

*Supporting Informations for*

# Luminescent sensor based on Ln(III) ternary complex for NAD(P)H detection

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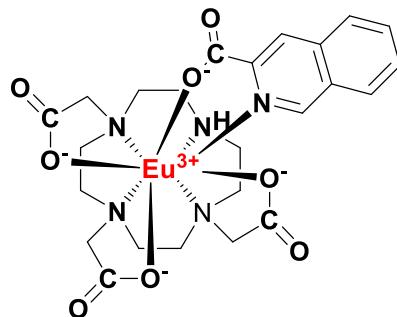
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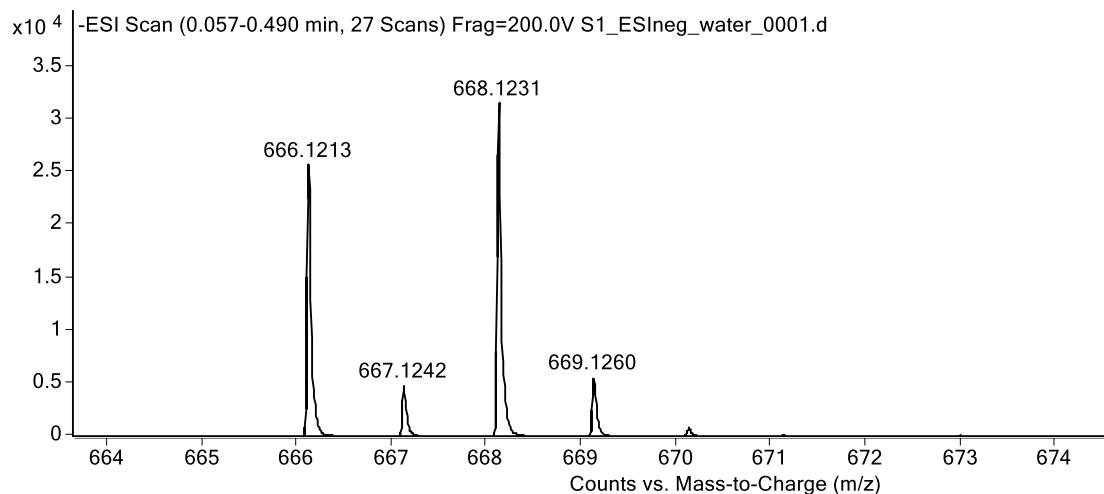
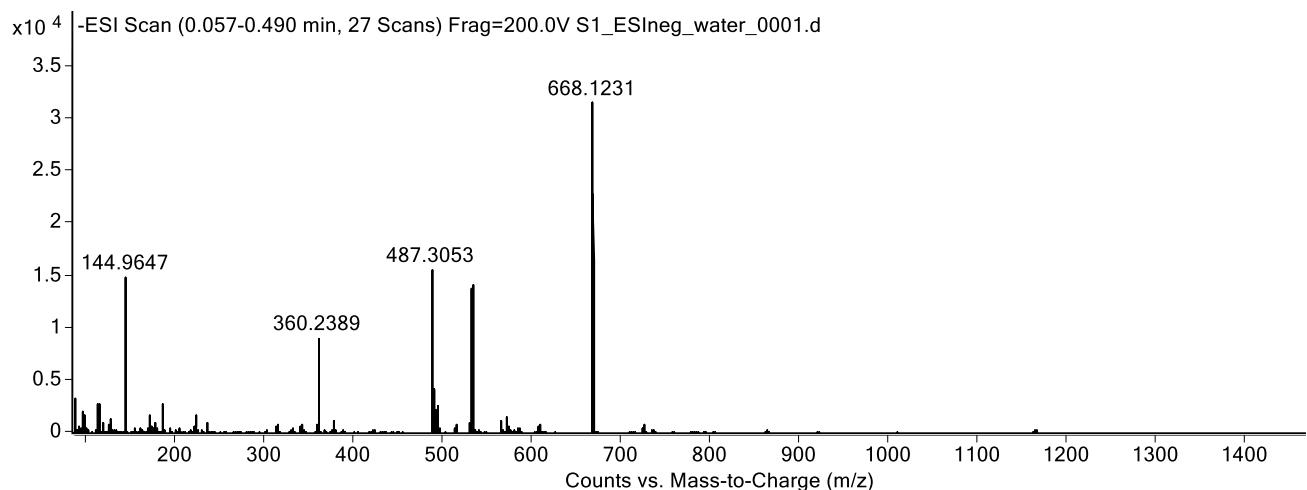
expected mass:  $[\text{Eu}(\text{DO3A})(\text{IQCA})]^+ = \underline{\underline{668.1236}}$

