# Potent MOR Agonists from 2'-Hydroxy-5,9-dimethyl- $N$-phenethyl Substituted-6,7-benzomorphans and from C8-Hydroxy, Methylene and Methyl Derivatives of $N$-Phenethylnormetazocine 

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#### Abstract

Dimethyl-6,7-benzomorphan (normetazocine) derivatives with a para-OH or ortho-F substituent in the aromatic ring of the $N$-phenethyl moiety were synthesized and found to have subnanomolar potency at MOR, and both were fully efficacious in vitro. These new compounds, (1R,5R,9R)-6,11-dimethyl-3-(2-fluorophenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin-8ol and (1R,5R,9R)-6,11-dimethyl-3-(4-hydroxyphenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d] azocin-8-ol, were more potent than the unsubstituted compound $N$-phenethylnormetazocine and about 30 or 40 times more potent than morphine, respectively. A variety of substituents in the ortho, meta, or para position in the aromatic ring of the $N$-phenethyl moiety were synthesized, 25 of these compounds, and found to have varying effects on potency and efficacy as determined by the forskolin-induced cAMP accumulation assay. The $N$-phenethyl moiety was also modified by increasing chain length to form a $N$-phenylpropyl side chain with and without a para-nitro moiety, and by an $N$-cinnamyl side chain. Also, an indole ethylamine normetazocine was synthesized to replace the $N$-phenethylamine side chain in normetazocine. The phenylpropylamine, propenylamine (cinnamyl) and the para-nitropropylamine had little or no MOR potency. The indole-ethylamine on the normetazocine nucleus, however, had moderate potency ( $\mathrm{MOR} \mathrm{EC}_{50}=12 \mathrm{nM}$ ), and was fully efficacious ( $\% \mathrm{E}_{\max }=102 \%$ ) in the cAMP assay. Retention of the $N$-phenethyl moiety and the addition of alkyl and alkenyl moieties on C8 in (-)-N-phenethylnormetazocine gave a C8-methylene derivative that had subnanomolar potency at MOR and a C8-methyl analog that had nanomolar potency. Five C8-substituted compounds were synthesized.


Keywords: opioid; $N$-phenethyl analogs of 6,7-benzomorphans; $N$-phenethylnormetazocine derivatives; C8-alkyl substituted 6,7-benzomorphans; C8-alkenyl substituted 6,7-benzomorphans; inhibition of forskolin-induced cAMP accumulation assay (cAMP assay)

## 1. Introduction

The 6,7-benzomorphans (Figure 1) have been of interest for the past several decades and analgesics based on them have been found to be clinically useful [1,2]. The 6,7benzomorphan structure is derived from the morphinans. It lacks the morphinan C-ring,
making the 6,7-benzomorphans less complex (Figure 1) than the morphinan or epoxymorphinan opioid structures. This structural modification did not necessarily lead to improved analgesics: e.g., a prototypical 6,7-benzomorphan, metazocine, $N$-methyl-5, 9-dimethyl-6,7benzomorphan, not only acts at the mu opioid receptor (MOR) essentially like morphine, but also has dysphoric and hallucinogenic side-effects attributed to interaction with the kappa opioid receptor (KOR). Several of the derivatives of the 6,7-benzomorphans have been found to act as MOR agonists or antagonists, depending on the N -substituent. The racemic phenazocine ( $N$-phenethyl relative of metazocine, Figure 1) was found to have milder side-effects in vivo (e.g., physical dependence, sedation, depression) than equivalent doses of morphine. Pentazocine, the $N$-dimethylallyl relative of metazocine, acts at both MOR and KOR and has been found to be a clinically useful analgesic with milder opioid-like side-effects. It is in clinical use and has been combined with naloxone to lessen its illicit use. The (+)-N-allylnormetazocine (SKF 10.047, Alazocine) is the prototypical sigma-1 receptor agonist [3] and does not interact at opioid receptors. Its (-)-enantiomer was found to be a MOR antagonist [4].


Morphine (Epoxymorphinan)


Morphinan


Common name:
(1R,5R,9R-meta-Hydroxy-
5,9-dimethyl-
6,7-benzomorphan
((-)-normetazocine)


IUPAC Name:
( $2 R, 6 R, 11 R$ )-6,11-dimethyl-
1,2,3,4,5,6-hexahydro-
2,6-methanobenzo[d]azocin-8-ol


Figure 1. Molecular structures of morphine, morphinan, and the 6,7-benzomorphans-metazocine, phenazocine and pentazocine.

A considerable amount of effort has been expended on the synthesis of derivatives of the classical opioids [5], as well as the 6,7-benzomorphans [6-12], in an attempt to ameliorate their undesirable side-effects, with relatively little success [13].

Various alkyl moieties have been added to the 5,9-positions in the benzomorphan, cis and trans, and this has influenced the potency of the compound, but the side-effects were those noted for the classical opioids. When the phenolic hydroxyl was converted to a carbonamide function, it was found to have increased potency and, most interestingly, some of those benzomorphans were partial agonists in the [ ${ }^{35}$ S]GTPgS (GTP) assay [14,15]. Depending on the alkyl substituents at C5 and C9 and the stereochemistry of the molecule, changing the N -substituent has been shown to have a considerable effect on the benzomorphan [12], sometimes improving its potency and, in a few instances, changing its receptor interaction to that of a bifunctional ligand (MOR- $\sigma 1$ ) [16].

We have found that substituents on the aromatic ring of the $N$-phenethyl substituent can modify the potency and efficacy of a different type of compound, norhydromorphone, as measured in the inhibition of forskolin-induced cAMP accumulation (cAMP) and $\left[{ }^{35} \mathrm{~S}\right] \mathrm{GTPgS}$ (GTP) functional assays [17]. The $p$-chlorophenethyl moiety was found to transform norhydromorphone into a bifunctional ligand that had subnanomolar potency at MOR and DOR in the cAMP assay, and that compound did not depress respiration in a rodent assay. We posed the question of whether we could obtain similar results with substituents on the aromatic ring in the $N$-phenethyl moiety of the 6,7 -benzomorphans.

Many N-substituents have been synthesized in the 6,7-benzomorphans, including $N$-methyl through decyl [10], heterocyclic derivatives of the $N$-methyl substituent [18], cyanoalkyl, $N$-allyl and $N$-alkynyl [13], $N$-napthyl and quinolyl substituents, $N$-phenylpro panamide [12] and $N$-propyl amides [19]. $N$-Phenethyl-6,7-benzomorphans are very wellknown $[17,20]$, but a study of the in vitro effect of substituents on the aromatic ring in the $N$-phenethyl substituent is novel. In this work, a series of $(-)$ - $N$-phenethylnormetazocines were synthesized with electron-withdrawing or -donating substituents on the aromatic ring of the $N$-phenethyl moiety to determine whether they could alter the functional effects of the 6,7-benzomorphans as measured in the cAMP assay and, possibly, convert them to partial agonists, compounds that are not as efficacious as morphine, with the hope that this would lessen their side-effects. Some partial agonists have been noted to have less effect than fully efficacious agonists like heroin or fentanyl on respiratory depression [21,22], a major cause of death due to opioid overdose.

The effect of substituents at C9 in the 5-(3-hydroxy)phenylmorphans, a position comparable to the C8 in normetazocine (Figure 1), has been extensively explored [20-22]. Substituents at C9 in the phenylmorphans and C8 in the benzomorphans are relatively close to the tertiary nitrogen atom. Substituents at C8 in 6,7-benzomorphan enantiomers are not well-known, and C8-alkyl and alkenyl-substituted $N$-phenethylnormetazocine enantiomers have not been previously examined. The ( - )-N-phenethylnormetazocine enantiomers with a C8-methyl or C8 methylene substituent were found to be very potent MOR agonists and were fully efficacious.

## 2. Results and Discussion

### 2.1. Optical Resolution of $( \pm)$-Normetazocine

Starting with racemic normetazocine (rac-1, Scheme 1) [2], optical resolution was most easily achieved through treatment with (+)-3-bromo-8-camphorsulphonic acid (bromocamphorsulfonic acid) [23]. This resulted in enantiomerically pure (-)-normetazocine as the bromocamphorsulfonate salt (2). The optical purity was determined by NMR using $(R)-(+)$-methylbenzyl isocyanate for chiral derivatization in $\mathrm{CDCl}_{3} / \mathrm{MeOD}(1 / 1)$.

rac-1



(-)-2.bromocamphorsulfonate

Scheme 1. Optical resolution of racemic normetazocine.

### 2.2. Substituents on the Aromatic Ring in the N-Phenethyl Side Chain

With the optically resolved (-)-normetazocine in hand, a series of N -substituted analogs were prepared using procedures in the literature (Scheme 2) [17]. The resolved $(-)-2$ was used as the bromocamphorsulfonate salt since it was found that the use of the isolated free base resulted in poorer yields. $\mathrm{NaHCO}_{3}$ (3 equivalents) was added to the reaction to generate the free base in situ. Heating the reaction at $100^{\circ} \mathrm{C}$ overnight in DMF gave the desired N -substituted analogs in good yields. Various substituted $N$-phenethyl bromides and other miscellaneous alkyl bromides were used as alkylating agents.

(-)-2.bromocamphorsulfonate

$(-)-3-(-)-23$

Scheme 2. Synthesis of N-phenethyl substituted 6,7-benzomorphans.
All of the (-)-N-phenethyl-substituted compounds (4-nitro (3), 3-nitro (4), 2-nitro (5), 4-fluoro (6), 3-fluoro (7), 2-fluoro (8), 4-trifluoromethyl (9), 3-trifluoromethyl (10), 2trifluoromethyl (11), 4-bromo (12), 3-bromo (13), 2-bromo (14), 4-chloro (15), 3-chloro (16), 2-chloro (17), 4-methoxy (18), 4-methyl (19), 4-hydroxy (20), 2,4-dichloro (21), and 2,6dichloro (22), and the unsubstituted compound (23)), were converted to their HCl salts by dissolving the free base in hot alcohol (isopropanol, ethanol, or methanol) and then adding concentrated hydrochloric acid. The HCl salt crystallized after stirring the solution for a few hours, employing a slow diffusion method with diethyl ether. The HCl salts were recrystallized from hot isopropanol.

### 2.3. Modification of the N-Phenethyl Side Chain

Several other types of $(-)$-N-substituted compounds were synthesized (Scheme 3) and evaluated for their potency and efficacy (Table 1): an $N$-phenylpropyl substituent ( - )-24, an $N$-cinnamyl substituent $(-)-\mathbf{2 5}$, and an indole $(-)$-26.

(-)-2•bromocamphorsulfonate

()-25 (Renylpropyl)
(-)-25 (Cinnamyl)
(-)-26 (2-(1H-Indol-3-yl)ethyl)

Scheme 3. Synthesis of modified (-)-N-phenethyl relatives (-)-24-(-)-26.
We then turned to a piperidine analog, a compound bearing a 1-methylpiperidin-4ylethyl side chain. (-)-Normetazocine, as the bromocamphorsulfonate salt ((-)-2॰bromoca mphorsulfonate), was treated with 2-(1-methylpiperidin-4-yl)acetic acid using HATU as a coupling agent. The resulting amide 27 then underwent a borane reduction [24] to give the desired 1-methylpiperidin-4-ylethyl analog (-)-28 (Scheme 4).


Scheme 4. Synthesis of the 1-methyl piperidine analog (-)-28.
A different method was used to synthesize an $N$-(4-nitrophenylpropyl)-substituted analog $(-)-31$ because of the unavailability of 4-nitrophenylpropyl bromide. The synthesis was achieved by $O$-mesylating 3-(4-nitrophenyl)propan-1-ol, followed by a nucleophilic substitution with ( - )-normetazocine (Scheme 5).


(-)-31
Scheme 5. Synthesis of the 4-nitrophenylpropyl substituent (-)-31.

### 2.4. Synthesis of C8-Oxo-Normetazocine

A large-scale synthesis and subsequent optical resolution of a C8-oxo key intermediate (rac-2'-methoxy-8-oxo- N -phenethylnormetazocine, rac-34) was undertaken to obtain the starting material for the synthesis of alkenyl and alkyl substituents at that C8 position. The synthesis of the $2^{\prime}$-methoxy racemate rac- 34 proceeded according to the literature [25,26] (for the comparable phenol), in good overall yield via a three-step process from normetazocine
(1) (Scheme 6). Boc protection of the secondary amine in normetazocine, followed by methylation of the phenol and subsequent removal of the Boc group, provided 32 in $98 \%$ yield. The C8 position was oxidized using chromium trioxide to form the ketone 33 ( $63 \%$ ). Finally, the key intermediate, rac-34, was synthesized via N -alkylation of the 8 -oxo- $\mathbf{2}^{\prime}$ -methoxy-normetazocine (34) with phenethyl bromide in high yield ( $80 \%$ ). The ketone in 34 was found to be resistant to olefination under Wittig or HWE conditions as well as with oxo-nucleophiles. It did, however, react with organometallics such as alkyl lithiates and Grignard reagents.


Scheme 6. Synthesis of rac m-methoxy-8-oxo-N-phenethylnormetazocine (rac-34).

### 2.5. Synthesis of the Optical Isomers of C8-oxo m-Methoxy-N-phenethyl-6,7-benzomorphan

While we had successfully resolved normetazocine based on methodology in the literature [27], we preferred to carry the synthesis of the C8-substituted compounds up to rac-34 (Scheme 6) and resolve that compound because that would enable us to run initial exploratory reactions on the available racemic compound. After screening conditions for crystallization, a combination of DMF and acetone was found to be ideal for the optical resolution of rac-34 with (+)-tartaric acid as the chiral acid (Scheme 7). Crystallization occurred after cooling to $0^{\circ} \mathrm{C}$. Conversion to the free base gave, after crystallization, $1 S, 5 R, 9 R(-)$-34. Its (+)-enantiomer ( $1 R, 5 S, 9 S-(+)-34)$ was obtained after treatment with $(-)$ tartaric acid of the rac-34 base obtained from the residual (+)-tartrate salt of $(-)$ - 34 . These enantiomers had optical rotations of $-41.3^{\circ}$ and $+44.4^{\circ}$, respectively. The resolution of rac- 34 was scaled up to provide the needed quantity, and both enantiomers were collected at greater than $90 \%$ yield (Scheme 7).


Scheme 7. Resolution of rac-34 to provide $1 S, 5 R, 9 R-(-)-34$ and $1 R, 5 S, 9 S-(+)-34$.

With the free bases in hand, a way to quantitatively confirm the enantiomeric excess was needed, and several NMR chiral solvating reagents were studied for that purpose (Figure 2).


2,2,2-Trifluoro-1-phenylethan-1-ol


1-(Anthracen-9-yl)-
2,2,2-trifluoroethan-1-ol


3-Methyl-5-nitro-N-(1phenylethyl)benzamide


2-Amino-2-phenylethan-1-ol


2-Methoxy-
2-phenylacetic acid


Diphenyl(pyrrolidin-2-yl)methanol

Figure 2. NMR chiral shift reagents.
The only reagent that was found to be successful for the separation of the enantiomers of 34 so that they could be clearly observed in their NMR spectra was 1-(anthracen-9-yl)-2,2,2-trifluoroethan-1-ol. The two peaks that were most resolved in the ${ }^{1} \mathrm{H}-\mathrm{NMR}$ spectra and that could be used for determining enantiomeric excess were the methyl doublet and the farthest down-field aryl signal. After a single recrystallization, an enantiomeric excess of $>98 \%$ was achieved for both enantiomers (see Supplementary Materials for NMR spectra). With a successful resolution method, we needed to determine the absolute configuration of the two enantiomers. Although tartaric acid resolved rac-34 well, it did not provide a proper crystalline form for X-ray spectroscopic determination. A small amount of each enantiomer base was dissolved in acetone and seven drops of concentrated hydrobromic acid added. As the acetone evaporated under ambient pressure, colorless needles formed. The crystals of the HBr salt of (+)-34 were used for the X-ray crystallographic analysis (Figure 3). The crystal structure analysis confirmed both the overall structure and the absolute configuration for the enantiomeric compounds.



Figure 3. X-ray crystal structure of the HBr salt of $(2 R, 6 S, 11 S)-(+)-34 . \mathrm{HBr}$. The ellipsoids are shown at the $50 \%$ probability level. The crystal data, atomic coordinates, etc., have been cited in the Supplementary Materials.

With the enantiopure $(-)$ - and $(+)-34$ in hand, we pursued the synthesis of the C8hydroxy, methylene and methyl compounds.

The C8-hydroxy compounds were synthesized from the resolved $N$-phenethyl C-8 ketones $(-)$ - and (+)-34 using lithium aluminum hydride to give the phenolic ethers $(-)$ and (+)-35 (Scheme 8). O-Demethylation with $\mathrm{BBr}_{3}$ gave the desired phenols ( - )- and (+)-36.


Scheme 8. Synthesis of (-)- and (+)-C8-hydroxy benzomorphans.
The reaction of $(-)$ - and $(+)-34$ with methyl lithium (Scheme 9) gave the intermediate C8-hydroxymethyl compounds ( - )- and (+)-37 that could be converted to the phenolic C8-vinyl compounds (-)- and (+)-38 on treatment with $\mathrm{BBr}_{3}$. Catalytic reduction of the vinyl compounds gave the desired C8-methyl enantiomers ( - )- and (+)-39.



Scheme 9. Synthesis of C8-methylene ((-)-38 and (+)-38) and methyl ((-)-39 and (+)-39) compounds.

### 2.6. Forskolin-Induced cAMP Accumulation Assay for In Vitro Determination of the Potency and

 Efficacy of the BenzomorphansBoth electron-withdrawing (e.g., $\left.\mathrm{NO}_{2},(-)-3\right)$ and -donating (e.g., $\left.\mathrm{OH},(-)-20\right)$ substituents at the para position led to highly MOR potent compounds in vitro (Table 1) with $\mathrm{EC}_{50}$ values in the subnanomolar range $\mathrm{EC}_{50}=0.3 \mathrm{nM}$ and 0.13 nM , respectively. Some of the ortho-substituents (e.g., 2-F (-)-8) were also extremely potent MOR agonists $\left(\mathrm{EC}_{50}=0.2 \mathrm{nM}\right)$. The para-OH $(-)-20$ and the ortho- $\mathrm{F}((-)-8)$ compounds were among the few synthesized compounds that were more potent than the unsubstituted benzomorphan $(-)-23\left(\mathrm{EC}_{50}=0.27 \mathrm{nM}\right)$. The meta-substituted compounds were in general much less potent than the comparable ortho- and para-analogs, except for the meta-fluoro analog ( - )-7 $\left(\mathrm{EC}_{50}=0.6 \mathrm{nM}\right)$ that was more potent than the para-fluoro analog $(-)-6\left(\mathrm{EC}_{50}=1.8 \mathrm{nM}\right)$. Addition of a second chloro substituent 2,4-dichloro, $21\left(\mathrm{EC}_{50}=5 \mathrm{nM}\right)$ or 2,6-dichloro, 22 $\left(\mathrm{EC}_{50}=4 \mathrm{nM}\right)$ on the aromatic ring reduced the potency of the ortho $\left(17, \mathrm{EC}_{50}=1.4 \mathrm{nM}\right)$ or para ( $15, \mathrm{EC}_{50}=1.3 \mathrm{nM}$ ) monochloro-compound. No clear pattern between agonist potency and electron donating or withdrawal effects was observed. A steric effect due
to bulky substituents was also not observed (e.g., the bulky para- $\operatorname{Br}\left(12, \mathrm{EC}_{50}=0.7 \mathrm{nM}\right)$ and the less bulky para-fluoro substituent $\left(6, \mathrm{EC}_{50}=1.8 \mathrm{nM}\right)$ ). A second example of that could be seen with the bulky para-nitro (3) substituted compound having subnanomolar MOR affinity $\left(\mathrm{EC}_{50}=0.3 \mathrm{nM}\right)$. However, the bulky trifluoromethyl at the para position was less potent $\left(9, \mathrm{EC}_{50}=2 \mathrm{nM}\right)$ than compounds with bromo, chloro, or nitro substituents in the para position on the aromatic ring, and the dichloro compounds (21 and 22) were also less potent than the para-monochloro compound 15. A steric effect that might influence potency at MOR was not clearly observed. Three compounds showed an increase in DOR potency, compared with the unsubstituted $N$-phenethylnormetazocine (23). The para-nitro, para-trifluoromethyl, para-bromo, and para-chloro substituted compounds (3,9,12, and $\mathbf{1 5}$, respectively) were several-fold more potent than the unsubstituted compound 23 , and these substituents did not modify efficacy-they were all partial DOR agonists. All of the examined compounds were less potent at KOR than the unsubstituted compound 23. However, 23 had very low efficacy at KOR ( $\mathrm{EC}_{50}=3.4 \mathrm{nM}, \% \mathrm{E}_{\max }=25 \%$ ), and most of the compounds with substituents in the aromatic ring in the N -phenethyl side chain had greater efficacy and would, theoretically, be more likely to show the dysphoric and hallucinogenic side-effects known to occur with KOR agonists than the unsubstituted compound 23.

