

SUPPLEMENTARY MATERIALS

Figure S1. Distribution diagrams with deprotonation routes of the (a) delphinidin; (b) delphinidin 3-*O*-glucoside; (c) peonidin; (d) peonidin 3-*O*-glucoside; (e) petunidin; (f) petunidin 3-*O*-glucoside.

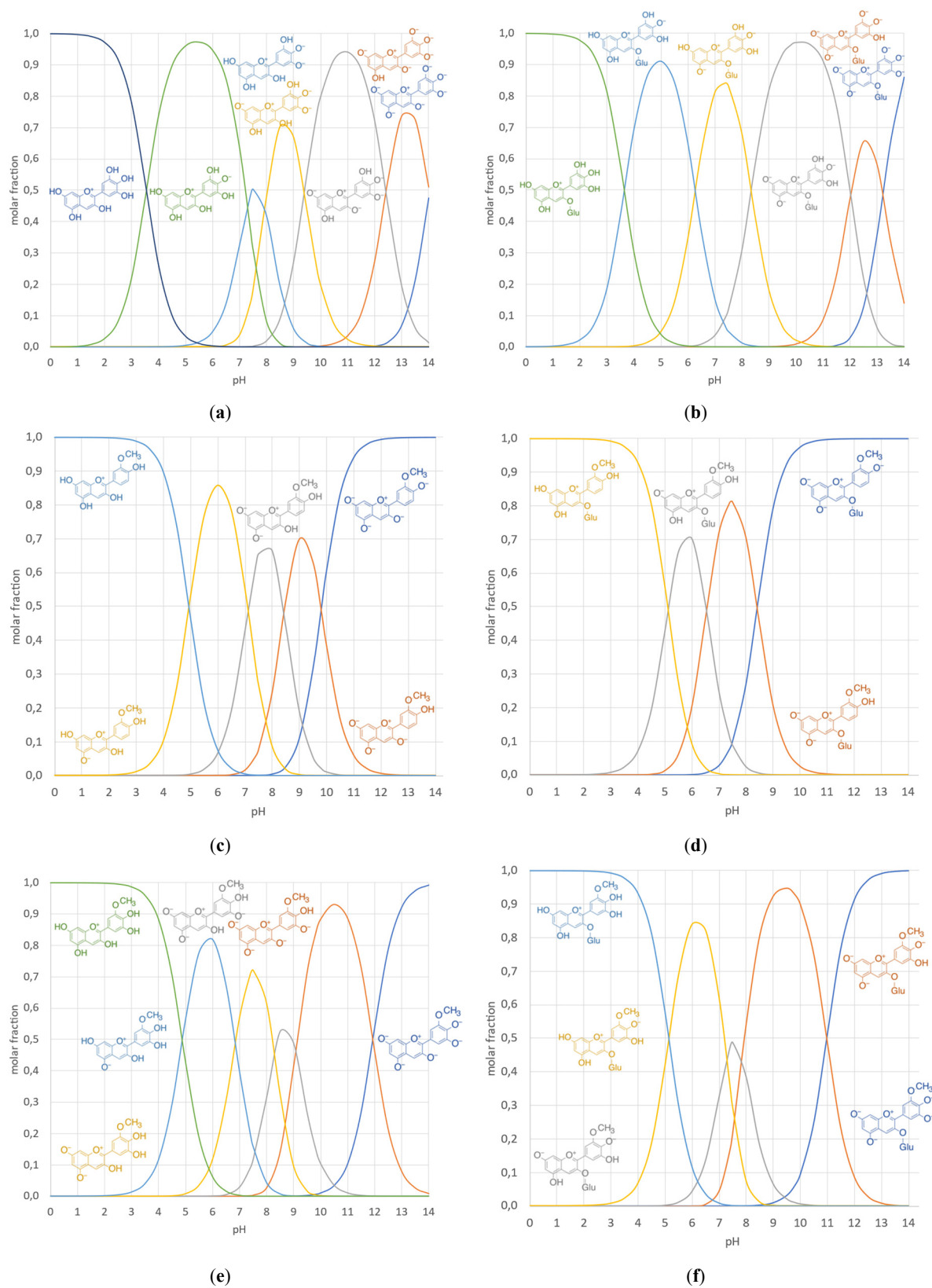
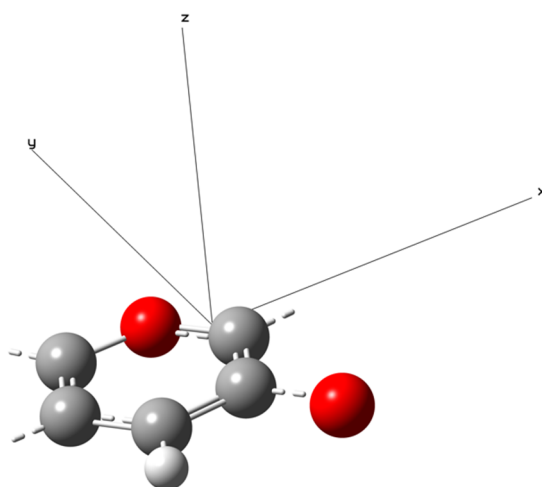


