

Supplementary Materials

Table S1. Crystallographic data of [Cd(TMPP)Pyz] (1).

Formula	C ₅₂ H ₄₀ CdN ₆ O ₄
Formula weight	925.30
Temperature (K)	150
Space group	P21/c
Crystal system	monoclinic
a/ Å	23.0965(10)
b/ Å	17.6690(7)
c/ Å	12.4340(5)
α /(deg)	90
β /(deg)	99.551(2)
γ /(deg)	90
Volume/Å ³	5003.9(4)
Z	4
Density (g/ cm ⁻³)	1.228
Crystal size (mm ³)	0.360 × 0.290 × 0.250
Absorption coefficient (mm ⁻¹)	0.484
F(000)	1896.0
Reflections collected	37209
Independent reflections	11304[R(int) = 0.0284]
Data/restraints/parameters	11304/0/572
Index ranges (h, k, l)	-28, 29; -22, 22; -13, 16
Theta range for data collection θ (°)	5.836 to 54.92°
S [Goodness of fit]	1.030
R1 a[Fo>4σ(Fo)]	0.0355
wR2b ([all data])	0.0879
Largest diff. peak/hole / e Å ⁻³	0.60/-0.59
CCDC	2069722

^a: R1 = $\sum ||F_o| - |F_c|| / \sum |F_o|$. b: wR2 = $\{\sum [w(|F_o|^2 - |F_c|^2)^2] / \sum [w(|F_o|^2)^2]\}^{1/2}$.

Table S2. Selected bond lengths (Å) and dihedral angles (°) for [Cd(TMPP)(Pyz)] (1).

Cadmium coordination polyhedron		
Bond lengths (Å)	Experimental	Theoretical
Cd-N1	2.1742(18)	2.22
Cd-N2	2.1813(17)	2.23
Cd-N3	2.1949(17)	2.23
Cd-N4	2.2010(17)	2.22
Cd-N5	2.3696(19)	2.46
Angles (°)	Experimental	Theoretical

N1-Cd-N2	86.61(6)	85.56
N1-Cd-N3	149.41(7)	143.91
N1-Cd-N4	86.37(6)	84.78
N1-Cd-N5	115.70(7)	124.60
N2-Cd-N3	86.22(6)	84.06
Pyrazine (axial ligand)		
Bond lengths (Å)	Experimental	Theoretical
N5-C62	1.338(3)	1.34
N5-C66	1.332(3)	1.34
C62-C63	1.372(4)	1.39
N6-C63	1.336(4)	1.34
N6-C65	1.323(4)	1.34
Angles (°)	Experimental	Theoretical
N5-C66-C65	121.4(2)	122.08
N5-C62-C63	121.6(2)	122.09
C62-N5-C66	116.3(2)	115.83
C63-N6-C65	115.5(2)	115.83
N6-C63-C62	122.4(3)	122.08
N6-C65-C66	122.9(3)	122.09

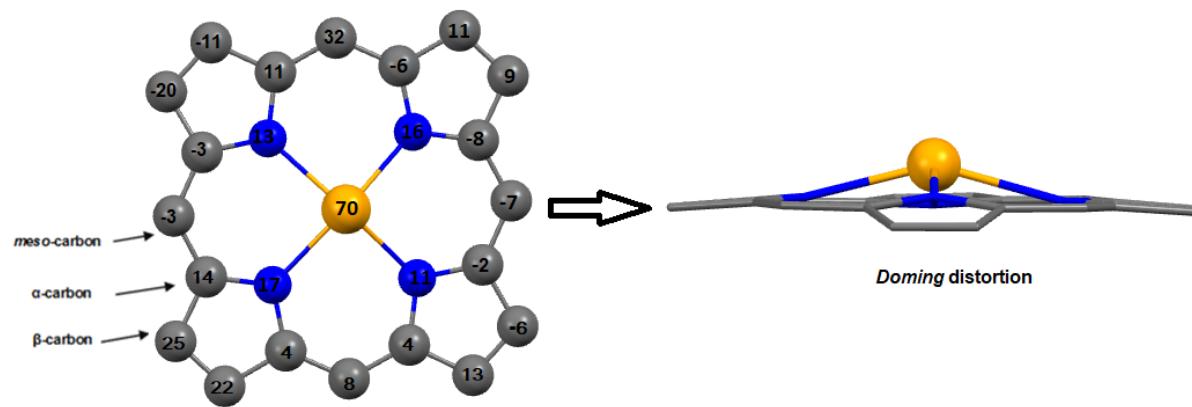


Figure S1. Formal diagrams of the porphyrinato core showing the out-of-plane localization of Cd(II) metal ion in [Cd(TMPP)(Pyz)] (**1**).

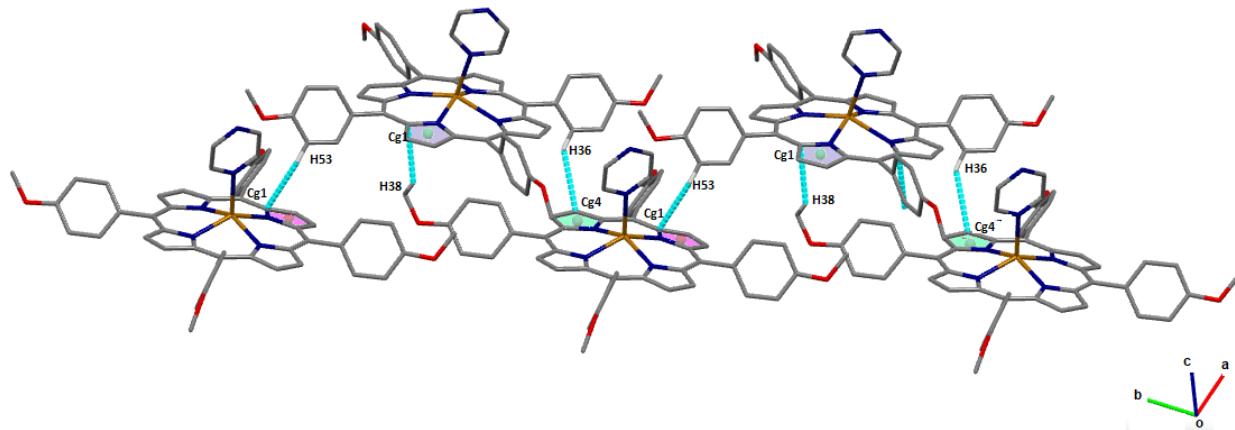


Figure S2. Schematic representation showing the C-H...Cg intermolecular interactions contacts are shown as dashed lines.

