

# Supplementary Materials

## Immobilization of Ionic Liquid on a Covalent Organic Framework for Effectively Catalyzing Cycloaddition of CO<sub>2</sub> to Epoxides

Qianqian Yan <sup>1,†</sup>, Hao Liang <sup>1,†</sup>, Shenglin Wang <sup>1</sup>, Hui Hu <sup>1</sup>, Xiaofang Su <sup>1,\*</sup>, Songtao Xiao <sup>2,\*</sup>, Huanjun Xu <sup>3</sup>,  
Xuechao Jing <sup>4</sup>, Fei Lu <sup>1</sup> and Yanan Gao <sup>1,\*</sup>

1 Key Laboratory of Ministry of Education for Advanced Materials in Tropical Island Resources, Hainan University, No. 58, Renmin Avenue, Haikou 570228, China

2 China Institute of Atomic Energy, Beijing 102413, China

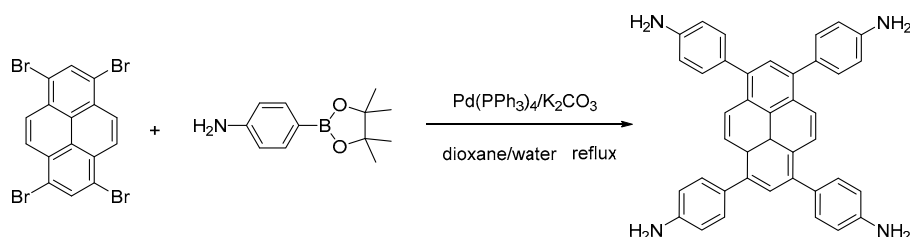
3 School of Science, Qiongtai Normal University, Haikou 571127, China

4 Liaocheng Luxi Polycarbonate Co., Ltd., Liaocheng 252000, China

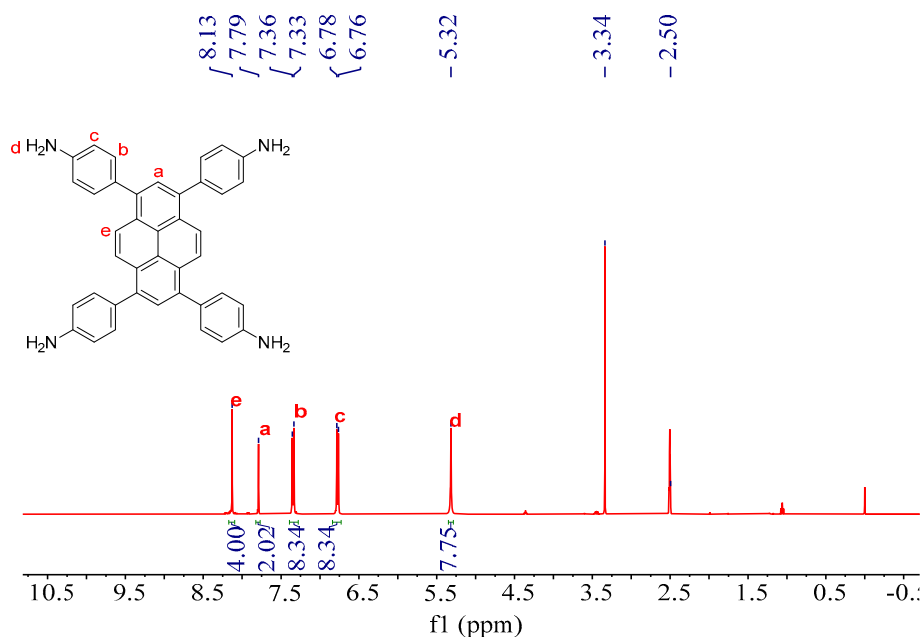
\* Correspondence: sxf@hainanu.edu.cn (X.S.);  
xiao200112@163.com (S.X.); ygao@hainanu.edu.cn (Y.G.)

† These authors contributed equally to this work.

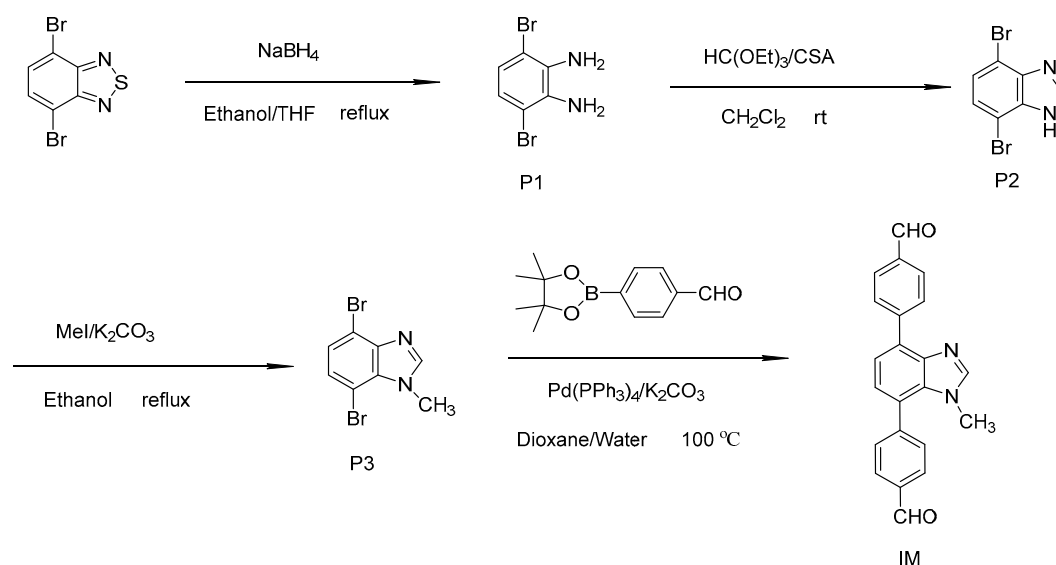
## Synthesis of 4,4',4'',4'''-(Pyrene-1,3,6,8-tetrayl) tetraaniline (PyTTA)



PyTTA was synthesized according to a literature method with modification [54]. To a 250 mL three-necked round flask were added 1,3,6,8-tetrabromopyrene (1.48 g, 2.86 mmol), 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)aniline (3.01 g, 13.7 mmol), palladium tetrakis(triphenylphosphine) (0.33 g, 0.29 mmol), and potassium carbonate (2.18 g, 15.7 mmol). The air in the flask was slowly removed by  $\text{N}_2$  flow and a mixture of dioxane (32 mL) and  $\text{H}_2\text{O}$  (8 mL) was added to the flask. The reaction was stirred under reflux (115 °C) under  $\text{N}_2$  protection for 3 days. After cooling to room temperature, 25 mL water was added to the system and the solid was collected by filtration, and washed thoroughly with  $\text{H}_2\text{O}$  until the filtrate was colorless. The obtained solid was further washed thoroughly with methanol to afford PyTTA in 74% yield.  $^1\text{H}$  NMR (400 MHz, DMSO)  $\delta$  8.14 (s, 4H), 7.80 (s, 2H), 7.35 (d,  $J = 8.3$  Hz, 8H), 6.78 (d,  $J = 8.4$  Hz, 8H), 5.32 (s, 8H).

