## Supplementary materials

Manuscript Title: "Hydrogel films based on chitosan and oxidized carboxymethylcellulose optimized for the controlled release of curcumin with applications in treating dermatological conditions"

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## Figure S1:



Figure S1: Schematic representation of $\mathrm{CMC}^{\prime}$ s oxidation reaction under the sodium periodate action.

## Figure S2:



Figure S2. The structure of the hydrogel films based on chitosan and oxidized carboxymethyl cellulose - schematic presentation

Figure S3


Figure S3. Antioxidant activity determination expressed by IC50 values for the analyzed samples using the DPPH assay.

## Figure S4:



Figure S4: The CS calibration curve determined with ninhydrine test

Figure S5


(a)
(c)

Figure S5. Calibration curves of curcumin in ethanol (a), phosphate buffer at $\mathrm{pH}=7.4$ (b), and acetate buffer at $\mathrm{pH}=5.5$ (c)

## Table S1.

Table SI. The CI values (\%) for samples obtained by chemical cross-linking and physical interaction between CS and CMCOx, respectively, by the CS amino groups' interaction with CMC's carboxylic groups.

| Samples | Molar <br> ratios | CIchemical cross-linking and <br> physical interactions (\%) | $\mathrm{CI}_{\text {physical }}$ <br> interactions $(\%)$ | CIchemical cros-linking and physical <br> interactions <br> CIChemical cross-linking (Shiff base) $(\%)$ |
| :---: | :---: | :---: | :---: | :---: |
| 1 | 2 | 3 | 4 | 5 |
| P1 | $0.25: 1$ | $42.27 \pm 0.1$ | $16.47 \pm 2.5$ | 25.8 |
| P2 | $0.375: 1$ | $49.05 \pm 9.3$ | $17.97 \pm 2.3$ | 31.08 |
| P3 | $0.5: 1$ | $61.83 \pm 7.3$ | $23.89 \pm 3.7$ | 37.94 |

