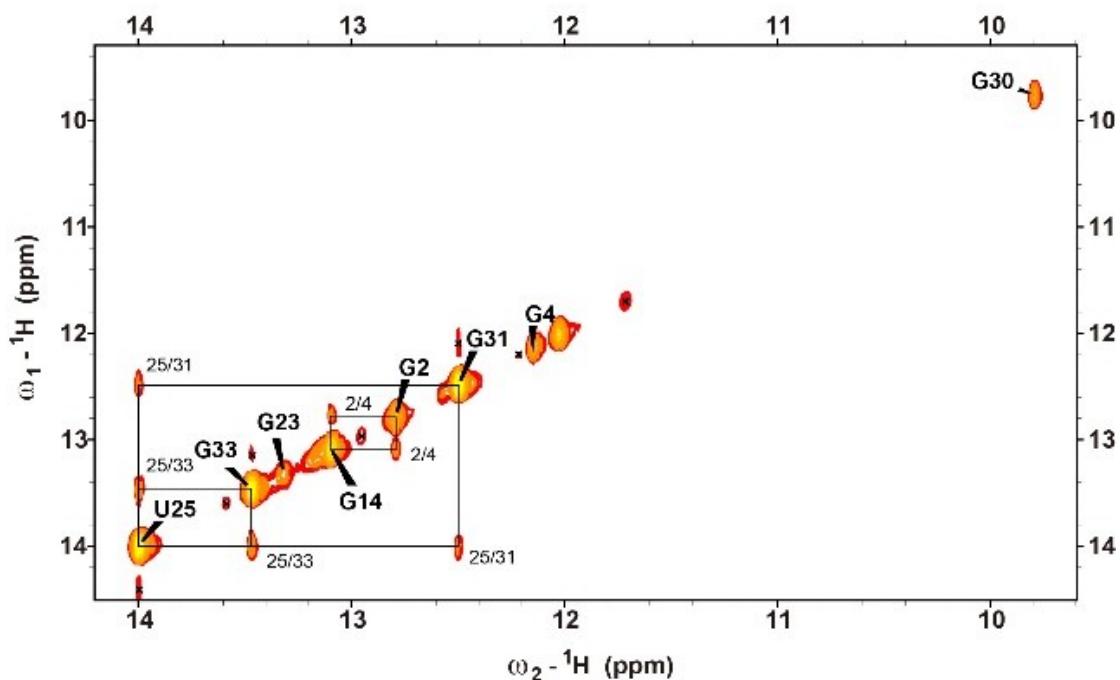
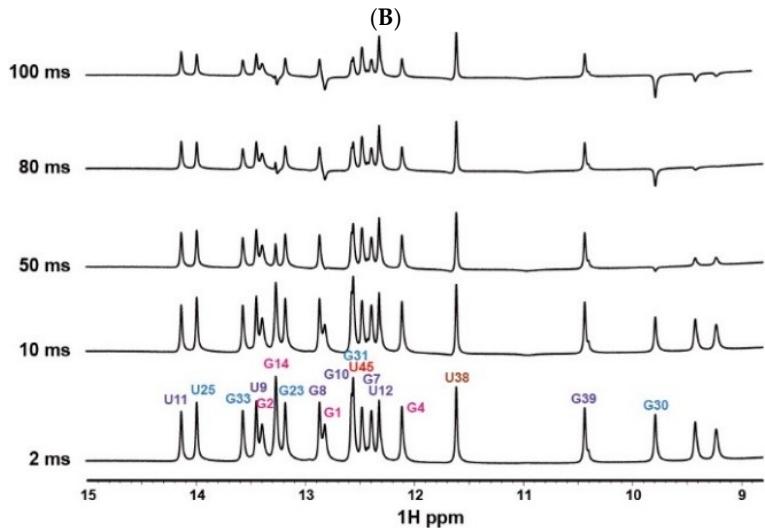
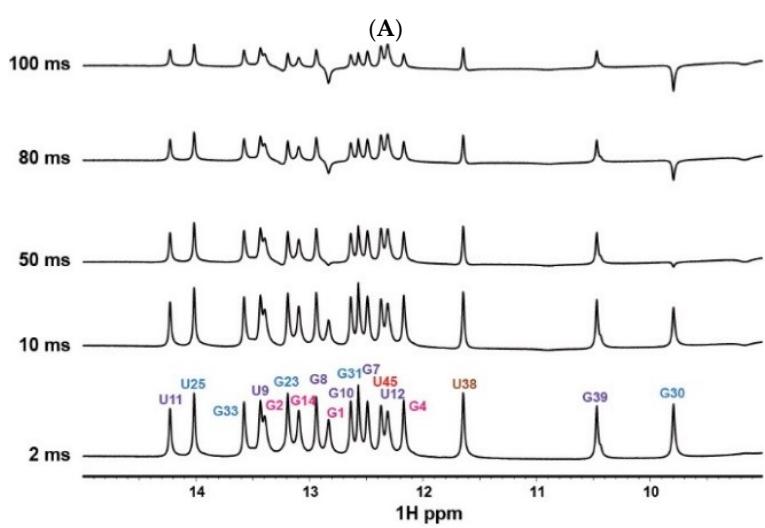
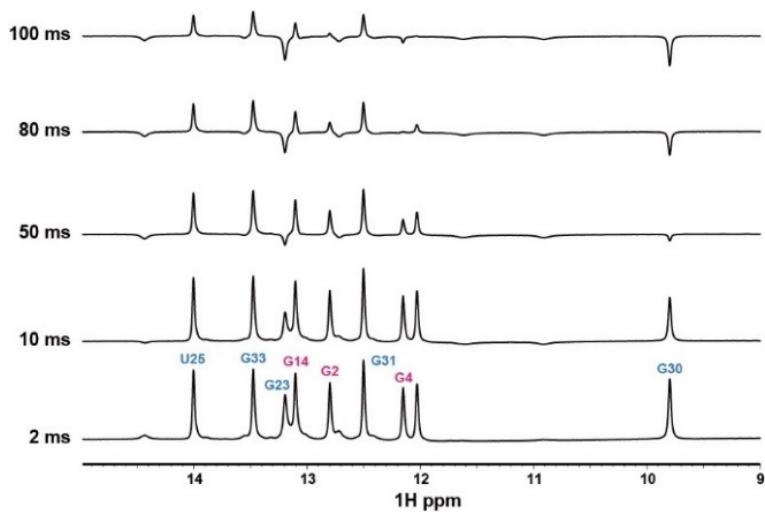


