



Communication

# The Influence of Copper and Zinc on Photosynthesis and Phenolic Levels in Basil (*Ocimum basilicum* L.), Borage (*Borago officinalis* L.), Common Nettle (*Urtica dioica* L.) and Peppermint (*Mentha piperita* L.)

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**Abstract:** This work is aimed at relationships which govern zinc and copper uptake by four popular medicinal herbs: basil (*Ocimum basilicum* L.), borage (*Borago officinalis* L.), common nettle (*Urtica dioica* L.) and peppermint (*Mentha piperita* L.). They are often grown in soils with significant copper or zinc levels. Herbs were cultivated by a pot method in controlled conditions. Manganese, iron, copper and zinc concentrations were determined by High-Resolution Continuum Source Flame Atomic Absorption Spectrometry. The efficiency of photosynthesis was estimated by measuring the chlorophyll content, water use efficiency, net photosynthesis, intercellular CO<sub>2</sub>, stomatal conductance, and transpiration rate. Phenolic compounds were determined by the Folin–Ciocalteu method. Analysis of variance showed that herbs grown in soil treated with copper exhibited a lower iron content in roots, while manganese behaved in the opposite way. The only exception was borage, where a decrease in the manganese content in roots was observed. Both copper and zinc supplementations increased the total content of phenolics, while the highest increases were observed for common nettle and basil. Peppermint and borage responded less to supplementation. In the majority of samples, zinc and copper did not significantly affect the photosynthesis. Herbal extracts from common nettle and basil had unique antioxidant properties and may be good free radical scavengers.

**Keywords:** heavy metals; polyphenols; herbs; photosynthesis



**Citation:** Adamczyk-Szabela, D.; Wolf, W.M. The Influence of Copper and Zinc on Photosynthesis and Phenolic Levels in Basil (*Ocimum basilicum* L.), Borage (*Borago officinalis* L.), Common Nettle (*Urtica dioica* L.) and Peppermint (*Mentha piperita* L.). *Int. J. Mol. Sci.* **2024**, *25*, 3612. <https://doi.org/10.3390/ijms25073612>

Academic Editors: Ilya Vladimirovich Seregin and Anna D. Kozhevnikova

Received: 31 January 2024

Revised: 20 March 2024

Accepted: 22 March 2024

Published: 23 March 2024



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## 1. Introduction

Heavy metals play a significant role in the growth and development of plants. Nevertheless, either their deficiency or excess can cause disorders in plant growth and development by affecting important physiological processes in plants [1].

The main sources of heavy metal contamination of soil are mining, exhaust emissions, sewage irrigation [2] and the continuously growing chemicalization of agriculture. However, in many places around the world, deficiencies of essential heavy metals are observed in agricultural crops. This phenomenon is often observed in alkaline soils, where the availability of microelements for plants may be quite limited indeed [3].

Zinc (Zn) and copper (Cu) are essential plant micronutrients [4]. Zinc is a structural component of zinc finger proteins and is pivotal for the synthesis of photosynthetic pigments like chlorophyll [5]. Food and Agriculture Organization of the United Nations (FAO) projects on the assessment of the level of microelements in agricultural soils around the world raise the problem of global zinc deficiency in crops [6]. Increasing the soil concentration of zinc in soils with a substantial deficit of this element brings a number of agronomic benefits in plant cultivation and productivity.

Copper acts as an essential cofactor of several enzymes that play key functions in plant cell metabolism including the respiration, photosynthesis, and scavenging of reactive

oxidative species (ROS) [7,8]. On the other hand, the redox properties of copper may contribute to its toxicity. Redox cycles between  $\text{Cu}^{2+}$  and  $\text{Cu}^+$  can increase the production of highly toxic hydroxyl radicals with subsequent damage to cells at the level of lipids, membranes, nucleic acids, proteins and other biomolecules [9]. Copper usually binds to proteins and has the ability to initiate oxidative damage and disrupt important cellular processes [10].

Therefore, significant changes in the content of copper or zinc in the soil are reflected by their respective levels in plant tissues. Above certain concentrations, those elements act as stressors and prompt the production of ROS [11]. In response, plants synthesize phenolic compounds which are effective free radicals scavengers [12].

Medicinal herbs and food species are important sources of bioactive compounds. Following the World Health Organization (WHO), almost 80% of populations heavily depend on herbal therapies [13]. In the polluted environment inhabited by people living in stressing conditions, there is a continuous demand for cheap and safe antioxidants with pronounced free radicals scavenging ability. They are represented by phenolic acids, phenolic diterpenes, flavonoids, and essential oils [14]. Plants containing these compounds are widely used in gastronomy, cosmetic industries, perfumery, and the pharmaceutical industry as well as in herbal medicine [15]. Preliminary studies indicate that they respond to heavy metal stress with the diverse magnitudes [16,17].

Basil (*Ocimum basilicum* L.) from the *Lamiaceae* family is a valuable herb that is used in medicine, food processing and cosmetics [18,19]. It contains a significant level of phenolic acids and flavonoid glycosides [20]. Basil has antispasmodic, antidiabetic, antibacterial, antifungal and antioxidant properties [21,22]. Borage (*Borago officinalis* L.) is an annual herb in the flowering plant family *Boraginaceae*. Its medicinal value is highly appreciated by either the contemporary pharmaceutical industry or traditional medicine. It is used as an effective anti-inflammatory agent in the prevention of colds, bronchitis and respiratory infections [23,24]. Moreover, borage lowers blood cholesterol levels and helps to fight digestive and cardiovascular disorders [25,26]. Common nettle (*Urtica dioica* L.) is a spice from the nettle family (*Urticaceae*). It is a medicinal, cosmetic, edible and feed plant [27]. Nettle has antihemorrhagic properties, increases the number of red blood cells, regulates sugar levels and replenishes the deficiencies of vitamins and mineral salts [28]. Peppermint (*Mentha piperita* L.) belongs to the *Lamiaceae* family in the genus *Mentha* [29]. The herbal raw material are leaves (*Menthae piperitae* folium) containing mainly mint oil, ascorbic acid, carotene, rutin, apigenin, betaine, oleanic and ursolic acids. Peppermint is used for gastrointestinal ailments, migraines and upper respiratory tract diseases. It has antibacterial and calming properties [30,31]. The quality of herbs, their extracts and essential oils strongly depends on the plants' development and their growing conditions [32]. Herb crops require appropriate fertilization, and the number of permissible plant protection products is very limited.

The goal of this work was to describe relationships which govern zinc and copper uptake by four popular medicinal herbs. The plant metabolism was to be assessed by gasometric analysis. Finally, we aimed at cultivation conditions which prompt the synthesis of phenolic compounds. The latter are useful medicinal substances and may be used as plant stress indicators.

## 2. Results

### 2.1. Soil Analysis

The soil used for growing herbs was acidic (pH 5.3) and belonged to organic soils (86%). The concentrations of heavy metals in this soil did not exceed the limit values (Table 1) in agreement with the international standards [33,34].

**Table 1.** Heavy metals content in the soil used in the work. The number of samples  $n = 5$ , probability level  $p = 0.05$ .

Metals	Bioavailable Forms ( $\mu\text{g/g}$ )	Pseudo-Total Forms ( $\mu\text{g/g}$ )	Limit Values ( $\mu\text{g/g}$ ) *
Mn	$75.7 \pm 1.0$	$108 \pm 0.6$	not applicable
Fe	$1323 \pm 46$	$2035 \pm 190$	not applicable
Cu	$9.46 \pm 0.06$	$16.7 \pm 0.1$	150
Zn	$61.8 \pm 0.8$	$93.3 \pm 0.39$	300

\* according to [33,34].

