

## Supplementary materials

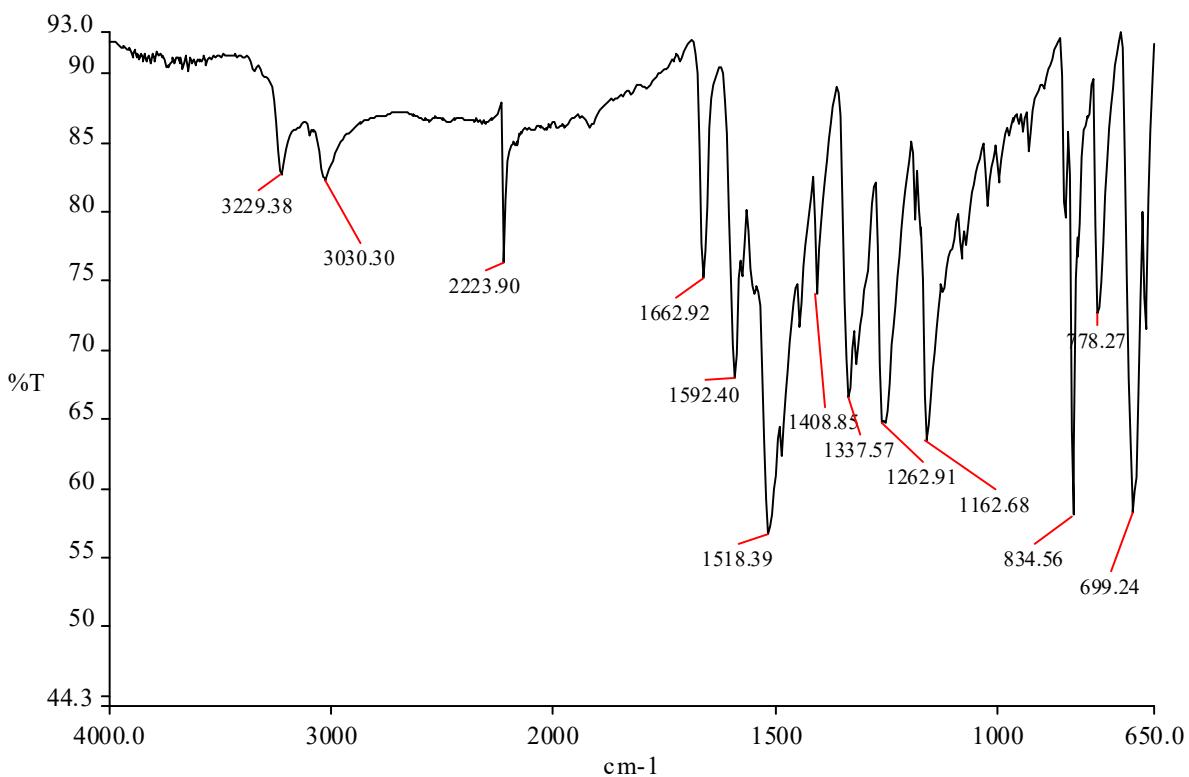
### Synthesis, crystal structure and cyclic voltammetric behavior of N-Aroyl-N'-(4'-cyanophenyl)thioureas

