

Supporting information

Nano-sized NiO immobilized on Au/CNT for benzyl alcohol oxidation:

Influences of hybrid structure and interface

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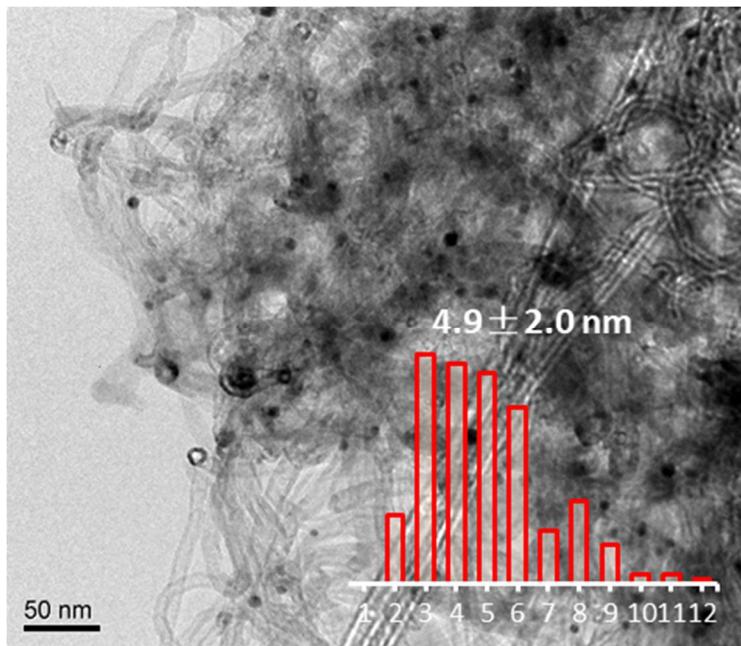


Figure S1. TEM image and size distribution of Au/CNT catalyst.

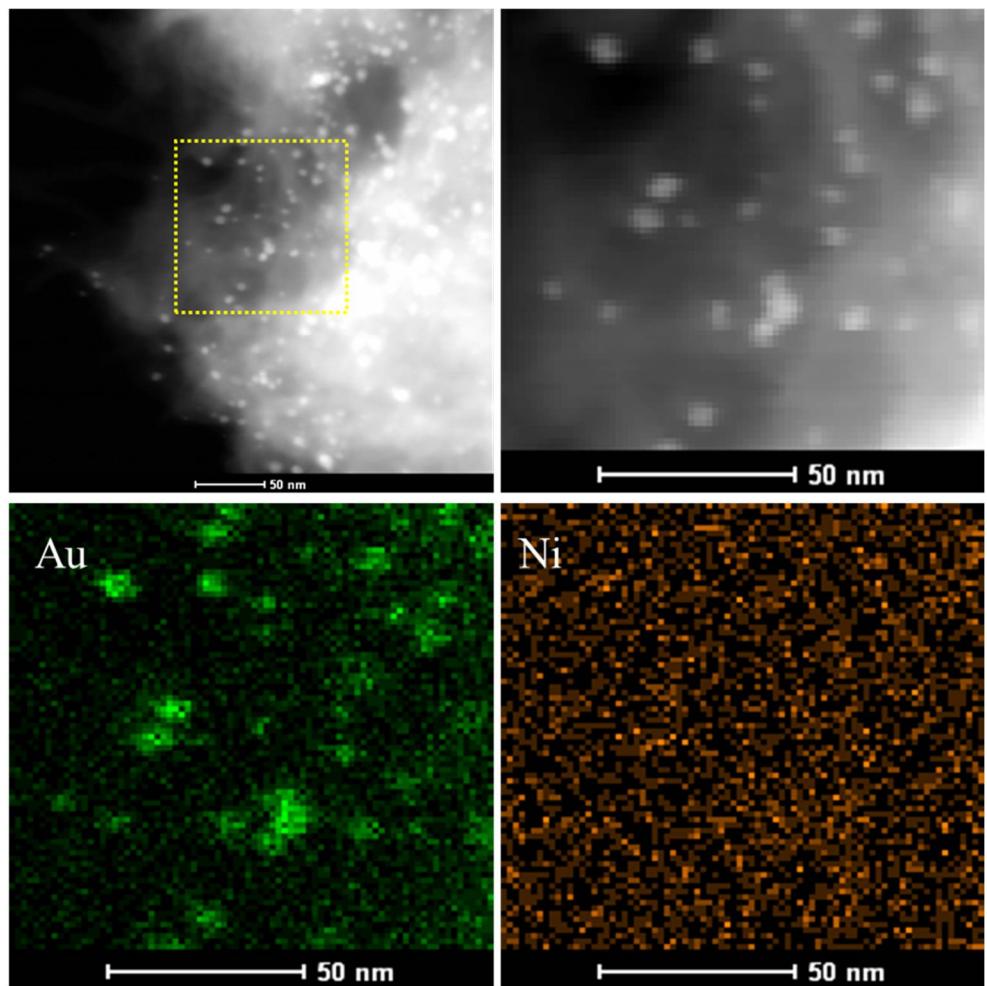


Figure S2. HAADF-STEM element mappings of the AuNi₈/CNT catalyst.