## 2.2. Plants Analysis

### 2.2.1. Heavy Metals Uptake by Herbs

Heavy metal concentrations in the above-ground parts and roots of herbs are presented in Tables 2 and 3. Iron, copper and zinc accumulated in the roots of all cultivated plants, while manganese accumulated mainly in the above-ground parts. The only exception is peppermint. To assess the impact of all applied treatments on the concentration of heavy metals in cultivated herbs, one-way ANOVA analysis was used at the probability level  $p = 0.05$  (Table 4). Calculations showed that both copper and zinc influence the content of manganese and iron in all tested plants. Copper supplementation triggers a decrease in the manganese content in the above-ground parts of borage, common nettle and peppermint. The exception was basil, where an increase in the content of the latter element was observed. It is notable that the addition of copper limited the iron levels in the above-ground parts of all herbs. Zinc supplementation decreased the manganese and iron contents in the above-ground parts of herbs. The ANOVA analysis (Table 5) clearly showed that herbs which had been grown in the soil treated with copper exhibited a lower iron content in roots. Manganese behaved in the opposite way. The only exception was borage, where a decrease in the manganese content in the roots was observed.

**Table 2.** Heavy metals contents in above-ground parts of herbs cultivated in soils under copper and zinc supplementations. The number of samples  $n = 5$ , probability level  $p = 0.05$ . 50Cu = 50  $\mu\text{g/g}$  Cu; 50Zn = 50  $\mu\text{g/g}$  Zn. Specific pairs of letters as given in parentheses illustrate the statistically significant differences between treatments as computed with the Tukey's HSD test for separate metal and plant combinations.

Treatments	Metal Contents in Above-Ground Parts ( $\mu\text{g/g}$ )			
	Mn	Fe	Cu	Zn
<b>Basil</b>				
Control	$59.7 \pm 3.1$ (aa)	$148 \pm 11$ (ba)	$10.8 \pm 1.8$ (ca)	$73 \pm 5$ (da)
50Cu	$66.3 \pm 4.7$ (aa)	$121 \pm 6$ (bb)	$47.2 \pm 3.1$ (cb)	$52 \pm 7$ (db)
50Zn	$45.1 \pm 3.9$ (ab)	$104 \pm 7$ (bc)	$7.61 \pm 1.22$ (ca)	$191 \pm 10$ (dc)
<b>Borage</b>				
Control	$52.8 \pm 2.5$ (ea)	$80.4 \pm 6.2$ (fa)	$10.1 \pm 0.8$ (ga)	$56.7 \pm 4.8$ (ha)
50Cu	$29.8 \pm 2.1$ (eb)	$71.2 \pm 5.1$ (fb)	$29.4 \pm 1.9$ (gb)	$60.2 \pm 5.0$ (hb)
50Zn	$30.6 \pm 2.1$ (eb)	$65.4 \pm 6.8$ (fc)	$8.53 \pm 0.88$ (gc)	$138 \pm 8$ (hc)
<b>Common nettle</b>				
Control	$69.7 \pm 6.7$ (ia)	$103 \pm 6$ (ja)	$7.42 \pm 0.44$ (ka)	$29.1 \pm 2.8$ (la)
50Cu	$58.2 \pm 7.1$ (ib)	$95 \pm 7$ (jb)	$28.5 \pm 1.9$ (kb)	$33.8 \pm 2.8$ (la)
50Zn	$47.5 \pm 6.3$ (ic)	$101 \pm 7$ (ja,jb)	$8.12 \pm 0.73$ (ka)	$94.6 \pm 9.1$ (lb)
<b>Peppermint</b>				
Control	$63.3 \pm 3.1$ (ma)	$115 \pm 11$ (na)	$7.35 \pm 0.72$ (oa)	$48.6 \pm 3.7$ (pa)
50Cu	$49.6 \pm 3.7$ (mb)	$108 \pm 9$ (na,nb)	$31.8 \pm 5.8$ (ob)	$34.2 \pm 3.3$ (pb)
50Zn	$33.7 \pm 2.9$ (mc)	$92.6 \pm 7$ (nb)	$6.39 \pm 2.52$ (oa)	$105 \pm 5$ (pc)

**Table 3.** Heavy metals content in roots of herbs cultivated in soils under copper and zinc supplementation. The number of samples  $n = 5$ , probability level  $p = 0.05$ . 50Cu = 50  $\mu\text{g/g}$  Cu; 50Zn = 50  $\mu\text{g/g}$  Zn. Specific pairs of letters as given in parentheses illustrate the statistically significant differences between treatments as computed with Tukey’s HSD test for separate metal and plant combinations.

Treatments	Metal Content in Roots ( $\mu\text{g/g}$ )			
	Mn	Fe	Cu	Zn
<b>Basil</b>				
Control	31.2 $\pm$ 2.1 (aa)	198 $\pm$ 6 (ba)	15.9 $\pm$ 1.9 (ca)	109 $\pm$ 6 (da)
50Cu	70.5 $\pm$ 4.3 (ab)	181 $\pm$ 5 (bb)	113 $\pm$ 5 (cb)	92.1 $\pm$ 7.2 (db)
50Zn	17.1 $\pm$ 3.4 (ac)	138 $\pm$ 7 (bc)	13.2 $\pm$ 2.0 (cd)	287 $\pm$ 12 (dc)
<b>Borage</b>				
Control	40.8 $\pm$ 2.5 (ea)	114 $\pm$ 6 (fa)	11.5 $\pm$ 0.8 (ga)	62.7 $\pm$ 4.8 (ha)
50Cu	32.4 $\pm$ 2.1 (eb)	85.8 $\pm$ 5.1 (fb)	37.4 $\pm$ 1.9 (gb)	70.2 $\pm$ 5.0 (hb)
50Zn	40.6 $\pm$ 1.9 (ea)	75.4 $\pm$ 6.8 (fc)	10.1 $\pm$ 0.9 (gc)	154 $\pm$ 8 (hc)
<b>Common nettle</b>				
Control	49.7 $\pm$ 6.7 (ia)	172 $\pm$ 6 (ja)	9.42 $\pm$ 0.44 (ka)	31.1 $\pm$ 2.8 (la)
50Cu	68.2 $\pm$ 7.1 (ib)	125 $\pm$ 7 (jb)	28.5 $\pm$ 1.9 (kb)	23.8 $\pm$ 2.8 (lb)
50Zn	57.5 $\pm$ 6.3 (ic)	137 $\pm$ 7 (jc)	7.12 $\pm$ 0.73 (kc)	114 $\pm$ 9 (lc)
<b>Peppermint</b>				
Control	71.3 $\pm$ 7.1 (ma)	175 $\pm$ 9 (na)	8.34 $\pm$ 0.78 (oa)	63.6 $\pm$ 7.7 (pa)
50Cu	91.6 $\pm$ 8.7 (mb)	128 $\pm$ 7 (nb)	43.8 $\pm$ 5.8 (ob)	44.2 $\pm$ 3.3 (pb)
50Zn	83.7 $\pm$ 2.9 (mc)	142 $\pm$ 8 (nc)	9.32 $\pm$ 2.56 (oa)	115 $\pm$ 6 (pc)

**Table 4.** The one-way ANOVA for manganese, iron, copper, and zinc contents in above-ground parts of herbs cultivated in soil under copper and zinc supplementations. Critical Snedecor’s F value is  $F_{\text{cryt}} = 3.8853$ , probability level  $p = 0.05$ .