Table S1. Proton affinities (kcal mol⁻¹) of anthocyanidins and anthocyanins investigated along with the reaction site. The lowest values are underlined.

		1+	0	1–	2–	3–	4–
Dp	C3	25.8	29.1				
	C3'	26.4	37.2	39.8	45.4	45.2	
	C4'	17.3					
	C5	22.6	39.1	32.4	35.7		
	C5'	32.1	29.7	41.4	46.4	45.6	50.3
	C7	22.4	29.3	31.2			
Dp 3-glc	C3	<i>-Glu</i>	<i>-Glu</i>	<i>-Glu</i>	<i>-Glu</i>	<i>-Glu</i>	<i>-Glu</i>
	C3'	33.5	37.3	40.1	44.0		
	C4'	17.6					
	C5	21.4	25.9				
	C5'	29.2	38.0	41.4	44.6	47.7	
	C7	25.8	29.0	32.4			
Pn	C3	23.7	29.0	32.7			
	C4'	26.4	30.7	33.8	37.0		
	C5	21.7					
	C7	23.4	28.5				
Pn 3-glc	C3	<i>-Glu</i>	<i>-Glu</i>	<i>-Glu</i>	<i>-Glu</i>		
	C4'	22.8	30.1	32.7			
	C5	23.9	26.7				
	C7	22.3					
Pt	C3	23.2	28.7	33.3	34.7		
	C3'	28.8	30.3	32.3			
	C4'	24.2	28.3	32.6	41.3	43.7	
	C5	21.5					
	C7	22.7	27.7				
Pt-3-glc	C3	<i>-Glu</i>	<i>-Glu</i>	<i>-Glu</i>	<i>-Glu</i>	<i>-Glu</i>	
	C3'	29.8	35.6	37.9	40.7		
	C4'	22.4					
	C5	23.7	29.1	30.9			
	C7	23.5	29.1				

Table S2. Dipole momenta (μ , in D) and their corresponding vector in 3D-dimensions, obtained for the aglycones and their glycosides in pentyl ethanoate (cation form) and water (neutral form).



	pentyl ethanoate		water	
	+		0	
	μ [D]	vector (x,y,z)	μ [D]	(x,y,z)
Dp	10.2	(-10.1, 0.3, 1.1)	22.5 (6)	(-21.9, -5.3, 0)
Dp 3-glc	6.8	(-0.0, 5.4, -4.1)	14.1 (2)	(-11.7, 6.3, -4.9)
Pn	2.6	(1.7, 1.7, 1.1)	15.5 (4)	(-13.4, 7.7, 0.3)
Pn-3-glc	12.0	(-11.6, -2.6, 1.9)	11.2 (1)	(-9.8, -4.8, -2.6)
Pt	4.7	(-3.4, 3.2, -0.5)	15.3 (3)	(11.3, 10.3, 0.8)
Pt 3-glc	10.6	(7.6, 7.2, -2.0)	19.5 (5)	(-7.5, 17.9, -1.9)

Table S3. The reactivity of the studied anthocyanidins and their glycosides in SET and HAT mechanisms towards common radicals in water.