Table S3. Selected intermolecular C-H···N and C-H··· π interactions for $[\text{Cd}(\text{TMPP})(\text{Pyz})]$ (**1**).

D-H···A a	Symmetry of A	D-H···A (Å)	D-H···A (°)
C48-H48···N6	1+x,-1+y, z	2.676(2)	148
C65-H65···N6	1-x, 1+y, z	2.962(3)	143
C38-H38···Cg1b	X,1/2-Y,-1/2+Z	3.661(3)	158
C53-H53···Cg4c	X,3/2-Y,1/2+Z	3.639(4)	148

a: D is the donor atom and A is the acceptor atom, b: Cg1 is a pyrrole centroid (i.e. C20-C19-C18-C1-N1), and c: Cg4 is a pyrrole centroid (i.e. C16-C15-C14-C13-N4).

Table S4. Total interaction energies for $[\text{Cd}(\text{TMPP})(\text{Pyz})]$ (**1**) (KJ.mol⁻¹) at the B3LYP/DGDZVP basis set

	N	Sym op	R	E_{ele}	E_{pol}	E_{dis}	E_{rep}	E_{tot}
	2	x, y, z	17.67	7.6	-1.3	-24.8	0.0	-14.5
	2	x, -y+1/2, z+1/2	12.01	-15.8	-5.0	-71.4	0.0	-82.6
	2	x, -y+1/2, z+1/2	9.66	-47.9	-8.1	-123.1	103.1	-100.2
	2	x, y, z	12.43	-2.3	-0.9	-12.0	0.0	-13.6
	2	-x, y+1/2, -z+1/2	16.94	-3.9	-1.1	-16.0	0.0	-18.8
	1	-x, -y, -z	18.03	-2.7	-1.4	-25.9	0.0	-26.4

1	-x, -y, -z	10.55	-40.5	-8.1	-100.2	84.3	-83.9
2	-x, y+1/2, -z+1/2	16.59	-2.6	-1.3	-24.7	0.0	-25.2
2	-x, y+1/2, -z+1/2	13.28	-3.5	-1.1	-11.7	0.0	-14.7
1	-x, -y, -z	12.92	-7.1	-1.0	-9.5	0.0	-16.6

Table S5. Total interaction energies for H₂-TMPP (KJ.mol⁻¹) at the B3LYP/DGDZVP basis set.

	N	Sym op	R (Å)	<i>E</i> _{ele}	<i>E</i> _{pol}	<i>E</i> _{dis}	<i>E</i> _{rep}	<i>E</i> _{tot}
	2	-x, y, -z+1/2	9.83	-20.8	-3.6	-62.4	47.7	-49.5
	4	-x, y, -z+1/2	16.41	-0.3	-1.0	-18.1	0.0	-16.7
	2	x, y, z	17.07	0.8	-0.2	-4.9	0.0	-3.6
	2	x, y, z	13.93	-15.3	-2.5	-40.8	0.0	-53.6
	2	-x, y, -z+1/2	8.68	-56.7	-8.5	-130.1	176.7	-70.4
	2	x, y, z	15.70	-3.9	-3.1	-46.2	0.0	-46.6

Table S6. Atomic percentage contribution to the distance between the surface and the internal and external nucleus to the surface for H₂-TMPP and [Cd(TMPP)(Pyz)] (**1**).

% Contribution											
H ₂ -TMPP											
Inside-Outside						Inside-Outside					
C-H	C-N	C-O	O-H	N-H	N-O	H-C	N-C	O-C	H-O	H-N	O-N
12.8	0	0.4	5.4	2.7	0	11.0	0	0.3	4.9	2.2	0
% Contribution											
[Cd(TMPP)(Pyz)] (1)											
Inside-Outside						Inside-Outside					
C-H	C-N	C-O	N-O	N-H	O-H	H-C	N-C	O-C	O-N	H-N	H-O
13.6	0	0.5	0.1	4.3	4.9	11.5	0	0.4	0.1	3.3	4.3

