

Supporting Information

2,3,5-*Tri-O-Benzyl-D-Xylofuranose*

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1. Crystal Data, Data Collection and IAM Structure Refinement Details of Compound 1 (α/β , ca 89:11) Collected at Room Temperature.

Table S1. Crystal data, data collection and IAM structure refinement details of compound 1 collected at room temperature using copper radiation (statistics for Mo are recalled in comparison, right column)

Identification code	2,3,5-Tri-O-benzyl- α,β -D-xylofuranose (α,β)-1 (α/β ca. 89:11)	
Empirical formula, (weight)	$C_{26}H_{28}O_5$, (420.48)	<i>Id.</i>
Temperature (K)	293(2)	100(2)
Diffractometer Rigaku®	mm007+Spider2	XtaLabPro mm003+Pilatus200k
Wavelength (Å)	1.54187	0.71073
Crystal system, space group	$P2_12_12_1$, Orthorhombic	<i>Id.</i>
Unit cell dimensions (Å, $\alpha=\beta=\gamma=90^\circ$)	a = 6.9071(1) b = 13.4121(3) c = 24.3907(17)	a = 6.9002(1) b = 13.1783(2) c = 23.9751(5)
Volume (Å ³)	2259.52(17)	2180.13(6)
Z, Calculated density (Mg/m ³)	4, 1.236	4, 1.281
Absorption coefficient (mm ⁻¹)	0.686	0.088
F(000)	896	<i>Id.</i>
Theta range for data collection (°)	3.624 to 68.193	2.98 to 30.25
Limiting indices	-8 ≤ h ≤ 8, -16 ≤ k ≤ 16, -29 ≤ l ≤ 26	-9 ≤ h ≤ 9, -17 ≤ k ≤ 18, -33 ≤ l ≤ 33
Reflections collected / unique R(int)	29687 / 4145 0.0448	55919 / 6102 0.0370
Completeness to θ_{max} ($iUCR$) (%)	100	99.8
Absorption correction method	Multi-scan 1.000 and 0.808	Gauss. & M-scan 1.000 and 0.302
Refinement method	IAM (<i>Shelxl software</i>)	<i>Id.</i>
Data / restraints / parameters	4145 / 426 / 421	6102 / 416 / 419
Goodness-of-fit on F^2	1.109	1.028
Final R indices [$I > 2\sigma(I)$]	R1, wR2	0.0372, 0.0974
Final R indices (all data)	R1, wR2	0.0410, 0.1032
Absolute structure parameter ^[S1]	-0.03(7)	-0.02(15)
Largest diff. peak and hole (e.Å ⁻³)	0.111 and -0.158	0.273 and -0.192

