

Supporting Info

Synthesis and biological assessment of 4,1-benzothiazepines with neuroprotective activity on the Ca²⁺ overload for the treatment of neurodegenerative diseases and stroke

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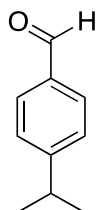
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General procedure for the synthesis of isopropylbenzaldehydes

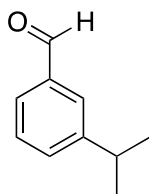
Similar to what has been described, (Anajota, 2015) with slight modifications. To a solution of the corresponding bromoisopropylbenzene (1 equiv) in dry THF (15 mL) at $-78\text{ }^{\circ}\text{C}$ under argon, *n*-butyllithium (2.5 M in hexanes, 1.1 equiv) was added dropwise and the resulting mixture was stirred at $-78\text{ }^{\circ}\text{C}$ for 30 min. Then, *N,N*-dimethylformamide (1.1 equiv) was added and the mixture stirred at $-78\text{ }^{\circ}\text{C}$ for 30 min more, then allowed to warm it up $-20\text{ }^{\circ}\text{C}$ and stirred until the disappearance of reagents monitored by TLC (2–5 h). Reaction was stopped with H_2O (30 mL) and extracted with diethyl ether ($3 \times 50\text{ mL}$). The combined organic layer was dried over anhydrous MgSO_4 , filtered, and the solvent removed by vacuum obtaining a colorless oil that did not need further purification.

- *p*-isopropylbenzaldehyde



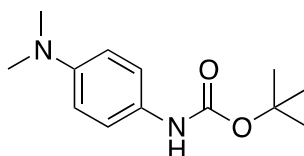
Following the *General procedure for the synthesis of isopropylbenzaldehydes*, reaction of 1-bromo-4-isopropylbenzene (2.51 mmol, 500 mg) [*n*-BuLi (2.76 mmol) and *N,N*-dimethylformamide (2.76 mmol, 202 mg, 214 μL)] yielded *p*-isopropylbenzaldehyde as a yellow oil (372 mg, quantitative yield) with spectral data according to literature [1].

- *meta*-isopropylbenzaldehyde



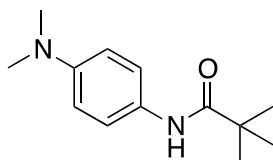
Following the *General procedure for the synthesis of isopropylbenzaldehydes*, reaction of 1-bromo-3-isopropylbenzene (3.07 mmol, 611 mg) [*n*-BuLi (3.38 mmol) and *N,N*-dimethylformamide (3.38 mmol, 247 mg, 261 μ L)] yielded *m*-isopropylbenzaldehyde as a yellow oil (453 mg, quantitative yield) with spectral data according to literature [1].

Synthesis of 4-dimethylamino-*N*-tert-butoxycarbonylaniline (19).



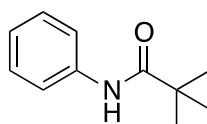
Similar to what we have described for other Boc-protected anilines, (Anajota, 2015) with slight modifications. *N,N*-dimethyl-*p*-phenylenediamine (1 g, 7.34 mmol) and tert-butyl dicarbonate (1.76 g, 8.08 mmol) were dissolved in dry tetrahydrofuran (THF) under Ar, and triethylamine (810 mg, 1.12 mL, 8.08 mmol) was injected slowly. The reaction mixture was stirred at room temperature for 2 h, after then solvent was evaporated. The crude was dissolved in ethyl acetate (30 mL) and washed with saturated NaHCO₃ (3 \times 10 mL) and brine (3 \times 10 mL). The combined organic layer was dried over Na₂SO₄, filtered and evaporated, obtaining a solid in quantitative yield with spectral data according to the literature [2].

Synthesis of *N*-[4-(dimethylamino)phenyl]-2,2-dimethylpropanamide (22).



N,N-dimethyl-*p*-phenylenediamine (1 g, 7.34 mmol) was dissolved in dry CH₂Cl₂ (11 mL), and triethylamine (1.28 mL, 928 mg, 9.175 mmol) was slowly added under Ar. The mixture was cooled down to 0 °C and pivaloyl chloride (0.99 mL, 973 mg, 8.07 mmol) was injected dropwise, allowing the reaction to stir at 0 °C for 15 min. Then, the reaction mixture was warmed up to rt and stirred for 3 h, monitoring reaction evolution by TLC. Reaction was terminated by addition of water (10 mL) and washed with saturated NaHCO₃ (3 × 10 mL). Combined organic layer was dried using anhydrous Na₂SO₄, filtered, and the solvent evaporated under vacuum. Compound **22** (1.53 g, 95%) was obtained as an intense purple-colored solid. ¹H NMR (300 MHz, CDCl₃) δ 7.35 (d, *J* = 9.0 Hz, 2H, H₂, H₆), 6.69 (d, *J* = 9.0 Hz, 2H, H₃, H₅), 7.20 (bs, 1H, NH), 2.90 [s, 6H, N(CH₃)₂], 1.29 [s, 9H, C(CH₃)₃].

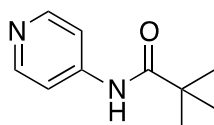
Synthesis of *N*-phenyl-2,2-dimethylpropanamide (**29**).



Aniline (511 mg, 5.49 mmol) was dissolved in dry CH₂Cl₂ (9 mL), and triethylamine (0.96 mL, 694 mg, 6.86 mmol) was slowly added under Ar. The mixture was cooled down to 0 °C and pivaloyl chloride (0.67 mL, 662 mg, 5.49 mmol) was injected dropwise, allowing the reaction to stir at 0 °C for 15 min. Then, the reaction mixture was warmed up to rt and stirred for 3 h, monitoring reaction evolution by TLC. Reaction was terminated by addition of water (10 mL) and washed with saturated NaHCO₃ (3 × 10 mL). Combined organic layer was dried using anhydrous Na₂SO₄,

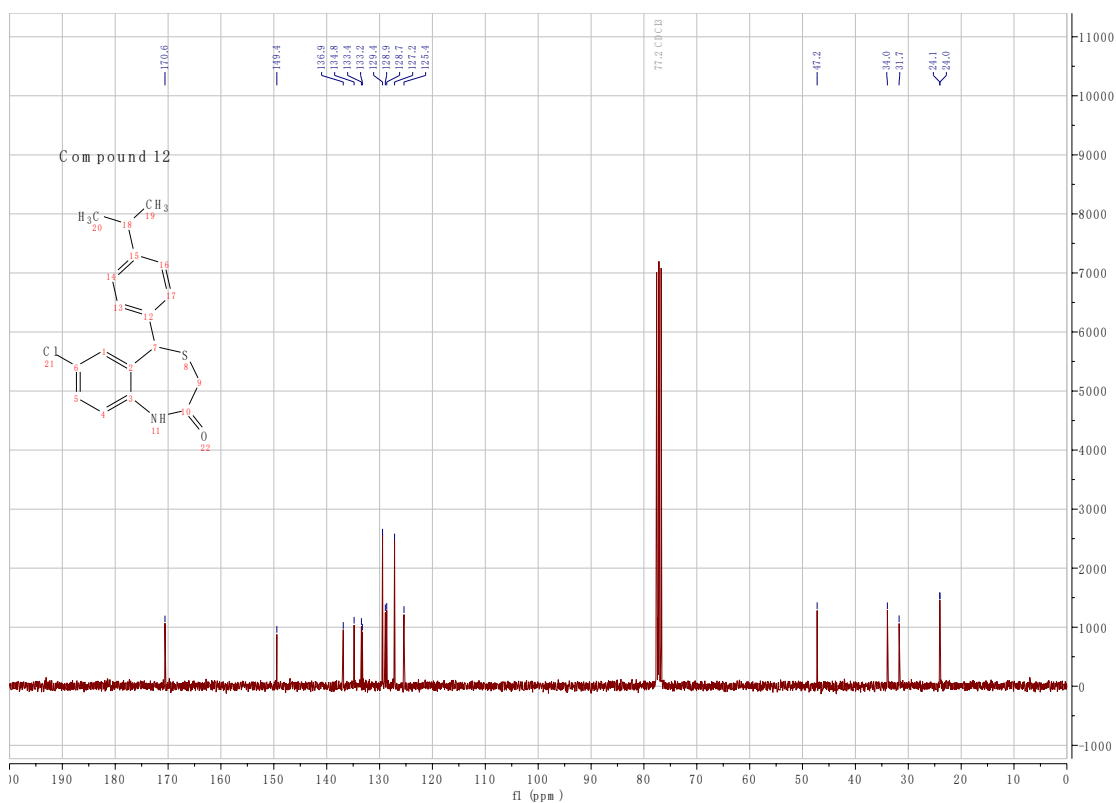
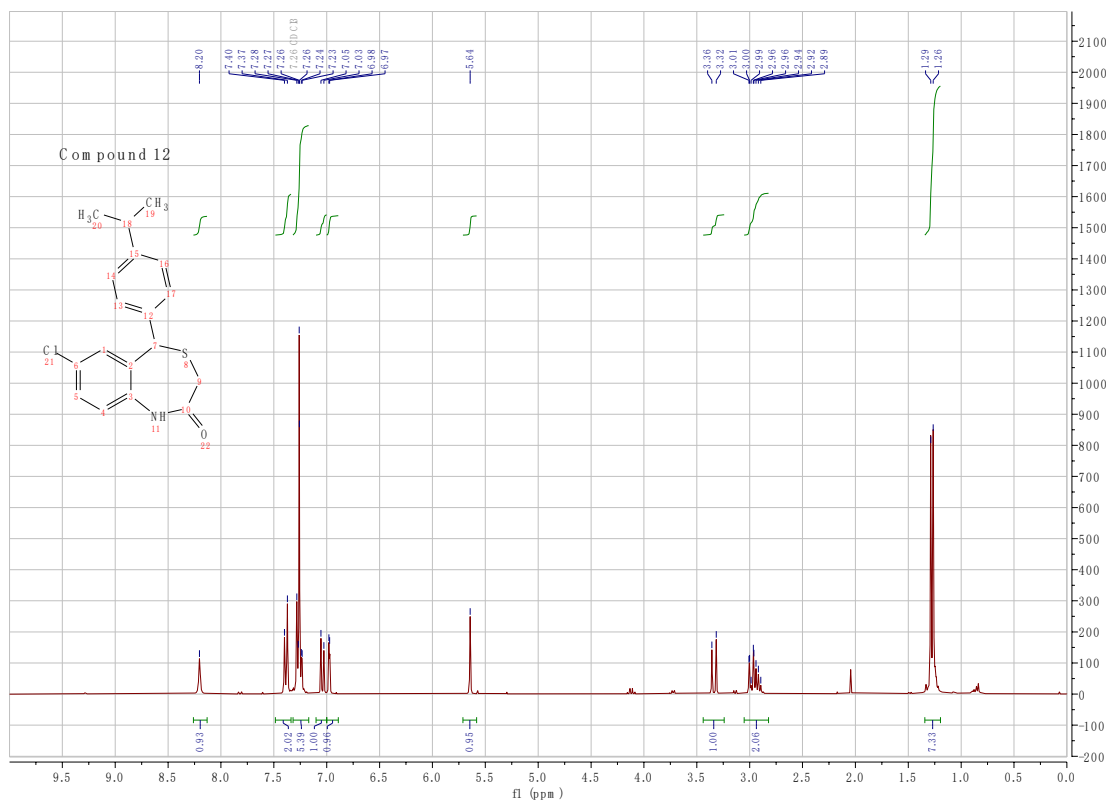
filtered, and the solvent evaporated under vacuum. Compound **29** (974 mg, > 99%) was obtained as a yellow solid, possessing spectral data according with the literature [3].