The N-phenethyl side-chain-modified benzomorphans (-)-24, (-)-25, (-)-28 and (-)31 (Table 1) were found to have little or no potency in vitro at MOR. The indole (-)-26, however, was fully efficacious and had moderate MOR agonist activity ( $\mathrm{EC}_{50}=12 \mathrm{nM}$, $\left.\% \mathrm{E}_{\max }=102 \%\right)$. Apparently, the addition of a single methylene in the side chain of the $N$-phenethyl moiety, to give an $N$-phenylpropyl derivative ( - )-24, changed the highly MOR potent $N$-phenethyl compound ( - )-23 ( $\mathrm{EC}_{50}=0.27, \% \mathrm{E}_{\max }=101 \%$ ) into a relatively inactive compound $(-)-24\left(\mathrm{EC}_{50}=69 \mathrm{nM}\right)$. Introduction of a double-bond in the side chain to give the cinnamyl derivative $(-)-25$, decreased the potency even more $\left(\mathrm{EC}_{50}=244 \mathrm{nM}\right)$. The addition of a para-nitro moiety to the $N$-phenylpropyl side chain gave a totally inactive compound (-)-31 (MOR agonist $\mathrm{EC}_{50} \geq 10,000 \mathrm{nM}$ ), in contrast with the para-nitro $N$ phenethyl compound (-)-3 with a MOR $\mathrm{EC}_{50}=0.3 \mathrm{nM}$. The saturated 1-methylpiperidine substituent (-)-28 was also inactive at MOR, but showed high agonist potency at KOR, although with very little efficacy $\left(\mathrm{EC}_{50}=0.05 \mathrm{nM}, \% \mathrm{E}_{\max }=15 \%\right)$.

All the C8 compounds were fully efficacious. As expected, the (+)-C8-compounds $(+)-36,(+)-38$, and (+)-39 were much less potent than the levorotatory compounds ( - )-38 and (-)-39 at MOR. This was expected, since dextrorotatory benzomorphans are usually much less active than their levorotatory relatives. Two (+)-compounds, the C8-hydroxy and methyl derivatives $(+)-36$ and (+)-39, had moderate MOR potency $\left(\mathrm{EC}_{50}=25.2\right.$ and 21.7 nM , respectively) and the (+)-methylene compound (+)-38 was considerably less potent $\left(E C_{50}=70 \mathrm{nM}\right)$. The methylene compound $(-)-38$ was found to have MOR subnanomolar potency $\left(\mathrm{EC}_{50}=0.23 \mathrm{nM}\right)$; it was about 25 times more potent than morphine. The C8-methyl compound (-)-39 was also quite potent, with an $\mathrm{EC}_{50}=1.2 \mathrm{nM}$. Both the C8-methylene $((-)-38)$ and methyl ((-)-39) compounds had moderate potency at DOR, and neither was as potent as the unsubstituted compound 23 at DOR or KOR. Both compounds had nearly full efficacy at DOR, and both had little efficacy at KOR.

Table 1. Opioid Receptor Activity Measured in the Forskolin-induced cAMP Accumulation Assay a ${ }^{\text {a }}$

|  |  |  | MOR | MOR | DOR | DOR | KOR | KOR |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | Agonist | Antagonist ${ }^{\text {b }}$ | Agonist | Antagonist ${ }^{\text {c }}$ | Agonist | Antagonist ${ }^{\text {d }}$ |
| Compound Number (3-31 (-)-Enantiomers) |  | Structure | $\begin{gathered} \mathrm{EC}_{50} \pm \text { SEM (nM) } \\ \left(\% \mathrm{E}_{\max } \pm \mathbf{S E M}\right) \end{gathered}$ | $\begin{gathered} \mathrm{IC}_{50} \pm \text { SEM (nM) } \\ \left(\% \mathrm{I}_{\max } \pm \mathbf{S E M}\right) \end{gathered}$ | $\begin{gathered} \mathrm{EC}_{50} \pm \text { SEM (nM) } \\ \left(\% \mathrm{E}_{\max } \pm \mathrm{SEM}\right) \end{gathered}$ | $\begin{gathered} \mathrm{IC}_{50} \pm \mathbf{S E M}(\mathrm{nM}) \\ \left(\% \mathrm{I}_{\max } \pm \mathbf{S E M}\right) \end{gathered}$ | $\begin{gathered} \mathrm{EC}_{50} \pm \mathbf{S E M}(\mathrm{nM}) \\ \left(\% \mathrm{E}_{\max } \pm \mathbf{S E M}\right) \end{gathered}$ | $\begin{gathered} \mathrm{IC}_{50} \pm \mathrm{SEM}(\mathrm{nM}) \\ \left(\% \mathrm{I}_{\max } \pm \mathrm{SEM}\right) \end{gathered}$ |
|  |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |
| 3 | 02-0011 | $4-\mathrm{NO}_{2}$ | $\begin{gathered} 0.3 \pm 0.05 \\ (101 \pm 1 \%) \end{gathered}$ | N/D | $\begin{gathered} 2.7 \pm 1.0 \\ (88 \pm 5 \%) \end{gathered}$ | N/D | $\begin{gathered} 13 \pm 0.5 \\ (46 \pm 7 \%) \end{gathered}$ | N/D |
| 4 | 02-0015 | $3-\mathrm{NO}_{2}$ | $\begin{gathered} 26 \pm 4.0 \\ (71 \pm 12 \%) \end{gathered}$ | N/D | N/D | N/D | N/D | N/D |
| 5 | 02-0018 | 2- $\mathrm{NO}_{2}$ | $\begin{gathered} 2.7 \pm 0.4 \\ (101 \pm 0.2 \%) \end{gathered}$ | N/D | $\begin{gathered} 37 \pm 6 \\ (73 \pm 14 \%) \end{gathered}$ | N/D | $\begin{gathered} 16 \pm 4 \\ (23 \pm 8 \%) \end{gathered}$ | $\begin{aligned} & 717 \pm 132 \\ & (89 \pm 5 \%) \end{aligned}$ |
| 6 | 02-0006 | 4-F | $\begin{gathered} 1.8 \pm 0.3 \\ (103 \pm 0.2 \%) \\ \hline \end{gathered}$ | N/D | $\begin{gathered} 12 \pm 3 \\ (73 \pm 11 \%) \end{gathered}$ | N/D | $\begin{gathered} 7 \pm 3 \\ (34 \pm 6 \%) \end{gathered}$ | $\begin{gathered} 368 \pm 92 \\ (71 \pm 2 \%) \end{gathered}$ |
| 7 | 02-0023 | 3-F | $\begin{gathered} 0.6 \pm 0.1 \\ (100 \pm 1 \%) \end{gathered}$ | N/D | $\begin{gathered} 11 \pm 4 \\ (86 \pm 5 \%) \end{gathered}$ | N/D | >10,000 | $\begin{gathered} 274 \pm 83 \\ (87 \pm 2 \%) \end{gathered}$ |
| 8 | 02-0020 | 2-F | $\begin{gathered} 0.2 \pm 0.1 \\ (101 \pm 1 \%) \end{gathered}$ | N/D | $\begin{gathered} 6 \pm 2 \\ (87 \pm 5 \%) \end{gathered}$ | N/D | $\begin{gathered} 6 \pm 1 \\ (45 \pm 7 \%) \end{gathered}$ | N/D |
| 9 | 02-0017 | $4-\mathrm{CF}_{3}$ | $\begin{aligned} 2 & \pm 1 \\ (102 & \pm 1 \%) \end{aligned}$ | N/D | $\begin{gathered} 3 \pm 1 \\ (72 \pm 14 \%) \end{gathered}$ | N/D | $\begin{gathered} 23 \pm 5 \\ (100 \pm 2 \%) \end{gathered}$ | N/D |
| 10 | 02-0024 | $3-\mathrm{CF}_{3}$ | $\begin{gathered} 167 \pm 72 \\ (63 \pm 13 \%) \end{gathered}$ | N/D | N/D | N/D | N/D | N/D |
| 11 | 02-0008 | $2-\mathrm{CF}_{3}$ | $\begin{gathered} 6 \pm 0.2 \\ (102 \pm 1 \%) \end{gathered}$ | N/D | $\begin{gathered} 24 \pm 5 \\ (77 \pm 9 \%) \end{gathered}$ | N/D | $\begin{gathered} 36 \pm 3 \\ (72 \pm 6 \%) \end{gathered}$ | N/D |

Table 1. Cont.

|  |  |  | $\begin{gathered} \text { MOR } \\ \hline \text { Agonist } \end{gathered}$ | MOR ${ }^{\text {Antagonist }}{ }^{\text {b }}$ | DOR <br> Agonist | $\begin{gathered} \hline \text { DOR } \\ \hline \text { Antagonist }^{\mathrm{c}} \end{gathered}$ | $\begin{gathered} \hline \text { KOR } \\ \hline \text { Agonist } \end{gathered}$ | KORAntagonist $^{\mathrm{d}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |  |  |  |
|  | nber mers) | Structure | $\begin{gathered} \mathrm{EC}_{50} \pm \mathrm{SEM}(\mathrm{nM}) \\ \left(\% \mathrm{E}_{\max } \pm \mathrm{SEM}\right) \end{gathered}$ | $\begin{aligned} & \mathrm{IC}_{50} \pm \text { SEM (nM) } \\ & \left(\% \mathrm{I}_{\max } \pm \mathbf{S E M}\right) \end{aligned}$ | $\begin{gathered} \mathrm{EC}_{50} \pm \text { SEM (nM) } \\ \left(\% \mathrm{E}_{\max } \pm \mathrm{SEM}\right) \end{gathered}$ | $\begin{aligned} & \mathrm{IC}_{50} \pm \mathbf{S E M}(\mathrm{nM}) \\ & \left(\% \mathrm{I}_{\max } \pm \mathbf{S E M}\right) \end{aligned}$ | $\begin{gathered} \mathrm{EC}_{50} \pm \text { SEM (nM) } \\ \left(\% \mathrm{E}_{\max } \pm \mathrm{SEM}\right) \end{gathered}$ | $\begin{aligned} & \mathrm{IC}_{50} \pm \text { SEM (nM) } \\ & \left(\% \mathbf{I}_{\max } \pm \mathbf{S E M}\right) \end{aligned}$ |
| 12 | 02-0010 | $4-\mathrm{Br}$ | $\begin{gathered} 0.7 \pm 0.2 \\ (102 \pm 1 \%) \end{gathered}$ | N/D | $\begin{aligned} & 1.1 \pm 0.3 \\ & (89 \pm 4 \%) \end{aligned}$ | N/D | $\begin{gathered} 5 \pm 2 \\ (100 \pm 1 \%) \end{gathered}$ | N/D |
| 13 | 02-0014 | $3-\mathrm{Br}$ | $\begin{gathered} 18 \pm 7 \\ (78 \pm 9 \%) \end{gathered}$ | N/D | $\begin{gathered} 110 \pm 24 \\ (51 \pm 4 \%) \end{gathered}$ | N/D | >10,000 | $\begin{gathered} 90 \pm 37 \\ (99 \pm 2 \%) \end{gathered}$ |
| 14 | 02-0021 | $2-\mathrm{Br}$ | $\begin{gathered} 2 \pm 0.4 \\ (101 \pm 2 \%) \end{gathered}$ | N/D | $\begin{gathered} 13 \pm 2 \\ (89 \pm 6 \%) \end{gathered}$ | N/D | $\begin{gathered} 16 \pm 2 \\ (80 \pm 6 \%) \end{gathered}$ | N/D |
| 15 | 02-0009 | $4-\mathrm{Cl}$ | $\begin{gathered} 1.3 \pm 0.4 \\ (103 \pm 1 \%) \end{gathered}$ | N/D | $\begin{gathered} 1.2 \pm 0.5 \\ (89 \pm 6 \%) \end{gathered}$ | N/D | $\begin{gathered} 5 \pm 1 \\ (97 \pm 1 \%) \end{gathered}$ | N/D |
| 16 | 02-0012 | $3-\mathrm{Cl}$ | $\begin{gathered} 13 \pm 2 \\ (95 \pm 2 \%) \end{gathered}$ | N/D | $\begin{gathered} 93 \pm 25 \\ (64 \pm 6 \%) \end{gathered}$ | N/D | >10,000 | $\begin{gathered} 111 \pm 43 \\ (108 \pm 10 \%) \end{gathered}$ |
| 17 | 02-0022 | $2-\mathrm{Cl}$ | $\begin{gathered} 1.4 \pm 0.3 \\ (101 \pm 1 \%) \\ \hline \end{gathered}$ | N/D | $\begin{gathered} 15 \pm 4.0 \\ (87 \pm 5 \%) \end{gathered}$ | N/D | $\begin{gathered} 8 \pm 1 \\ (57 \pm 11 \%) \end{gathered}$ | N/D |
| 18 | 02-0007 | 4-OMe | $\begin{gathered} \hline 1.2 \pm 0.4 \\ (103 \pm 0.3 \%) \end{gathered}$ | N/D | $\begin{gathered} 4 \pm 1 \\ (87 \pm 6 \%) \end{gathered}$ | N/D | $\begin{gathered} 9 \pm 2 \\ (89 \pm 3 \%) \end{gathered}$ | N/D |
| 19 | 02-0005 | 4-Me | $\begin{gathered} 1.4 \pm 0.4 \\ (103 \pm 1 \%) \end{gathered}$ | N/D | $\begin{gathered} 5 \pm 2 \\ (87 \pm 5 \%) \end{gathered}$ | N/D | $\begin{gathered} 13 \pm 3 \\ (93 \pm 3 \%) \end{gathered}$ | N/D |
| 20 | 02-0019 | 4-OH | $\begin{gathered} 0.13 \pm 0.02 \\ (102 \pm 0.4 \%) \end{gathered}$ | N/D | $\begin{gathered} 5 \pm 1 \\ (84 \pm 5 \%) \end{gathered}$ | N/D | $\begin{gathered} 8 \pm 3 \\ (97 \pm 1 \%) \end{gathered}$ | N/D |
| 21 | 02-0025 | 2,4-dichloro | $\begin{gathered} 5 \pm 3 \\ (103 \pm 1 \%) \end{gathered}$ | N/D | $\begin{gathered} 37 \pm 9 \\ (86 \pm 12 \%) \end{gathered}$ | N/D | $\begin{gathered} 42 \pm 10 \\ (93 \pm 5 \%) \end{gathered}$ | N/D |
| 22 | 02-0026 | 2,6-dichloro | $\begin{gathered} 4 \pm 1 \\ (99 \pm 2 \%) \end{gathered}$ | N/D | $\begin{gathered} 59 \pm 25 \\ (78 \pm 11 \%) \end{gathered}$ | N/D | >10,000 | $\begin{gathered} 65 \pm 24 \\ (87 \pm 7 \%) \end{gathered}$ |
| 23 | 02-013 | H | $\begin{gathered} 0.27 \pm 0.1 \\ (101 \pm 0.3 \%) \end{gathered}$ | N/D | $\begin{gathered} 4 \pm 1 \\ (85 \pm 6 \%) \end{gathered}$ | N/D | $\begin{gathered} 3.4 \pm 0.4 \\ (25 \pm 5 \%) \end{gathered}$ | $\begin{gathered} 281 \pm 94 \\ (81 \pm 5 \%) \end{gathered}$ |

Table 1. Cont.

|  |  | MOR | MOR | DOR | DOR | KOR | KOR |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | Agonist | Antagonist ${ }^{\text {b }}$ | Agonist | Antagonist ${ }^{\text {c }}$ | Agonist | Antagonist ${ }^{\text {d }}$ |
| Compound Number (3-31 (-)-Enantiomers) | Structure | $\begin{gathered} \mathrm{EC}_{50} \pm \text { SEM (nM) } \\ \left(\% \mathrm{E}_{\text {max }} \pm\right. \text { SEM) } \end{gathered}$ | $\begin{gathered} \mathrm{IC}_{50} \pm \text { SEM (nM) } \\ \left(\% \mathrm{I}_{\mathrm{max}} \pm \mathbf{S E M}\right) \end{gathered}$ | $\begin{gathered} \mathrm{EC}_{50} \pm \text { SEM (nM) } \\ \left(\% \mathrm{E}_{\max } \pm\right. \text { SEM) } \end{gathered}$ | $\begin{gathered} \mathrm{IC}_{50} \pm \text { SEM (nM) } \\ \left(\% \mathrm{I}_{\text {max }} \pm \mathbf{S E M}\right) \end{gathered}$ | $\begin{gathered} \mathrm{EC}_{50} \pm \text { SEM (nM) } \\ \left(\% \mathrm{E}_{\max } \pm\right. \text { SEM) } \end{gathered}$ | $\begin{gathered} \mathrm{IC}_{50} \pm \text { SEM (nM) } \\ \left(\% \mathrm{I}_{\max } \pm \mathbf{S E M}\right) \end{gathered}$ |
|  |  $N-\mathrm{R}_{1}=$ |  |  |  |  |  |  |
| 24 02-0027 |  | $\left.\begin{array}{c} 69 \pm 13 \\ (100 \end{array} \pm 1 \%\right)$ | N/D | N/D | N/D | N/D | N/D |
| 25 02-0029 |  | $\begin{gathered} 244 \pm 58 \\ (101 \pm 1 \%) \end{gathered}$ | N/D | N/D | N/D | N/D | N/D |
| 26 02-0034 |  | $\begin{gathered} 12 \pm 1 \\ (102 \pm 1 \%) \end{gathered}$ | N/D | $\begin{gathered} 343 \pm 35 \\ (91 \pm 5 \%) \end{gathered}$ | N/D | >10,000 | >10,000 |
| 28 02-0057 |  | >10,000 | $>10,000$ | >10,000 | >10,000 | $\begin{gathered} 0.05 \pm 0.02 \\ (15 \pm 3 \%) \end{gathered}$ | >10,000 |
| 31 02-0033 |  | >10,000 | >10,000 | N/D | N/D | N/D | N/D |
|  | C8-Substituted |  |  |  |  |  |  |
| $(+)-36$ GW-S-01-73 |  | $\begin{gathered} 25.2 \pm 5.9 \\ (103 \pm 0.3 \%) \end{gathered}$ | N/D | $\begin{gathered} 1906 \pm 232 \\ (97 \pm 4.2 \%) \end{gathered}$ | N/D | >10,000 | N/D |

Table 1. Cont.