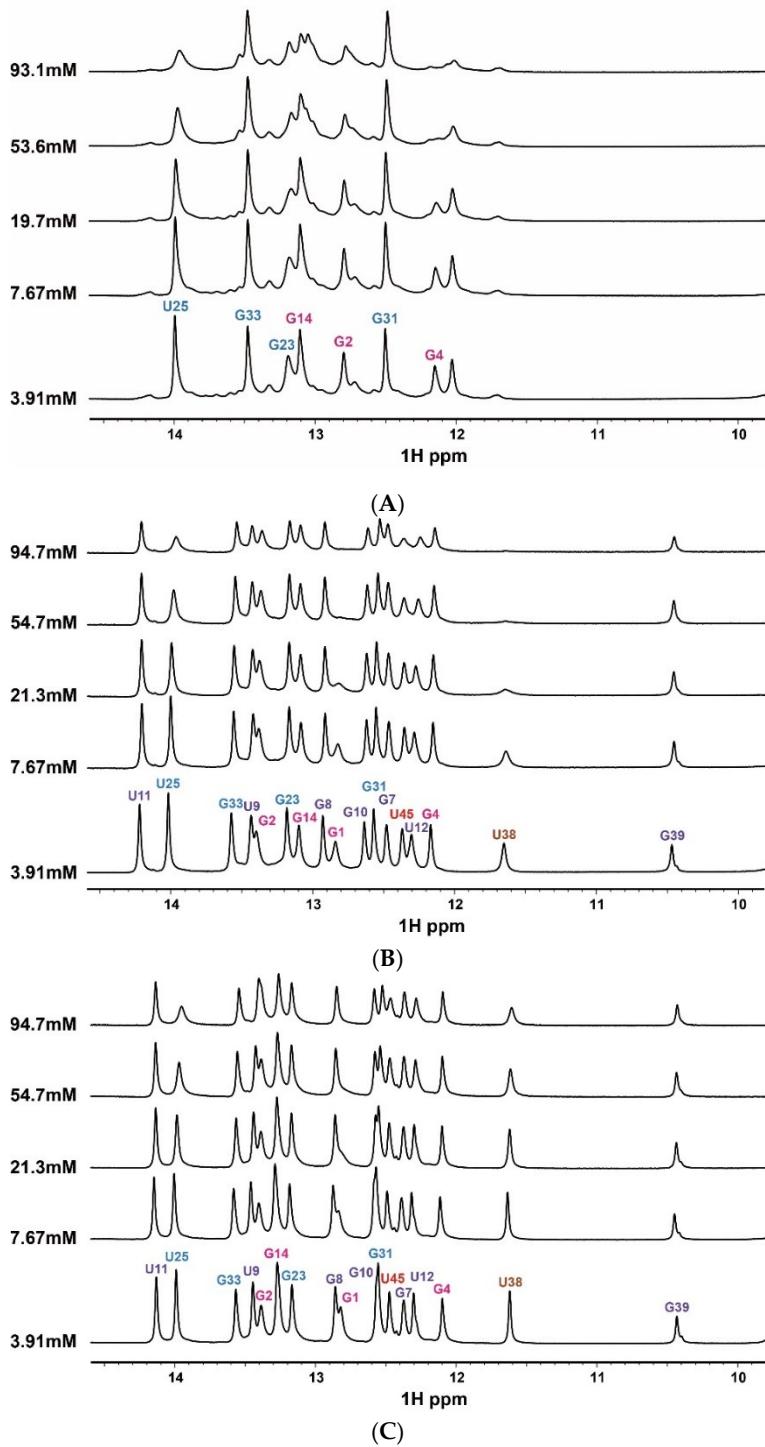
## Supplementary Materials



**Figure S1.** Imino proton resonance assignments of fluoride riboswitch in the free state by Watergate NOESY spectra at 25 °C.



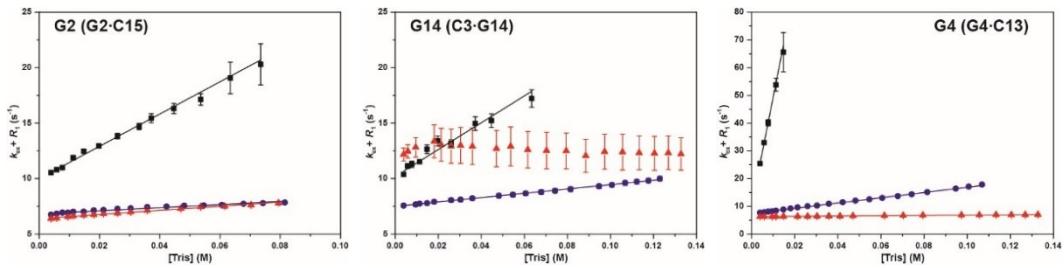
**Figure S2.** 1D  $^1\text{H}$  spectra of water magnetization transfer experiments showing imino protons of the fluoride riboswitch in the (A) free, (B) apo and (C) holo states. The control spectrum with no selective water inversion is indicated at the bottom. The delay times between the selective water inversion and acquisition pulses are shown on the left side of the spectra.



