

# Supplementary Materials: Synthesis of Catechol Derived Rosamine Dyes and Their Reactivity toward Biogenic Amines

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# NMR spectrum of 2,3-dibenzyloxybenzaldehyde and 3,4-dibenzyloxybenzaldehyde

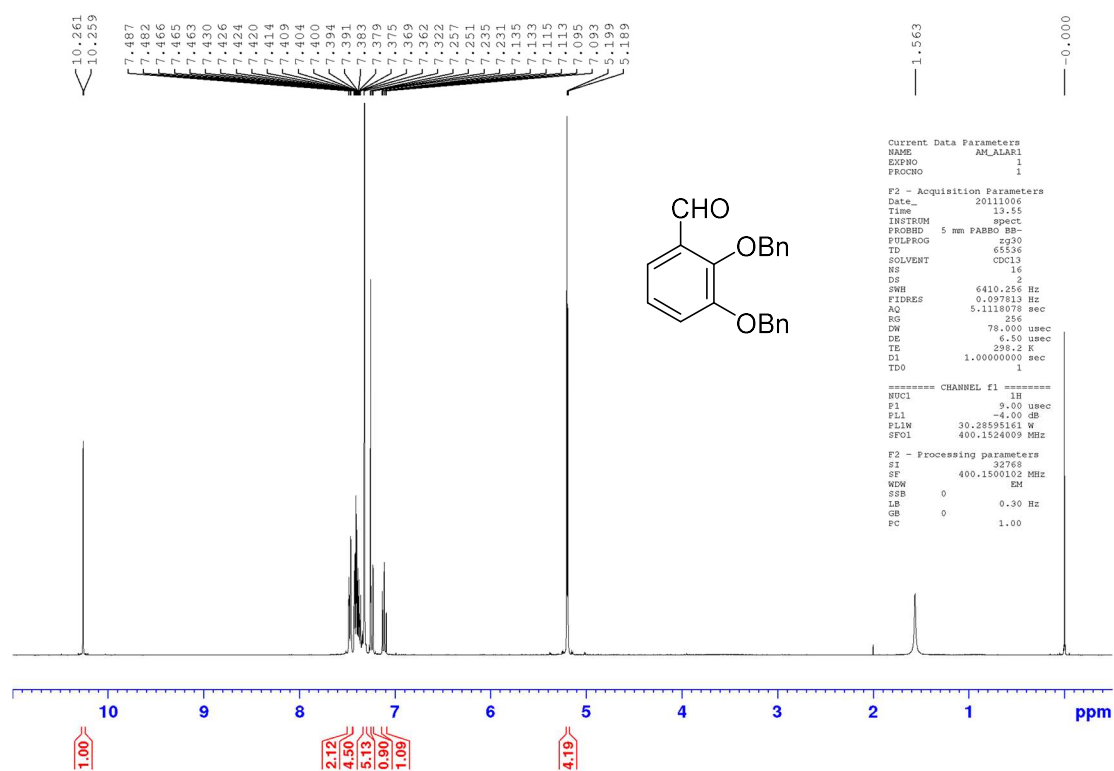


Figure S1. <sup>1</sup>H NMR spectrum (400.15 MHz, CDCl<sub>3</sub>) of 2,3-dibenzyloxybenzaldehyde.

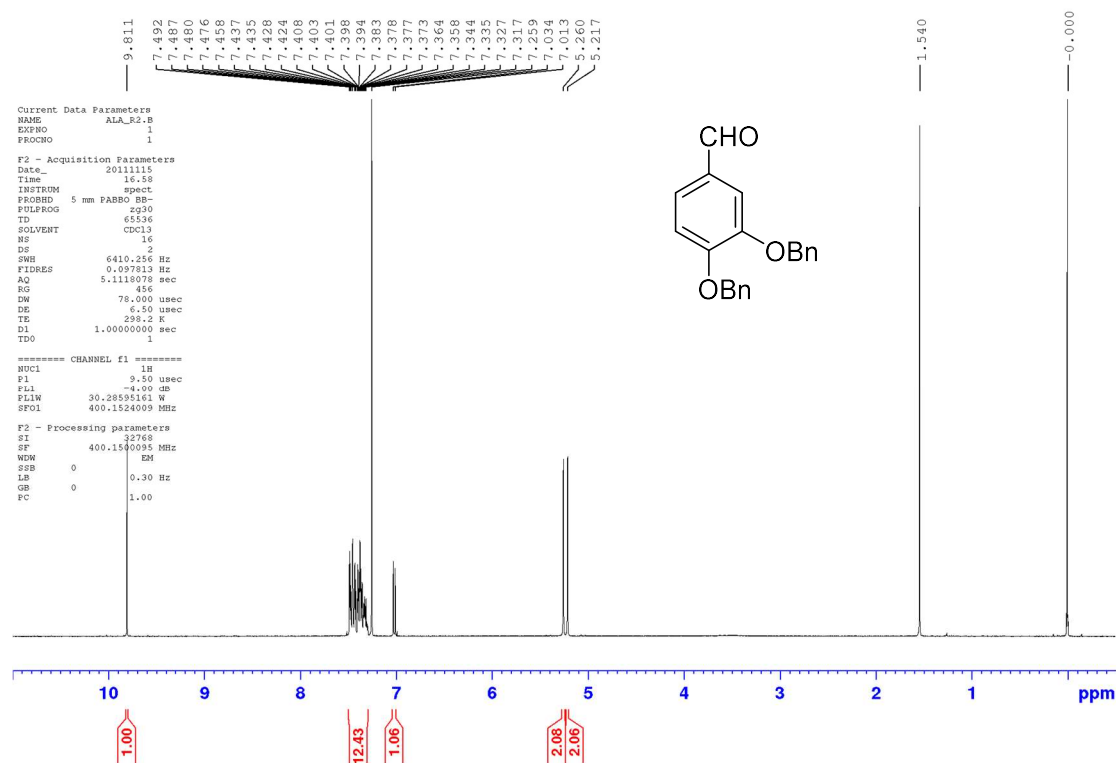


Figure S2.  $^1\text{H}$  NMR spectrum (400.15 MHz,  $\text{CDCl}_3$ ) of 3,4-dibenzyloxybenzaldehyde.

### NMR spectra of rosamine 1

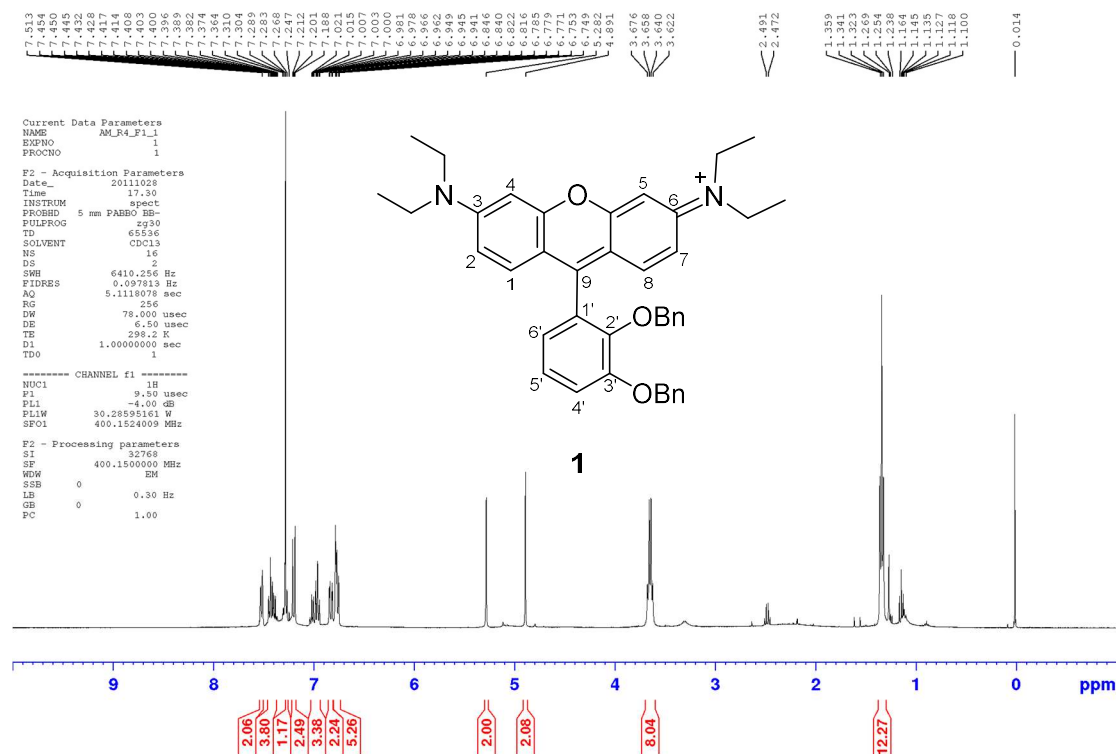


Figure S3.  $^1\text{H}$  NMR spectrum (400.15 MHz,  $\text{CDCl}_3$ ) of 1.



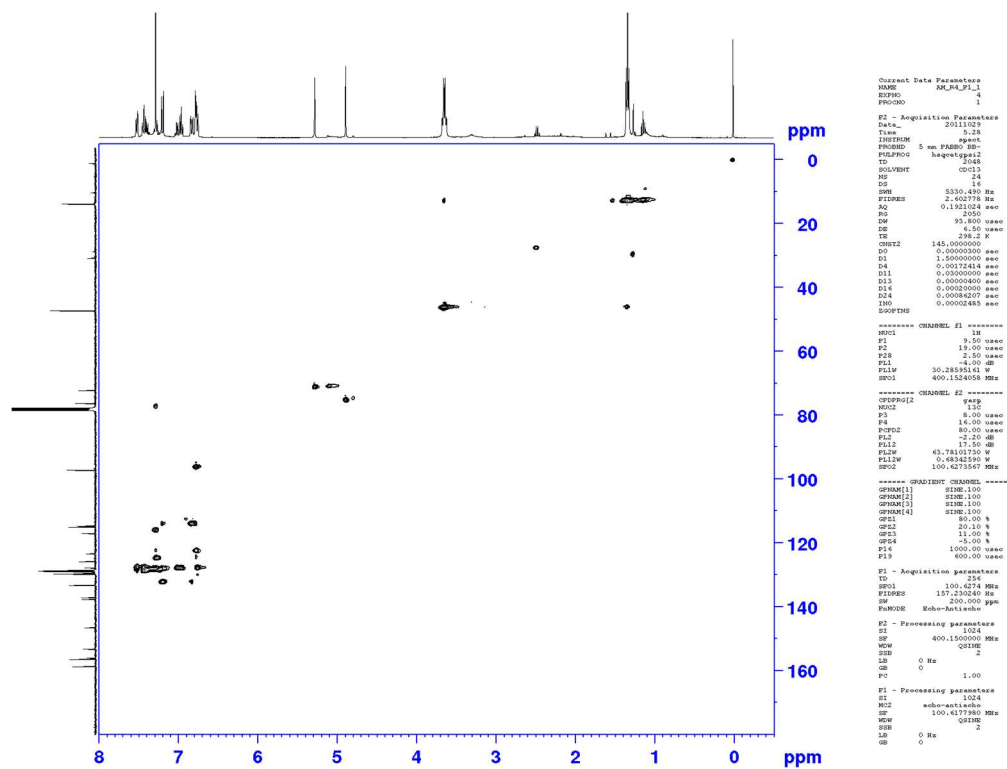


Figure S6. HSQC spectrum of 1.

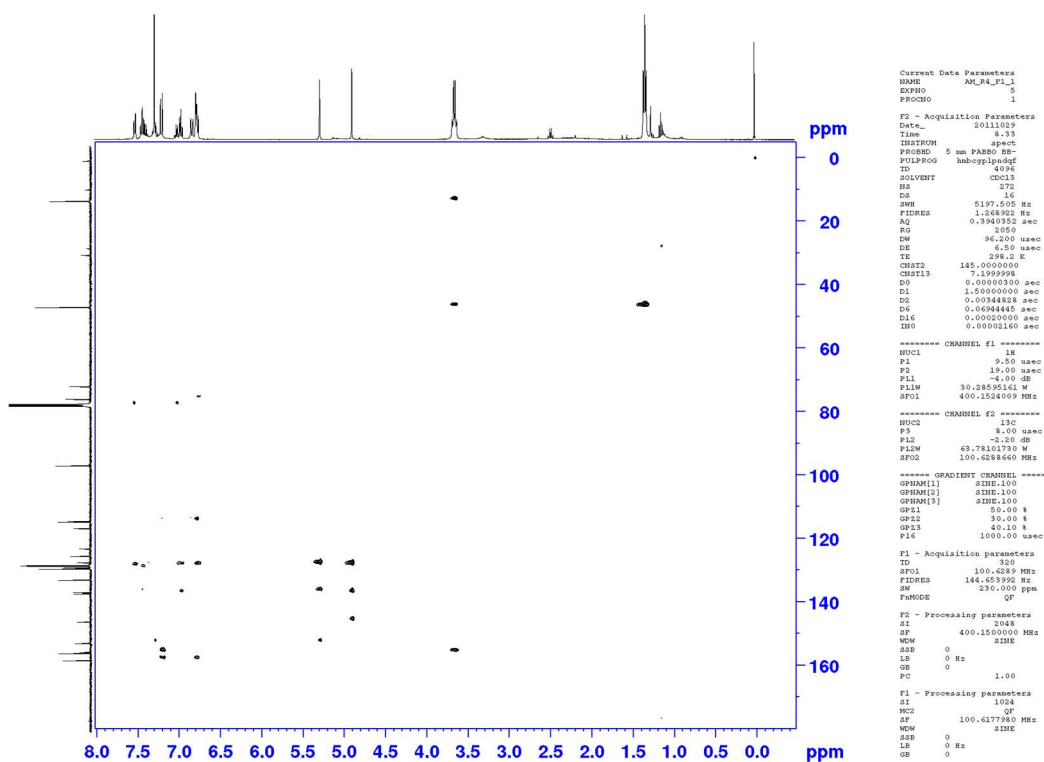


Figure S7. HMBC spectrum of 1.