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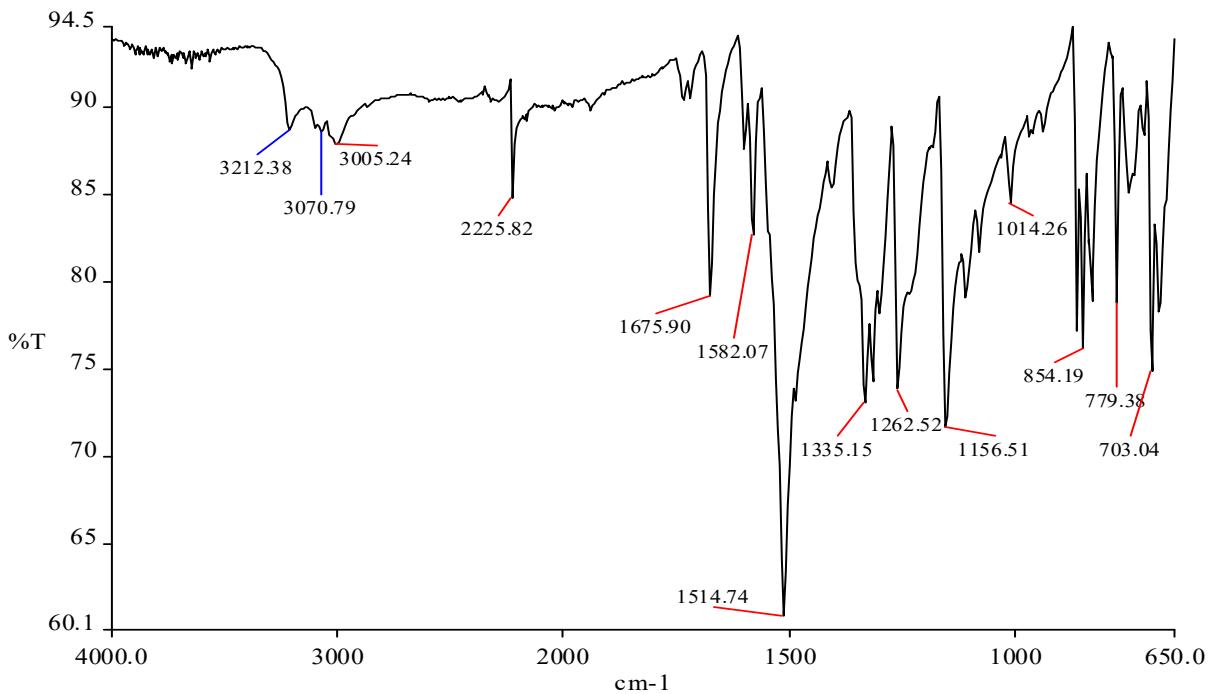
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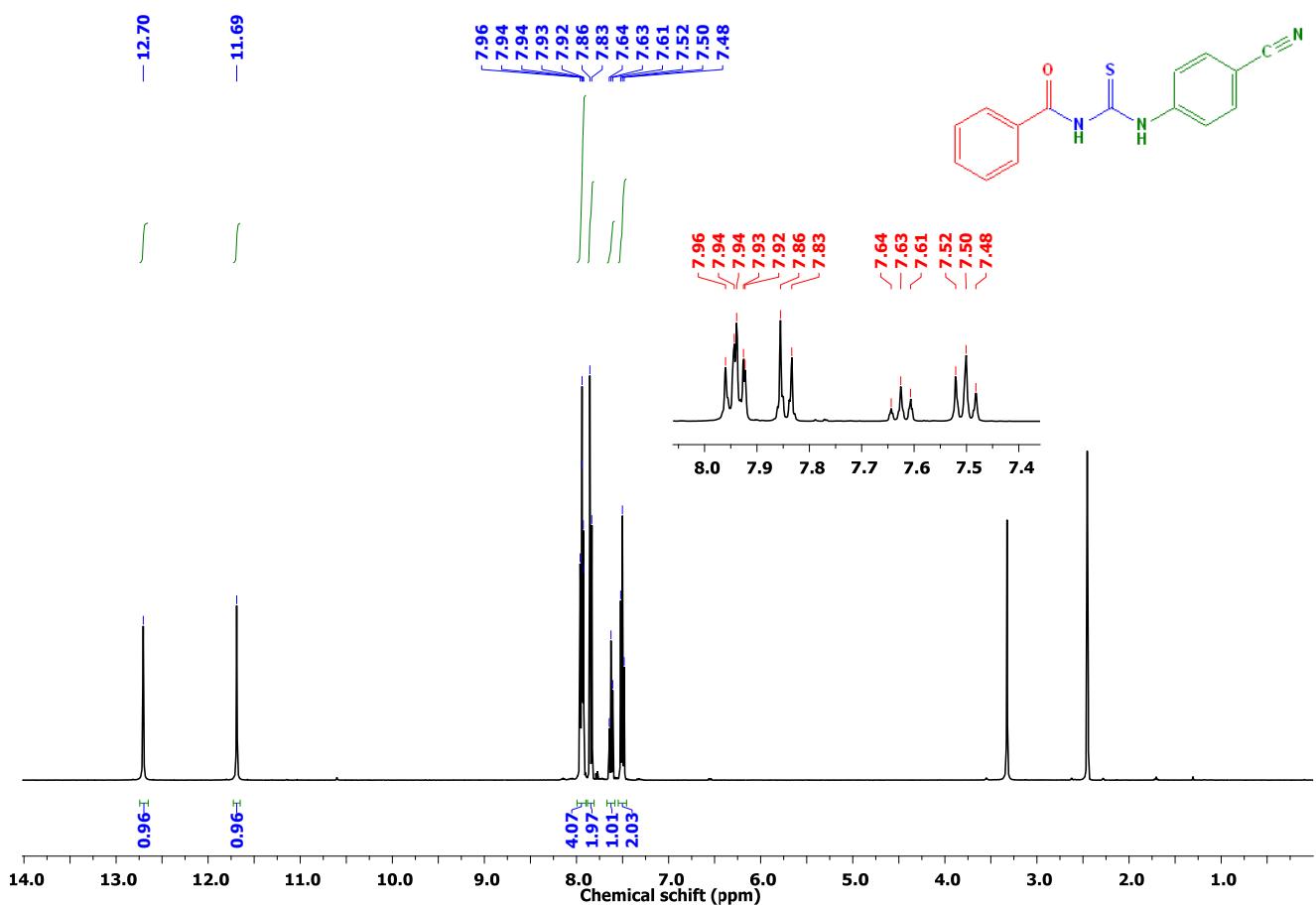