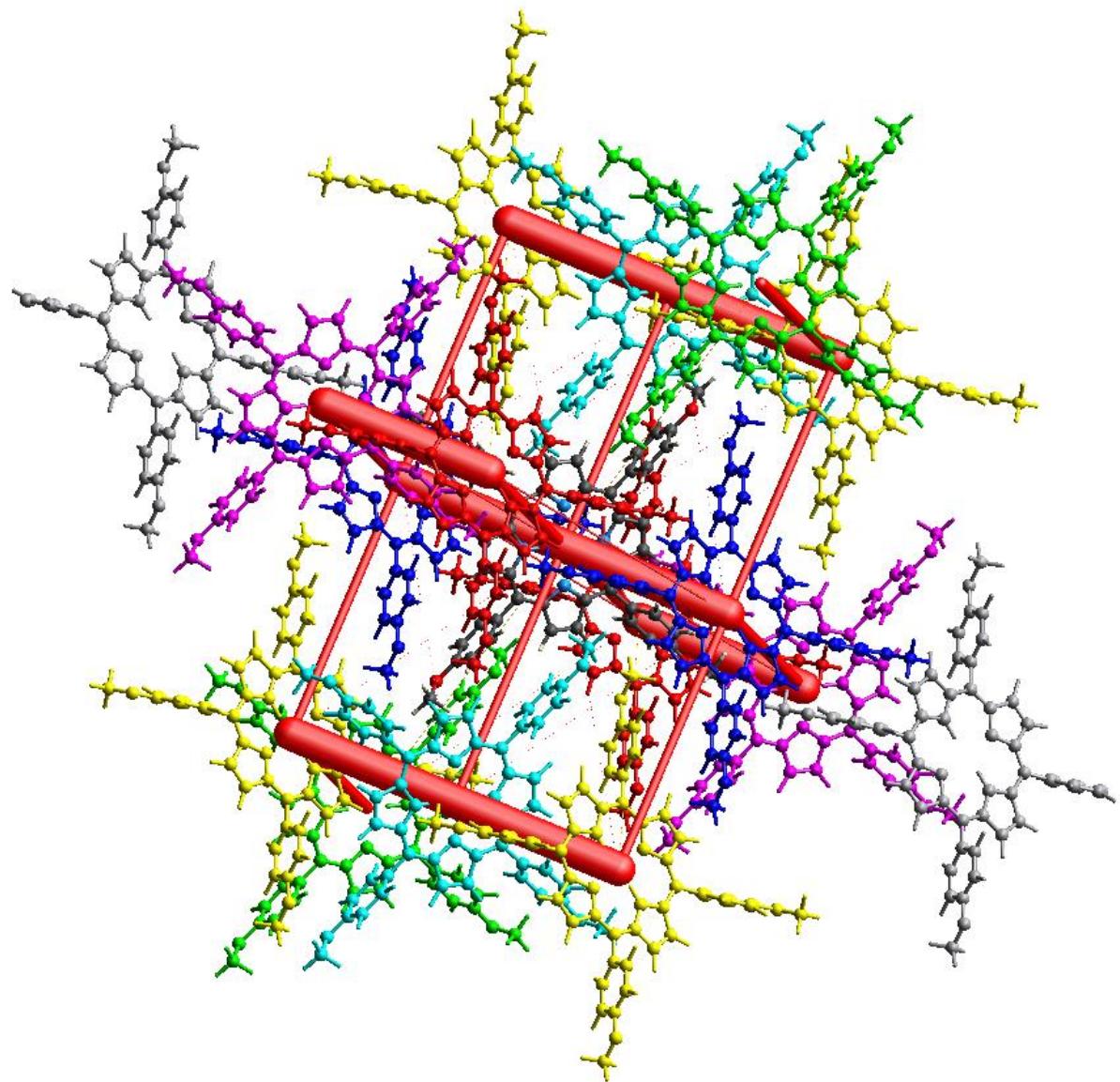


Figure S3. Coulombic interacting energies for interacting molecules in the H₂-TMPP crystal.

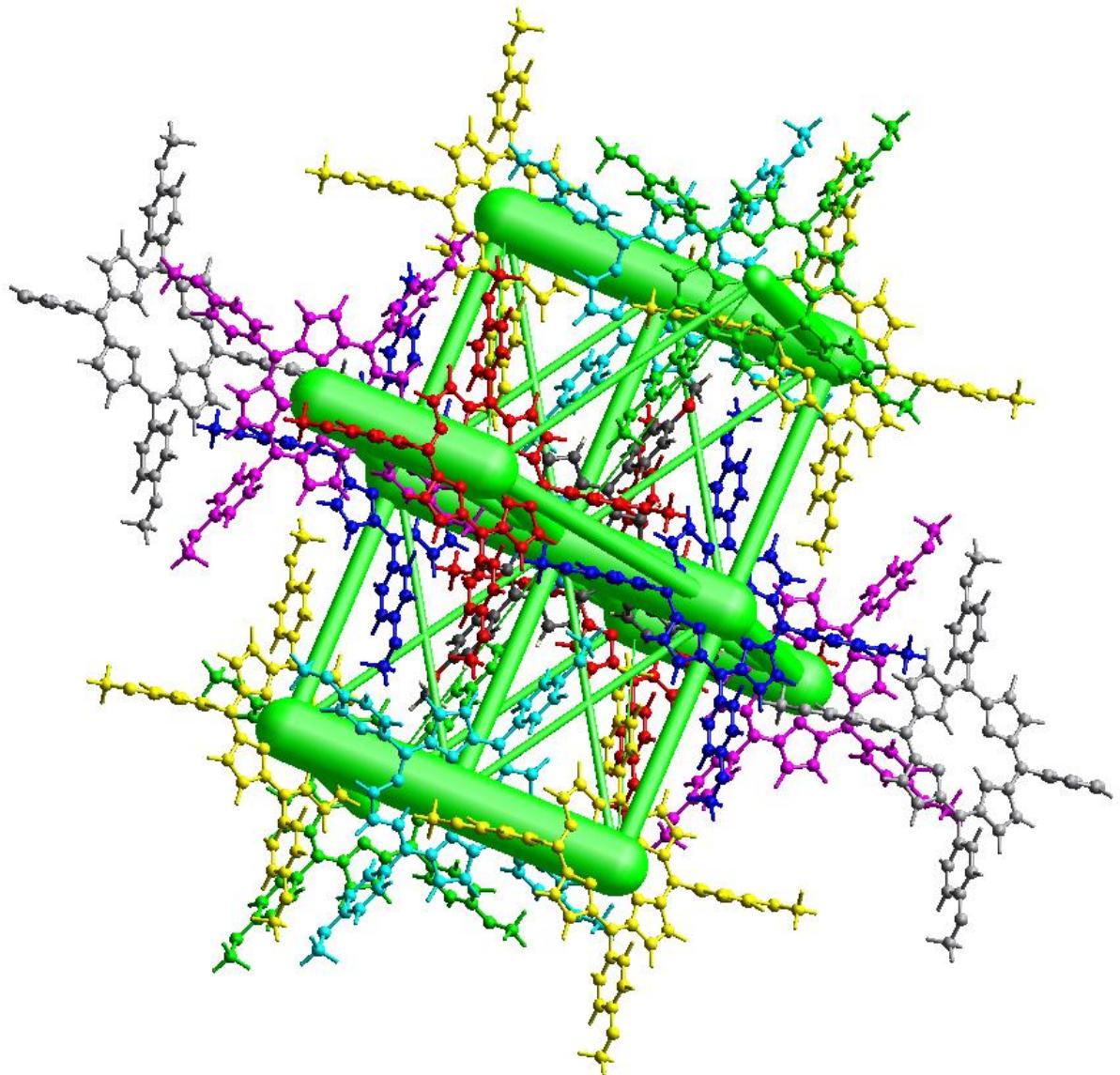


Figure S4. Dispersion interacting energies between molecules in the H₂-TMPP crystal.