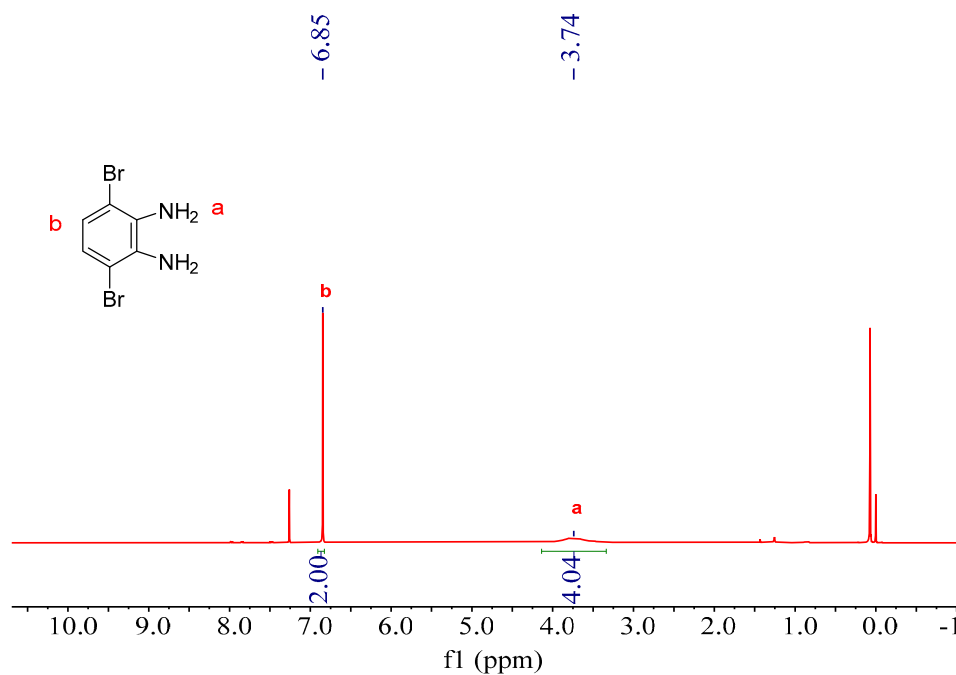


## Synthesis of 5,6-bis(4-formylbenzyl)-1-methyl-1H-benzimidazole (IM)



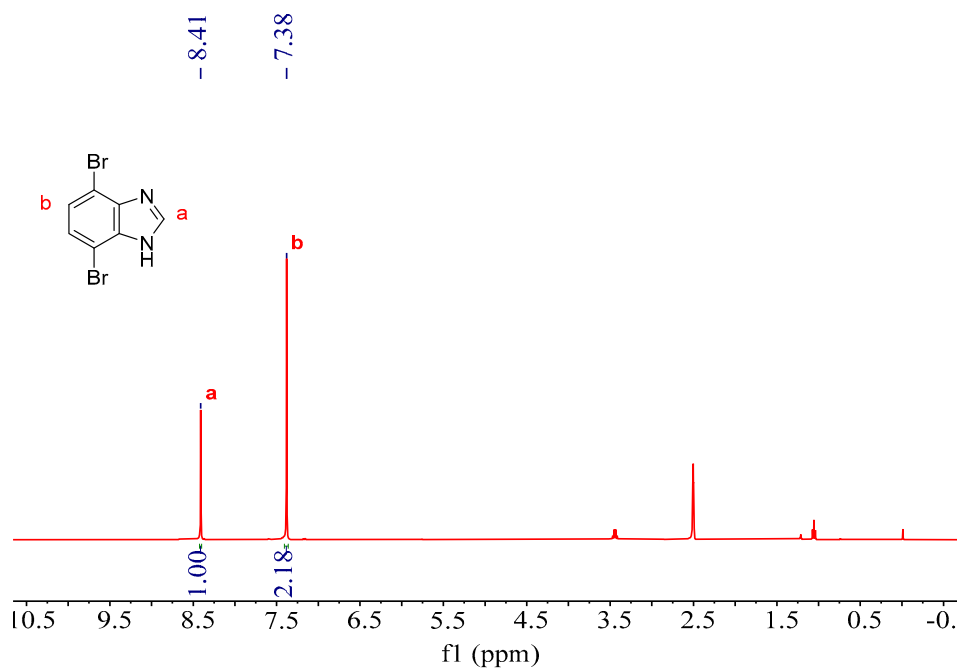
**4,5-Dibromo-1,2-benzenediamine. (P1).** To a 500 mL two-neck round-bottom flask were added 4,7-dibromobenzothiadiazole (3.5 g, 11.9 mmol), anhydrous THF (40 mL) and anhydrous EtOH (400 mL) under  $\text{N}_2$  protection. The mixture was cooled down to  $0\text{ }^\circ\text{C}$  with vigorous stirring.  $\text{NaBH}_4$  (2.65 g, 70 mmol for each portion) was added three times (total 8.0 g) every hour. (Be careful! Add slowly. Exothermic nature of the reaction tends to cause bumping. Gas evolution was also observed.) After consumption of 4,7-dibromobenzothiadiazole was confirmed by TLC analysis, the temperature of the mixture was cooled to  $0\text{ }^\circ\text{C}$ . After slow addition of water (100 mL) and vigorous stirring for 10 min, gummy precipitate was filtered off using Celite. Organic solvent was evaporated and product was extracted with dichloromethane three times. Combined organic layer was washed with water and brine and dried over  $\text{MgSO}_4$ . The extract was filtered off, evaporated, and the crude mixture was purified with short pad silica gel chromatography (eluent: hexane/acetone = 5/1). Combined solution was evaporated to give **P1** as an orange solid in 86.3% yield.

$^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  6.85 (s, 2H), 3.74 (s, 4H).