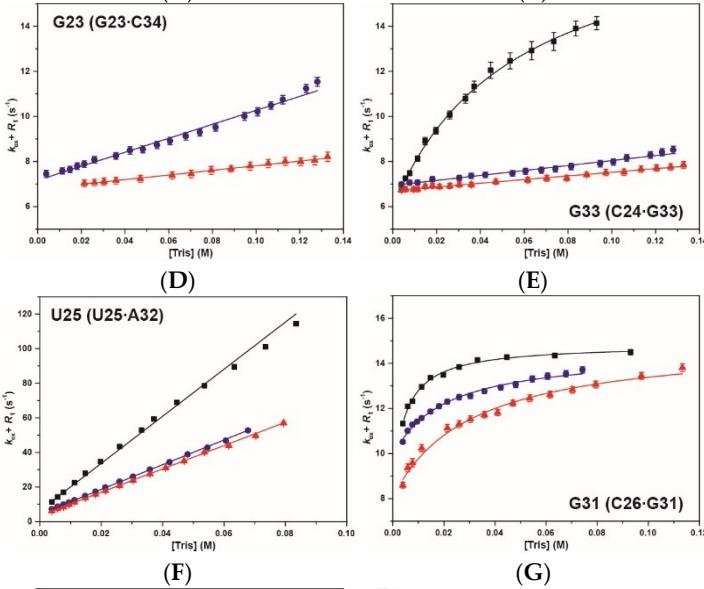
**Figure S3.** 1D spectra of *CrcB* aptamer measured with increasing Tris concentration in a 90% H<sub>2</sub>O/10% D<sub>2</sub>O solution containing 10mM Tris (pH 8.0 at 25 °C), 50mM KCl, 50uM EDTA (pH 8.0). The Tris concentrations are shown to the left of each spectrum. (A) free, (B) apo(add 2mM MgCl<sub>2</sub>) and (C) holo(add 2mM MgCl<sub>2</sub> and 10mM NaF).