	Basil	Borage	Common Nettle	Peppermint
	Above-Ground Parts			
Mn	$p = 3.85 \times 10^{-6}$ F = 41.9357	$p = 1.47 \times 10^{-8}$ F = 115.2513	$p = 4.95 \times 10^{-8}$ F = 93.0262	$p = 1.70 \times 10^{-9}$ F = 167.6868
Fe	$p = 1.21 \times 10^{-8}$ F = 119.1513	$p = 2.48 \times 10^{-7}$ F = 69.7012	$p = 2.02 \times 10^{-2}$ F = 5.4899	$p = 2.12 \times 10^{-3}$ F = 10.7391
Cu	$p = 6.45 \times 10^{-11}$ F = 387.6501	$p = 1.52 \times 10^{-9}$ F = 204.5946	$p = 5.19 \times 10^{-10}$ F = 254.4199	$p = 5.04 \times 10^{-11}$ F = 407.4172
Zn	$p = 1.17 \times 10^{-11}$ F = 546.103	$p = 3.21 \times 10^{-9}$ F = 175.8112	$p = 1.60 \times 10^{-10}$ F = 322.7082	$p = 1.11 \times 10^{-9}$ F = 218.2207

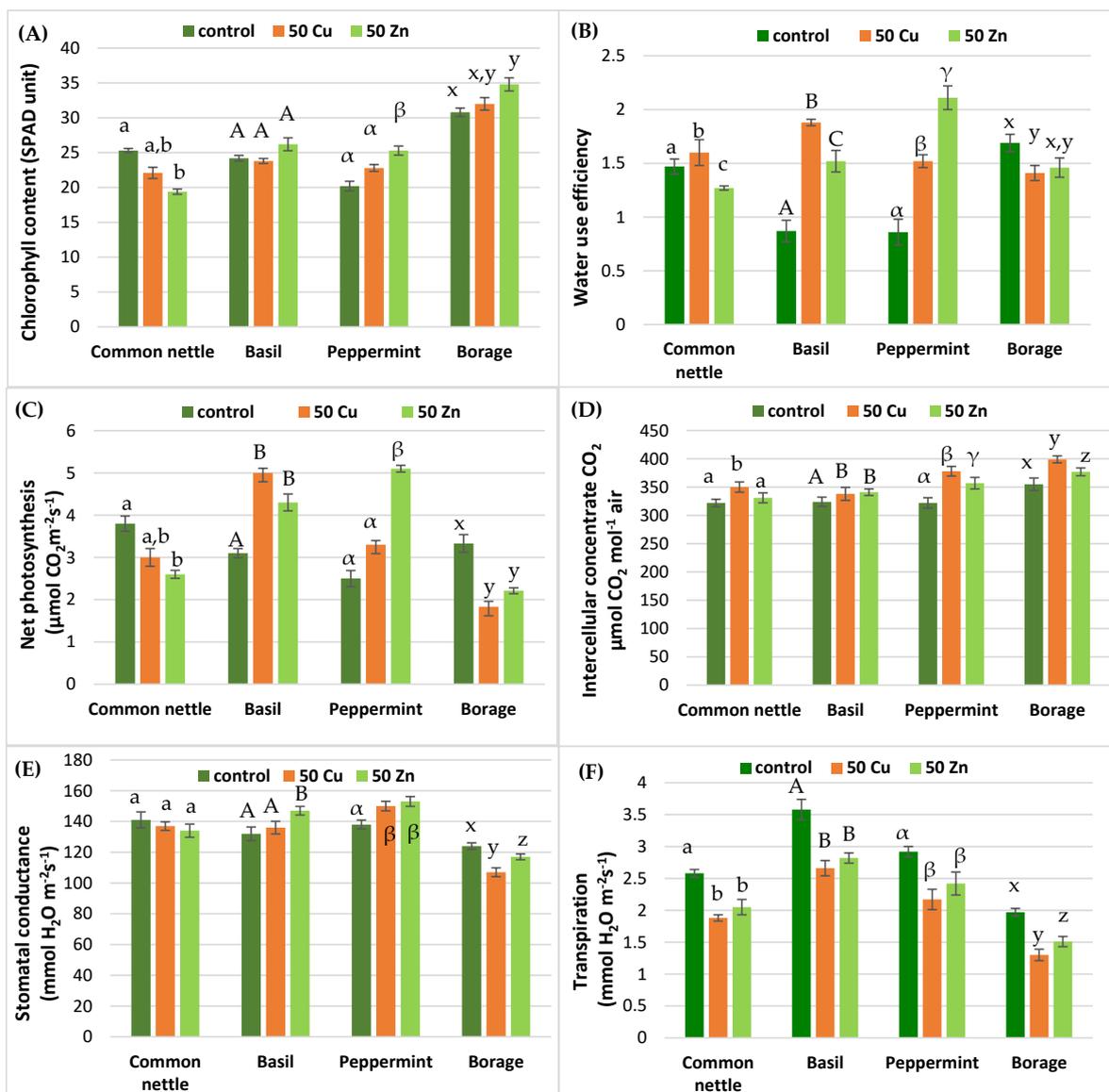
**Table 5.** The one-way ANOVA for manganese, iron, copper, and zinc contents in roots of herbs cultivated in soil under copper and zinc supplementations. Critical Snedecor’s F value is  $F_{\text{cryt}} = 3.8853$ , probability level  $p = 0.05$ .

	Basil	Borage	Common Nettle	Peppermint
	Roots			
Mn	$p = 8.23 \times 10^{-13}$ F = 613.7908	$p = 2.92 \times 10^{-4}$ F = 17.3007	$p = 1.31 \times 10^{-7}$ F = 78.1844	$p = 5.46 \times 10^{-7}$ F = 60.3707
Fe	$p = 7.51 \times 10^{-9}$ F = 129.5797	$p = 2.69 \times 10^{-8}$ F = 103.6419	$p = 1.14 \times 10^{-9}$ F = 179.6911	$p = 8.17 \times 10^{-8}$ F = 85.0893
Cu	$p = 5.49 \times 10^{-13}$ F = 1010.294	$p = 3.28 \times 10^{-8}$ F = 109.3599	$p = 4.32 \times 10^{-10}$ F = 264.0011	$p = 1.04 \times 10^{-7}$ F = 86.1670
Zn	$p = 5.72 \times 10^{-11}$ F = 397.1455	$p = 4.88 \times 10^{-11}$ F = 409.9500	$p = 7.27 \times 10^{-9}$ F = 148.8850	$p = 1.17 \times 10^{-7}$ F = 84.1723