observed mass:  $[\text{Eu}(\text{DO3A})(\text{IQCA})]^+ = \underline{\underline{668.1231}}$

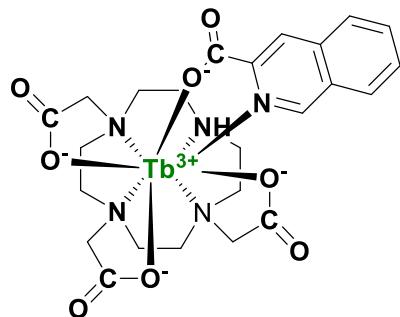
mass accuracy = 0.7 ppm

$\text{C}_{24}\text{H}_{29}\text{O}_8\text{N}_5\text{Eu}$

**exact mass: 668.1230**



**Fig. S1.** The HR-MS for ternary  $[\text{Eu}(\text{DO3A})(\text{IQCA})]^+$  complex



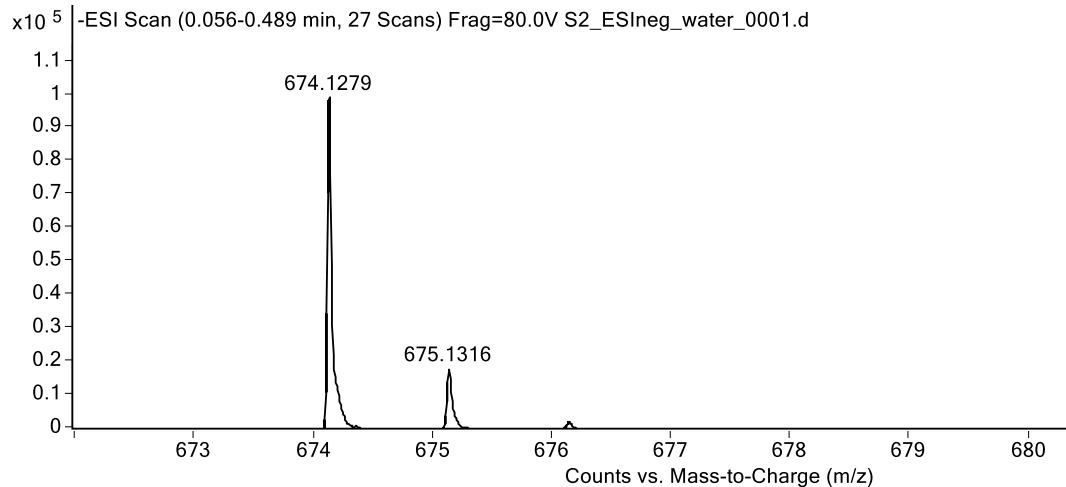
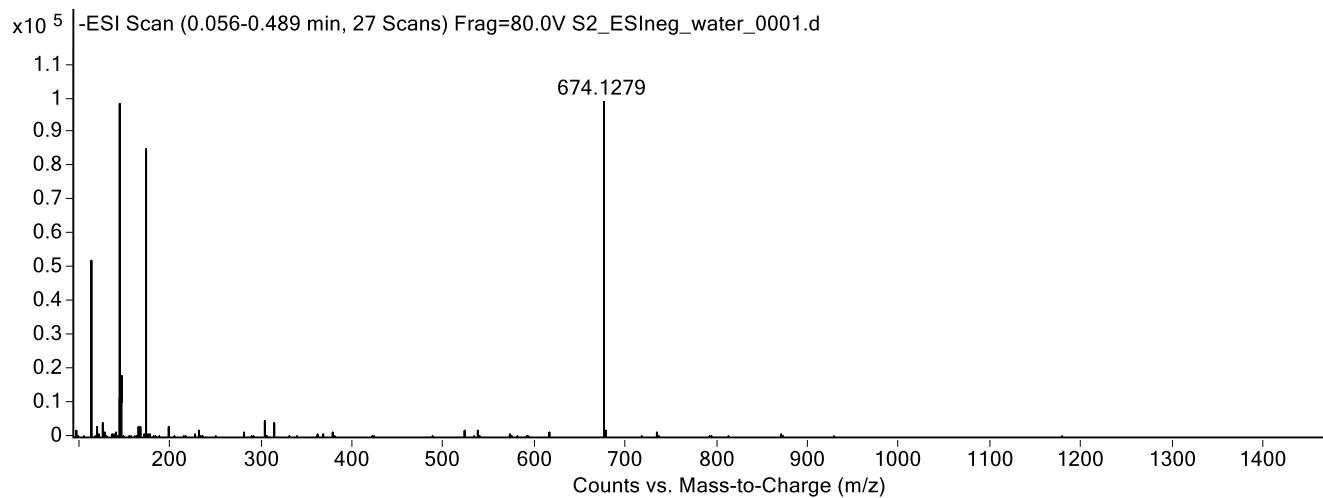
expected mass:  $[\text{Tb}(\text{DO3A})(\text{IQCA})]^- = \underline{\underline{674.1275}}$