P1

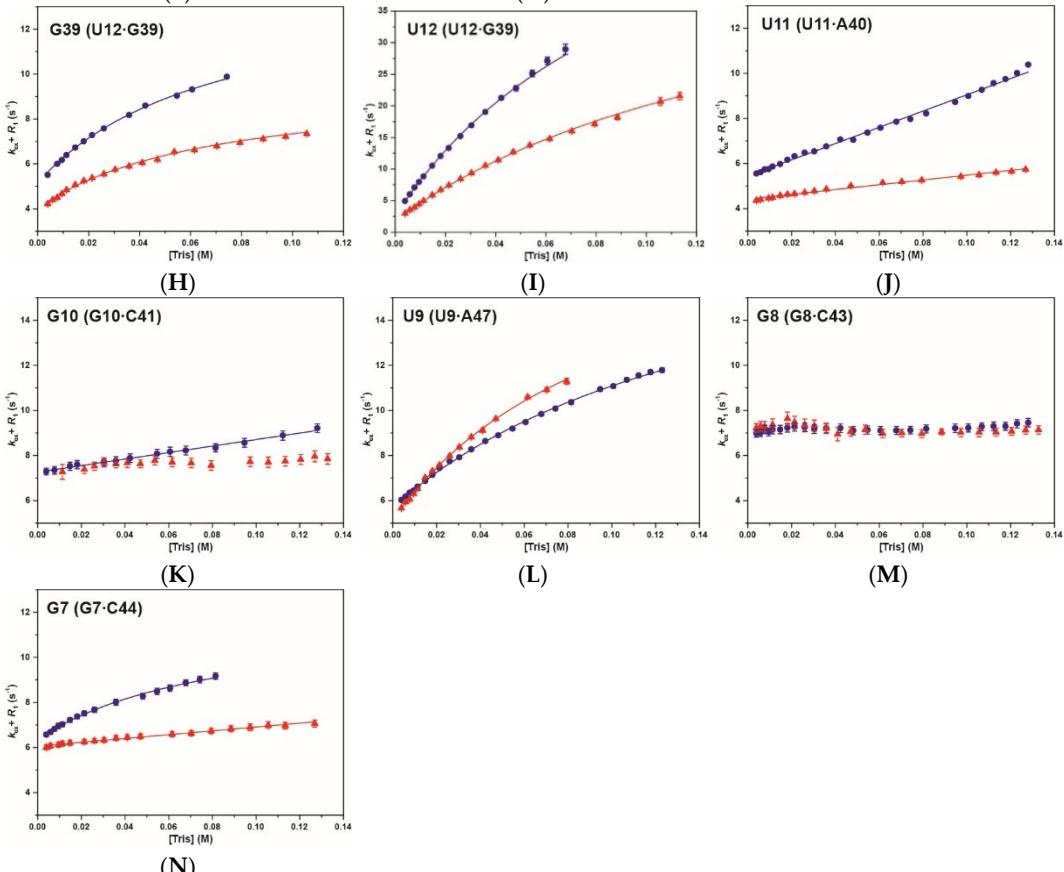
■ Free ● Apo ▲ Holo

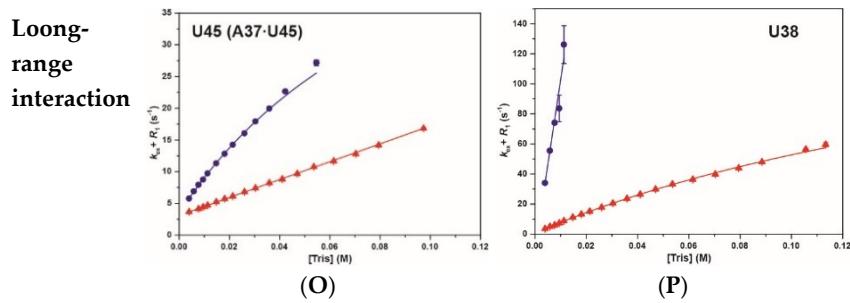


P2



P3





**Figure S4.** Hydrogen exchange data of the *CrcB* aptamer. The  $R_{1a}$  of the imino protons as a function of the Tris concentration in NMR buffer at 25 °C. The location of the imino protons in the *CrcB* aptamer is marked on the left.

**Supplementary table 1.** Hydrogen exchange rate constants ( $k_{\text{ex}}$ , s<sup>-1</sup>) of the *CrcB* aptamer.

Base pair	Imino proton	Free	Apo	Holo
G1·C16	G1	- <sup>a</sup>	17.3 ± 0.09	15.8 ± 0.1
G2·C15	G2	7.47 ± 0.02	4.73 ± 0.02	4.56 ± 0.04
C3·G14	G14	5.63 ± 0.03	5.83 ± 0.05	9.22 ± 0.1
G4·C13	G4	9.62 ± 0.03	5.45 ± 0.03	4.77 ± 0.04
G23·C34	G23	20.9 ± 0.1	5.65 ± 0.09	4.74 ± 0.03
C24·G33	G33	4.44 ± 0.01	4.85 ± 0.03	3.68 ± 0.08
U25·A32	U25	4.72 ± 0.02	4.29 ± 0.02	4.22 ± 0.02