Delphinidin

Neutral

R-OH + OH* -> R-OH+* + OH-	[]	R-OH + OOH* -> R-OH+* + OOH-	[kcal/mol]
adiabatic	-9,2	adiabatic	14,2
vertical	-5,1	vertical	27,5
activation	1,6	activation	14,2
rate constants	8.37E+9	rate constants	2.36E+02
R-OH + OH* -> R-O* + H2O		R-OH + OOH* -> R-O* + H2O2	
C3	-45,4	C3	-7,7
C3'	-39,8	C3'	-2,2
C5	-41,1	C5	-3,4
C5'	-40,2	C5'	-2,5
C7	-41,5	C7	-3,9
R-OH + CH3O* -> R-OH+* + CH3O-	[kcal/mol]		
adiabatic	10,8		
vertical	17,7		
activation	11,4		
rate constants	2,96E+04		
R-OH + CH3O* -> R-O* + CH3OH			
C3	-27,2		
C3'	-21,6		
C5	-22,9		
C5'	-21,9		
C7	-23,3		
R-OH + SH* -> R-OH+* + SH-	[kcal/mol]		
adiabatic	6,4		
vertical	11,2		
activation	6,5		
rate constants	1,01E+08		
R-OH + SH* -> R-O* + H2S			
C3	-31,0		
C3'	-25,4		
C5	-26,7		
C5'	-25,7		
C7	-27,1		

Anion

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol]	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol]
<i>adiabatic</i> -23,1	<i>adiabatic</i> 0,2
<i>vertical</i> -17,1	<i>vertical</i> 15,6
<i>activation</i> 12,2	<i>activation</i> 4,0
<i>rate constants</i> 7,27E+03	<i>rate constants</i> 3.99E+09
$R-OH + OH^* \rightarrow R-O^* + H_2O$	$R-OH + OOH^* \rightarrow R-O^* + H_2O_2$
C3' -43,9	C3' -6,3
C5 -51,4	C5 -13,7
C5' -45,4	C5' -7,8
C7 -51,9	C7 -14,2
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol]	
<i>adiabatic</i> -3,1	
<i>vertical</i> 5,7	
<i>activation</i> 0,9	
<i>rate constants</i> 7.85E+9	
$R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$	
C3' -25,7	
C5 -33,2	
C5' -27,2	
C7 -33,7	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol]	
<i>adiabatic</i> -7,5	
<i>vertical</i> -0,7	
<i>activation</i> 0,0	
<i>rate constants</i> 8.52E+9	
$R-OH + SH^* \rightarrow R-O^* + H_2S$	
C3' -29,5	
C5 -37,0	
C5' -31,0	
C7 -37,5	

Dianion

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -31,6 <i>vertical</i> -25,0 <i>activation</i> 23,7 <i>rate constants</i> 2,74E-05 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C3' -49,2 C5 -54,0 C5' -48,7	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> -8,3 <i>vertical</i> 7,7 <i>activation</i> 0,9 <i>rate constants</i> 8.03E+9 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C3' -11,5 C5 -16,4 C5' -11,0
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> -11,7 <i>vertical</i> -2,2 <i>activation</i> 0,1 <i>rate constants</i> 7.92E+9 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C3' -31,0 C5 -35,8 C5' -30,5	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> -16,1 <i>vertical</i> -8,6 <i>activation</i> 2,5 <i>rate constants</i> 7.84E+9 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C3' -34,8 C5 -39,6 C5' -34,3	

