

## Supporting Information

### **2,3,5-*Tri-O-Benzyl-D*-Xylofuranose**

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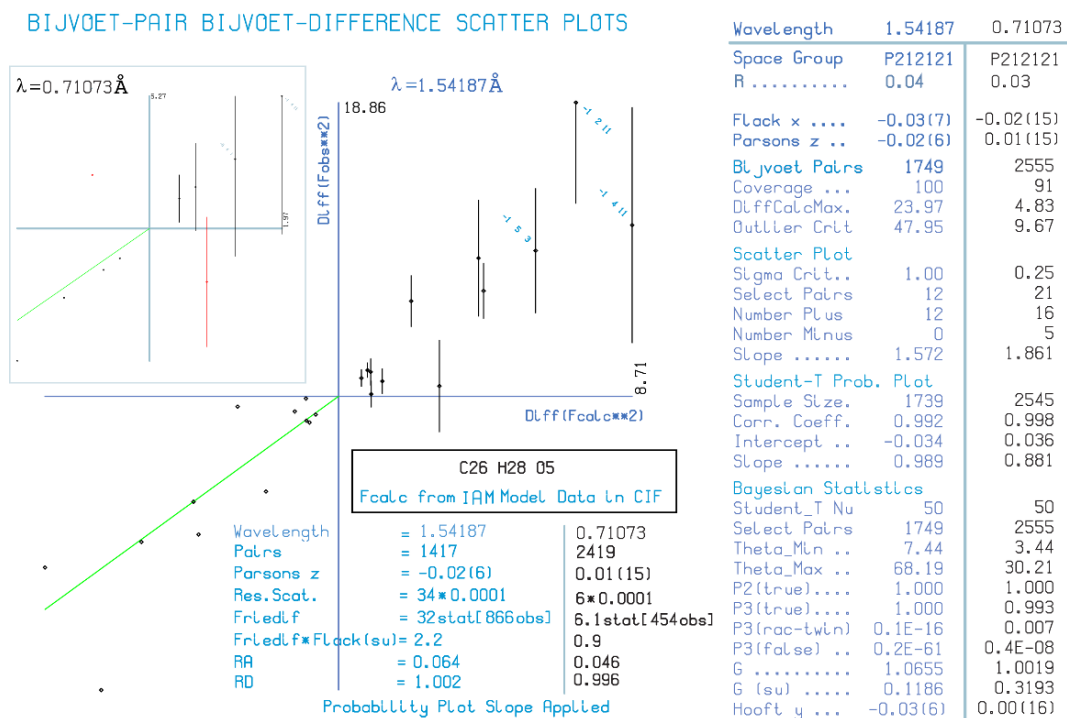
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# 1. Crystal Data, Data Collection and IAM Structure Refinement Details of Compound 1 ( $\alpha/\beta$ , *ca* 89:11) Collected at Room Temperature.

**Table S1.** Crystal data, data collection and IAM structure refinement details of compound 1 collected at room temperature using copper radiation (statistics for Mo are recalled in comparison, right column)

Identification code	2,3,5-Tri-O-benzyl- $\alpha,\beta$ -D-xylofuranose ( $\alpha,\beta$ )-1 ( $\alpha/\beta$ <i>ca.</i> 89:11)	
Empirical formula, (weight)	$C_{26}H_{28}O_5$ , (420.48)	<i>Id.</i>
Temperature (K)	293(2)	100(2)
Diffractometer Rigaku®	mm007+Spider2	<i>XtaLabPro</i> <i>mm003+Pilatus200k</i>
Wavelength (Å)	1.54187	0.71073
Crystal system, space group	$P2_12_12_1$ , Orthorhombic	<i>Id.</i>
Unit cell dimensions (Å, $\alpha=\beta=\gamma=90^\circ$ )	$a = 6.9071(1)$ $b = 13.4121(3)$ $c = 24.3907(17)$	$a = 6.9002(1)$ $b = 13.1783(2)$ $c = 23.9751(5)$
Volume (Å <sup>3</sup> )	2259.52(17)	2180.13(6)
Z, Calculated density (Mg/m <sup>3</sup> )	4, 1.236	4, 1.281
Absorption coefficient (mm <sup>-1</sup> )	0.686	0.088
$F(000)$	896	<i>Id.</i>
Theta range for data collection (°)	3.624 to 68.193	2.98 to 30.25
Limiting indices	$-8 \leq h \leq 8$ , $-16 \leq k \leq 16$ , $-29 \leq l \leq 26$	$-9 \leq h \leq 9$ , $-17 \leq k \leq 18$ , $-33 \leq l \leq 33$
Reflections collected / unique $R(int)$	29687 / 4145 0.0448	55919 / 6102 0.0370
Completeness to $\theta_{max}$ ( $iUCR$ ) (%)	100	99.8
Absorption correction method	Multi-scan 1.000 and 0.808	<i>Gauss. &amp; M-scan</i> <i>1.000 and 0.302</i>
Refinement method	IAM ( <i>Shelxl software</i> )	<i>Id.</i>
Data / restraints / parameters	4145 / 426 / 421	6102 / 416 / 419
Goodness-of-fit on $F^2$	1.109	1.028
Final R indices [ $I > 2\sigma(I)$ ]	$R1$ , $wR2$	0.0372, 0.0974
Final R indices (all data)	$R1$ , $wR2$	0.0338, 0.0896
Absolute structure parameter <sup>[S1]</sup>	0.0410, 0.1032	0.0352, 0.0903
Largest diff. peak and hole (e.Å <sup>-3</sup> )	-0.03(7)	-0.02(15)
	0.111 and -0.158	0.273 and -0.192

