

Short Note

# 6-[(4-Chlorophenyl)(1*H*-1,2,4-triazol-1-yl)methyl]-3-methyl-2(3*H*)-benzoxazolone

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**Abstract:** A new hybrid molecule containing a triazole and a benzoxazolone ring was synthesized. The structure of 6-[(4-chlorophenyl)(1*H*-1,2,4-triazol-1-yl) methyl]-3-methyl-2(3*H*)-benzoxazolone was confirmed by IR, <sup>1</sup>H-, <sup>13</sup>C-NMR, MS and elemental analysis.

**Keywords:** triazole; 2(3*H*)-benzoxazolone; aromatase inhibitors; antifungal agents

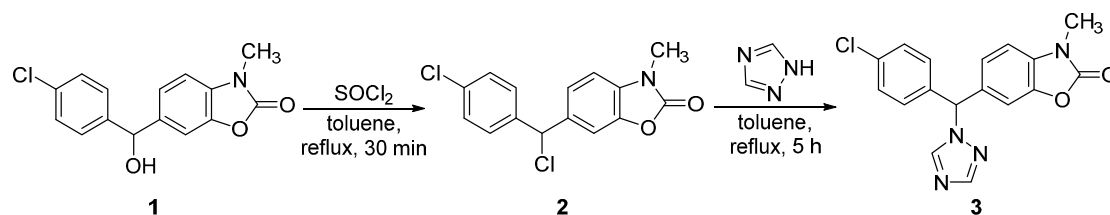
## 1. Introduction

Triazoles are an important class of heterocyclic compounds with antimicrobial, antimycotic, antimycobacterial, antiprotozoal, anticonvulsant and anticancer activity [1]. Commonly used triazole antifungal drugs include *fluconazole*, *itraconazole* and *voriconazole* [2,3]. These compounds have a broad spectrum of activity against both superficial and invasive fungal infections [4]. Other clinically useful triazole derivatives, such as *anastrozole*, *letrozole* and *vorozole*, act as non-steroidal aromatase inhibitors and find use in the treatment of estrogen dependent breast cancer [5,6].

Given the wide presence of the triazole moiety in bioactive compounds with diverse mechanisms of action, here, we propose the synthesis of 6-[(4-chlorophenyl)(1*H*-1,2,4-triazol-1-yl) methyl]-3-methyl-2(3*H*)-benzoxazolone (**3**). Structurally related hybrid compounds combining an imidazole fragment with a benzazole ring have been previously synthesized as aromatase inhibitors [7–9] and antifungal agents [10,11]. However, fewer molecules containing the triazole and benzoxazolone moieties have been reported. As a third ring in the target molecule, we chose the 4-chlorophenyl moiety, as the presence of chlorine atom on aromatic positions 2 or 4 in some cases improved the effectiveness in azole antifungal agents [12]. To the best of our knowledge, triazole derivative **3** has not been previously prepared.

## 2. Results

The synthesis of the title compound **3** was achieved in two steps, as presented in Scheme 1. Reaction of secondary alcohol **1** [8,11] with thionyl chloride led to the corresponding chloride **2**, which was used in the second stage without further purification. Treatment of **2** with an excess of 1,2,4-triazole provided **3** and triazole hydrochloride as a by-product. Compound **3** was isolated in good yield and purified by recrystallization. Compounds **1–3** possess a chiral center and can exist as a pair of enantiomers. We used racemic reagent **1** and the target compound **3** was correspondingly obtained as a mixture of both enantiomers. Their separation could be achieved by HPLC methods using polysaccharide-based stationary phases [13,14].



**Scheme 1.** Synthesis of 6-[(4-chlorophenyl)(1H-1,2,4-triazol-1-yl)methyl]-3-methyl-2(3H)-benzoxazolone (**3**).

The structure of **3** was confirmed by IR,  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR, MS and elemental analysis. All data are in accordance with the assumed structure. In the  $^1\text{H}$ -NMR spectrum two singlets corresponding to the triazole protons (7.98 and 8.03 ppm) were observed. In addition to the signals for aromatic protons in the 6.95–7.36 ppm range, a characteristic singlet for the methine proton was observed at 6.74 ppm. A singlet corresponding to the methyl group of benzoxazolone heterocycle was observed at 3.40 ppm (Figure S1).

