

## Supporting Information

### Rational Construction of a Responsive Azo-Functionalized Porous Organic Framework for CO<sub>2</sub> Sorption

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**Figure S1.** Changes in the adsorption spectra of azobenzene in CH<sub>2</sub>Cl<sub>2</sub> over the time during the irradiation with 365 nm light

**Figure S2.** BET surface area plots

**Table S1.** The BET surface area report of JJU-1

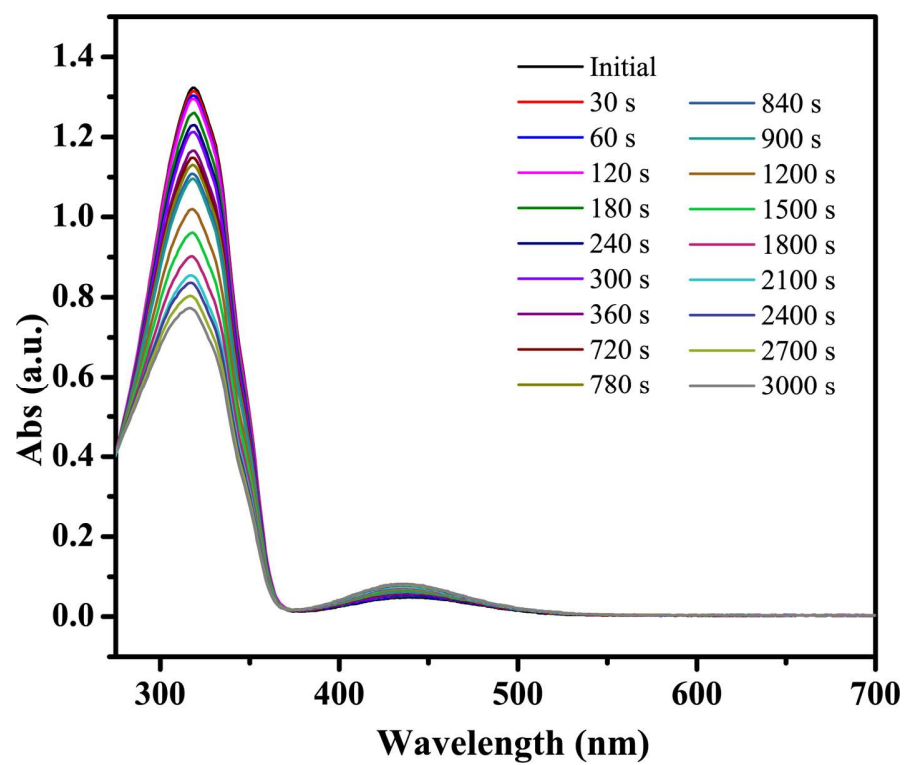
**Figure S3.** The plots of pressures in function of gas uptakes and the parameters (Virial-type expression) for the calculation of heats of adsorption of CO<sub>2</sub> for (a) initial JJU-1 and (b) JJU-1-1st UV

**Figure S4.** The reversible CO<sub>2</sub> adsorption isotherms at 273 K by UV irradiation and thermal treatment of JJU-1