observed mass:  $[\text{Tb}(\text{DO3A})(\text{IQCA})]^- = \underline{\underline{674.1279}}$

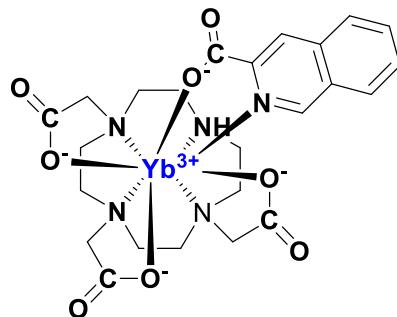
mass accuracy = - 0.6 ppm

$\text{C}_{24}\text{H}_{29}\text{O}_8\text{N}_5\text{Tb}$

exact mass: **674.1270**



**Fig. S2.** The HR-MS for ternary  $[\text{Tb}(\text{DO3A})(\text{IQCA})]^-$  complex



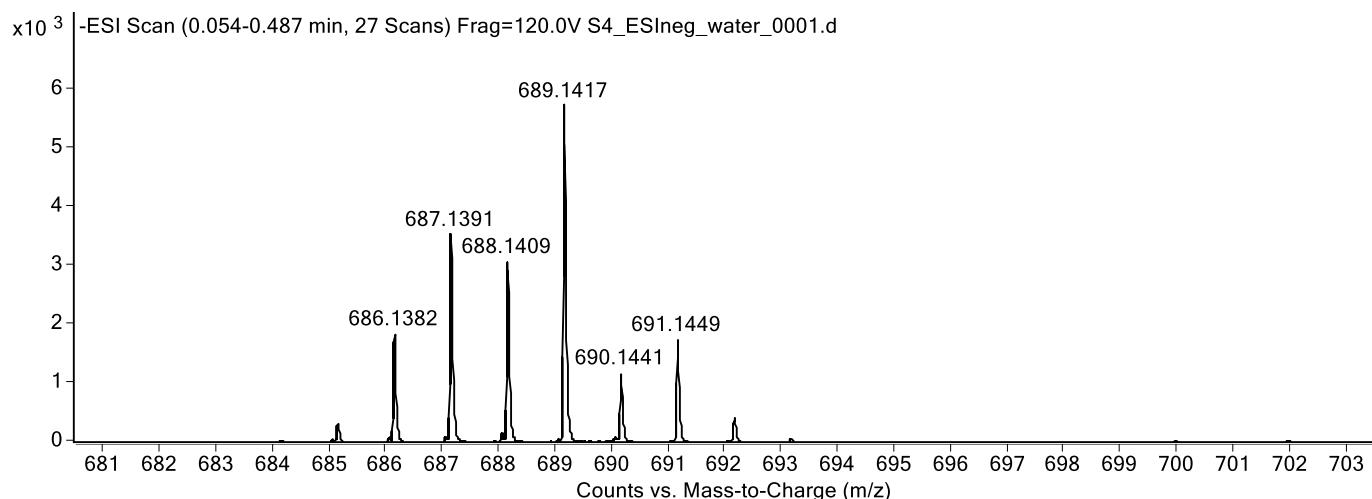
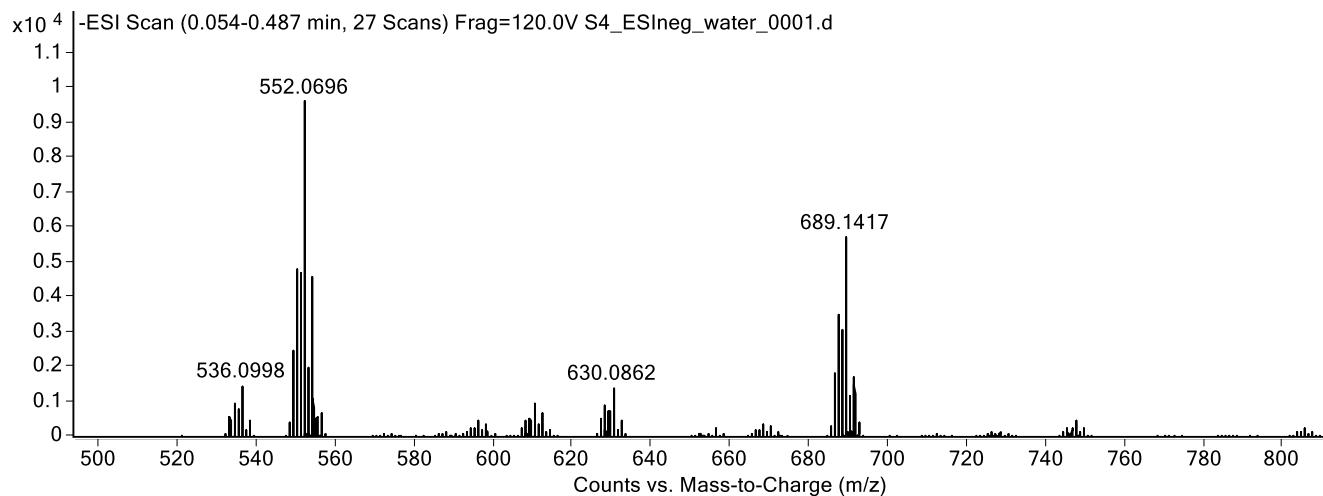
expected mass:  $[\text{Yb}(\text{DO3A})(\text{IQCA})]^- = \underline{\underline{689.1414}}$

observed mass:  $[\text{Yb}(\text{DO3A})(\text{IQCA})]^- = \underline{\underline{689.1417}}$