Trianion

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol]		$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol]	
<i>adiabatic</i>	-38,3	<i>adiabatic</i>	-15,0
<i>vertical</i>	-31,2	<i>vertical</i>	1,5
<i>activation</i>	34,8	<i>activation</i>	0,0
<i>rate constants</i>	2,03E-13	<i>rate constants</i>	8.07E+9
$R-OH + OH^* \rightarrow R-O^* + H_2O$		$R-OH + OOH^* \rightarrow R-O^* + H_2O_2$	
C3'	-53,3	C3'	-15,7
C5'	-52,6	C5'	-15,0
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol]			
<i>adiabatic</i>	-18,4		
<i>vertical</i>	-8,4		
<i>activation</i>	-0,7		
<i>rate constants</i>	7.74E+9		
$R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$			
C3'	-35,1		
C5'	-34,4		
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol]			
<i>adiabatic</i>	-22,8		
<i>vertical</i>	-14,8		
<i>activation</i>	6,9		
<i>rate constants</i>	5,96E+07		
$R-OH + SH^* \rightarrow R-O^* + H_2S$			
C3'	-38,9		
C5'	-38,2		

Delphinidin 3-O-glucoside

neutral

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -9,0 <i>vertical</i> -4,9 <i>activation</i> 1,5 <i>rate constants</i> 8.93+09 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C3' -41,1 C5 -42,0 C5' -41,9 C7 -40,7	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> 14,3 <i>vertical</i> 27,8 <i>activation</i> 14,3 <i>rate constants</i> 2,00E+02 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C3' -3,4 C5 -4,3 C5' -4,2 C7 -3,0
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> 10,9 <i>vertical</i> 17,9 <i>activation</i> 11,4 <i>rate constants</i> 2,54E+04 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C3' -22,8 C5 -23,8 C5' -23,7 C7 -22,4	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> 6,6 <i>vertical</i> 11,5 <i>activation</i> 6,8 <i>rate constants</i> 6.97+7 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C3' -26,6 C5 -27,6 C5' -27,5 C7 -26,2	

Anion

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -16,5 <i>vertical</i> -11,3 <i>activation</i> 6,1 <i>rate constants</i> 1,92E+08 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C3' -44,9 C5' -44,5 C7 -41,0	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> 6,8 <i>vertical</i> 21,4 <i>activation</i> 7,8 <i>rate constants</i> 1,11E+07 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C3' -7,3 C5' -6,8 C7 -3,3
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> 3,5 <i>vertical</i> 11,5 <i>activation</i> 4,1 <i>rate constants</i> 3.42E+09 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C3' -26,7 C5' -26,3 C7 -22,8	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> -0,9 <i>vertical</i> 5,1 <i>activation</i> 1,08 <i>rate constants</i> 9,01E+9 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C3' -30,5 C5' -30,1 C7 -26,6	

Dianion

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -22,1 <i>vertical</i> -16,4 <i>activation</i> 11,8 <i>rate constants</i> 1,40E+04 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C3' -47,2 C5' -47,0	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> 1,3 <i>vertical</i> 16,2 <i>activation</i> 4,4 <i>rate constants</i> 2.75E+09 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C3' -9,5 C5' -9,3
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> -2,1 <i>vertical</i> 6,4 <i>activation</i> 1,20 <i>rate constants</i> 8,22E+9 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C3' -28,9 C5' -28,7	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> -6,5 <i>vertical</i> -0,1 <i>activation</i> 0,0 <i>rate constants</i> 9.12E+9 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C3' -32,8 C5' -32,5	