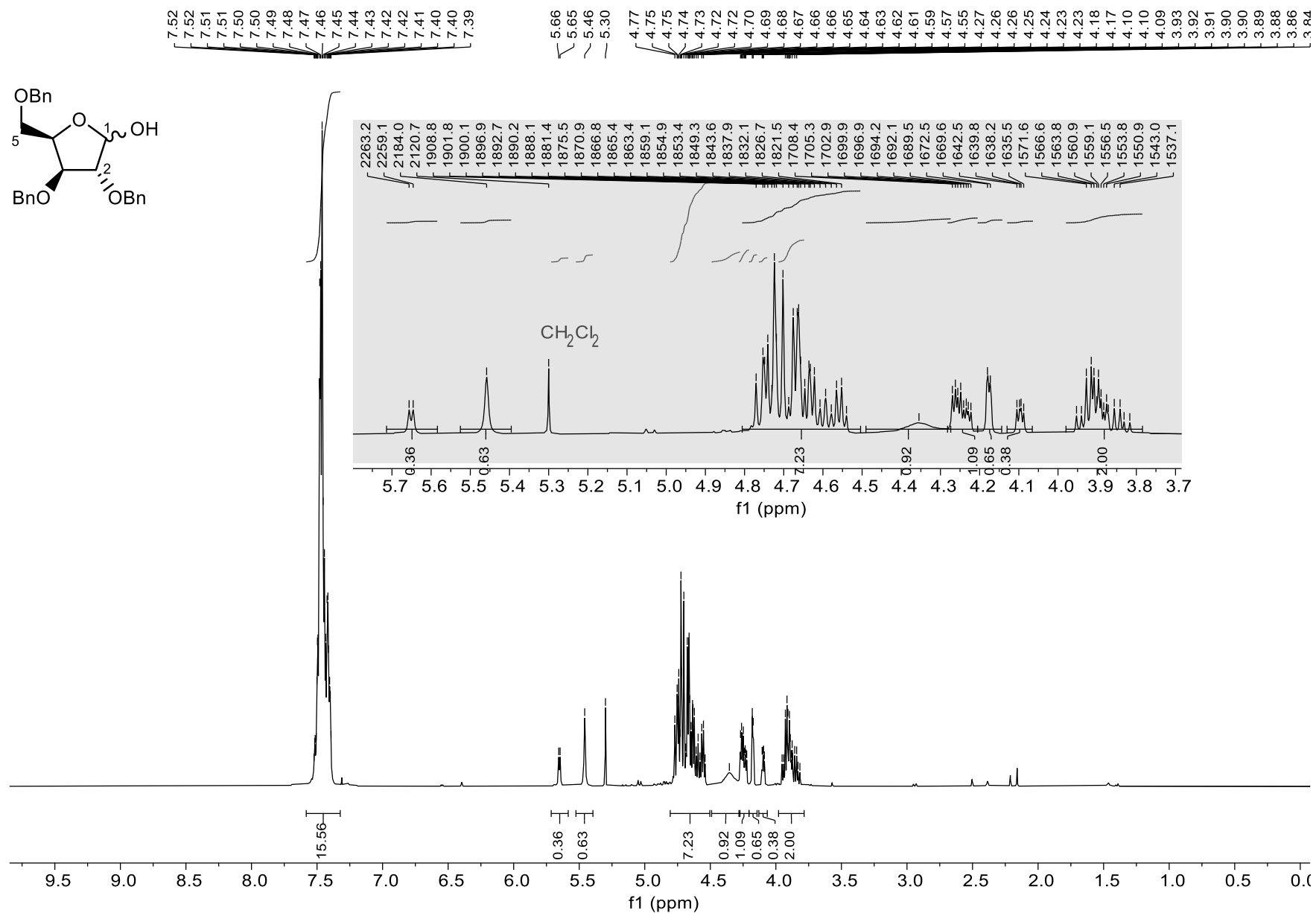
## 2. Bijvoet Analyses.