Table 1. Cont.

|  |  | MOR | MOR | DOR | DOR | KOR | KOR |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | Agonist | Antagonist ${ }^{\text {b }}$ | Agonist | Antagonist ${ }^{\text {c }}$ | Agonist | Antagonist ${ }^{\text {d }}$ |
| Compound Number (3-31 (-)-Enantiomers) | Structure | $\begin{gathered} \mathrm{EC}_{50} \pm \text { SEM (nM) } \\ \left(\% \mathrm{E}_{\max } \pm \mathbf{S E M}\right) \end{gathered}$ | $\begin{gathered} \mathrm{IC}_{50} \pm \text { SEM (nM) } \\ \left(\% \mathrm{I}_{\max } \pm \mathbf{S E M}\right) \end{gathered}$ | $\begin{gathered} \mathrm{EC}_{50} \pm \text { SEM (nM) } \\ \left(\% \mathrm{E}_{\max } \pm \mathbf{S E M}\right) \end{gathered}$ | $\begin{gathered} \mathrm{IC}_{50} \pm \mathbf{S E M}(\mathrm{nM}) \\ \left(\% \mathbf{I}_{\max } \pm \mathbf{S E M}\right) \end{gathered}$ | $\begin{gathered} \mathrm{EC}_{50} \pm \text { SEM (nM) } \\ \left(\% \mathrm{E}_{\max } \pm \mathbf{S E M}\right) \end{gathered}$ | $\begin{gathered} \mathrm{IC}_{50} \pm \mathbf{S E M}(\mathrm{nM}) \\ \left(\% \mathrm{I}_{\max } \pm \mathbf{S E M}\right) \end{gathered}$ |
|  | Naltrexone | $\begin{gathered} 2.1 \pm 1.2 \\ (30 \pm 6 \%) \end{gathered}$ | $\begin{gathered} 11 \pm 1 \\ (104 \pm 1 \%) \end{gathered}$ | >10,000 | $\begin{gathered} 295 \pm 48 \\ (99 \pm 1 \%) \end{gathered}$ | $\begin{gathered} 0.6 \pm 0.3 \\ (57 \pm 7 \%) \end{gathered}$ | $\begin{gathered} 6 \pm 1 \\ (41 \pm 7 \%) \end{gathered}$ |
|  | nor-BNI |  |  |  |  |  | $\begin{gathered} 2.3 \pm 0.3 \\ (102 \pm 1 \%) \end{gathered}$ |

${ }^{\text {a }}$ Inhibition of forskolin-induced cAMP accumulation; cAMP Hunter ${ }^{\mathrm{TM}}$ Chinese hamster ovary cells (CHO-K1) that express human $\mu$-opioid receptor (OPRM1), human k -opioid receptor (OPRK1), and human $\delta$-opioid receptor (OPRD1) were used for the forskolin-induced cAMP accumulation assay to determine potency and efficacy of the compounds following the previously established methods [28]; to determine \% efficacy in forskolin-induced cAMP assays, data were blank subtracted with the vehicle control, followed by normalization to the forskolin control. Data were then analyzed in GraphPad Prism 8 (GraphPad, LaJolla, CA, USA) using nonlinear regression; values are expressed as the mean $\pm$ SEM of at least three independent experiments; N/D = not determined (MOR antagonist activity was not determined when a compound had potent, fully efficacious MOR agonist activity; DOR and KOR agonist activity were not determined if the MOR agonist activity was less than 20 nM ). ${ }^{\mathrm{b}}$ MOR antagonist potency ( $\mathrm{IC}_{50}$ ) determined versus $\mathrm{EC}_{90}$ of fentanyl; degree of antagonism ( $\mathrm{I}_{\text {max }}$ )
 determined versus $\mathrm{EC}_{90}$ of U 50488 H ; degree of antagonism (Imax) normalized to nor-BNI.

## 3. Materials and Methods

### 3.1. General Information

All reactions were performed in oven-dried glassware under an argon atmosphere unless otherwise noted. Proton ( ${ }^{1} \mathrm{H}$ NMR) and carbon ( ${ }^{13} \mathrm{C}$ NMR) spectra were recorded on a Varian Gemini-400 spectrometer at 400 MHz for ${ }^{1} \mathrm{H}$ NMR and 101 MHz for ${ }^{13} \mathrm{C}$ NMR. Mass spectra (HRMS) were recorded on a Waters (Mitford, MA, USA) Xevo-G X5 QTof. The optical rotation data were obtained on a PerkinElmer polarimeter model 341, and melting points were obtained using a Thomas-Hoover melting point apparatus. Thin layer chromatography (TLC) was performed on a 250 mm Analtech GHLF. Visualization was accomplished under UV or by staining in an iodine chamber. Flash column chromatography was performed with Fluka silica gel 60 (mesh 220-400). Elemental analyses were performed by Robertson Microlit Laboratories, Ledgewood, NJ, USA.

### 3.2. Synthesis

Optical resolution of rac-normetazocine (rac-1). rac-Normetazocine was optically resolved using a procedure in the literature [23] to obtain (-)-normetazocine ( - )-1. racNormetazocine (rac-1, 10.6 g , 1 equiv, 48.8 mmol ) was suspended in $\mathrm{H}_{2} \mathrm{O}(56 \mathrm{~mL})$. Then, $37 \% \mathrm{HCl}(1.78 \mathrm{~g}, 4.06 \mathrm{~mL}, 12$ molar, 1 equiv, 48.8 mmol$)$ was added to the solution, followed by (+)-3-bromo-8-camphorsulfonic acid ammonium salt ( $16.0 \mathrm{~g}, 1$ equiv, 48.8 mmol ) and 1,3-dioxolane ( 5.6 mL ). The solution was heated to $100^{\circ} \mathrm{C}$ until all of the solids dissolved. The solution was allowed to cool to room temperature and stirred for 1 h . The crystals that formed were filtered and washed with $10 \%$ 1,3-dioxolane in $\mathrm{H}_{2} \mathrm{O}$. The salt was recrystallized $2 x$ from $12 \%$ 1,3-dioxolane in $\mathrm{H}_{2} \mathrm{O}$ and dried under high vacuum at $60^{\circ} \mathrm{C}$ overnight. NMR studies using (R)-(+)-methylbenzyl isocyanate in $\mathrm{CDCl}_{3}: \mathrm{CD}_{3} \mathrm{OD}(1: 1)$ indicated $>99 \%$ optical purity.

