

SUPPLEMENTARY MATERIALS

Morphological, anatomical, and phytochemical studies of *Carlina acaulis* L. cypsela

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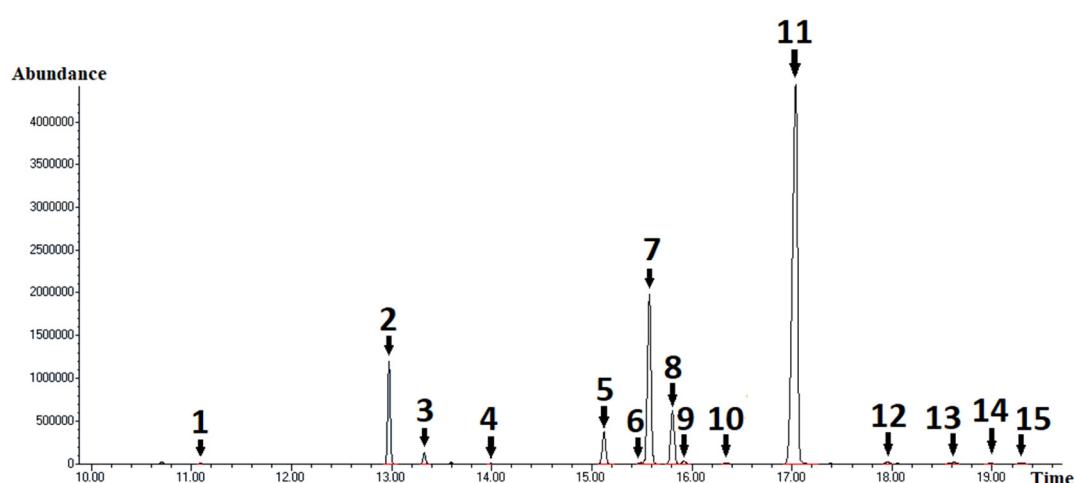


Figure 1. Example of a GC-MS chromatogram of *Carlina acaulis* cypsela oil. 1 – Myristic acid, 2 – Palmitic acid, 3 – cis-5-hexadecenoic acid ?, 4 – Margaric acid, 5 – Stearic acid, 6 – Unidentified compound, 7 – 5-Octadecenoic acid, 8 – Oleic acid, 9 – 8-Octadecenoic acid or 11- Octadecenoic acid, 10 – 9,12-Octadecadienoic acid, 11 – Linoleic acid, 12 – Arachidic acid, 13 – alpha-Linolenic acid, 14 – 11,14,17-Eicosadienoic acid or 9,12-Octadecadienoic acid, 15 – Mangiferic acid. HP-88 Agilent capillary column (60 m × 0.25 mm; 0.20 µm film thickness). The oven temperature was programmed from 110 °C to 190 °C with 8 °C/min, hold for 2 min at 110 °C and 13 min at 190 °C. The temperature of the injector was 250 °C. Helium was used as a carrier gas at a flow rate of 1.2 mL/min A quadrupole mass spectrometer with electron ionization (EI) at 70 eV and with a full scan type acquisition mode (50 m/z to 500 m/z) was used as a detector connected with the GC. The temperature of the MS source and the MS quadrupole was set to 230 °C and 150 °C, respectively.