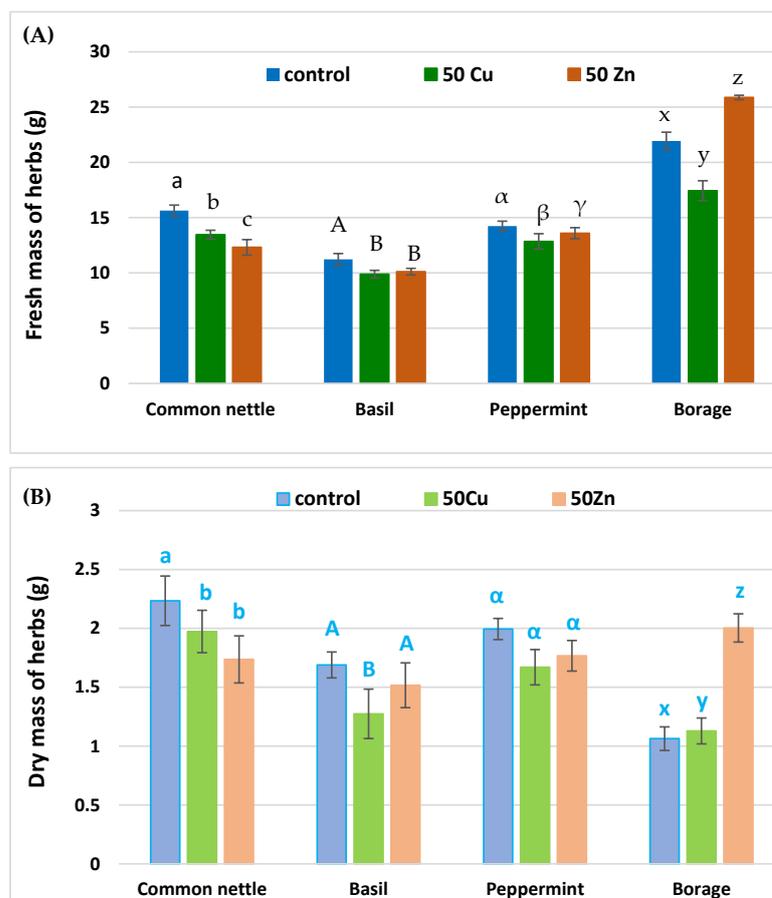
2.2.2. Photosynthesis Parameters

The metabolism of herbs was estimated using photosynthesis indicators, i.e., the index of chlorophyll content in leaves, water use efficiency (WUE), net photosynthesis activity,

stomatal conductance, transpiration rate and intercellular CO<sub>2</sub> concentration (Figure 1). Those parameters clearly showed that herbs were in reasonable growth conditions. The influence of copper or zinc on the chlorophyll content in herbs varies greatly and was dependent on the plant species. A clear decrease was observed only for the common nettle. In peppermint and borage grown in soil with the addition of metals, the content chlorophyll increased. The WUE clearly showed that the analyzed herbs reacted on metals introduced into the soil in quite different ways. The photosynthesis parameters of the analyzed herbs treated with copper or zinc were strictly dependent on the plant species. A common nettle grown on soil supplemented with copper or zinc showed a decrease in the intensity of the photosynthesis process. The opposite situation was observed for peppermint. Basil and borage behaved in a more diverse way. An increase in intercellular CO<sub>2</sub> content was observed in all analyzed herbs after supplementation with copper and zinc. The mass of common nettle, peppermint and basil-treated zinc or copper (Figure 2) decreased slightly. The only exception was borage grown in soil supplemented with zinc.



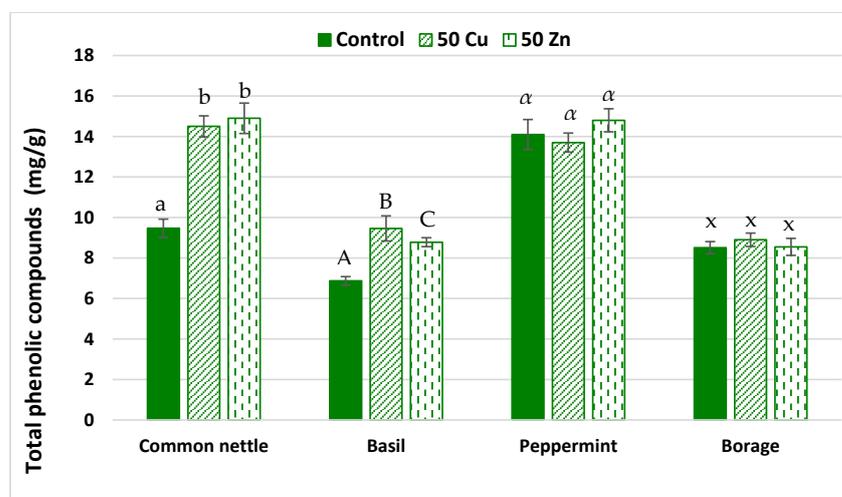
**Figure 1.** The chlorophyll content (A), water use efficiency (B), net photosynthesis (C), intercellular concentration CO<sub>2</sub> (D), stomatal conductance (E), transpiration (F). Specific letters illustrate the statistically significant differences as computed with Tukey's HSD test. Error bars correspond to the S.D. The number of samples  $n = 5$ , probability level  $p = 0.05$ .



**Figure 2.** The fresh mass of herbs (A) and dry mass of herbs (B) cultivated on soil with additives. Specific letters demonstrate the statistically significant differences as computed with Tukey's HSD test. Error bars correspond to the S.D. The number of samples  $n = 5$ , probability level  $p = 0.05$ .

### 2.2.3. Total Phenolic Compounds Content

Figure 3 presents the total phenolic compounds (TPCs) content as determined by the Folin–Ciocalteu method. Both copper and zinc supplementation increased the total content of phenolic compounds in stinging nettle and basil.



**Figure 3.** The total phenolic compounds in herbs cultivated in soil with copper and zinc supplementation. Specific letters demonstrate the statistically significant differences as computed with Tukey's HSD test. Error bars correspond to the S.D. The number of samples  $n = 5$ , probability level  $p = 0.05$ .

The content of phenolic compounds in common nettle increased by 53% after copper supplementation and by 57% when the zinc treatment was applied. For basil, those values were 38% (copper supplementation) and 28% (zinc supplementation), respectively, while no significant differences were observed for peppermint and borage.

### 3. Discussion

It is well known that heavy metals in the soil environment tend to migrate to the rhizosphere and are subjected to the uptake by plant roots there. Subsequently, they are transported via xylem and phloem to the upper parts of the plant [35]. Our investigations clearly show that the impact of heavy metals on herbs is quite diverse and heavily depends on individual species. Significant antagonistic interactions between zinc and copper were observed in all of the herbs investigated. In particular, the zinc uptake could be inhibited by the increased copper content in roots. This result may indicate that the similar carrier sites are involved in the absorption and transport mechanisms of both metals [36].

In particular, Behtash et al. [37] examined the effect of copper and zinc on the morpho-physiological characteristics of spring squash. Increased zinc content was observed in the presence of small copper concentrations, while the low zinc levels prompted the copper uptake. The authors pointed out that copper in high concentrations directly had affected the reaction centers of the photosystem and had interfered with photosynthetic processes by annihilating the photosynthetic pigments, membrane stability and photosynthetic enzymes.

It is common knowledge that any disturbances in the photosynthesis process adversely affect the quality of crops. For this reason, the antagonistic interaction of zinc and copper may significantly reduce the effects of copper toxicity. However, we did not observe this effect along the applied copper supplementation. Zinc contents in the aerial parts of cultivated herbs varied greatly depending on the plant species [37]. The diverse interactions of copper and zinc have been described many times. Le et al. [38] examined the interaction of both those elements and found that the presence of zinc reduces the toxicity of copper, while the presence of copper does not affect the toxicity of zinc. Moreover, results may be biased by interactions with other ions, e.g., zinc is transported through root cells by the ZIP family of transporters, which can also bind iron [39–41].

Kabata-Pendias and Pendias [42] report that zinc interferes more with the uptake and transport of iron than copper. This is likely due to competition between  $Zn^{2+}$  and  $Fe^{2+}$  ions during uptake by plant roots and interference in chelation processes. High levels of copper in the plant decrease the iron content. The optimal copper–iron ratio varies for different plant species. In particular, the uptake and transport of iron are greatly influenced by the concentrations and proportions of other heavy metals [43]. Iron is considered an element that plays a key role in the mechanisms of photosynthesis [44].

