## SUPPLEMENTARY INFORMATION

## Fluorescence assay for the determination of D-panthenol based on novel ringfused 2-pyridone derivative

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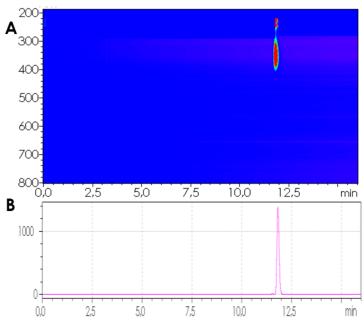
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**Table S1**. Molar ratios for reactions of citric acid in glass vial reactors at 160 °C for 3 hours with DP (A), 3A1P (B), PAN (C).

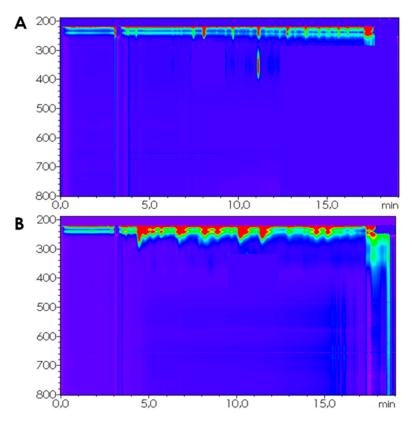
	Molar ratio CA:DP	Mass of CA [mg]	Mass of DP [mg]
	10:1	451.76	48.24
	6:1	424.47	75.53
	4:1	394.66	105.34
Α	2:1	325.98	174.02
	1:1	241.81	258.19
	1:2	159.47	340.53
	1:4	94.86	405.14
	1:6	67.51	432.49
	1:10	42.82	457.18

	Molar ratio CA:3A1P	Mass of CA [mg]	Mass of 3A1P [mg]		Molar ratio CA:PAN	Mass of CA [mg]	Mass of PAN [mg]
	10:1	481.20	18.8	-	10:1	468.29	31.71
	6:1	469.44	30.56		6:1	449.30	50.70
	4:1	455.52	44.48		4:1	427.62	72.38
_	2:1	418.30	81.70	_	2:1	373.54	126.46
В	1:1	359.55	140.45	C	1:1	298.14	201.86
	1:2	280.70	219.30		1:2	212.39	287.61
	1:4	195.12	304.88		1:4	134.83	365.17
	1:6	149.53	350.47		1:6	98.77	401.23
	1:10	101.91	398.09		1:10	64.34	435.66

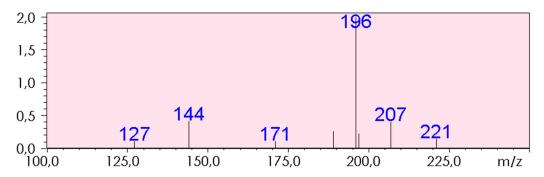
В



**Figure S1**. LC-DAD chromatogram of pure fluorophore separated from the reaction mixture (retention time: 11.8 min.) (**A**). LC chromatogram of the pure fluorophore calculated from a maximum plot of 190-800 nm (retention time: 11.8 min.) (**B**).



**Figure S2.** LC-DAD chromatogram of CA:3A1P reaction mixture (1:1 mol/mol. 160 °C. 180 min) (**A**). and LC-DAD chromatogram of CA:PAN reaction mixture (1:1 mol/mol. 160 °C. 180 min) (**B**).



**Figure S3.** The mass spectrum of the fraction with a retention time of 11.8 minutes formed in a reaction with a molar ratio from CA:31AP 1:1.

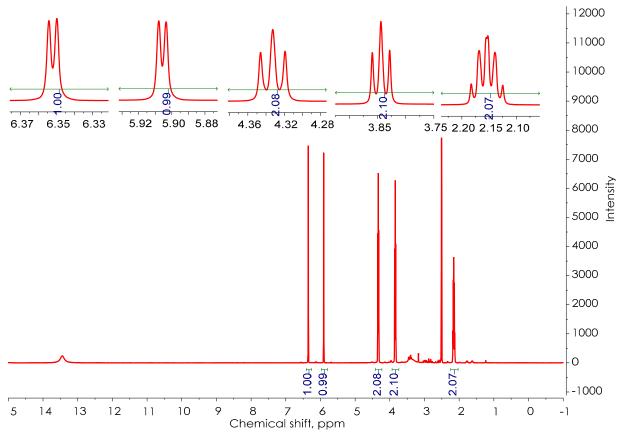
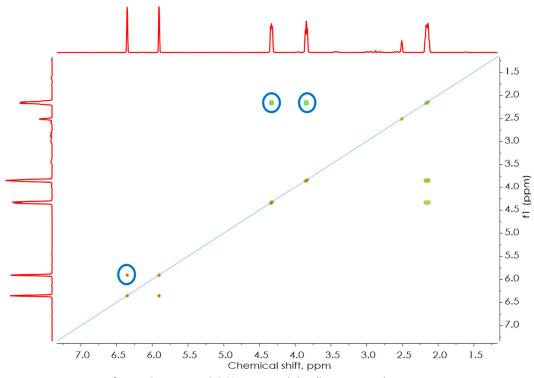


Figure S4. <sup>1</sup>H NMR of the fluorescent fraction.