C26·G31	G31	5.12 ± 0.02	5.80 ± 0.03	5.42 ± 0.04
U27·G30	G30	17.4 ± 0.04	16.9 ± 0.1	17.0 ± 0.09
U12·G39	G39	- <sup>a</sup>	4.53 ± 0.03	3.74 ± 0.04
U12·G39	U12	- <sup>a</sup>	3.10 ± 0.04	1.94 ± 0.02
U11·A40	U11	- <sup>a</sup>	4.33 ± 0.02	3.28 ± 0.02
G10·C41	G10	- <sup>a</sup>	5.56 ± 0.03	5.86 ± 0.07
U9·A42	U9	- <sup>a</sup>	4.39 ± 0.03	4.22 ± 0.03
G8·C43	G8	- <sup>a</sup>	4.97 ± 0.03	5.09 ± 0.05
G7·C44	G7	- <sup>a</sup>	4.88 ± 0.03	4.56 ± 0.04
A37·U45	U45	- <sup>a</sup>	3.70 ± 0.03	2.79 ± 0.03
	U38	- <sup>a</sup>	4.44 ± 0.07	2.28 ± 0.07

<sup>a</sup> No imino proton resonance.

**Supplementary table 2.** Base-pair dissociation constants ( $K_{\text{op}}$ ), base-pair lifetimes ( $\tau_0 = 1/k_{\text{op}}$ ) and lifetimes for base-pair opening ( $\tau_{\text{open}} = 1/k_{\text{cl}}$ ) of the *CrcB* motif determined by the Tris-catalyzed NMR exchange experiments at 25 °C<sup>a</sup>.