It is quite common knowledge that heavy metals induce stress to plants and prompt multidirectional metabolic disorders which further limit photosynthesis and biomass productivity. Thus, the reduced weight of plants grown in the soil with increased copper or zinc levels may result from oxidative stress, a limited absorption of essential elements and reduced metabolism [45]. Nevertheless, it should be remembered that the response of plants to stress is closely related to plant species and the level of heavy metals in the cultivation environment. Our research clearly shows that common nettle is the most sensitive to the presence of these additives. However, modifications of the root system may also affect the final weight of particular herbs [46].

Heavy metals may limit the uptake of nutrients, the content of photosynthetic pigments, enzymatic activity and protein biosynthesis [47]. However, several studies indicate that the plant stress induced by toxic elements may induce secondary metabolism [48,49]. The plant response to stress caused by heavy metals affects their stomata. The latter are specialized pores in the epidermis of plant cells involved in the process of photosynthesis, respiration and transpiration. According to the literature data, the exposure of plants to heavy metals may cause damage to the structure and function of stomata and ultimately lead to changes in the physiology and ecology of plants [50]. The reasons for stomatal

closure in plants are complex and partially may be related to the concentration of heavy metals and the duration of this contamination [51,52]. Moreover, heavy metals may affect the carbon dioxide concentrations in the plant tissues [53]. In response, plants may increase stomatal conductance to meet their respiratory needs. In addition, the accumulation of heavy metals can lead to the leakage of potassium ions from the plant, negatively affecting the plant's ability to regulate stomatal closure [54]. We clearly observed this phenomenon in basil and peppermint after supplementation with copper and zinc (Figure 1). In all of the analyzed herbs grown in soil with the addition of metals, an increase in intercellular CO<sub>2</sub> content was observed.

Physiological processes related to water absorption and nutrient uptake significantly influence the growth and development of plants [55,56]. The literature data [57] clearly show that a significant increase in the content of heavy metals in the cultivation media reduces the water content in plant organs. However, particular plant species use specific mechanisms to maintain plant water balance [58]. Our results showed clearly that WUE values changed substantially for each cultivated plant after the copper or zinc addition.

Upon uptake, heavy metals penetrating plant tissues prompt an increase in reactive oxygen species (ROS), which further lead to redox balance disturbance [59,60]. Increased levels of superoxide anion (O<sup>2-</sup>) and hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) cause a further peroxidation of membrane lipids and the destruction of oxidative cells [61]. ROS are typically produced in plant cell chloroplasts, mitochondria, and subcellular structures [62]. In response to stress, plants may adapt to it or reduce the ROS level by changing the activity of antioxidant enzymes in their tissues [63,64]. Low levels of ROS can activate plant defense reactions, which leads to an increased activity of antioxidant enzymes. However, when ROS levels exceed the ability of the antioxidant defense system, oxidative damage occurs [65].

Polyphenols are compounds naturally existing in plants, and they are involved in natural plant antioxidant activity. They are scavengers of free radicals either in plant or human body environments [66]. Our results clearly showed that the addition of copper or zinc to the soil increased the total content of phenolic compounds in common nettle and basil. However, they responded to applied supplementations in diverse ways. It is well documented that plants grown in contaminated soils react to stress conditions by synthesizing phenolic compounds [66–68]. Moreover, Mleczek et al. [41] determined that only high levels of copper and zinc prompted a significant increase in the total phenolics.

## 4. Materials and Methods

### 4.1. Soil Analysis and Preparation for Cultivation

The soil used to grow herbs was taken from the agricultural area located in Lagiewniki, Poland, according to standards [69]. The soil was dried and passed through a 2 mm stainless steel sieve. The potentiometric method was used in pH determination [70]. The content of organic matter was estimated by the gravimetric method [71]. The bioavailable forms of metals were analyzed in 0.5 mol/L HCl extracts. Pseudo-total metals contents in soil were determined in solution obtained by microwave decomposition in a mixed HNO<sub>3</sub> (65%) and HCl (36%). The Anton Paar Multiwave 3000 (Graz, Austria) apparatus was used. Metal concentrations were determined by the High Resolution-Continuum Source Flame Atomic Absorption Spectrometry (HR-CS FAAS) with the Analytik Jena ContrAA 300 (Jena, Germany) apparatus. Each sample was analyzed five times.

The air-dried soil was weighed (200 g) into plastic containers. The first series of soil without added metals was a reference (Control). The remaining two series were supplemented with solution of zinc or copper in the form of Cu(NO<sub>3</sub>)<sub>2</sub> and Zn(NO<sub>3</sub>)<sub>2</sub>. The metal concentrations in these samples were increased by 50 µg/g Cu (50Cu) or 50 µg/g Zn (50Zn), respectively. Copper and zinc supplementations were calculated to be consistent with the amounts of those elements which are being introduced to soil with fertilizers used in herbal agriculture.

#### 4.2. Plant Material Preparation

Four herbs were cultivated, i.e., basil, borage, common nettle, and peppermint. Each plant was grown in three series of five replicates (control, samples with the addition of 50 µg/g Cu (50Cu) and samples with the addition of 50 µg/g Zn (50Zn)). The seeds of all plants came from the P.H. Legutko company. Herbs were cultivated in a greenhouse under controlled conditions: temperatures  $23 \pm 2$  and  $16 \pm 2$  °C for day and night, respectively; the relative humidity was limited to 70–75%; the photosynthetic active radiation (PAR) during the 16 h photoperiod was restricted to 400 µmol/m<sup>2</sup> s. Plant cultivation was carried out for three months. After determining the parameters of photosynthesis, all plants were cut, dried in the air and homogenized.

#### 4.3. Photosynthesis Parameters

All measurements were collected five times from plants from each pot. The content of chlorophyll in leaves was measured by Konica Minolta SPAD-502 (Tokyo, Japan) in the red and the near-infrared regions. The activity of net photosynthesis ( $P_N$ ), the stomatal conductance ( $G_S$ ), the intercellular concentration of carbon dioxide ( $C_i$ ), and the transpiration ( $E$ ) were measured with the gas analyzer TPS-2 (Portable Photosynthesis System, Amesbury MA, USA).

#### 4.4. Determination of Total Phenolic Compounds in Herbs by Folin–Ciocalteu Method

The commonly used Folin–Ciocalteu method with gallic acid as the standard was used to measure the total phenolics content in herbs [72]. The boiling water (50 mL) was used to extract dried herbs (1 g). The absorbance measurement was made at a wavelength of 765 nm on the Specol 11, Carl Zeiss (Jena, Germany) instrument.

#### 4.5. Determination of Heavy Metals in Herbs

The metals in roots and above-ground parts of herbs contents were measured in mineralizates of plant. The same protocol as used for soil analysis was applied.

The quality assurance and quality control (QA/QC) of metals in plant samples were assessed by determining the metal content in the INCT-MPH-2 certified reference material containing a mixture of selected Polish herbs (Table S1).

#### 4.6. Data Analysis

All parameters were determined in parallel for five independent samples. The statistical analysis was made separately for each herb. Bartlett and Hartley tests were used to check the equality of variances (STATISTICA 10 PL package). The normality of the data sets was assessed using the Shapiro–Wilk test [73,74]. The Tukey HDS post hoc test was used to assess statistically significant differences among individual parameters. One-way ANOVA was used to identify significant differences in manganese, iron, copper and zinc contents in the above-ground parts and roots of herbs cultivated in soil with copper and zinc supplementations.

### 5. Conclusions

The addition of copper or zinc to the soil increased the polyphenol content in common nettle and basil. However, this influence varied substantially depending on the plant species. In most plant samples, it did not significantly affect the photosynthesis. Herbal extracts from common nettle and basil had unique antioxidant properties and may be good free radical scavengers. Investigations on increasing the content of polyphenols in plants are important for their common medical use. On the other hand, determining the safe levels of metals supplemented to the soil in order to increase phenolic compounds in herbs is crucial to obtain proper pharmaceutical materials.