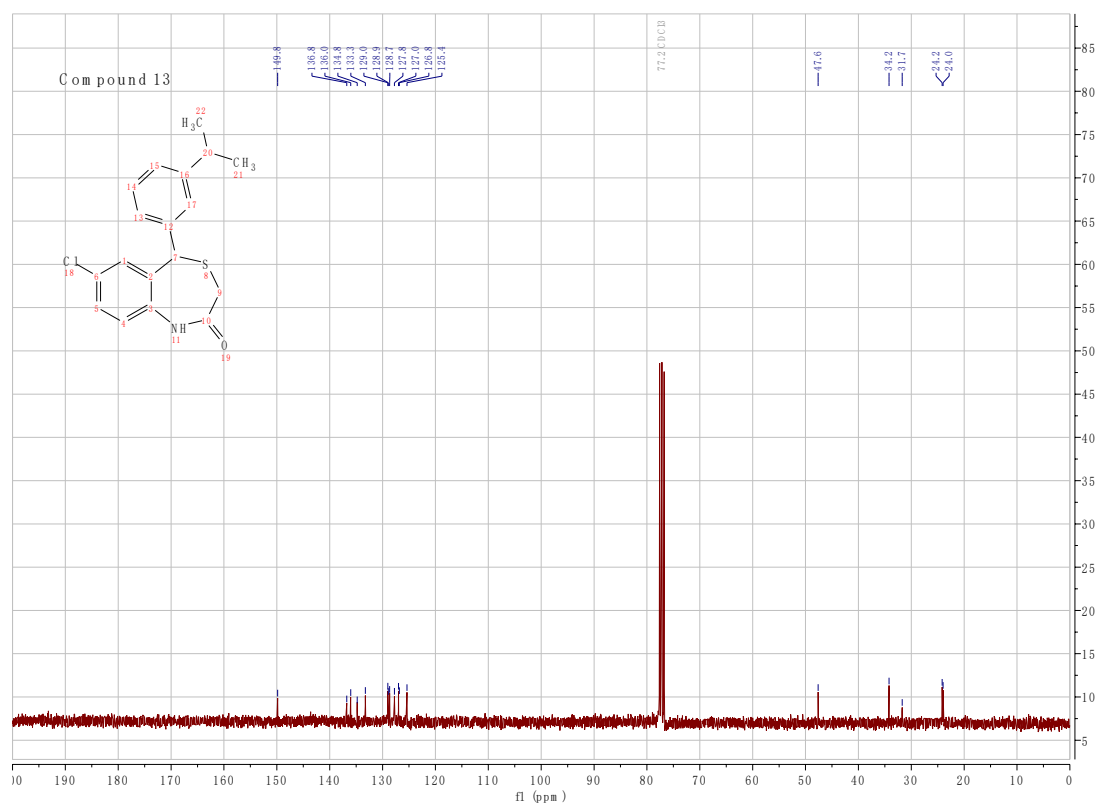
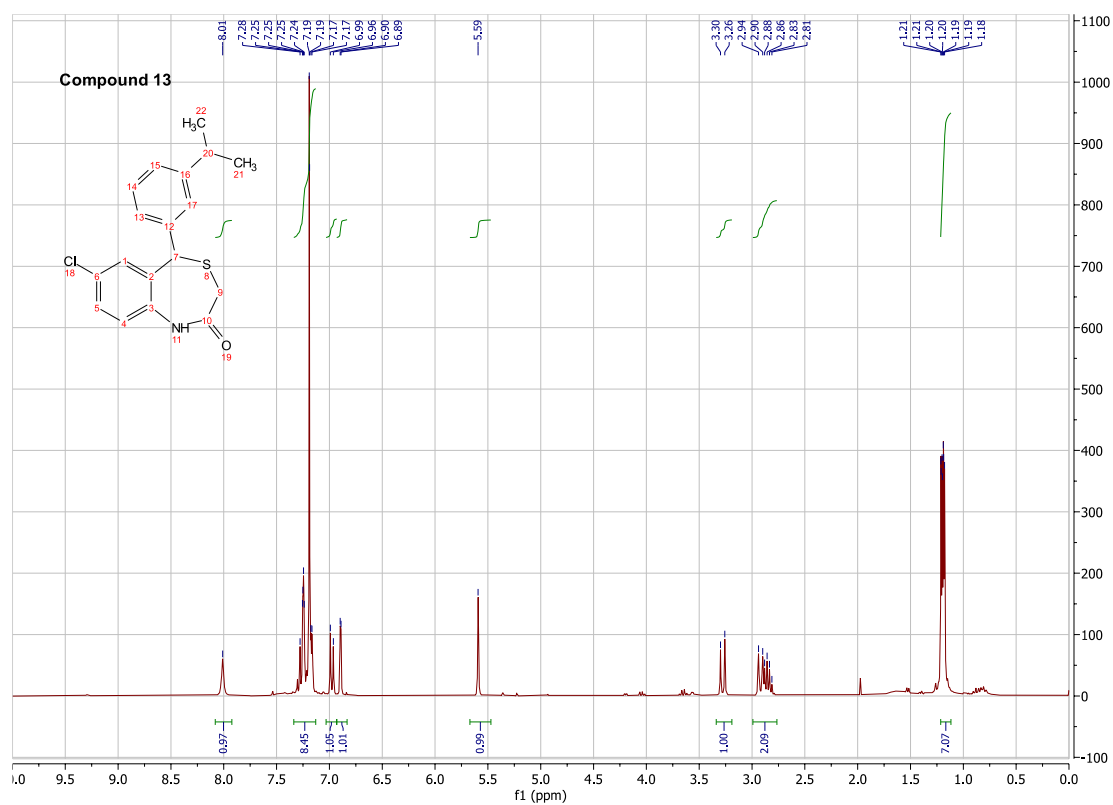
Synthesis of 2,2-dimethyl-*N*-(4-pyridinyl)propanamide (34).