## NMR spectra of rosamine 2

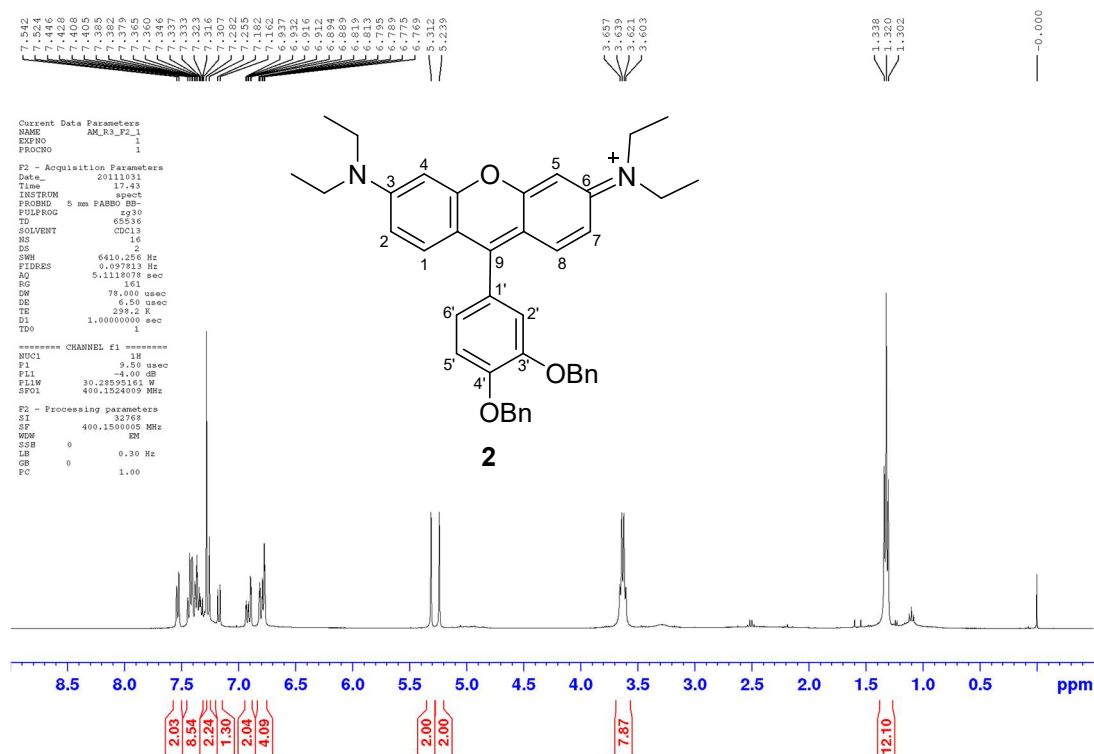


Figure S8.  $^1\text{H}$  NMR spectrum (400.15 MHz,  $\text{CDCl}_3$ ) of 2.

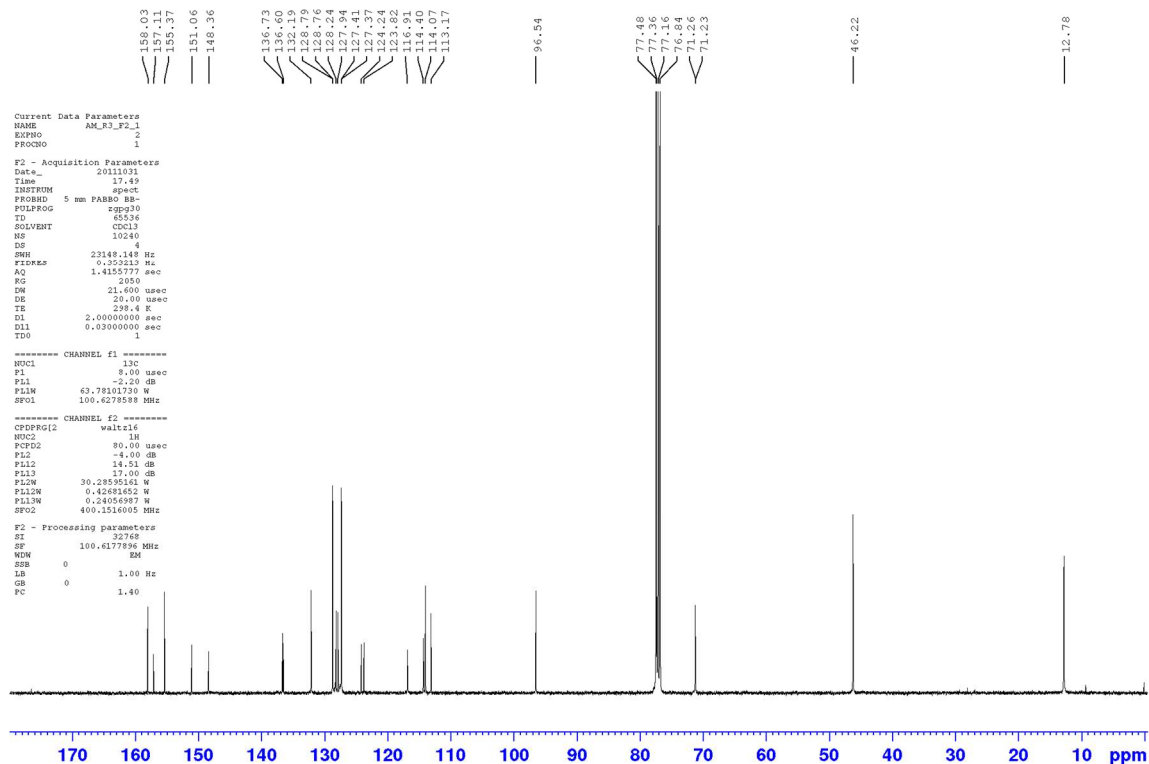


Figure S9.  $^{13}\text{C}$  NMR spectrum (100.63 MHz,  $\text{CDCl}_3$ ) of 2.

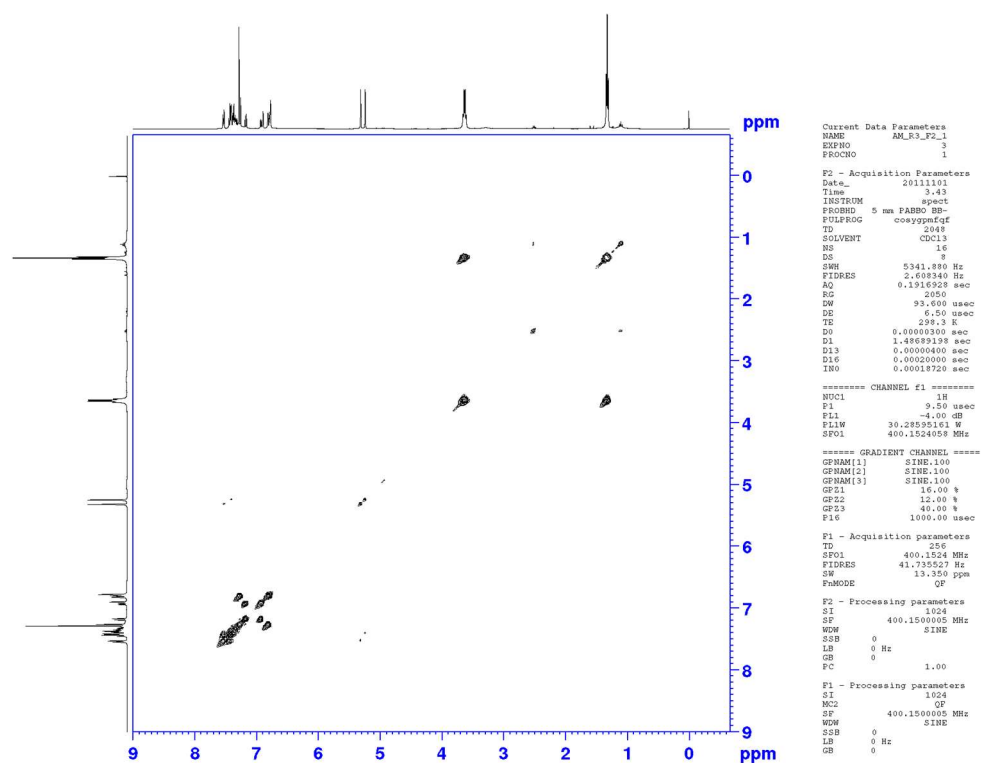


Figure S10. COSY spectrum of 2.

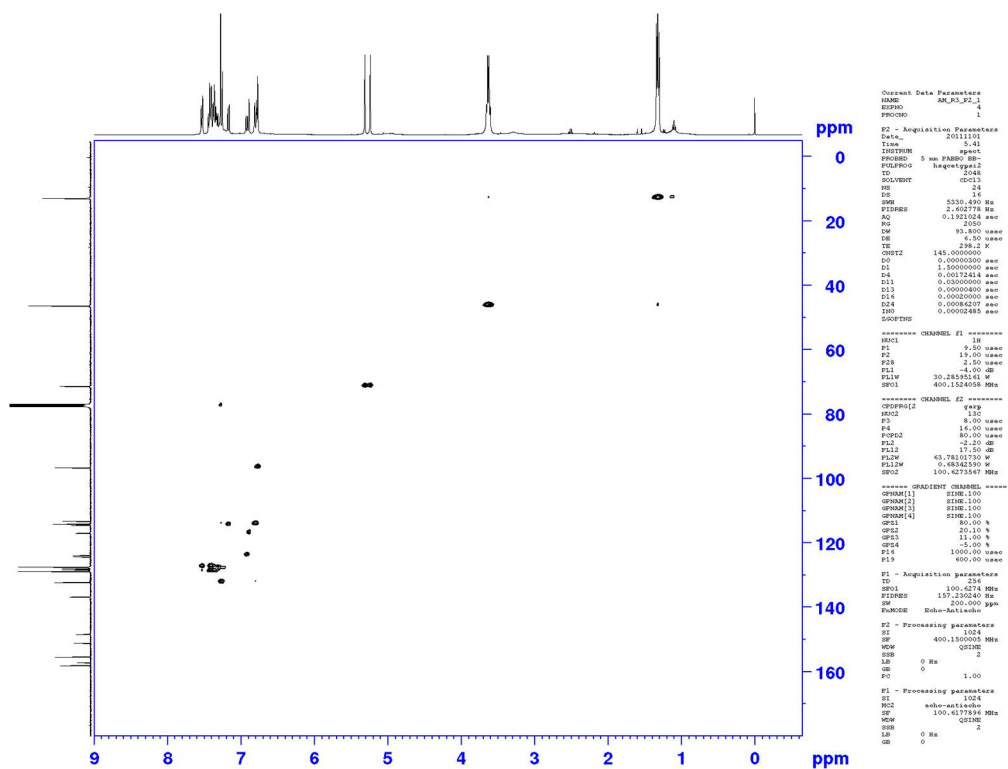


Figure S11. HSQC spectrum of 2.