General procedure for N -alkylation of (-)-normetazocine. Optically resolved (-)2•bromocamphorsulfonate salt ( 1 equiv, 1.89 mmol ), sodium bicarbonate ( 3 equiv, 3.69 mmol ) and the alkyl bromide ( 1.5 equiv, 1.84 mmol ) were combined in anhydrous DMF ( 15 mL ). The reaction was stirred at $100^{\circ} \mathrm{C}$ for 20 h . The reaction was cooled to room temperature, and $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ was added. The mixture was transferred to a separatory funnel and extracted with $\mathrm{CHCl}_{3}(3 \times 20 \mathrm{~mL})$. The organic layers were combined and washed with $\mathrm{H}_{2} \mathrm{O}(5 \times 10 \mathrm{~mL})$ and brine $(20 \mathrm{~mL})$. The organic layer was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and filtered. The solvent was removed, and the residual material was purified by flash chromatography (gradient $1-10 \% \mathrm{MeOH} / \mathrm{NH}_{4} \mathrm{OH}$ in $\mathrm{CHCl}_{3}$ ).
(1R,5R,9R)-6,11-Dimethyl-3-(4-nitrophenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin8 -ol $((-)-3)$. General procedure was used, and the alkylation was achieved with 4-nitrophen ethyl bromide. The product was obtained as a pale-yellow solid ( $300 \mathrm{mg}, 67 \%$ ), $\mathrm{mp}(\mathrm{HCl}$ salt) $273-274{ }^{\circ} \mathrm{C}(\mathrm{dec}) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.43-7.40(\mathrm{~m}, 2 \mathrm{H}), 7.17-7.14(\mathrm{~m}$, $2 \mathrm{H}), 6.89(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.69(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.56(\mathrm{dd}, J=8.2,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.02$ (dd, $J=5.5,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.92(\mathrm{~d}, J=18.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.80-2.74(\mathrm{~m}, 3 \mathrm{H}), 2.72-2.61(\mathrm{~m}, 3 \mathrm{H}), 2.13(\mathrm{td}$, $J=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.88(\mathrm{dd}, J=7.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.81(\mathrm{td}, J=12.9,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.34(\mathrm{~s}$, $3 \mathrm{H}), 1.32(\mathrm{t}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.87(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.101 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right): \delta 155.4$, $148.4,146.5,142.2,129.4,127.7,126.7,123.0,112.7,111.6,57.6,55.7,45.5,41.5,41.2,35.9,33.3$, 24.4, 22.8, 13.0. HRMS (ES+) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{~N}_{2} \mathrm{O}_{3}, 367.2022(\mathrm{M}+\mathrm{H})^{+}$; found, 367.2016. $[\alpha]^{20}{ }_{\mathrm{D}}-94.8^{\circ}$ (c 1.17, $\mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)$ ). Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{ClN}_{2} \mathrm{O}_{3} \bullet 0.15 \mathrm{C}_{3} \mathrm{H}_{8} \mathrm{O}$ : C, 65.46; H, 6.90; N, 6.80. Found: C, 65.60; H, 6.94; N, 6.66.
(1R,5R,9R)-6,11-Dimethyl-3-(3-nitrophenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin8 -ol $((-)-4)$. General procedure was used, and the alkylation was achieved with 3-nitrophen ethyl bromide. The product was obtained as a pale-yellow solid $(276 \mathrm{mg}, 81 \%), \mathrm{mp}(\mathrm{HCl}$ salt) $259-262{ }^{\circ} \mathrm{C}$ (dec). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz ; CD3OD): $\delta 8.16$ (t, $\left.J=1.8 \mathrm{~Hz}, 1 \mathrm{H}\right), 8.10-8.07$ $(\mathrm{m}, 1 \mathrm{H}), 7.67(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.54(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.91(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.70(\mathrm{~d}$, $J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.57(\mathrm{dd}, J=8.2,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.04(\mathrm{dd}, J=5.5,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.95(\mathrm{dt}, J=10.4$,
$4.2 \mathrm{~Hz}, 3 \mathrm{H}), 2.89-2.81(\mathrm{~m}, 1 \mathrm{H}), 2.78-2.64(\mathrm{~m}, 3 \mathrm{H}), 2.18(\mathrm{td}, J=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.89(\mathrm{dd}$, $J=7.1,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.83(\mathrm{td}, J=12.9,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.36(\mathrm{~s}, 3 \mathrm{H}), 1.32(\mathrm{~d}, J=20.7 \mathrm{~Hz}, 1 \mathrm{H}), 0.88$ $(\mathrm{d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; CD $\left.{ }_{3} \mathrm{OD}\right): ~ \delta 155.4,148.3,142.5,142.2,134.9,129.1$, $127.7,126.6,123.1,120.7,112.8,111.6,57.6,55.8,45.5,41.5,41.2,35.9,32.9,24.5,22.8,13.0$. HRMS (ES+) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{~N}_{2} \mathrm{O}_{3}, 367.2022(\mathrm{M}+\mathrm{H})^{+}$; found, $367.2022[\alpha]^{20}{ }_{\mathrm{D}}{ }^{-94.5^{\circ}(c)}$ 1.1, $\left.\mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)\right)$. Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{ClN}_{2} \mathrm{O}_{3} \bullet 0.25 \mathrm{C}_{3} \mathrm{H}_{8} \mathrm{O}: \mathrm{C}, 65.38 ; \mathrm{H}, 6.99$; N, 6.70. Found: C, 65.19; H, 6.80; N, 6.82.
(1R,5R,9R)-6,11-Dimethyl-3-(2-nitrophenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin8 -ol ((-)-5). General procedure was used, and the alkylation was achieved with 2-nitrophen ethyl bromide. The product was obtained as a pale-yellow solid ( $265 \mathrm{mg}, 76 \%$ ), $\mathrm{mp}(\mathrm{HCl}$ salt) $237-239{ }^{\circ} \mathrm{C}$ (dec). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.89$ (dd, $J=8.2,1.1 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.59 (td, $J=7.5,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.47$ (dd, $J=7.7,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.44-7.40(\mathrm{~m}, 1 \mathrm{H}), 6.88(\mathrm{~d}, J=8.3 \mathrm{~Hz}$, $1 \mathrm{H}), 6.68(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.55(\mathrm{dd}, J=8.2,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.13$ (ddd, $J=12.5,10.5,5.9 \mathrm{~Hz}$, $1 \mathrm{H}), 3.03-2.94(\mathrm{~m}, 2 \mathrm{H}), 2.91(\mathrm{~d}, J=18.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.86-2.79(\mathrm{~m}, 1 \mathrm{H}), 2.71-2.64(\mathrm{~m}, 2 \mathrm{H}), 2.59$ (dd, $J=11.9,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.14(\mathrm{td}, J=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.87(\mathrm{dd}, J=7.1,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.81$ $(\mathrm{td}, J=12.8,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.33(\mathrm{~s}, 3 \mathrm{H}), 1.29(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 0.84(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; CD ${ }_{3}$ OD): $\delta 155.3,149.7,142.3,134.7,132.7,132.3,127.7,127.2,126.8,124.1$, 112.7, 111.6, 57.7, 55.6, 45.6, 41.4, 41.0, 35.9, 30.5, 24.5, 23.1, 13.1. HRMS (ES+) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{~N}_{2} \mathrm{O}_{3}, 367.2022(\mathrm{M}+\mathrm{H})^{+}$; found, 367.2020. $[\alpha]^{20}{ }_{\mathrm{D}}-87.8^{\circ}\left(c 1.4, \mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)\right)$. Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{ClN}_{2} \mathrm{O}_{3} \bullet 0.3 \mathrm{H}_{2} \mathrm{O} \bullet 0.5 \mathrm{C}_{3} \mathrm{H}_{8} \mathrm{O}: \mathrm{C}, 64.39 ; \mathrm{H}, 7.27 ; \mathrm{N}, 6.39$. Found: C, 64.34; H, 7.32; N, 6.42.
(2R,6R,11R)-6,11-Dimethyl-3-(4-fluorophenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azoc in-8-ol ((-)-6). General procedure was used, and the alkylation was achieved with 4fluorophenethyl bromide. The product was obtained as a pale-yellow solid ( $500 \mathrm{mg}, 83 \%$ ), $\mathrm{mp}\left(\mathrm{HCl}\right.$ salt) $303-304{ }^{\circ} \mathrm{C}$ (dec). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right): ~ \delta 7.23-7.20(\mathrm{~m}, 2 \mathrm{H}), 7.00-6.95$ $(\mathrm{m}, 2 \mathrm{H}), 6.88(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.67(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.54(\mathrm{dd}, J=8.2,2.6 \mathrm{~Hz}, 1 \mathrm{H})$, $3.02(\mathrm{dd}, J=5.6,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.91(\mathrm{~d}, J=18.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.80-2.59(\mathrm{~m}, 6 \mathrm{H}), 2.12(\mathrm{td}, J=12.4$, $3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.87(\mathrm{dd}, J=7.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.80(\mathrm{td}, J=12.9,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.33(\mathrm{~s}, 3 \mathrm{H}), 1.30$ $(\mathrm{t}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 0.85(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; CD $\left.{ }_{3} \mathrm{OD}\right): \delta 160.2$ (2), 155.4, $142.2,135.90$ (2), 129.84 (2), 127.7, 126.6, 114.5 (2), 112.8, 111.6, 57.3, 56.6, 47.87, 47.83, 47.5, $45.6,41.4,41.1,35.9,32.5,24.5,22.5,13.1$. HRMS ( $\mathrm{ES}^{+}$) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{FNO}, 340.2077$ $(\mathrm{M}+\mathrm{H})^{+}$; found, $340.2075[\alpha]^{20}{ }_{\mathrm{D}}-109.7^{\circ}$ ( с $\left.1.0, \mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)\right)$. Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{ClFNO}: \mathrm{C}, 70.29 ; \mathrm{H}, 7.24 ; \mathrm{N}, 3.73$. Found: C, 70.55; H, 7.23; N, 3.71.
(1R,5R,9R)-6,11-Dimethyl-3-(3-fluorophenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin-8ol ((-)-7). General procedure was used, and the alkylation was achieved with 3-fluorophen ethyl bromide. The product was obtained as a pale-yellow solid ( $200 \mathrm{mg}, 64 \%$ ), $\mathrm{mp}(\mathrm{HCl}$ salt) $268-270{ }^{\circ} \mathrm{C}(\mathrm{dec}) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.26$ (td, $\left.J=7.9,6.1 \mathrm{~Hz}, 1 \mathrm{H}\right), 7.03(\mathrm{~d}$, $J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.97(\mathrm{dt}, J=10.1,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.92-6.88(\mathrm{~m}, 2 \mathrm{H}), 6.67(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.55$ (dd, $J=8.2,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.07(\mathrm{dd}, J=5.6,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.92(\mathrm{~d}, J=18.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.85-2.78$ $(\mathrm{m}, 3 \mathrm{H}), 2.75-2.63(\mathrm{~m}, 3 \mathrm{H}), 2.17(\mathrm{td}, J=12.5,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.89(\mathrm{tt}, J=7.0,3.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.81$ ( $\mathrm{td}, J=13.0,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.35(\mathrm{dd}, J=3.1,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 1.33(\mathrm{~s}, 3 \mathrm{H}), 0.86(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR (101 MHz; CD ${ }_{3} \mathrm{OD}$ ): $\delta 162.9$ (2), 155.5, 142.6 (2), 142.0, 129.7 (2), 127.7, 126.3, 124.1 (2), 115.0 (2), 112.9, 112.4 (2), 111.6, 57.6, 56.0, 45.6, 41.2, 40.9, 35.8, 32.9, 24.4, 22.6, 13.0. HRMS ( $\mathrm{ES}^{+}$) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{FNO},(\mathrm{M}+\mathrm{H})^{+} 340.2077$; found, $340.2071[\alpha]^{20}{ }_{\mathrm{D}}-99.1^{\circ}$ (c 0.97, $\left.\mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)\right)$. Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27}$ ClFNO: C, $70.13 ; \mathrm{H}, 7.25 ; \mathrm{N}, 3.72$. Found: C, 69.90; H, 6.96; N, 3.75.
(1R,5R,9R)-6,11-Dimethyl-3-(2-fluorophenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin-8ol ((-)-8). General procedure was used, and the alkylation was achieved with 2-fluorophene thyl bromide. The product was obtained as a pale-yellow solid ( $200 \mathrm{mg}, 62 \%$ ), mp ( HCl salt) $263-265{ }^{\circ} \mathrm{C}$ (dec). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.25$ (td, $J=7.6,1.6 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.19 (ddd, $J=7.6,5.5,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.09-7.05(\mathrm{~m}, 1 \mathrm{H}), 7.04-6.99(\mathrm{~m}, 1 \mathrm{H}), 6.87(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H})$, $6.67(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.54(\mathrm{dd}, J=8.2,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.05(\mathrm{dd}, J=5.5,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.93-2.84$
(m, 2H), 2.78 (ddt, $J=19.8,12.3,6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.70-2.67(\mathrm{~m}, 1 \mathrm{H}), 2.65-2.61(\mathrm{~m}, 2 \mathrm{H}), 2.13$ $(\mathrm{td}, J=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.88(\mathrm{dd}, J=7.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.81(\mathrm{td}, J=12.9,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.32$ ( $\mathrm{s}, 3 \mathrm{H}$ ), $1.30(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 0.85(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.101 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right): \delta$ 161.1 (2), 155.4, 142.1, 130.7 (2), 127.8 (2), 127.7, 126.5 (2), 126.4, 123.9 (2), 114.8, 114.6, 112.8, 111.6, 57.4, 54.9, 45.6, 41.3, 40.9, 35.8, 26.59, 26.57, 24.4, 22.6, 13.1. HRMS (ES ${ }^{+}$) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{FNO}, 340.2077(\mathrm{M}+\mathrm{H})^{+}$; found, $340.2082[\alpha]^{20} \mathrm{D}^{-107.1^{\circ}}$ (c 1.0, $\mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)$ ). Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{ClFNO}: ~ \mathrm{C}, 70.29 ; \mathrm{H}, 7.24 ; \mathrm{N}, 3.73$. Found: C, $70.13 ; \mathrm{H}, 6.94 ; \mathrm{N}, 3.75$.
(1R,5R,9R)-6,11-Dimethyl-3-(4-trifluoromethylphenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobe nzo[d]azocin-8-ol ((-)-9). General procedure was used, and the alkylation was achieved with 4-trifluoromethylphenethyl bromide. The product was obtained as a pale-yellow solid ( $260 \mathrm{mg}, 59 \%$ ) mp ( HCl salt) $283-285{ }^{\circ} \mathrm{C}$ (dec). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.58$ (d, $J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.43(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 6.90(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.70(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H})$, 6.57 (dd, $J=8.2,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.04(\mathrm{dd}, J=5.6,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.96-2.87(\mathrm{~m}, 3 \mathrm{H}), 2.86-2.77(\mathrm{~m}$, $1 \mathrm{H}), 2.74-2.62(\mathrm{~m}, 3 \mathrm{H}), 2.15(\mathrm{td}, J=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.90(\mathrm{td}, J=7.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.83$ (td, $J=12.8,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.35(\mathrm{~s}, 3 \mathrm{H}), 1.33(\mathrm{t}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.88(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, CD 3 OD): d 155.4, 144.8, 142.2, 129.0, 128.0 (4), 127.5, 126.6, 124.8 (4), 124.4 (4), $112.8,111.6,57.4,56.0,45.5,41.4,41.1,35.9,33.2,24.5,22.6,13.1$. HRMS (ES ${ }^{+}$) Calcd. For $\mathrm{C}_{23} \mathrm{H}_{27} \mathrm{NOF}_{3}, 390.2045(\mathrm{M}+\mathrm{H})^{+}$; found, $390.2043[\alpha]^{20}{ }_{\mathrm{D}}-87.5^{\circ}\left(c 1.0, \mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)\right.$ ). Anal. Calcd. For $\mathrm{C}_{23} \mathrm{H}_{27} \mathrm{ClF}_{3} \mathrm{NO} \bullet 0.05 \mathrm{C}_{3} \mathrm{H}_{8} \mathrm{O}: \mathrm{C}, 64.83 ; \mathrm{H}, 6.44 ; \mathrm{N}, 3.27$. Found: C, 64.84; H, 6.30; N, 3.10.
(1R,5R,9R)-6,11-Dimethyl-3-(3-trifluoromethylphenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanoben zo[d]azocin-8-ol ((-)-10). General procedure was used, and the alkylation was achieved with 3-trifluoromethylphenethyl bromide. The product was obtained as a pale-yellow solid (280 mg, 77\%), mp (HCl salt) $282.2-284.0^{\circ} \mathrm{C}$ (dec). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.53$ (s, $1 \mathrm{H}), 7.49-7.42(\mathrm{~m}, 3 \mathrm{H}), 6.87(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.67(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.54$ (dd, $J=8.2$, $2.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.00(\mathrm{dd}, J=5.3,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.92-2.84(\mathrm{~m}, 3 \mathrm{H}), 2.80-2.75(\mathrm{~m}, 1 \mathrm{H}), 2.73-2.58(\mathrm{~m}$, $3 \mathrm{H}), 2.13(\mathrm{td}, J=12.4,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 1.86(\mathrm{dd}, J=7.0,2.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.79(\mathrm{td}, J=12.9,4.7 \mathrm{~Hz}$, $1 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H}), 1.30(\mathrm{~s}, 1 \mathrm{H}), 0.84(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.101 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right): \delta 155.4$, $142.2,141.5,132.2,130.2$ (4), 128.8, 127.7, 126.7, 125.0 (4), 124.3 (4), 122.5 (4), 112.8, 111.6, $57.5,56.1,45.5,41.5,41.1,35.9,33.1,24.5,22.7,13.1$. HRMS (ES+) Calcd. For $\mathrm{C}_{23} \mathrm{H}_{27} \mathrm{~F}_{3} \mathrm{NO}$, $(\mathrm{M}+\mathrm{H})^{+} 390.2045$; found, $390.2040[\alpha]^{20}{ }_{\mathrm{D}}-91.0^{\circ}\left(c 1.2, \mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)\right)$. Anal. Calcd. For $\mathrm{C}_{23} \mathrm{H}_{27} \mathrm{ClF}_{3} \mathrm{NO}: \mathrm{C}, 64.86 ; \mathrm{H}, 6.39 ; \mathrm{N}, 3.29$. Found: C, 64.64; H, 6.12; N, 3.23.
(1R,5R,9R)-6,11-Dimethyl-3-(2-trifluoromethylphenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanoben zo[d]azocin-8-ol ((-)-11). General procedure was used, and the alkylation was achieved with 2-trifluoromethylphenethyl bromide. The product was obtained as a pale-yellow solid ( $100 \mathrm{mg}, 14 \%$ ), mp ( HCl salt) $215-217{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.63$ (d, $J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.53(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.44(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.35(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H})$, $6.87(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.67(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.54(\mathrm{dd}, J=8.2,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.02-2.94$ $(\mathrm{m}, 3 \mathrm{H}), 2.89(\mathrm{~d}, J=18.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.77(\mathrm{td}, J=11.6,5.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.70-2.61(\mathrm{~m}, 3 \mathrm{H}), 2.14$ ( $\mathrm{td}, J=12.4,3.1 \mathrm{~Hz}, 1 \mathrm{H}$ ), $1.89(\mathrm{dd}, J=7.0,2.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.82(\mathrm{td}, J=12.9,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.33$ $(\mathrm{s}, 3 \mathrm{H}), 1.29(\mathrm{~d}, J=18.1 \mathrm{~Hz}, 1 \mathrm{H}), 0.86(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; CD $\left.{ }_{3} \mathrm{OD}\right): \delta$ $155.4,142.2,138.5,131.9,131.6,128.2,127.7,126.6,126.5$ (4), 126.21, 125.4 (4), 112.8, 111.6, $57.7,56.5,48.2,47.8,46.9,45.6,41.4,41.0,35.9,30.3,24.5,22.8,13.1$. HRMS (ES ${ }^{+}$) Calcd. For $\mathrm{C}_{23} \mathrm{H}_{27} \mathrm{~F}_{3} \mathrm{NO}_{2}, 390.2045(\mathrm{M}+\mathrm{H})^{+}$; found, $390.2050[\alpha]^{20}{ }_{\mathrm{D}}-109.8^{\circ}\left(\right.$ c $1.0, \mathrm{CHCl}_{3} / \mathrm{MeOH}$ (9/1)). Anal. Calcd. For $\mathrm{C}_{23} \mathrm{H}_{27} \mathrm{ClF}_{3} \mathrm{NO}_{2} \bullet 0.6 \mathrm{C}_{3} \mathrm{H}_{8} \mathrm{O}: \mathrm{C}, 64.48 ; \mathrm{H}, 6.94 ; \mathrm{N}, 3.03$. Found: C, 64.51; H, 6.95; N, 3.03.
(1R,5R,9R)-6,11-Dimethyl-3-(4-bromophenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin-8ol ((-)-12). General procedure was used, and the alkylation was achieved with 4-bromophen ethyl bromide. The product was obtained as a pale-yellow solid ( $311 \mathrm{mg}, 63 \%$ ) , mp ( HCl salt) $292-293{ }^{\circ} \mathrm{C}(\mathrm{dec}) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.43-7.40(\mathrm{~m}, 2 \mathrm{H}), 7.17-7.14(\mathrm{~m}$, $2 \mathrm{H}), 6.89(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.69(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.56(\mathrm{dd}, J=8.2,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.02$ (dd, $J=5.5,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.92(\mathrm{~d}, J=18.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.80-2.74(\mathrm{~m}, 3 \mathrm{H}), 2.72-2.61(\mathrm{~m}, 3 \mathrm{H})$,
$2.13(\mathrm{td}, J=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.88(\mathrm{dd}, J=7.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.81(\mathrm{td}, J=12.9,4.7 \mathrm{~Hz}$, $1 \mathrm{H}), 1.34(\mathrm{~s}, 3 \mathrm{H}), 1.32(\mathrm{t}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.87(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; $\mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 155.4,142.2,139.3,131.1,130.3,127.7,126.6,119.4,112.8,111.6,57.4,56.2,45.6$, 41.4, 41.1, 35.9, 32.7, 24.4, 22.6, 13.1. HRMS (ES ${ }^{+}$) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{BrNO}, 400.1276$ $(\mathrm{M}+\mathrm{H})^{+}$; found, $400.1271[\alpha]^{20}{ }_{\mathrm{D}}-105.5^{\circ}\left(c 1.0, \mathrm{CHCl}_{3} / \mathrm{MeOH},(9 / 1)\right)$. Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{BrClNO} \bullet 0.45 \mathrm{C}_{3} \mathrm{H}_{8} \mathrm{O}: \mathrm{C}, 60.46 ; \mathrm{H}, 6.65 ; \mathrm{N}, 3.02$. Found: C, 60.40; H, 6.58; N, 2.96.
(1R,5R,9R)-6,11-Dimethyl-3-(3-bromophenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin-8ol ((-)-13). General procedure was used, and the alkylation was achieved with 3-bromophe nethyl bromide. The product was obtained as a pale-yellow solid ( $276 \mathrm{mg}, 81 \%$ ), mp ( HCl salt) $255-257{ }^{\circ} \mathrm{C}$ (dec). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.42(\mathrm{~s}, 1 \mathrm{H}), 7.34(\mathrm{dt}, J=6.5$, $2.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.21-7.19(\mathrm{~m}, 2 \mathrm{H}), 6.90(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.69(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.56(\mathrm{dd}$, $J=8.2,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.02(\mathrm{dd}, J=5.5,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.92(\mathrm{~d}, J=18.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.81-2.75(\mathrm{~m}$, $3 \mathrm{H}), 2.72-2.60(\mathrm{~m}, 3 \mathrm{H}), 2.14(\mathrm{td}, J=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.88(\mathrm{dd}, J=7.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.82$ $(\mathrm{dt}, J=12.9,6.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.34(\mathrm{~s}, 3 \mathrm{H}), 1.32(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 0.87(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; CD 3 OD): $\delta 155.4,142.8,142.2,131.3,129.8,128.8,127.7,127.2,126.7,121.9$, 112.8, 111.6, 57.4, 56.2, 45.6, 41.4, 41.1, 35.9, 33.0, 24.5, 22.6, 13.1. HRMS (ES+) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{NOBr}, 400.1276(\mathrm{M}+\mathrm{H})^{+}$; found, 400.1277. $[\alpha]^{20}{ }_{\mathrm{D}}-92.5^{\circ}\left(c 1.0, \mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)\right.$ ). Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{BrClNO}: ~ C, ~ 60.49 ; ~ H, ~ 6.23 ; ~ N, ~ 3.21 . ~ F o u n d: ~ C, ~ 60.67 ; ~ H, ~ 6.33 ; ~ N, ~ 3.16 . ~$
(1R,5R,9R)-6,11-Dimethyl-3-(2-bromophenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin-8ol ((-)-14). General procedure was used, and the alkylation was achieved with 2-bromophen ethyl bromide. The product was obtained as a pale-yellow solid ( $200 \mathrm{mg}, 59 \%$ ), mp ( HCl salt) $233-235{ }^{\circ} \mathrm{C}$ (dec). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.51$ (dd, $J=8.0,1.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.31-7.23(\mathrm{~m}, 2 \mathrm{H}), 7.10-7.06(\mathrm{~m}, 1 \mathrm{H}), 6.86(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.67(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.54$ (dd, $J=8.2,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.05(\mathrm{dd}, J=5.5,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.00-2.87(\mathrm{~m}, 3 \mathrm{H}), 2.76(\mathrm{dt}, J=11.8$, $5.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.71(\mathrm{t}, J=4.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.66-2.57(\mathrm{~m}, 2 \mathrm{H}), 2.15(\mathrm{td}, J=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.89(\mathrm{qd}$, $J=7.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.81(\mathrm{td}, J=12.9,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H}), 1.30(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 0.85$ (d, $J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; CD 3 OD$)$ : $\delta 155.4,142.1,139.1,132.4,130.7,127.85$, $127.71,127.5,126.6,123.8,112.8,111.6,57.5,54.7,45.7,41.3,40.9,35.9,33.6,24.5,22.9,13.1$. HRMS (ES ${ }^{+}$) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{BrNO}, 400.1276(\mathrm{M}+\mathrm{H})^{+}$; found, 400.1278. $[\alpha]^{20}{ }_{\mathrm{D}}-88.2^{\circ}(c$ $\left.0.85, \mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)\right)$. Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{BrClNO} \bullet 0.1 \mathrm{H}_{2} \mathrm{O} \bullet 0.2 \mathrm{C}_{3} \mathrm{H}_{8} \mathrm{O}: \mathrm{C}$, $60.24 ; \mathrm{H}, 6.44 ; \mathrm{N}, 3.11$. Found: C, 60.32; H, 6.37; N, 3.02.
(1R,5R,9R)-6,11-Dimethyl-3-(4-chlorophenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin-8ol ((-)-15). General procedure was used, and the alkylation was achieved with 4-chlorophen ethyl bromide. The product ((-)-15) was obtained as a pale-yellow solid ( $310 \mathrm{mg}, 71 \%$ ), mp (HCl salt) 300-302 ${ }^{\circ} \mathrm{C}$ (dec). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right): ~ \delta 7.26-7.23(\mathrm{~m}, 2 \mathrm{H}), 7.21-7.18$ $(\mathrm{m}, 2 \mathrm{H}), 6.87(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.67(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.54(\mathrm{dd}, J=8.2,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.01$ (dd, $J=5.6,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.91(\mathrm{~d}, J=18.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.80-2.72(\mathrm{~m}, 3 \mathrm{H}), 2.68(\mathrm{dd}, J=9.9,4.1 \mathrm{~Hz}$, $1 \mathrm{H}), 2.64-2.62(\mathrm{~m}, 1 \mathrm{H}), 2.61-2.58(\mathrm{~m}, 1 \mathrm{H}), 2.12(\mathrm{td}, J=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.86(\mathrm{dd}, J=7.0$, $3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.80(\mathrm{td}, J=12.9,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H}), 1.30(\mathrm{t}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.85(\mathrm{~d}, J=$ $7.0 \mathrm{~Hz}, 3 \mathrm{H}$ ). ${ }^{13} \mathrm{C}$ NMR (101 MHz; CD ${ }_{3} \mathrm{OD}$ ): $\delta 155.4,142.2,138.8,131.5,129.9,128.0,127.7$, $126.6,112.8,111.6,57.4,56.3,45.6,41.4,41.1,35.9,32.7,24.5,22.6,13.1$. HRMS $^{\left(E S^{+}\right.}$) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{ClNO}, 356.1781(\mathrm{M}+\mathrm{H})^{+}$; found, 356.1779. $[\alpha]^{20}{ }_{\mathrm{D}}-105.0^{\circ}\left(c 1.0, \mathrm{CHCl}_{3} / \mathrm{MeOH}\right.$ (9/1)). Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{Cl}_{2} \mathrm{NO}: \mathrm{C}, 67.35 ; \mathrm{H}, 6.94 ; \mathrm{N}, 3.57$. Found: C, 67.38; H, 6.69; N, 3.57.
(1R,5R,9R)-6,11-Dimethyl-3-(3-chlorophenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin-8ol ((-)-16). General procedure was used, and the alkylation was achieved with 3-chlorophen ethyl bromide. The product was obtained as a pale-yellow solid ( $300 \mathrm{mg}, 69 \%$ ), $\mathrm{mp}(\mathrm{HCl}$ salt) $258-260{ }^{\circ} \mathrm{C}(\mathrm{dec}) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.26-7.22(\mathrm{~m}, 2 \mathrm{H}), 7.18-7.13(\mathrm{~m}$, 2H), $6.88(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.67(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.54(\mathrm{dd}, J=8.2,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.00(\mathrm{dd}$, $J=5.6,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.90(\mathrm{~d}, J=18.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.80-2.72(\mathrm{~m}, 3 \mathrm{H}), 2.67(\mathrm{dd}, J=5.5,1.8 \mathrm{~Hz}$, $1 \mathrm{H}), 2.65-2.57(\mathrm{~m}, 2 \mathrm{H}), 2.12(\mathrm{td}, J=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.85(\mathrm{dt}, J=7.5,4.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.79(\mathrm{dt}$, $J=12.9,6.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H}), 1.30(\mathrm{dd}, J=2.8,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 0.85(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .13 \mathrm{C}$