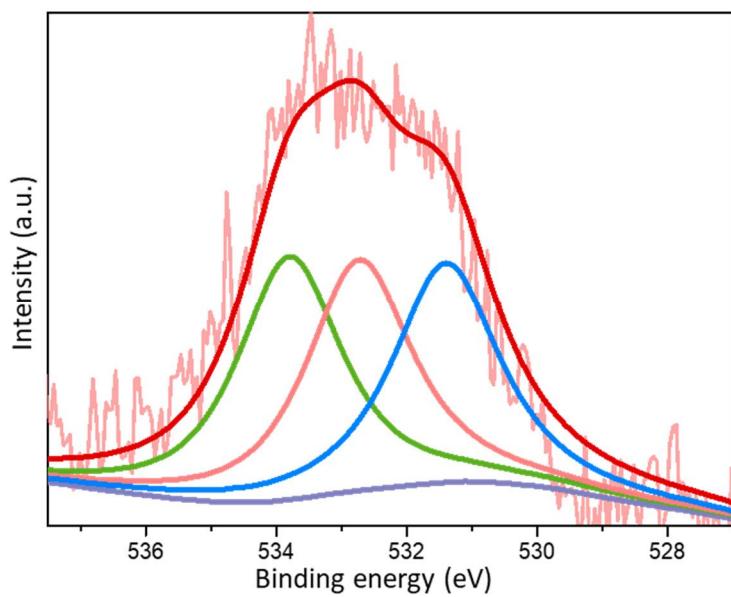


Figure S3. XPS spectra of the O 1s (c) core level of Au/CNT reference sample.



Figure S4. A typical GC analysis graph by Au reflecting the chromatographic peaks of benzaldehyde, benzyl alcohol, and biphenyl (substance) in the reaction mixture.

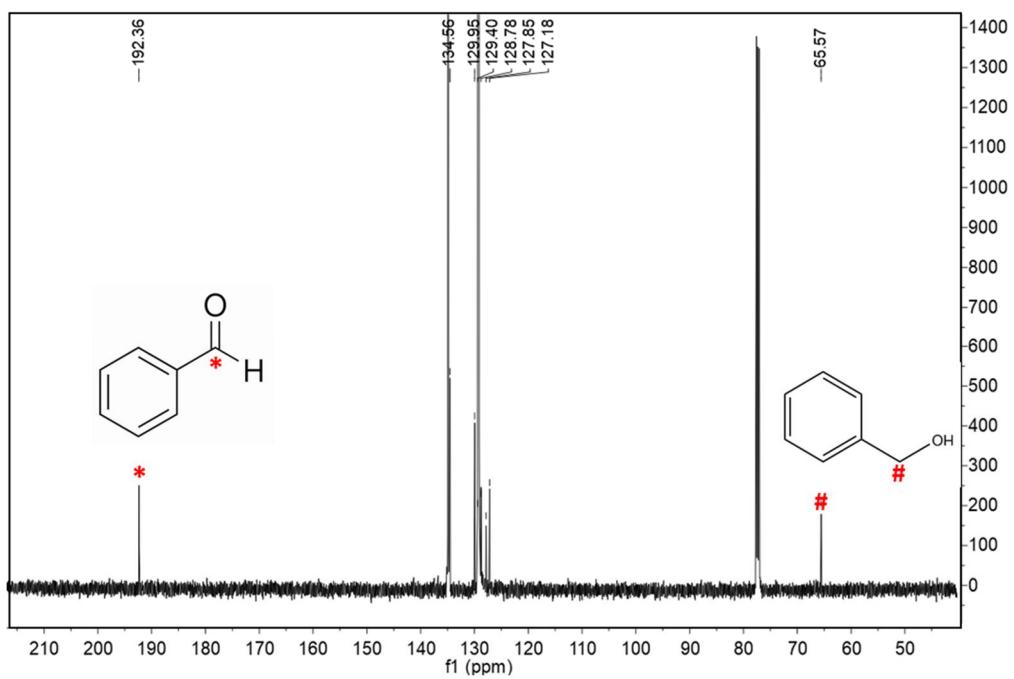


Figure S5. ^{13}C NMR spectra of typical reaction mixture.

Table S1 Surface composition of different gold catalysts.

Catalyst	Au 4f		Ni 2p		Au/C% ^a	Ni/Au ^b
	Au ⁰ (%)	Au ^{δ+} (%)	Ni ⁰ (%)	Ni ²⁺ (%)		
Au/CNT	100	0	n.a.	n.a.	3.1	n.a.
AuNi ₂ /CNT	100	0	n.a.	n.a.	2.9	n.a.
AuNi ₈ /CNT	90.1	9.9	17.4	82.6	3.4	11.2
AuNi ₁₂ /CNT	86.1	13.9	0	100	3.3	14.8

^a Surface weight ratio of Au/C by XPS results.^b Surface atomic ratio of Ni/Au by XPS results.

Table S2 Comparison of specific rate in the current work and the reported literature

Catalyst	Reaction temperature	Specific rate ($\mu\text{mol/g/s}$)	Literature
Au/CNT	50 °C	163	This work
AuNi ₈ /CNT	50 °C	185	This work
AuNi ₁₂ /CNT	50 °C	208	This work
AuNi ₁₆ /CNT	50 °C	182	This work
0.5Au-CeZr	100 °C	370	[1]
RMC-Au	130 °C	212	[2]
Au/CZ	80 °C	106	[3]
Au-Pd/TiO ₂	90 °C	1239	[4]
Au-Pd/TiO ₂ -001	120 °C	26002	[5]

References

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