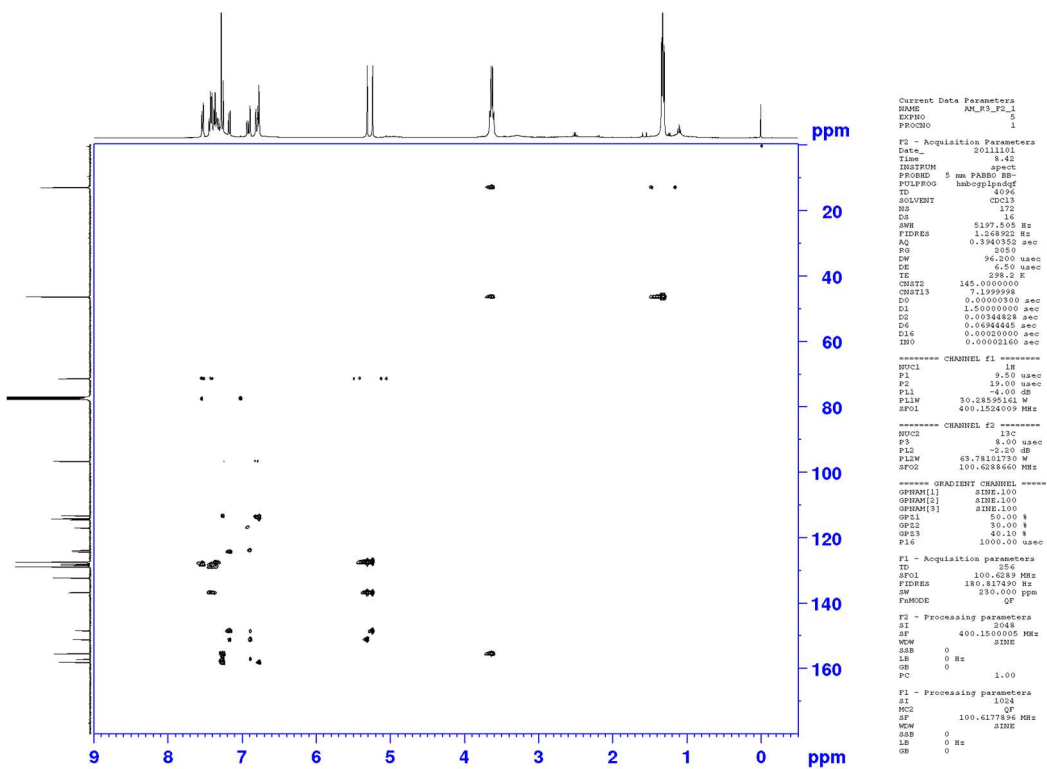


Figure S12. HMBC spectrum of **2**.

### NMR spectra of rosamine 3

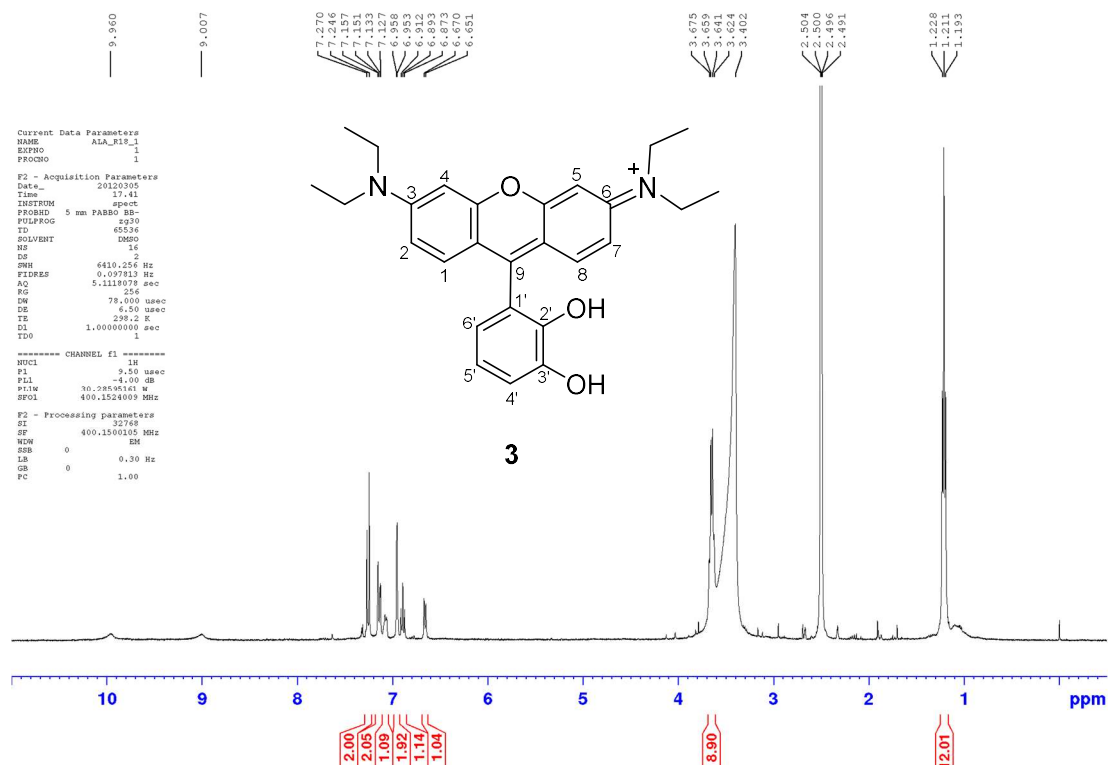


Figure S13.  $^1\text{H}$  NMR spectrum (400.15 MHz, DMSO- $d_6$ ) of **3**.



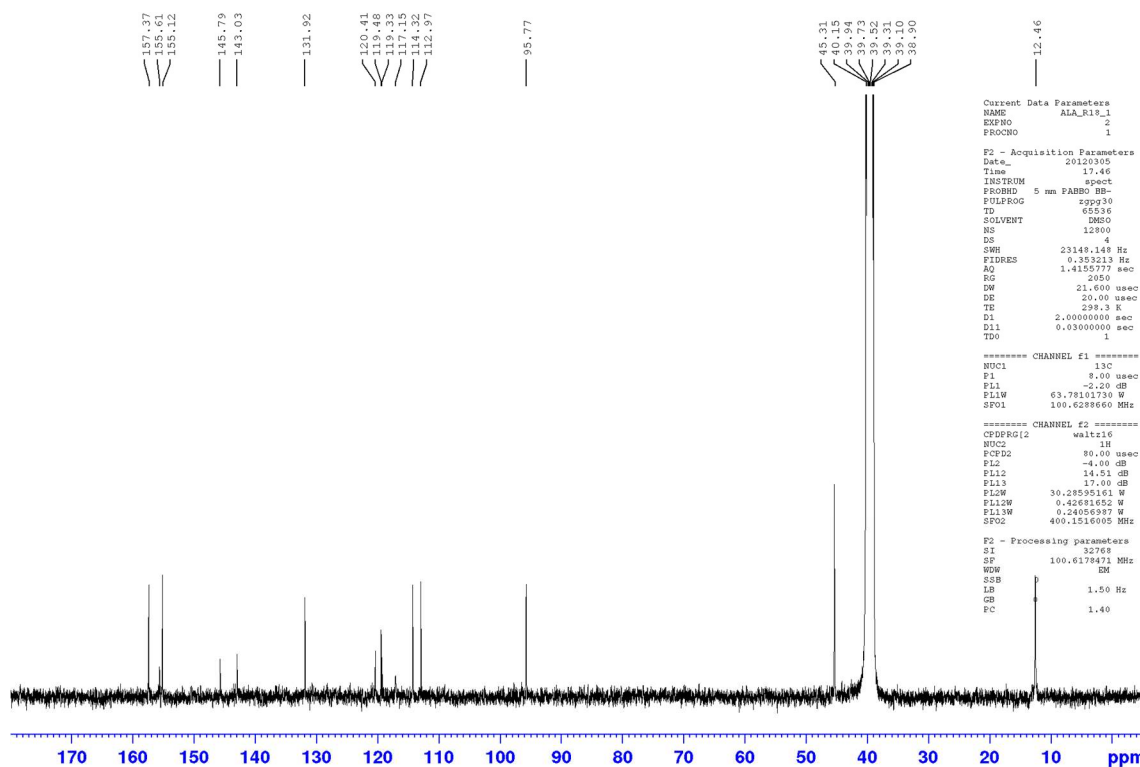


Figure S14.  $^{13}\text{C}$  NMR spectrum (100.63 MHz, DMSO- $d_6$ ) of **3**.

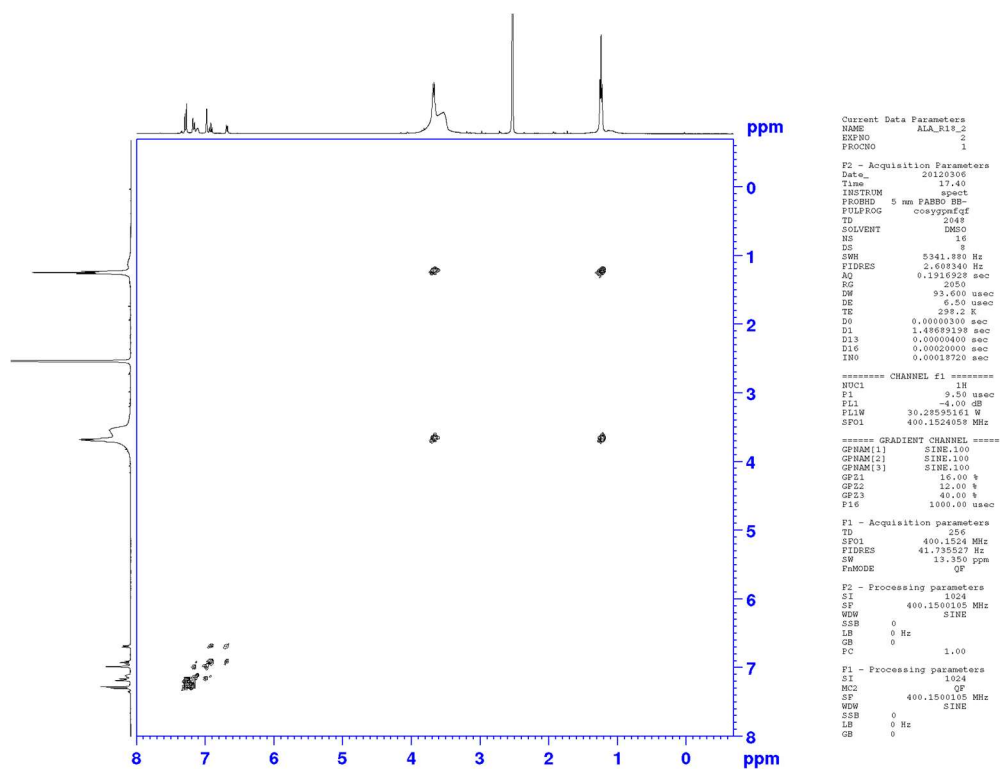
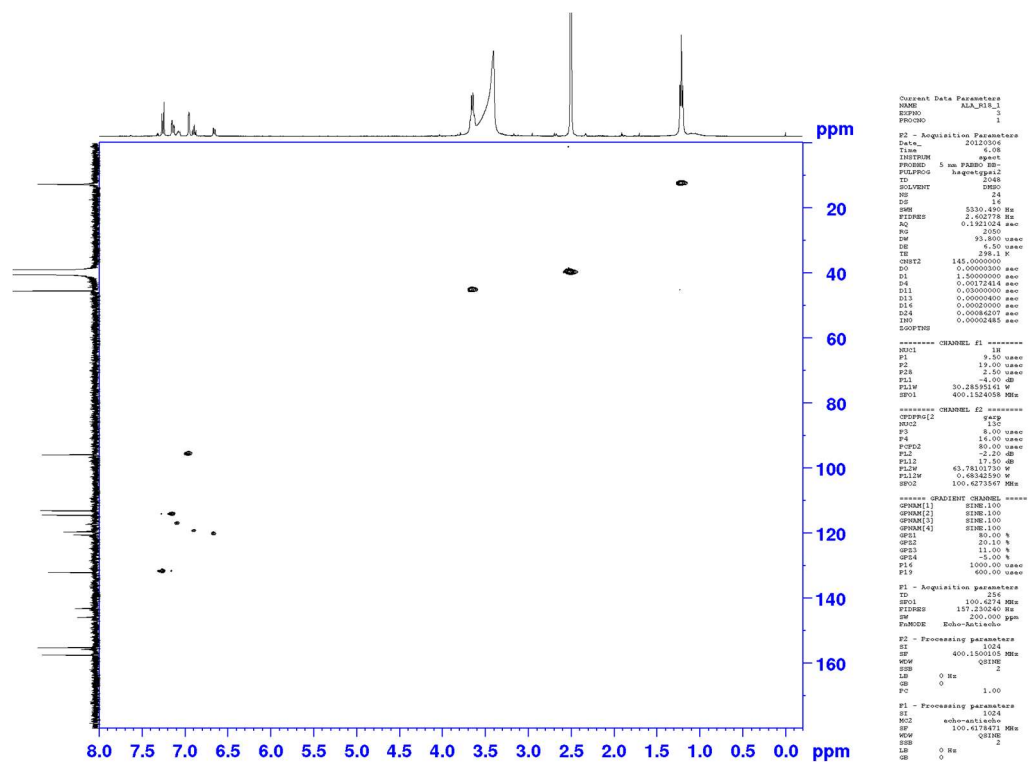
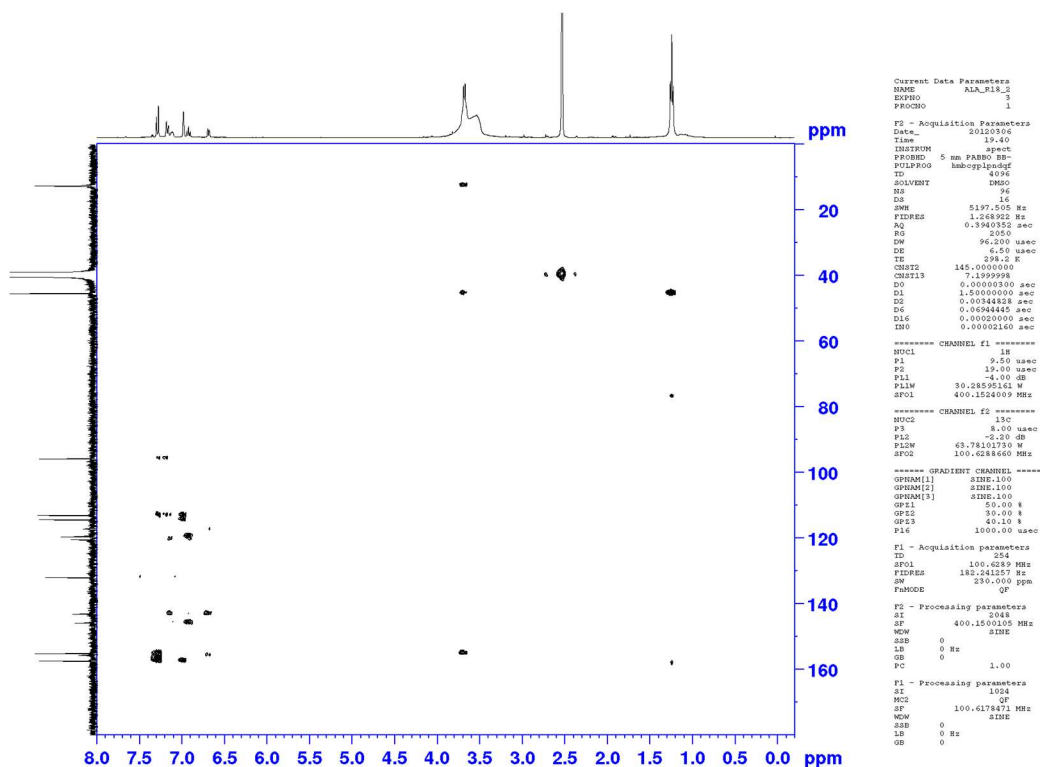


