument (Thermo Finnigan, Italy). Melting points (uncorrected) were measured on a Kofler micro hot-stage apparatus.

1*H*-Pyrazole (**1**) and Al_2O_3 were commercial ones. 1-Phenylprop-2-yn-1-one (**2**) was obtained according to the method from [14].

Synthesis of 1-phenyl-3,3-di(1H-pyrazol-1-yl)propan-1-one (3). 1H-pyrazole (**1**) (408 mg, 6 mmol) and 1-phenylprop-2-yn-1-one (**2**) (390 mg, 3 mmol) were carefully ground together with alumina (8 g, ~10-fold amount by weight) for 5 min and allowed to stay at rt for 24 h. Then, the solid reaction mixture was placed on top of an Al₂O₃-packed column and successively eluted with *n*-hexane and systems of *n*-hexane with diethyl ether (*n*-hexane/diethyl ether with a gradient from 3:1, 1:1, 1:3) to afford 1-phenyl-3,3-di(1H-pyrazol-1-yl)propan-1-one (**3**). Yield 415 mg (52%), yellow crystals (suitable for X-ray analysis), mp 70 °C. IR spectrum (KBr), ν , cm⁻¹: 3313, 3114, 2922, 1663 (C=O), 1615, 1570 (C=N, C=C). ¹H NMR (CDCl₃, ppm): δ 8.00–7.98 (m, 2H, *o*-Ph), 7.65–7.64 (m, 2H, H-4, pyrazole), 7.60–7.57 (m, 1H, *p*-Ph), 7.54–7.53 (m, 2H, H-5, pyrazole), 7.48–7.44 (m, 2H, *m*-Ph), 7.17 (t, *J* = 6.6 Hz, 1H, CH–CH₂), 6.28–6.26 (m, 2H, H-3, pyrazole), 4.39 (d, *J* = 6.6 Hz, 2H, CH–CH₂). ¹³C NMR (CDCl₃, ppm): 194.7 (C=O), 140.3 (2C), 135.9, 133.7, 129.1 (2C), 128.7 (2C), 128.2 (2C), 106.6 (2C), 71.3, 42.2. Anal. calcd. for C₁₅H₁₄N₄O (%): C, 67.65; H, 5.30; N, 21.04. Found (%): C, 67.79; H, 5.43; N, 21.19.

Supplementary Materials: The followings can be downloaded online. Copies of ¹H NMR, ¹³C NMR and X-ray diffraction analysis.

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