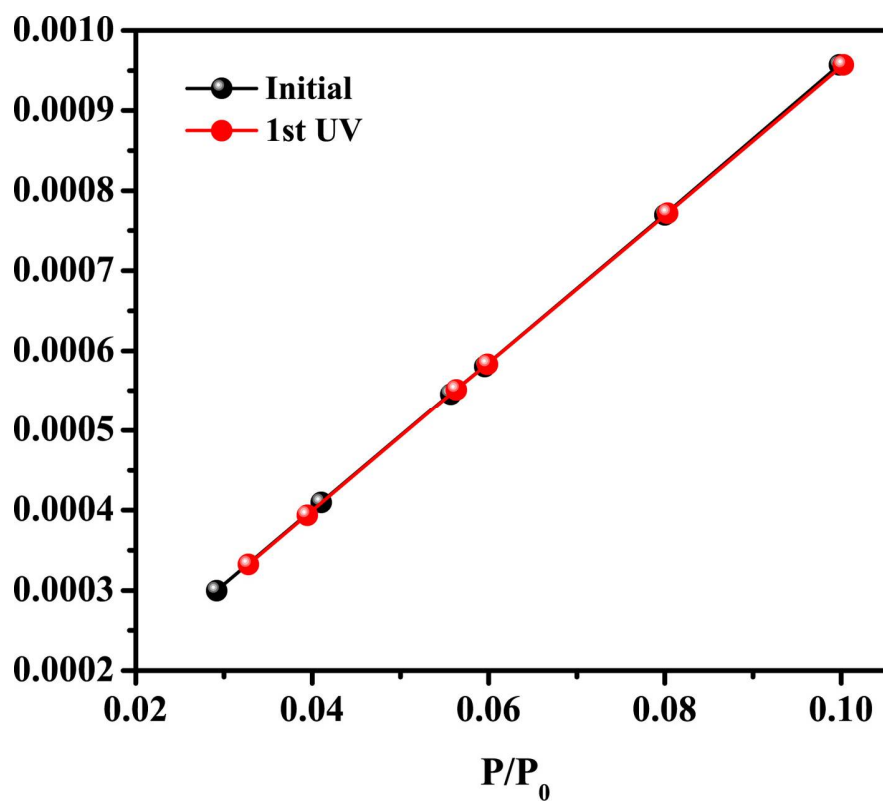
**Figure S5.** The schematic diagram of the structure changes during the reversible *trans/cis* transformation of azo functional group under UV irradiation and thermal regeneration

**Table S2.** BET surface area reports and CO<sub>2</sub> uptakes of some similar POFs

**References**



**Figure S1.** Changes in the adsorption spectra of azobenzene in  $\text{CH}_2\text{Cl}_2$  over the time during the irradiation with 365 nm light.