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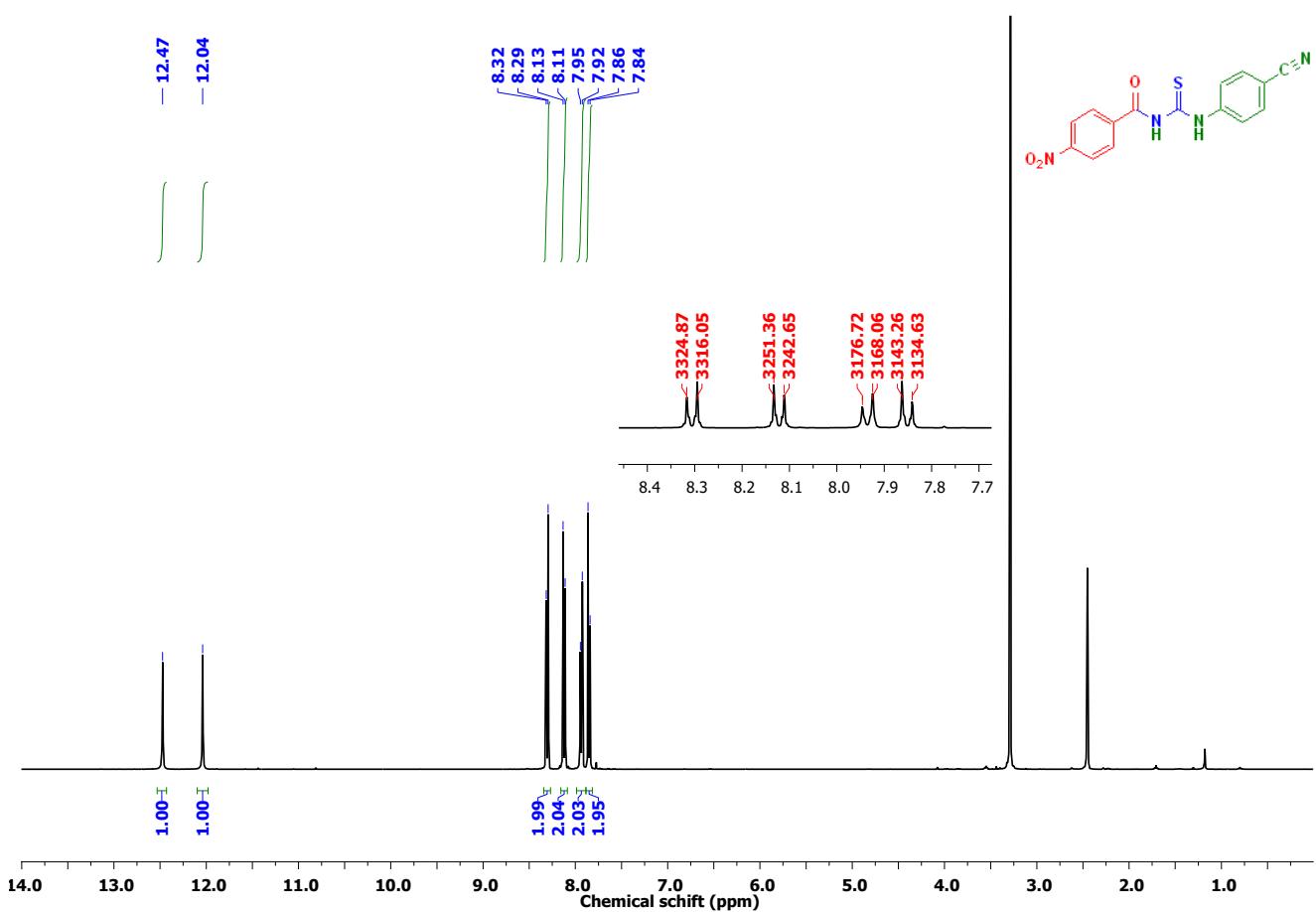
**Figure S1.** The FT-IR spectrum of N-benzoyl-N'-(4-cyanophenyl)-thiourea (**1**)