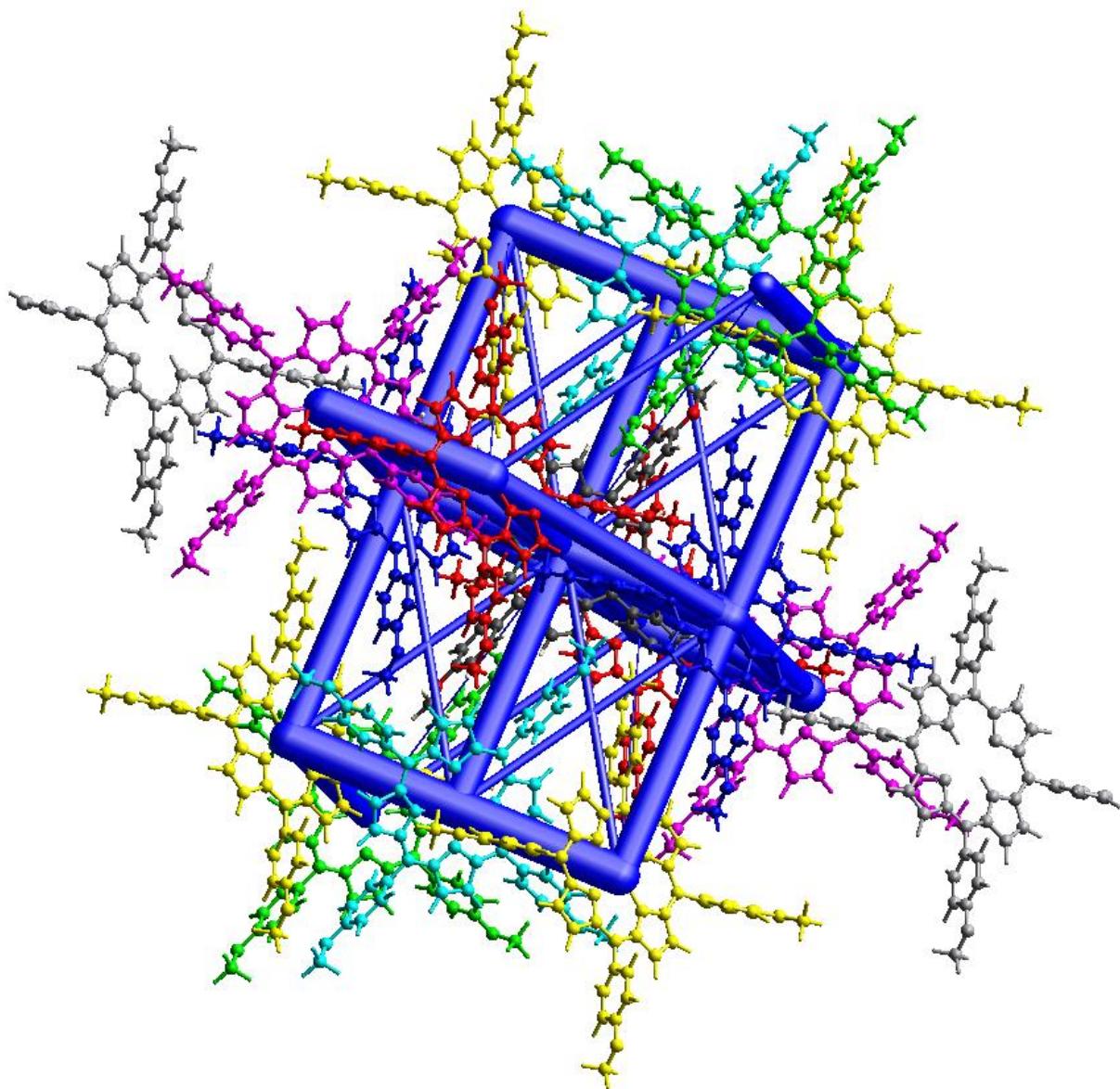


Figure S5. Total interacting energies between molecules in the H₂-TMPP crystal.

