

Supporting Information

Equation S1

$$F_c = F_m e^{(A_1 + A_2)/2} \quad (S1)$$

Where F_c and F_m represent fluorescence intensity values for correction and determination, respectively. A_1 and A_2 represent quercetin's absorbance levels at fluorescence excitation and emission wavelengths, respectively.

Equations S2 – S4

$$\frac{1}{v} = \frac{K_m}{V_{\max}} \left(1 + \frac{[I]}{K_i}\right) \frac{1}{[S]} + \frac{1}{V_{\max}} \left(1 + \frac{[I]}{\alpha K_i}\right) \quad (S2)$$

$$\text{Slop} = \frac{K_m}{V_{\max}} + \frac{K_m [I]}{V_{\max} K_i} \quad (S3)$$

$$\text{Y-intercept} = \frac{1}{V_{\max}^{\text{app}}} = \frac{1}{V_{\max}} + \frac{1}{\alpha K_i V_{\max}} [I] \quad (S4)$$

In these formulas, $[I]$ represent inhibitor content, whereas $[S]$ stands for substrate content; v and V_{\max} represent enzymatic reaction rate and maximum catalytic reaction rate, respectively; K_m is the Michaelis–Menten constant; K_i and αK_i stand for the equilibrium constants of quercetin binding to the free AChE and AChE–substrate complex, and α is the apparent coefficient.

Equations S5

$$\frac{F_0}{F} = 1 + K_{sv} [Q] = 1 + K_q \tau_0 [Q] \quad (S5)$$

In the formula, F_0 stands for system fluorescence intensity with AChE alone, whereas F indicates that with quercetin added, respectively, and K_{sv} stands for the Stern–Volmer quenching constant, which represents fluorescence quenching efficiency, $[Q]$ denotes quercetin content, K_q indicates bimolecular quenching rate constant, whereas τ_0 denotes the average fluorophore lifetime, about 10^{-8} s.

Equations S6

$$\log \frac{F_0 - F}{F} = n \log K_a - n \log \frac{1}{[Q_t] - \frac{(F_0 - F)[P_t]}{F_0}} \quad (\text{S6})$$

$[Q_t]$ and $[P_t]$ denote quercetin and AChE concentrations, respectively. F_0 and F are the same as in Equation S 5.

Equations S7 – S8

$$\log K_a = -\frac{\Delta H^\circ}{2.303RT} + \frac{\Delta S^\circ}{2.303R} \quad (\text{S7})$$

$$\Delta G^\circ = \Delta H^\circ - T\Delta S^\circ \quad (\text{S8})$$

where K_a and R suggest binding and gas ($8.314 \text{ J mol}^{-1} \text{ K}^{-1}$) constants, respectively.

Temperature (T) was set to 25, 31, and 37 °C.

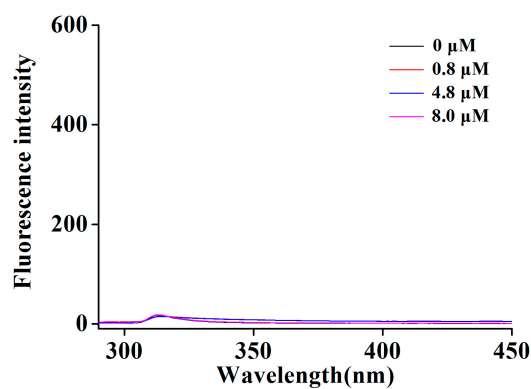


Figure S1. Fluorescence spectra of quercetin at varying concentrations (pH 7.6, $T = 25^\circ\text{C}$). $c(\text{quercetin}) = 0, 0.8, 4.8$ and $8.0 \mu\text{M}$, respectively.

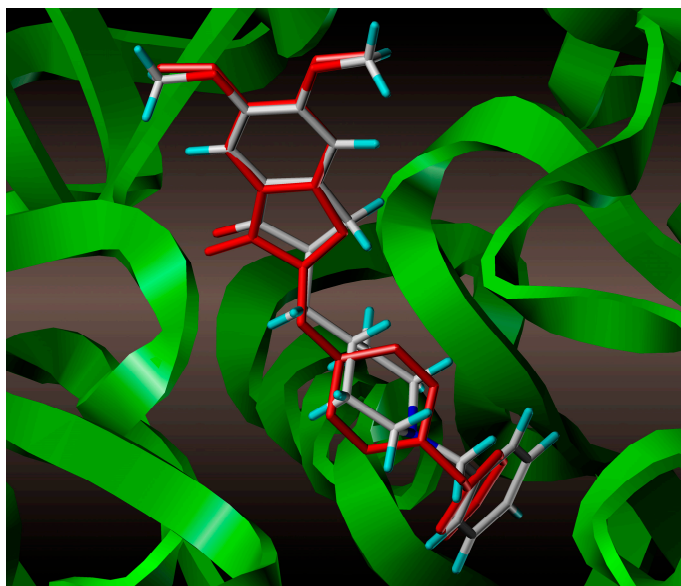


Figure S2. Docked conformer of donepezil (white) and original pose of co-crystallized donepezil (red).