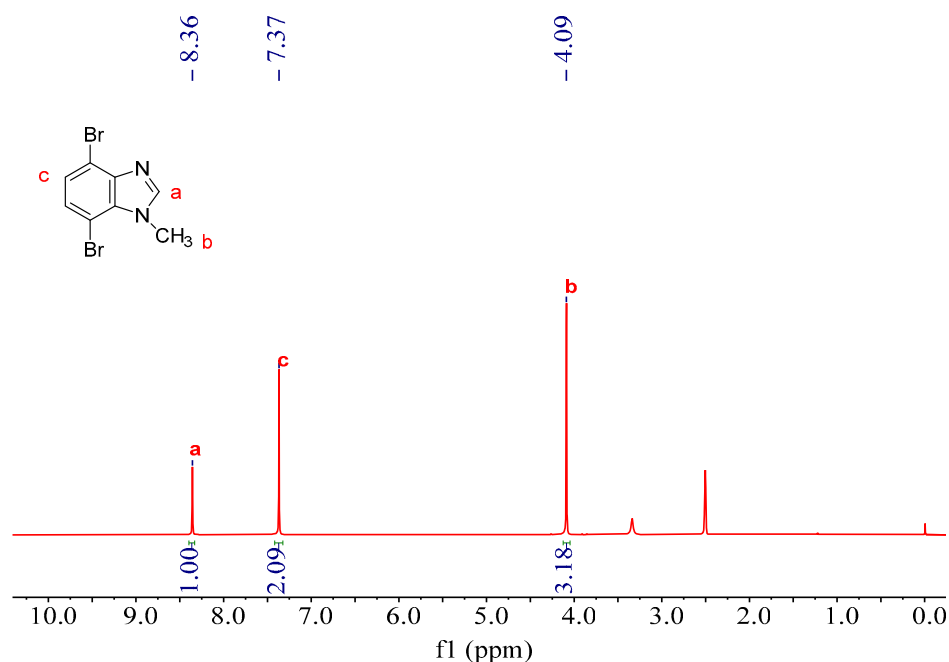
**5,6-Dibromobenzimidazole (P2).** The newly prepared diamine **P1** (2.16g, 8.11 mol) was immediately used for the next step. To the **P1** dissolved in CH<sub>2</sub>Cl<sub>2</sub> (90 mL) were added HC(OEt)<sub>3</sub> (1.35 mL, 12.39 mmol) and sulfamic acid (27.5 mg). The mixture was stirred for 3 hours and powder precipitate formed. Solvent was evaporated and the residue was rinsed with ether. After drying under air, **P2** was obtained as a white powder in 67% yield.

<sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 8.41 (s, 1H), 7.38 (s, 2H).



**5,6-Dibromo-1-methyl-1*H*-Benzimidazole (P3).** To a 250 mL flask were added **P2** (1.97 g, 7.14 mmol), K<sub>2</sub>CO<sub>3</sub> (2.96 g, 21.4 mmol) and EtOH (50 mL). The mixture was heated to 50 °C. To the hot mixture, MeI (1.0 mL, 16.2 mmol) was added dropwise and the mixture was maintained at reflux for 3 h. After consumption of **P2** was confirmed by TLC analysis, the mixture was cooled to room temperature. After addition of water (20 mL) and evaporation of EtOH, the powdered precipitate was collected, washed with water and hexane/Et<sub>2</sub>O (1/1) successively, and dried under vacuum at 60 °C to produce **P3** as a yellow powder in 86% yield.

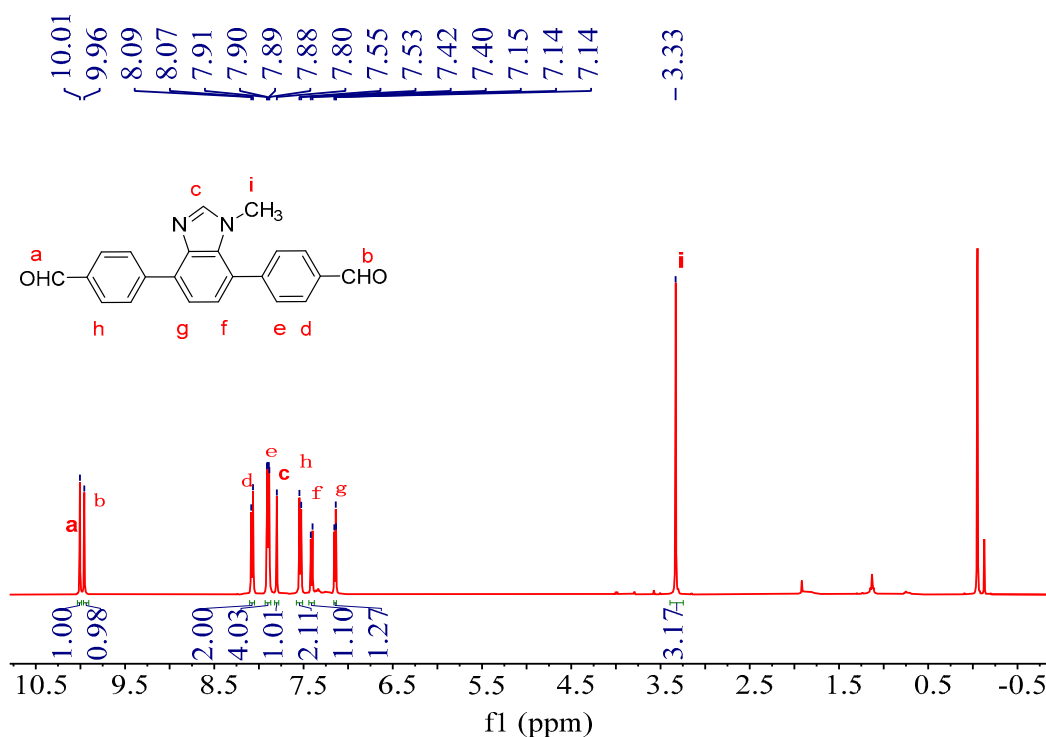
<sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 8.36 (s, 1H), 7.37 (s, 2H), 4.09 (s, 3H).