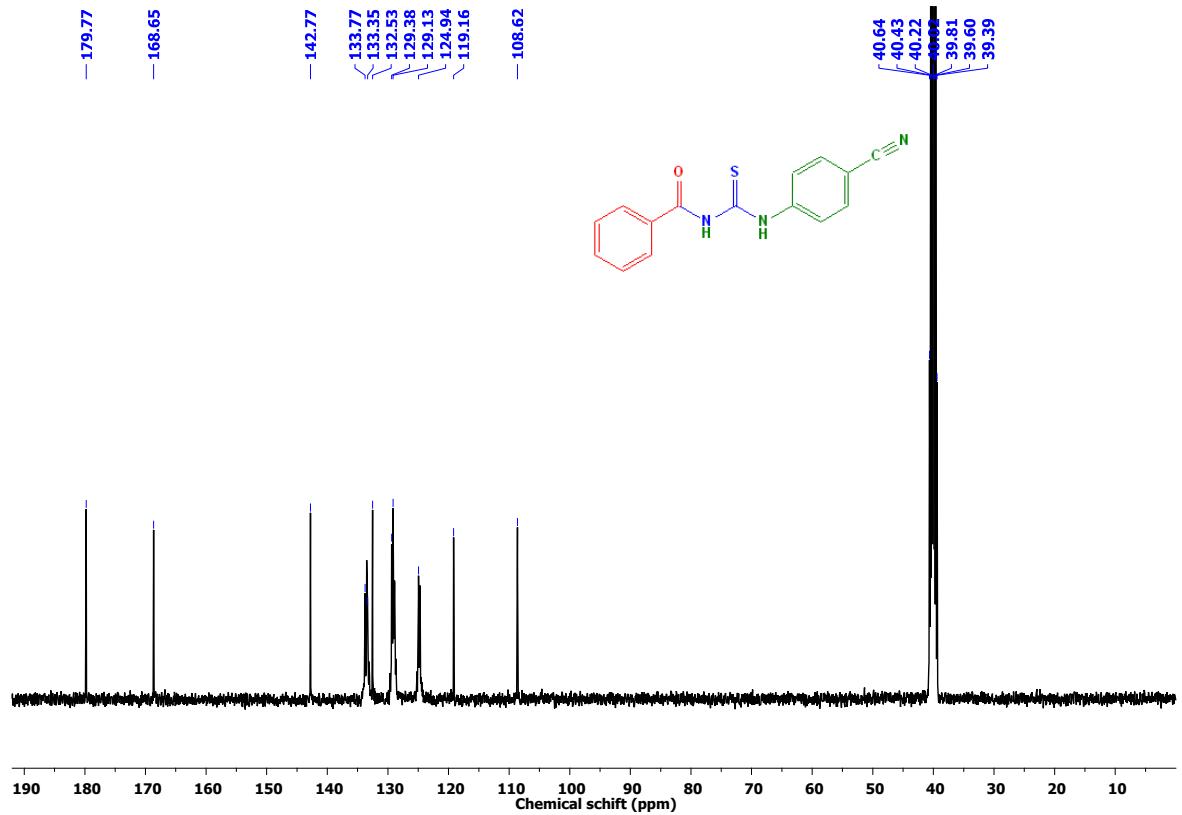
**Figure S2.** The FT-IR spectrum of N-(4-nitrobenzoyl)-N'-(4-cyanophenyl)-thiourea (**2**)