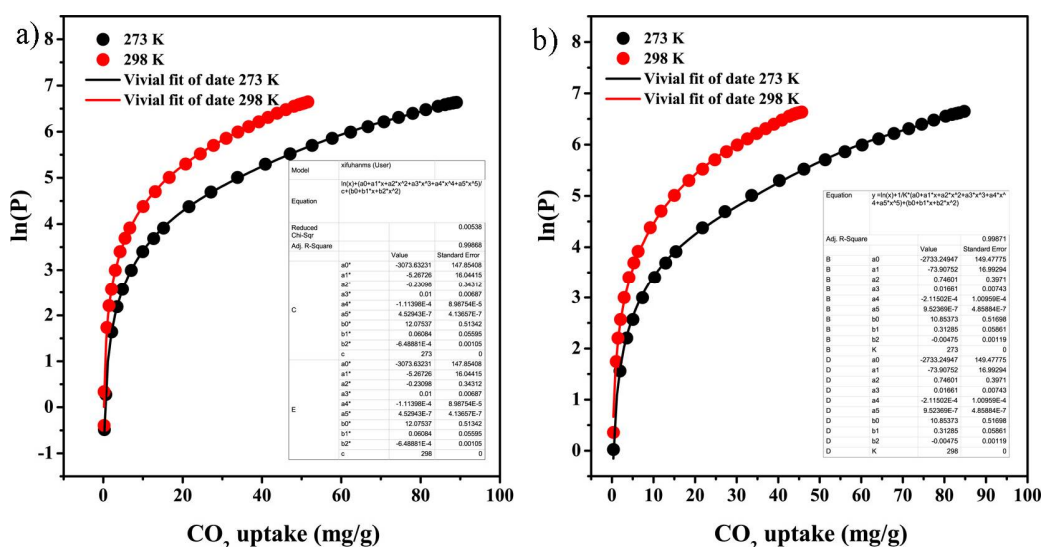
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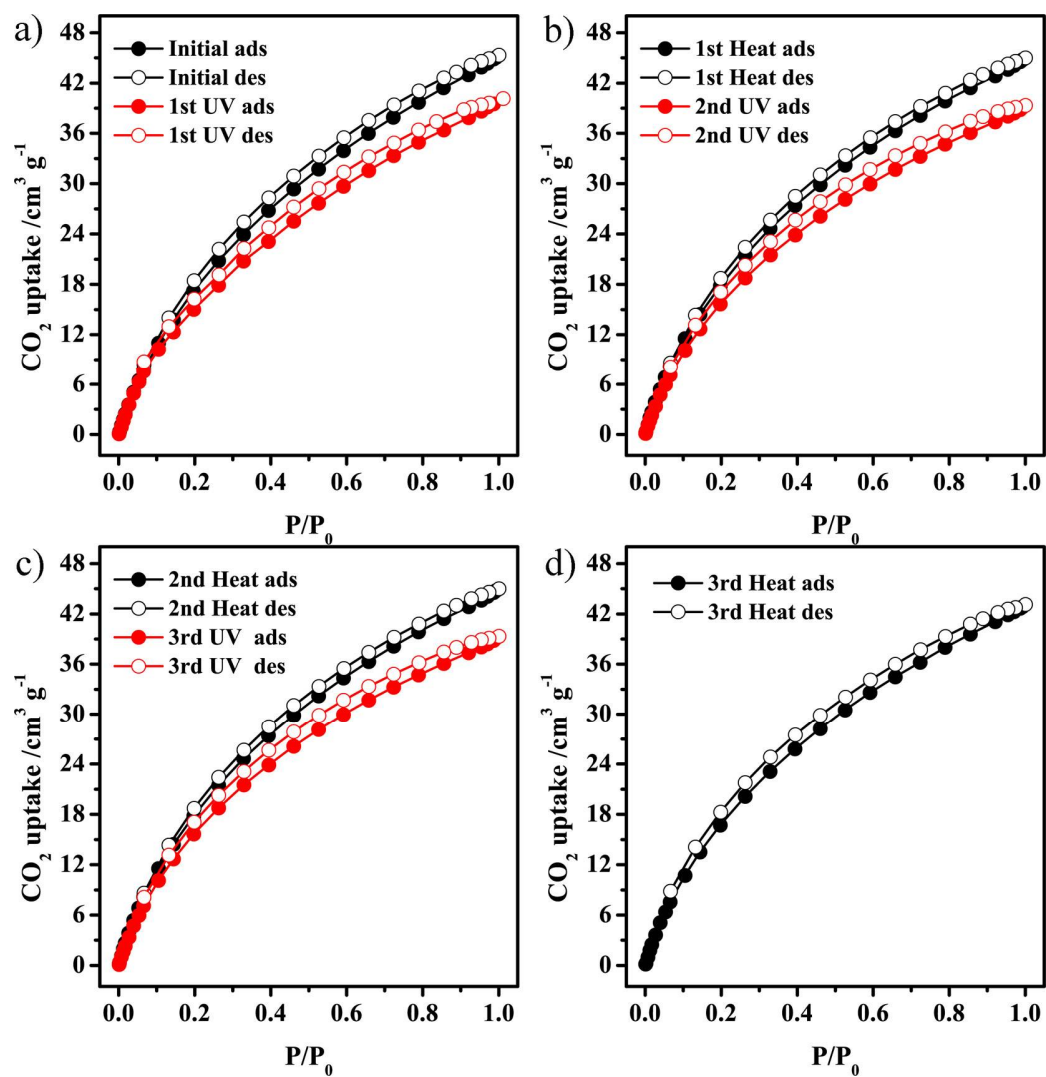
JJU-1	BET [m <sup>2</sup> g <sup>-1</sup> ]	Slope [g cm <sup>-3</sup> ]	Y-intercept [g cm <sup>-3</sup> ]	C	Qm [cm <sup>3</sup> g <sup>-1</sup> ]	Correlation coefficient
Initial	467.09	0.009290	0.000028	327.398390	107.3127	0.9999919
1st UV	468.68	0.009258	0.000029	318.840491	107.6778	0.9999978