Substructure	Base pair	Imino proton		Free	Apo	Holo
P1	G2·C15	G2	$K_{\text{op}} (\times 10^6)$	$1.26 \pm 0.04$	$0.12 \pm 0.007$	$0.16 \pm 0.005$
			$\tau_0 (\text{ms})$	n.d. <sup>b</sup>	n.d. <sup>b</sup>	n.d. <sup>b</sup>
			$\tau_{\text{op}} (\text{ns})$			
			$k_{\text{int}} (\times 10^6)$			
	C3·G14	G14	$K_{\text{op}} (\times 10^6)$	$1.04 \pm 0.09$	$0.17 \pm 0.002$	$< 0.01 \times 10^{-6}$
			$\tau_0 (\text{ms})$	n.d. <sup>b</sup>	n.d. <sup>b</sup>	n.d. <sup>b</sup>
			$\tau_{\text{op}} (\text{ns})$			
			$k_{\text{int}} (\times 10^6)$			
	G4·C13	G4	$K_{\text{op}} (\times 10^6)$	$34 \pm 0.4$	$0.82 \pm 0.01$	$0.047 \pm 0.001$
			$\tau_0 (\text{ms})$	n.d. <sup>b</sup>	n.d. <sup>b</sup>	n.d. <sup>b</sup>
			$\tau_{\text{op}} (\text{ns})$			
			$k_{\text{int}} (\times 10^6)$			
P2	G23·C34	G23	$K_{\text{op}} (\times 10^6)$	n.d. <sup>d</sup>	$0.27 \pm 0.007$	$0.089 \pm 0.002$
			$\tau_0 (\text{ms})$		n.d. <sup>b</sup>	n.d. <sup>b</sup>
			$\tau_{\text{op}} (\text{ns})$			
			$k_{\text{int}} (\times 10^6)$			
	C24·G33	G33	$K_{\text{op}} (\times 10^6)$	$4.29 \pm 0.5$	$0.095 \pm 0.003$	$0.070 \pm 0.002$
			$\tau_0 (\text{ms})$	$52 \pm 1$	n.d. <sup>b</sup>	n.d. <sup>b</sup>
			$\tau_{\text{op}} (\text{ns})$	$224 \pm 27$		
			$k_{\text{int}} (\times 10^6)$	$2.03 \pm 0.2$		
	U25·A32	U25	$K_{\text{op}} (\times 10^6)$	$10 \pm 0.1$	$5.32 \pm 0.03$	$5.03 \pm 0.03$
			$\tau_0 (\text{ms})$	n.d. <sup>b</sup>	n.d. <sup>b</sup>	n.d. <sup>b</sup>
			$\tau_{\text{op}} (\text{ns})$			
			$k_{\text{int}} (\times 10^6)$			
P3	C26·G31	G31	$K_{\text{op}} (\times 10^6)$	$63 \pm 35$	$20 \pm 11$	$9.17 \pm 4$
			$\tau_0 (\text{ms})$	$67 \pm 0.4$	$69 \pm 0.8$	$67 \pm 1$
			$\tau_{\text{op}} (\text{ns})$	$4243 \pm 2388$	$1359 \pm 762$	$611 \pm 253$
			$k_{\text{int}} (\times 10^6)$	$0.31 \pm 0.1$	$1.57 \pm 0.2$	$190 \pm 0.3$
	U12·G39	G39	$K_{\text{op}} (\times 10^6)$	n.d. <sup>e</sup>	$2.72 \pm 0.4$	$2.10 \pm 1$
			$\tau_0 (\text{ms})$		$70 \pm 2$	$104 \pm 6$
			$\tau_{\text{op}} (\text{ns})$		$192 \pm 32$	$219 \pm 107$
			$k_{\text{int}} (\times 10^6)$		$2.98 \pm 0.3$	$3.27 \pm 1$
	U12·G39	U12	$K_{\text{op}} (\times 10^6)$	n.d. <sup>e</sup>	$4.93 \pm 0.4$	$2.34 \pm 0.7$
			$\tau_0 (\text{ms})$		$15 \pm 0.6$	$20 \pm 4$
			$\tau_{\text{op}} (\text{ns})$		$72 \pm 6$	$47 \pm 16$
			$k_{\text{int}} (\times 10^6)$		$0.55 \pm 0.03$	$0.83 \pm 0.2$
P4	U11·A40	U11	$K_{\text{op}} (\times 10^6)$	n.d. <sup>e</sup>	$0.27 \pm 0.004$	$0.079 \pm 0.002$
			$\tau_0 (\text{ms})$		n.d. <sup>b</sup>	n.d. <sup>b</sup>
			$\tau_{\text{op}} (\text{ns})$			
			$k_{\text{int}} (\times 10^6)$			