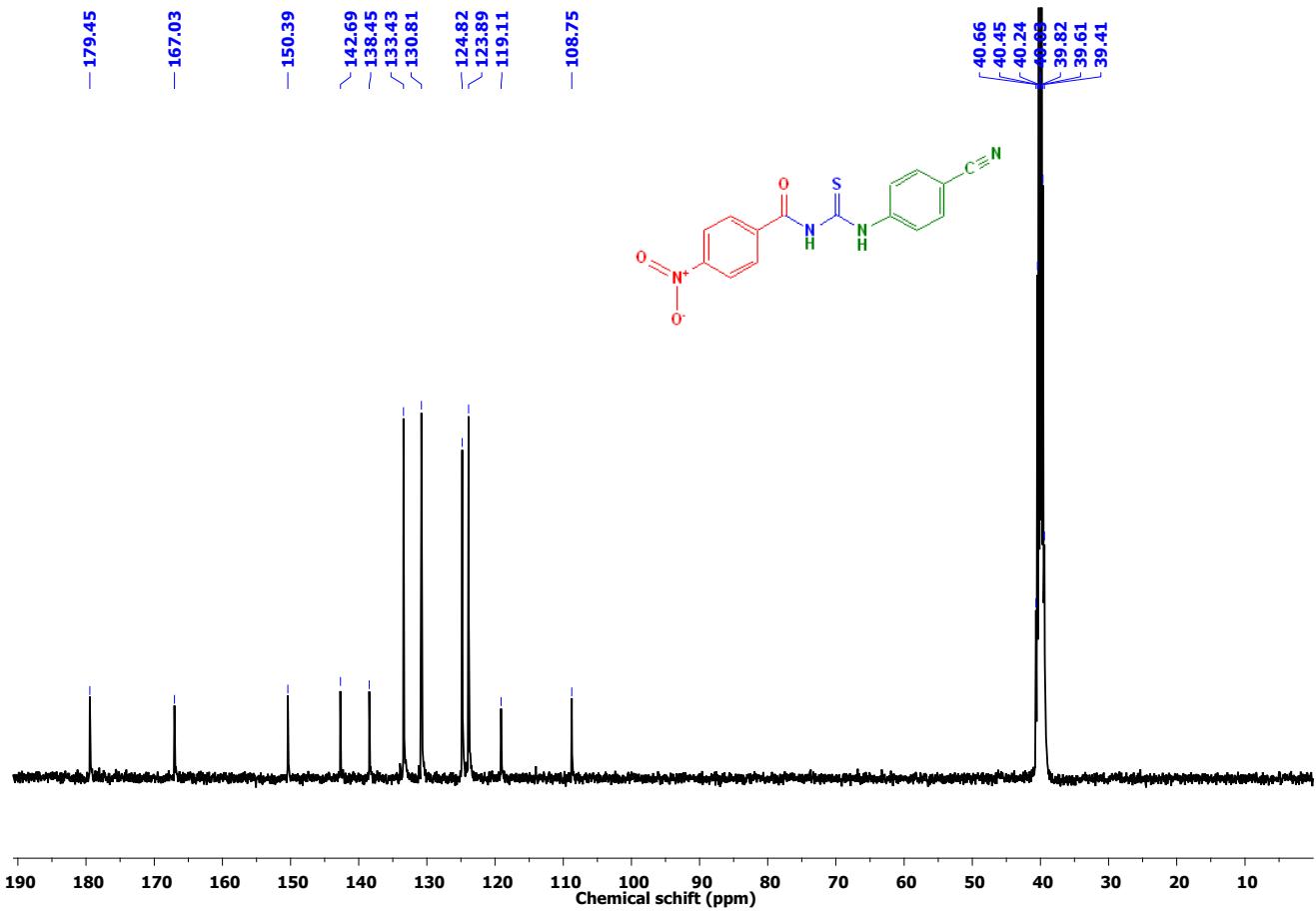
**Figure S3.** The <sup>1</sup>H-NMR spectrum of the N-benzoyl-N'-(4-cyanophenyl)-thiourea (**1**)