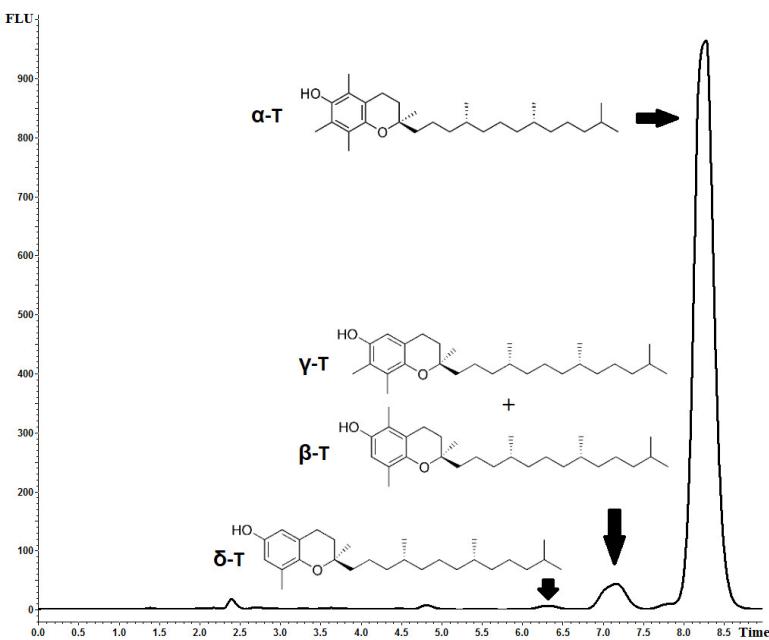