Peonidin

Cation

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> 13,0 <i>vertical</i> 17,5 <i>activation</i> 17,0 <i>rate constants</i> 2,09E+00 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C3 -37,2 C4' -35,0 C5 -33,3 C7 -28,5	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> 36,4 <i>vertical</i> 50,2 <i>activation</i> 45,7 <i>rate constants</i> 2,12E-21 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C3 0,5 C4' 2,7 C5 4,4 C7 9,1
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> 33,0 <i>vertical</i> 40,3 <i>activation</i> 55,6 <i>rate constants</i> 1,05E-28 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C3 -19,0 C4' -16,8 C5 -15,0 C7 -10,3	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> 28,6 <i>vertical</i> 33,9 <i>activation</i> 54,21 <i>rate constants</i> 1,13E-27 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C3 -22,8 C4' -20,6 C5 -18,8 C7 -14,1	

Neutral

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -3,6 <i>vertical</i> -0,7 <i>activation</i> 0,0 <i>rate constants</i> 8,48E+09 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C3 -44,2 C4' -37,2 C7 -32,5	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> 19,8 <i>vertical</i> 32,0 <i>activation</i> 21,0 <i>rate constants</i> 2,57E-03 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C3 -6,6 C4' 0,4 C7 5,1
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> 16,4 <i>vertical</i> 22,1 <i>activation</i> 21,4 <i>rate constants</i> 1,23E-03 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C3 -26,0 C4' -19,0 C7 -14,3	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> 12,0 <i>vertical</i> 15,7 <i>activation</i> 16,65 <i>rate constants</i> 3,84E+00 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C3 -29,8 C4' -22,8 C7 -18,1	

Anion

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -9,6 <i>vertical</i> -7,0 <i>activation</i> 4,7100 <i>rate constants</i> 1,74E+09 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C3 -48,7 C4' -39,8	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> 13,7 <i>vertical</i> 25,6 <i>activation</i> 13,7700 <i>rate constants</i> 5,01E+02 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C3 -11,1 C4' -2,1
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> 10,3 <i>vertical</i> 15,8 <i>activation</i> 11,3500 <i>rate constants</i> 2,98E+04 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C3 -30,5 C4' -21,5	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> 5,9 <i>vertical</i> 9,3 <i>activation</i> 6,4 <i>rate constants</i> 1,35E+08 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C3 -34,3 C4' -25,3	

Dianion

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -30,0 <i>vertical</i> -26,2 <i>activation</i> 45,16 <i>rate constants</i> 4,86E-21 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C4' -54,3	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> -6,7 <i>vertical</i> 6,5 <i>activation</i> 0,80 <i>rate constants</i> 8.04E+09 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C4' -16,6
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> -10,0 <i>vertical</i> -3,4 <i>activation</i> 0,44 <i>rate constants</i> 7.91E+09 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C4' -36,1	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> -14,4 <i>vertical</i> -9,8 <i>activation</i> 5,22 <i>rate constants</i> 8.35E+08 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C4' -39,9	

Peonidin 3-O-glucoside

Neutral

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -1,9 <i>vertical</i> 2,9 <i>activation</i> 0,4 <i>rate constants</i> 9.09E+09 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C4' -37,6 C5 -31,5	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> 21,4 <i>vertical</i> 35,6 <i>activation</i> 22,3 <i>rate constants</i> 2,73E-04 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C4' 0,1 C5 6,2
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> 18,1 <i>vertical</i> 25,7 <i>activation</i> 21,7 <i>rate constants</i> 7,34E-04 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C4' -19,3 C5 -13,3	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> 13,7 <i>vertical</i> 19,3 <i>activation</i> 16,63 <i>rate constants</i> 4,01E+00 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C4' -23,2 C5 -17,1	

Anion

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -9,8 <i>vertical</i> -6,1 <i>activation</i> 2,5100 <i>rate constants</i> 8,25E+9 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C4' -39,6	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> 13,5 <i>vertical</i> 26,6 <i>activation</i> 13,5000 <i>rate constants</i> 7,84E+02 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C4' -1,9
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> 10,2 <i>vertical</i> 16,7 <i>activation</i> 10,7300 <i>rate constants</i> 8,51E+04 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C4' -21,4	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> 5,8 <i>vertical</i> 10,3 <i>activation</i> 5,89 <i>rate constants</i> 2,87E+08 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C4' -25,2	

