

A.1. Results

Table S1. Texture properties of the methyl substituted silica gels.

Sample (mol% MTES)	Hysteresis Type	Acid Catalysts	DFT (nm)	$S_{\text{BET}}^{\text{a}}$ (m ² /g)	$S_{\text{BET}}^{\text{b}}$ (m ² /g)	V_{micro} (cm ³ /g)	V_{total} (cm ³ /g)
0	H2a	A	4.72	448	494	0.133	0.373
	H3	B	32.65	291	283	0.110	0.562
5	H2a	A	3.06	563	608	0.168	0.385
	H2a	B	13.46	669	696	0.238	1.304
10	H2a	A	3.06	556	604	0.166	0.360
	H2a	B	7.3	777	813	0.266	0.945
20	H2a	A	3.06	588	627	0.221	0.335
	H2a	B	5.09	859	893	0.307	0.720
30	H2a	A	2.5	541	613	0.239	0.304
	H2a	B	4.89	930	976	0.324	0.728
40	H4	A	1.38	527	603	0.248	0.299
	H2a	B	5.09	1059	1115	0.356	0.931
50	H4	A	1.38	457	527	0.230	0.252
	H2a	B	5.29	851	893	0.303	0.799
60	H4	A	2.03	616	695	0.269	0.351
	H2a	B	4.89	895	938	0.332	0.704
70	H4	A	2.03	409	465	0.197	0.2378
	H4	B	2.03	694	722	0.275	0.482
80	H4	A	2.03	251	265	0.115	0.158
	H4	B	2.11	424	438	0.189	0.282

a – single point, b - multi point

Table S2. Average values of the pores and primary particles size obtained in small angle scattering measurements by fitting data with the unified exponential/power-law model.

MTES content (%)	Diameter (nm)			
	series A		series B	
	SANS	SAXS	SANS	SAXS
0	10.35 ± 0.16	8.11 ± 0.19	53.59 ± 6.02	30.68 ± 4.35
5	7.29 ± 0.09	6.34 ± 0.10	15.13 ± 0.35	11.39 ± 0.50
10	5.64 ± 0.08	5.36 ± 0.04	9.68 ± 0.17	8.37 ± 0.19
20	4.74 ± 0.11	4.65 ± 0.07	6.84 ± 0.14	6.34 ± 0.09
30	8.17 ± 1.06	5.59 ± 1.02	6.48 ± 0.18	5.85 ± 0.15
40	12.0 ± 2.21	8.60 ± 2.13	8.55 ± 0.31	7.59 ± 0.28
50	15.02 ± 2.28	28.08 ± 2.18	10.54 ± 0.27	10.45 ± 0.24
60	5.04 ± 0.63	5.38 ± 1.15	9.12 ± 0.45	9.28 ± 0.50
70	11.17 ± 5.33	7.59 ± 3.12	8.85 ± 0.86	8.48 ± 0.91
80	7.32 ± 1.32	6.89 ± 1.29	11.10 ± 4.25	9.69 ± 3.78

Table S3. XRD maxima position shift with the MTES mole percent

Sample	Position	FWHM	Height
A0	22.70	8.65	276.3
A5	22.65	8.65	260.7
A10	22.66	8.92	249.8
A20	22.61	9.42	226.5
A30	22.64	10.15	216.4
A40	22.57	10.87	211.9
A50	22.56	12.2	211.8
A60	22.47	12.06	209.2
A70	22.45	11.3	219.8
A80	22.34	11.43	216.3
B0	22.86	9.19	277.3
B5	22.70	9.06	240.9
B10	22.8	9.51	250.8
B20	22.81	10.01	244.2
B30	2.77	10.13	228.2
B40	22.72	9.98	207.2
B50	22.65	10.22	213.8
B60	22.51	10.78	215.5
B70	22.48	10.92	217.5
B80	22.32	11.07	211.4

Table S4. Quantitative analysis of ^{29}Si MAS NMR spectra.

