

# Enhanced Cycling Performance of Rechargeable Zinc–Air Flow Batteries Using Potassium Persulfate as Electrolyte Additive

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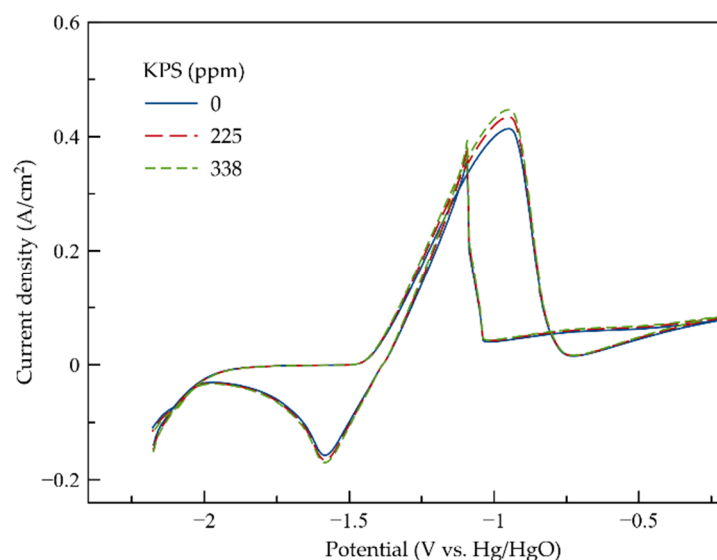
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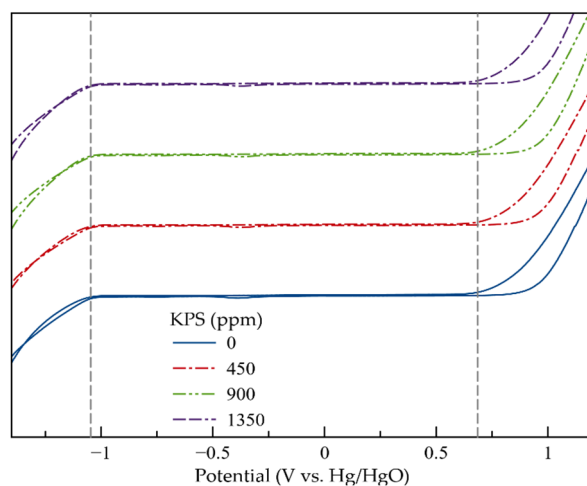
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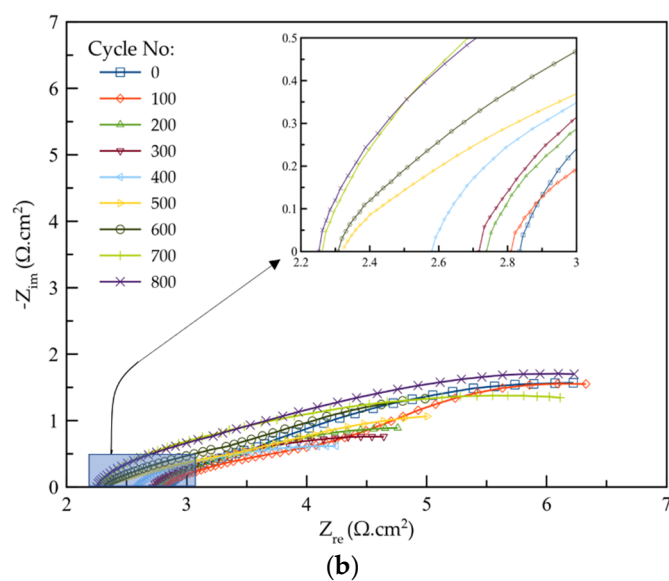
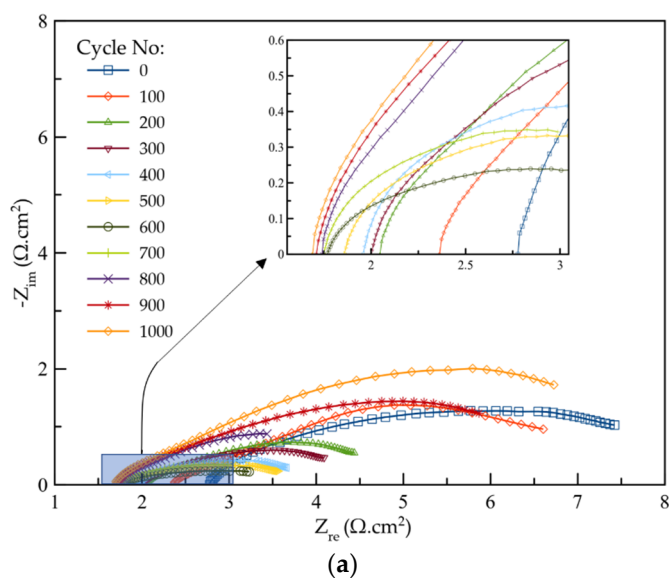
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**Figure S1.** Cyclic voltammograms of Zn in a cell containing 0, 225 and 338 ppm KPS additive at scan rate 0.07 V/s.



**Figure S2.** Electrochemical windows in the presence of 0, 450, 900 and 1350 ppm KPS additive.



**Figure S3.** Nyquist plots of the cycling impact on the impedance of a cell containing (a) 900 ppm KPS and (b) 1350 ppm KPS additive to KOH/ZnO electrolyte at initial state and after every 100 charge/discharge cycle until the 800th cycle, at a frequency range from 100 kHz to 0.1 Hz, having an alternate current amplitude of 10 mV around OCV.

**Table S1.** Resistance values of cell impedance after every 100 cycles of CV for open cell with the electrolyte containing 900 ppm KPS.

Concentration		Cycle Number										
KPS (ppm)		0	100	200	300	400	500	600	700	800	900	1000
900	$R_s(\Omega)$ ( $\pm 1\%$ )	2.77	2.36	2.04	2.00	1.95	1.85	1.76	1.75	1.74	1.71	1.69
	$R_{ct}(\Omega)$ ( $\pm 1\%$ )	8.76	7.05	7.93	7.36	7.15	6.89	6.57	6.97	10.92	11.52	13.14

**Table S2.** Resistance values of cell impedance after every 100 cycles of CV for open cell with the electrolyte containing 1350 ppm KPS.

Concentration		Cycle Number										
KPS (ppm)		0	100	200	300	400	500	600	700	800	900	1000
1350	$R_s(\Omega)$ ( $\pm 1\%$ )	2.83	2.81	2.73	2.71	2.57	2.32	2.30	2.26	2.25	-	-
	$R_{ct}(\Omega)$ ( $\pm 1\%$ )	14.31	13.72	7.75	7.84	8.32	8.86	10.17	13.01	14.41	-	-