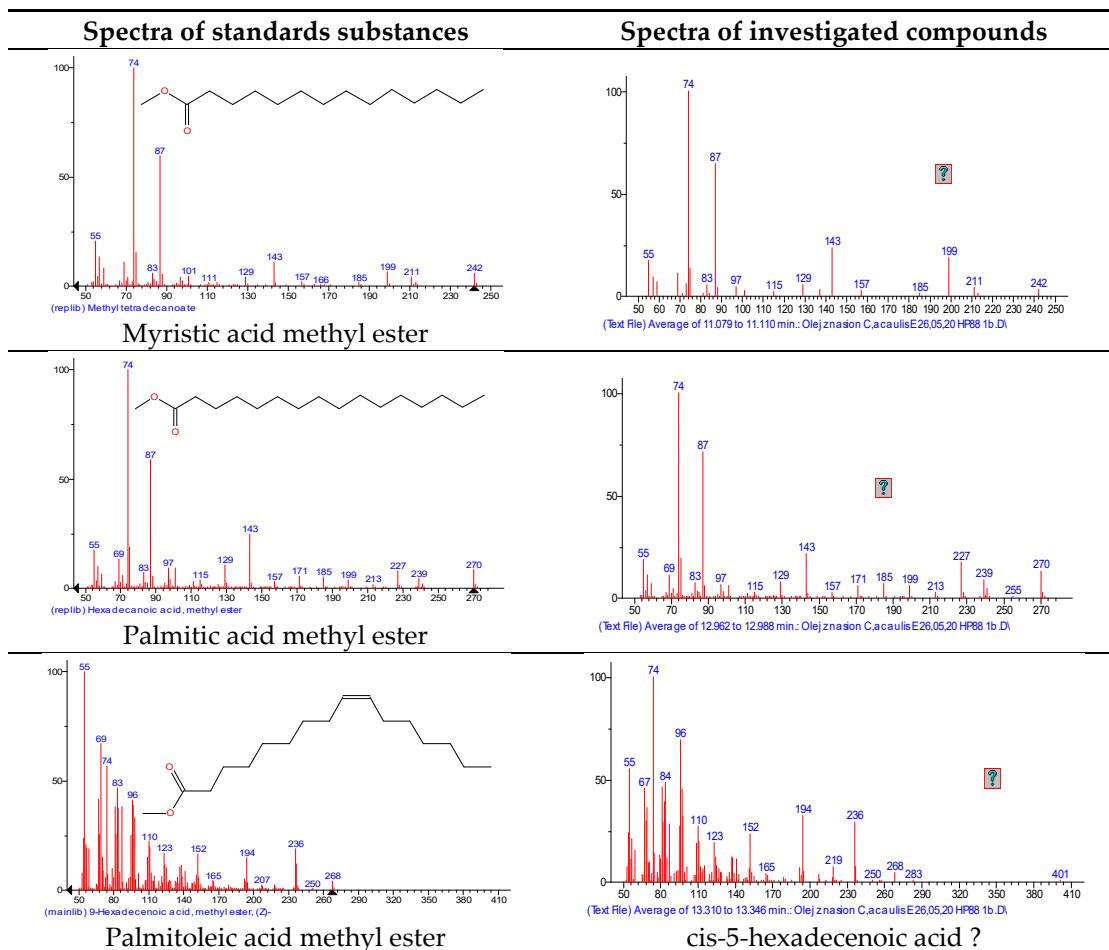
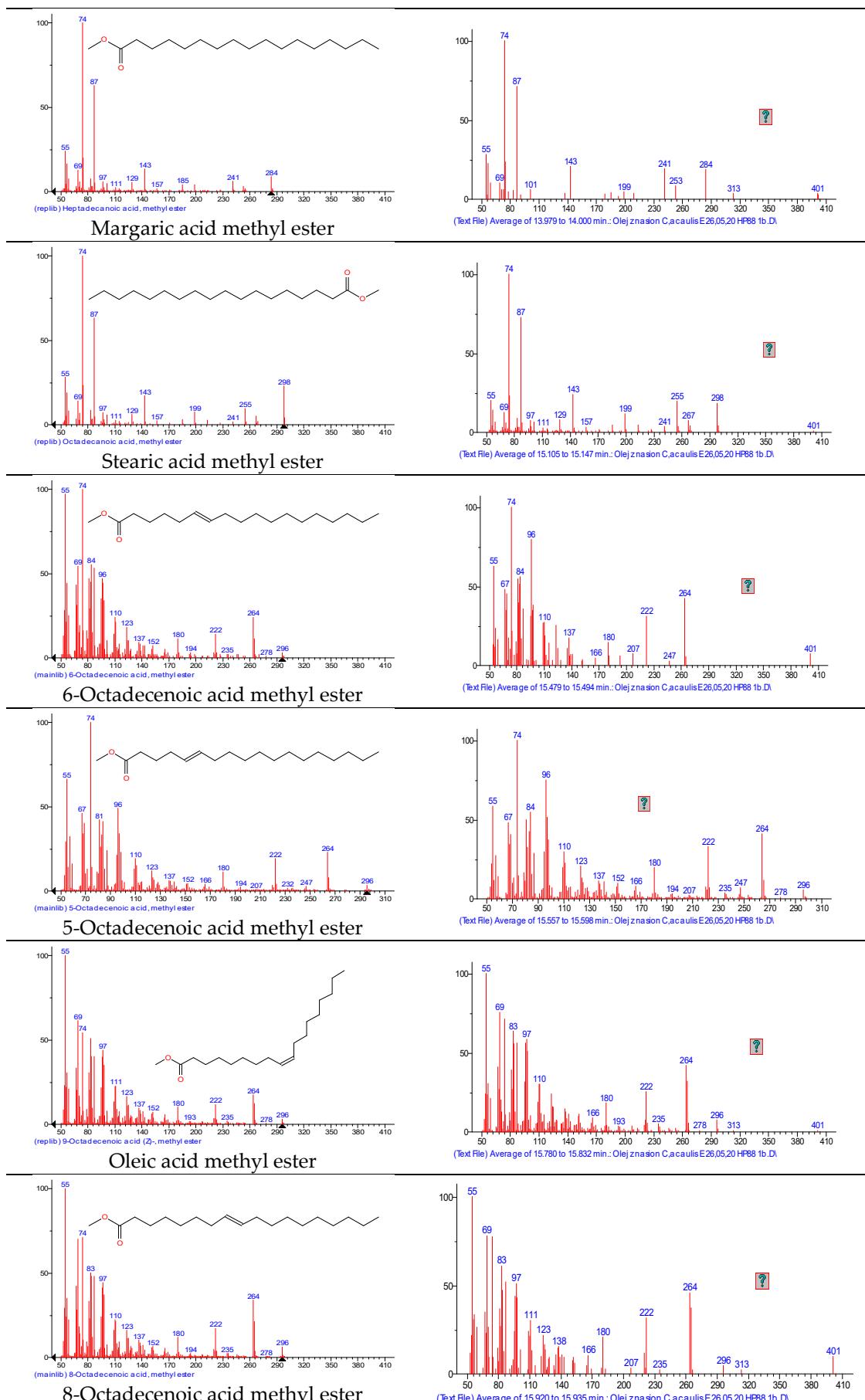