2. Bijvoet Analyses.

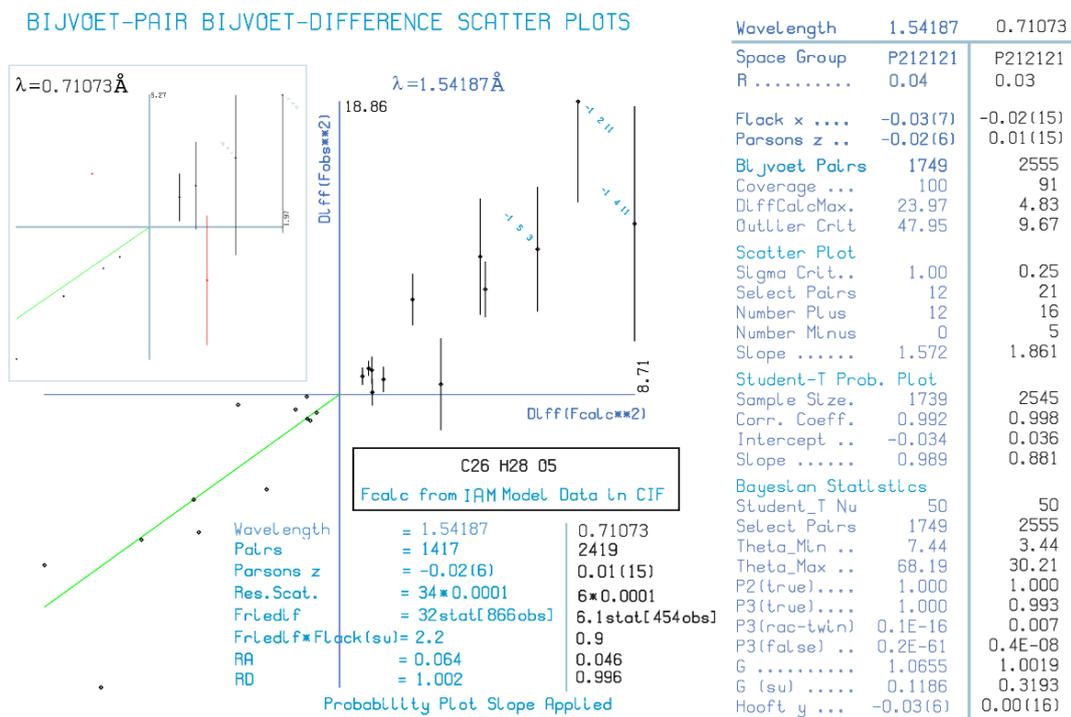


Figure S1. Summary of Bijvoet analyses performed with the program *Platon* ^[S2] (using IAM models derived with two different radiation energies, see Table S1), assessing the absolute configuration of **1** by anomalous dispersion.

3. Copies of ^1H and ^{13}C NMR Spectra of Compound 1.

Figure S2. ^1H NMR (400 MHz, CDCl_3) Spectrum of Compound 1 (α/β ca. 2:3).

