Excited State Lifetimes of Sulfur-Substituted DNA and RNA Monomers Probed Using the Femtosecond Fluorescence Up-Conversion Technique

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Supplementary Material

Figure S1. Normalized ground-state absorption spectra of the investigated thiobase monomers in phosphate buffer saline (PBS) solution at pH 7.4 and in acetonitrile (ACN).



Figure S2. Emission spectra of the investigated thiobase monomers in phosphate buffer saline (PBS) solution at pH 7.4 or in acetonitrile (ACN). Excitation was performed at 268 nm in airsaturated conditions. The overtone at ca. 536 ± 15 nm has been masked in each plot. Other experimental and instrument conditions are reported in Section 4.2. The additional emission bands observed for 4tUra, 4tUrd, 2,4dtUra, 2,4dtThy around 550 nm are assigned to phosphorescence emission because their intensity increases/decreases when the solutions are purged with N₂/O₂ (see also Figure S3 below). Dual emission (i.e., fluorescence + phosphorescence) has been reported previously for 4tThd in different solvents.¹⁻²



Figure S3. Emission spectra for 2,4dtUra and 4tUra in phosphate buffer saline (PBS) solution at pH 7.4 at selected excitation wavelengths and under N₂-, air-, or O₂-saturated conditions. The overtone at ca. 536 ± 15 nm has been masked from the top panels. We note that a recent study has assigned the emission band at ca. 550 nm in 2,4dtUra to fluorescence from a high-energy ${}^{1}\pi_{s}\pi^{*}$ state,³ in contrast to our assignment to phosphorescence emission from the ${}^{3}\pi_{s}\pi^{*}$ state for both 4tUra and 2,4dtUra.



Figure S4. Normalized isotropic fluorescence up conversion signals for selected thiobases in acetonitrile (ACN) or in phosphate buffer saline (PBS) solution at pH 7.4 at the emission wavelength of 425 nm upon excitation at 362 nm.



Figure S5. Normalized isotropic fluorescence up conversion signals for selected thiobases in acetonitrile at the emission wavelengths of 450, 525, or 532 nm upon excitation at 362 nm.

$\lambda_{\text{exc}} = 362 \text{ nm}$ IRF: 300 fs	$\lambda_{em} = 425 \text{ nm}$ Isotropic (ps)	$\lambda_{em} = 450 \text{ nm}$ Isotropic (ps)	$\lambda_{em} = 525 \text{ nm}$ Isotropic (ps)	$\lambda_{em} = 532 \text{ nm}$ Isotropic (ps)
4tUra	0.84 ± 0.03	0.89 ± 0.04	0.9 ± 0.2	-
4tUrd	0.76 ± 0.02	0.76 ± 0.02	0.93 ± 0.10	-
4tThd	0.50 ± 0.03	0.51 ± 0.02	0.55 ± 0.05	-
2,4dtThy	0.16 ± 0.01	0.16 ± 0.02	0.29 ± 0.08	-
2,4dtUra	0.23 ± 0.01	-	-	0.26 ± 0.05
2,4dtUrd	0.18 ± 0.01	-	-	0.20 ± 0.07

Table S1. Fluorescence lifetimes of selected thiobases in acetonitrile.

Supplementary References

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