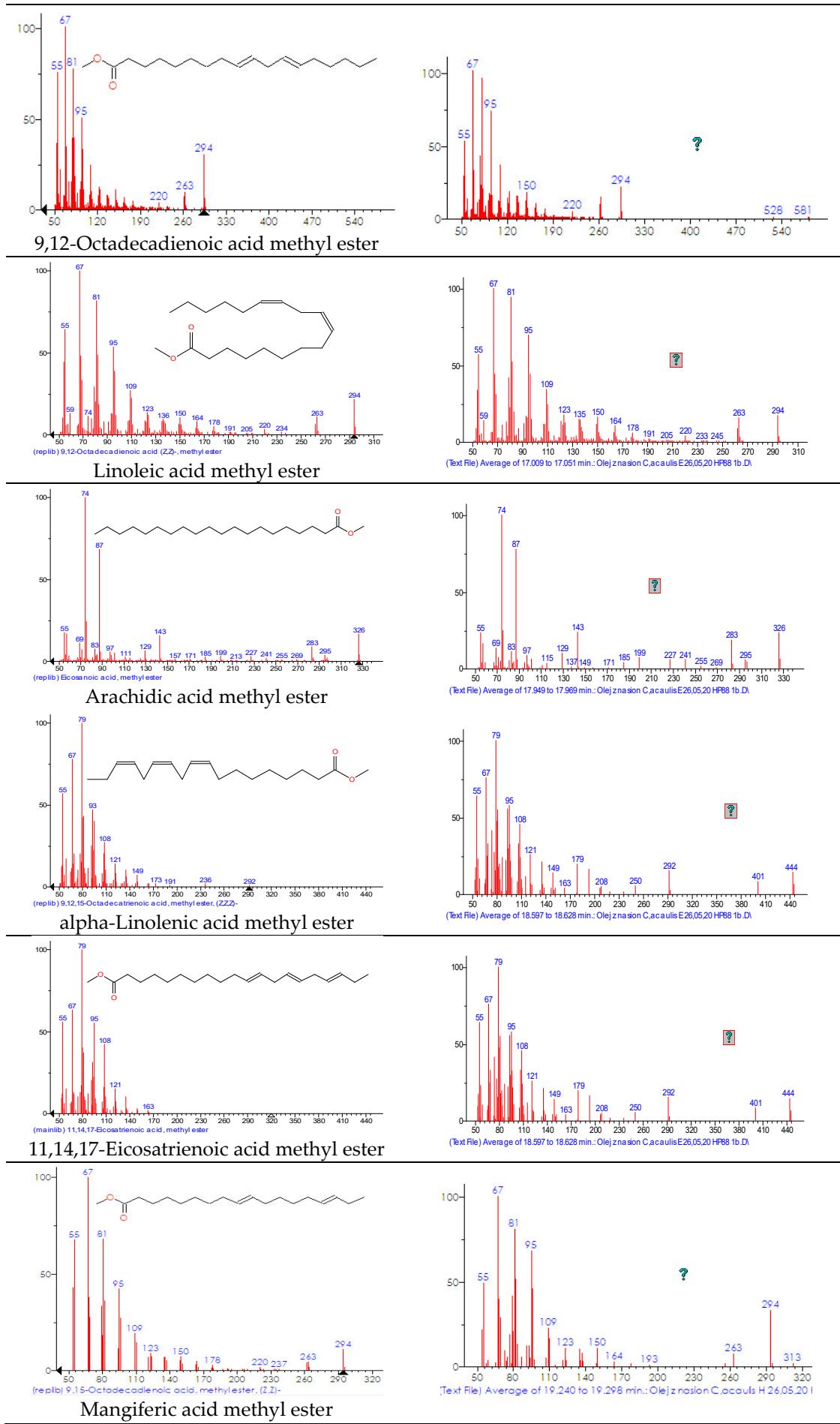