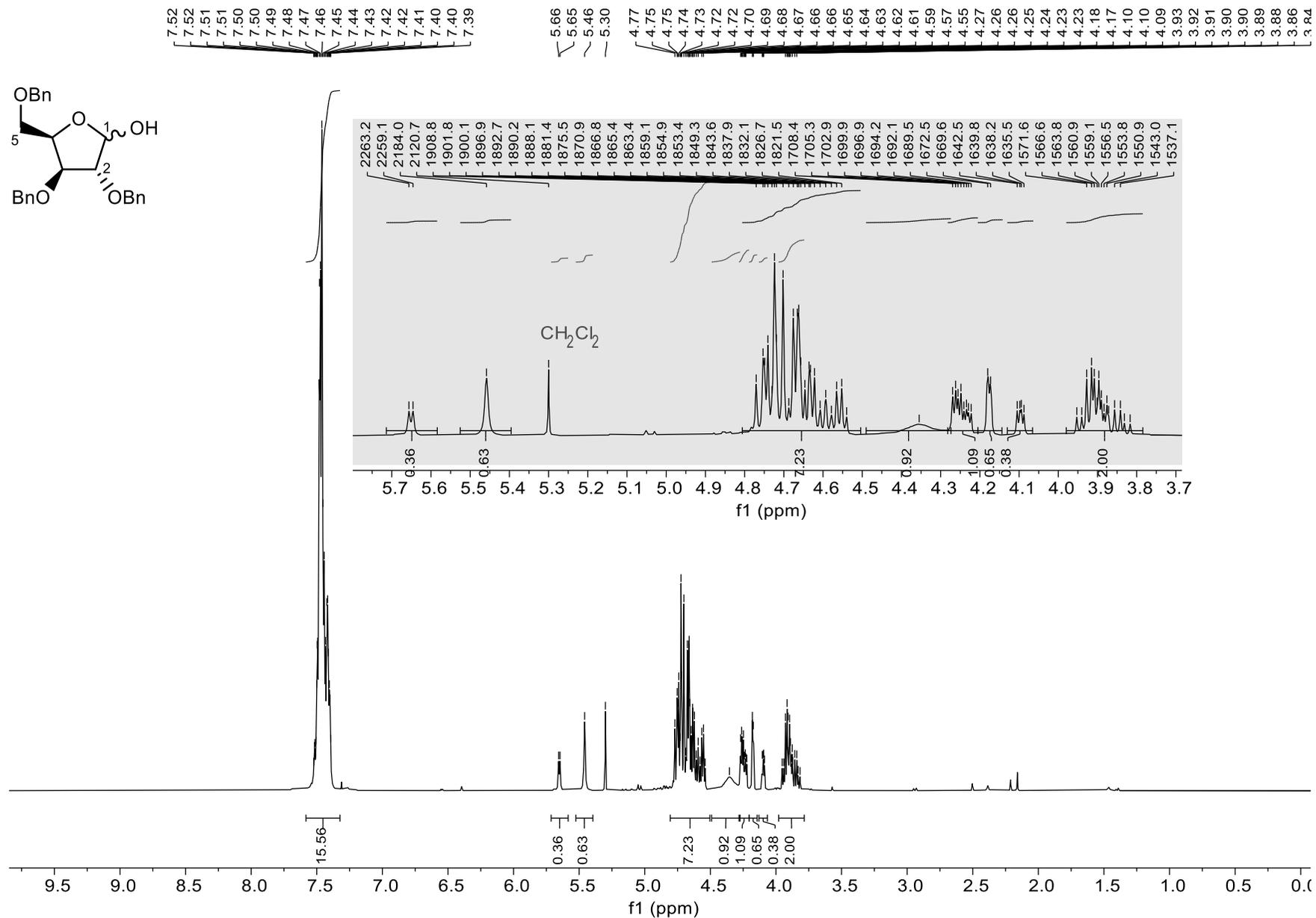
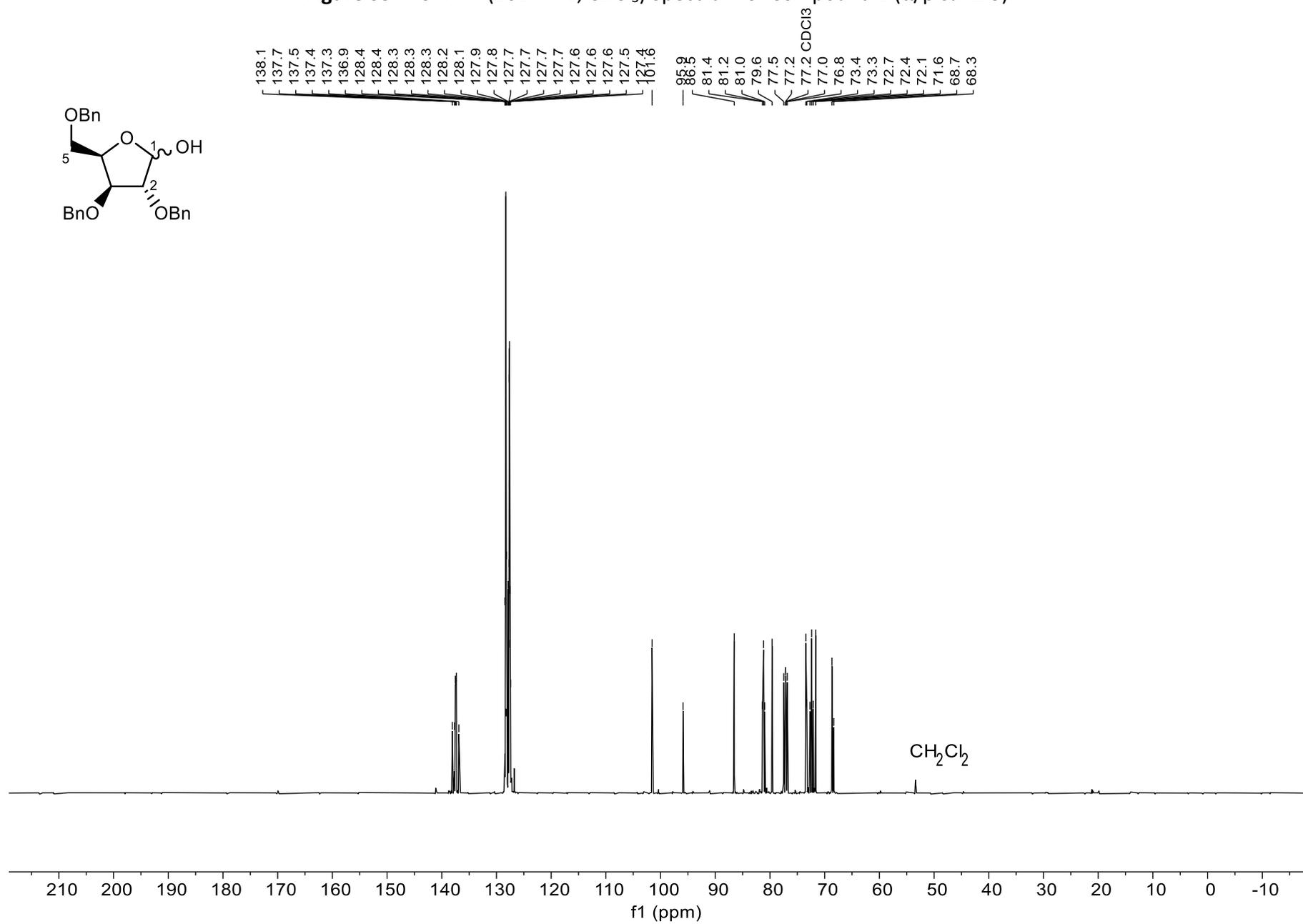


Figure S3. ^{13}C NMR (101 MHz, CDCl_3) Spectrum of Compound **1** (α/β *ca.* 2:3).

4. References.

- S1. Parsons, S.; Flack, H.D.; Wagner, T. Use of intensity quotients and differences in absolute structure refinement. *Acta Cryst.* **2013**, *B69*, 249–259. doi:10.1107/S2052519213010014
- S2. Spek, A. L. Structure validation in chemical crystallography. *Acta Cryst.* **2009**, *D65*, 148–155. doi:10.1107/S090744490804362X