**Figure S4.** The <sup>1</sup>H-NMR spectrum of the N-(4-nitrobenzoyl)-N'-(4-cyanophenyl)-thiourea (**2**)



**Figure S5.** The  $^{13}\text{C}$ -NMR spectrum of the N-benzoyl-N'-(4-cyanophenyl)-thiourea (1)



**Figure S6.** The  $^{13}\text{C}$ -NMR spectrum of the N-(4-nitrobenzoyl)-N'-(4-cyanophenyl)-thiourea (**2**)

**Table S1.** Crystal data and structure refinement parameters for the N-Benzoyl-N'-(4-cyanophenyl)-thiourea (**1**).

<i>Crystal data</i>	
Chemical formula	C <sub>15</sub> H <sub>11</sub> N <sub>3</sub> OS
Mr	281.33
Crystal system, space group	Triclinic, <i>P</i> -1
Temperature (K)	296
a, b, c (Å)	4.0684 (3), 12.3410 (11), 14.7486 (13)
α, β, γ (°)	69.009 (3) 89.918 (3), 83.018 (3)
V (Å <sup>3</sup> )	685.52 (10)
Z	2
D <sub>x</sub> mg/mm	1.363
Radiation type, λ, Å	MoK <sub>a</sub> , , 0.71073
No. of reflections for cell measurement	7562
θ range (°) for cell measurement	2.7 - 21.4
μ (mm <sup>-1</sup> )	0.23
F(000)	292
Crystal shape	Block, colourless
Crystal size (mm)	0.30 × 0.145 × 0.12
Index ranges	-5 ≤ h ≤ 5, -19 ≤ k ≤ 19, -16 ≤ l ≤ 16
<i>Data collection</i>	
Diffractometer	BrukerAPEX-II CCD
Absorption correction	Multi-scan
No. of measured, independent and observed [I > 2σ(I)] reflections	34359, 3430, 2170
R <sub>int</sub>	0.069
Θ <sub>max</sub> , Θ <sub>min</sub> (°)	28.3, 1.5
<i>Refinement</i>	
R[F <sup>2</sup> > 2σ(F <sup>2</sup> )], wR(F <sup>2</sup> ), S	064, 0.213, 1.10
No. of reflections	3430
No. of parameters	181
H-atom treatment	H-atom parameters constrained w = 1/[σ <sup>2</sup> (F <sub>o</sub> <sup>2</sup> ) + (0.1242P) <sup>2</sup> + 0.0055P] where P = (F <sub>o</sub> <sup>2</sup> + 2F <sub>c</sub> <sup>2</sup> )/3
(Δ/σ) <sub>max</sub>	< 0.001
Δρ <sub>max</sub> , Δρ <sub>min</sub> (e Å <sup>-3</sup> )	0.61, -0.38

Computer programs: BrukerAPEX2, BrukerSAINT, SHELXT 2014/4 (Sheldrick, 2014), SHELXL2016/6 (Sheldrick, 2016).

**Table S2.** The optimized geometry (bond lengths (Å), bond angles (°) and torsion angles (°) of the the *N*-Benzoyl-*N'*-(4'-cyanophenyl)thiourea (**1**)

Geometric parameters (Å, °)			
<b>C1—C2</b>	<b>1.387 (4)</b>	<b>C8—S1</b>	<b>1.650 (3)</b>
<b>C1—C6</b>	<b>1.392 (4)</b>	<b>C9—C10</b>	<b>1.387 (4)</b>
<b>C1—H1</b>	<b>0.9300</b>	<b>C9—C14</b>	<b>1.392 (4)</b>
<b>C2—C3</b>	<b>1.371 (5)</b>	<b>C9—N2</b>	<b>1.414 (3)</b>
<b>C2—H2</b>	<b>0.9300</b>	<b>C10—C11</b>	<b>1.383 (4)</b>
<b>C3—C4</b>	<b>1.370 (4)</b>	<b>C10—H10</b>	<b>0.9300</b>
<b>C3—H3</b>	<b>0.9300</b>	<b>C11—C12</b>	<b>1.380 (4)</b>
<b>C4—C5</b>	<b>1.384 (4)</b>	<b>C11—H11</b>	<b>0.9300</b>
<b>C4—H4</b>	<b>0.9300</b>	<b>C12—C13</b>	<b>1.385 (4)</b>
<b>C5—C6</b>	<b>1.382 (4)</b>	<b>C12—C15</b>	<b>1.443 (4)</b>
<b>C5—H5</b>	<b>0.9300</b>	<b>C13—C14</b>	<b>1.373 (4)</b>
<b>C6—C7</b>	<b>1.489 (4)</b>	<b>C13—H13</b>	<b>0.9300</b>
<b>C7—O1</b>	<b>1.220 (3)</b>	<b>C14—H14</b>	<b>0.9300</b>
<b>C7—N1</b>	<b>1.380 (3)</b>	<b>C15—N3</b>	<b>1.140 (4)</b>
<b>C8—N2</b>	<b>1.342 (3)</b>	<b>N1—H1A</b>	<b>0.8600</b>
<b>C8—N1</b>	<b>1.400 (3)</b>	<b>N2—H2A</b>	<b>0.8600</b>