NMR (101 MHz; CD ${ }_{3}$ OD): $\delta 155.4,142.5,142.2,133.7,129.5,128.3,127.7,126.74,126.69$, $125.8,112.8,111.6,57.4,56.2,45.6,41.5,41.1,35.9,33.1,24.5,22.6,13.1$. HRMS (ES+) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{NOCl}, 356.1781(\mathrm{M}+\mathrm{H})^{+}$; found, 356.1785. $[\alpha]^{20}{ }_{\mathrm{D}}-104^{\circ}$ (c 1.1, $\mathrm{CHCl}_{3} / \mathrm{MeOH}$ (9/1)). Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{Cl}_{2} \mathrm{NO}: ~ \mathrm{C}, 67.35 ; \mathrm{H}, 6.94 ; \mathrm{N}, 3.57$. Found: C, 67.37; H, 6.72; N, 3.50.
(1R,5R,9R)-6,11-Dimethyl-3-(2-chlorophenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin-8ol ((-)-17). General procedure was used, and the alkylation was achieved with 2-chlorophen ethyl bromide. The product was obtained as a pale-yellow solid ( $250 \mathrm{mg}, 74 \%$ ), $\mathrm{Mp}(\mathrm{HCl}$ salt) $249-251{ }^{\circ} \mathrm{C}$ (dec). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.31$ (ddd, $J=15.6,7.6,1.5 \mathrm{~Hz}, 2 \mathrm{H}$ ), $7.23-7.14(\mathrm{~m}, 2 \mathrm{H}), 6.86(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.67(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.54(\mathrm{dd}, J=8.2,2.5 \mathrm{~Hz}$, $1 \mathrm{H}), 3.04(\mathrm{dd}, J=5.4,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.00-2.86(\mathrm{~m}, 3 \mathrm{H}), 2.75(\mathrm{td}, J=11.6,5.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.72-2.58$ $(\mathrm{m}, 3 \mathrm{H}), 2.14(\mathrm{td}, J=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.88(\mathrm{dd}, J=7.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.81(\mathrm{td}, J=13.0,4.7 \mathrm{~Hz}$, $1 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H}), 1.30(\mathrm{~s}, 1 \mathrm{H}), 0.85(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta$ $155.4,142.2,137.4,133.5,130.7,129.1,127.70,127.61,126.9,126.6,112.8,111.6,57.5,54.6$, $45.7,41.3,40.9,35.9,31.1,24.5,22.8,13.1$. HRMS (ES+) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{ClNO}, 356.1781$ $(\mathrm{M}+\mathrm{H})^{+}$; found, 356.1781. [ $\left.\alpha\right]^{20}{ }_{\mathrm{D}}-90.6^{\circ}\left(c 1.2, \mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)\right)$. Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{27} \mathrm{Cl}_{2} \mathrm{NO} \bullet 0.15 \mathrm{H}_{2} \mathrm{O} \bullet 0.1 \mathrm{C}_{3} \mathrm{H}_{8} \mathrm{O}: \mathrm{C}, 66.78 ; \mathrm{H}, 7.06 ; \mathrm{N}, 3.49$. Found: C, 66.87; H, 6.97; N, 3.39.
(1R,5R,9R)-6,11-Dimethyl-3-(4-methoxyphenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin8 -ol ((-)-18). General procedure was used, and the alkylation was achieved with 4methoxyphenethyl bromide. The product was obtained as a pale-yellow solid ( 465 mg , $82 \%), \mathrm{mp}\left(\mathrm{HCl}\right.$ salt) $164-166{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right): \delta 7.12-7.09(\mathrm{~m}, 2 \mathrm{H}), 6.87$ (d, $J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.83-6.79(\mathrm{~m}, 2 \mathrm{H}), 6.67(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.54(\mathrm{dd}, J=8.2,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.73$ $(\mathrm{s}, 3 \mathrm{H}), 3.03(\mathrm{dd}, J=5.6,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.91(\mathrm{~d}, J=18.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.73-2.59(\mathrm{~m}, 6 \mathrm{H}), 2.12$ (td, $J=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.87(\mathrm{dd}, J=7.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.80(\mathrm{td}, J=12.9,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.33(\mathrm{~s}, 3 \mathrm{H})$, $1.30(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 0.85(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.101 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right): \delta 158.2,155.4$, 142.1, 131.8, 129.1, 127.7, 126.6, 113.5, 112.8, 111.6, 57.3, 56.8, 54.2, 45.7, 41.3, 41.0, 35.9, 32.4, 24.4, 22.5, 13.1. HRMS ( $\mathrm{ES}^{+}$) Calcd. For $\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{NO}_{2}, 352.2277(\mathrm{M}+\mathrm{H})^{+}$; found, 352.2278. $[\alpha]^{20}{ }_{\mathrm{D}}-109.0^{\circ}\left(c 1.0, \mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)\right.$ ). Anal. Calcd. For $\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{ClNO}_{2} \bullet 0.5 \mathrm{H}_{2} \mathrm{O}: \mathrm{C}$, 69.59; H, 7.87; N, 3.53. Found: C, 69.88; H, 8.22; N, 3.23.
(1R,5R,9R)-6,11-Dimethyl-3-(4-methylphenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin8 -ol ((-)-19). General procedure was used, and the alkylation was achieved with 4methylphenethyl bromide. The product was obtained as a pale-yellow solid ( 460 mg , $72 \%$ ), mp (HCl salt) $305-306{ }^{\circ} \mathrm{C}$ (dec). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right): \delta 7.09(\mathrm{~s}, 4 \mathrm{H}), 6.89(\mathrm{~d}$, $J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.69(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.56(\mathrm{dd}, J=8.2,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.05(\mathrm{dd}, J=5.7,3.1 \mathrm{~Hz}$, $1 \mathrm{H}), 2.93(\mathrm{~d}, J=18.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.79-2.60(\mathrm{~m}, 6 \mathrm{H}), 2.28(\mathrm{~s}, 3 \mathrm{H}), 2.13(\mathrm{td}, J=12.4,3.3 \mathrm{~Hz}, 1 \mathrm{H})$, $1.90(\mathrm{td}, J=6.9,2.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.82(\mathrm{td}, J=12.9,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.35(\mathrm{~s}, 3 \mathrm{H}), 1.32(\mathrm{t}, J=2.5 \mathrm{~Hz}$, $1 \mathrm{H}), 0.87(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; $\left.\mathrm{CD}_{3} \mathrm{OD}\right): \delta 156.9,143.7,138.3,136.8,130.1$, 129.6, 129.2, 128.1, 114.3, 113.1, 58.7, 58.2, 47.2, 42.9, 42.5, 37.4, 34.4, 26.0, 24.0, 21.1, 14.6. HRMS (ES ${ }^{+}$) Calcd. For $\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{NO}, 336.2327(\mathrm{M}+\mathrm{H})^{+}$; found, 336.2324. $[\alpha]^{20}{ }_{\mathrm{D}}-110.8^{\circ}(c$ 1.0, $\left.\mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)\right)$. Anal. Calcd. For $\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{ClNO} \bullet 0.05 \mathrm{H}_{2} \mathrm{O}: \mathrm{C}, 74.09 ; \mathrm{H}, 8.14 ; \mathrm{N}$, 3.76. Found: C, 74.10; H, 8.18; N, 3.74.
(1R,5R,9R)-6,11-Dimethyl-3-(4-hydroxyphenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin8 -ol ((-)-20). General procedure was used, and the alkylation was achieved with 4hydroxyphenethyl bromide. The product was obtained as a pale-yellow solid ( 170 mg , $52 \%$ ) mp ( HCl salt) $300-302{ }^{\circ} \mathrm{C}(\mathrm{dec}) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.03-6.99(\mathrm{~m}, 2 \mathrm{H})$, $6.88(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.70-6.66(\mathrm{~m}, 3 \mathrm{H}), 6.54(\mathrm{dd}, J=8.2,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.03(\mathrm{dd}, J=5.5$, $3.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.96-2.83(\mathrm{~m}, 2 \mathrm{H}), 2.73-2.59(\mathrm{~m}, 6 \mathrm{H}), 2.11(\mathrm{td}, J=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.87(\mathrm{dt}$, $J=7.0,3.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.80(\mathrm{td}, J=12.9,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H}), 1.30(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H})$, $0.85(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; CD $\left.{ }_{3} \mathrm{OD}\right): ~ \delta 155.40,155.35,142.1,130.5,129.1$, 127.7, 126.6, 114.8, 112.8, 111.6, 57.2, 56.9, 45.7, 41.3, 40.9, 35.9, 32.4, 24.4, 22.4, 13.1. HRMS (ES+) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{NO}_{2}, 338.2120(\mathrm{M}+\mathrm{H})^{+}$; found, 338.2122. $[\alpha]^{20}{ }_{\mathrm{D}}-120^{\circ}$ (c 1.0,
$\mathrm{CHCl}_{3} / \mathrm{MeOH}$ (9/1Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{ClNO}_{2} \bullet 0.05 \mathrm{H}_{2} \mathrm{O} \bullet 0.2 \mathrm{C}_{3} \mathrm{H}_{8} \mathrm{O}: \mathrm{C}, 70.17 ; \mathrm{H}$, $7.74 ;$ N, 3.62. Found: C, 70.24; H, 7.83; N, 3.70.
(1R,5R,9R)-6,11-Dimethyl-3-(2,4-dichlorophenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d] az ocin-8-ol ((-)-21). General procedure was used, and the alkylation was achieved with 2,4dichlorophenethyl bromide. The product was obtained as a pale-yellow solid ( 220 mg , $61 \%$ ), Mp ( HCl salt) $240-241{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.39(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H})$, 7.28 (d, $J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.23$ (dd, $J=8.2,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.86(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.67$ (d, $J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.54(\mathrm{dd}, J=8.2,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.01(\mathrm{dt}, J=5.6,2.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.96-2.89(\mathrm{~m}$, $2 \mathrm{H}), 2.87-2.83(\mathrm{~m}, 1 \mathrm{H}), 2.77-2.71(\mathrm{~m}, 1 \mathrm{H}), 2.68(\mathrm{t}, J=9.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.63-2.56(\mathrm{~m}, 2 \mathrm{H}), 2.13$ $(\mathrm{td}, J=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.87(\mathrm{tt}, J=7.0,3.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.79(\mathrm{td}, J=12.9,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.32(\mathrm{~s}$, $3 \mathrm{H}), 1.29-1.25(\mathrm{~m}, 1 \mathrm{H}), 0.84(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.101 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right): \delta 155.4,142.2$, 136.4, 134.3, 132.4, 131.8, 128.6, 127.7, 127.0, 126.6, 112.8, 111.6, 57.6, 54.3, 45.6, 41.4, 41.0, 35.9, 30.6, 24.5, 22.9, 13.1. HRMS (ES+) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{26} \mathrm{Cl}_{2} \mathrm{NO},(\mathrm{M}+\mathrm{H})^{+} 390.1391$; found, 390.1387. $[\alpha]^{20}{ }_{\mathrm{D}}-95.6^{\circ}$ (c 1.1, $\mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)$ ). Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{26} \mathrm{Cl}_{3} \mathrm{NO} \bullet 0.55$ $\mathrm{H}_{2} \mathrm{O}: \mathrm{C}, 61.56 ; \mathrm{H}, 6.09$; N, 3.12. Found: C, 61.59; H, 6.17; N, 3.03.
(1R,5R,9R)-6,11-Dimethyl-3-(2,6-dichlorophenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]az ocin-8-ol ((-)-22). General procedure was used, and the alkylation was achieved with 2,6dichlorophenethyl bromide. The product was obtained as a pale-yellow solid ( 245 mg , $66 \%$ ), mp ( HCl salt) $241-243^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.34(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.17 $(\mathrm{t}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.86(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.67(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.54(\mathrm{dd}, J=8.2,2.5 \mathrm{~Hz}$, $1 \mathrm{H}), 3.21-3.13(\mathrm{~m}, 2 \mathrm{H}), 3.10(\mathrm{t}, J=4.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.93(\mathrm{~d}, J=18.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.73(\mathrm{dtd}, J=17.4$, $11.5,5.4 \mathrm{~Hz}, 3 \mathrm{H}), 2.59(\mathrm{td}, J=11.9,5.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.19(\mathrm{td}, J=12.4,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 1.93-1.87(\mathrm{~m}$, $1 \mathrm{H}), 1.83(\mathrm{dt}, J=12.9,6.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.34-1.31(\mathrm{~m}, 4 \mathrm{H}), 0.86(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; CD 3 OD): $\delta 155.4,142.1,135.18,135.07,128.26,128.08,127.7,126.4,112.8,111.6$, $57.7,52.2,45.8,41.2,40.7,35.8,28.8,24.4,23.0,13.1$. HRMS (ES+) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{26} \mathrm{Cl}_{2} \mathrm{NO}$, $(\mathrm{M}+\mathrm{H})^{+}$390.1391; found, 390.1389. [ $\left.\alpha\right]^{20}{ }_{\mathrm{D}}-89.1^{\circ}\left(c 1.1, \mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)\right)$. Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{26} \mathrm{Cl}_{3} \mathrm{NO} \bullet 0.4 \mathrm{H}_{2} \mathrm{O} \bullet 0.2 \mathrm{C}_{3} \mathrm{H}_{8} \mathrm{O}: \mathrm{C}, 60.86 ; \mathrm{H}, 6.42 ; \mathrm{N}, 3.14$. Found: C, 60.92; H , 6.35 ; N, 3.07.
(1R,5R,9R)-6,11-Dimethyl-3-(phenethyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin-8-ol $((-)-23)$. General procedure was used, and the alkylation was achieved with phenethyl bromide. The product was obtained as a pale-yellow solid ( $244 \mathrm{mg}, 64 \%$ ) , mp ( HCl salt) $277-280^{\circ} \mathrm{C}$ (dec). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.27-7.19(\mathrm{~m}, 4 \mathrm{H}), 7.18-7.14(\mathrm{~m}, 1 \mathrm{H}), 6.88$ $(\mathrm{d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.67(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.55(\mathrm{dd}, J=8.2,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.07(\mathrm{dd}, J=5.5$, $3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.97-2.90(\mathrm{~m}, 1 \mathrm{H}), 2.83-2.76(\mathrm{~m}, 3 \mathrm{H}), 2.73-2.63(\mathrm{~m}, 3 \mathrm{H}), 2.15(\mathrm{td}, J=12.5,3.2 \mathrm{~Hz}$, $1 \mathrm{H}), 1.88$ (ddd, $J=6.4,5.8,4.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.81(\mathrm{dt}, J=13.0,6.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.33(\mathrm{~s}, 3 \mathrm{H}), 1.31(\mathrm{~d}$, $J=2.8 \mathrm{~Hz}, 1 \mathrm{H}), 0.86(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; CD3OD): $\delta 155.4,142.1,139.8$, $128.24,128.05,127.7,126.5,125.8,112.8,111.6,57.4,56.5,45.7,41.2,40.9,35.8,33.3,24.4,22.5$, 13.0. HRMS (ES+) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{NO}, 322.2171(\mathrm{M}+\mathrm{H})^{+}$; found, 322.2170. $[\alpha]^{20}{ }_{\mathrm{D}}-101.5^{\circ}$ (c 1.1, $\mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)$ ). Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{ClNO} \bullet 0.05 \mathrm{H} 2 \mathrm{O}: \mathrm{C}, 73.64 ; \mathrm{H}, 7.89$; N, 3.90. Found: C, 73.67; H, 7.79; N, 3.83.
(1R,5R,9R)-6,11-Dimethyl-3-(phenylpropyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin-8ol ((-)-24). General procedure was used, and the alkylation was achieved with phenylpropyl bromide. The product ( - )-24 was obtained as a pale-yellow solid ( $255 \mathrm{mg}, 80 \%$ ), $\mathrm{mp}\left(\mathrm{HCl}\right.$ salt) $209-211{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.26-7.22(\mathrm{~m}, 2 \mathrm{H}), 7.19-7.12(\mathrm{~m}$, $3 \mathrm{H}), 6.87(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.67(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.55(\mathrm{dd}, J=8.3,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.06$ (dd, $J=5.5,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.87(\mathrm{~d}, J=18.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.72-2.56(\mathrm{~m}, 6 \mathrm{H}), 2.15(\mathrm{td}, J=12.6,3.3 \mathrm{~Hz}, 1 \mathrm{H})$, $1.90-1.75(\mathrm{~m}, 4 \mathrm{H}), 1.34-1.31(\mathrm{~m}, 4 \mathrm{H}), 0.84(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.101 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right)$ : $\delta 155.6,141.8,141.5,128.0,127.8,125.9,125.6,113.0,111.6,57.4,53.7,45.8,40.8,40.5,35.7$, 33.2, 28.1, 24.3, 22.5, 12.9. HRMS (ES+) Calcd. For $\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{NO},(\mathrm{M}+\mathrm{H})^{+} 336.2327$; found, 336.2329. $[\alpha]^{20}{ }_{\mathrm{D}}-73.1^{\circ}$ (c 1.1, $\mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)$ ). Anal. Calcd. For $\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{ClNO} \bullet 0.2$ $\mathrm{H}_{2} \mathrm{O}: \mathrm{C}, 73.56 ; \mathrm{H}, 8.16$; N, 3.73. Found: C, 73.57; H, 8.02; N, 3.64.
(1R,5R,9R)-6,11-Dimethyl-3-(cinnamyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin-8-ol $((-)-25)$. General procedure was used, and the alkylation was achieved with cinnamyl bromide. The product ( - )-25 was obtained as an off-white foam ( $200 \mathrm{mg}, 66 \%$ ), $\mathrm{mp}(\mathrm{HCl}$ salt) $162-164{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.39-7.37$ (m, 2H), 7.29-7.25 (m, 2H), $7.21-7.17(\mathrm{~m}, 1 \mathrm{H}), 6.91(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.67(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.60-6.55(\mathrm{~m}, 2 \mathrm{H}), 6.26(\mathrm{dt}$, $J=15.8,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.37(\mathrm{ddd}, J=13.4,7.1,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.01-2.95(\mathrm{~m}, 2 \mathrm{H}), 2.65(\mathrm{dd}, J=18.4$, $6.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.59-2.55(\mathrm{~m}, 1 \mathrm{H}), 2.12(\mathrm{td}, J=12.5,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.88$ (dd, $J=7.0,3.1 \mathrm{~Hz}, 1 \mathrm{H})$, $1.79(\mathrm{td}, J=12.9,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H}), 1.30(\mathrm{t}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.83(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR (101 MHz; CD ${ }_{3} \mathrm{OD}$ ): $\delta 155.4,142.2,136.8,133.4,128.2,127.7,127.2,126.6,125.9$, $125.4,112.8,111.6,57.2,56.8,45.4,41.3,41.0,35.9,24.5,22.5,13.1$. HRMS (ES+) Calcd. For $\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{NO},(\mathrm{M}+\mathrm{H})^{+} 334.2171$; found, 334.2171. $[\alpha]^{20}{ }_{\mathrm{D}}-118.3^{\circ}\left(c 1.1, \mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)\right)$. Anal. Calcd. For $\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{ClNO} \bullet 0.5 \mathrm{H}_{2} \mathrm{O}: \mathrm{C}, 72.9 ; \mathrm{H}, 7.71 ; \mathrm{N}, 3.70$. Found: C, 73.00; H, 7.52; N, 3.48.
(1R,5R,9R)-3-(2-(1H-Indol-3-yl)ethyl)-6,11-dimethyl-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]a zocin-8-ol ((-)-26). General procedure was used, and the alkylation was achieved with 3 -(2-bromoethyl)-1H-indole. The product was obtained as a pale-yellow solid ( 250 mg , $80 \%$ ), mp 115-118 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.52(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.30(\mathrm{~d}$, $J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.08-7.04(\mathrm{~m}, 2 \mathrm{H}), 6.98(\mathrm{td}, J=7.5,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.87(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.68$ $(\mathrm{d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.55(\mathrm{dd}, J=8.2,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.13(\mathrm{dd}, J=5.5,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.98-2.92$ $(\mathrm{m}, 3 \mathrm{H}), 2.90-2.85(\mathrm{~m}, 1 \mathrm{H}), 2.81(\mathrm{dd}, J=10.7,4.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.71-2.65(\mathrm{~m}, 2 \mathrm{H}), 2.19-2.13(\mathrm{~m}$, $1 \mathrm{H}), 1.92(\mathrm{dd}, J=7.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.85(\mathrm{dt}, J=13.0,6.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.35-1.32(\mathrm{~m}, 4 \mathrm{H}), 0.87(\mathrm{~d}$, $J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; CD ${ }_{3} \mathrm{OD}$ ): $\delta 155.4,142.1,136.7,127.7,127.2,126.5,121.6$, $120.9,118.1,117.7,112.8,112.3,111.6,110.8,57.2,55.4,45.8,41.2,40.9,35.9,24.4,22.63,22.50$, 13.1. HRMS (ES+) Calcd. For $\mathrm{C}_{24} \mathrm{H}_{29} \mathrm{~N}_{2} \mathrm{O},(\mathrm{M}+\mathrm{H})^{+} 361.2280$; found, 361.2285. $[\alpha]^{20}{ }_{\mathrm{D}}-78.5^{\circ}$ (c 1.0, $\mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)$ ). Anal. Calcd. For $\mathrm{C}_{24} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O} \bullet 0.7 \mathrm{H}_{2} \mathrm{O} \bullet 0.25 \mathrm{CHCl}_{3}$ : C, 72.28; H, 7.42; N, 6.95. Found: C, 72.37; H, 7.51; N, 6.80.