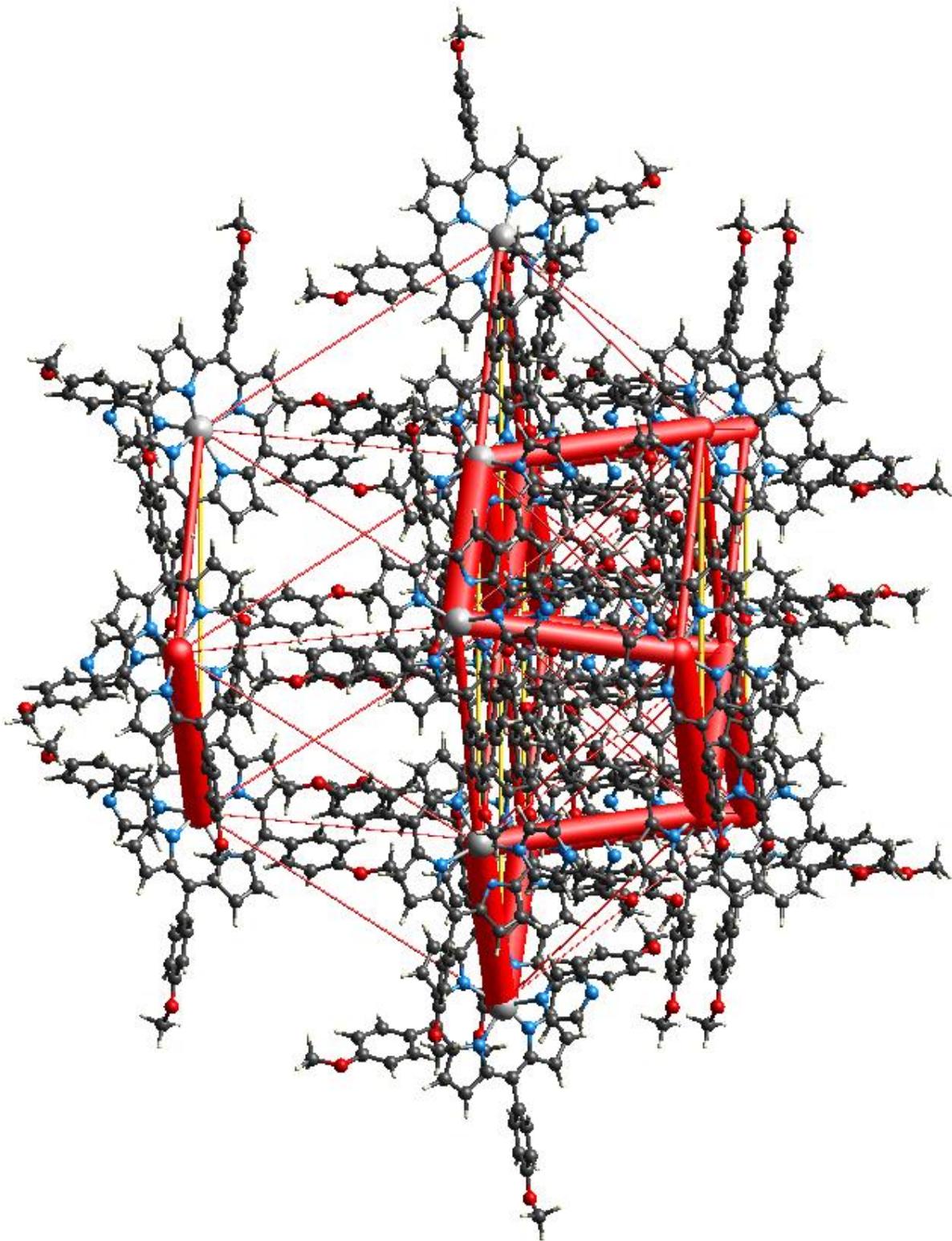


Figure S6. Coulomb interacting energies between molecules in the $[\text{Cd(TMPP)}(\text{Pyz})]$ crystal.

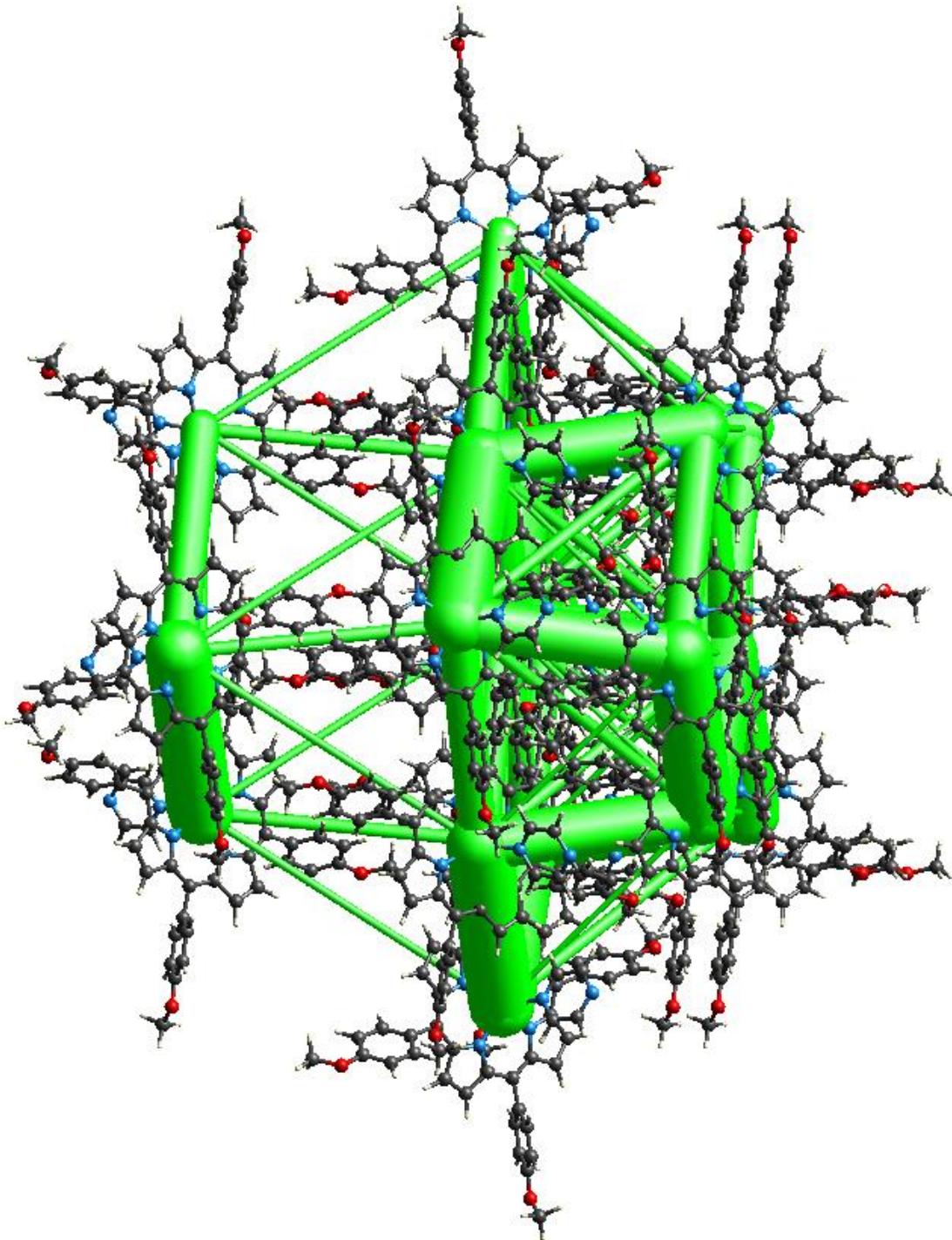


Figure S7. Dispersion interacting energies between molecules in the $[\text{Cd}(\text{TMPP})(\text{Pyz})]$ crystal.