G10·C41	G10	$K_{\text{op}} (\times 10^6)$	n.d. <sup>e</sup>	$0.12 \pm 0.003$	$< 0.01 \times 10^{-6}$ <sup>c</sup>
		$\tau_0$ (ms)		n.d. <sup>b</sup>	n.d. <sup>b</sup>
		$\tau_{\text{op}}$ (ns)			
		$k_{\text{int}} (\times 10^6)$			
U9·A42	U9	$K_{\text{op}} (\times 10^6)$	n.d. <sup>e</sup>	$1.28 \pm 0.1$	$1.88 \pm 0.2$
		$\tau_0$ (ms)		$51 \pm 1$	$51 \pm 2$
		$\tau_{\text{op}}$ (ns)		$65 \pm 5$	$96 \pm 12$
		$k_{\text{int}} (\times 10^6)$		$6.25 \pm 0.3$	$3.69 \pm 0.3$
G8·C43	G8	$K_{\text{op}} (\times 10^6)$	n.d. <sup>e</sup>	$< 0.01 \times 10^{-6}$	$< 0.01 \times 10^{-6}$
		$\tau_0$ (ms)		n.d. <sup>b</sup>	n.d. <sup>b</sup>
		$\tau_{\text{op}}$ (ns)			
		$k_{\text{int}} (\times 10^6)$			
G7·C44	G7	$K_{\text{op}} (\times 10^6)$	n.d. <sup>e</sup>	$2.61 \pm 0.8$	$0.073 \pm 0.002$
		$\tau_0$ (ms)		$84 \pm 3$	n.d. <sup>b</sup>
		$\tau_{\text{op}}$ (ns)		$219 \pm 69$	
		$k_{\text{int}} (\times 10^6)$		$5.20 \pm 0.7$	
Tertiary interactions	A37·U45	$K_{\text{op}} (\times 10^6)$	n.d. <sup>e</sup>	$5.05 \pm 0.8$	$1.05 \pm 0.007$
		$\tau_0$ (ms)		$15 \pm 1$	n.d. <sup>b</sup>
		$\tau_{\text{op}}$ (ns)		$73 \pm 14$	
		$k_{\text{int}} (\times 10^6)$		$0.72 \pm 0.1$	
U38·C41 and A40	U38	$K_{\text{op}} (\times 10^6)$	n.d. <sup>e</sup>	$84 \pm 0.8$	$5.52 \pm 0.6$
		$\tau_0$ (ms)		n.d. <sup>b</sup>	$5.61 \pm 0.4$
		$\tau_{\text{op}}$ (ns)			$31 \pm 4$
		$k_{\text{int}} (\times 10^6)$			$0.091 \pm 0.02$

<sup>a</sup> Parameters used in the calculation:  $k_{\text{coll}} = 1.5 \times 10^9 \text{ s}^{-1}$ ,  $pK_a(\text{G-NH1}) = 9.24$ ,  $pK_a(\text{U-NH3}) = 9.20$ ,  $pK_a(\text{Tris}, 25^\circ\text{C}) = 8.192$ ; sample conditions: 10 mM Tris (pH 8.0 at 25 °C), 50 mM KCl, 50 μM EDTA (pH 8.0) (for free state)/adding 2 mM MgCl<sub>2</sub> (for apo state)/2 mM MgCl<sub>2</sub>, 10 mM NaF (for holo state). [Tris] total = 10–339 mM, 25 °C. The errors for these values were determined from the curve fitting using Equation (6) and linear fitting using Equation (7). <sup>b</sup> Not determined. <sup>c</sup> These resonances are partially overlapped with another resonance and this overlap may lead to a systematic error in  $K_{\text{op}}$ . <sup>d</sup> Not available because the imino proton resonance disappeared. <sup>e</sup> No imino proton resonance.