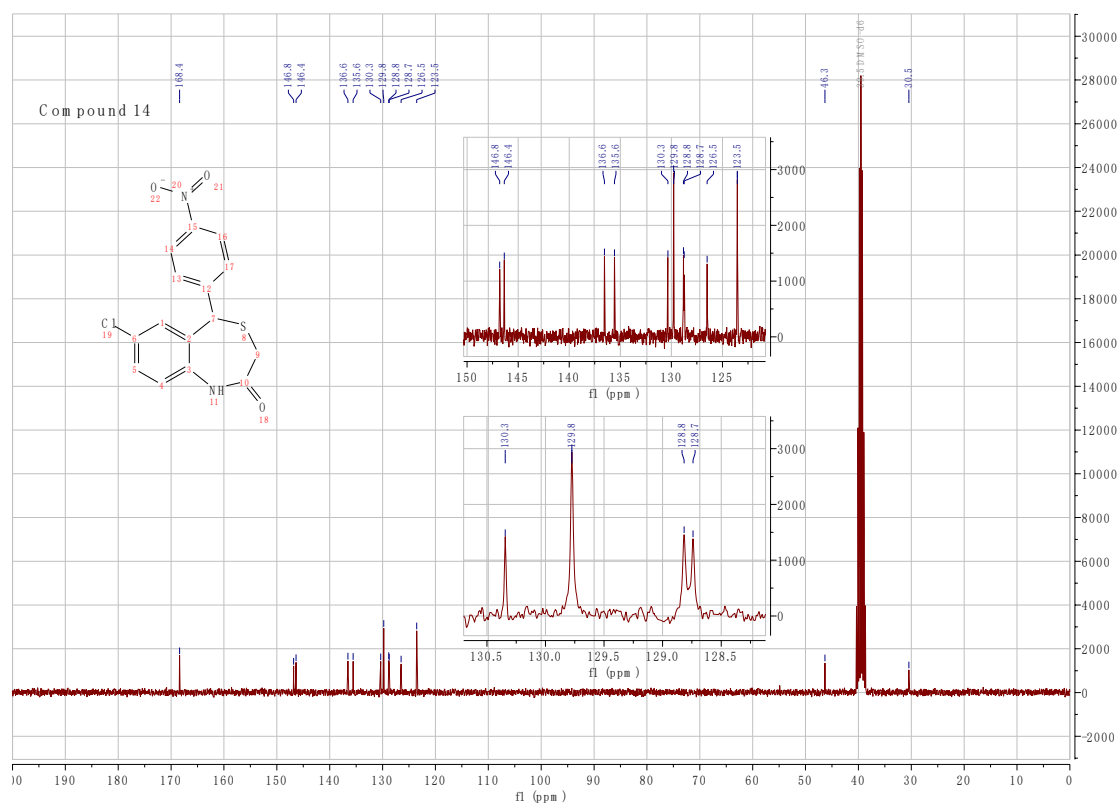
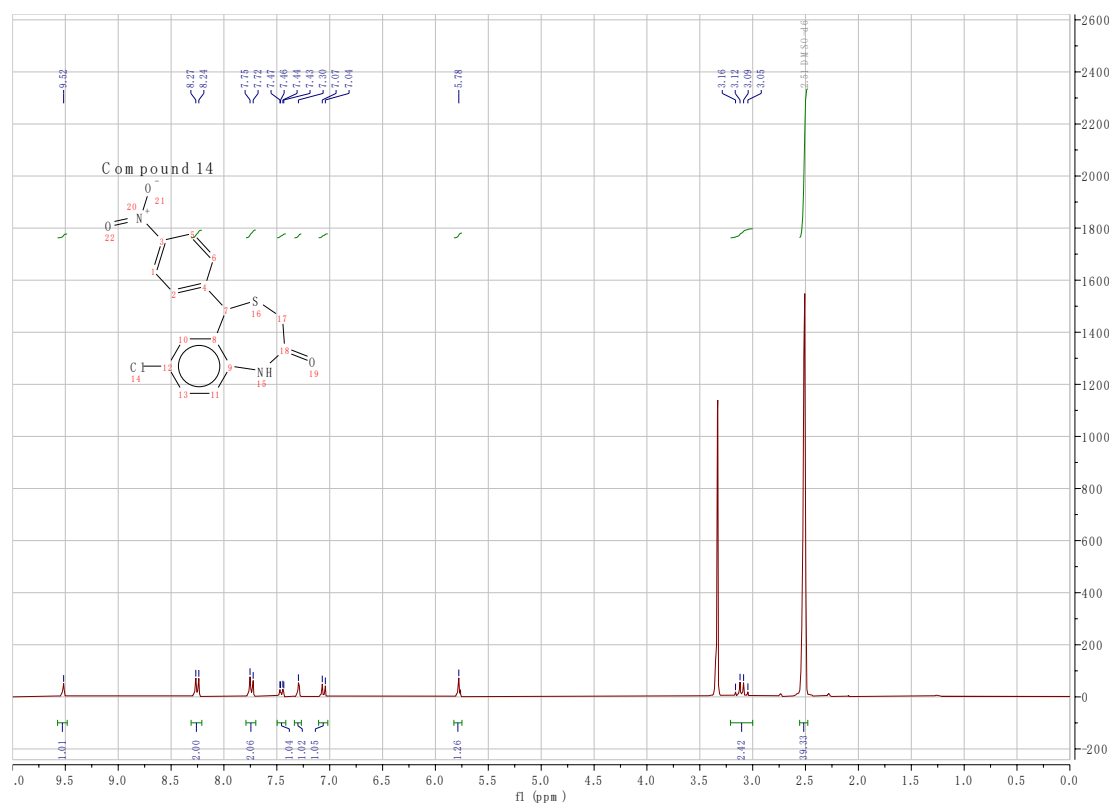


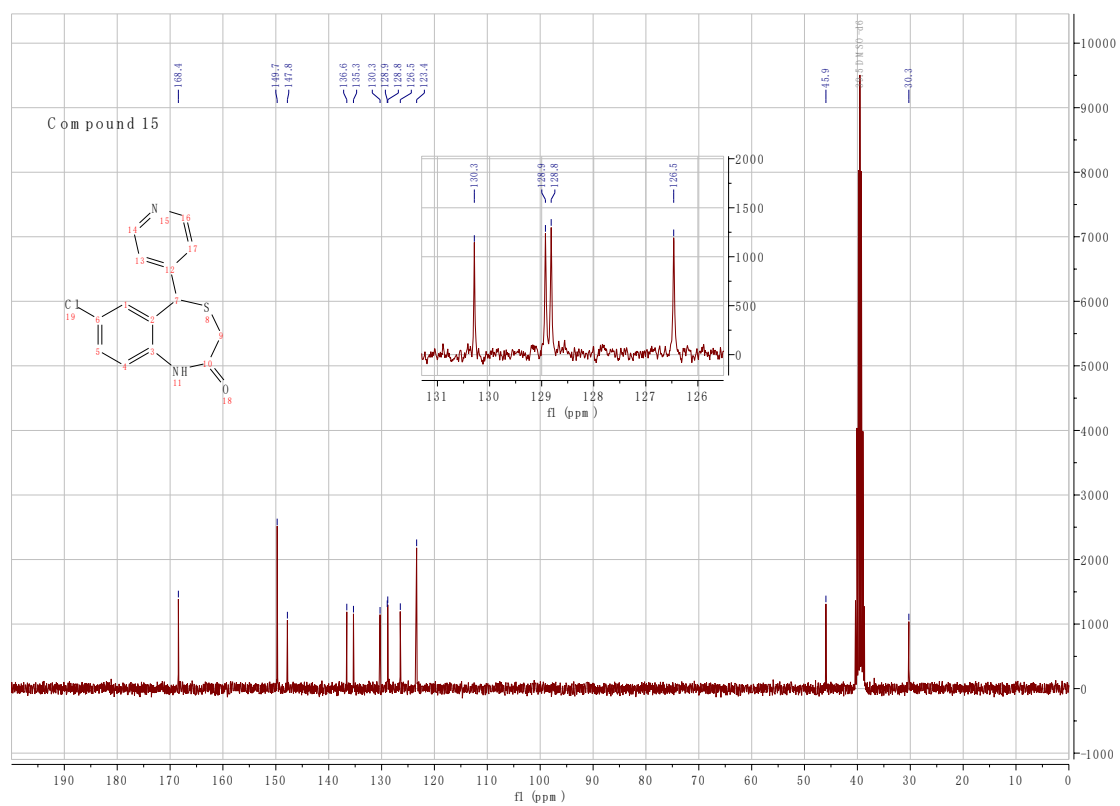
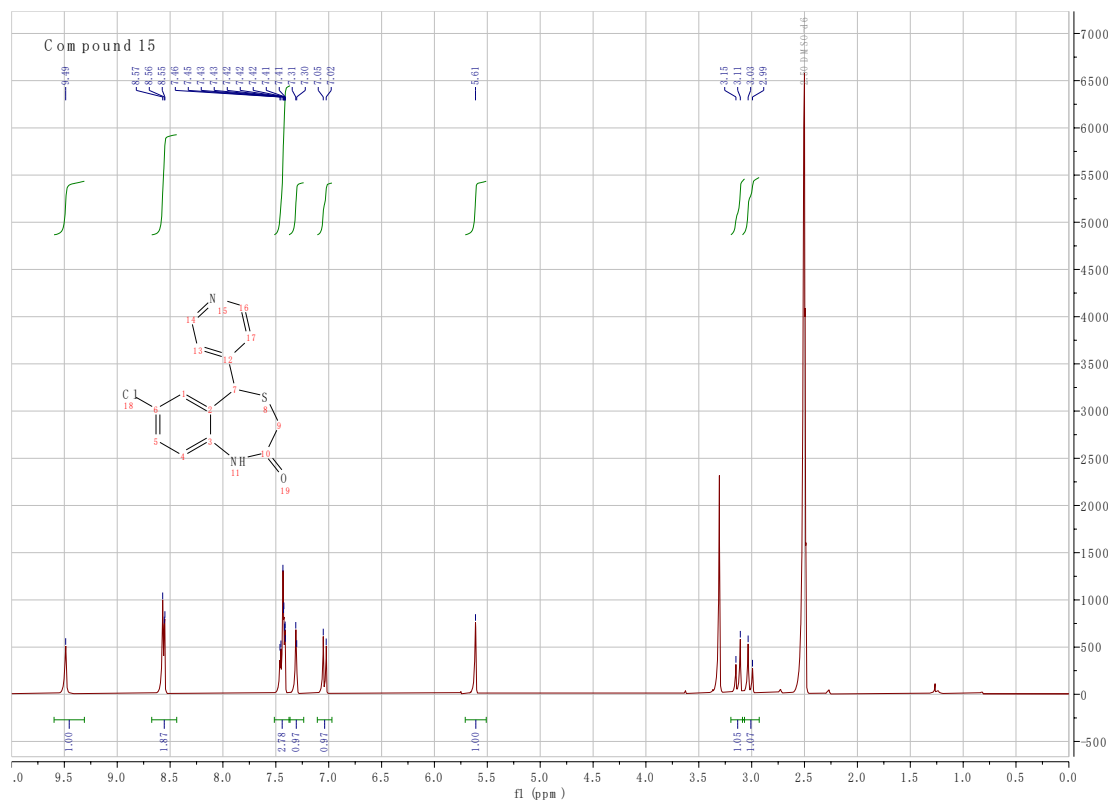
Similarly to what has been described [4] with slight modifications, 4-aminopyridine (2 g, 20.42 mmol) was dissolved in dry CH₂Cl₂ (30 mL), and triethylamine (3.6 mL, 2.6 g, 25.69 mmol) was slowly added under Ar. The mixture was cooled down to –20 °C and pivaloyl chloride (2.84 mL, 2.78 g, 23.06 mmol) was injected dropwise, allowing the reaction to stir at –20 °C for 75 min, monitoring reaction evolution by TLC. Reaction was terminated by addition of water (30 mL) and washed with saturated NaHCO₃ (3 × 50 mL). Combined organic layer was dried using anhydrous Na₂SO₄, filtered, and the solvent evaporated under vacuum. The crude was purified by automatized flash chromatography (Biotage Isolare One) using ethyl acetate/*n*-hexane mixtures as eluent to give compound **34** (2.2 g, 65%) was obtained as a white solid, possessing spectral data according with the literature [4].