Figure S15. COSY spectrum of **3**.

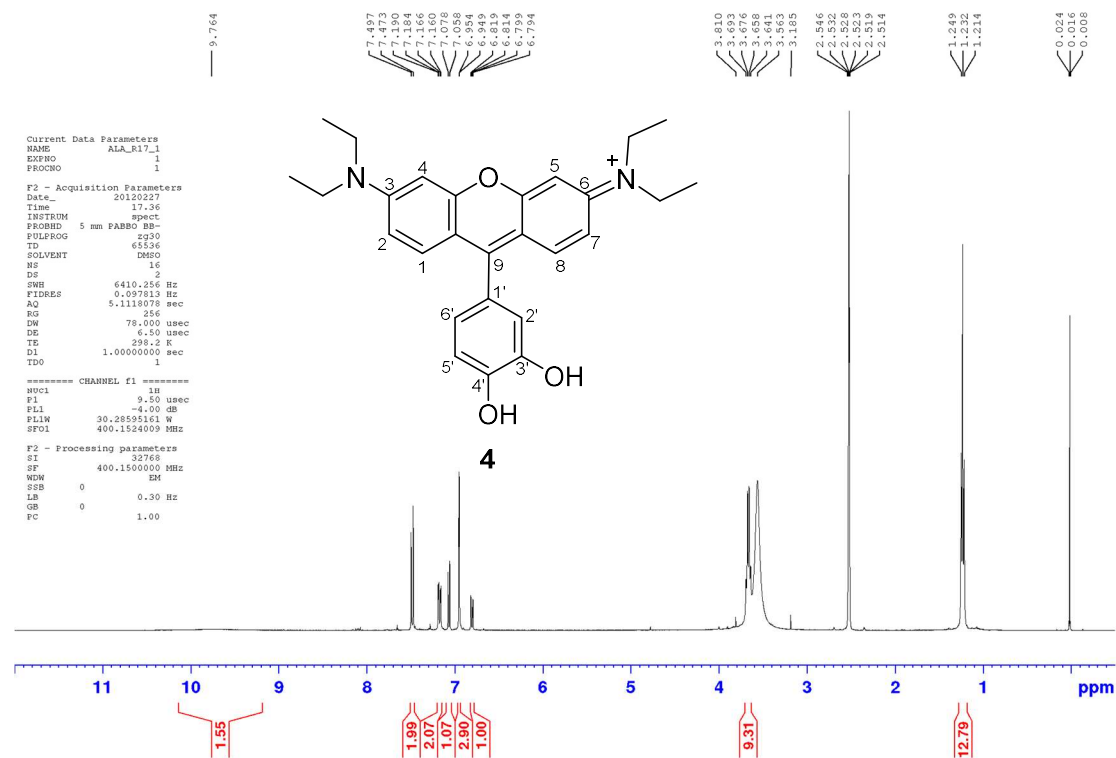


**Figure S16.** HSQC spectrum of **3**.



**Figure S17.** HMBC spectrum of **3**.

# NMR spectra of rosamine 4



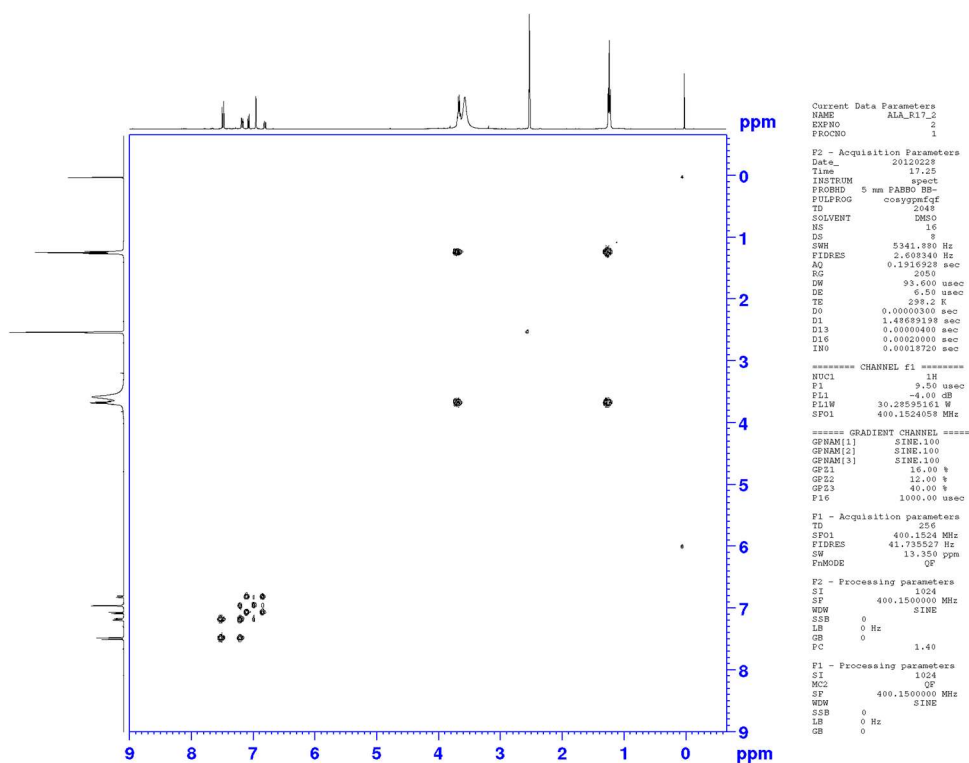


Figure S20. COSY spectrum of 4.

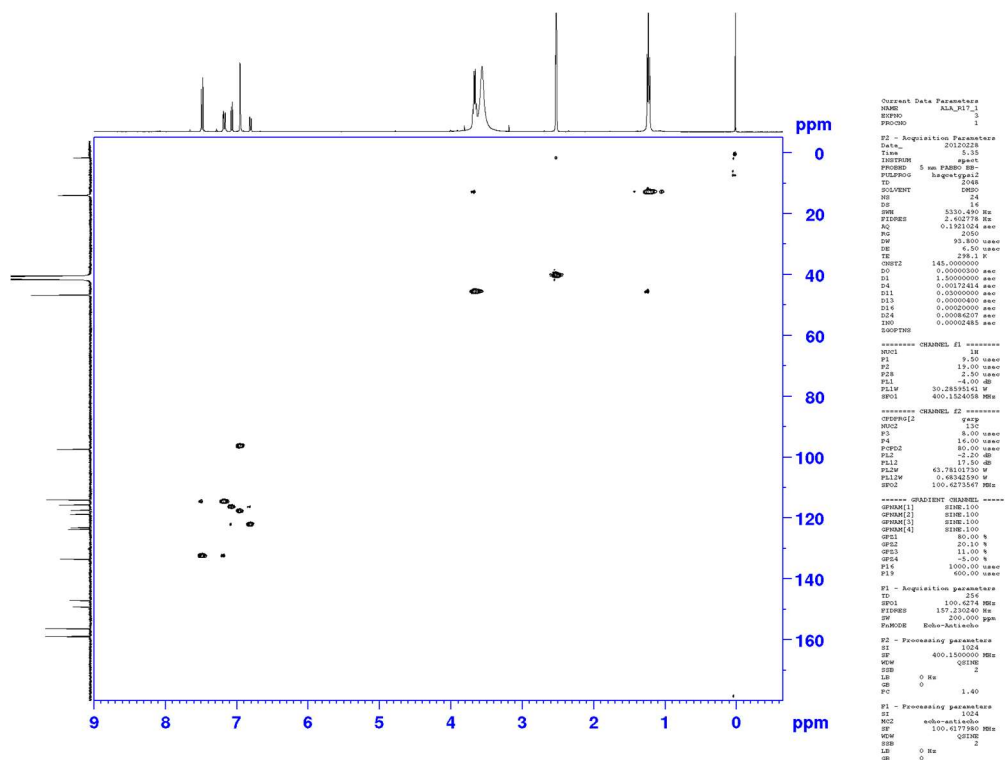


Figure S21. HSQC spectrum of 4.

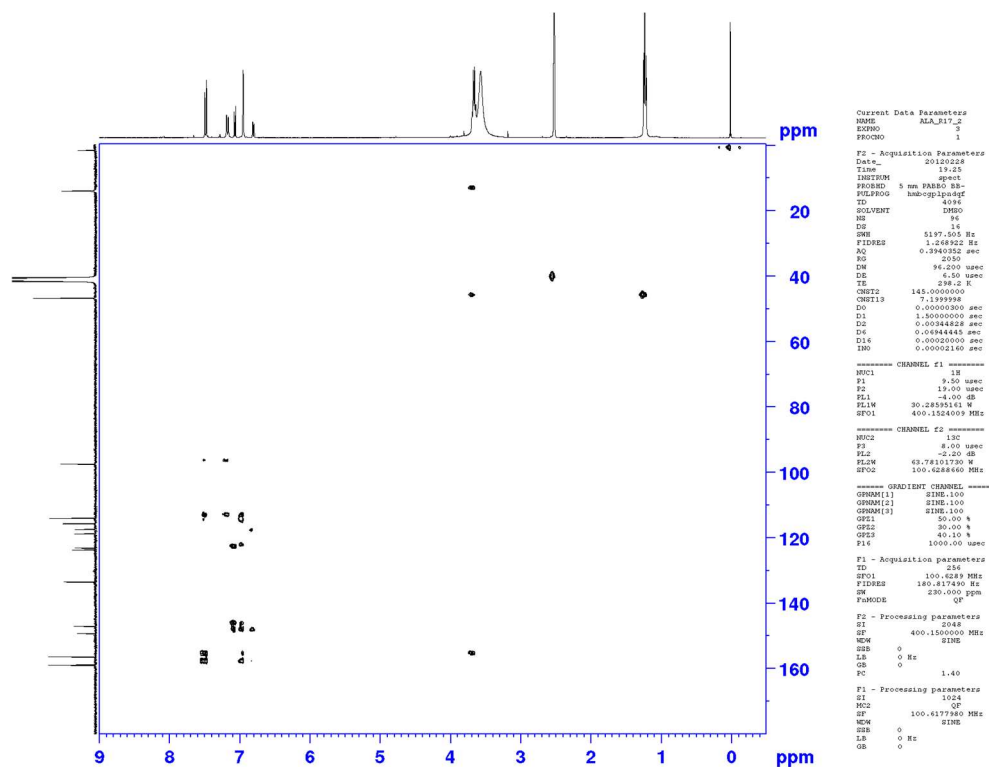


Figure S22. HMBC spectrum of 4.