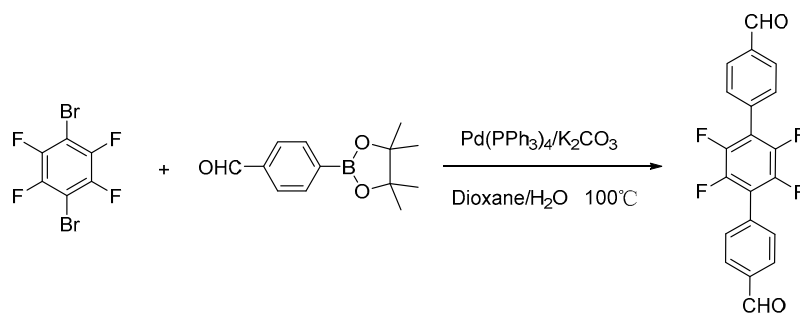
**5,6-Bis(4-formylbenzyl)-1-methyl-1*H*-benzimidazole (IM).** To a 500 mL three-neck bound bottle flask were added **P3** (1.93 g, 6.67 mmol), 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)benzaldehyde (4.67 g, 15.35 mmol), Pd(PPh<sub>3</sub>)<sub>4</sub> (385 mg, 0.33 mmol) and K<sub>2</sub>CO<sub>3</sub> (2.76 g, 20 mmol) in 50 mL of 1,4-dioxane and 12 mL of water with vigorous stirring under N<sub>2</sub> protection. The reaction was heated to 100 °C overnight under nitrogen atmosphere. The color of the reaction was changed from slight yellow to dark brown. The mixture was cooled to room temperature. Water was added and organic compounds were extracted with ethyl acetate three times. The combined organic layer was washed with brine and dried over anhydrous MgSO<sub>4</sub>. The extract was

filtered through short pad basic aluminum oxide and evaporated. The obtained residue was rinsed with hexane/Et<sub>2</sub>O (2/1) to give **IM** as a brown powder (65% yield).

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 10.14 (s, 1H), 10.09 (s, 1H), 8.21 (d, *J* = 8.1 Hz, 2H), 8.03 (d, *J* = 11.6 Hz, 4H), 7.93 (s, 1H), 7.67 (d, *J* = 8.0 Hz, 2H), 7.54 (d, *J* = 7.7 Hz, 1H), 7.27 (s, 1H), 3.46 (s, 3H).



#### Synthesis of 2'3'5'6'-tetrafluoro-[1,1':4',1''-terphenyl]-4,4'-dicarbaldehyde (**4F**)

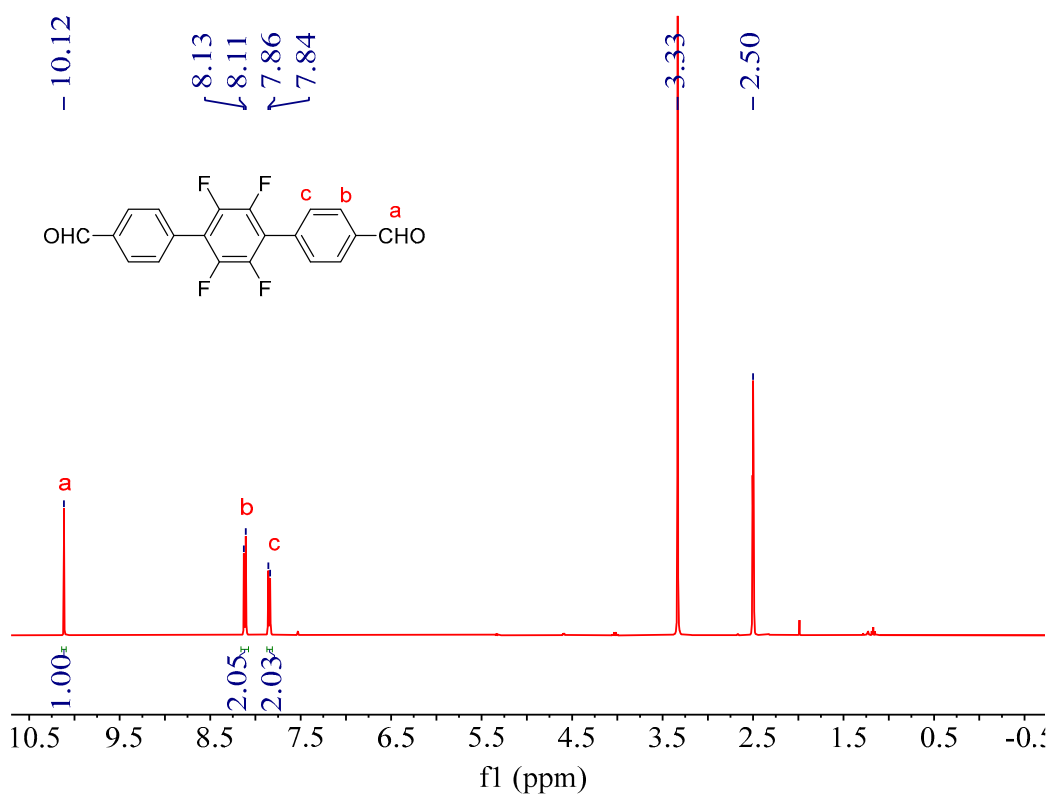


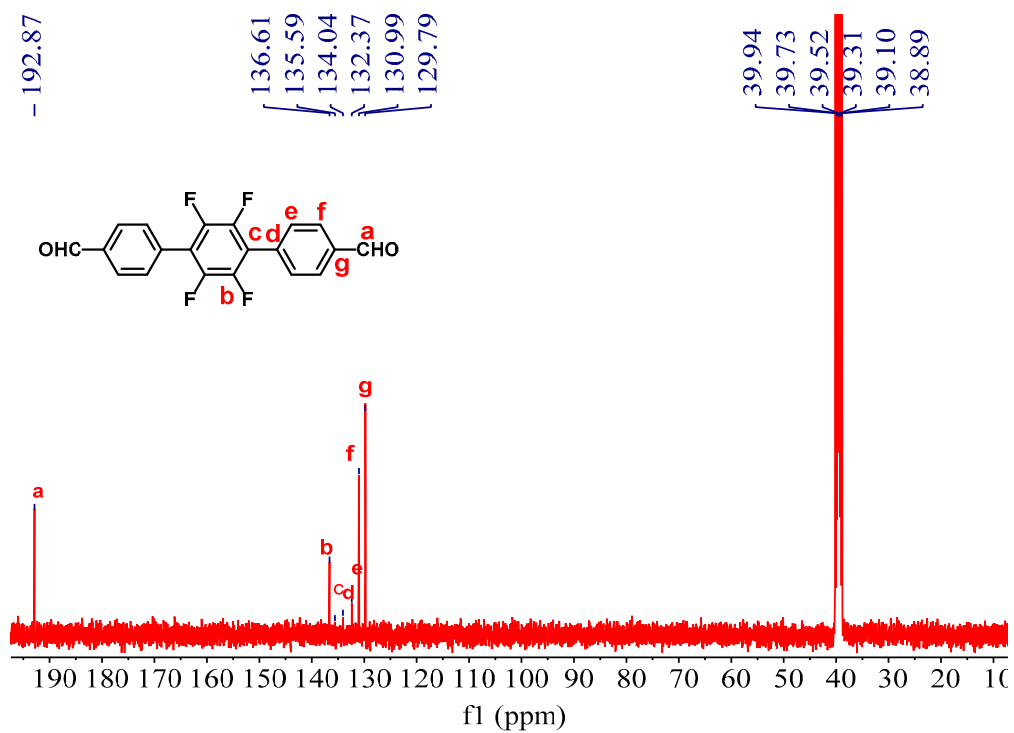
**4F** was synthesized according to a literature method with modification [55]. To a 500 mL three-neck round bottle flask were added 1,4-dibromotetrafluorobenzene (1.65 g, 5.36 mmol), K<sub>2</sub>CO<sub>3</sub> (2.63 g, 19.0 mmol), 4-formylphenylboronic acid pinacol cyclic ester (3.36 g, 14.48 mmol) and palladium tetrakis(triphenylphosphine) (0.35 g, 0.30

