

# Synthesis and Characterization of Iron Bispyridine Bisdicyanamide, $\text{Fe}[\text{C}_5\text{H}_5\text{N}]_2[\text{N}(\text{CN})_2]_2$

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**Table S1.** Data of the single crystal refinement of  $\text{Fe}[\text{C}_5\text{H}_5\text{N}]_2[\text{N}(\text{CN})_2]_2$ .

Chemical formula	$\text{FeC}_{14}\text{N}_8\text{H}_{10}$
Formula weight (g/mol)	346.15
Crystal system	monoclinic
Space group (no.)	$I2/m$ (12)
Temperature (K)	100(2)
$a$ (Å)	7.453(7)
$b$ (Å)	13.167(13)
$c$ (Å)	8.522(6)
$\beta$ (°)	114.98(6)
$V$ (Å <sup>3</sup> )	758.1(12)
$Z$	2
Crystal shape and color	brown-beige block
$\rho_{\text{calc}}$ (g·cm <sup>-3</sup> )	1.516
Radiation, $\lambda$ (Å)	Mo-K $\alpha_1$ , 0.71073
Diffractometer	Bruker APEX CCD
Absorption correction	Multi-scan, SADABS 2014/5
$T_{\text{min}}, T_{\text{max}}$	0.4236, 0.7461
No. of measured, independent, and observed [ $I > 2\sigma(I)$ ] reflections	5814, 1159, 676
$R_1$	15.29%
$R_{1\text{obs}}$	7.42%
$R_{\text{all}}$	17.08%
$R_{1\text{all}}$	14.45%
GOF <sub>obs</sub>	1.057
No. of parameters, restraint, constraints	58, 0, 3
H-atom treatment	Constrained refinement

**Table S2.** Data of Rietveld Refinement of powderous [C<sub>5</sub>H<sub>5</sub>N]<sub>2</sub>[N(CN)<sub>2</sub>]<sub>2</sub> sample shown in Figure 2.

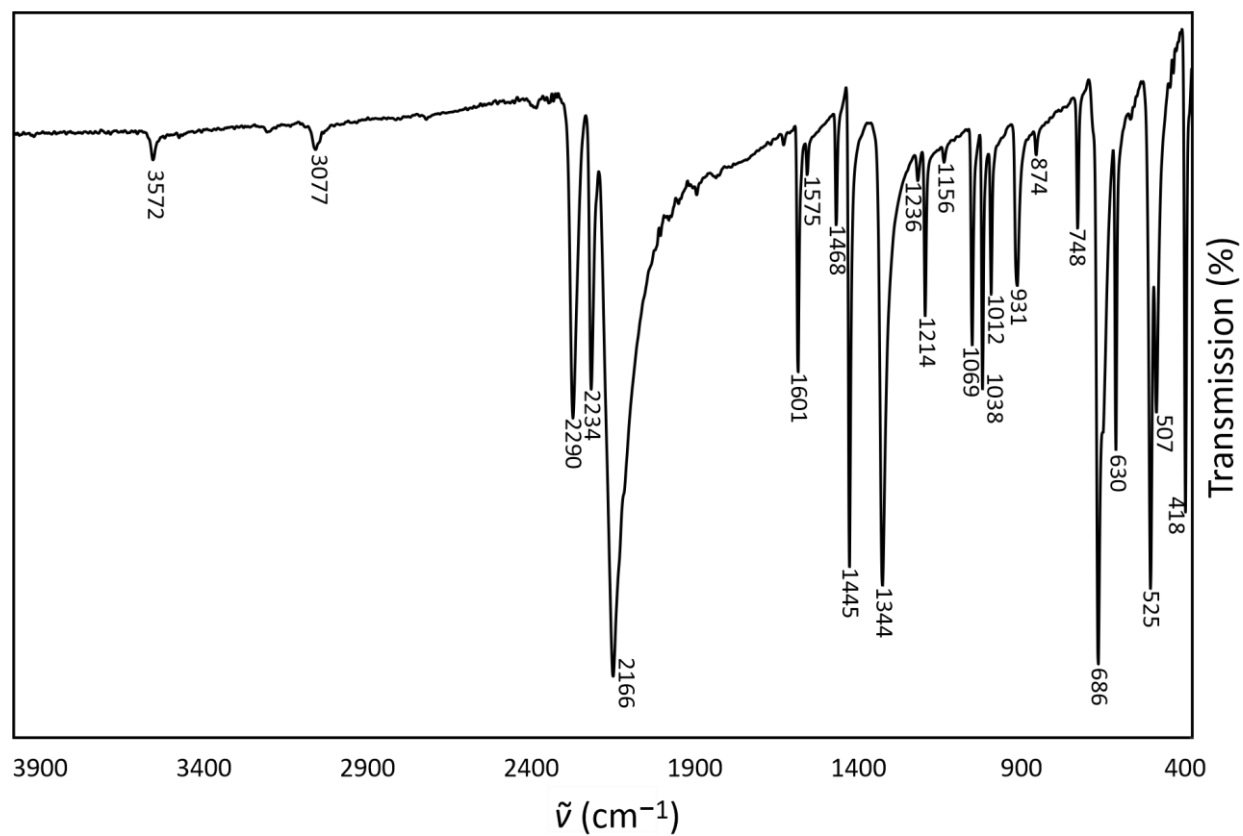
Chemical formula	FeC <sub>14</sub> N <sub>8</sub> H <sub>10</sub>	$\rho_{\text{calc}}$ (g·cm <sup>-3</sup> )	1.479
Formula weight	346.15	Radiation, $\lambda$ (Å)	Mo-K $\alpha_1$ , 0.71073
Crystal System	monoclinic	Profile function	pseudo-Voigt w. axial divergence asymmetry
Space group (No.)	<i>I</i> 2/ <i>m</i> (12)	<i>U</i>	5.616(38)
Temperature (K)	300	<i>V</i>	−0.675(16)
<i>a</i> (Å)	7.437(99)	<i>W</i>	0.022(38)
<i>b</i> (Å)	13.300(82)	<i>S<sub>L</sub>/S<sub>D</sub></i>	0.027(3)
<i>c</i> (Å)	8.652(34)	$\eta$	1.0
$\beta$ (°)	114.68(1)	<i>R<sub>p</sub></i> (%)	4.34
<i>V</i> (Å <sup>3</sup> )	777.36(17)	<i>R<sub>wp</sub></i> (%)	2.16
<i>Z</i>	2	<i>X</i> <sup>2</sup>	12.9