### MS spectra of 1, 2, 3 and 4

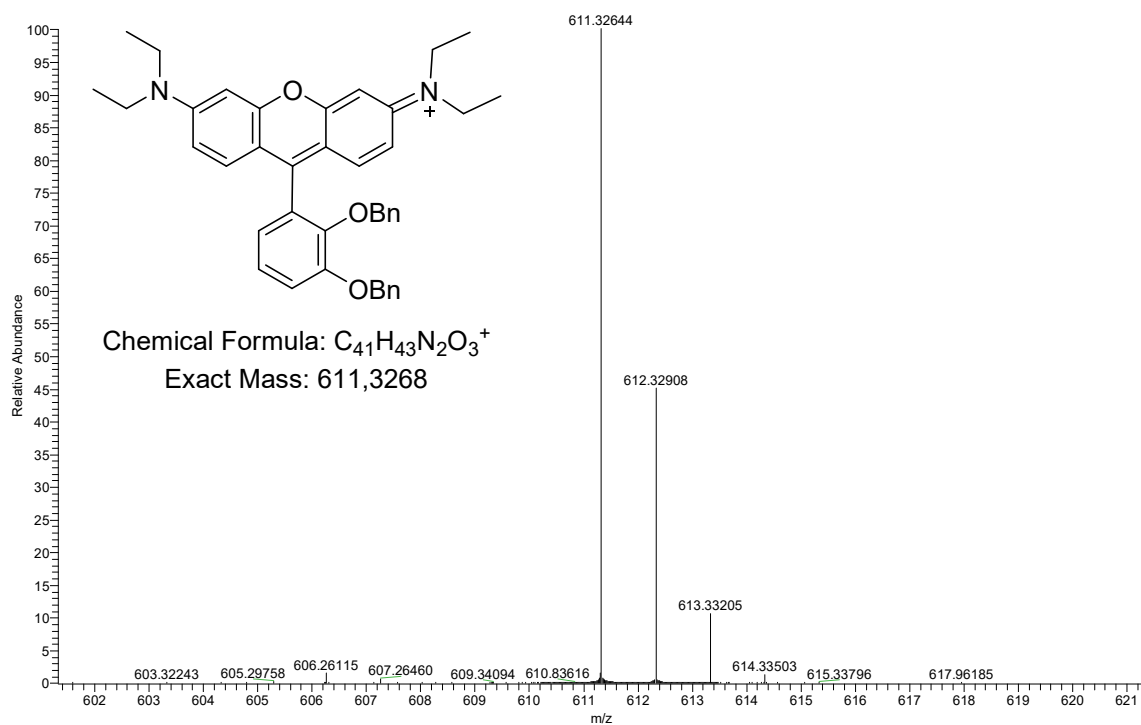
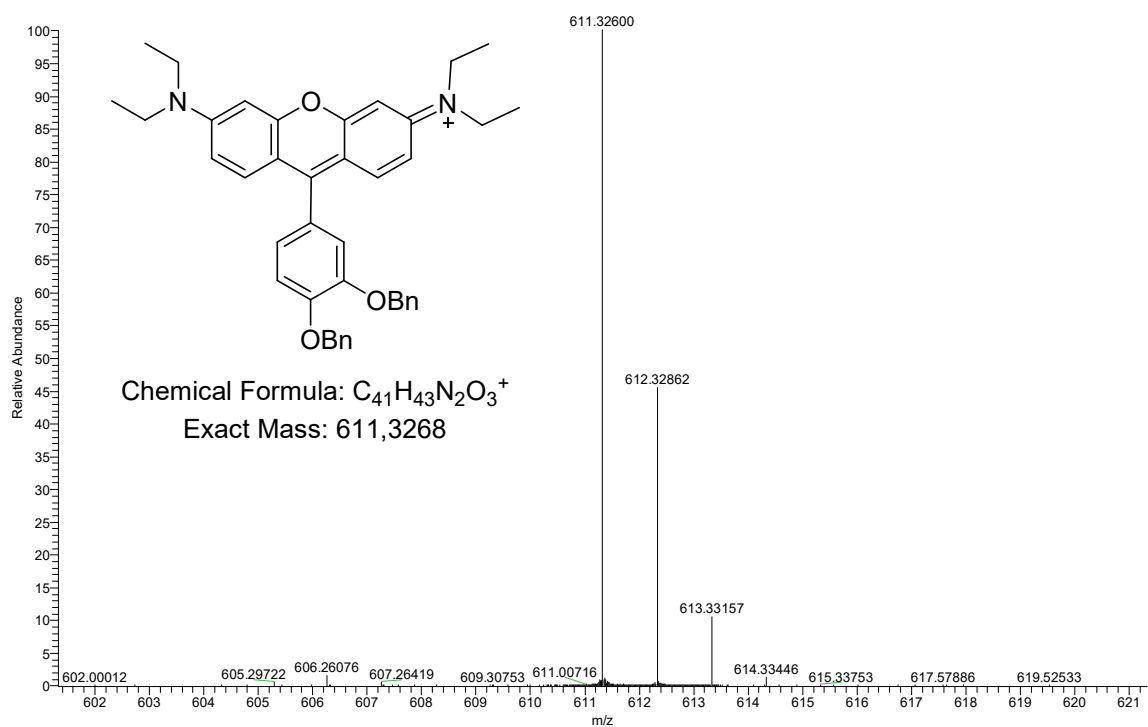
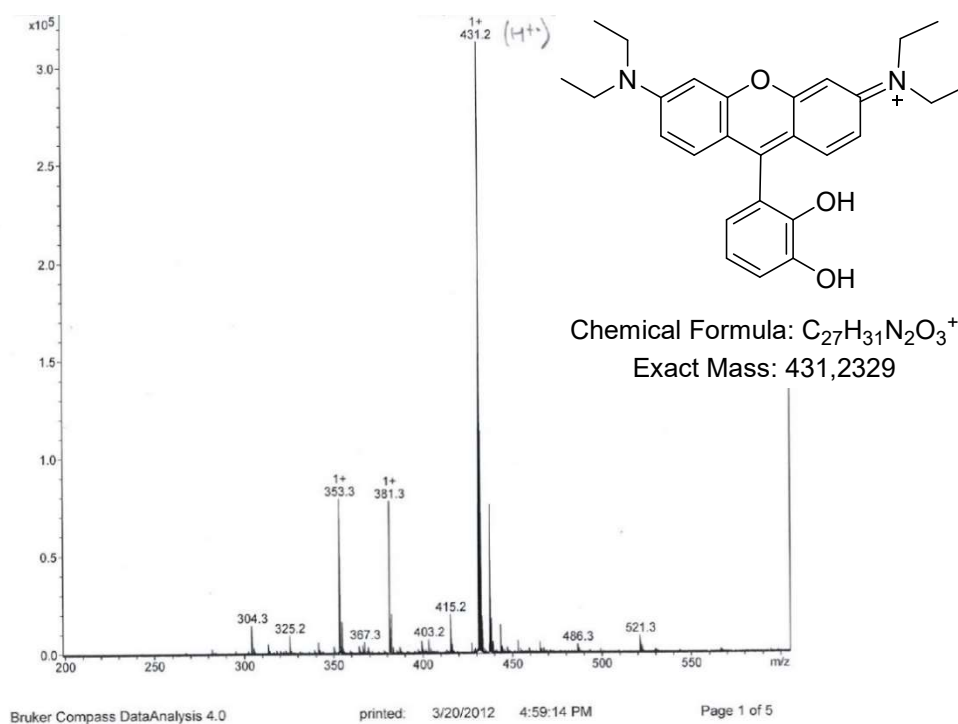