<b>C2—C1—C6</b>	<b>119.3 (3)</b>	<b>C10—C9—N2</b>	<b>125.2 (2)</b>
<b>C2—C1—H1</b>	<b>120.3</b>	<b>C14—C9—N2</b>	<b>115.3 (2)</b>
<b>C6—C1—H1</b>	<b>120.3</b>	<b>C11—C10—C9</b>	<b>119.4 (3)</b>
<b>C3—C2—C1</b>	<b>120.4 (3)</b>	<b>C11—C10—H10</b>	<b>120.3</b>
<b>C3—C2—H2</b>	<b>119.8</b>	<b>C9—C10—H10</b>	<b>120.3</b>
<b>C1—C2—H2</b>	<b>119.8</b>	<b>C12—C11—C10</b>	<b>121.0 (3)</b>
<b>C4—C3—C2</b>	<b>120.5 (3)</b>	<b>C12—C11—H11</b>	<b>119.5</b>
<b>C4—C3—H3</b>	<b>119.8</b>	<b>C10—C11—H11</b>	<b>119.5</b>
<b>C2—C3—H3</b>	<b>119.8</b>	<b>C11—C12—C13</b>	<b>119.6 (3)</b>
<b>C3—C4—C5</b>	<b>119.9 (3)</b>	<b>C11—C12—C15</b>	<b>120.3 (3)</b>
<b>C3—C4—H4</b>	<b>120.1</b>	<b>C13—C12—C15</b>	<b>120.1 (3)</b>
<b>C5—C4—H4</b>	<b>120.1</b>	<b>C14—C13—C12</b>	<b>120.0 (3)</b>
<b>C6—C5—C4</b>	<b>120.3 (3)</b>	<b>C14—C13—H13</b>	<b>120.0</b>
<b>C6—C5—H5</b>	<b>119.9</b>	<b>C12—C13—H13</b>	<b>120.0</b>
<b>C4—C5—H5</b>	<b>119.9</b>	<b>C13—C14—C9</b>	<b>120.6 (3)</b>
<b>C5—C6—C1</b>	<b>119.6 (3)</b>	<b>C13—C14—H14</b>	<b>119.7</b>
<b>C5—C6—C7</b>	<b>122.6 (2)</b>	<b>C9—C14—H14</b>	<b>119.7</b>
<b>C1—C6—C7</b>	<b>117.7 (3)</b>	<b>N3—C15—C12</b>	<b>179.7 (5)</b>
<b>O1—C7—N1</b>	<b>122.4 (2)</b>	<b>C7—N1—C8</b>	<b>129.1 (2)</b>

<b>O1—C7—C6</b>	<b>122.4 (2)</b>	<b>C7—N1—H1A</b>	<b>115.4</b>
<b>N1—C7—C6</b>	<b>115.2 (2)</b>	<b>C8—N1—H1A</b>	<b>115.4</b>
<b>N2—C8—N1</b>	<b>114.3 (2)</b>	<b>C8—N2—C9</b>	<b>131.8 (2)</b>
<b>N2—C8—S1</b>	<b>127.8 (2)</b>	<b>C8—N2—H2A</b>	<b>114.1</b>
<b>N1—C8—S1</b>	<b>117.88 (19)</b>	<b>C9—N2—H2A</b>	<b>114.1</b>
<b>C10—C9—C14</b>	<b>119.5 (2)</b>		