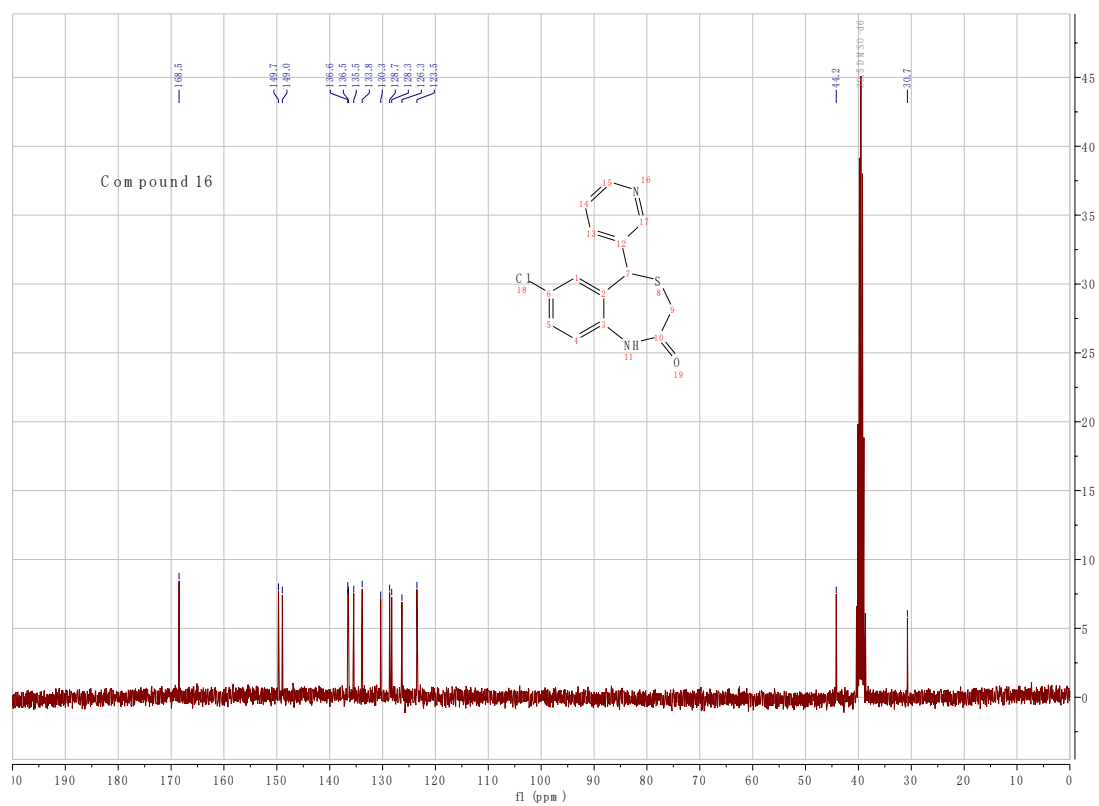
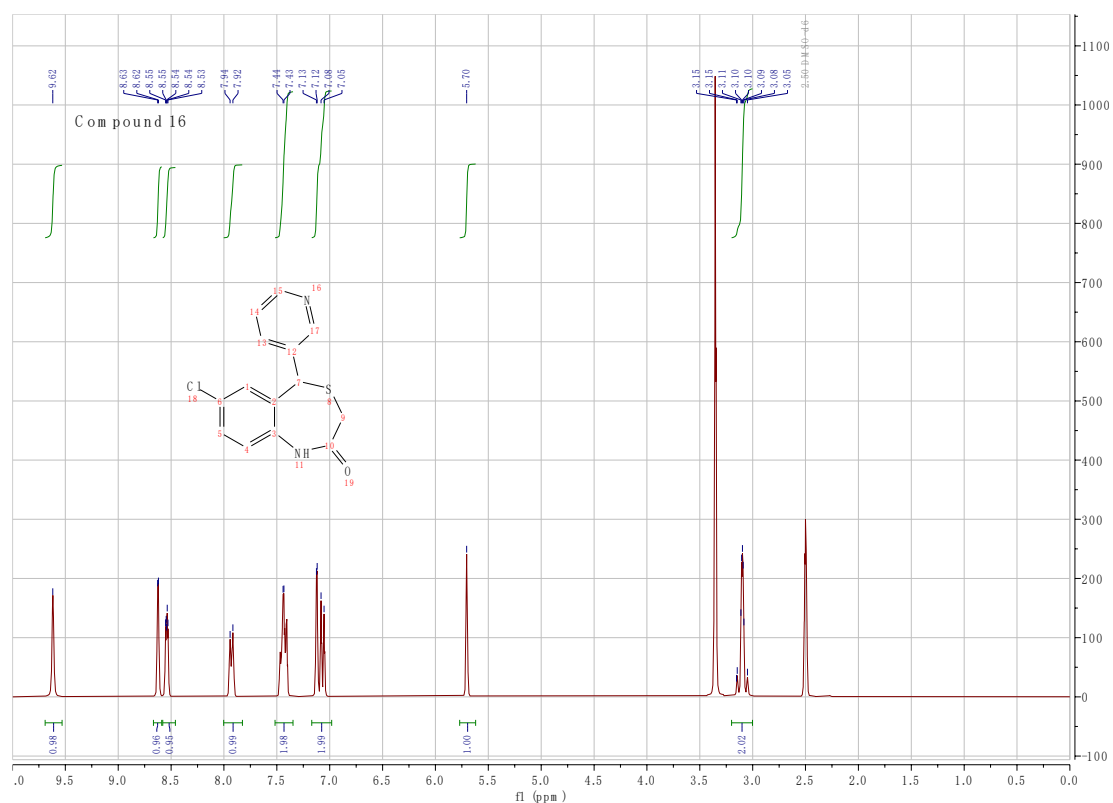
Spectral data

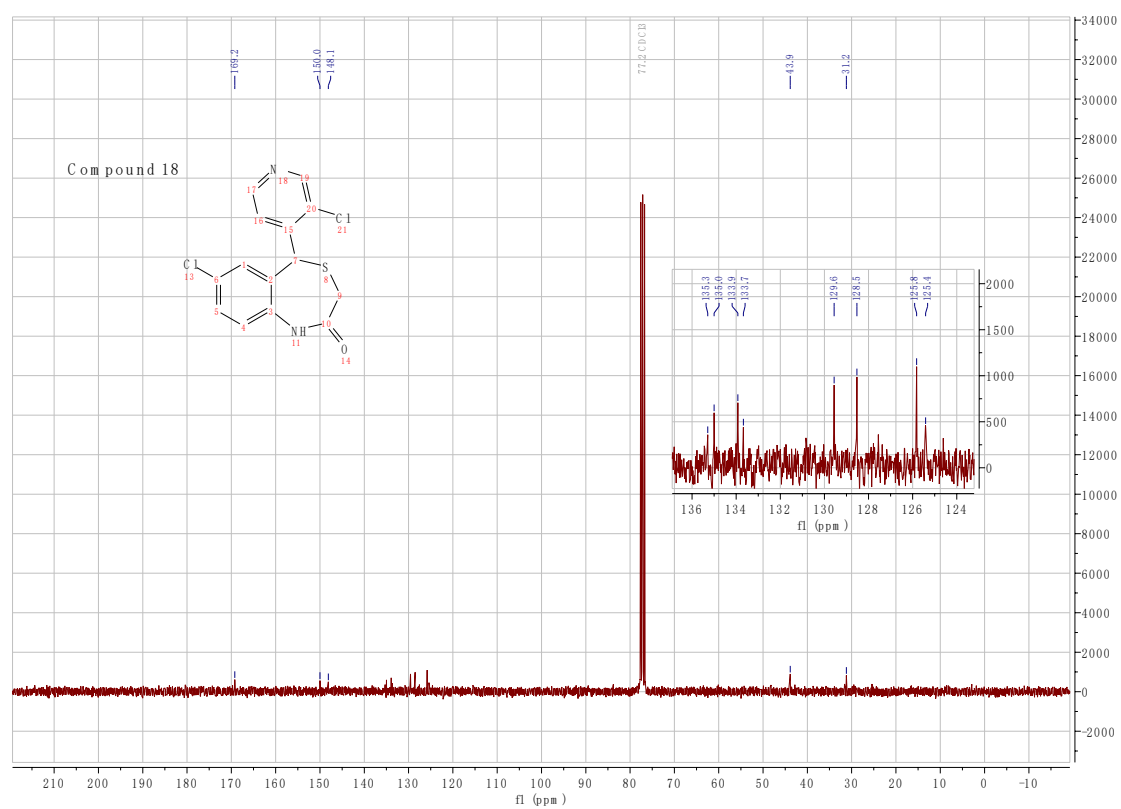
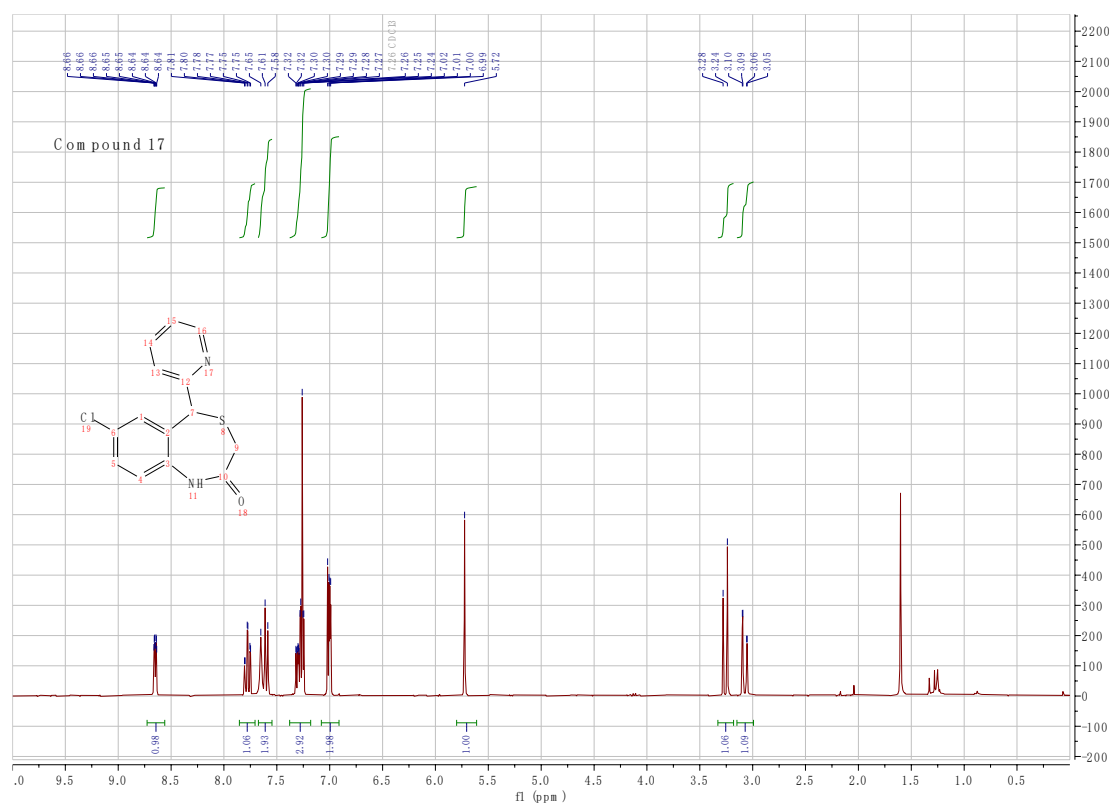