Sample	^{29}Si MAS NMR											
	Q ⁴ (%)	Q ³ (%)	Q ² (%)	Q (%)	T ³ (%)	T ² (%)	T ¹ (%)	T (%)	Q ⁴ + T ³ (%)	Q ⁴ + T ³ + Q ³ (%)	Q ³ +Q ² + T ² (%)	Q ² + T ² (%)
B0	70.5	28.70	0.82	100	0	0	-	0	70.53	99.19	29.48	0.82
B5	63.3	32.30	1.51	97.20	T3+T2 = 2.82	-	2.82	66.16	98.49	33.84	1.51	
B10	65.6	22.10	0	87.70	10.20	2.11	-	12.31	75.78	97.89	24.22	2.11
B40	43.6	14.80	0.62	58.9	26.50	14.51	-	41.05	70.11	84.86	29.88	15.13
B60	36.6	4.82	0	41.40	40.50	18.10	-	58.58	77.08	81.90	22.92	18.10
B80	17.4	1.58	0	19.90	61.18	18.90	0.91	80.07	78.61	80.19	20.47	18.89

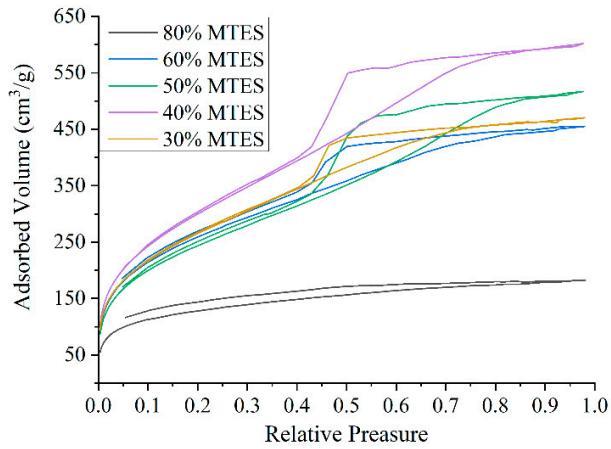


Figure S1. Evolution of nitrogen adsorption isotherms with 30-80% of MTES content.

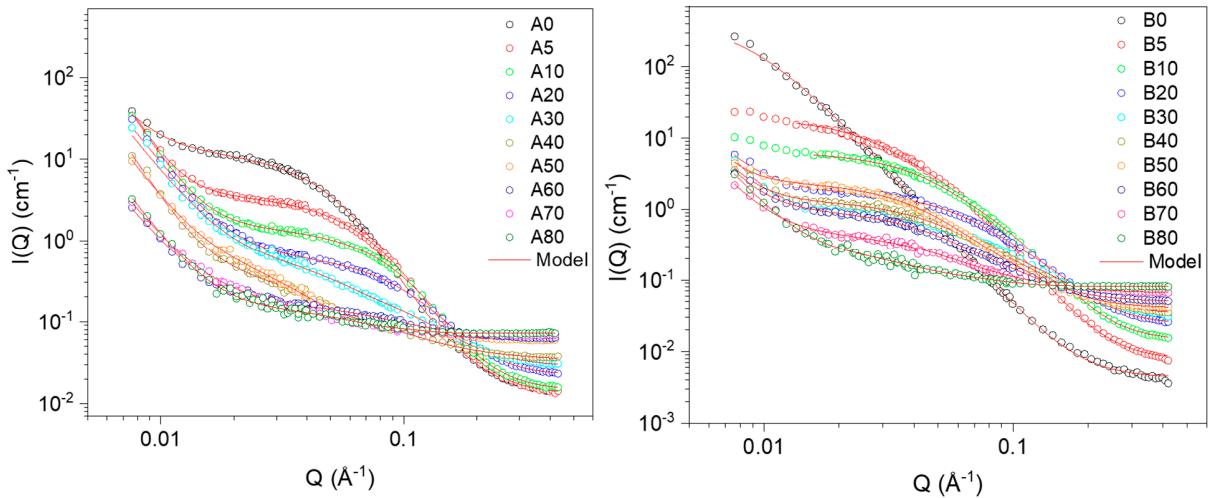


Figure S2. SANS profiles of series A (left) and B (right) as the amount of MTES in the reaction mixture increases from 5 % (A5/B5) up to 80 % (A80/B80). The SANS spectra of the TEOS silica precursor are also displayed for comparison (A0/B0). The applied model is also displayed as red lines.