Dianion

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol]	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol]
<i>adiabatic</i> -20,9	<i>adiabatic</i> 2,4
<i>vertical</i> -16,0	<i>vertical</i> 16,7
<i>activation</i> 13,0600	<i>activation</i> 4,8800
<i>rate constants</i> 1,65E+03	<i>rate constants</i> 1,39E+09
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol]	
<i>adiabatic</i> -0,9	
<i>vertical</i> 6,8	
<i>activation</i> 1,5000	
<i>rate constants</i> 8.21E+9	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol]	
<i>adiabatic</i> -5,3	
<i>vertical</i> 0,4	
<i>activation</i> 0,0100	
<i>rate constants</i> 9.08E+09	

Petunidin
Neutral

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -2,7 <i>vertical</i> -0,1 <i>activation</i> 0,0000 <i>rate constants</i> 8.45E+09 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C3 -44,0 C3' -36,4 C4' -38,7 C7 -32,6	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> 20,6 <i>vertical</i> 32,5 <i>activation</i> 22,1900 <i>rate constants</i> 3,36E-04 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C3 -6,3 C3' 1,2 C4' -1,0 C7 5,1
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> 17,3 <i>vertical</i> 22,6 <i>activation</i> 24,0900 <i>rate constants</i> 1,35E-05 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C3 -25,7 C3' -18,2 C4' -20,4 C7 -14,4	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> 12,9 <i>vertical</i> 16,2 <i>activation</i> 19,88 <i>rate constants</i> 1,65E-02 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C3 -29,6 C3' -22,0 C4' -24,2 C7 -18,2	

Anion

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -8,9 <i>vertical</i> -7,1 <i>activation</i> 7,0000 <i>rate constants</i> 4,58E+07 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C3 -47,9 C3' -37,5 C4' -40,6	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> 14,4 <i>vertical</i> 25,6 <i>activation</i> 14,6300 <i>rate constants</i> 1,17E+02 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C3 -10,3 C3' 0,2 C4' -2,9
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> 11,1 <i>vertical</i> 15,7 <i>activation</i> 13,4000 <i>rate constants</i> 9,39E+02 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C3 -29,7 C3' -19,2 C4' -22,4	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> 6,7 <i>vertical</i> 9,3 <i>activation</i> 8,32 <i>rate constants</i> 4,97E+06 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C3 -33,5 C3' -23,0 C4' -26,2	

Dianion

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -18,3 <i>vertical</i> -13,4 <i>activation</i> 9,1600 <i>rate constants</i> 1,19E+06 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C3 -49,3 C4' -49,2	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> 5,0 <i>vertical</i> 19,2 <i>activation</i> 6,4900 <i>rate constants</i> 1,09E+08 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C3 -11,6 C4' -11,6
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> 1,7 <i>vertical</i> 9,4 <i>activation</i> 2,8700 <i>rate constants</i> 4,90E+10 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C3 -31,1 C4' -31,0	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> -2,7 <i>vertical</i> 2,9 <i>activation</i> 0,38 <i>rate constants</i> 8.44E+09 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C3 -34,9 C4' -34,8	

Trianion

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -32,6 <i>vertical</i> -27,8 <i>activation</i> 40,2500 <i>rate constants</i> 1,93E-17 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C4' -54,1	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> -9,3 <i>vertical</i> 4,9 <i>activation</i> 0,4200 <i>rate constants</i> 8.04E+09 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C4' -16,4
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> -12,6 <i>vertical</i> -5,0 <i>activation</i> 0,8200 <i>rate constants</i> 7.81E+09 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C4' -35,8	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> -17,0 <i>vertical</i> -11,4 <i>activation</i> 5,8 <i>rate constants</i> 3,47E+08 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C4' -39,7	