**Supplementary Materials:** The supporting information can be downloaded at <https://www.mdpi.com/article/10.3390/ijms25073612/s1>.

**Author Contributions:** Conceptualization, D.A.-S. and W.M.W.; methodology, D.A.-S. and W.M.W.; software, D.A.-S. and W.M.W.; validation, D.A.-S. and W.M.W.; formal analysis, D.A.-S. and W.M.W.; investigation, D.A.-S.; resources, D.A.-S. and W.M.W.; data curation, D.A.-S. and W.M.W.; writing—original draft preparation, D.A.-S. and W.M.W.; writing—review and editing, D.A.-S. and W.M.W.; visualization, D.A.-S. and W.M.W.; supervision, D.A.-S. and W.M.W.; project administration, D.A.-S. and W.M.W.; funding acquisition, D.A.-S. and W.M.W. All authors have read and agreed to the published version of the manuscript.

**Funding:** This research received no external funding.

**Institutional Review Board Statement:** Not applicable.

**Informed Consent Statement:** Not applicable.

**Data Availability Statement:** The data sets used and analyzed during the current study available from the corresponding author on reasonable request.

**Acknowledgments:** The European University Foundation is acknowledged for advising on the legal and social dimension of this study and Jakub Kubicki for advice and help during heavy metal determinations.

**Conflicts of Interest:** The authors declare no conflicts of interest.

## References

1. Arif, N.; Yadav, V.; Singh, S.; Singh, S.; Ahmad, P.; Mishra, R.K.; Sharma, S.; Tripathi, D.K.; Dubey, N.K.; Chauhan, D.K. Influence of high and low levels of plant-beneficial heavy metal ions on plant growth and development. *Front. Environ. Sci.* **2016**, *4*, 69. [\[CrossRef\]](#)
2. He, H.Y.; Li, Y.; He, L.-F. The central role of hydrogen sulfide in plant responses to toxic metal stress. *Ecotoxicol. Environ. Saf.* **2018**, *157*, 403–408. [\[CrossRef\]](#)
3. Seeda Abou, M.A.; Abou El-Nour, E.A.A.; Yassen, A.A.; Gad, M.M.; Sahar, M.Z. Interaction of copper, zinc, and their importance in plant physiology: Review, Acquisition and Transport. *Middle East J. Appl. Sci.* **2020**, *10*, 407–434.
4. Rizwan, M.; Ali, S.; Maqbool, A. A critical review on the effects of zinc at toxic levels of cadmium in plants. *Environ. Sci. Pollut. Res.* **2019**, *26*, 6279–6289. [\[CrossRef\]](#)
5. Han, G.; Qiao, Z.; Li, Y.; Wang, C.; Wang, B. The roles of CCCH zinc-finger proteins in plant abiotic stress tolerance. *Int. J. Mol. Sci.* **2021**, *22*, 8327. [\[CrossRef\]](#) [\[PubMed\]](#)
6. Alloway, B.J. Soil factors associated with zinc deficiency in crops and humans. *Environ. Geochem. Health* **2009**, *3*, 537–548. [\[CrossRef\]](#)
7. Pietrini, F.; Carnevale, M.; Beni, C.; Zacchini, M.; Gallucci, F.; Santangelo, E. Effect of different copper levels on growth and morpho-physiological parameters in Giant Reed (*Arundo donax* L.) in semi-hydroponic mesocosm experiment. *Water* **2019**, *11*, 1837. [\[CrossRef\]](#)
8. Badiia, O.; Yssaad, H.A.R.; Topcuoglu, B. Effect of heavy metals (copper and zinc) on proline, polyphenols and flavonoids content of tomato (*Lycopersicon esculentum* mill.). *Plant Arch.* **2020**, *20*, 2125–2137.
9. Ninh, A.P.; Xing, G.; Miller, C.J.; Waite, T.D. Fenton-like copper redox chemistry revisited: Hydrogen peroxide and superoxide mediation of copper-catalyzed oxidant production. *J. Catal.* **2013**, *301*, 54–64.
10. Bernal, M.; Ramiro, M.V.; Cases, R.; Picorel, R.; Yruela, I. Excess copper effect on growth, chloroplast ultrastructure, oxygen-evolution activity and chlorophyll fluorescence in Glycine max cell suspensions. *Physiol. Plant.* **2006**, *127*, 312–325. [\[CrossRef\]](#)
11. Shams, M.; Ekinci, M.; Turan, M.; Dursun, A.; Kul, R.; Yildirim, E. Growth, nutrient uptake and enzyme activity response of Lettuce (*Lactucasativa* L.) to excess copper. *Environ. Sustain.* **2019**, *2*, 67–73. [\[CrossRef\]](#)
12. Giannakoula, A.; Therios, I.; Chatzissavvidis, C. Effect of lead and copper on photosynthetic apparatus in citrus (*Citrus aurantium* L.) plants. The role of antioxidants in oxidative damage as a response to heavy metal stress. *Plants* **2021**, *10*, 155. [\[CrossRef\]](#)
13. Sahito, S.R.; Memon, M.A.; Kazi, T.G.; Kazi, G.H. Evaluation of mineral contents in medicinal plant *Azadirachta indica* (Neem). *J. Chem. Soc. Pak.* **2003**, *25*, 139–143.
14. Biezanowska-Kopeć, R.; Piatkowska, E. Total polyphenols and antioxidant properties of selected fresh and dried herbs and spices. *Appl. Sci.* **2022**, *12*, 4876. [\[CrossRef\]](#)
15. Pateiro, M.; Gomez-Salazar, J.A.; Jaime-Patlan, M.; Sosa-Morales, M.E.; Lorenzo, J.M. Plant extracts obtained with green solvents as natural antioxidants in fresh meat products. *Antioxidants* **2021**, *10*, 181. [\[CrossRef\]](#)
16. Adamczyk-Szabela, D.; Lisowska, K.; Romanowska-Duda, Z.; Wolf, W.M. Associated effects of cadmium and copper alter the heavy metals uptake by *Melissa officinalis*. *Molecules* **2019**, *24*, 2458. [\[CrossRef\]](#)
17. Adamczyk-Szabela, D.; Lisowska, K.; Romanowska-Duda, Z.; Wolf, W.M. Combined cadmium-zinc interactions alter manganese, lead, copper uptake by *Melissa officinalis*. *Sci. Rep.* **2020**, *10*, 1675. [\[CrossRef\]](#)