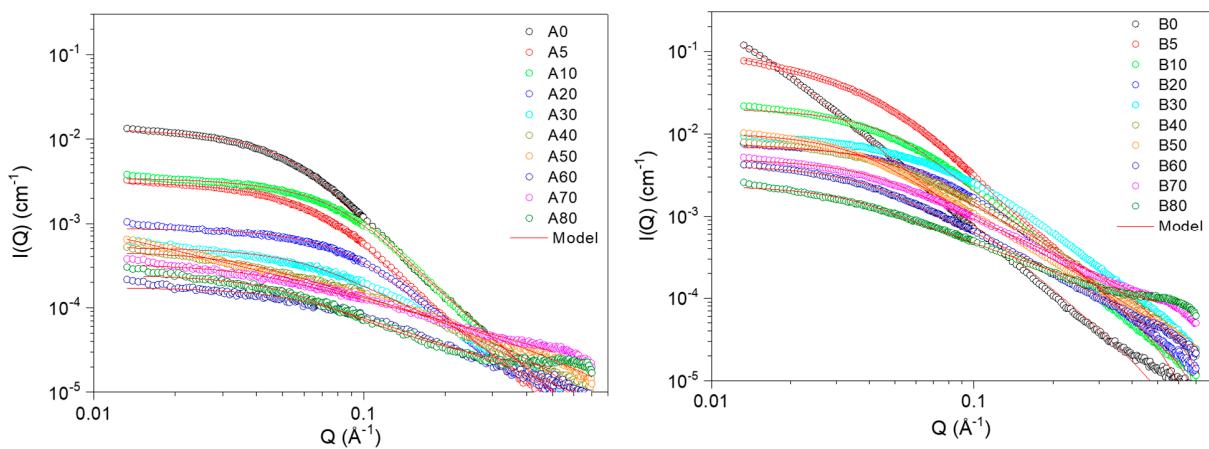


Figure S3. SAXS profiles obtained for series A (left) and B (right), as the amount of MTES in the reaction mixture increases from 5 % (A5/B5) up to 80 % (A80/B80). The SAXS spectra of the TEOS silica precursor are also displayed for comparison (A0/B0). The applied model is also displayed as red lines.

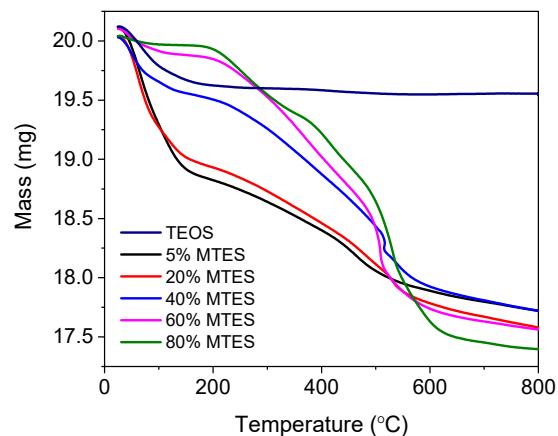
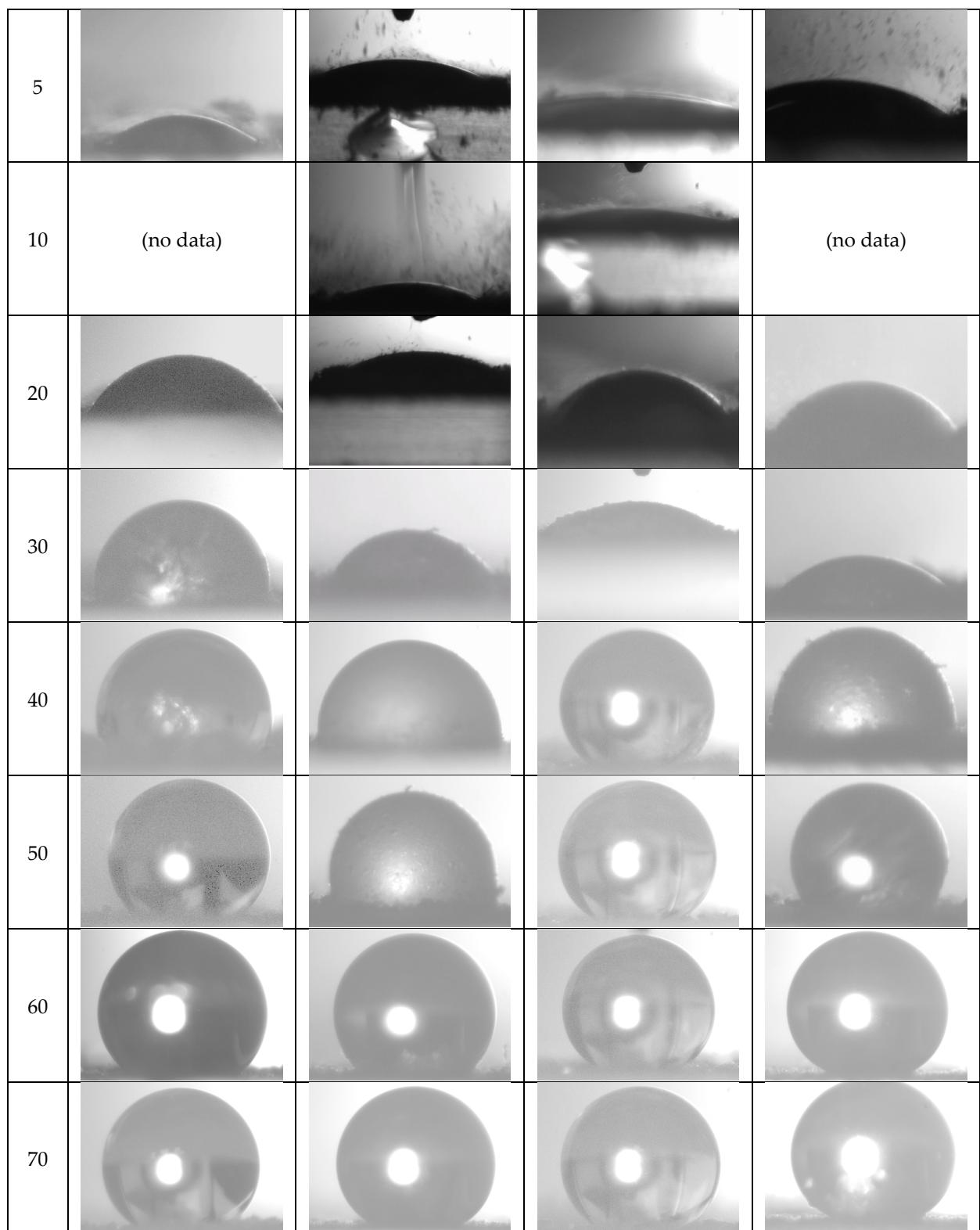