### 3. Experimental Section

#### 3.1. General Information

All reagents were used as received without further purification. Compound **1** was prepared by reduction of the corresponding ketone with sodium borohydride as described previously [8,11]. The reaction was monitored by thin-layer chromatography (TLC) on silica gel plates (Kieselgel 60 F<sub>254</sub>), using cyclohexane/ethyl acetate (3:2) and ethyl acetate/isopropanol (3:1) as eluents. The purity of the final compound **3** was determined by GC-MS analyses on an Agilent 6890 system with MSD 5973, using electron-impact ionization (EI) at 70 eV on a HP-5MS column (30 m  $\times$  0.25 mm  $\times$  0.25  $\mu\text{m}$ ). Helium was used as a carrier gas with 0.8 mL/min flow rate. The temperature programmed mode was used (from 60  $^\circ\text{C}$  for 2 min, then with 10  $^\circ\text{C}/\text{min}$  to 300  $^\circ\text{C}$  for 10 min). The sample was introduced in splitless injection mode. IR spectrum ( $\text{CHCl}_3$ ) was recorded on a Shimadzu FTIR-8400S spectrometer (Kyoto, Japan). NMR spectra were obtained on a Bruker Avance III HD 500 (Rheinstetten, Germany) in  $\text{CDCl}_3$ . Chemical shifts are given in parts per million ( $\delta$ ) relative to the solvent peak ( $\delta$  7.26 ppm for  $^1\text{H}$ ;  $\delta$  77.16 ppm for  $^{13}\text{C}$ ). Coupling constants ( $J$ ) were measured in hertz (Hz). Elemental analysis (C, H, N) was performed on a VARIO EL III Elemental analyzer (Hanau, Germany) and the obtained results were within  $\pm 0.4\%$  of the theoretical values. Melting point of **3** was determined on a Boetius hot-stage microscope and was uncorrected.

#### 3.2. Synthesis of 6-[(4-Chlorophenyl)(1H-1,2,4-triazol-1-yl)methyl]-3-methyl-2(3H)-benzoxazolone (**3**)

Thionyl chloride (1 mL, 14 mmol) was added to a solution of 6-[(4-chlorophenyl)(hydroxy)methyl]-3-methyl-2(3H)-benzoxazolone (**1**, 1.40 g, 5 mmol) in toluene (10 mL). The mixture was refluxed for 30 min and then the excess of thionyl chloride was evaporated under reduced pressure. The obtained oil of **2** was dissolved in toluene (15 mL) and 1,2,4-triazole (0.7 g, 10 mmol) was added. The mixture was refluxed for 5 h (TLC) and the solvent was evaporated *in vacuo*. Water (20 mL) was added and the obtained precipitate was filtered off and dried. Recrystallization from toluene afforded **3** (68%, 1.15 g) as white crystals, m.p.: 185–186  $^\circ\text{C}$ . IR ( $\text{CHCl}_3$ ): 1755 ( $\text{C}=\text{O}$ )  $\text{cm}^{-1}$ .  $^1\text{H}$ -NMR (500 MHz,  $\text{CDCl}_3$ ):  $\delta$  3.40 (s, 3H,  $\text{NCH}_3$ ), 6.74 (s, 1H, CH), 6.95 (d, 1H, arom. H,  $J$  = 8.0 Hz), 6.99–7.01 (m, 2H, arom. H), 7.06 (d, 2H, arom. H,  $J$  = 8.4 Hz), 7.36 (d, 2H, arom. H,  $J$  = 8.4 Hz), 7.98 (s, 1H, triazole H), 8.03 (s, 1H, triazole H).  $^{13}\text{C}$ -NMR (125.8 MHz,  $\text{CDCl}_3$ ):  $\delta$  28.2 ( $\text{NCH}_3$ ), 66.7 (CH), 108.2, 109.9, 123.9, 129.2, 129.3, 132.2, 132.3, 134.9, 136.2, 142.9 (arom. C), 143.4, 152.6 (triazole C), 154.5 (CO). Anal. calcd. for  $\text{C}_{17}\text{H}_{13}\text{ClN}_4\text{O}_2$  (340.77): C, 59.92; H, 3.85; N, 16.44. Found: C, 60.25; H, 3.58; N, 16.23. MS (EI):  $[\text{M}]^+ m/z$  = 340(38),  $[\text{M} + 2]^+ m/z$  = 342(13), 272(100), 274(32), 152(35), 178(21), 207(20), 236(19).

**Supplementary Materials:** Copies of the  $^1\text{H}$ -,  $^{13}\text{C}$ -NMR and MS spectra as well as the molfile for compound 3 can be found at <http://www.mdpi.com/1422-8599/2016/2/M901>.

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**Author Contributions:** O.I.P. designed the experiments; M.S.G. and Y.B.I. performed the experiments and contributed to manuscript; M.S.G. and O.I.P. analyzed the NMR spectral data and wrote the manuscript; C.D.C. provided and analyzed the mass spectra. All authors read and approved the final manuscript.

**Conflicts of Interest:** The authors declare no conflict of interest.

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