mmol). The air in the flask was slowly removed by N<sub>2</sub> flow and a mixture of dioxane (50 mL) and H<sub>2</sub>O (12 mL) was added to the flask under N<sub>2</sub> protection. The reaction was stirred under reflux (100 °C) under N<sub>2</sub> environment for 24 hours. After cooling to room temperature, solid was precipitated and separated by filtration. The solid was washed by water, ethyl acetate, and ethanol, successively. After drying at 80 °C under vacuum overnight, a white solid product **4F** was obtained in 80% yield.

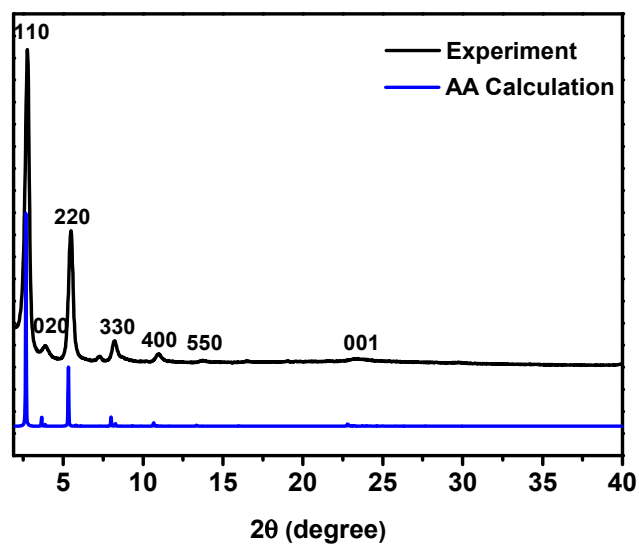
<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 10.12 (s, 2H), 8.06 (d, J = 8.4 Hz, 4H), 7.73 (d, J = 8.0 Hz, 4H).

<sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>) δ 192.87, 136.61, 135.59, 134.04, 132.37, 130.99, 129.79.