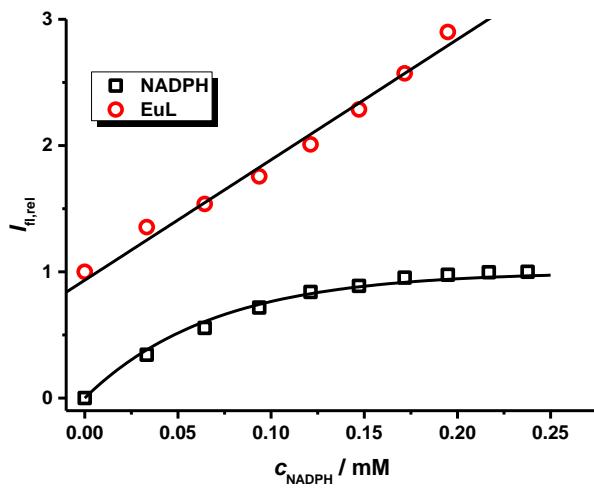
mass accuracy = - 0.4 ppm

$\text{C}_{24}\text{H}_{29}\text{O}_8\text{N}_5\text{Yb}$

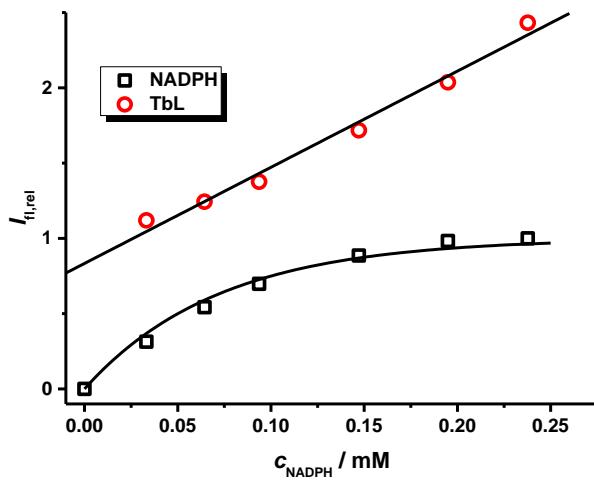
**exact mass: 689.1405**



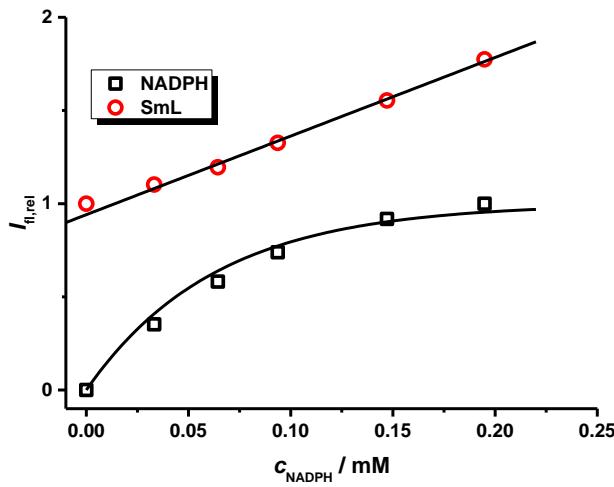
**Fig. S3.** The HR-MS for ternary  $[\text{Yb}(\text{DO3A})(\text{IQCA})]^-$  complex



