

## Supporting Information

### **Preparation of Multicolour Solid Fluorescent Carbon Dots for Light-Emitting Diodes Using Phenylethylamine As a Co-Carbonization Agent**

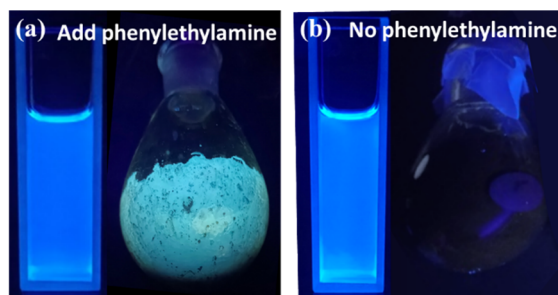
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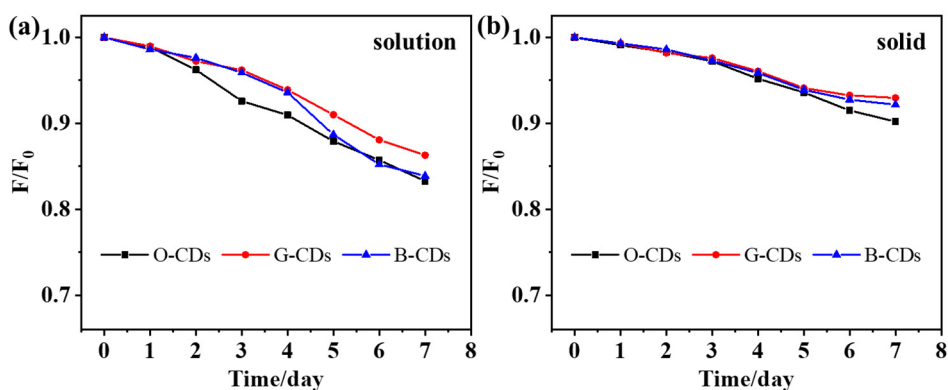
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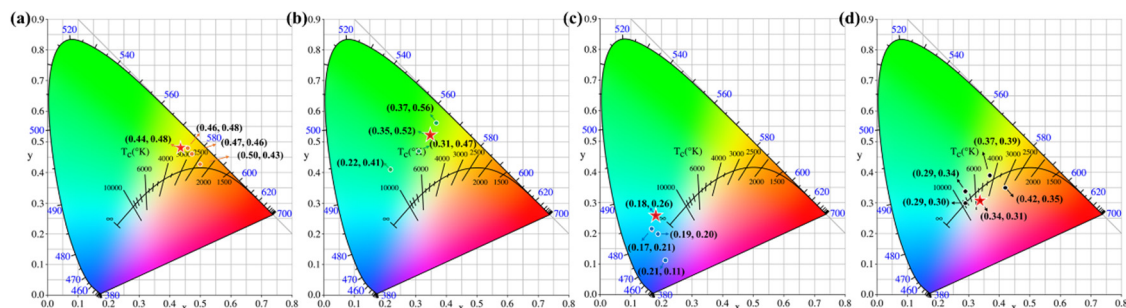
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**Figure S1.** CDs prepared in reaction systems with phenylethylamine (a) and without phenylethylamine (b).



**Figure S2.** Decay curve of PL intensity of CDs in solution (a) and solid state (b) with increasing visible light time.



**Figure S3.** CIE 1931 diagram comparing the color coordinates of (a) orange, (b) green, (c) blue and (d) white LEDs reported in the literature with the LEDs in this study. Note: The red star represents the color-coordinate position of the LED prepared in this study.

**Table S1.** Comparison of process parameters and performance of O-CDs, G-CDs and B-CDs.

Synthetic Method			PL Color		Stability		QY	
Sample	Reaction Precursor	Reaction Solution	Solution	Solid	Solution	Solid	Solution	Solid
O-CDs	Citric acid	methanol	green	orange	83%	90%	36.7%	13.9%
G-CDs	+ urea + phenylethy	ethanol	green	green	86%	93%	69.6%	35.7%
B-CDs	lamine	water	blue	blue	84%	92%	67.3%	2.6%

**Table S2.** The preparation method and fluorescence QY of solid fluorescent CDs have been reported.

Preparation Method	CDs PL Color	QY (%)	References
Assemble BaSO <sub>4</sub> on the CDs surface	green	27.0	[25]
Scattered in PVP matrix	red	41.3	[31]
One step hydrothermal method	white	2.0	[30]
One step hydrothermal method	green	26.0	[22]
One step hydrothermal method	white	35.0	[23]

**Table S3.** CIE coordinates of LEDs with different luminescence colors reported in the literature.

LED Color	CIE (X, Y)	References
Blue	0.17, 0.21	[14]
	0.21, 0.11	[16]
	0.19, 0.20	[54]
Green	0.22, 0.41	[15]
	0.31, 0.47	[54]
	0.37, 0.56	[55]
Orange	0.47, 0.46	[7]
	0.50, 0.43	[15]
	0.46, 0.48	[54]
White	0.37, 0.39	[14]
	0.29, 0.30	[16]
	0.29, 0.34	[23]
	0.34, 0.31	[24]
	0.42, 0.35	[47]