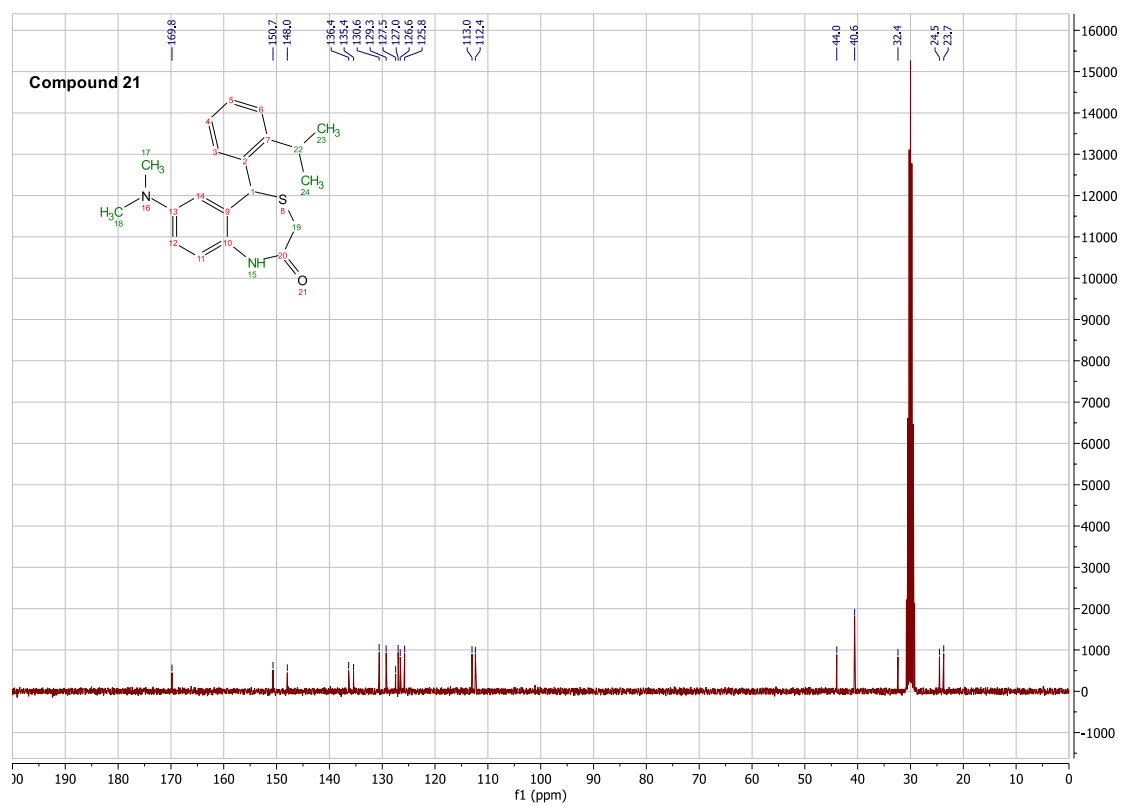
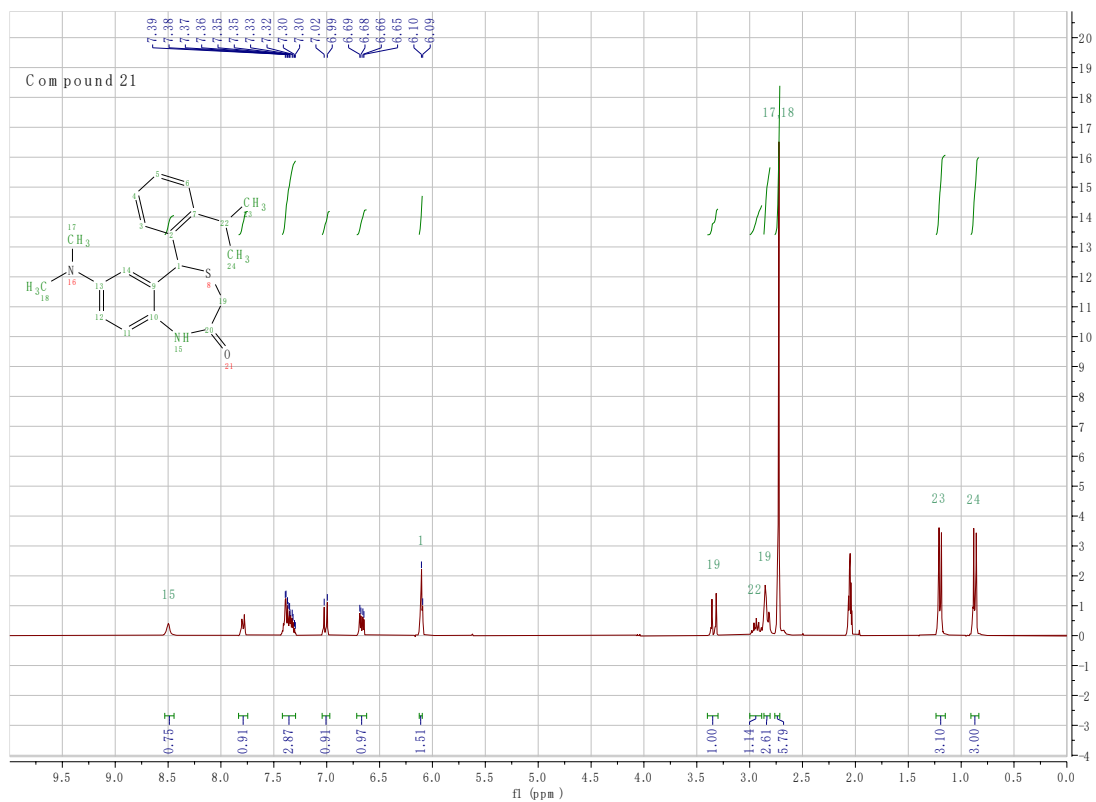