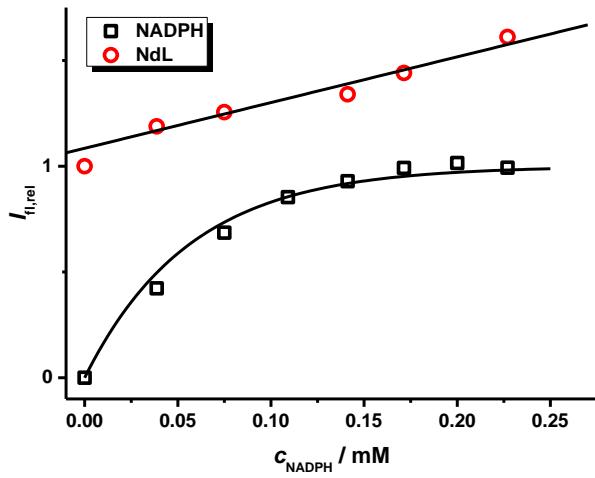
**Fig. S4.** The quenching effect of NADPH on luminescence intensity of ternary Eu(III) complex with DO3A macrocyclic and IQCA ligands ( $\lambda_{\text{exc}} = 325 \text{ nm}$ , pH = 7.5 (40 mM HEPES),  $c_{\text{Ln}} \sim 0.2 \text{ mM}$ ,  $c_{\text{DO3A}} \sim 0.3 \text{ mM}$ ,  $c_{\text{IQCA}} \sim 0.3 \text{ mM}$ ).  $\lambda_{\text{em,NADPH}} = 460 \text{ nm}$ ,  $\lambda_{\text{em,EuL}} = 618 \text{ nm}$



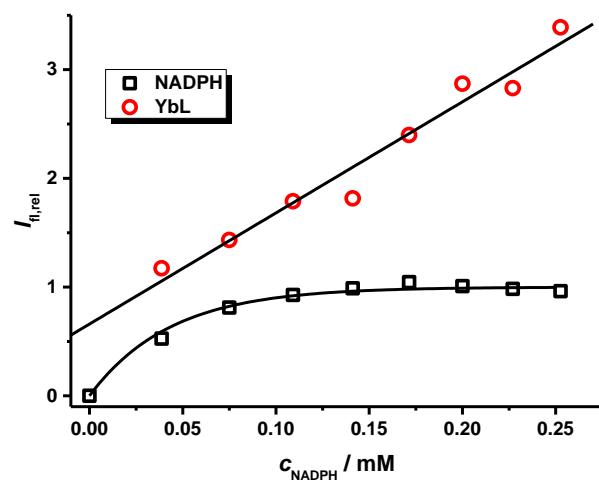
**Fig. S5.** The quenching effect of NADPH on luminescence intensity of ternary Tb(III) complex with DO3A macrocyclic and IQCA ligands ( $\lambda_{\text{exc}} = 325 \text{ nm}$ , pH = 7.5 (40 mM HEPES),  $c_{\text{Ln}} \sim 0.2 \text{ mM}$ ,  $c_{\text{DO3A}} \sim 0.3 \text{ mM}$ ,  $c_{\text{IQCA}} \sim 0.3 \text{ mM}$ ).  $\lambda_{\text{em,NADPH}} = 460 \text{ nm}$ ,  $\lambda_{\text{em,TbL}} = 545 \text{ nm}$ .



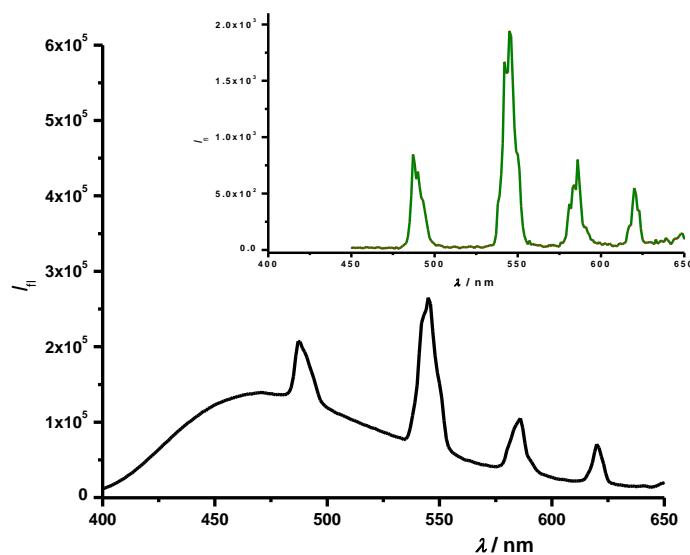
**Fig. S6.** The quenching effect of NADPH on luminescence intensity of ternary Sm(III) complex with DO3A macrocyclic and IQCA ligands ( $\lambda_{\text{exc}} = 325 \text{ nm}$ , pH = 7.5 (40 mM HEPES),  $c_{\text{Ln}} \sim 0.2 \text{ mM}$ ,  $c_{\text{DO3A}} \sim 0.3 \text{ mM}$ ,  $c_{\text{IQCA}} \sim 0.3 \text{ mM}$ ).  $\lambda_{\text{em,NADPH}} = 460 \text{ nm}$ ,  $\lambda_{\text{em,SmL}} = 597 \text{ nm}$