18. Mostafavi, S.; Asadi-Gharneh, H.A.; Miransari, M. The phytochemical variability of fatty acids in basil seeds (*Ocimum Basilicum* L.) affected by genotype and geographical differences. *Food Chem.* **2019**, *276*, 700–706. [CrossRef] [PubMed]
19. Shahrajabian, M.H.; Wenli, S.; Qi, C. Chemical components and pharmacological benefits of basil (*Ocimum basilicum*): A review. *Int. J. Food Prop.* **2020**, *23*, 1961–1970. [CrossRef]
20. Javanmardi, J.; Khalighi, A.; Kashi, A.; Bais, H.P.; Vivanco, J.M. Chemical characterization of basil (*Ocimum basilicum* L.) found in local accessions and used in traditional medicines in Iran. *J. Agric. Food Chem.* **2002**, *50*, 5878–5888. [CrossRef] [PubMed]
21. Mousavi, L.; Salleha, M.R.; Murugaiyah, V. Phytochemical and bioactive compounds identification of *Ocimum tenuiflorum* leaves of methanol extract and its fraction with an antidiabetic potential. *Int. J. Food Prop.* **2018**, *21*, 2390–2399. [CrossRef]
22. Ahmed, A.F.; Attia, F.A.K.; Liu, Z.; Li, C.; Wei, J.; Kang, W. Antioxidant activity and total phenolic content of essential oils and extracts of sweet basil (*Ocimum basilicum* L.) plants. *Food Sci. Hum. Wellness* **2019**, *8*, 299–305. [CrossRef]
23. Bulgari, R.; Morgutti, S.; Cocetta, G.; Negrini, N.; Farris, S.; Calcante, A.; Spinardi, A.; Ferrari, E.; Mignani, I.; Oberti, R.; et al. Evaluation of borage extracts as potential biostimulant using a phenomic, agronomic, physiological, and biochemical approach. *Front. Plant Sci.* **2017**, *8*, 935. [CrossRef]
24. Montaner, C.; Zufiaurre, R.; Movila, M.; Mallor, C. Evaluation of borage (*Borago officinalis* L.) genotypes for nutraceutical value based on leaves fatty acids composition. *Foods* **2022**, *11*, 16. [CrossRef]
25. Gupta, M.; Singh, S. *Borago officinalis* L.: An important medicinal plant of Mediterranean region: A review. *Int. J. Pharm. Sci. Rev. Res.* **2010**, *5*, 27–34.
26. Sheikhzadeh, P.; Zare, N.; Mahmoudi, F. The synergistic effects of hydro and hormone priming on seed germination, antioxidant activity and cadmium tolerance in borage. *Acta Bot. Croat.* **2021**, *80*, 18–28. [CrossRef]
27. Genc, Z.; Yarat, A.; Tunali-Akbay, T.; Sener, G. The effect of stinging nettle (*Urtica dioica*) seed oil on experimental colitis in rats. *J. Med. Food* **2011**, *14*, 1554–1561. [CrossRef]
28. Repajic, M.; Cegledi, E.; Zoric, Z.; Pedisic, S.; Elez Garofulic, I.; Radman, S.; Palcic, I.; Dragovic-Uzelac, V. Bioactive compounds in Wild Nettle (*Urtica dioica* L.) leaves and stalks: Polyphenols and pigments upon seasonal and habitat variations. *Foods* **2021**, *10*, 190. [CrossRef]
29. Beigi, M.; Toriki-Harchegani, M.; Pirbalouti, A.G. Quantity and chemical composition of essential oil of Peppermint (*Mentha piperita* L.) leaves under different drying methods. *Int. J. Food Prop.* **2018**, *1*, 267–276. [CrossRef]
30. Bensabah, F.; Houbairi, S.; Essahli, M.; Lamiri, A.; Naja, J. Chemical composition and inhibitory effect of the essential oil from mentha spicata irrigated by wastewater on the corrosion of aluminum in 1 molar hydrochloric acid. *Port. Elect. Chim. Acta* **2013**, *31*, 195–206. [CrossRef]
31. Zeljkovic, S.C.; Šišková, J.; Komzáková, K.; De Diego, N.; Kaffková, K.; Tarkowski, P. Phenolic compounds and biological activity of selected Mentha Specie. *Plants* **2021**, *10*, 550. [CrossRef]
32. Adamczyk-Szabela, D.; Chrzescijanska, E.; Zielenkiewicz, P.; Wolf, W. Antioxidant activity and photosynthesis efficiency in Melissa officinalis subjected to heavy metals stress. *Molecules* **2023**, *28*, 2642. [CrossRef]
33. He, Z.; Shentu, J.; Yang, X.; Baligar, V.C.; Zhang, T.; Stoffella, P.J. Heavy metal contamination of soils: Sources, indicators, and assessment. *J. Environ. Ind.* **2015**, *9*, 17–18.
34. Regulation of the Minister of Environment. Item 1395, 1 August 2016. Available online: <https://isap.sejm.gov.pl/isap.nsf/DocDetails.xsp?id=wdu20160001395> (accessed on 19 March 2024).
35. Page, V.; Feller, U. Heavy metals in crop plants: Transport and redistribution processes on the whole plant level. *Agronomy* **2015**, *5*, 447–463. [CrossRef]
36. Skiba, E.; Adamczyk-Szabela, D.; Wolf, W.M. *Metal-Based Nanoparticles' Interactions with Plants, Nanotechnology in the Life Sciences*; Springer Nature: Cham, Switzerland, 2021; pp. 145–169.
37. Behtash, F.; Abedini, F.; Ahmadi, H.; Mosavi, S.B.; Aghaee, A.; Morshedloo, M.R.; Lorenzo, J.M. Zinc application mitigates copper toxicity by regulating Cu uptake, activity of antioxidant enzymes, and improving physiological characteristics in summer squash. *Antioxidants* **2022**, *11*, 1688. [CrossRef]
38. Le, T.T.Y.; Vijver, M.G.; Kinraide, T.B.; Peijnenburg, W.J.G.M.; Hendriks, A.J. Modelling metal–metal interactions and metal toxicity to lettuce *Lactuca sativa* following mixture exposure ( $\text{Cu}^{2+}$ – $\text{Zn}^{2+}$  and  $\text{Cu}^{2+}$ – $\text{Ag}^{+}$ ). *Environ. Pollut.* **2013**, *176*, 185–192. [CrossRef]
39. Andresen, E.; Peiter, E.; Küpper, H. Trace metal metabolism in plants. *J. Exp. Bot.* **2018**, *69*, 909–954. [CrossRef]
40. Walker, E.L.; Connolly, E.L. Time to pump iron: Iron-deficiency signaling mechanisms of higher plants. *Curr. Opin. Plant Biol.* **2008**, *11*, 530–535. [CrossRef]
41. Mleczek, M.; Budka, A.; Gasecka, M.; Budzyńska, S.; Drzewiecka, K.; Magdziak, Z.; Rutkowski, P.; Goliński, P.; Niedzielski, P. Copper, lead and zinc interactions during phytoextraction using *Acer platanoides* L.: A pot trial. *Environ. Sci. Pollut. Res.* **2023**, *30*, 27191–27207. [CrossRef]
42. Kabata Pendias, A.; Pendias, H. *Trace Elements in Soils and Plants*, 3rd ed.; CRC Press: Boca Raton, FL, USA; London, UK; New York, NY, USA; Washington, DC, USA, 2001.
43. Kobayashi, T.; Nozoye, T.; Naoko, K.; Nishizawa, I. Iron transport and its regulation in plants. *Free Radic. Biol. Med.* **2019**, *133*, 11–20. [CrossRef]
44. Ning, X.; Lin, M.; Huang, G.; Mao, J.; Gao, Z.; Wang, X. Research progress on iron absorption, transport, and molecular regulation strategy in plants. *Front. Plant Sci.* **2023**, *14*, 1190768. [CrossRef]