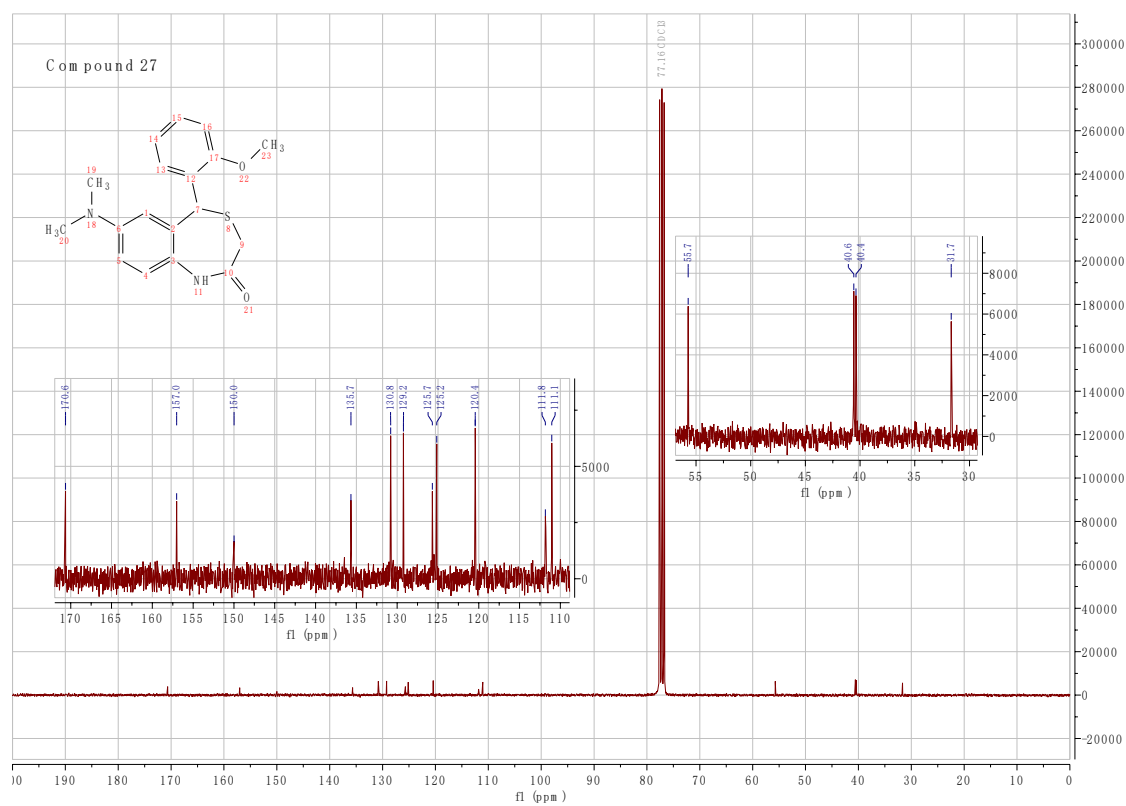
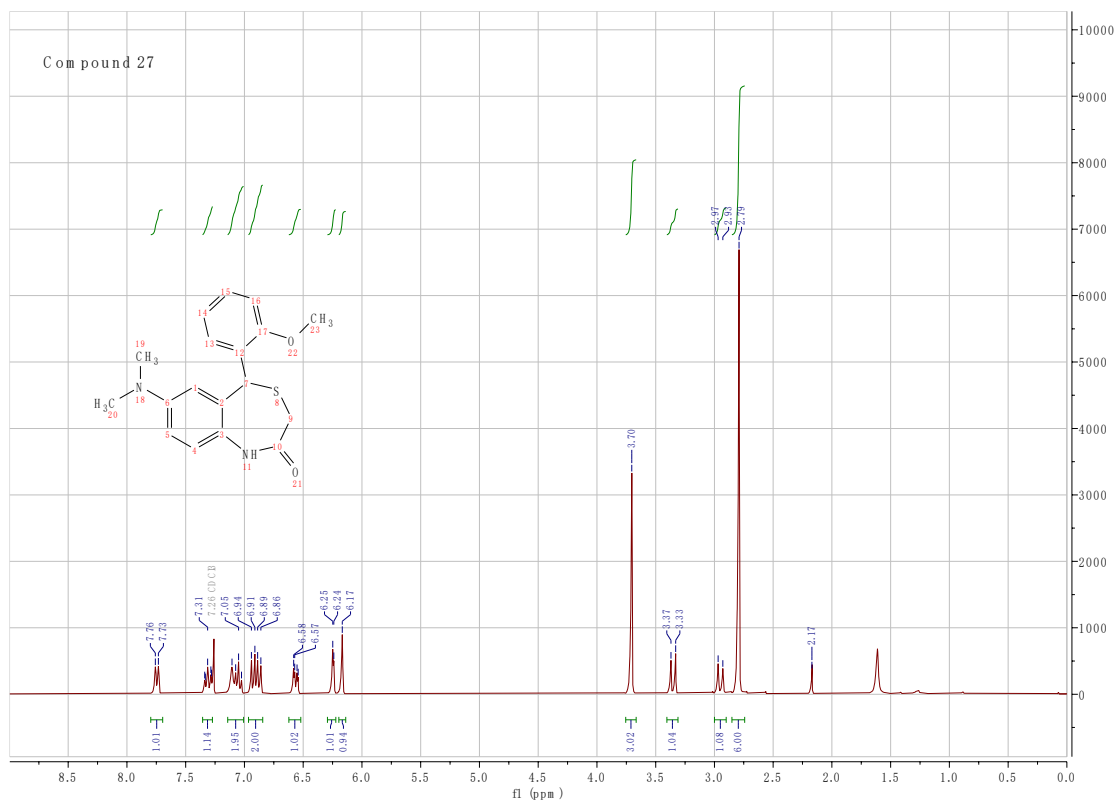


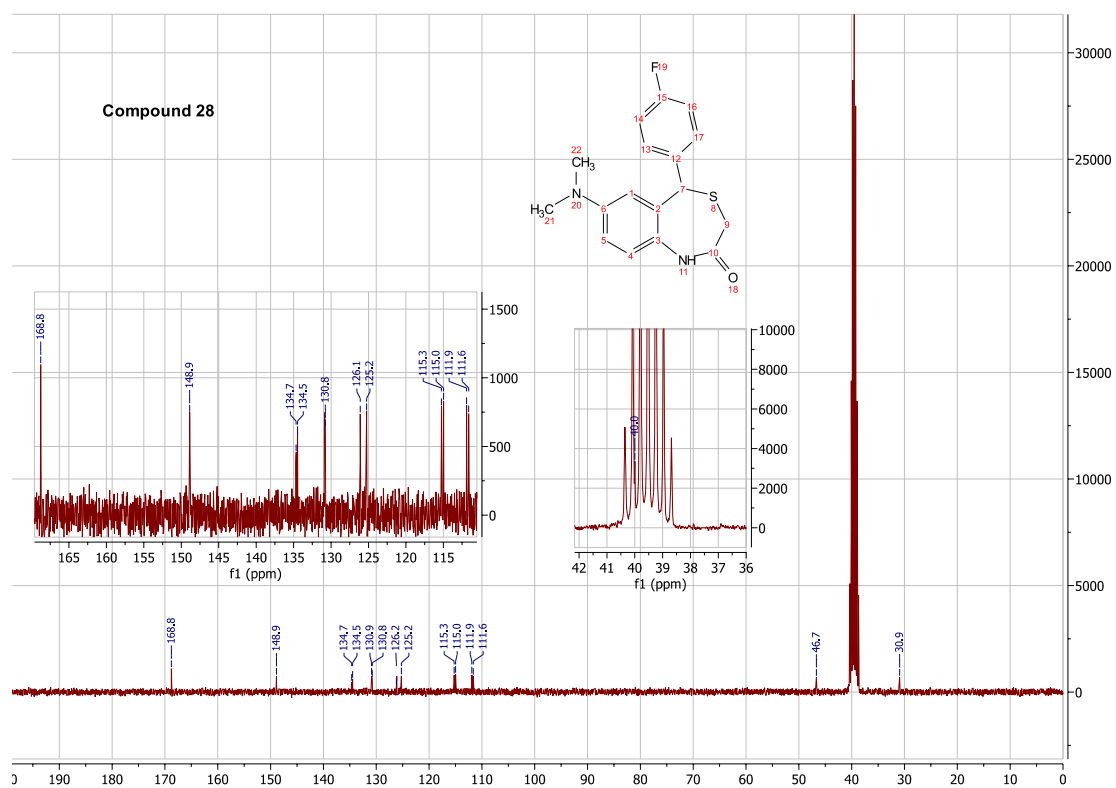
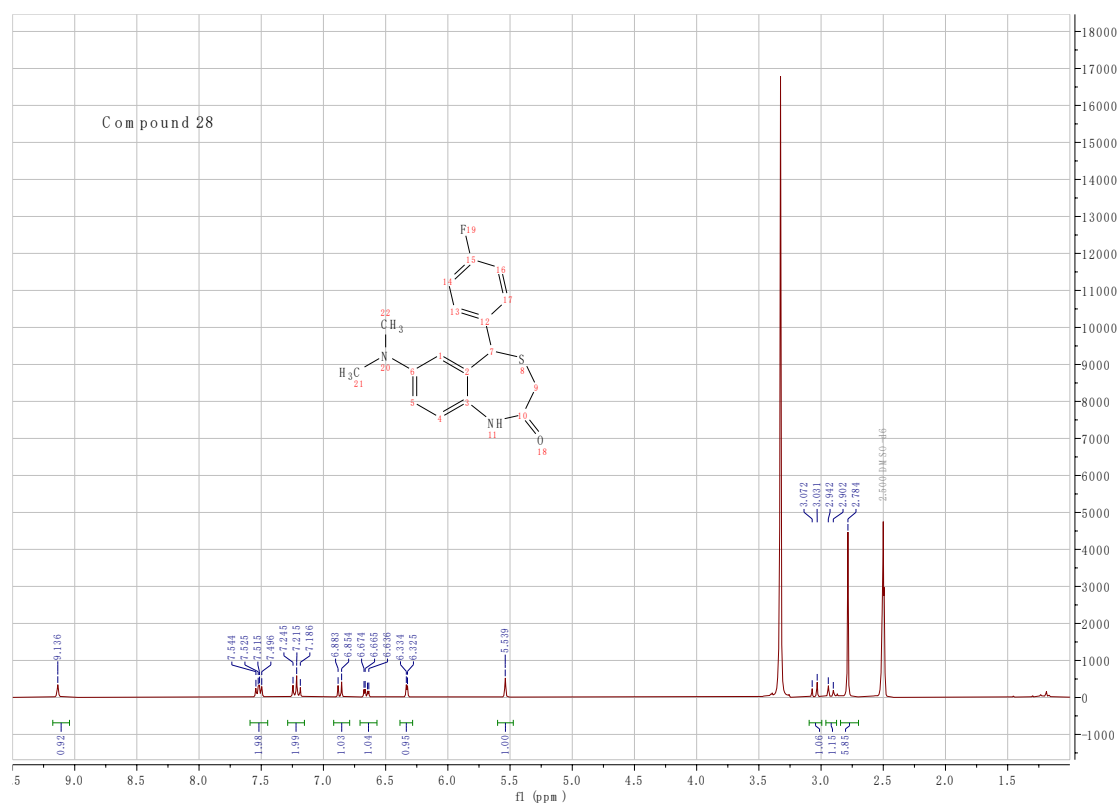