Figure S4. TG curves of the silica xerogels with different methyl content

MTES content (%)	series A		series B	
	as-received	ground	as-received	ground
0				



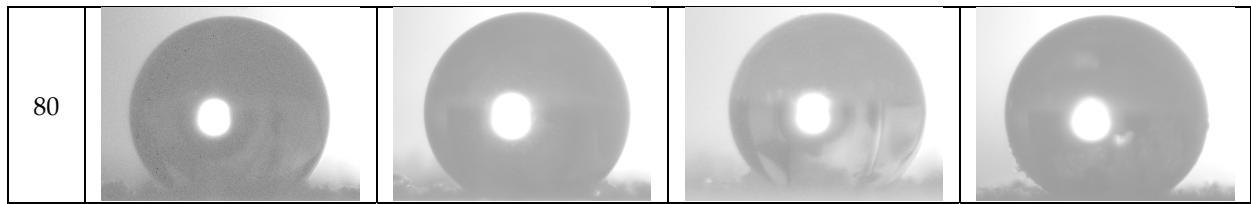


Figure S5. Representative contact angle images, one for each sample type, series A, respectively B, both as-received and ground.

Mathematical modelling of drug release

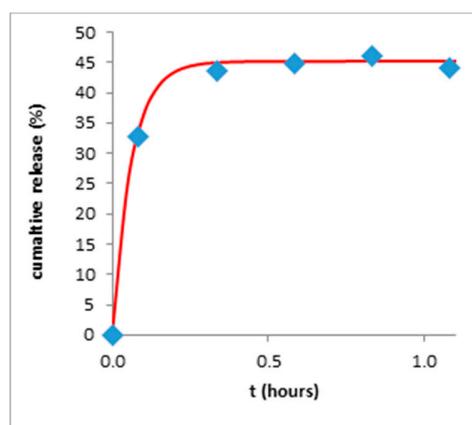


Figure S6. Fitted drug release curve of carrier A40 silica measured in HCl medium.