<b>C6—C1—C2—C3</b>	<b>-0.6 (5)</b>	<b>C10—C11—C12—C13</b>	<b>0.2 (5)</b>
<b>C1—C2—C3—C4</b>	<b>-0.6 (5)</b>	<b>C10—C11—C12—C15</b>	<b>179.6 (3)</b>
<b>C2—C3—C4—C5</b>	<b>0.9 (5)</b>	<b>C11—C12—C13—C14</b>	<b>0.2 (4)</b>
<b>C3—C4—C5—C6</b>	<b>0.0 (4)</b>	<b>C15—C12—C13—C14</b>	<b>-179.3 (3)</b>
<b>C4—C5—C6—C1</b>	<b>-1.2 (4)</b>	<b>C12—C13—C14—C9</b>	<b>0.0 (4)</b>
<b>C4—C5—C6—C7</b>	<b>-179.3 (3)</b>	<b>C10—C9—C14—C13</b>	<b>-0.5 (4)</b>
<b>C2—C1—C6—C5</b>	<b>1.5 (4)</b>	<b>N2—C9—C14—C13</b>	<b>178.6 (3)</b>
<b>C2—C1—C6—C7</b>	<b>179.6 (3)</b>	<b>O1—C7—N1—C8</b>	<b>-2.9 (4)</b>
<b>C5—C6—C7—O1</b>	<b>148.7 (3)</b>	<b>C6—C7—N1—C8</b>	<b>177.7 (2)</b>
<b>C1—C6—C7—O1</b>	<b>-29.4 (4)</b>	<b>N2—C8—N1—C7</b>	<b>-7.0 (4)</b>
<b>C5—C6—C7—N1</b>	<b>-31.9 (4)</b>	<b>S1—C8—N1—C7</b>	<b>173.7 (2)</b>
<b>C1—C6—C7—N1</b>	<b>150.0 (2)</b>	<b>N1—C8—N2—C9</b>	<b>179.9 (3)</b>
<b>C14—C9—C10—C11</b>	<b>0.9 (5)</b>	<b>S1—C8—N2—C9</b>	<b>-0.9 (4)</b>
<b>N2—C9—C10—C11</b>	<b>-178.1 (3)</b>	<b>C10—C9—N2—C8</b>	<b>-4.4 (5)</b>
<b>C9—C10—C11—C12</b>	<b>-0.7 (5)</b>	<b>C14—C9—N2—C8</b>	<b>176.6 (3)</b>

**Table S3.** Hydrogen bonds, intermolecular interactions and symmetry operations the *N*-Benzoyl-*N'*-(4'-cyanophenyl)thiourea (**1**) (Å°, °)

<b>D—H···A</b>	<b>D—H</b>	<b>H···A</b>	<b>D···A</b>	<b>D—H···A</b>
N1—H1A···S1 <sup>i</sup>	0.86	2.69	3.518 (2)	163
N2—H2A···O1	0.86	1.92	2.646 (3)	141
C5—H5···S1 <sup>i</sup>	0.93	2.87	3.239 (3)	105
C5—H5···S1 <sup>ii</sup>	0.93	2.87	3.773 (3)	164
C10—H10···S1	0.93	2.52	3.186 (3)	129

*Symmetry codes:* (i) -x+2, -y+2, -z+1; (ii) -x+1, -y+2, -z+1.

**Table S4.** Mulliken charges of title compounds

	Compound (1)	Compound (2)
Atoms	Charges	Charges
C1	-0.071	-0.087
C2	-0.184	-0.145
C3	-0.107	0.289
C4	-0.163	-0.116
C5	-0.082	-0.102
C6	-0.163	-0.125
C7	0.541	0.545
C8	0.203	0.200
C9	0.390	0.406
C10	-0.101	-0.089
C11	-0.089	-0.100
C12	0.070	0.072
C13	-0.066	-0.066
C14	-0.266	-0.265
C15	-0.241	-0.240
N1	-0.206	-0.699
N2	-0.823	-0.820
N3	-0.096	-0.090
N4	-	0.013
S	-0.082	-0.064
O1	-0.457	-0.445
O2	-	-0.253
O3	-	-0.255
H1	0.176	0.189
H2	0.156	0.219
H3	0.157	-
H4	0.195	0.219
H5	0.250	0.208
H1A	0.393	0.394
H2A	0.406	0.406
H10	0.183	0.186
H11	0.185	0.256
H12	0.187	0.187
H13		0.187