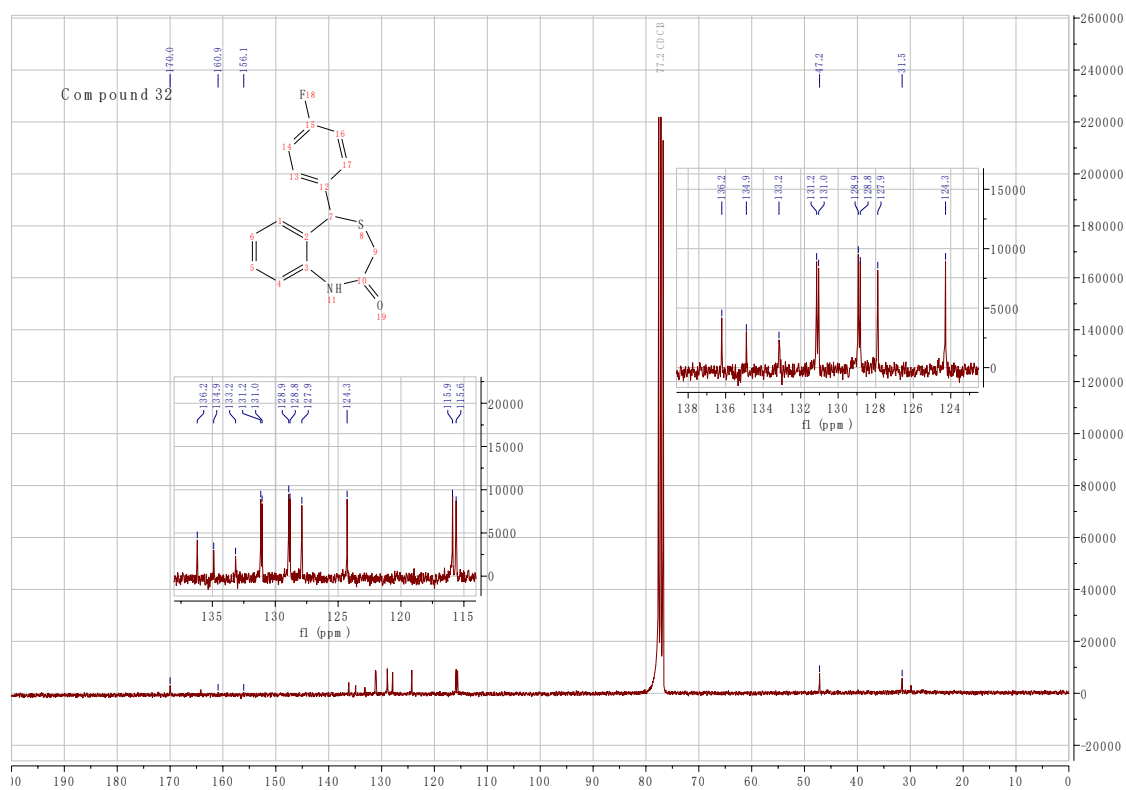
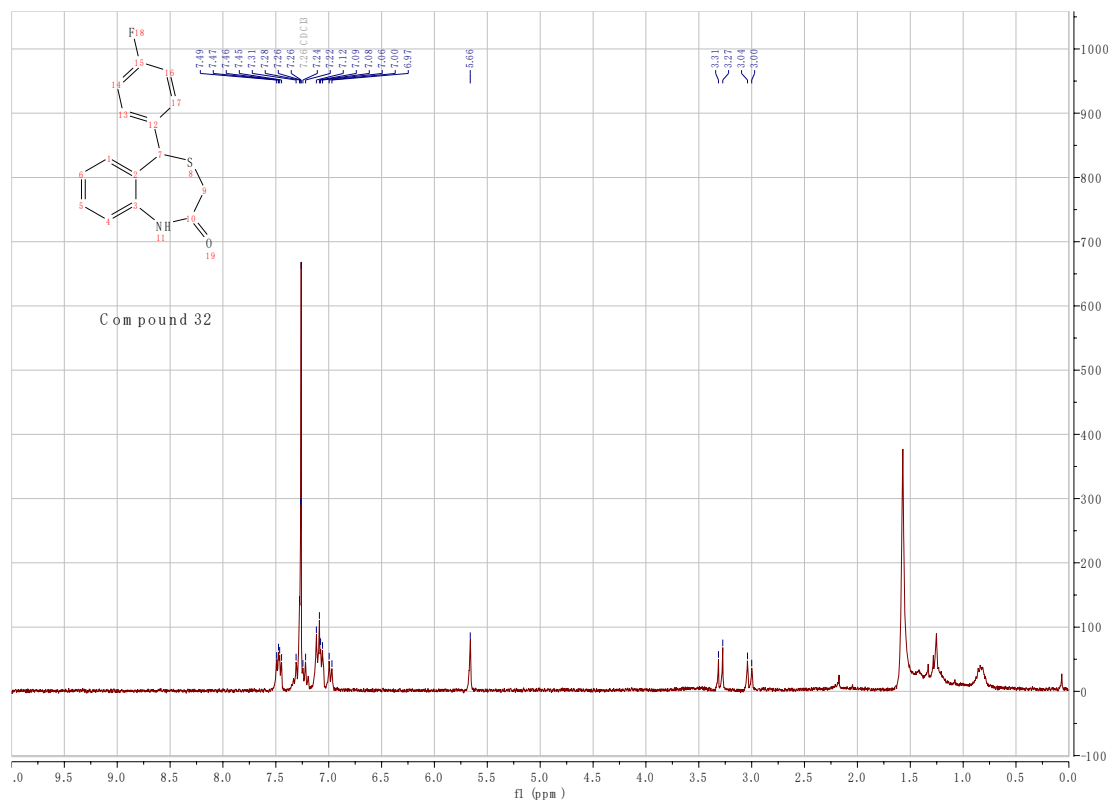


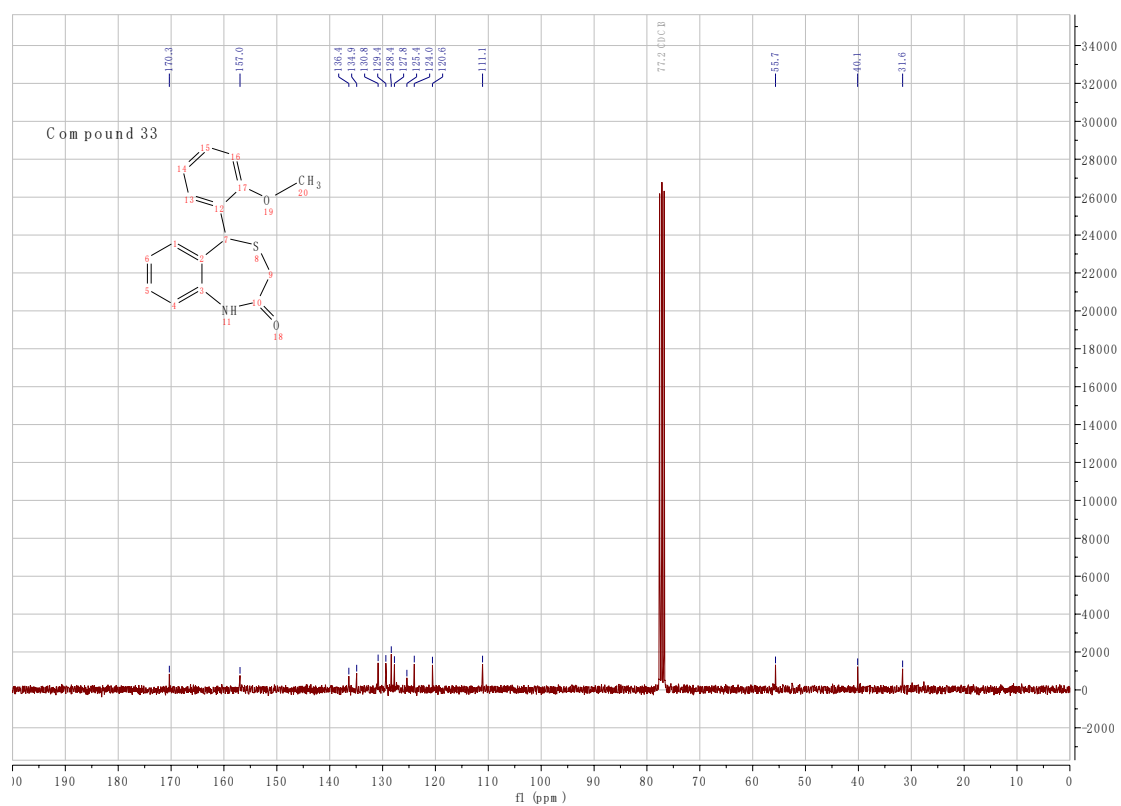
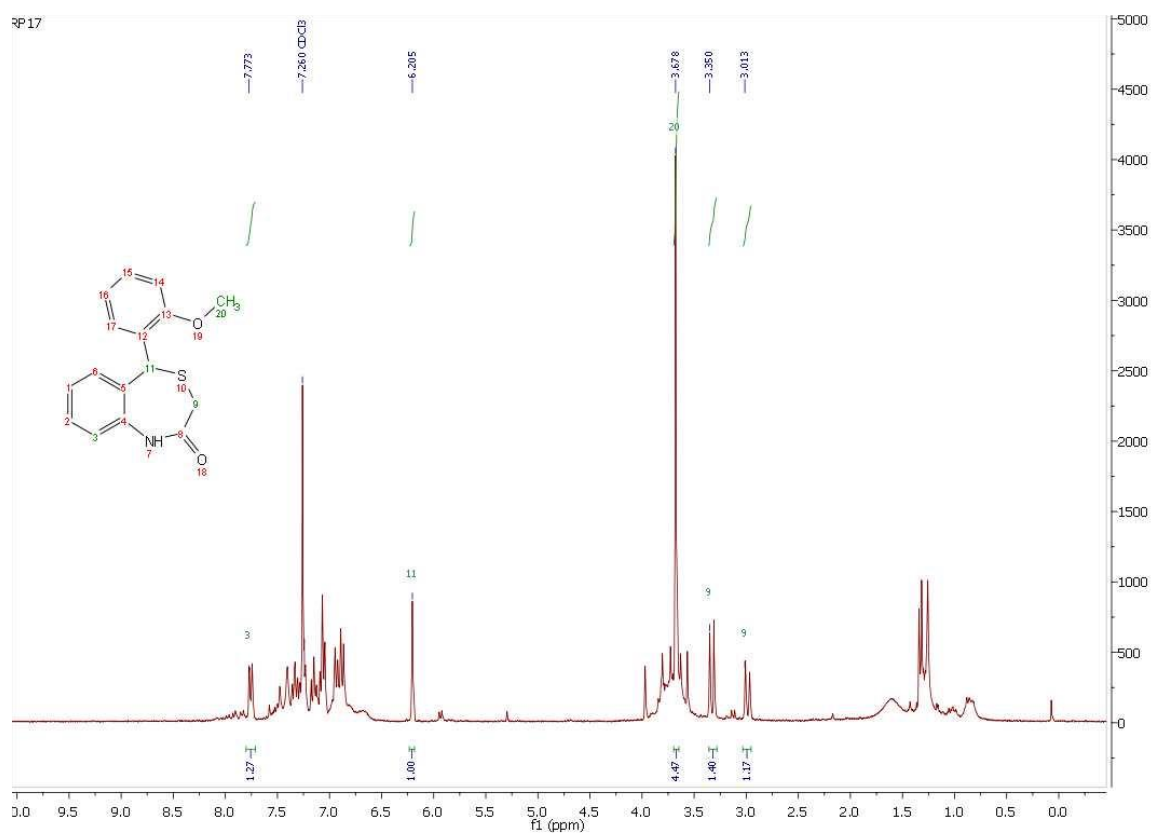