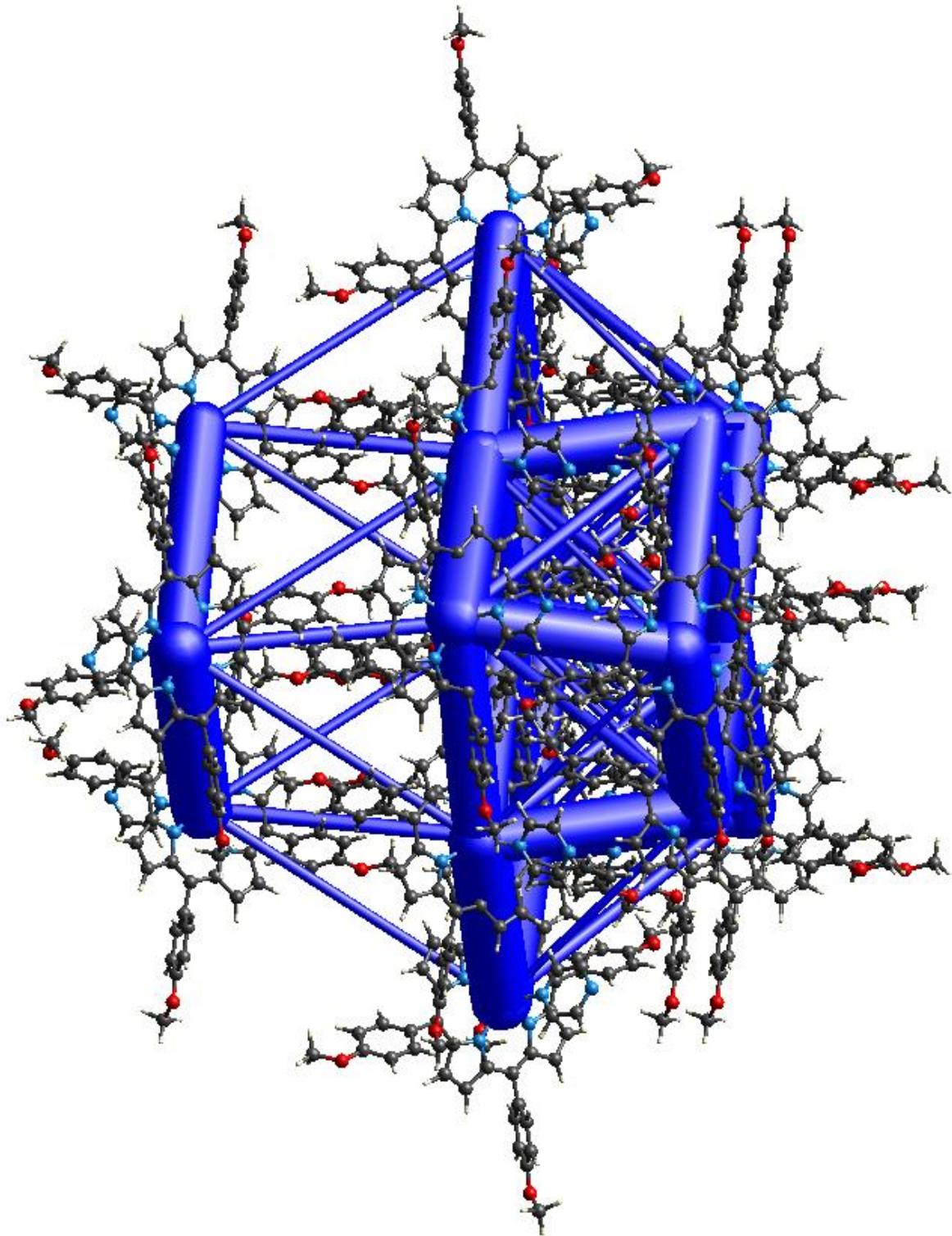


Figure S8. Total interacting energies between molecules in the $[\text{Cd}(\text{TMPP})(\text{Pyz})]$ crystal.

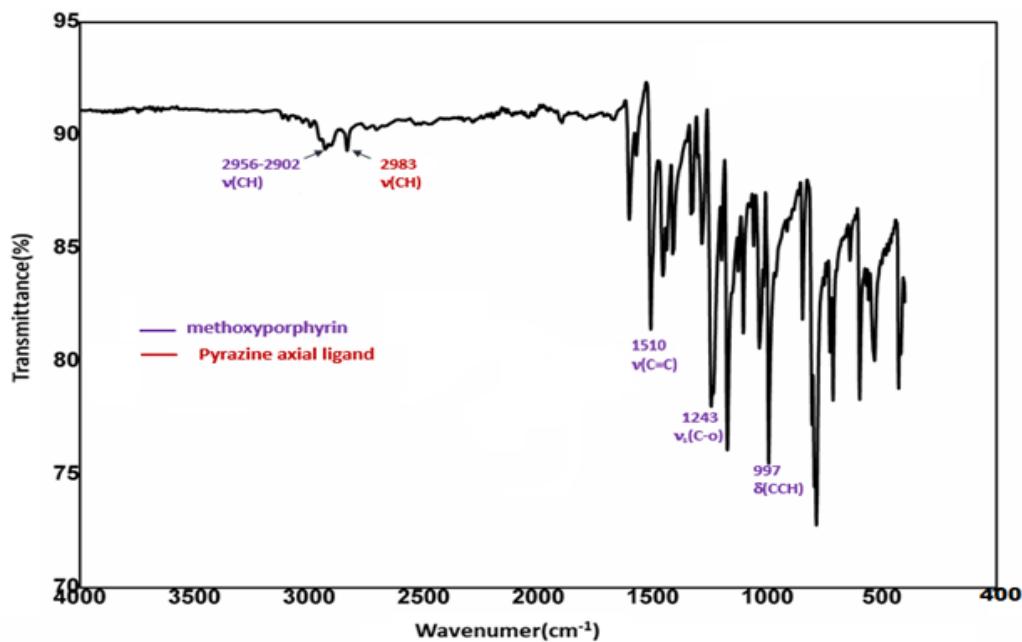


Figure S9. FTIR spectrum of solid $[\text{Cd}(\text{TMPP})(\text{Pyz})]$ (**1**).