1-((1R,5R,9R)-8-Hydroxy-6,11-dimethyl-1,4,5,6-tetrahydro-2,6-methanobenzo[d]azocin-3(2H)-yl)-2-(1-methylpiperidin-4-yl)ethan-1-one (-)-27). To a stirred solution of 2-(1-methylpiperidin-4yl)acetic acid ( $219 \mathrm{mg}, 1.05$ equiv, 1.39 mmol ), $\mathrm{N}, \mathrm{N}$-diisopropylethylamine ( $685 \mathrm{mg}, 923 \mu \mathrm{~L}$, 4 equiv, 5.30 mmol ) and ( - )-2ebromocamphorsulfonate ( 700 mg , 1 equiv, 1.32 mmol ) in DMF at $0{ }^{\circ} \mathrm{C}$, HATU ( $529 \mathrm{mg}, 1.05$ equiv, 1.39 mmol ) was added. The reaction was stirred for 16 h . The reaction was diluted with water, and saturated $\mathrm{NaHCO}_{3}$ was added to it to adjust the pH to $\sim 8.5$. The aqueous solution was extracted with $\mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)$. The organic extractions were combined and washed with $\mathrm{H}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$ to remove DMF. The organic extractions were then washed with brine $(2 \times 20 \mathrm{~mL})$. The organic layer was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and the solvent was removed in vacuo. The crude residue was purified by flash chromatography ( $2-25 \% \mathrm{CMA} \mathrm{in}^{\mathrm{CHCl}_{3}}$ ) to give ( - )-27 as an off-white foam ( $400 \mathrm{mg}, 97 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 6.87(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.66$ $(\mathrm{d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.54(\mathrm{dd}, J=8.2,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.96(\mathrm{dd}, J=5.6,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.91-2.83(\mathrm{~m}$, $3 \mathrm{H}), 2.67-2.47(\mathrm{~m}, 4 \mathrm{H}), 2.24(\mathrm{~s}, 3 \mathrm{H}), 2.08(\mathrm{td}, J=12.5,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.00(\mathrm{t}, J=11.0 \mathrm{~Hz}, 2 \mathrm{H})$, 1.84 (qd, $J=7.5,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.74$ (ddd, $J=15.4,11.3,3.9 \mathrm{~Hz}, 3 \mathrm{H}), 1.45(\mathrm{dd}, J=10.2,5.7 \mathrm{~Hz}$, 2H), 1.31-1.26 (m, 7H), 0.84 (d, $J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 155.4,142.1$, 127.7, 126.5, 112.8, 111.6, 57.0, 55.2, 52.0, 45.7, 44.9, 41.3, 40.9, 35.9, 33.5, 33.1, 31.64, 31.56, 24.4, 22.3, 13.1. HRMS (ES+) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{33} \mathrm{~N}_{2} \mathrm{O}_{2}$, (M+H)+ 357.2542; found, 357.2539.
(1R,5R,9R)-6,11-Dimethyl-3-(2-(1-methylpiperidin-4-yl)ethyl)-1,2,3,4,5,6-hexahydro-2,6-methano benzo[d]azocin-8-ol ((-)-28). Amide (-)-27 ( 0.307 g , 1 equiv, $861 \mu \mathrm{~mol}$ ) was dissolved in THF $(7 \mathrm{~mL})$. To this, a freshly prepared borane solution ( $2.58 \mathrm{~mL}, 4$ molar, 12 equiv, 10.03 mmol ) was added. The solution was heated at $60^{\circ} \mathrm{C}$ for 24 h . The reaction was quenched with anhydrous MeOH at $0^{\circ} \mathrm{C}$ and stirred for 30 min . The solvent was removed in vacuo and then redissolved in MeOH and acidified with 2 N HCl . The mixture was refluxed for 4 h . The reaction was cooled to room temperature and then basified with $\mathrm{NH}_{4} \mathrm{OH}$ solution to pH 9. The aqueous solution was extracted with $\mathrm{CHCl}_{3}: \mathrm{MeOH}(9: 1)$. The organic layers were washed with brine and dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and solvent removed
in vacuo to give ( - )-28 as a white solid ( $170 \mathrm{mg}, 58 \%$ ), mp ( HCl salt) $309-310{ }^{\circ} \mathrm{C}$ (dec). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 6.87$ (d, $\left.J=8.3 \mathrm{~Hz}, 1 \mathrm{H}\right), 6.66(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.54$ (dd, $J=8.2,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.96(\mathrm{dd}, J=5.6,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.91-2.83(\mathrm{~m}, 3 \mathrm{H}), 2.67-2.47(\mathrm{~m}, 4 \mathrm{H})$, $2.24(\mathrm{~s}, 3 \mathrm{H}), 2.08(\mathrm{td}, J=12.5,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.00(\mathrm{t}, J=11.0 \mathrm{~Hz}, 2 \mathrm{H}), 1.84(\mathrm{qd}, J=7.5,3.2$ Hz, 1H), 1.74 (ddd, $J=15.4,11.3,3.9 \mathrm{~Hz}, 3 \mathrm{H}), 1.45$ (dd, $J=10.2,5.7 \mathrm{~Hz}, 2 \mathrm{H}), 1.31-1.26$ $(\mathrm{m}, 7 \mathrm{H}), 0.84(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; CD 3 OD$): \delta 155.4,142.1,127.7,126.5$, $112.8,111.6,57.0,55.2,52.0,45.7,44.9,41.3,40.9,35.9,33.5,33.1,31.64,31.56,24.4,22.3,13.1$. HRMS (ES+) Calcd. For $\mathrm{C}_{22} \mathrm{H}_{35} \mathrm{~N}_{2} \mathrm{O},(\mathrm{M}+\mathrm{H})^{+} 343.2749$; found, 343.2746. [ $\left.\alpha\right]^{20}{ }_{\mathrm{D}}-87.5^{\circ}$ (c 1.1, $\mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)$ ). Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{36} \mathrm{Cl}_{2} \mathrm{~N}_{2} \mathrm{O} \bullet 0.2 \mathrm{C}_{3} \mathrm{H}_{8} \mathrm{O} \bullet 1.15 \mathrm{H}_{2} \mathrm{O}: \mathrm{C}$, $60.57 ;$ H, 8.97 ; N, 6.25. Found: C, 60.57 ; H, 8.99; N, 6.26.
(1R,5R,9R)-6,11-Dimethyl-3-(3-(4-nitrophenyl)propyl)-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d] azocin-8-ol ((-)-31). General procedure was used, and the alkylation was achieved with 30 as the alkylating agent (obtained from 3-(4-nitrophenyl)propan-1-ol, 29). The product (-)-31 was obtained as a pale-yellow solid ( $252 \mathrm{mg}, 70 \%$ ), mp ( HCl salt) $265-266{ }^{\circ} \mathrm{C}$ (dec). ${ }^{1} \mathrm{H}$ NMR (400 MHz; CD $\left.{ }_{3} \mathrm{OD}\right): ~ \delta 8.15-8.11(\mathrm{~m}, 2 \mathrm{H}), 7.46-7.42(\mathrm{~m}, 2 \mathrm{H}), 6.85(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H})$, $6.65(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.53(\mathrm{dd}, J=8.2,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.92(\mathrm{dd}, J=5.6,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.85(\mathrm{~d}$, $J=18.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.76(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.65-2.45(\mathrm{~m}, 4 \mathrm{H}), 2.06(\mathrm{td}, J=12.4,3.2 \mathrm{~Hz}, 1 \mathrm{H})$, $1.90-1.80(\mathrm{~m}, 3 \mathrm{H}), 1.76(\mathrm{td}, J=12.9,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.30(\mathrm{~s}, 3 \mathrm{H}), 1.27(\mathrm{t}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 0.82(\mathrm{~d}$, $J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; CD ${ }_{3} \mathrm{OD}$ ): $\delta 155.4,150.2,146.3,142.2,129.11,129.06$, 127.6, 126.6, 123.1, 112.7, 111.6, 57.2, 53.7, 45.6, 41.4, 41.0, 35.9, 33.1, 28.0, 24.5, 22.5, 13.1. HRMS (ES+) Calcd. For $\mathrm{C}_{23} \mathrm{H}_{29} \mathrm{~N}_{2} \mathrm{O}_{3},(\mathrm{M}+\mathrm{H})^{+} 381.2178$; found, 381.2174. [ $\left.\alpha\right]^{20}{ }_{\mathrm{D}}-77.5^{\circ}$ (c 1.1, $\mathrm{CHCl}_{3} / \mathrm{MeOH}(9 / 1)$ ). Anal. Calcd. For $\mathrm{C}_{23} \mathrm{H}_{29} \mathrm{ClN}_{2} \mathrm{O}_{3} \bullet 0.05 \mathrm{H}_{2} \mathrm{O}: \mathrm{C}, 66.11 ; \mathrm{H}, 7.02 ; \mathrm{N}$, 6.70. Found: C, 65.97; H, 6.75; N, 6.62.

3-(4-Nitrophenyl)propyl methanesulfonate (30). 3-(4-Nitrophenyl)propan-1-ol (510 mg, 1 equiv, 2.81 mmol ) was dissolved in anhydrous dichloromethane ( 15 mL ) and cooled to $0{ }^{\circ} \mathrm{C}$. Triethylamine ( $427 \mathrm{mg}, 588 \mu \mathrm{~L}, 1.5$ equiv, 4.22 mmol ) and methanesulfonyl chloride ( 355 mg , $241 \mu \mathrm{~L}, 1.1$ equiv, 3.10 mmol ) were added to it, and the reaction was stirred at $0^{\circ} \mathrm{C}$ for 1 h . The reaction was diluted with water and extracted with dichloromethane ( 15 mL ). The organic layer was washed with saturated $\mathrm{NaHCO}_{3}(15 \mathrm{~mL})$ and brine ( 15 mL ) and dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The organic layer was filtered, and the solvent was removed in vacuo to give a pale-yellow oil. The crude product was purified by flash chromatography ( $0-80 \%$ EtOAc/hexanes) to give 30 as a pale-yellow oil that solidified on standing ( 700 mg , $96 \%) .{ }^{1} \mathrm{HNMR}\left(400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}\right): ~ \delta 8.16-8.13(\mathrm{~m}, 2 \mathrm{H}), 7.47-7.44(\mathrm{~m}, 2 \mathrm{H}), 4.24(\mathrm{t}, \mathrm{J}=6.2 \mathrm{~Hz}$, $2 \mathrm{H}), 3.05(\mathrm{~s}, 3 \mathrm{H}), 2.86(\mathrm{t}, \mathrm{J}=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 2.11-2.04(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 148.9,146.5,129.2,123.2,69.1,35.6,31.0,30.0$. HRMS (ES+) Calcd. For $\mathrm{C}_{10} \mathrm{H}_{13} \mathrm{NO}_{5} \mathrm{~S}$, $(\mathrm{M}+\mathrm{Na})^{+}$282.0412; found, 282.0413 .
rac-8-Methoxy-6,11-dimethyl-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocine (32) [29]. 6,11-Dimethyl-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin-8-ol (rac-normetazocine, (rac-1), 10 g , 1 equiv, 46 mmol ) and potassium carbonate ( $13 \mathrm{~g}, 2$ equiv, 92 mmol ) were added to $\mathrm{H}_{2} \mathrm{O}$ ( 0.15 L ), and the mixture was cooled to $0{ }^{\circ} \mathrm{C}$ for 20 min . In a separate Erlenmeyer flask, di-tert-butyl dicarbonate ( $10 \mathrm{~g}, 11 \mathrm{~mL}, 1$ equiv, 46 mmol ) was dissolved in 1,4-dioxane ( 75 mL ). This solution was added to the aqueous mixture, and the reaction was warmed at $23^{\circ} \mathrm{C}$ over the course of 18 h . The organic products were extracted with $\mathrm{CHCl}_{3} 3 \mathrm{x}$, and the organic solution was washed with 1 M HCl and brine. The combined organic phase was dried over sodium sulfate and concentrated in vacuo to provide tert-butyl 8-hydroxy-6,11-dimethyl-1,4,5,6-tetrahydro-2,6-methanobenzo[d]azocine-3(2H)-carboxylate ( $15 \mathrm{~g}, 47 \mathrm{mmol}, 100 \%$ ) as a yellow / orange oil. The tert-butyl compound, as a crude oil ( 15 g , 1 equiv, 46 mmol ), was dissolved in $2 \mathrm{~N} \mathrm{NaOH}\left(94 \mathrm{~mL}\right.$ ) and cooled to $0^{\circ} \mathrm{C}$ for 20 min . Dimethyl sulfate ( 5.8 g , 4.4 mL , 1 equiv, 46 mmol ) was added, and the mixture kept at $23^{\circ} \mathrm{C}$ for 4 h . Diethyl ether was added to dilute the reaction, and the aqueous phase extracted $3 x$ with diethyl ether. The combined organic phase was washed with brine and concentrated in vacuo to a white foam. The foam was dissolved in $1 \mathrm{M} \mathrm{HCl}(0.11 \mathrm{~L})$ and methanol $(0.11 \mathrm{~L})$ and kept at $50^{\circ} \mathrm{C}$ until TLC showed full conversion. Volatiles and $\mathrm{H}_{2} \mathrm{O}$ were removed in vacuo to provide
rac-normetazocine methyl ether 32 ( $7.65 \mathrm{~g}, 33.1 \mathrm{mmol}, 72 \%$ ) as a foam. ${ }^{1} \mathrm{H}-\mathrm{NMR}(400 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right): \delta 7.06(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.80(\mathrm{~d}, J=2.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.76(\mathrm{dd}, J=8.4,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.79$ ( $\mathrm{s}, 3 \mathrm{H}$ ), 3.23-3.05 (m, 3H), 2.74 (td, $J=13.3,3.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.15(\mathrm{dd}, J=7.0,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 1.99$ (dd, $J=13.7,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.51(\mathrm{dd}, J=14.0,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.42(\mathrm{~s}, 3 \mathrm{H}), 0.88(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H})$. ${ }^{13}$ C NMR (101 MHz; cdcl3): $\delta 158.8,140.4,128.7,125.0,111.9,111.5,55.2,52.2,38.6,37.71$, $37.54,35.5,27.1,24.8,13.1$. The NMR spectra corresponded to those in the literature [29].
rac-8-Methoxy-6,11-dimethyl-3,4,5,6-tetrahydro-2,6-methanobenzo[d]azocin-1(2H)-one (33). In an argon-charged flame-dried flask, $\mathrm{CrO}_{3}(7.5 \mathrm{~g}, 1.5$ equiv, 74.6 mmol ) was dissolved in $10: 1$ $\mathrm{H}_{2} \mathrm{O}: \mathrm{H}_{2} \mathrm{SO}_{4}, 330 \mathrm{~mL}$. To this, the methyl ether 32 ( $13 \mathrm{~g}, 1$ equiv, 57.4 mmol ) in 330 mL of $10: 1 \mathrm{H}_{2} \mathrm{O}: \mathrm{H}_{2} \mathrm{SO}_{4}$, was added rapidly, and the mixture was refluxed for 5 h . The mixture was cooled and basified with $\mathrm{NH}_{4} \mathrm{OH}$ to pH 9 . The chromium precipitate (grey solid) was filtered off. The filtrate was extracted $2 x$ with EtOAc. The organic phase was washed with brine, dried, and concentrated in vacuo. The residue was purified via flash chromatography with CMA (50:45:5) to provide 8.9 g ( $63 \%$ yield) of the ketone 33 as a yellow syrup. ${ }^{1} \mathrm{H}-\mathrm{NMR}$ ( $400 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ): $\delta 8.05(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.87(\mathrm{dd}, J=8.7,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.81(\mathrm{~d}, J=2.5 \mathrm{~Hz}$, $1 \mathrm{H}), 3.89(\mathrm{~d}, J=6.1 \mathrm{~Hz}, 3 \mathrm{H}), 3.28(\mathrm{~d}, J=2.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.75(\mathrm{ddd}, J=12.0,5.3,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.60$ ( $\mathrm{td}, J=12.4,3.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.18-2.08 (m, 2H), $1.92(\mathrm{td}, J=12.9,5.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.49$ (ddd, $J=13.0$, $3.2,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.44(\mathrm{~s}, 3 \mathrm{H}), 0.87(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; $\left.\mathrm{CDCl}_{3}\right): \delta 198.8$, $164.8,149.4,128.1,127.6,112.0,111.6,63.2,55.3,43.6,40.8,39.2,37.8,25.9,14.2,13.7$. HRMS: [ $\left.\mathrm{C}_{15} \mathrm{H}_{19} \mathrm{NO}_{2}\right] \mathrm{H}^{+}$Calcd: 246.1494; Found: 246.1490. NMR spectra corresponded to those in the literature [29].
rac-8-Methoxy-6,11-dimethyl-3-phenethyl-3,4,5,6-tetrahydro-2,6-methanobenzo[d]azocin-1(2H)-one (rac-34). In a flame-dried flask, the C8-ketone 33 ( $8.9 \mathrm{~g}, 1$ equiv, 36 mmol ), (2-bromoethyl) benzene ( $20 \mathrm{~g}, 15 \mathrm{~mL}, 3$ equiv, 0.11 mol ), and potassium carbonate ( $20 \mathrm{~g}, 4$ equiv, 0.15 mol ) were added to acetonitrile $(120 \mathrm{~mL})$. The mixture was heated at $80^{\circ} \mathrm{C}$ for 18 h . The reaction mixture was filtered through celite to remove the remaining potassium carbonate. The filtrate was concentrated in vacuo, and the residual material was purified by flash chromatography using $30 \%$ ethyl acetate in hexanes. The N-phenethyl compound rac-34 (10.2 g, $29.2 \mathrm{mmol}, 80 \%$ ) was isolated as a light brown oil. The oil was concentrated overnight in vacuo to give a light-yellow solid. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right): \delta 7.99(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H})$, $7.25-7.20(\mathrm{~m}, 4 \mathrm{H}), 7.15(\mathrm{~s}, 1 \mathrm{H}), 6.82(\mathrm{dd}, J=8.7,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.77(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.84$ $(\mathrm{d}, J=5.0 \mathrm{~Hz}, 3 \mathrm{H}), 3.23(\mathrm{~d}, J=2.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.01-2.94(\mathrm{~m}, 1 \mathrm{H}), 2.80-2.73(\mathrm{~m}, 3 \mathrm{H}), 2.56-2.49$ $(\mathrm{m}, 1 \mathrm{H}), 2.10(\mathrm{~d}, J=10.0 \mathrm{~Hz}, 3 \mathrm{H}), 1.40(\mathrm{~s}, 3 \mathrm{H}), 0.87(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz; $\left.\mathrm{CDCl}_{3}\right): \delta 194.9,164.7,149.1,140.4,128.8,128.2,127.95,127.83,125.9,111.9,67.4,57.1,55.4$, $46.3,44.1,41.6,37.5,34.0,25.6,13.7$. NMR spectra agreed with those in the literature [29].