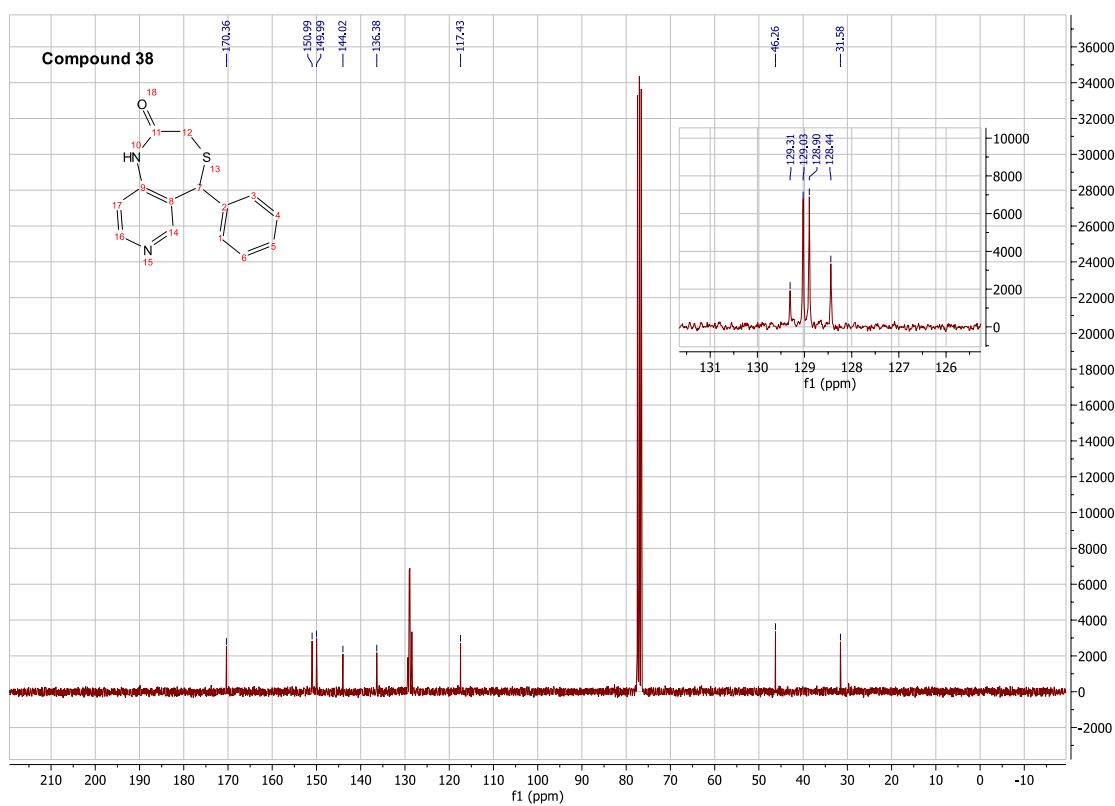
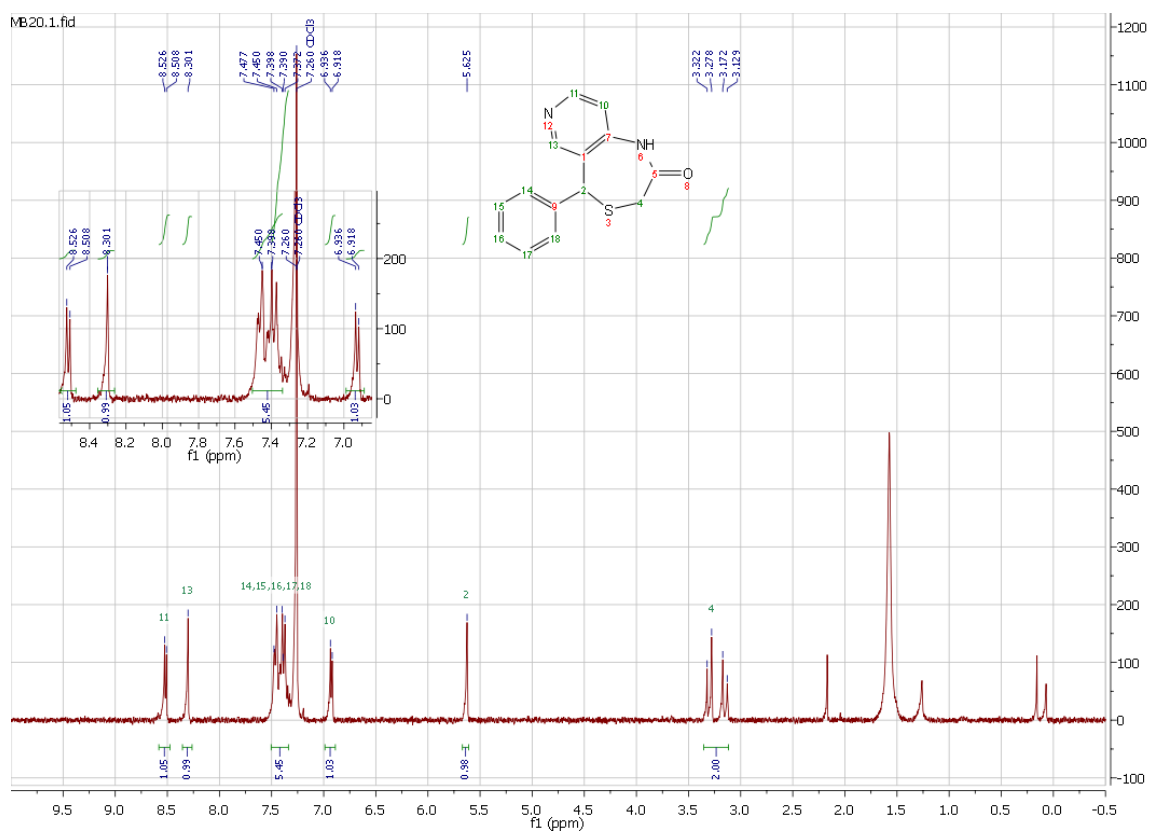
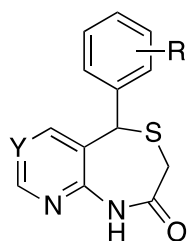
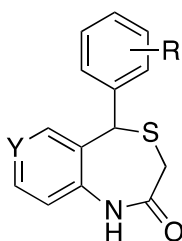


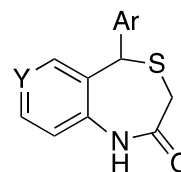
Table S1. Prediction of log P and other Lipinski's parameters.



ITH12662



12–14, 21, 26–28, 32, 33, 38



15–18

COMPOUNDS	Y	R	AR	DON. ^A	ACC. ^B	PSA (Å ²) ^C	CLOG P
CGP37157	C–Cl	2-Cl	-	1	2	54.4	4.36
ITH12575	C–Cl	2- ⁱ Pr	-	1	2	54.4	5.00
12	C–Cl	4- ⁱ Pr	-	1	2	54.4	5.00
13	C–Cl	3- ⁱ Pr	-	1	2	54.4	5.00
14	C–Cl	4-NO ₂	-	1	4	100.2	3.70
21	C–NMe ₂	2- ⁱ Pr	-	1	3	57.6	4.51
26	C–NMe ₂	2-Cl	-	1	3	57.6	3.86
27	C–NMe ₂	2-OMe	-	1	4	66.9	3.10
28	C–NMe ₂	4-F	-	1	3	57.6	3.40
32	C–H	4-F	-	1	2	54.4	3.30
33	C–H	2-OMe	-	1	3	63.6	2.99
38	N	H	-	1	3	67.3	1.93
15	C–Cl	H	4-y	1	3	67.3	2.54
16	C–Cl	H	3-y	1	3	67.3	2.54
17	C–Cl	H	2-y	1	3	67.3	2.67
18	C–Cl	2-Cl	4-y	1	3	67.3	3.14
ITH12662	C–Cl	4-F	-	1	3	67.3	3.28

^a Number of H-bonds-donating groups; ^b Number of H-bonds-accepting groups; ^c Polar surface Area (according to the Hopkins-modified Lipinski's rules of five, PSA of drug-like compounds should be below 150).

References and notes

- [1] Han, C.; Buchwald, S.L. Negishi Coupling of Secondary Alkylzinc Halides with Aryl Bromides and Chlorides. *J Am Chem Soc* **2009**, *131*, 7532-7533.
- [2] Ma, F.; Xie, X.; Zhang, L.; Peng, Z.; Ding, L.; Fu, L.; Zhang, Z. Palladium-Catalyzed Amidation of Aryl Halides using 2-Dialkylphosphino-2'-Alkoxy-1,1'-Binaphthyl as Ligands. *J Org Chem* **2012**, *77*, 5279-5285.
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- [4] Turner, J.A. Regiospecific Electrophilic Substitution of Aminopyridines: Ortho Lithiation of 2-, 3-, and 4-(Pivaloylamino)Pyridines. *J. Org. Chem.* **1983**, *48*, 3401-3408.