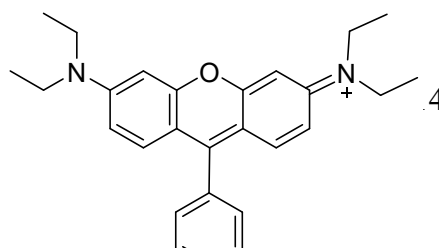
Figure S23. MS spectrum of 1

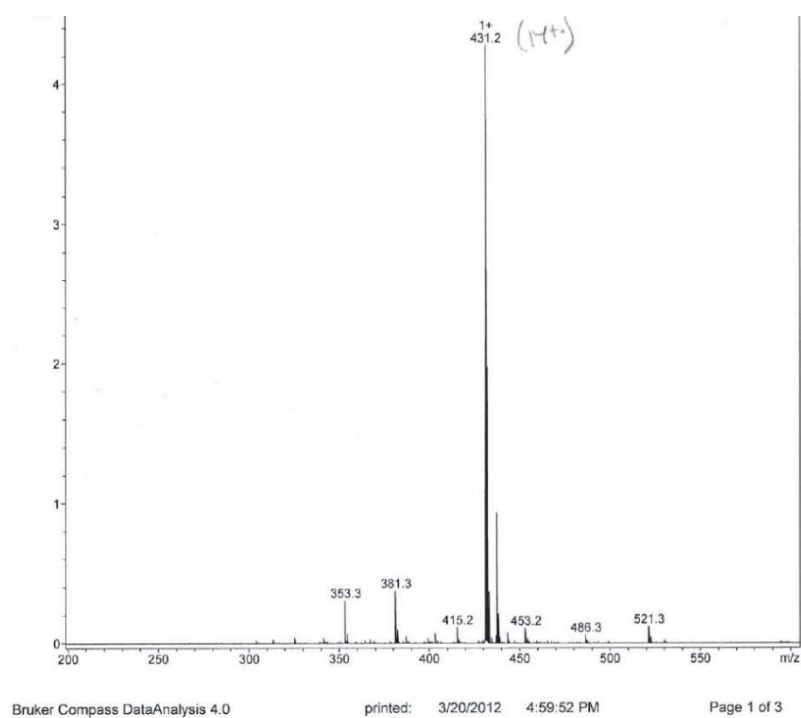


**Figure S24.** MS spectrum of **2**



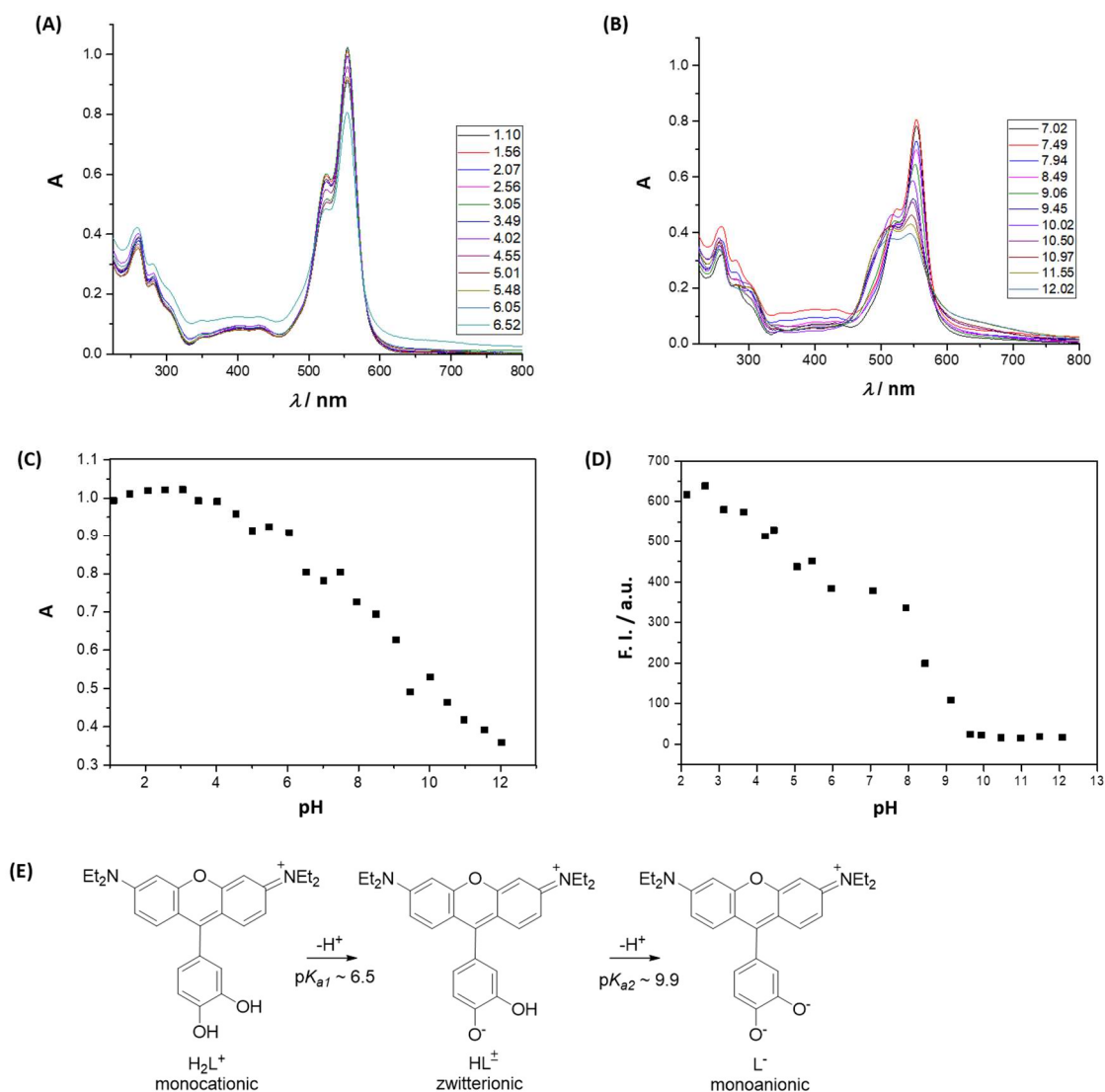
**Figure S25.** MS spectrum of **3**





**Figure S26.** MS spectrum of **4**

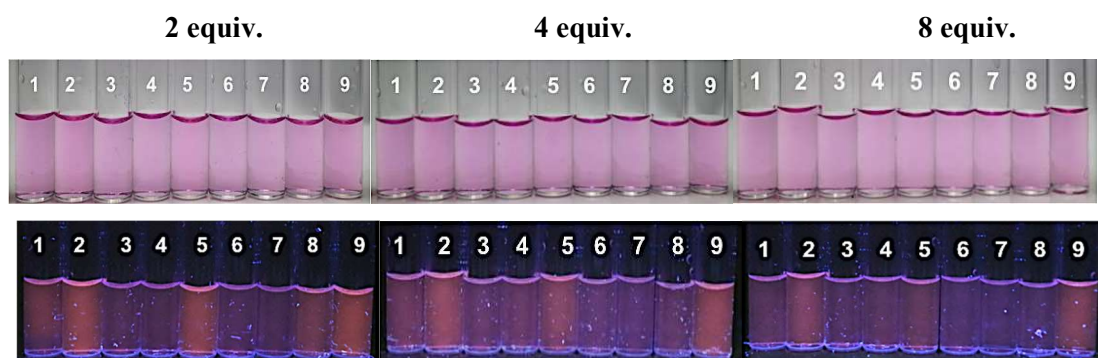
**The influence of pH variation in the fluorescence intensity of rosamine 4.**



**Figure S27.** The influence of pH variation in the fluorescence intensity of rosamine 4. A – Absorption spectra for pH < 7; B – Absorption spectra for pH  $\geq$  7; C – Graphical representation of the pH in function of the maximum in absorbance; D – Graphical representation of the pH as a function of the maximum in fluorescence intensity; E – Structures and  $pK_a$ 's of rosamine 4 deprotonation.

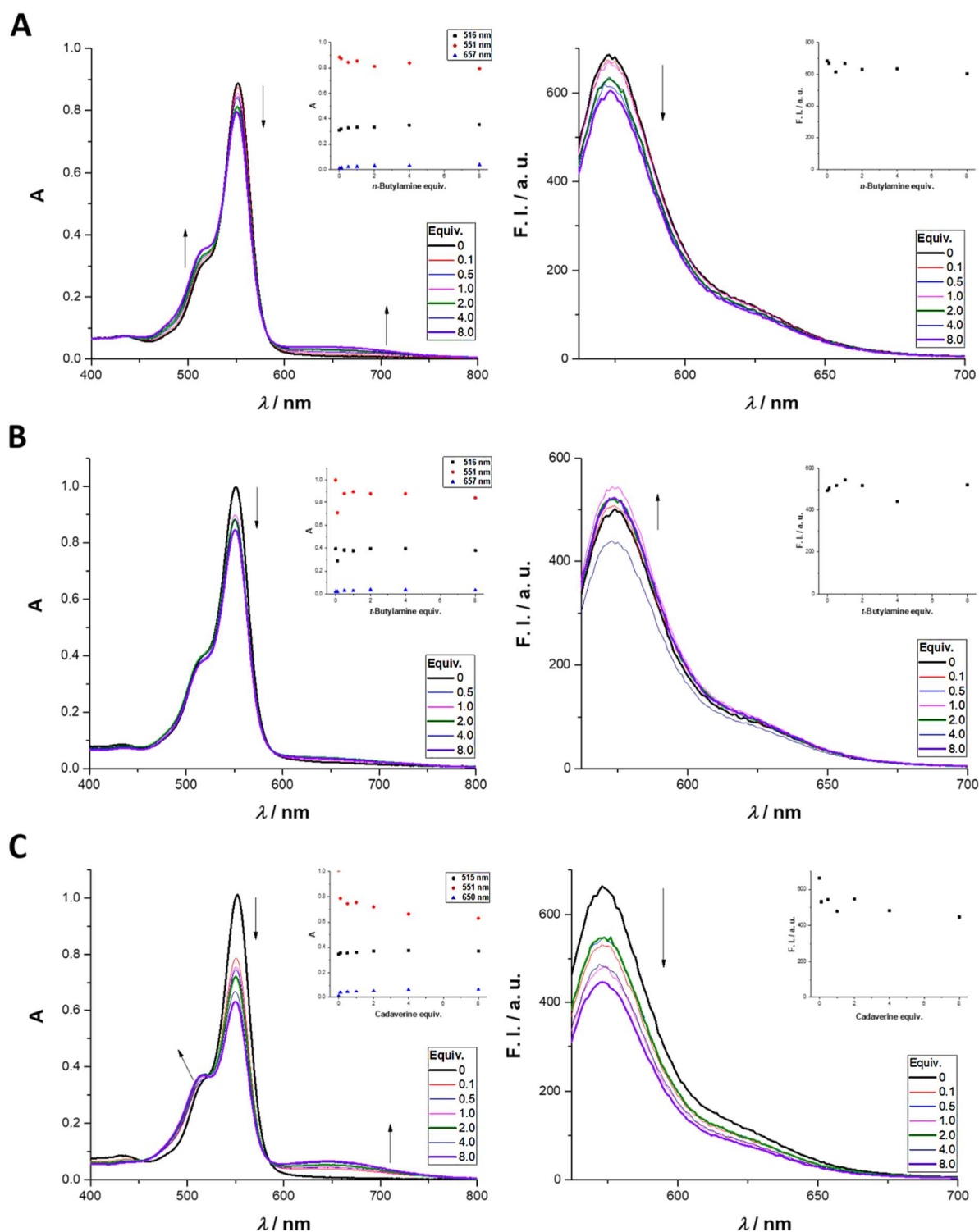
**Detection study of rosamine 4 with different biogenic amines in solution (preliminary study)**



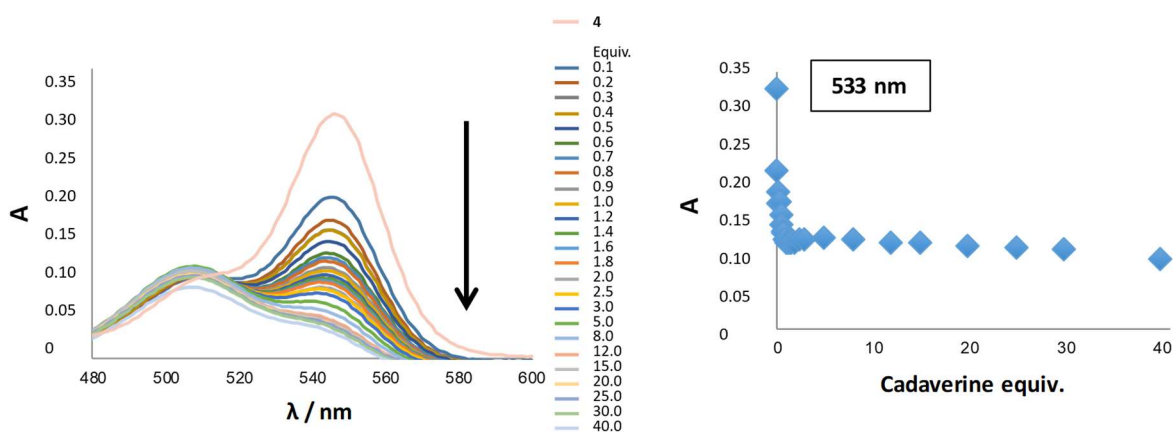


**Figure S28.** Samples with 2, 4 and 8 equiv. of 1-histamine, 2-tyramine, 3-cadaverine, 4-putrescine, 5-phenylethylamine, 6-spermidine, 7-spermine, 8-butylamine, 9- rosamine **4**, observed under visible (top) and UV light (bottom).

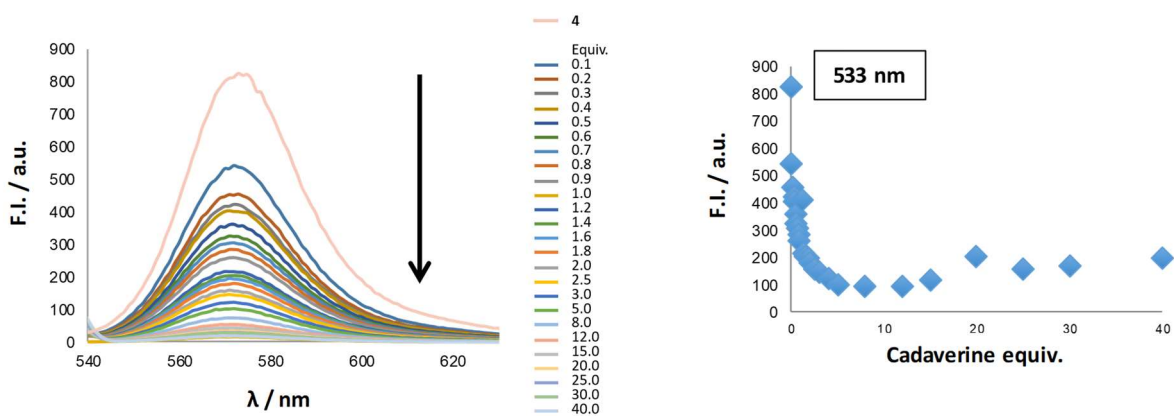
### Detection study of rosamine 4 with different amines in solution (CH<sub>3</sub>CN)



**Figure S29.** Absorption (left) and emission (right) spectra of **4** upon addition (0 – 8 equiv.) of *n*-butylamine (A), *t*-butylamine (B) and cadaverine (C) in CH<sub>3</sub>CN (the inset graphs represent the variation in absorbance (right) and fluorescence intensity (left) for the wavelengths that present more variation in intensity.  $[\mathbf{4}] = 35 \mu\text{M}$  and  $\lambda_{\text{exc}} = 551 \text{ nm}$ ).



**Figure S30.** Absorption (left) spectrum of **4** upon addition (0.1–40 equiv.) of cadaverine in CH<sub>3</sub>CN and graph representing the variation in absorbance (right) intensity at 533 nm [**4**] = 5  $\mu$ M.



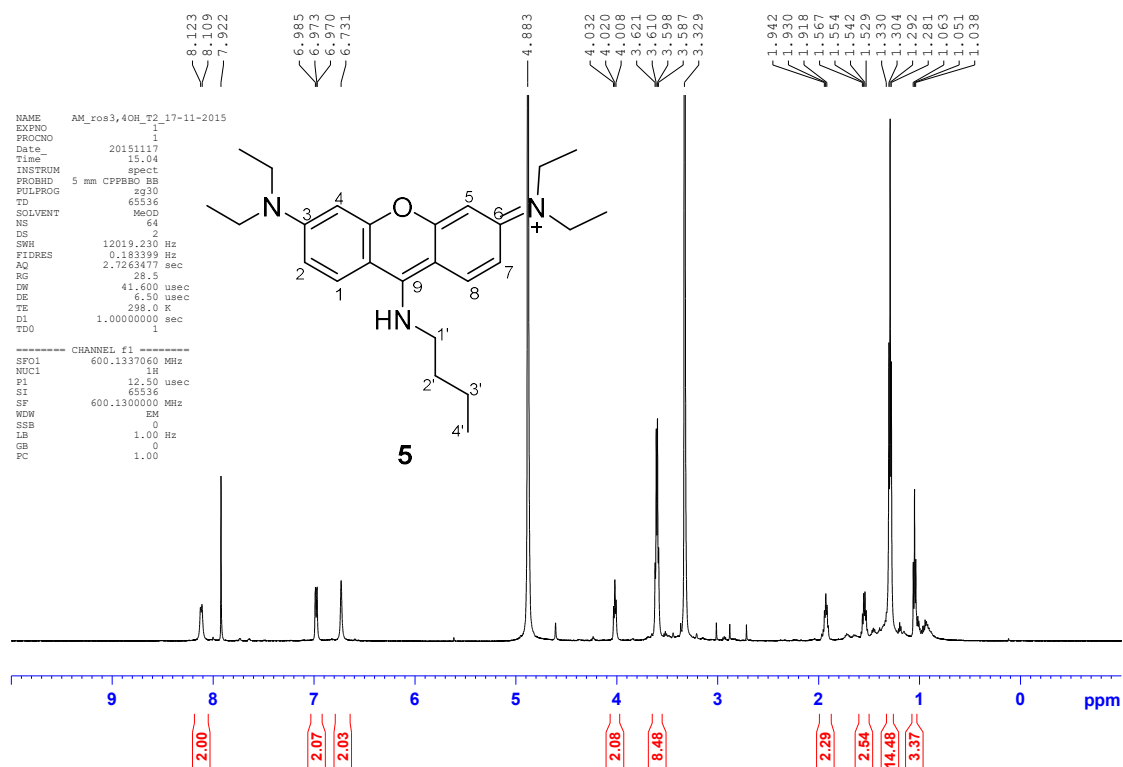
**Figure S31.** Emission (left) spectrum of **4** upon addition (0.1–40 equiv.) of cadaverine in CH<sub>3</sub>CN and graph representing the variation in emission (right) intensity [**4**] = 5  $\mu$ M and  $\lambda_{\text{exc}}$  = 533 nm.

