

## Supplementary Materials

**Table S1.** Apparent weight average molar masses  $M_{w,app}$  and polydispersity  $\mathcal{D}$  values obtained from sedimentation equilibrium of Butvar B-98.

Concentration (mg/ml)	$M_{w,app}$ (kDa)	$\mathcal{D}$ ( $M_{z,app}/M_{w,app}$ )
0.5	51.5	2.7
0.75	39.2	2.4
1.0	50.9	1.8
1.5	50.2	2.1
2.0	43.8	1.8
3.0	41.1	1.7
4.0	39.4	1.6

**Table S2.** Apparent weight average molar masses  $M_{w,app}$  and polydispersity  $\mathcal{D}$  values for PS36000 in isopropanol at different concentrations.

Concentration (mg/mL)	<sup>a</sup> $M_{w,app}$ (kDa)	<sup>b</sup> $M_{w,app}$ (kDa)	$\mathcal{D}$ ( $M_{z,app}/M_{w,app}$ )
0.5	48.7	44.4	1.0
1.0	49.0	38.2	1.3
2.0	35.6	34.4	1.6
3.0	29.1	28.9	1.0
4.0	27.7	26.8	1.3

a: M\* method. b: hinge point method.

**Table S3.** Apparent weight average molar masses  $M_{w,app}$  and polydispersity  $\mathcal{D}$  values for PS18000 in isopropanol at different concentrations.

Concentration (mg/mL)	<sup>a</sup> $M_{w,app}$ (kDa)	<sup>b</sup> $M_{w,app}$ (kDa)	$\mathcal{D}$ ( $M_{z,app}/M_{w,app}$ )
1.0	33.6	31.0	1.3
2.0	29.8	25.2	1.3
3.0	22.9	18.5	1.3
4.0	22.3	19.9	1.4

a: M\* method. b: hinge point method.

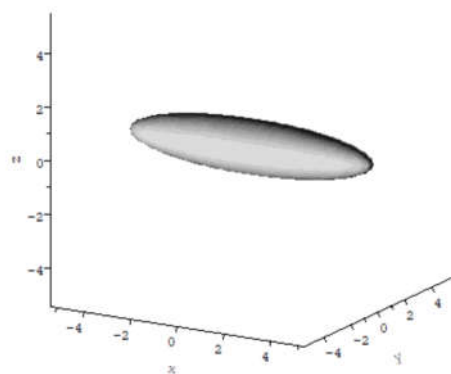
**Table S4.** Apparent weight average molar masses  $M_{w,app}$  and polydispersity  $\mathcal{D}$  values for PS4200 in isopropanol at different concentrations.

Concentration (mg/mL)	<sup>a</sup> $M_{w,app}$ (kDa)	<sup>b</sup> $M_{w,app}$ (kDa)	$\mathcal{D}$ ( $M_z/M_{w,app}$ )
1.0	5.9	5.2	3.8
1.5	4.6	3.8	5.4
2.0	5.7	5.1	2.9
3.0	5.1	4.5	2.7
4.0	3.9	3.4	5.1

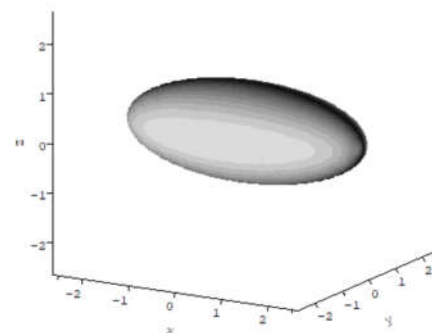
a: M\* method. b: hinge point method.

**Table S5.** Apparent weight average molar masses  $M_{w,app}$  (M\* method) and polydispersity  $\mathcal{D}$  values for PS550 in isopropanol.

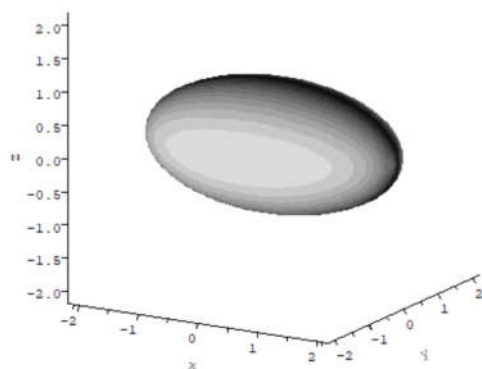
Concentration (mg/mL)	$M_{w,app}$ (kDa)	$\mathcal{D}$ ( $M_z/M_{w,app}$ )
3.0	1.6	1.8
4.0	1.5	1.8

**(a)**

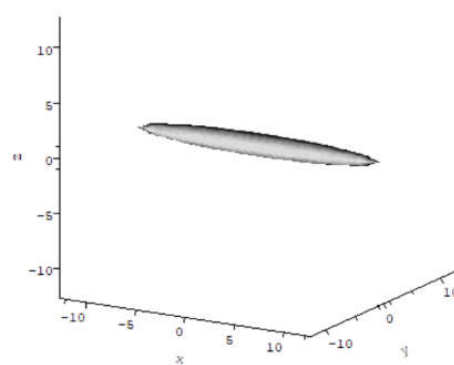
$$a/b = 5.19 \quad b/c = 1.00$$

**(b)**

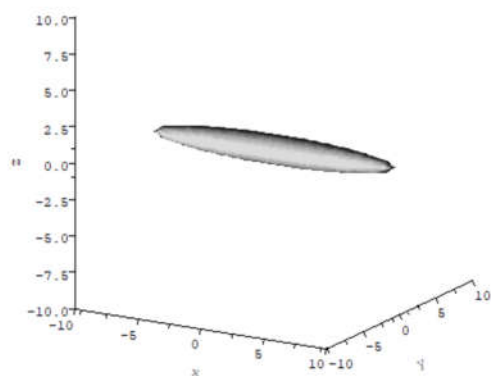
$$a/b = 2.52 \quad b/c = 1.00$$

**(c)**

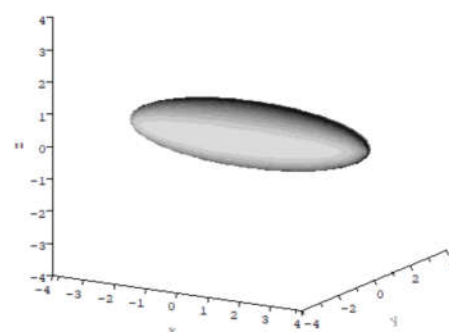
$$a/b = 2.08 \quad b/c = 1.00$$

**(d)**

$$a/b = 12.10 \quad b/c = 1.00$$

**(e)**

$$a/b = 9.55 \quad b/c = 1.00$$

**(f)**

$$a/b = 3.81 \quad b/c = 1.00$$

**Figure S1.** Conformations (equivalent hydrodynamic ellipsoids) for  $\nu$  at  $(v_s/\bar{v}) = 1.2$  using the programme ELLIPS1. (a) = PS18000 in isopropanol; (b) = PS4200 in isopropanol; (c) = PS550 in isopropanol; (d) P36000 in turpentine; (e) = PS18000 in turpentine; (f) = P4200 in turpentine.