**Table S3.** Atomic positions from single crystal refinement.

Atom	Site	<i>x</i>	<i>y</i>	<i>z</i>
Fe1	2 <i>a</i>	0	0	0
N1	4 <i>i</i>	0.1390(6)	0	0.2837(6)
N2	8 <i>j</i>	0.2083(5)	0.1147(3)	0.9998(4)
N3	4 <i>h</i>	½	0.2015(4)	0
C1	8 <i>j</i>	0.1869(10)	0.0842(4)	0.3740(6)
C2	8 <i>j</i>	0.2840(11)	0.0866(5)	0.5525(7)
C3	4 <i>i</i>	0.3363(9)	0	0.6439(8)
C4	8 <i>j</i>	0.3483(6)	0.1515(3)	0.9998(5)
H1	8 <i>j</i>	0.1530	0.1470	0.3136
H2	8 <i>j</i>	0.3141	0.1500	0.6110
H3	4 <i>i</i>	0.4063	0	0.7665

**Table S4.** Anisotropic displacement parameters for iron, carbon and nitrogen and isotropic displacement parameters for hydrogen in 1 from single crystal refinement.

Atom	<i>U</i> <sub>11</sub>	<i>U</i> <sub>22</sub>	<i>U</i> <sub>33</sub>	<i>U</i> <sub>12</sub>	<i>U</i> <sub>13</sub>	<i>U</i> <sub>23</sub>
Fe1	0.0145(6)	0.0243(7)	0.0210(7)	0	0.0090(5)	0
N1	0.018(2)	0.024(3)	0.024(3)	0	0.010(2)	0
N2	0.023(2)	0.029(2)	0.027(2)	0.0057(15)	0.0097(17)	0.0016(15)
N3	0.029(3)	0.012(3)	0.069(4)	0	0.033(3)	0
C1	0.114(5)	0.030(3)	0.023(3)	−0.003(2)	−0.004(3)	0.034(3)
C2	0.138(7)	0.037(4)	0.029(3)	−0.011(3)	0.000(3)	0.040(4)
C3	0.023(3)	0.040(4)	0.027(3)	0	0.009(3)	0
C4	0.024(2)	0.016(2)	0.022(2)	0.0011(16)	0.0097(18)	0.0056(16)
<i>U</i> <sub>iso</sub>						
H1	0.080					
H2	0.096					
H3	0.037					



**Figure S1.** ATR-IR measurement of **1** between 4000 and 400  $\text{cm}^{-1}$ .

Table S5. ATR-IR for **1** with all assigned vibrations below 2500 cm<sup>-1</sup>.

$\tilde{\nu}$ (cm <sup>-1</sup> ) measured	$\tilde{\nu}$ (cm <sup>-1</sup> ) calculated	$\tilde{\nu}$ (cm <sup>-1</sup> ) Fe[N(CN) <sub>2</sub> ] <sub>2</sub> calculated	$\tilde{\nu}$ (cm <sup>-1</sup> ) Pyridine calculated	Vibration
418	420	435	404	$\gamma_s(\text{C-C}) + \nu(\text{Fe-N}) + \nu(\text{NC-N-CN})$
507/525	490/505	470	-	$\sigma_{as}(\text{N-C}\equiv\text{N})$
630/686	625/685	665	-	$\delta_s(\text{C-C})$
748	745	-	702/748	$\delta_s(\text{C-H})$
874*	-	-	-	$\gamma_s(\text{C-C}) + \nu(\text{Fe-N}) + \nu(\text{NC-N-CN})$
931	925	955	910	$\nu_s(\text{N-C})$
1012	1010	-	986	$\delta_{as}(\text{C-C})$
1038	1035	-	1045	$\delta_s(\text{C-H})$
1069	1070	-	1057	$\sigma_{as}(\text{N-C}\equiv\text{N}) + \sigma_s(\text{N-C}\equiv\text{N})$
1154*	-	-	-	$\delta_s(\text{C-C})$
1214	1201	-	1174	$\delta_{as}(\text{C-H})$
1234*	-	-	-	$\delta(\text{C-C})$
1344	1331	1310	-	$\nu_{as}(\text{N-C})$
1445/1469/1601	1435/1465/1596	-	1411/1452/1583	$\nu_s(\text{C-C})$
1574*	-	-	-	$\gamma_s(\text{C-H})$
1647*	-	-	-	$\nu_s(\text{N-C})$
2166*	-	-	-	$\sigma_{as}(\text{N-C}\equiv\text{N}) + \sigma_s(\text{N-C}\equiv\text{N})$
2234	-	2275	-	$\nu_{as}(\text{N-C}) + \nu_s(\text{N-C})$
2290	2239/2280	2305	-	$\nu_s(\text{N}\equiv\text{C})$
2409*	-	-	-	$\delta_{as}(\text{C-H})$

\*Overtone, cannot be calculated by DFT.