Optimized optical resolution of rac-8-methoxy-6,11-dimethyl-3-phenethyl-3,4,5,6-tetra hydro-2,6-methanobenzo[d]azocin-1(2H)-one (rac-34) to obtain $2 S, 6 R, 11 R-(-)-34$ and $2 R, 6 S$, $11 S-(+)-34(1 S, 5 R, 9 R-(-)-34$ and $1 R, 5 S, 9 S-(+)-34$ in benzomorphan nomenclature). The N -phenethyl compound rac-34 ( $3.14 \mathrm{~g}, 2$ equiv, 8.99 mmol ) as an orange oil was dissolved in acetone $(4 \mathrm{~mL})$. In a separate Erlenmeyer flask, L-tartaric acid ( $1.35 \mathrm{~g}, 2$ equiv, 8.99 mmol ) was dissolved in $\mathrm{N}, \mathrm{N}$-dimethylformamide ( 3 mL ) with stirring and heating to $80^{\circ} \mathrm{C}$, and the acetone solution of rac-34 was added. The combined solution was allowed to cool to $0^{\circ} \mathrm{C}$, and crystallization occurred after scratching the solution with a glass rod, providing a white crystalline tartrate salt of $(-)-34(1.33 \mathrm{~g}), \mathrm{mp} .197-199^{\circ} \mathrm{C} ;[\alpha]_{25}{ }^{\mathrm{D}}=-48.6^{\circ}$ (c 0.44, $\mathrm{MeOH})$. The salt was collected via filtration and free-based with concentrated $\mathrm{NH}_{4} \mathrm{OH}$. The free base was extracted with $\mathrm{CHCl}_{3}$, and the aqueous phase was extracted 2 x with $\mathrm{CHCl}_{3}$. The combined organic phases were washed $2 x$ with saturated $\mathrm{Na}_{2} \mathrm{CO}_{3}$. After drying over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and solvent removal in vacuo, ( - ) $-34(1.05 \mathrm{~g}, 3.00 \mathrm{mmol}, 67 \%)$ was obtained as a yellow oil (see Figure 1 for IUPAC numbering of benzomorphans), $[\alpha]_{25}{ }^{\mathrm{D}}-41.3^{\circ}$ (c 1.08, $\mathrm{MeOH}) .{ }^{1} \mathrm{HNMR}\left(400 \mathrm{MHz} ; \mathrm{C}_{6} \mathrm{H}_{6}\right): \delta 8.34(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.21(\mathrm{dd}, J=16.8,7.4 \mathrm{~Hz}$, $15 \mathrm{H}), 7.08(\mathrm{t}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.74(\mathrm{~s}, 1 \mathrm{H}), 6.50(\mathrm{dd}, J=8.6,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.34(\mathrm{~s}, 1 \mathrm{H}), 3.21(\mathrm{~s}$, $3 \mathrm{H}), 3.16-3.01(\mathrm{~m}, 2 \mathrm{H}), 2.84-2.77(\mathrm{~m}, 1 \mathrm{H}), 2.65(\mathrm{ddd}, J=12.2,9.1,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.53-2.49(\mathrm{~m}$, $1 \mathrm{H}), 2.17-2.10(\mathrm{~m}, 1 \mathrm{H}), 1.94-1.90(\mathrm{~m}, 1 \mathrm{H}), 1.79-1.71(\mathrm{~m}, 1 \mathrm{H}), 1.24-1.21(\mathrm{~m}, 1 \mathrm{H}), 1.08(\mathrm{~s}, 3 \mathrm{H})$,
$0.74(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H})$. HRMS: $\left[\mathrm{C}_{23} \mathrm{H}_{27} \mathrm{NO}_{2}\right] \mathrm{H}^{+}$Calculated: 350.2120; Found: 350.2123. The mother liquor after isolation of $(-)-34$ was free-based with ammonia water and extracted 3 x with $\mathrm{CHCl}_{3}$. After drying and removal of solvent in vacuo, 2.1 g of rac-34 base was collected and dissolved in 3 mL of acetone. D-tartaric acid ( 0.9 g ) was dissolved in 2 mL of DMF and added to the acetone solution. Crystallization was achieved using the same procedure as with (-)-34 to give the D-tartrate salt of (+)-34 as a solid, mp 197-199 ${ }^{\circ} \mathrm{C}$, $[\alpha]_{25}{ }^{\mathrm{D}}=+46.9^{\circ}(c 0.48, \mathrm{MeOH})$. After conversion to the free base, $(+)-34$ was isolated as a yellow oil, $[\alpha]_{25}{ }^{\mathrm{D}}+44.4^{\circ}(c 1.65, \mathrm{MeOH}) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz} ; \mathrm{C}_{6} \mathrm{H}_{6}\right): \delta 8.27(\mathrm{~d}, J=8.6 \mathrm{~Hz}$, $1 \mathrm{H}), 7.18-7.10(\mathrm{~m}, 8 \mathrm{H}), 7.04-7.01(\mathrm{~m}, 1 \mathrm{H}), 6.68(\mathrm{~s}, 1 \mathrm{H}), 6.45(\mathrm{dd}, J=8.7,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.11(\mathrm{~s}$, $1 \mathrm{H}), 3.28(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.15(\mathrm{~s}, 3 \mathrm{H}), 3.10-2.95(\mathrm{~m}, 2 \mathrm{H}), 2.74$ (ddd, $J=13.3,8.5,5.3 \mathrm{~Hz}$, $1 \mathrm{H}), 2.59(\mathrm{ddd}, J=12.1,9.0,5.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.45(\mathrm{dd}, J=11.9,4.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.08(\mathrm{td}, J=12.4$, $3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.89-1.84(\mathrm{~m}, 1 \mathrm{H}), 1.69(\mathrm{td}, J=12.8,5.1 \mathrm{~Hz}, 1 \mathrm{H}), 1.17(\mathrm{~d}, J=12.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.02(\mathrm{~s}$, 3H), $0.68(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz;} \mathrm{C}_{6} \mathrm{D}_{6}$ ): $\delta 193.6,164.5,148.6,140.7,128.8$, 128.5, 128.2, 127.8, 127.51, 127.39, 125.8, 112.2, 111.1, 67.3, 56.9, 54.4, 46.2, 43.9, 41.6, 37.2, 34.0, 25.1, 13.3. HRMS: [ $\left.\mathrm{C}_{23} \mathrm{H}_{27} \mathrm{NO}_{2}\right] \mathrm{H}^{+}$Calcd: 350.2120; Found: 350.2118. The absolute configuration of (+)-34 was established as $2 R, 6 S, 11 S$ by X-ray crystallographic analysis of the HBr salt (Figure 3). Anal. Calcd. For $\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{BrNO}_{2} \bullet 0.8 \mathrm{H}_{2} \mathrm{O} \bullet 0.25 \mathrm{CH}_{4} \mathrm{O}: \mathrm{C} 61.67 ; \mathrm{H}$ 6.81; N 3.09. Found C 61.71; H 6.69; N 2.97.
(2R,6S,11S)-8-Methoxy-6,11-dimethyl-3-phenethyl-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azo cin-1-ol ( $(2 R, 6 S, 11 S)-(+)-35)$. In a flame-dried flask, (+)-34 ( $0.580 \mathrm{~g}, 1$ equiv, 1.66 mmol ) was dissolved in THF $(10 \mathrm{~mL})$ and cooled to $0^{\circ} \mathrm{C}$. Lithium aluminum hydride ( $75.6 \mathrm{mg}, 1.2$ equiv, 1.99 mmol ) was added, and the reaction was allowed to warm slowly to room temperature. The reaction was quenched successively with $\mathrm{iPrOH}, \mathrm{MeOH}$ and water, followed by dilution with $\mathrm{CHCl}_{3}$. The heterogeneous mixture was diluted further with brine and extracted $3 x$ with $\mathrm{CHCl}_{3}$. The combined organic phases were washed with ammonium chloride, dried, and the concentrated residue was purified by flash chromatography. The alcohol (+)-35 ( $0.508 \mathrm{~g}, 1.45 \mathrm{mmol}, 87.1 \%$ ) was isolated as a colorless oil, and only a single diastereomer was observed in the NMR. The optical rotation was essentially the same as that determined for $(-)-35$, except for the sign of rotation, as expected for an enantiomeric pair. $[\alpha]_{25}{ }^{\mathrm{D}}=+52.3^{\circ}(\mathrm{c}, 2.8, \mathrm{MeOH}) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right): \delta 7.47(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H})$, $7.29(\mathrm{t}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.22-7.18(\mathrm{~m}, 3 \mathrm{H}), 6.80(\mathrm{dd}, J=8.5,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.69(\mathrm{~d}, J=2.6 \mathrm{~Hz}$, $1 \mathrm{H}), 4.62(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.14-3.08(\mathrm{~m}, 1 \mathrm{H}), 2.93-2.86(\mathrm{~m}, 2 \mathrm{H}), 2.78(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H})$, $2.49(\mathrm{dd}, J=9.9,2.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.08-2.02(\mathrm{~m}, 1 \mathrm{H}), 1.83-1.76(\mathrm{~m}, 1 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H}), 1.01(\mathrm{dt}$, $J=13.0,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 0.82(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(101 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta 158.86,141.26$, $140.04,132.89,128.70,128.38,126.09,111.22,110.89,63.80,62.96,55.98,55.11,43.61,37.61$, 37.54, 37.19, 35.95, 25.98, 13.03. HRMS [ $\left.\mathrm{C}_{23} \mathrm{H}_{29} \mathrm{NO}_{2}\right] \mathrm{H}^{+}$Calcd: 352.2277; Found: 352.2277.
(2S,6R,11R)-8-Methoxy-6,11-dimethyl-3-phenethyl-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azo cin-1-ol ( $2 S, 6 R, 11 R-(-)-35)$. In a flame-dried flask, ( - )-34 ( $0.550 \mathrm{~g}, 1$ equiv, 1.57 mmol ) was dissolved in THF $(10 \mathrm{~mL})$ and cooled to $0^{\circ} \mathrm{C}$. Lithium aluminum hydride $(71.7 \mathrm{mg}, 1.2$ equiv, 1.89 mmol ) was added, and the reaction was allowed to warm slowly to room temperature. The reaction was quenched successively with $\mathrm{iPrOH}, \mathrm{MeOH}$ and $\mathrm{H}_{2} \mathrm{O}$, followed by addition of $\mathrm{CHCl}_{3}$. Brine was added to the heterogeneous mixture, and it was extracted $3 x$ with $\mathrm{CHCl}_{3}$. The combined organic phases were washed with $\mathrm{NH}_{4} \mathrm{OH}$, dried, and purified by flash chromatography. The alcohol ( - )-35 ( $0.398 \mathrm{~g}, 1.13 \mathrm{mmol}, 71.9 \%$ ) was isolated as a colorless oil, $[\alpha]_{25}{ }^{\mathrm{D}}-51.7^{\circ}(\mathrm{c}, 2.08, \mathrm{MeOH}) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right): \delta 7.47(\mathrm{~d}$, $J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.29(\mathrm{t}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.22-7.18(\mathrm{~m}, 3 \mathrm{H}), 6.80(\mathrm{dd}, J=8.5,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.69$ $(\mathrm{d}, J=2.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.62(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.11(\mathrm{dt}, J=13.0,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.93-2.86(\mathrm{~m}, 2 \mathrm{H})$, $2.78(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.49(\mathrm{dd}, J=9.7,2.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.08-2.02(\mathrm{~m}, 1 \mathrm{H}), 1.83-1.76(\mathrm{~m}, 1 \mathrm{H})$, $1.32(\mathrm{~s}, 3 \mathrm{H}), 1.01(\mathrm{dt}, J=13.0,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 0.82(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(101 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): \delta 158.87,141.25,140.04,132.89,128.69,128.38,126.08,111.23,110.89,63.81,62.96$, $55.98,55.11,43.62,37.61,37.54,37.20,35.96,25.98,13.03$. HRMS: $\left[\mathrm{C}_{23} \mathrm{H}_{29} \mathrm{NO}_{2}\right] \mathrm{H}^{+}$Calcd: 352.2277; Found: 352.2279.
(2R,6S,11S)-6,11-Dimethyl-3-phenethyl-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocine-1,8-diol $(2 R, 6 S, 11 S-(+)-36)$. In a flame-dried flask, the aromatic ether (+)-35 ( $0.493 \mathrm{~g}, 1$ equiv, $1.40 \mathrm{mmol})$ was dissolved in dichloromethane $(10 \mathrm{~mL})$ and cooled to $-78^{\circ} \mathrm{C}$. To this solution, boron tribromide ( $703 \mathrm{mg}, 265 \mu \mathrm{~L}, 2$ equiv, 2.81 mmol ) was added, and the reaction was allowed to warm to room temperature. The reaction was cooled to $0^{\circ} \mathrm{C}$ and quenched with MeOH followed by 2 M HCl . The reaction was allowed to stir for 1 h and then basified with a saturated sodium bicarbonate solution. The aqueous phase was separated and extracted $3 x$ with $\mathrm{CHCl}_{3}$, followed by washing the combined organic phases with a saturated sodium bicarbonate solution. After drying, the organic phase was concentrated in vacuo, and the residue purified via flash chromatography, eluting with $7 \%$ CMA in $\mathrm{CHCl}_{3}$, to give the phenol (+)-36 ( $0.312 \mathrm{~g}, 925 \mu \mathrm{~mol}, 65.9 \%$ ) as a white foam, $[\alpha]_{25}{ }^{\mathrm{D}}+58.1^{\circ}$ (c 1.14, MeOH). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right): \delta 7.40(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.31-7.18(\mathrm{~m}, 8 \mathrm{H})$, $6.72-6.69(\mathrm{~m}, 2 \mathrm{H}), 6.63(\mathrm{~d}, J=2.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.62(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.54(\mathrm{~s}, 1 \mathrm{H}), 3.11(\mathrm{td}$, $J=10.8,5.3 \mathrm{~Hz}, 2 \mathrm{H}), 2.93-2.86(\mathrm{~m}, 3 \mathrm{H}), 2.78(\mathrm{q}, J=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.51(\mathrm{dd}, J=8.9,2.2 \mathrm{~Hz}$, $2 \mathrm{H}), 2.04(\mathrm{dq}, J=10.6,3.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.83-1.75(\mathrm{~m}, 2 \mathrm{H}), 1.34(\mathrm{~s}, 1 \mathrm{H}), 1.31(\mathrm{~d}, J=12.9 \mathrm{~Hz}, 4 \mathrm{H})$, $1.01(\mathrm{dt}, J=13.0,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 0.87(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 1 \mathrm{H}), 0.82(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ (101 MHz, $\mathrm{CDCl}_{3}$ ): $\delta 154.99,141.50,139.99,132.58,130.07,128.75,128.68,128.62,128.40$, 128.36, 126.11, 113.64, 111.53, 63.81, 62.91, 55.97, 43.60, 37.52, 37.49, 37.14, 35.93, 25.95, 13.00. HRMS: $\left[\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{NO}_{2}\right]^{+}$Calcd: 338.2120; Found: 338.2116. The HCl salt of the phenol (+)-36 was formed by dissolving the free base in a minimal amount of acetone, followed by the addition of 6 drops of concentrated HCl . The solution was concentrated to a thick wax that was triturated in diethyl ether at room temperature overnight to provide the HCl salt as a free-flowing off-white solid. Anal. Calcd. For $\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{ClNO}_{2} \bullet 0.7 \mathrm{H}_{2} \mathrm{O}: \mathrm{C}, 68.36 ; \mathrm{H}, 7.67$; N, 3.62. Found C, 68.53; H, 7.81; N, 3.45.
(2S,6R,11R)-6,11-Dimethyl-3-phenethyl-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocine-1,8-diol $(2 S, 6 R, 11 R)-(-)-36$. In a flame-dried flask, the aromatic ether ((-)-35 (0.396 g, 1 equiv, $1.13 \mathrm{mmol})$ was dissolved in dichloromethane ( 10 mL ) and cooled to $-78^{\circ} \mathrm{C}$. To this solution, boron tribromide ( $0.564 \mathrm{mg}, 213 \mu \mathrm{~L}, 2$ equiv, 2.25 mmol ) was added, and the reaction was allowed to warm to room temperature. The reaction was cooled to $0^{\circ} \mathrm{C}$ and quenched with MeOH followed by 2 M HCl . The reaction was allowed to stir for 1 h and then basified with a saturated sodium bicarbonate solution. The aqueous phase was extracted 3 x with $\mathrm{CHCl}_{3}$, and the combined organic phases were washed with a saturated sodium bicarbonate solution. After drying, the organic phase was purified by flash chromatography, eluting with $7 \%$ CMA in $\mathrm{CHCl}_{3}$, to give ( - )-36 as a white foam ( $0.190 \mathrm{~g}, 563 \mu \mathrm{~mol}, 50.0 \%$ ). $[\alpha]_{25}{ }^{\mathrm{D}}$ $-57.4^{\circ}(c 1.45, \mathrm{MeOH}){ }^{1} \mathrm{H}-\mathrm{NMR}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right): \delta 7.40(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.31-7.16(\mathrm{~m}$, $8 \mathrm{H}), 6.70$ (td, $J=6.0,2.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.63(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.62(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.17$ (s, 1H), $3.54(\mathrm{~s}, 1 \mathrm{H}), 3.11(\mathrm{dt}, J=13.0,6.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.89(\mathrm{dq}, J=11.4,7.2 \mathrm{~Hz}, 3 \mathrm{H}), 2.82-2.74(\mathrm{~m}, 3 \mathrm{H})$, $2.50(\mathrm{t}, ~ J=4.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.04(\mathrm{dtd}, J=14.2,7.2,3.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.86-1.75(\mathrm{~m}, 2 \mathrm{H}), 1.36-1.33(\mathrm{~m}$, $2 \mathrm{H}), 1.30(\mathrm{~s}, 3 \mathrm{H}), 1.01(\mathrm{dt}, J=12.9,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 0.88(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 0.82(\mathrm{~d}, J=7.1 \mathrm{~Hz}$, $3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 155.12,141.48,139.97,132.39,130.07,128.75,128.68$, $128.61,128.40,128.37,126.12,126.09,113.69,111.73,111.56,73.50,63.83,62.90,59.67,57.78$, $55.96,43.60,37.49,37.13,35.92,25.96,14.38,13.00$.
(2S,6R,11R)-8-Methoxy-1,6,11-trimethyl-3-phenethyl-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]a zocin-1-ol ( $2 S, 6 R, 11 R-(-)-37)$. In an argon-charged, flame-dried flask, ((-)-34 ( $0.5 \mathrm{~g}, 1$ equiv, 1 mmol ) was dissolved in THF ( 15 mL ) and cooled to $-78^{\circ} \mathrm{C}$. Methyllithium ( $0.09 \mathrm{~g}, 3 \mathrm{~mL}$, 1.3 molar, 3 equiv, 4 mmol ) was added, and the reaction was warmed to room temperature overnight. The reaction mixture was quenched with a concentrated $\mathrm{NH}_{4} \mathrm{Cl}$ solution, and it was extracted with $\mathrm{CHCl}_{3}$. The aqueous phase was separated and extracted with $\mathrm{CHCl}_{3}$ $2 x$, and the combined organic phases were washed with brine. The organic phase was then dried with sodium sulfate, filtered, and purified via flash chromatography.to give $(-)-37$ as a dark amber oil ( $0.454 \mathrm{~g}, 1.24 \mathrm{mmol}, 90 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ): $\delta 7.52$ $(\mathrm{d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.32-7.25(\mathrm{~m}, 3 \mathrm{H}), 7.24-7.19(\mathrm{~m}, 3 \mathrm{H}), 6.83(\mathrm{dt}, J=8.6,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.69$ $(\mathrm{d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.80(\mathrm{~d}, J=6.4 \mathrm{~Hz}, 3 \mathrm{H}), 3.14-3.07(\mathrm{~m}, 1 \mathrm{H}), 2.98-2.89(\mathrm{~m}, 2 \mathrm{H}), 2.82-2.73$
$(\mathrm{m}, 3 \mathrm{H}), 2.66-2.42(\mathrm{~m}, 4 \mathrm{H}), 2.15(\mathrm{qd}, J=7.2,3.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.82(\mathrm{td}, J=13.2,4.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.54$ (d, $J=7.7 \mathrm{~Hz}, 3 \mathrm{H}), 1.34(\mathrm{~d}, J=10.3 \mathrm{~Hz}, 3 \mathrm{H}), 0.95(\mathrm{t}, J=8.6 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(101 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ): $\delta 128.85,128.70,128.42,128.15,127.79,126.15,111.64,110.83,69.26,67.29,60.31$, $55.76,55.10,55.08,43.38,43.34,37.97,37.95,36.99,35.99,33.62,26.32,23.69,16.35$. HRMS: [ $\left.\mathrm{C}_{24} \mathrm{H}_{32} \mathrm{NO}_{2}\right]^{+}$Calcd: 366.2433; found: 366.2437.
(2R,6S,11S)-8-Methoxy-1,6,11-trimethyl-3-phenethyl-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]a zocin-1-ol ( $2 R, 6 S, 11 S-(+)-37)$. In an argon-charged flame-dried flask, (+)-34 ( $0.5 \mathrm{~g}, 1$ equiv, 1 mmol ) was dissolved in THF ( 15 mL ) and cooled to $-78^{\circ} \mathrm{C}$. Methyllithium ( $0.09 \mathrm{~g}, 3 \mathrm{~mL}$, 1.3 molar, 3 equiv, 4 mmol ) was added, and the reaction was allowed to warm to room temperature. The reaction mixture was quenched with a concentrated aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ solution and extracted with $\mathrm{CHCl}_{3}$. The aqueous phase was separated and extracted 2 x with $\mathrm{CHCl}_{3}$, and the combined organic phases were washed with brine. The organic phase was then dried with sodium sulfate, filtered, and purified by flash chromatography to give $(+)-37$ as a dark amber oil ( $0.51 \mathrm{~g}, 1.4 \mathrm{mmol}, 100 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ): $\delta 7.52$ (d, $J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.31-7.23(\mathrm{~m}, 3 \mathrm{H}), 7.23-7.19(\mathrm{~m}, 3 \mathrm{H}), 6.84(\mathrm{dt}, J=8.6,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.69$ (d, $J=2.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.79(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 3 \mathrm{H}), 3.13-3.06(\mathrm{~m}, 1 \mathrm{H}), 2.98-2.88(\mathrm{~m}, 2 \mathrm{H}), 2.84-2.77$ $(\mathrm{m}, 2 \mathrm{H}), 2.65(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.59-2.45(\mathrm{~m}, 2 \mathrm{H}), 2.15(\mathrm{qd}, J=7.0,3.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.81(\mathrm{td}$, $J=13.2,4.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.53(\mathrm{~s}, 3 \mathrm{H}), 1.33(\mathrm{~s}, 3 \mathrm{H}), 1.07-1.04(\mathrm{~m}, 1 \mathrm{H}), 0.95(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 158.67,140.41,128.82,128.69,128.40,128.21,127.77,126.13$, $111.88,111.64,110.82,69.25,67.27,55.80,55.10,43.42,38.09,38.07,37.99,37.02,36.03,33.62$, 26.33, 16.36. HRMS: $\left[\mathrm{C}_{24} \mathrm{H}_{32} \mathrm{NO}_{2}\right]^{+}$Calcd: 366.2433; found: 366.2433.
(2R,6S,11S)-6,11-Dimethyl-1-methylene-3-phenethyl-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]az ocin-8-ol ( $2 R, 6 S, 11 S-(+)-38)$. In a flame-dried flask, (+)-37 ( $0.5 \mathrm{~g}, 1$ equiv, 1 mmol ) was dissolved in dichloromethane ( 7 mL ) and cooled to $-78{ }^{\circ} \mathrm{C}$. To this solution, boron tribromide ( $498 \mathrm{mg}, 0.188 \mathrm{~mL}, 1$ equiv, 1.99 mmol ) was added, and the reaction was allowed to warm to room temperature. The reaction was quenched with MeOH followed by an equal volume of 2 M HCl . The solution was heated to $75^{\circ} \mathrm{C}$ for 2 h , followed by addition of a sodium bicarbonate to a pH of 7.5-8. The aqueous phase was extracted with $\mathrm{CHCl}_{3} 3 x$. The combined organic phases were washed with a saturated sodium bicarbonate solution and brine, then dried with sodium sulfate. The organic phase was purified by flash chromatography, eluting with $15 \% \mathrm{CMA}$ in $\mathrm{CHCl}_{3}$ to give (+)-38. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 7.53$ (d, $J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.23-7.20(\mathrm{~m}, 2 \mathrm{H}), 7.13(\mathrm{td}, J=6.8,2.9 \mathrm{~Hz}, 3 \mathrm{H}), 6.67(\mathrm{~d}, J=2.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.60$ (dd, $J=8.6,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.68(\mathrm{~s}, 1 \mathrm{H}), 4.80(\mathrm{~s}, 1 \mathrm{H}), 3.33(\mathrm{~s}, 1 \mathrm{H}), 3.30(\mathrm{t}, J=3.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.83$ $(\mathrm{dq}, J=6.6,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.76-2.68(\mathrm{~m}, 2 \mathrm{H}), 2.61-2.57(\mathrm{~m}, 1 \mathrm{H}), 2.46-2.41(\mathrm{~m}, 1 \mathrm{H}), 1.97-1.90$ $(\mathrm{m}, 2 \mathrm{H}), 1.84(\mathrm{dt}, J=12.7,6.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.34-1.31(\mathrm{~m}, 3 \mathrm{H}), 0.78(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta 157.92,142.09,140.03,136.67,128.27,128.02,126.59,125.70,123.34$, $113.34,111.82,110.61,65.90,56.79,45.86,41.80,41.78,36.94,33.12,24.66,13.24$. HRMS: [ $\left.\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{NO}\right]^{+}$Calculated: 334.2171 ; Found: 334.2165 . The HCl salt of (+)-38 was obtained by dissolving the free base in a minimal amount of acetone. The solution was concentrated to a pale syrup in vacuo, followed by addition of diethyl ether. The mixture was stirred overnight to give $(+)-38 \bullet \mathrm{HCl}$ as a pale beige solid. Anal. Calcd. For $\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{ClNO} \bullet 0.7$ $\mathrm{H}_{2} \mathrm{O}$ C, $72.21 ; \mathrm{H}, 7.75 ; \mathrm{N}, 3.66$. Found C, 72.13; H, 7.55; N, 3.55.
(2S,6R,11R)-6,11-Dimethyl-1-methylene-3-phenethyl-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]a zocin-8-ol (2S,6R,11R-(-)-38). In a flame-dried flask, (-)-37 ( $0.454 \mathrm{~g}, 1$ equiv, 1.24 mmol ) was dissolved in dichloromethane ( 7 mL ) and cooled to $-78{ }^{\circ} \mathrm{C}$. To this solution, boron tribromide ( $615 \mathrm{mg}, 0.232 \mathrm{~mL}, 1.98$ equiv, 2.45 mmol ) was added, and the reaction was allowed to warm to room temperature. The reaction was quenched with MeOH followed by an equal volume of 2 M HCl . The solution was heated to $75^{\circ} \mathrm{C}$ for 2 h , followed by addition of a saturated sodium bicarbonate solution to a pH of 7.5-8. Chloroform was added, and the aqueous phase was extracted with $\mathrm{CHCl}_{3} 3 x$. The combined organic phase was washed with a saturated sodium bicarbonate solution and brine, then dried with sodium sulfate. The organic phase was purified by flash chromatography, eluting with $15 \%$ CMA in $\mathrm{CHCl}_{3}$ to give $(-)-38(0.125 \mathrm{~g}, 375 \mu \mathrm{~mol}, 30.2 \%) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta$
$7.55(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.25-7.22(\mathrm{~m}, 2 \mathrm{H}), 7.17-7.12(\mathrm{~m}, 3 \mathrm{H}), 6.67(\mathrm{~d}, J=2.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.60$ $(\mathrm{dd}, J=8.6,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.71(\mathrm{~s}, 1 \mathrm{H}), 4.84(\mathrm{~s}, 1 \mathrm{H}), 3.34(\mathrm{~d}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.86(\mathrm{dt}, J=11.6$, $5.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.79-2.68(\mathrm{~m}, 2 \mathrm{H}), 2.65-2.61(\mathrm{~m}, 1 \mathrm{H}), 2.47(\mathrm{td}, J=13.0,6.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.00-1.91(\mathrm{~m}$, $2 \mathrm{H}), 1.85(\mathrm{td}, J=12.8,4.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.34(\mathrm{~s}, 3 \mathrm{H}), 0.80(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 101 MHz , $\left.\mathrm{CD}_{3} \mathrm{OD}\right): ~ \delta 159.35,143.52,141.46,138.12,129.69,129.44,128.04,127.12,124.76,114.74,113.21$, 112.05, 67.35, 58.22, 47.30, 43.23, 43.21, 38.37, 34.53, 26.05, 14.61. HRMS: [ $\left.\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{NO}\right]^{+}$Calcd: 334.2171; found: 334.2165 . The $-(-)-38 \bullet \mathrm{HCl}$ salt was formed by dissolving the free base in a minimal amount of acetone. The solution was concentrated to a pale syrup, and diethyl ether was added. The mixture was stirred overnight to give $(-)-38 \bullet \mathrm{HCl}$ as a pale beige solid. Anal. Calcd. For $\mathrm{C}_{23} \mathrm{H}_{28} \mathrm{ClNO} \bullet 0.4 \mathrm{H}_{2} \mathrm{O}$ C, $73.25 ; \mathrm{H}, 7.70 ; \mathrm{N}, 3.71$. Found: C, 73.24; H, 7.82; N, 3.57.
(2R,6S,11S)-1,6,11-Trimethyl-3-phenethyl-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin-8-ol ( $2 R, 6 S, 11 S-(+)-39)$. In a Parr Shaker flask, (+)-38 ( $0.107 \mathrm{~g}, 1$ equiv, $321 \mu \mathrm{~mol}$ ), palladium on carbon ( $205 \mathrm{mg}, 5 \% \mathrm{wt}, 0.3$ equiv, $96.3 \mu \mathrm{~mol}$ ), and acetic acid ( $18.4 \mu \mathrm{~L}, 1$ equiv, $321 \mu \mathrm{~mol}$ ) were dissolved in methanol ( 10 mL ) under an argon atmosphere. The vessel was charged with 30 psi of hydrogen gas and shaken overnight at room temperature. The reaction mixture was filtered through celite and rinsed multiple times with methanol. The filtrate was purified via flash chromatography to give (+)-39 as a light brown foam ( $56 \mathrm{mg}, 0.17 \mathrm{mmol}$, $52 \%) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ): $\delta 7.29-7.25(\mathrm{~m}, 2 \mathrm{H}), 7.22-7.19(\mathrm{~m}, 3 \mathrm{H}), 7.05(\mathrm{~d}, J=9.2 \mathrm{~Hz}$, 1H), 6.76-6.73 (m, 2H), $3.22(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.05-2.93(\mathrm{~m}, 5 \mathrm{H}), 2.89(\mathrm{dd}, J=12.2,3.9 \mathrm{~Hz}$, $1 \mathrm{H}), 2.26(\mathrm{td}, J=12.6,3.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.19-2.13(\mathrm{~m}, 1 \mathrm{H}), 2.11(\mathrm{~s}, 2 \mathrm{H}), 1.95(\mathrm{td}, J=13.3,4.7 \mathrm{~Hz}$, $1 \mathrm{H}), 1.37(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}), 1.29(\mathrm{~s}, 3 \mathrm{H}), 0.87(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 155.80,141.25,138.45,130.41,128.76,128.61,128.20,126.57,114.10,112.31,64.12,55.48$, $45.33,40.38,39.36,36.13,32.46,25.47,24.12,15.36$. HRMS: $\left[\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{NO}\right]^{+}$Calcd: 336.2327; Found: 336.2327. The free base was dissolved in a minimal amount of acetone, and 6 drops of concentrated HCl was added. Upon stirring, a syrup formed. Diethyl ether was added, and the mixture was stirred overnight to give (+)-39. HCl as an off-white solid ( 32 mg ), mp $178-181^{\circ} \mathrm{C},[\alpha]_{25}{ }^{\mathrm{D}}=+70.7^{\circ}$ (c 0.39, MeOH). Anal. Calcd. For $\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{ClNO} \bullet 0.75 \mathrm{H}_{2} \mathrm{O} \mathrm{C}$, 71.67 H, $8.24 ;$ N, 3.63. Found: C, 71.83; H, 8.40; N:3.54.
(2S,6R,11R)-1,6,11-Trimethyl-3-phenethyl-1,2,3,4,5,6-hexahydro-2,6-methanobenzo[d]azocin-8-ol $(2 S, 6 R, 11 R-(-)-39)$. In a Parr Shaker flask, (-)-38 ( $0.120 \mathrm{~g}, 1$ equiv, $360 \mu \mathrm{~mol})$, palladium on carbon ( $230 \mathrm{mg}, 5 \% \mathrm{wt}, 0.3$ equiv, $108 \mu \mathrm{~mol}$ ), and acetic acid ( $20.6 \mu \mathrm{~L}$, 1 equiv, $360 \mu \mathrm{~mol}$ ) were added to methanol ( 10 mL ) under an argon atmosphere. The vessel was charged with 30 psi of hydrogen gas and shaken overnight at room temperature. The reaction mixture was filtered through celite and rinsed multiple times with methanol. The filtrate was purified via flash chromatography to give ( - )-39 as a light brown foam ( $71 \mathrm{mg}, 0.21 \mathrm{mmol}$, $59 \%),{ }^{1}{ }^{H}$ NMR (400 MHz; $\mathrm{CDCl}_{3}$ ): $\delta 7.31-7.27(\mathrm{~m}, 2 \mathrm{H}), 7.27-7.20(\mathrm{~m}, 3 \mathrm{H}), 7.07(\mathrm{~d}, J=8.2 \mathrm{~Hz}$, $1 \mathrm{H}), 6.83-6.81(\mathrm{~m}, 2 \mathrm{H}), 3.32(\mathrm{~s}, 1 \mathrm{H}), 3.13-3.00(\mathrm{~m}, 4 \mathrm{H}), 3.00-2.94(\mathrm{~m}, 1 \mathrm{H}), 2.41-2.35(\mathrm{~m}, 2 \mathrm{H})$, $2.16-2.08(\mathrm{~m}, 2 \mathrm{H}), 1.40(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}), 1.33(\mathrm{~s}, 2 \mathrm{H}), 0.89(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (101 MHz, $\mathrm{CDCl}_{3}$ ): $\delta 155.84,140.71,137.58,128.78,128.74,128.31,126.86,114.42,112.38$, $64.56,55.44,45.69,39.72,38.76,35.96,31.86,25.19,24.08,15.24$. HRMS: $\left[\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{NO}\right]^{+}$Calcd: 336.2327; Found: 336.2326. The free base was dissolved in a minimal amount of acetone, and 6 drops of concentrated HCl was added. Diethyl ether was added to the syrup that was obtained, and the mixture was stirred overnight to give $(-)-39 \bullet \mathrm{HCl}$ as an off-white solid ( 57 mg ), mp 181-184 ${ }^{\circ} \mathrm{C},[\alpha]_{25}{ }^{\mathrm{D}}-69.1^{\circ}$ (c 0.59, MeOH). Anal. Calcd. For $\mathrm{C}_{23} \mathrm{H}_{30} \mathrm{ClNO} \bullet 0.75$ $\mathrm{H}_{2} \mathrm{O}$ C, 71.67 ; H, $8.24 ; \mathrm{N}, 3.63$. Found: C, 71.73; H, 8.34; N, 3.48.