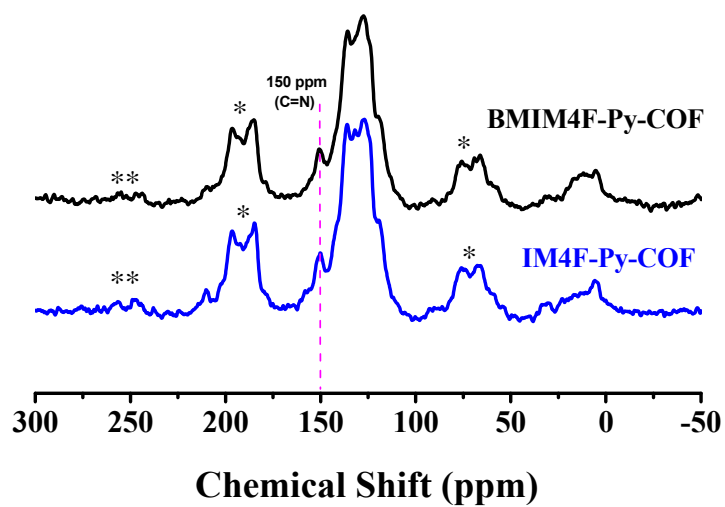




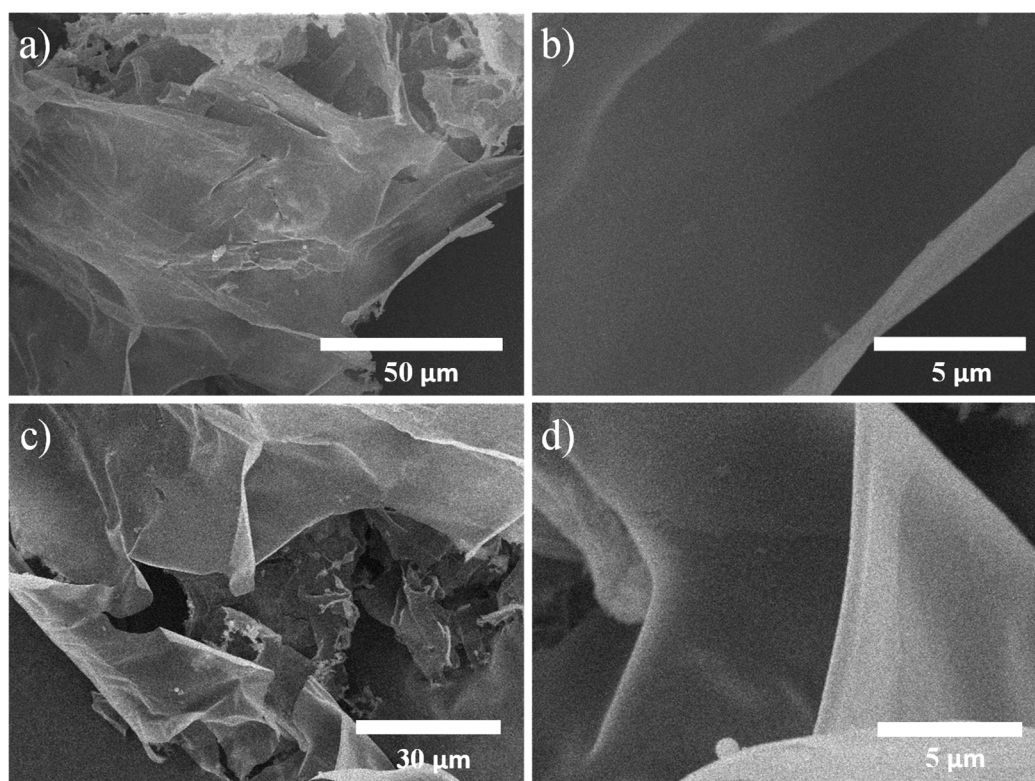




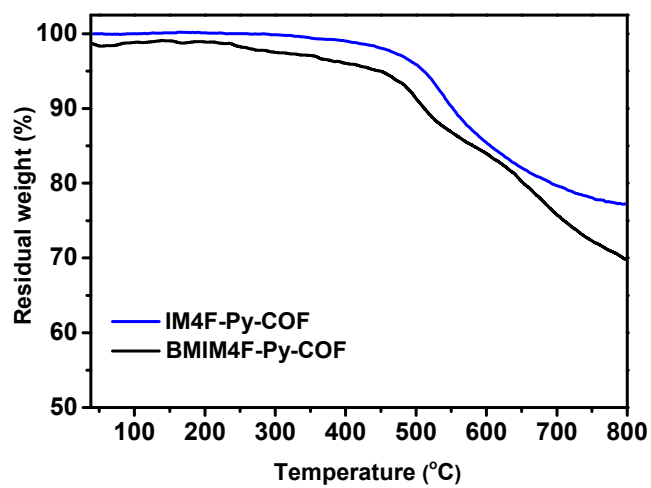
**Figure S1.** PXRD pattern of BMIM4F-Py-COF. Experimental pattern (black) and AA-stacking (blue).



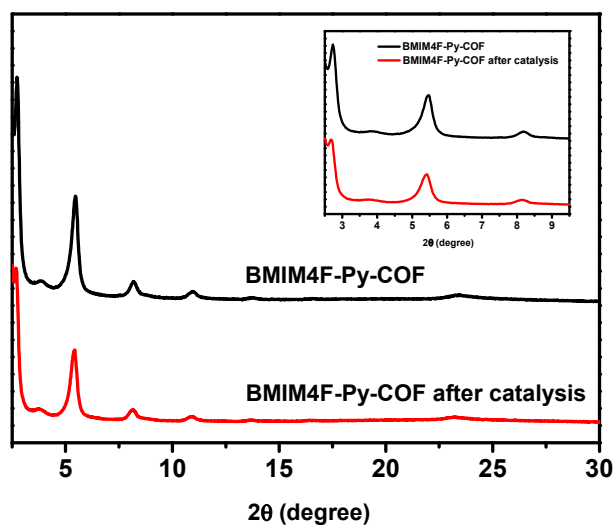
**Figure S2.** Solid-state  $^{13}\text{C}$  NMR spectra of IM4F-Py-COF (blue) and BMIM4F-Py-COF (black).



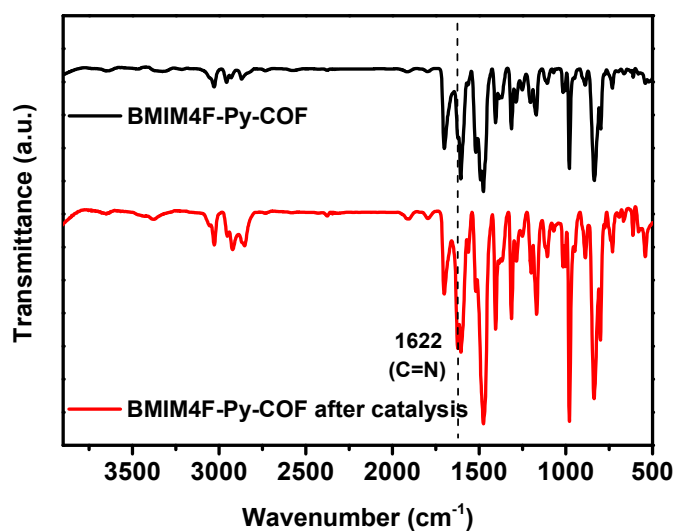
**Figure S3.** FE-SEM images of IM4F-Py-COF (a,b) and BMIM4F-Py-COF (c,d).



**Figure S4.** Thermogravimetric analysis of IM4F-Py-COF (blue) and BMIM4F-Py-COF (black).



**Figure S5.** Comparison of PXRD pattern of BMIM4F-Py-COF before (black) and after catalysis (red). The insert indicates an enlarged PXRD pattern.