Petunidin 3-O-glucoside

Cation

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> 10,8 <i>vertical</i> 15,7 <i>activation</i> 12,5800 <i>rate constants</i> 3,75E+03 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C3' -34,9 C4' -39,4 C5 -31,6 C7 -29,7	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> 34,1 <i>vertical</i> 48,3 <i>activation</i> 41,1900 <i>rate constants</i> 3,95E-18 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C3' 2,8 C4' -1,7 C5 6,0 C7 7,9
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> 30,8 <i>vertical</i> 38,5 <i>activation</i> 48,1300 <i>rate constants</i> 3,26E-23 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C3' -16,6 C4' -21,1 C5 -13,4 C7 -11,5	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> 26,4 <i>vertical</i> 32,0 <i>activation</i> 45,7100 <i>rate constants</i> 1,91E-21 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C3' -20,4 C4' -25,0 C5 -17,2 C7 -15,3	

Neutral

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -10,7 <i>vertical</i> -5,8 <i>activation</i> 1,7200 <i>rate constants</i> 8.85E+09 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C3' -43,4 C5 -40,6 C7 -42,4	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> 12,6 <i>vertical</i> 26,8 <i>activation</i> 12,6500 <i>rate constants</i> 3,34E+03 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C3' -5,7 C5 -2,9 C7 -4,7
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> 9,2 <i>vertical</i> 16,9 <i>activation</i> 9,2700 <i>rate constants</i> 9,89E+05 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C3' -25,1 C5 -22,3 C7 -24,2	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> 4,8 <i>vertical</i> 10,5 <i>activation</i> 4,84 <i>rate constants</i> 1,48E+09 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C3' -28,9 C5 -26,1 C7 -28,0	

MonoAnion

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -20,2 <i>vertical</i> -14,8 <i>activation</i> 10,1400 <i>rate constants</i> 2,29E+05 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C3' -48,3 C5 -44,2	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> 3,1 <i>vertical</i> 17,9 <i>activation</i> 5,4100 <i>rate constants</i> 6,23E+08 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C3' -10,6 C5 -6,6
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> -0,2 <i>vertical</i> 8,0 <i>activation</i> 1,9500 <i>rate constants</i> 7.97E+9 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C3' -30,1 C5 -26,0	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> -4,6 <i>vertical</i> 1,6 <i>activation</i> 0,1 <i>rate constants</i> 9.07E+09 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C3' -33,9 C5 -29,8	

Dianion

$R-OH + OH^* \rightarrow R-OH^{+*} + OH^-$ [kcal/mol] <i>adiabatic</i> -23,8 <i>vertical</i> -17,2 <i>activation</i> 11,2100 <i>rate constants</i> 3,79E+04 $R-OH + OH^* \rightarrow R-O^* + H_2O$ C3' -50,9	$R-OH + OOH^* \rightarrow R-OH^{+*} + OOH^-$ [kcal/mol] <i>adiabatic</i> -0,5 <i>vertical</i> 15,5 <i>activation</i> 3,7500 <i>rate constants</i> 4.81E+09 $R-OH + OOH^* \rightarrow R-O^* + H_2O_2$ C3' -13,2
$R-OH + CH_3O^* \rightarrow R-OH^{+*} + CH_3O^-$ [kcal/mol] <i>adiabatic</i> -3,8 <i>vertical</i> 5,6 <i>activation</i> 0,8300 <i>rate constants</i> 8.21E+09 $R-OH + CH_3O^* \rightarrow R-O^* + CH_3OH$ C3' -32,7	
$R-OH + SH^* \rightarrow R-OH^{+*} + SH^-$ [kcal/mol] <i>adiabatic</i> -8,2 <i>vertical</i> -0,8 <i>activation</i> 0,02 <i>rate constants</i> 9.07E+09 $R-OH + SH^* \rightarrow R-O^* + H_2S$ C3' -36,5	