#### Detection study of rosamine **4** with different biogenic amines in gas phase

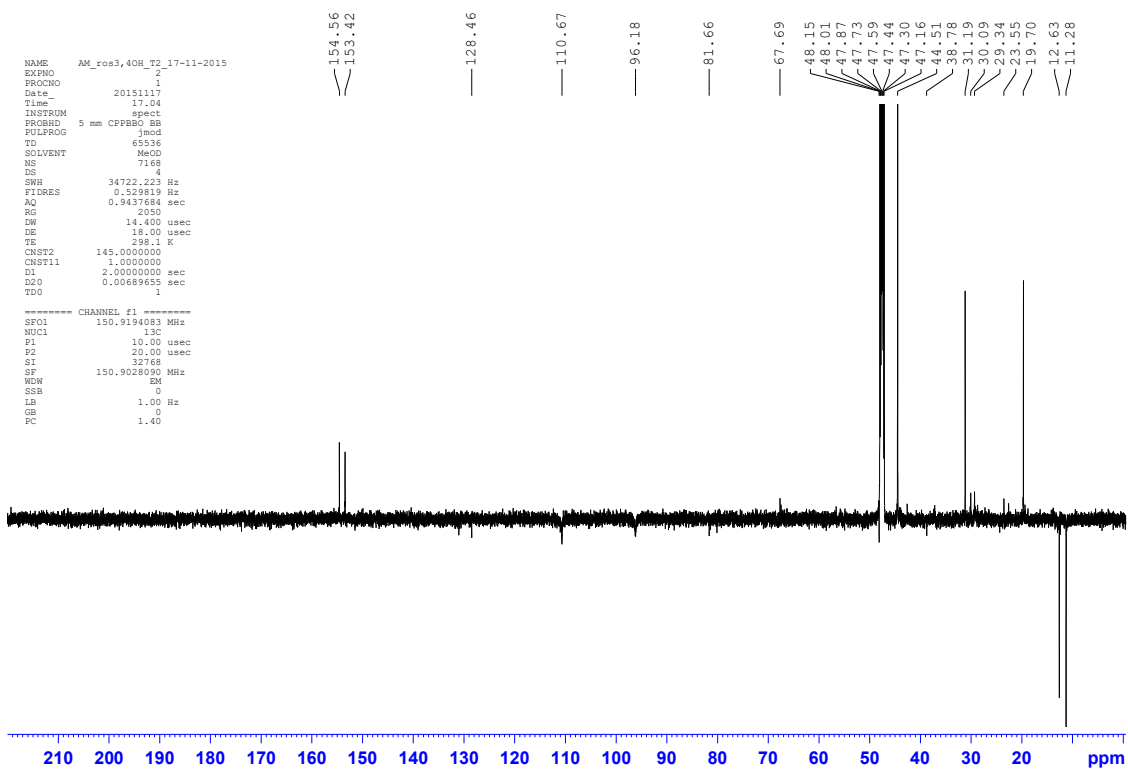


**Figure S32.** Commercial biogenic amine samples subjected to heat

## NMR spectra of aminopyronin 5

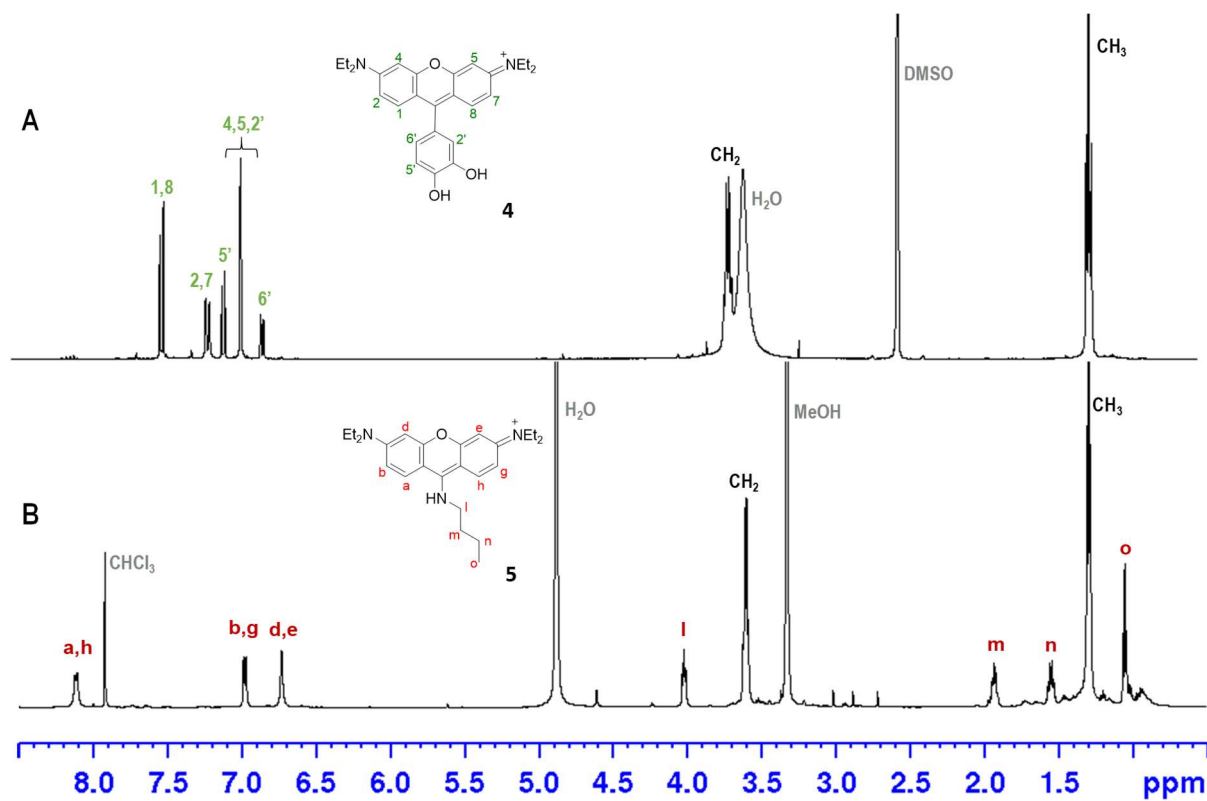


**Figure S33.** <sup>1</sup>H NMR spectrum (600.13 MHz, MeOD-d<sub>4</sub>) of **5**.



**Figure S34.** APT spectrum (150.90 MHz, MeOD-d<sub>4</sub>) of **5**.

**$^1\text{H}$  NMR spectrum comparison between rosamine 4 and aminopyronin 5**



**Figure S35.**  $^1\text{H}$  NMR spectra of **4** (A) and **5** (B).

### NMR spectrum of aminopyronin 6a

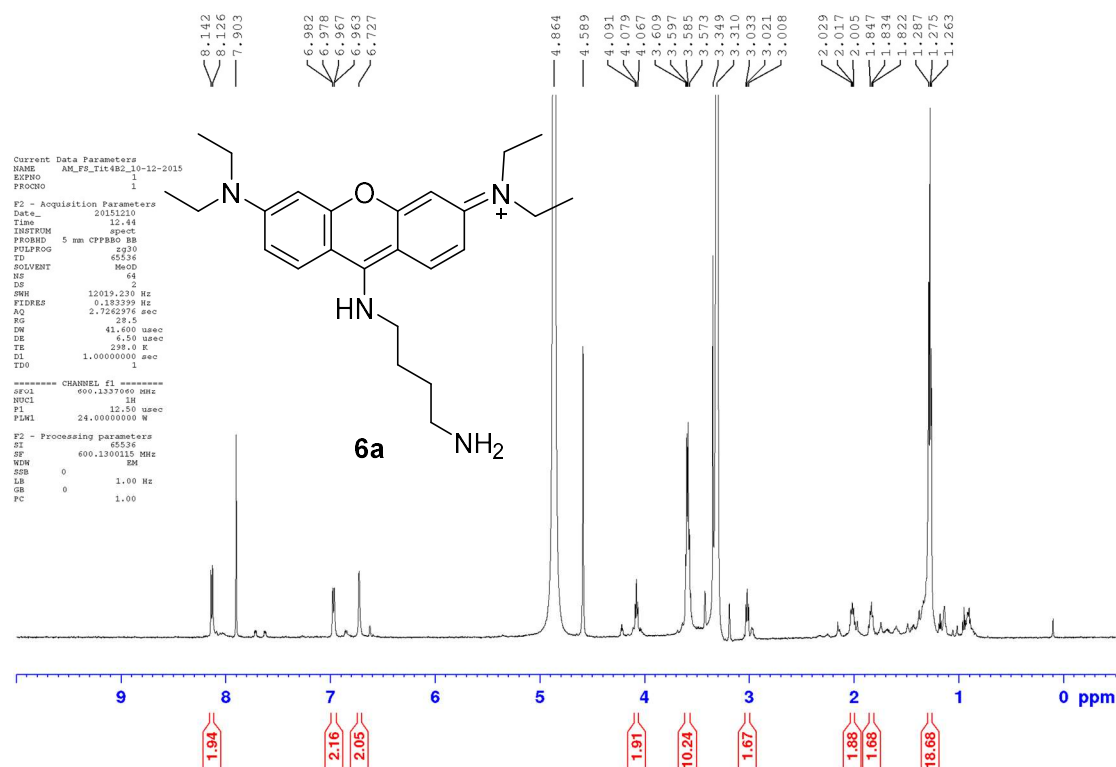


Figure S36. <sup>1</sup>H NMR spectrum (600.13 MHz, MeOD-d<sub>4</sub>) of 6a.

### NMR spectrum of aminopyronin 7a

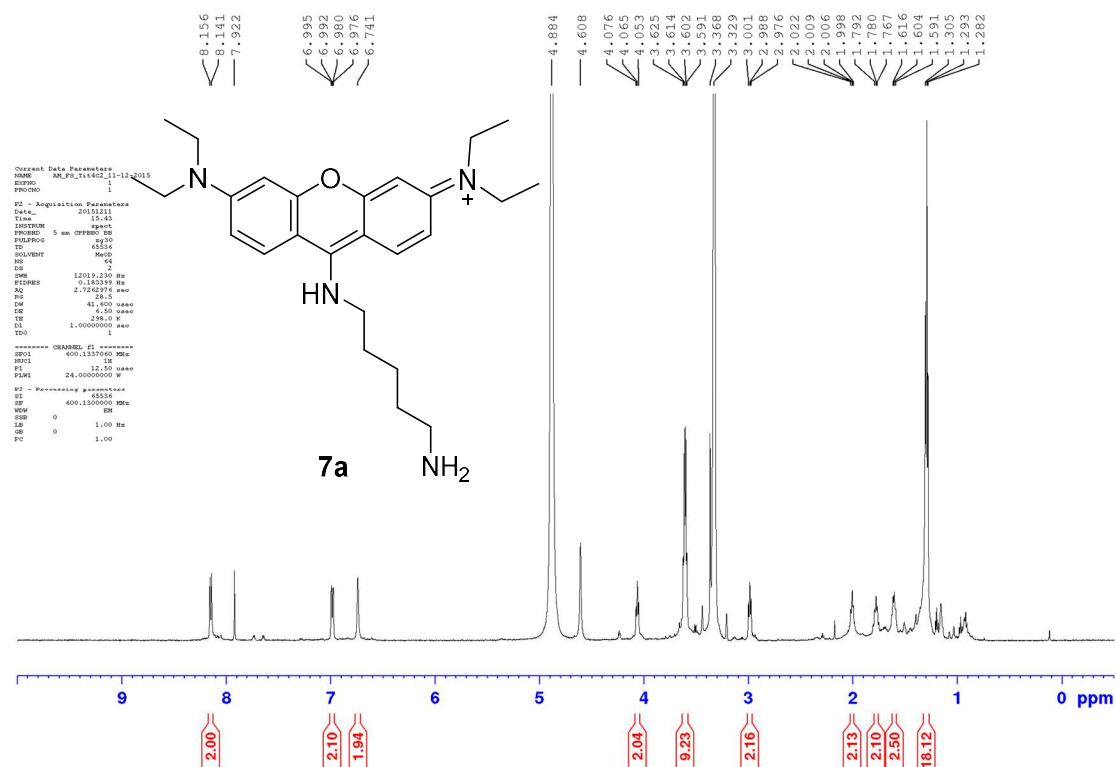


Figure S37. <sup>1</sup>H NMR spectrum (600.13 MHz, MeOD-d<sub>4</sub>) of 7a.