Table S7. Experimental and Theoretical IR assignments of the complex (**1**).

[Cd (TMPP)(Pyz)] (1)		
Experimental (cm^{-1})	Theoretical (cm^{-1})	Assignment
1243	1290	(O-CH ₃)
1510	1561	(C=C)
2833	3203	(C-H)Pyz
2956-2902	3215-3282	(C-H)porph
997	1009	((δ(CCH))porph

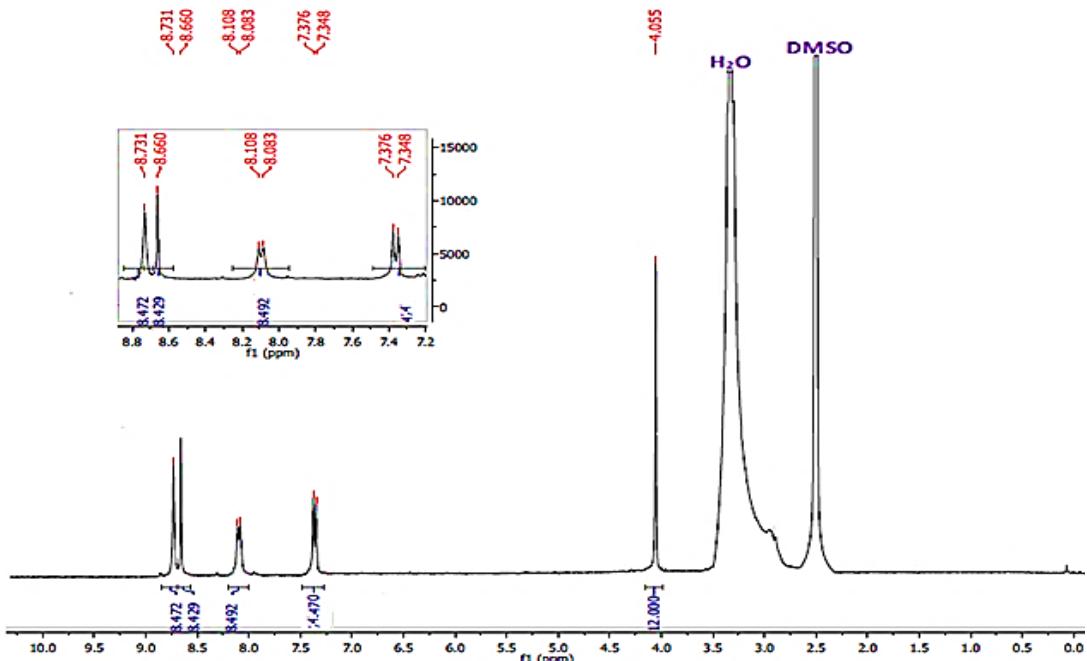


Figure S10. ^1H NMR spectrum (300 MHz, DMSO- d_6 , 298 K) of $[\text{Cd}(\text{TMPP})(\text{Pyz})]$ (1).

Table S8. ^1H NMR data (δ in ppm) of the $[\text{Cd}(\text{TMPP})(\text{Pyz})]$ complex in DMSO- d_6 .

H (ppm)	Experimental	Theoretical
H β -pyr	8.73	8.1
Ho-Ph/ Hm-Ph	8.10/ 8.08	6.1
p- OCH ₃	4.05	3.9
H(Pyz ligand)	7.34	8.2-8.7

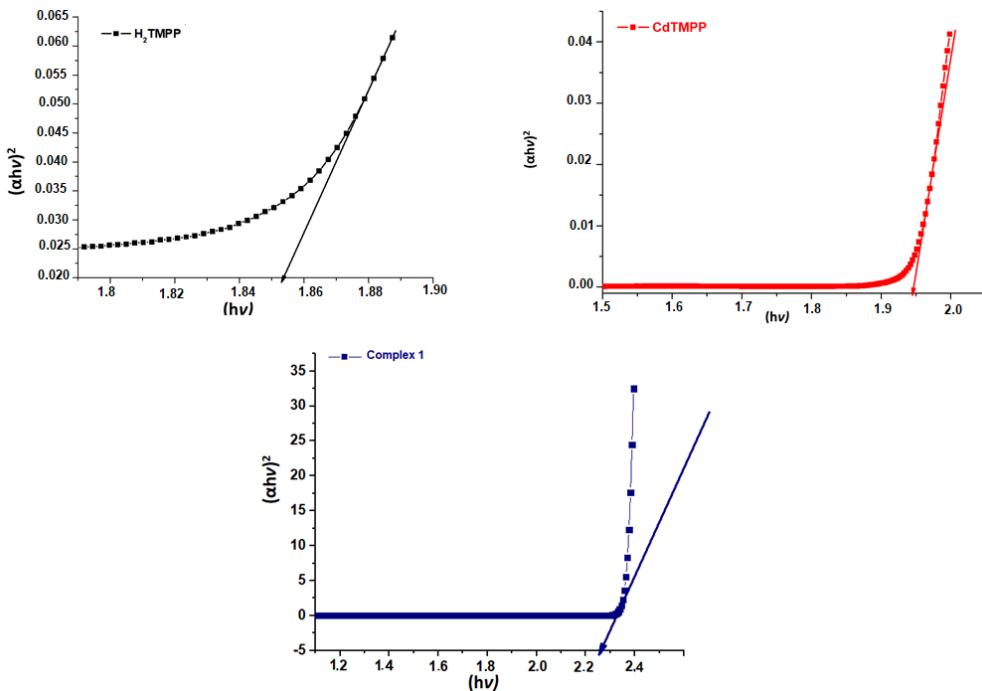


Figure S11. Plots of $(\alpha h\nu)^2$ vs photon energy E of H₂TMPP, CdTMPP and Complex 1. α is the absorption coefficient.