**Figure S6.** Comparison of FT-IR spectroscopy of BMIM4F-Py-COF before (black) and after catalysis (red).

**Table S1.** Fractional atomic coordinates for the unit cell of IM4F-Py-COF.

Triclinic P1							
a = 49.83 Å, b = 42.59 Å, c = 3.93 Å; $\alpha = 89.50^\circ$ , $\beta = 88.71^\circ$ , $\gamma = 90.06^\circ$							
atom	x	y	z	atom	x	y	z
C1	0.17137	0.3325	0.43195	N141	0.87962	0.36	0.4616
C2	0.14814	0.31593	0.53614	C142	0.87981	0.32987	0.4135
C3	0.15022	0.28399	0.62232	H143	0.93161	0.47572	0.68483
C4	0.02362	0.43703	0.52968	H144	0.92779	0.5587	0.18003
C5	0.02361	0.47017	0.56812	H145	0.88675	0.59063	0.17662
C6	0.04726	0.48723	0.62198	H146	0.92489	0.65984	0.79006
C7	0.04851	0.41739	0.51635	H147	0.965	0.62699	0.82272
C8	0.07119	0.58051	0.3033	C148	0.801	0.72269	0.47991
C9	0.09442	0.59894	0.29028	C149	0.77481	0.73897	0.48389
C10	0.09539	0.62757	0.45996	C150	0.7716	0.76734	0.65831
C11	0.0726	0.63759	0.64678	C151	0.82474	0.73929	0.40317
C12	0.04962	0.6187	0.6681	C152	0.80236	0.6905	0.55485
N13	0.11904	0.36079	0.46951	C153	0.75296	0.72597	0.31604
C14	0.12173	0.33153	0.55973	H154	0.86764	0.73715	0.34334
H15	0.06612	0.47574	0.66753	H155	0.82777	0.65022	0.61293
H16	0.0707	0.55927	0.15606	H156	0.82371	0.76418	0.34184
H17	0.11167	0.59141	0.14001	H157	0.78404	0.67764	0.61843
H18	0.07254	0.65927	0.78621	F158	0.79243	0.77809	0.84612
H19	0.03275	0.6264	0.82471	F159	0.75712	0.69979	0.123
C20	0.20237	0.71582	0.45558	H160	0.89856	0.31772	0.36227
C21	0.22924	0.73047	0.42588	C161	0.33118	0.82322	0.56917
C22	0.23254	0.76111	0.29569	C162	0.35451	0.812	0.40498
C23	0.18075	0.73222	0.60411	C163	0.35362	0.78329	0.23423
C24	0.19823	0.68528	0.3364	C164	0.4743	0.93698	0.53173
C25	0.25221	0.71458	0.52995	C165	0.47428	0.97013	0.57209
H26	0.13919	0.73095	0.75928	C166	0.45056	0.98719	0.63263
H27	0.17009	0.64774	0.27165	C167	0.44939	0.91731	0.51651
H28	0.18363	0.75568	0.70077	C168	0.42721	0.08017	0.3216
H29	0.21461	0.67247	0.21651	C169	0.40395	0.09846	0.31476
H30	0.21535	0.77422	0.20861	C170	0.40304	0.12756	0.47483
N31	0.25304	0.68509	0.70083	C171	0.42551	0.13795	0.65137
H32	0.10469	0.31782	0.64928	C172	0.4484	0.11907	0.67041
C33	0.67137	0.8325	0.43461	N173	0.37961	0.86001	0.45914
C34	0.64814	0.81594	0.53872	C174	0.3798	0.82988	0.4109
C35	0.65021	0.78401	0.62512	H175	0.43159	0.97572	0.68322
C36	0.52361	0.93703	0.52961	H176	0.42782	0.05869	0.17848
C37	0.5236	0.97017	0.56803	H177	0.38677	0.09062	0.17448
C38	0.54724	0.98722	0.62228	H178	0.42487	0.15986	0.78801

C39	0.5485	0.91739	0.51677	H179	0.46498	0.12701	0.82132
C40	0.5712	0.08051	0.30392	C180	0.30101	0.22269	0.47673
C41	0.59444	0.09894	0.29131	C181	0.27482	0.23897	0.48069
C42	0.5954	0.12756	0.4612	C182	0.27161	0.26734	0.65513
C43	0.57259	0.13757	0.64777	C183	0.32474	0.23928	0.40006
C44	0.5496	0.11869	0.66867	C184	0.30236	0.19049	0.55172
N45	0.61904	0.86078	0.4713	C185	0.25296	0.22597	0.31284
C46	0.62173	0.83154	0.56194	H186	0.36764	0.23714	0.34046
H47	0.56609	0.97573	0.66813	H187	0.32776	0.15021	0.60998
H48	0.57073	0.05927	0.15657	H188	0.32372	0.26418	0.33872
H49	0.6117	0.09141	0.14123	H189	0.28404	0.17763	0.61524
H50	0.57252	0.15925	0.7873	F190	0.29244	0.27809	0.84296
H51	0.53273	0.12637	0.82507	F191	0.25712	0.19979	0.11979
C52	0.70239	0.21578	0.45857	H192	0.39855	0.81773	0.35978
C53	0.72925	0.23045	0.42901	C193	0.17289	0.67127	0.36706
C54	0.73255	0.26109	0.29888	C194	0.15133	0.68751	0.52017
C55	0.68076	0.23218	0.60705	C195	0.15547	0.7181	0.63811
C56	0.69825	0.18525	0.33922	C196	0.02373	0.5701	0.52077
C57	0.75222	0.21456	0.53309	C197	0.02366	0.53693	0.56371
H58	0.63919	0.2309	0.76193	C198	0.04728	0.51984	0.61961
H59	0.67011	0.14772	0.27405	C199	0.0487	0.58966	0.50033
H60	0.68363	0.25564	0.70385	C200	0.07119	0.42631	0.31945
H61	0.71464	0.17245	0.21937	C201	0.09424	0.4076	0.31185
H62	0.71536	0.2742	0.21179	C202	0.09475	0.37889	0.48584
N63	0.75305	0.18506	0.7039	C203	0.07188	0.36931	0.67568
H64	0.60469	0.81784	0.65155	C204	0.0491	0.38844	0.69157
C65	0.8271	0.67504	0.55202	N205	0.11999	0.6452	0.4373
C66	0.8509	0.69166	0.47652	C206	0.12459	0.67281	0.56208
C67	0.84955	0.72391	0.40336	H207	0.06617	0.5313	0.66325
C68	0.97441	0.5701	0.52606	H208	0.07098	0.44754	0.16758
C69	0.97435	0.53692	0.56979	H209	0.11165	0.41487	0.16121
C70	0.95061	0.5198	0.63267	H210	0.07178	0.34788	0.82396
C71	0.9495	0.58969	0.50977	H211	0.03214	0.38104	0.84992
C72	0.92706	0.42687	0.33047	C212	0.19829	0.28551	0.50525
C73	0.90398	0.40821	0.32115	C213	0.22463	0.26964	0.49476
C74	0.90342	0.37872	0.47833	C214	0.22781	0.24126	0.32131
C75	0.92585	0.36839	0.65649	C215	0.1751	0.26886	0.60505
C76	0.94847	0.38766	0.68004	C216	0.19628	0.31739	0.41829
N77	0.87865	0.64533	0.46801	C217	0.2465	0.28275	0.6611
C78	0.87718	0.67576	0.47755	H218	0.13263	0.27077	0.70445
H79	0.93166	0.53122	0.68294	H219	0.17019	0.35711	0.36281
H80	0.92754	0.44859	0.18761	H220	0.17663	0.24415	0.67304
H81	0.88682	0.41599	0.17916	H221	0.21422	0.3303	0.33899