### 3.3. X-ray Crystal Structure Experimental Data

Single-crystal X-ray diffraction data on compound (+)-34 were collected using $\mathrm{Cu} \mathrm{K} \alpha$ radiation and a Bruker SMART APEX II CCD area detector. The crystal was prepared for data collection by coating with high-viscosity microscope oil. The oil-coated crystal was mounted on a micromesh mount (MiTeGen, Inc., Ithaca, NY, USA) and transferred to the diffractometer, and a data set collected at 293(2) K. The $0.097 \times 0.074 \times 0.060 \mathrm{~mm}^{3}$
crystal was monoclinic in space group $\mathrm{P} 2{ }_{1}$ with unit cell dimensions $\mathrm{a}=11.2668(6) \AA$, $\mathrm{b}=7.8053(4) \AA, \mathrm{c}=13.0428(6) \AA, \alpha=\gamma=90^{\circ}$ and $\beta=97.890(2)^{\circ}$. The final anisotropic full matrix least-squares refinement on $\mathrm{F}^{2}$ with 272 variables converged at $\mathrm{R}_{1}=5.68 \%$ for the observed data and $w R 2=15.01 \%$ for all data. The structure was solved by direct methods and refined by full-matrix least-squares on $\mathrm{F}^{2}$ values using the programs found in the SHELXL suite (Bruker, SHELXL v2014.7, 2014, Bruker AXS Inc., Madison, WI, USA). Corrections were applied for Lorentz, polarization, and absorption effects. Parameters refined included atomic coordinates and anisotropic thermal parameters for all nonhydrogen atoms. The H atoms were included using a riding model and direct assignment. The hydrogen atom found on the solvent methanol (O25) required a DFIX command to prevent migration. This is likely due to very minor positional disorder that could not be modeled to completion. Stereochemistry of C1 (R), C5 (S) and C9 (S) was determined via the orientation of the molecules. Complete information on data collection and refinement is available in the Supplementary Materials, Tables S1-S7.

Atomic coordinates for (+)-34 have been deposited with the Cambridge Crystallographic Data Centre, deposition number 2290877. Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge, CB2 1EZ, UK (fax: +44(0)-1223-336033 or e-mail: deposit@ccdc.cam.ac.uk).

### 3.4. In Vitro Assay

### 3.4.1. Cell Lines and Cell Culture

Cell lines and cell culture: cAMP Hunter ${ }^{\text {TM }}$ Chinese hamster ovary cells (CHO-K1) that express human $\mu$-opioid receptor (OPRM1), human $\kappa$-opioid receptor (OPRMK1), and human $\delta$-receptor (OPRMD1) were purchased from Eurofins DiscoverX (Fremont, CA) and used for the forskolin-induced cAMP accumulation assays [16]. All cell lines were maintained in F-12 media with $10 \%$ fetal bovine serum (Life Technologies, Grand Island, NY, USA), $1 \%$ penicillin/streptomycin/L-glutamine (Life Technologies), and $4.800 \mu \mathrm{~g} / \mathrm{mL}$ Geneticin (Mirus Bio, Madison, WI, USA). All cells were grown at $37{ }^{\circ} \mathrm{C}$ and $5 \% \mathrm{CO}_{2}$ in a humidified incubator.

### 3.4.2. Forskolin-Induced cAMP Accumulation Assays

Assays were performed as previously described [30]. Briefly, 10,000 cells/well of cells were plated in 384 -well tissue culture plates and incubated overnight at $37{ }^{\circ} \mathrm{C}$ in $5 \% \mathrm{CO}_{2}$. Stock solutions of the compound were prepared in DMSO at 5 mM concentration, and then 9 to 10 concentrations of 100X working solutions were prepared by serial dilution using DMSO. 5 X working solutions were subsequently prepared using assay buffer consisting of Hank's Buffered Salt Solution, HEPES, and forskolin. For the agonist assay, cells were incubated at $37^{\circ} \mathrm{C}$ with compounds for 30 min at a 1 X final concentration. For the antagonist assay [28], cells were incubated at $37^{\circ} \mathrm{C}$ with compounds for 15 min before 30 min incubation at $37^{\circ} \mathrm{C}$ with the selected agonist at their $\mathrm{EC}_{50}$ or $\mathrm{EC}_{90}$ dose. Detection was performed by using the HitHunter cAMP Assay for Small Molecules by DiscoverX according to the manufacturer's directions, and the BioTek Synergy H1 hybrid plate reader (BioTek, Winooski, VT, USA) and Gen5 Software version 2.01 were used to quantify luminescence (BioTek, Winooski, VT, USA). To determine \% efficacy in forskolin-induced cAMP assays, data were blank subtracted with the vehicle control, followed by normalization to the forskolin control. Data were then analyzed in GraphPad Prism 8 (GraphPad, LaJolla, CA, USA) using nonlinear regression. Values are expressed as the mean $\pm$ SEM of at least three independent experiments. Degree of antagonism ( $\mathrm{I}_{\max }$ ) was normalized to naltrexone (MOR, DOR) or nor-BNI (KOR).

## 4. Conclusions

Highly potent MOR agonists were found with electron withdrawing (e.g., $\mathrm{NO}_{2},(-)-3$, $\mathrm{EC}_{50}=0.3 \mathrm{nM}, \% \mathrm{E}_{\max }=101 \%$ ) and electron donating (e.g., $\mathrm{OH},(-)-20, \mathrm{EC}_{50}=0.13 \mathrm{nM}, \%$ $\mathrm{E}_{\max }=102 \%$ ) substituents at the para position in the aromatic ring of the $N$-phenethyl moiety.

An ortho-positioned substituent (e.g., $\mathrm{F},(-)-8)$ was also extremely potent $\left(\mathrm{EC}_{50}=0.2 \mathrm{nM}\right.$, $\% \mathrm{E}_{\max }=101 \%$ ). The meta-substituted compounds were in general much less potent than their comparable ortho- and para-analogs with one exception-the meta-fluoro analog (-)$7\left(\mathrm{EC}_{50}=0.6 \mathrm{nM}, \% \mathrm{E}_{\max }=100 \%\right)$ was more potent than the para-fluoro analog $(-)-6$ $\left(\mathrm{EC}_{50}=1.8 \mathrm{nM}, \% \mathrm{E}_{\max }=100 \%\right)$. A clear pattern between agonist potency and electron donating or withdrawal effects was not observed.

Steric effects were not clearly observable since potent compounds were found with both more bulky and less bulky substituents. Compounds with bulky ortho and meta substituents (e.g., 4 and 5 ) were generally less potent than a compound with a comparable para substituent (6). The bulkier substituents-the 4-nitro ((-)-3) and 4-bromo ((-)-12) moieties, were potent, as were the less bulky 4-hydroxy ((-)-20) and 4-chloro ((-)-15) substituents. Addition of a second chloro atom in the aromatic ring $((-)-21$ and $(-)-22)$ increased possible steric interactions and decreased potency-the monochloro substituent at the ortho $(-)-17$ or para $(-)-\mathbf{1 5}$ position was at least three-fold more potent than the dichloro compounds.

A remarkable specificity of the chain length of the $N$-phenethyl moiety was observed. A single extra methylene changed the highly potent $N$-phenethyl compound $(-)-23\left(\mathrm{EC}_{50}=0.27, \% \mathrm{E}_{\max }=101 \%\right)$ into a relatively inactive $N$-phenylpropyl compound $(-)-30\left(\mathrm{EC}_{50}=70 \mathrm{nM}, \% \mathrm{E}_{\max }=101 \%\right)$. The synthesis of levorotatory C8-substituted N phenethylnormetazocines gave potent compounds, with the C8-methylene-substituted $N$-phenethylnormetazocine having subnanomolar potency. All the potent C8-substituted compounds were fully efficacious in vitro. These data could be summarized by noting that most substituents on the aromatic ring in the phenethyl moiety did not increase potency at MOR when compared with the unsubstituted aromatic ring in 23 . All of the substituents that were examined reduced potency, with only two exceptions, and those only slightly increased the potency of the unsubstituted compound 23. Similarly, there were no major effects observable at DOR and KOR from compounds with the added substituents when compared with the unsubstituted compound 23. Although the examined substituents on the aromatic ring in the $N$-phenethyl side chain were, with only two minor exceptions, unable to increase MOR potency or efficacy, the length of that side chain was of extreme importance-the addition of a single methylene group, to form an $N$-phenylpropyl side chain, resulted in over a 250 -fold reduction in potency at MOR.

Supplementary Materials: The following supporting information can be downloaded at: https: / /www.mdpi.com/article/10.3390/molecules28237709/s1, Figures S1-S40: ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$-NMR spectra of novel compounds, and Tables S1-S7: Crystal data, atomic coordinates from X-ray crystallographic determination of (+)-34.

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#### Abstract

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Conflicts of Interest: The authors declare that they have no known competing financial interest or personal relationships that could have appeared to influence the work reported in this paper.


## Abbreviations

| MOR | mu-opioid receptor |
| :--- | :--- |
| DOR | delta-opioid receptor |
| KOR | kappa-opioid receptor |
| cAMP | cyclic adenosine monophosphate |
| cAMP assay | Inhibition of forskolin-induced cAMP accumulation assay |

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