45. Pande, P.; Anwar, M.; Chand, S.; Yadav, V.K.; Patra, D. Optimal level of iron and zinc in relation to its influence on herb yield and production of essential oil in menthol mint. *Commun. Soil Sci. Plant Anal.* **2007**, *38*, 561–578. [CrossRef]
46. Krzesłowska, M.; Timmers, A.C.J.; Mleczek, M.; Niedzielski, P.; Rabęda, I.; Woźny, A.; Goliński, P. Alterations of root architecture and cell wall modifications in *tilia cordata miller* (Linden) growing on mining sludge. *Environ. Pollut.* **2019**, *248*, 247–259. [CrossRef]
47. Nasim, S.A.; Dhir, B. Heavy metals alter the potency of medicinal plants. *Rev. Environ. Contam. Toxicol.* **2010**, *203*, 139–149.
48. Dresler, S.; Rutkowska, E.; Bednarek, W.; Stanislawski, G.; Kubrak, T.; Bogucka-Kocka, A.; Wójcik, M. Selected secondary metabolites in *Echium vulgare* L. populations from nonmetalliferous and metalliferous areas. *Phytochemistry* **2017**, *133*, 4–14. [CrossRef]
49. Clemens, S. Molecular mechanisms of plant metal tolerance and homeostasis. *Planta* **2001**, *212*, 475–486. [CrossRef]
50. Guo, Z.; Gao, Y.; Yuan, X.; Yuan, M.; Huang, L.; Wang, S.; Liu, C.; Duan, C. Effects of heavy metals on stomata in plants: A Review. *Int. J. Mol. Sci.* **2023**, *24*, 9302. [CrossRef]
51. Alkhatib, R.; Mheidat, M.; Abdo, N.; Tadros, M.; Al-Eitan, L.; Al-Hadid, K. Effect of lead on the physiological, biochemical and ultrastructural properties of *Leucaena leucocephala*. *Plant Biol.* **2019**, *21*, 1132–1139. [CrossRef]
52. Khudsar, T.; Arshi, A.; Siddiqi, T.O.; Mahmooduzzafar; Iqbal, M. Zinc-Induced Changes in Growth Characters, Foliar Properties and Zn-Accumulation Capacity of Pigeon Pea at Different Stages of Plant Growth. *J. Plant Nutr.* **2008**, *31*, 281–306. [CrossRef]
53. Vasilachi, I.C.; Stoleru, V.; Gavrilesco, M. Analysis of heavy metal impacts on cereal crop growth and development in contaminated soils. *Agriculture* **2023**, *13*, 1983. [CrossRef]
54. Xiong, T.; Zhang, T.; Dumat, C.; Sobanska, S.; Dappe, V.; Shahid, M.; Xian, Y.; Li, X.; Li, S. Airborne foliar transfer of particular metals in *Lactuca sativa* L.: Translocation, phytotoxicity, and bioaccessibility. *Environ. Sci. Pollut. Res. Int.* **2019**, *26*, 20064–20078. [CrossRef]
55. Liao, J.Q.; Li, Z.L.; Wang, J.S.; Tian, D.S.; Tian, D.; Niu, S.L. Nitrogen use efficiency of terrestrial plants in China: Geographic patterns, evolution, and determinants. *Ecol. Process.* **2021**, *10*, 69. [CrossRef]
56. Zeng, X. Water and nutrient use efficiencies of *Stipa purpurea* Griseb. along a precipitation gradient of the Tibetan plateau. *Plant Soil Environ.* **2023**, *69*, 230–237. [CrossRef]
57. Pandey, A.K.; Zoric, L.; Sun, T.; Karanovic, D.; Fang, P.; Borišev, M.; Wu, X.; Lukovic, J.; Xu, P. The anatomical basis of heavy metal responses in legumes and their impact on plant–rhizosphere interactions. *Plants* **2022**, *11*, 2554. [CrossRef]
58. Rucinska-Sobkowiak, R. Water relations in plants subjected to heavy metal stresses. *Acta Physiol. Plant* **2016**, *38*, 257. [CrossRef]
59. AbdElgawad, H.; Zinta, G.; Hamed, B.A.; Selim, S.; Beemster, G.; Hozzein, W.N.; Wadaan, M.A.M.; Asard, H.; Abuelsoud, W. Maize roots and shoots show distinct profiles of oxidative stress and antioxidant defense under heavy metal toxicity. *Environ. Pollut.* **2020**, *258*, 113705. [CrossRef]
60. Chen, L.; Zhang, D.; Yang, W.; Liu, Y.; Zhang, L.; Gao, S. Sex-specific responses of *Populus deltoides* to *Glomus* in tradices colonization and Cd pollution. *Chemosphere* **2016**, *155*, 196–206. [CrossRef]
61. Afzal, S.; Abdul Manap, A.S.; Attiq, A.; Albokhadaim, I.; Kandeel, M.; Alhojaily, S.M. From imbalance to impairment: The central role of reactive oxygen species in oxidative stress-induced disorders and therapeutic exploration. *Front. Pharmacol.* **2023**, *14*, 1269581. [CrossRef]
62. Bhaduri, A.M.; Fulekar, M. Antioxidant enzyme responses of plants to heavy metal stress. *Rev. Environ. Sci. Bio/Technol.* **2012**, *11*, 55–69. [CrossRef]
63. Feng, D.; Wang, R.; Sun, X.; Liu, L.; Liu, P.; Tang, J.; Zhang, C.; Liu, H. Heavy metal stress in plants: Ways to alleviate with exogenous substances. *Sci. Total Environ.* **2023**, *897*, 165397. [CrossRef]
64. Martins, L.L.; Mourato, M.P.; Cardoso, A.I.; Pinto, A.P.; Mota, A.M.; Gonçalves, M.; Plantarum, A. Oxidative stress induced by-cadmium in *Nicotiana tabacum* L.: Effects on growth parameters, oxidative damage and antioxidant responses in different plant parts. *Acta Physiol. Plant.* **2011**, *33*, 1375–1383. [CrossRef]
65. Chaki, M.; Begara-Morales, J.C.; Barroso, J.B. Oxidative stress in plants. *Antioxidants* **2020**, *9*, 481. [CrossRef]
66. Bertelli, A.; Biagi, M.; Corsini, M.; Bains, G.; Cappellucci, G.; Miraldi, E. Polyphenols: From theory to practice. *Foods* **2021**, *10*, 2595. [CrossRef]
67. Ullah, R.; Hadi, F.; Ahmad, S.; Ullah, A.J.; Rongliang, Q. Phytoremediation of lead and chromium contaminated soil improves with the endogenous phenolics and proline production in *Parthenium*, *Cannabis*, *Euphorbia*, and *Rumex* Species. *Water Air Soil Pollut.* **2019**, *230*, 40. [CrossRef]
68. Azzazy, M.F. Plant bioindicators of pollution in Sadat City, Western Nile Delta. *Egypt PLoS ONE* **2020**, *15*, e0226315. [CrossRef]
69. PN-ISO 10381-4; Soil Quality—Sampling—Part 4: Rules for Procedure during the Research Areas of Natural, Semi-Natural and Cultivated. International Organization for Standardization: Geneva, Switzerland, 2007.
70. PN-ISO 10390; Agricultural Chemical Analysis of the Soil. Determination of pH. International Organization for Standardization: Geneva, Switzerland, 1997. Available online: <http://sklep.pkn.pl/pn-iso-10390-1997p.html> (accessed on 20 April 2019).
71. Schumacher, B.A. *Methods for the Determination of Total Organic Carbon (TOC) in Soils and Sediments*; Environmental Protection Agency: Washington, DC, USA, 2002.
72. Ndhkala, A.R.; Kasiyamhuru, A.; Mupure, C.; Chitindingu, K.; Benhura, M.A.; Muchuweti, M. Phenolic composition of *Flacourtia indica*, *Opuntia megacantha* and *Sclerocarya birrea*. *Food Chem.* **2007**, *103*, 82–87. [CrossRef]

73. Goodson, D.Z. *Mathematical Methods for Physical and Analytical Chemistry*; Wiley: Hoboken, NJ, USA, 2011.
74. Razali, N.M.; Wah, Y.B. Power comparisons of Shapiro-Wilk, Kolmogorov-Smirnov, Lilliefors and Anderson-Darling tests. *J. Stat. Model. Anal.* **2011**, *2*, 21–33.

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