H82	0.92536	0.34598	0.78806	F222	0.20677	0.23018	0.13998
H83	0.96504	0.37966	0.83173	F223	0.24229	0.30897	0.85302
C84	0.80675	0.27639	0.40703	H224	0.10914	0.68565	0.69841
C85	0.78111	0.25885	0.39295	C225	0.67291	0.17124	0.36961
C86	0.77746	0.22746	0.50749	C226	0.65134	0.18748	0.52265
C87	0.82965	0.26625	0.22791	C227	0.65548	0.21806	0.6408
C88	0.80756	0.30545	0.57337	C228	0.52373	0.07009	0.52068
C89	0.75809	0.27514	0.28352	C229	0.52365	0.03692	0.56361
H90	0.87118	0.27465	0.10008	C230	0.54727	0.01983	0.6199
H91	0.83156	0.3453	0.70636	C231	0.5487	0.08965	0.50071
H92	0.82869	0.24544	0.07622	C232	0.5712	0.9263	0.32012
H93	0.78991	0.31429	0.70617	C233	0.59425	0.90759	0.31295
N94	0.79576	0.20568	0.65962	C234	0.59474	0.87889	0.48716
H95	0.75985	0.29907	0.19008	C235	0.57185	0.86932	0.67675
H96	0.89525	0.68994	0.47427	C236	0.54908	0.88845	0.69218
C97	0.3271	0.17504	0.54902	N237	0.62	0.14518	0.43897
C98	0.35091	0.19165	0.47363	C238	0.6246	0.17278	0.56421
C99	0.34956	0.22391	0.4004	H239	0.56616	0.03129	0.66384
C100	0.47441	0.0701	0.52509	H240	0.571	0.94752	0.16813
C101	0.47434	0.03692	0.56884	H241	0.61167	0.91485	0.1625
C102	0.45059	0.0198	0.63135	H242	0.57175	0.84789	0.82517
C103	0.4495	0.08969	0.50835	H243	0.5321	0.88105	0.85032
C104	0.42707	0.92688	0.3288	C244	0.69829	0.78552	0.50831
C105	0.40399	0.90822	0.31915	C245	0.72463	0.76965	0.49791
C106	0.40342	0.87873	0.47618	C246	0.72781	0.74126	0.32448
C107	0.42584	0.86838	0.65452	C247	0.6751	0.76887	0.60808
C108	0.44845	0.88765	0.67844	C248	0.69629	0.8174	0.42119
N109	0.37865	0.14533	0.46555	C249	0.7465	0.78275	0.66424
C110	0.37719	0.17576	0.47491	H250	0.63262	0.77079	0.7072
H111	0.43164	0.03121	0.68134	H251	0.6702	0.85712	0.36531
H112	0.42756	0.9486	0.18603	H252	0.67662	0.74416	0.67619
H113	0.38684	0.91601	0.17701	H253	0.71422	0.8303	0.34193
H114	0.42533	0.84597	0.78598	F254	0.70677	0.73018	0.14312
H115	0.46501	0.87965	0.83032	F255	0.74229	0.80898	0.8561
C116	0.30674	0.77642	0.40389	H256	0.60914	0.1856	0.70065
C117	0.2811	0.75888	0.38975	C257	-0.00103	0.48681	0.56589
C118	0.27745	0.72748	0.50431	C258	-0.00098	0.42149	0.4979
C119	0.32964	0.76628	0.22484	H259	-0.00086	0.39658	0.44522
C120	0.30755	0.80548	0.57027	C260	0.49896	0.9868	0.56535
C121	0.25808	0.77516	0.2803	C261	0.49901	0.92149	0.49738
H122	0.37118	0.77467	0.09724	H262	0.49914	0.89657	0.44471
H123	0.33154	0.84532	0.70347	C263	-0.00101	0.52027	0.56403
H124	0.32869	0.74546	0.07312	C264	-0.00087	0.58566	0.48922

H125	0.28989	0.81431	0.70301		H265	-0.00076	0.61062	0.43415
N126	0.29575	0.70571	0.65647		C266	0.49899	0.02027	0.5635
H127	0.25984	0.79909	0.18685		C267	0.49913	0.08566	0.48869
H128	0.39525	0.18994	0.47172		H268	0.49925	0.11062	0.43363
C129	0.83119	0.3232	0.57212		C269	0.27809	0.68088	0.77094
C130	0.85452	0.31198	0.40782		C270	0.32331	0.69877	0.5445
C131	0.85363	0.28327	0.23715		H271	0.28506	0.65977	0.89677
C132	0.9743	0.43698	0.5327		H272	0.32402	0.69423	0.26718
C133	0.97429	0.47013	0.57305		H273	0.33076	0.67734	0.67478
C134	0.95057	0.48719	0.63396		H274	0.33714	0.71806	0.60658
C135	0.9494	0.41731	0.51794		C275	0.7781	0.18085	0.77401
C136	0.92719	0.58017	0.32324		C276	0.82332	0.19875	0.54764
C137	0.90393	0.59847	0.31674		H277	0.78507	0.15974	0.89979
C138	0.90303	0.62756	0.47696		H278	0.82403	0.1942	0.27033
C139	0.92552	0.63794	0.65331		H279	0.83077	0.17732	0.67796
C140	0.94841	0.61906	0.67199		H280	0.83715	0.21804	0.60967

**Table S2.** Comparison with various metal-free catalysts in the performance of the cycloaddition of CO<sub>2</sub> to epichlorohydrin.

Catalysts	Temperature (°C)	Pressure (MPa)	Yield (%)	Ref
BMIM4F-Py-COF	110	4	97	This work
BMIM4F-Py-COF	120	1	94	This work
2,3-DhaTph	110	0.1	88	67
CTF-1	130	0.69	77	68
CTF-1-HSA	130	0.69	96	69
cCTF	90	1	95	69
PAD-3	110	1	99	70
DVB-HTA	120	1.2	90	71
DVB-BTA	120	1.2	84	71
POM3-IM	120	1	90	72
PRP-1+TBAB	60	0.1	89	73
[PTPP]50%-TD-COF	100	1	99	74
PIL-HPCOF-320–10–100	90	1	95	75
IP-1	100	0.1	76	76
IP-2	100	0.1	99	76
IP-1	80	0.1	52	76
IP-2	80	0.1	70	76
COF-JLU7/TBAB	80	0.1	96	77
DB10%-Pa-Tp	120	0.1	99	78
COF-IL	80	0.1	98	79
PAIL-1	25	0.5	10	80
PAIL-2	25	0.5	22	80



PAIL-3	25	0.5	41	80
PAIL-3	80	0.5	99	80
CTF-0-400-40-1	130	0.69	<5	81
CTF-0-400/600-20/20-5	130	0.69	93	81
CTF-CSU1	25	0.1	78	82
COF-JLU7	40	0.1	92	77
OMe-OH-TPBP-COF	40	0.1	91	83
COF-salen-Co	100	2	90	84
DB10%-Pa-Tp	120	0.1	99	85
COF-HNU3	100	2	96	86
AMIMBr@H2P-DHP COF	120	1	91	58
COF-HNU14	120	2	99	87