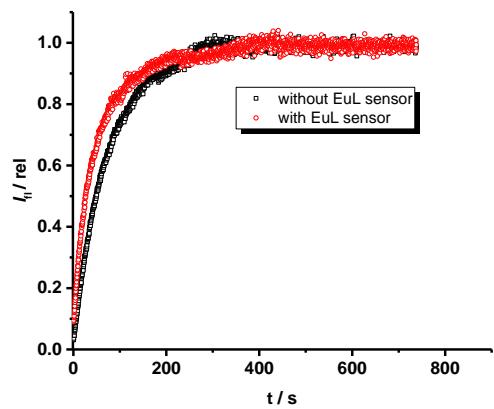
**Fig. S7.** The quenching effect of NADPH on luminescence intensity of ternary Nd(III) complex with DO3A macrocyclic and IQCA ligands ( $\lambda_{\text{exc}} = 325 \text{ nm}$ , pH = 7.5 (40 mM HEPES),  $c_{\text{Ln}} \sim 0.2 \text{ mM}$ ,  $c_{\text{DO3A}} \sim 0.35 \text{ mM}$ ,  $c_{\text{IQCA}} \sim 0.2 \text{ mM}$ ).  $\lambda_{\text{em,NADPH}} = 460 \text{ nm}$ ,  $\lambda_{\text{em,NdL}} = 880 \text{ nm}$ .



**Fig. S8.** The quenching effect of NADPH on luminescence intensity of ternary Yb(III) complex with DO3A macrocyclic and IQCA ligands ( $\lambda_{\text{exc}} = 325 \text{ nm}$ , pH = 7.5 (40 mM HEPES),  $c_{\text{Ln}} \sim 0.2 \text{ mM}$ ,  $c_{\text{DO3A}} \sim 0.35 \text{ mM}$ ,  $c_{\text{IQCA}} \sim 0.2 \text{ mM}$ ).  $\lambda_{\text{em,NADPH}} = 460 \text{ nm}$ ,  $\lambda_{\text{em,YbL}} = 980 \text{ nm}$



**Fig. S9.** The emission spectra of ternary Tb(III) complex with DO3A macrocyclic and IQCA ligands in presence of NADPH compound ( $\lambda_{\text{exc}} = 325 \text{ nm}$ , pH = 7.5 (40 mM HEPES),  $c_{\text{Tb}} \sim 0.2 \text{ mM}$ ,  $c_{\text{DO3A}} \sim 0.35 \text{ mM}$ ,  $c_{\text{IQCA}} \sim 0.2 \text{ mM}$ ). The corrected spectra have not been gated (lower picture) while upper in inset were gated by  $\Delta t = 50 \text{ ms}$ .



**Fig. S10** The example of time trace of fluorescence intensity for enzymatic reaction under the same experimental conditions ( $\text{pH} = 8.0$ ,  $c_{\text{EtOH}} = 0.1 \text{ M}$ ,  $c_{\text{EuLZ}} = 0.1 \text{ mM}$ ,  $c_{\text{NAD}} = 0.1 \text{ mM}$ ,  $\lambda_{\text{exc}} = 325 \text{ nm}$ ,  $\lambda_{\text{em}} = 460 \text{ nm}$ )