**Figure S5.** <sup>1</sup>H-<sup>1</sup>H COSY NMR of the fluorescent fraction.

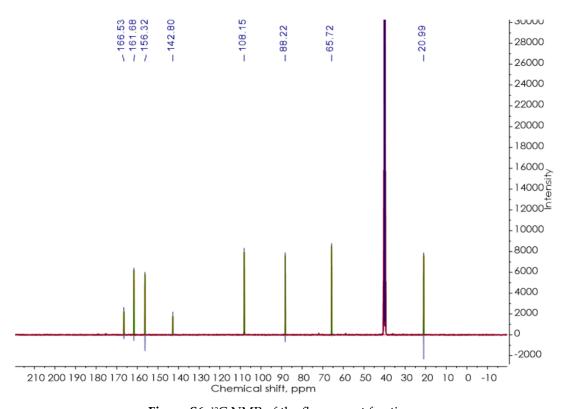


Figure S6.  $^{13}\text{C}$  NMR of the fluorescent fraction.

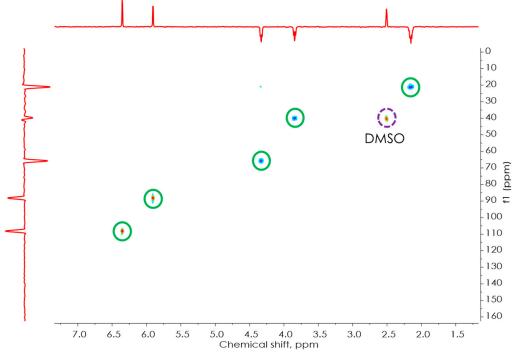
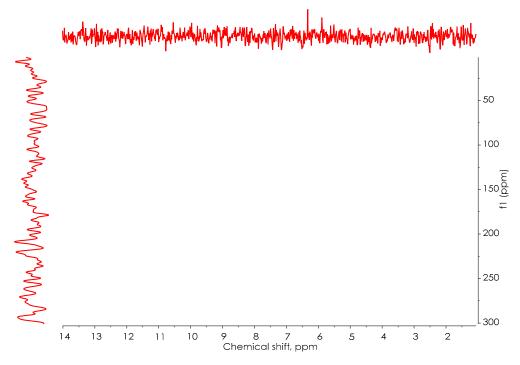
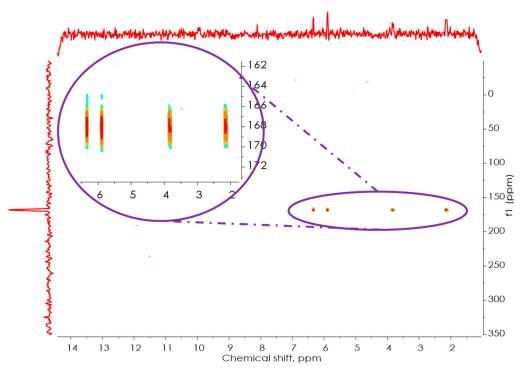


Figure S7.  $^1H^{-13}C$  HSQC NMR of the fluorescent fraction.



**Figure S8.** <sup>1</sup>H-<sup>15</sup>N HSQC of the fluorescent fraction.



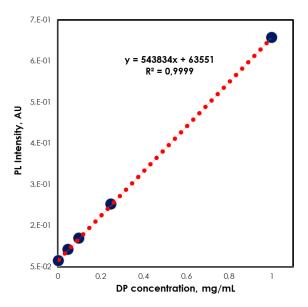
**Figure S9.** <sup>1</sup>H-<sup>15</sup>N HMBC of the fluorescent fraction.

**Table S2.** Couplings between the atoms in <sup>1</sup>H-<sup>13</sup>C HMBC NMR of the fluorescent fraction.

N.	Chemical shift	Chemical shift	Distance		
No.	(hydrogen) (ppm)	(carbon) (ppm)	(number of bonds)		
1	6.35	166.54	3		
2	5.90	166.54	3		
3	6.35	161.73	2		
4	3.84	161.73	3		
5	6.35	156.31	4		
6	5.90	156.31	2		
7	4.33	156.31	3		
8	3.84	156.31	3		
9	6.35	142.86	2		
10	5.90	142.86	2		
11	3.84	142.86	5		
12	5.90	108.14	3		
13	6.35	88.21	3		
14	3.84	65.65	3		
15	2.14	65.65	2		
16	4.33	40.25	3		
17	2.14	40.25	2		
18	4.33	20.95	2		
19	3.84	20.95	2		



Figure S10. The type of vessel used in the process of panthenol derivatization.



**Figure S11**. LC-MS calibration curve for the determination of panthenol in pharmaceutical formulations (calculated from the area under the curves).