## Calculation of isosteric heat of adsorption

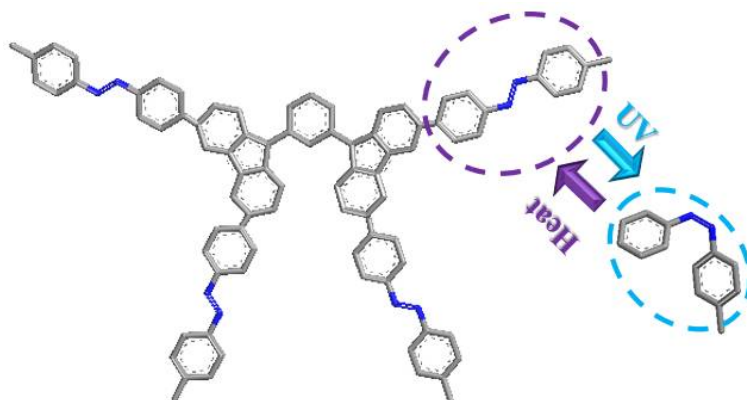
The CO<sub>2</sub> isosteric heat of adsorption of JJU-1 was calculated as a function of the adsorption isotherms at 273 K and 298 K. The data were modelled with a virial-type expression composed of parameters  $a_i$  and  $b_i$  (eq 1), and the heat of adsorption was calculated from the fitting parameters using eq 2, where  $P$  is the pressure,  $N$  is the amount adsorbed,  $T$  is the temperature,  $R$  is the universal gas constant, and  $m$  and  $n$  determine the number of terms required to adequately describe the isotherm.  $a_i$  and  $b_i$  are virial coefficients.



**Figure S3.** The plots of pressures in function of gas uptakes and the parameters (Virial-type expression) for the calculation of heats of adsorption of CO<sub>2</sub> for (a) initial JJU-1 and (b) JJU-1-1st UV.



**Figure S4.** The reversible CO<sub>2</sub> adsorption isotherms at 273 K by UV irradiation and thermal treatment of JJU-1.



**Figure S5.** The schematic diagram of the structure changes during the reversible *trans/cis* transformation of azo functional group under UV irradiation and thermal regeneration.

**Table S2.** BET surface area reports and CO<sub>2</sub> uptakes of some similar POFs.

<b>Materials</b>	<b>BET [m<sup>2</sup>/g]</b>	<b>V<sub>total</sub> [cm<sup>3</sup>/g]</b>	<b>CO<sub>2</sub> uptake [cm<sup>3</sup>/g]</b>	<b>Q<sub>st</sub> [kJ/mol]</b>	<b>Reference</b>
<b>Azo-MOP-1</b>	456	0.48	49.6	--	1
<b>PAF-36</b>	325	0.25	28.6	27.0	2
<b>PAF-36-UV</b>	385	0.27	31.2	28.4	2
<b>PAF-37</b>	443	0.27	26.3	36.8	2
<b>PAF-37-UV</b>	456	0.28	29.6	40.7	2
<b>TAP-1</b>	474	0.74	48.4	35.6	3
<b>UCBZ-1</b>	980	1.46	39.0	--	4
<b>UCBZ-1-UV</b>	--	--	50.0	--	4
<b>azo-COP-1</b>	608	0.39	54.7	29.3	5
<b>azo-COP-2</b>	703	0.42	56.2	24.8	5
<b>Azo-POF-1</b>	712	--	66.7	27.5	6
<b>Azo-POF-2</b>	439	--	43.0	26.6	6
<b>ALP-4</b>	862	0.50	41.2	28.2	7
<b>POF-Initial</b>	571	0.49	46.2	26.7	8
<b>POF-1st UV</b>	549	0.48	41.6	29.7	8
<b>JJU-1-Initial</b>	467	0.31	45.3	27.1	This work
<b>JJU-1st UV</b>	469	0.34	40.1	33.1	This work

## References

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