Table S9. UV-Visible data of H₂TMPP, [Cd(TMPP)], [Cd(TMPP)(Pyz)] and several [Cd(porph)(L)] porphyrin complexes .

Compound	Soret Band ^{a,b}	Q bands ^{a,b}	Eg	Ref.
			(eV)	
H ₂ -TMPP ^c	422 (463.2)	518 (14.13), 556 (9.7), 593 (4.3), 650 (4.8)	-	[61]
[Cd(TMPP)] ^c	436(367.7)	572(15.02), 616(15.87)	-	[61]
H ₂ -TMPP ^c	422(390.2)	518(14.10), 555(9.5), 594(3.9), 650(4.2)	1.85	This work
[Cd(TMPP)] ^c	437(125)	573(4.2), 617(2.4)	1.95	This work
H ₂ -TClPP ^c	420 (309)	517 (15.8), 553(9.2), 589 (5.6), 645 (4.1)	1.820	[36]
[Cd(TClPP)] ^c	433 (117)	568 (2.3), 609(2.6)	1.960	[36]
[Cd(T(p-Cl)4PP)(py)] ^c	429(301)	- -	-	[62]
[Cd(T(p-Cl)4PP)(DMF)] ^c	433(310)	567(18), 609(14)	-	[62]
[Cd(TPP)(2-NH ₂ -py)] ^d	433(310)	567(18), 609(14)	-	[25]
[Cd(TClPP)(morph)] ^c	436(126)	575(15.84), 620(3.6)	1.903	[36]
[Cd(TBPP)(2-MeHIm)] ^c	438 (109.65)	579 (14.65), 621 (3.5)	1.920	[22]
[Cd(TMPP)(Pyz)] ^c	440(113)	579(15.2) 623(3.2)	2.23	This work

^a: λ_{max} (nm), ^b: $\log \epsilon$. (ϵ . 10^{-3} L.cm⁻¹.mol⁻¹), ^c: in dichloromethane solvent, ^d: in chloroform solution.

Table S10. Photophysical data of our synthetic porphyrin species and a selection of Cd(porphyrin) complexes were recorded in CH_2Cl_2 .

Compound	Fluorescence a λ_{max} (nm)		Φf^b	τ_f (ns) ^c	$\Phi \Delta^d$	Ref.
	Q(0,1)	Q(0,0)				
H ₂ -TCIPP ^e	653	678.8	0.063	8.2	0.69	[63]
H ₂ -TCIPP	650	715	0.075	8.9	0.49	[36]
H ₂ -TBPP	650	717	0.08	8.9	-	[22]
H ₂ -TMPP	656	722	0.084	7.8	0.73	This work
[Cd(TPP)] ^f	620	655	0.065	-	0.69	[33]
[Cd(TMPP)] ^g	622	667	0.018	-	0.73	[61]
[Cd(TCIPP)]	619	650	0.06	1.6	0.41	[36]
[Cd(TCIPP)(morph)]	619	652	0.046	1.5	0.53	[36]
[Cd(TBPP)]	614	650	0.01	1.6	-	[22]
[Cd(TBPP)(2-MeIm)]	612	657	0.02	1.7	-	[22]
[Cd(TMPP)(DABCO)]	625	652	0.01	1.2	-	[21]
[Cd(TMPP)]	618	652	0.06	1.8	0.57	This work
[Cd(TMPP)(Pyz)](1)	618	652	0.03	1.6	0.13	This work

a: Fluorescence spectra of porphyrins were obtained as a function of $\lambda_{\text{ex}} = 437$ nm. b: Fluorescence quantum yield. c: Fluorescence lifetime (ns). d: Singlet oxygen production quantum yield. e: in toluene solvent, f: in EtOH solvent, g: in CH_2Cl_2 solvent.

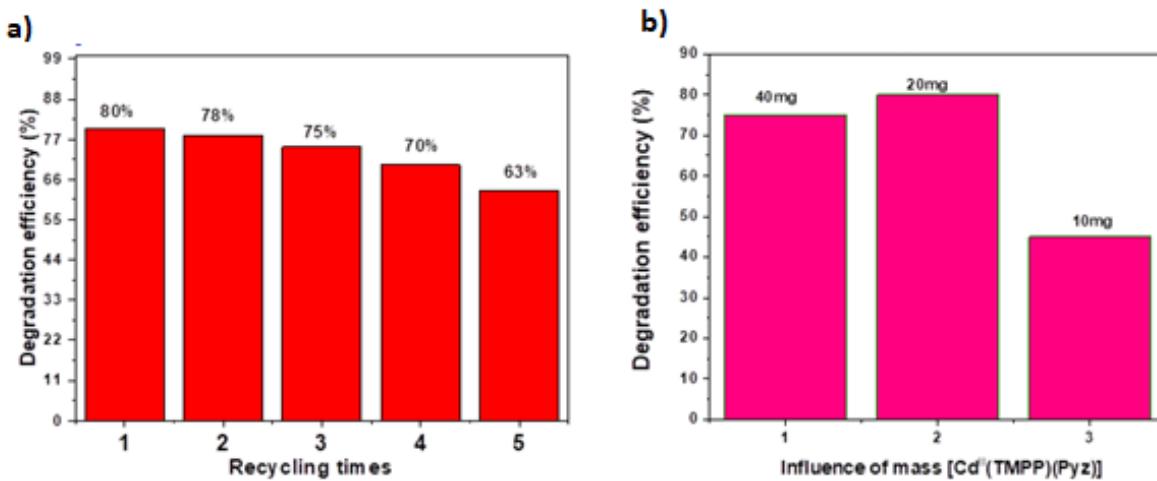


Figure S12. (a) Cycling experiments. (b) Influence of mass of $[\text{Cd}^{\text{II}}(\text{TMPP})(\text{Pyz})]$ (1).