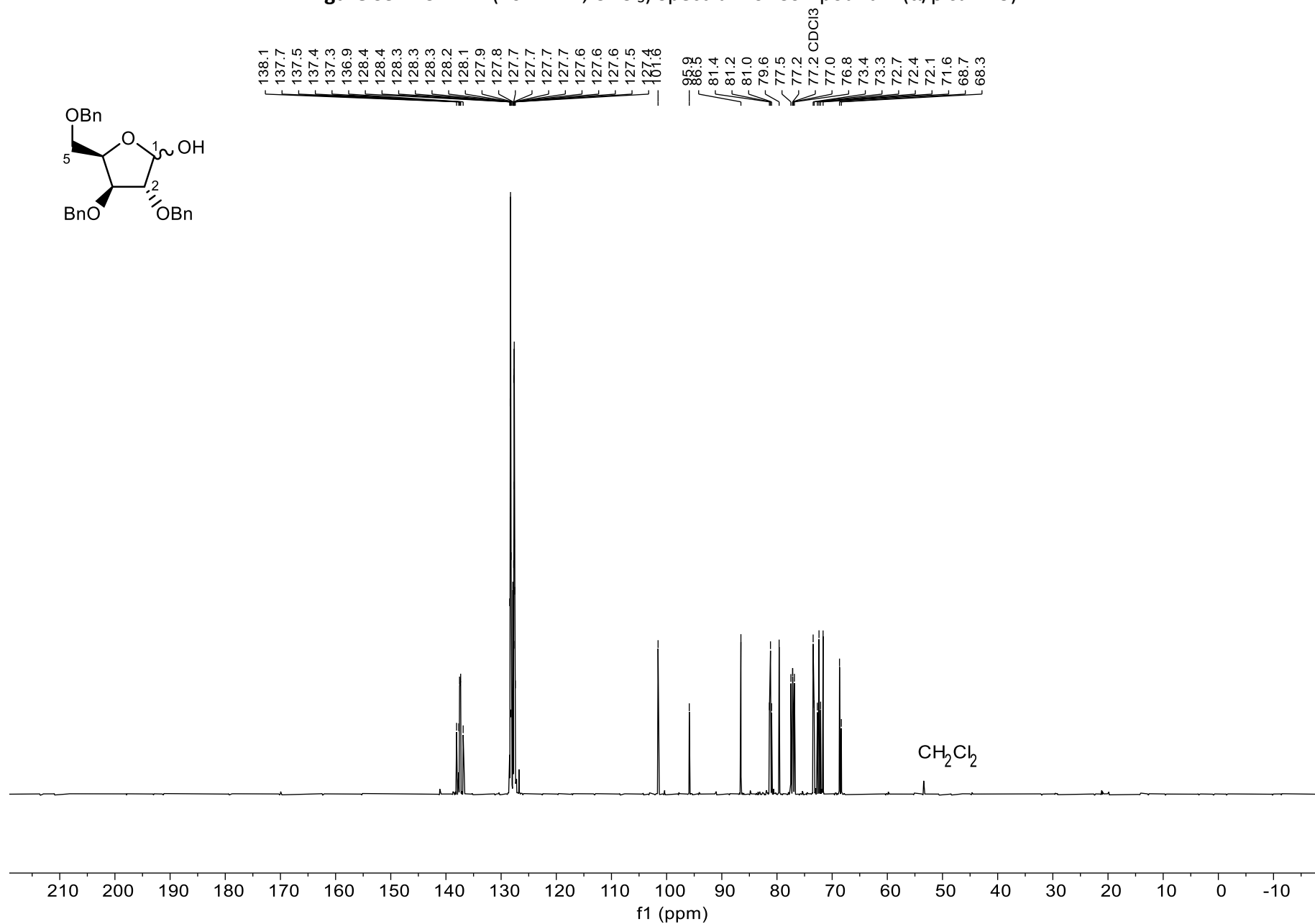


**Figure S1.** Summary of Bijvoet analyses performed with the program *Platon* <sup>[S2]</sup> (using IAM models derived with two different radiation energies, see Table S1), assessing the absolute configuration of **1** by anomalous dispersion.

### 3. Copies of $^1\text{H}$ and $^{13}\text{C}$ NMR Spectra of Compound 1.

Figure S2.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ) Spectrum of Compound 1 ( $\alpha/\beta$  ca. 2:3).



**Figure S3.**  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ ) Spectrum of Compound **1** ( $\alpha/\beta$  *ca.* 2:3).

#### 4. References.

- S1. Parsons, S.; Flack, H.D.; Wagner, T. Use of intensity quotients and differences in absolute structure refinement. *Acta Cryst.* **2013**, *B69*, 249–259. doi:10.1107/S2052519213010014
- S2. Spek, A. L. Structure validation in chemical crystallography. *Acta Cryst.* **2009**, *D65*, 148–155. doi:10.1107/S090744490804362X