Figure 2. Example of an HPLC-FLD chromatogram of *Carlina acaulis* cypsela oil. The analysis was performed on an RP18e LiChrospher 100 column (25 cm x 4.0 mm i.d., 5 μ m particle size). The column temperature was set to 30 °C. Acetonitrile and methanol (5:95 v/v) at a flow rate of 1.2 mL/min were used as eluent. $\lambda_{\text{ex}}=296$ nm, $\lambda_{\text{em}}=330$ nm. α -T – alpha-Tocopherol, β -T – beta-Tocopherol, γ -T – gamma-Tocopherol, δ -T – delta-Tocopherol.

Table 1. Mass spectra for standards (based on the NIST database) and investigated fatty acids.







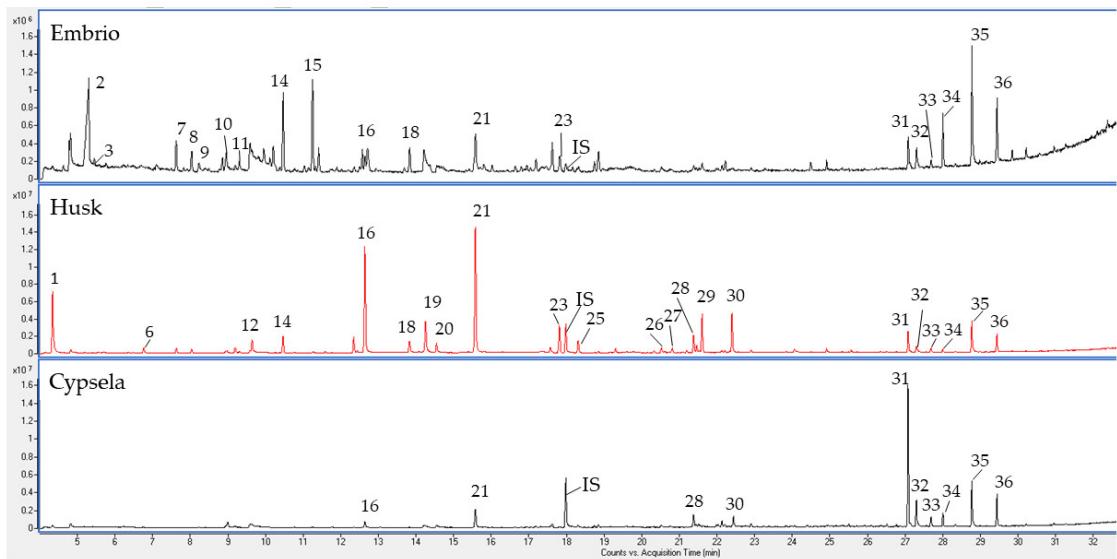


Figure 3. Example of a GC-MS chromatogram of volatile compounds in *Carlina acaulis* cypselae. 1 – hexanal, 2 – 3-Methylbutanoic acid, 3 - 2-Methylbutanoic acid, 4 – 2-Hexenal, 5 - Pentanoic acid, 6 – Heptanal, 7 – α -Thujene, 8 – Camphene, 9 – γ -Valerolactone, 10 – Hexanoic acid, 11 – 2-Pentylofuran, 12 – Octanal, 13 – Limonene, 14 – Eucalyptol, 15 – 4,5-Dimethylnonane, 16 – Nonanal, 17 – 2-Ethylhexanoic, 18 – Camphor, 19 – 2-Nonenal, 20 – Octanoic acid, 21 – Decanal, 22 – Nonanoic acid, 23 – Bornyl acetate, IS – 2-Undecanone (internal standard), 25 – Undecanal, 26 – Tertadecane, 27 – Dodecanal, 28 – trans-Geranylactone, 29 – Alloaromadendrene, 30 – Tridecanal, 31 – Isopropyl myristate, 32 – Farnesyl acetaldehyde, 33 – Phtalic acid, hept-4-yl isobutyl ester, 34 – Nonadecane, 35 – 2-Ethylhexyl octadecyl carbonate, 36 – Isopropyl palmitate, 37 – Heneicosane, 38 – 2-Ethylhexyl 4-methoxycinnamate. The analysis was performed using Agilent 7890B GC coupled with the 7000GC/TQ system (Agilent Technologies, Palo Alto, CA). Separation was carried out on an HP-5 MS column; 30m \times 0.25 mm \times 0.25 μ m (J&W, Agilent Technologies, Palo Alto, CA) at a constant helium flow of 1 mL/min. The injector temperature was set at 250°C and the sample was applied in a split mode (20:1). The temperature program was 50°C for 1 min, followed by 4°C/min to 130°C, 10°C/min to 280°C, and held isothermal for 2 min. The MS source was set at 230°C, the transfer line was 320°C, and the quadrupole temperature was 150°C. The electron ionization energy was set at 70 eV, scan range, m/z 30-400.

Table 2. Mass spectra of investigated volatile compounds.