### NMR spectrum of aminopyronin 7b

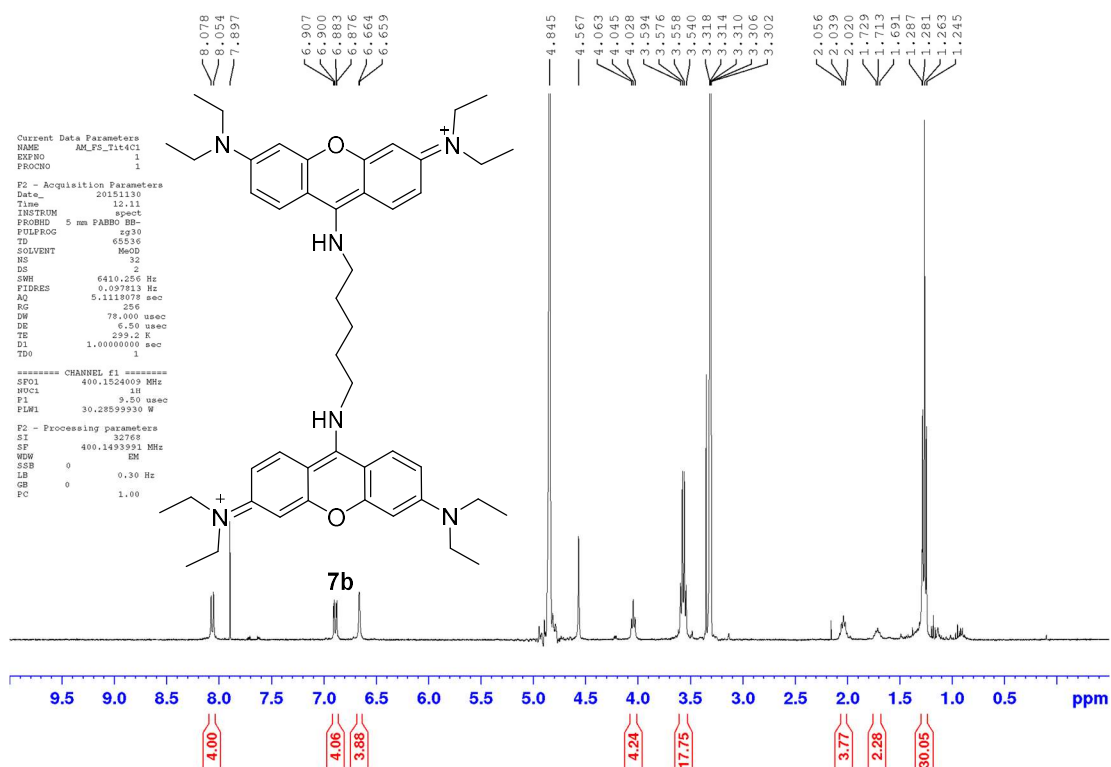


Figure S38.  $^1\text{H}$  NMR spectrum (400.15 MHz,  $\text{MeOD-d}_4$ ) of 7b.

### MS spectra of 5, 6a, 7a and 7b

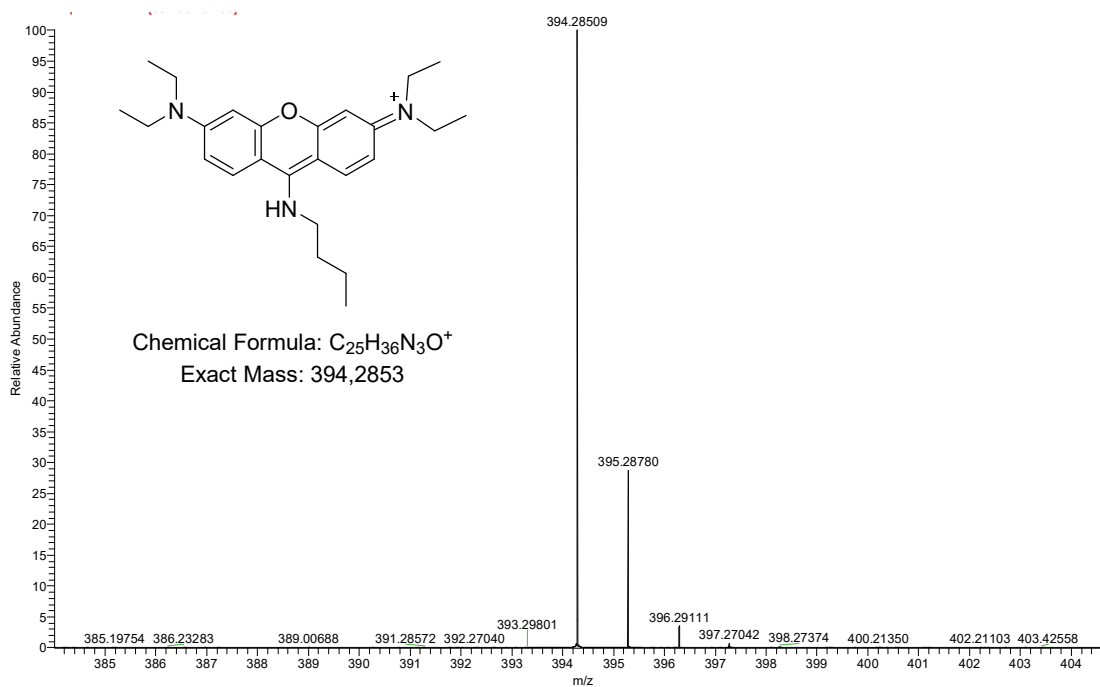
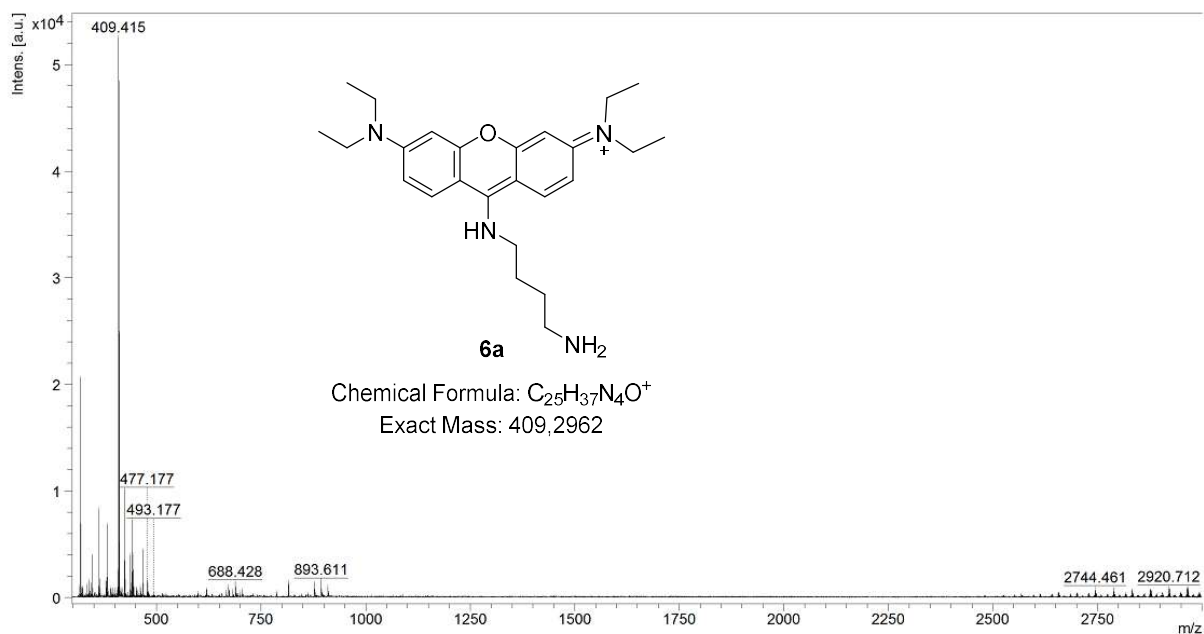
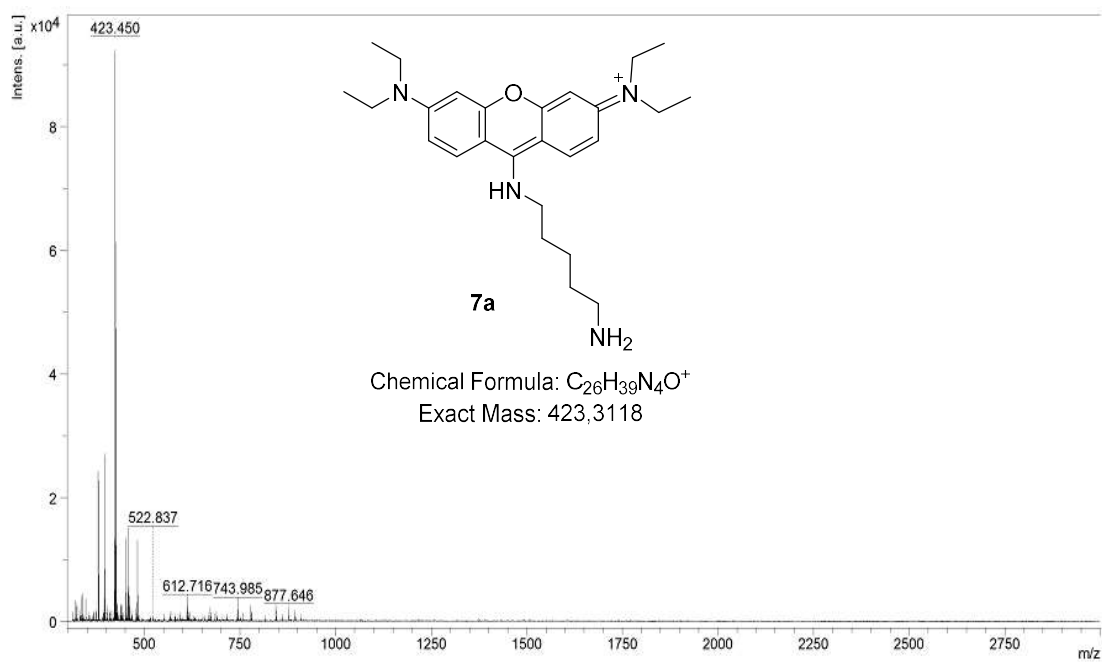


Figure S39. MS spectrum of 5.

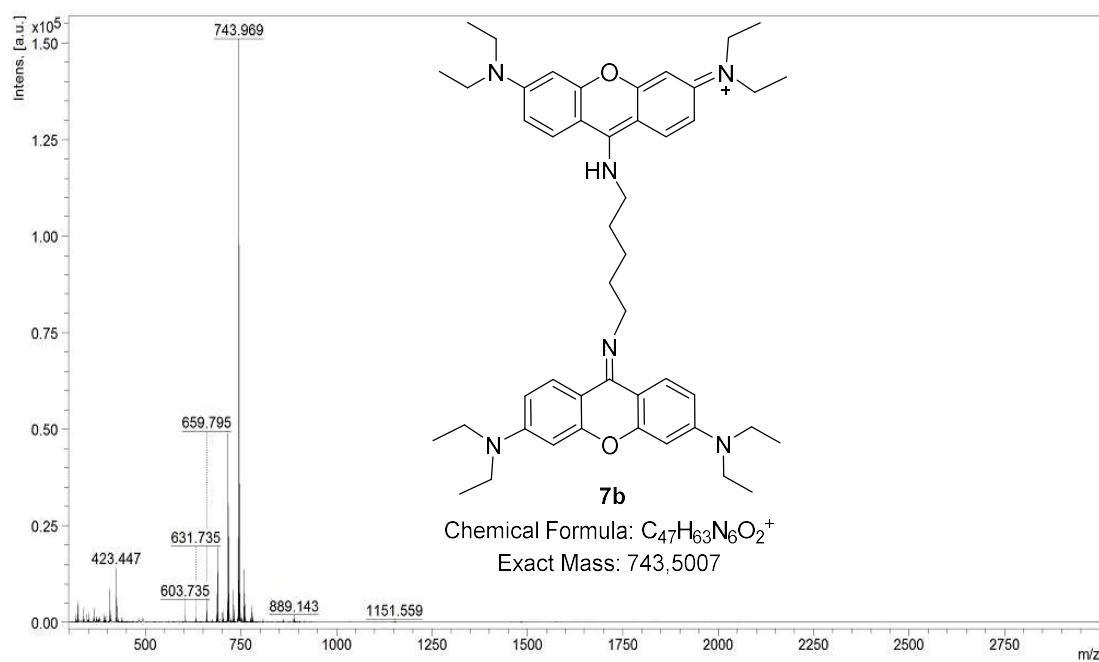


**Figure S40.** MS spectrum of **6a**.



**Figure S41.** MS spectrum of **7a**.





**Figure S42.** MS spectrum of **7b**.